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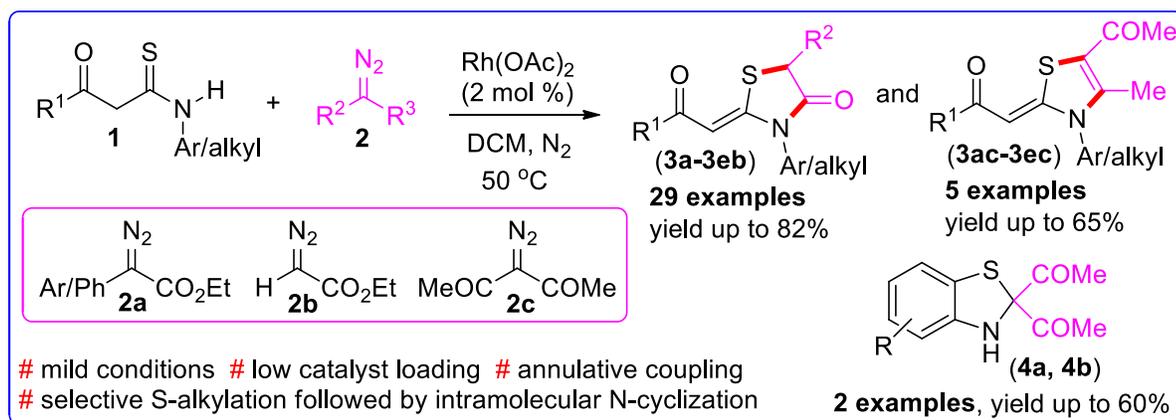
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Rhodium(II)-Catalyzed Annulative Coupling of β -Ketothioamides with α -Diazo compounds: Access to Highly Functionalized Thiazolidin-4-ones and Thiazolines

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ABSTRACT: An operationally simple and efficient one-pot protocol for the synthesis of highly functionalized thiazolidin-4-ones and thiazolines has been devised via Rh(OAc)_2 -catalyzed annulative coupling of β -ketothioamides with diazo compounds under mild reaction conditions for the first time. This double functionalization of diazo compounds proceeds via selective S-alkylation followed by intramolecular N-cyclization enabling the formation of C–S and C–N bonds at moderate temperature. Notably, the products possess *Z*-stereochemistry with regard to the exocyclic C=C double bond at the 2-position of the ring. Further, the synthetic utility of the strategy has been revealed to access 2,3-dihydrobenzo[*d*]thiazoles. Remarkably, atom-economy and tolerance of a wide range of functional groups are added characteristics to this strategy.

INTRODUCTION

The utility of nitrogen/sulfur heterocycles has become prevalent in medicinal and agricultural industries. Thiazolidine-4-one/thiazoline derivatives represent an important class of heterocycles, which are ubiquitous core structural motifs in many natural products, pharmaceuticals and functional materials.¹ The thiazolidine derivatives exhibit extensive pharmacological property such as anticancer,^{2a} antihistaminic,^{2b} anti-HIV^{2c} activities, and selective COX-2 inhibitors.^{2d} Thiazolidin-4-one scaffold has been considered as a magic moiety because it possesses almost all types of biological properties such as antifungal, antitubercular, antimicrobial, antioxidant, antibacterial, cytotoxic, anti-inflammatory, analgesic and anti-YFV (yellow fever virus) activities.³ Among the various available reports for the synthesis of thiazolidine-4-ones, the most common and general approach involves the reaction between aldehyde, amine (or Schiff's base) and mercaptoacetic acid.^{4a} Bolognese and co-workers^{4b} synthesized the thiazolidine 4-ones via microwave technique, while Lingampalle *et al.*^{4c} constructed thiazolidines utilizing ionic liquid. Jagodzinski *et al.*^{5a} and our group^{5b,c} synthesized the thiazolidin-4-ones by the reaction of β -ketothioamides with different coupling partners. Despite the aforesaid approaches for the synthesis of thiazolidin-4-ones, most of them suffer from one or more limitations such as use of excess base, lack of generality, pre-functionalization of starting materials, limited functional group tolerability, and poor yield. Therefore, it is still challenging and highly desirable to explore an operationally simple, efficient and widely applicable general approach for the synthesis of thiazolidin-4-one derivatives.

Over the past several decades, the decomposition of diazo compounds generating carbenes or metal-stabilized carbenoids provided synthetic chemist a valuable tool for the selective construction of both C-C and C-heteroatom bonds.⁶ Diazo compounds are amphiphilic, and the negatively polarized diazo carbon atom is nucleophilic, while the metal carbene species generated from a diazo compound has an electron-deficient carbene center.⁷ The utility of α -diazocarbonyl compounds as versatile synthon has been established in various areas of organic synthesis such as Wolff rearrangements,⁸ cyclopropanation reactions,^{9a-c} benzannulation,^{9d} cycloaddition chemistry,¹⁰ and C-H/heteroatom-H inser-

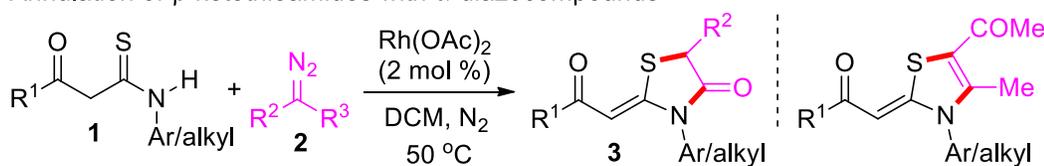
tion/functionalization reactions.¹¹ Furthermore, diazo compounds are widely used for the synthesis of heterocyclic frameworks.¹² Very recently, Wang *et al.*^{13a} reported sulfhydryl-directed iridium-catalyzed C–H/diazo coupling with naphthalene-1-thiols to construct naphtho fused thiopyrans. Koenigs *et al.*^{13b} reported iron-catalysed S-alkylation of thioether with diazoalkane. Rh(OAc)₂ introduced by Teyssie and co-workers¹⁴ in the early 1970s, is still the catalyst of choice for metal carbenoid formation.¹⁵

In contrast, the coupling reactions involved metal carbenes and thiocarbonyl (C=S) group are not often explored. In this regard, Zeng^{16a} and Moody^{16b} developed Rh-catalyzed intermolecular coupling reaction of carbonyl group (C=O) with diazo compounds, in which nucleophilic attack of carbonyl oxygen of amide to Rh-carbenes generated carbonyl ylides, which upon further intramolecular cyclization afforded oxazole derivatives. In addition to above approaches the reaction of diazo compounds with thioamides for the synthesis of heterocyclic motifs have been reported but most of them suffer from generality, poor yield of desired products and vigorous conditions.^{16c,d} To date, the intermolecular coupling reaction between diazo compounds and thiocarbonyl C=S of β -ketothioamides has not been explored, while this type of transformation could provide an alternative approach for the gathering of structurally diverse molecules.

The success of carbenoid insertion/annulation strategy relies on the judicious choice of starting material, which requires two orthogonal reactivity types: a site for insertion and a compatible electrophile, which will not undergo reaction with the metal carbenoid. One such simple substrate is β -ketothioamide that has been well-documented as an intriguing synthon to access valuable heterocyclic scaffolds.¹⁷ Site-selectivity in β -ketothioamide functionalizations has been achieved by leveraging several factors.¹⁸ Against this background and our experience towards the chemistry of β -ketothioamides,¹⁸ it was envisioned that the marriage of the carbenoid insertion/annulation with β -ketothioamides would provide access to interesting and potentially useful heterocyclic frameworks. To this end, herein we disclose the Rh(OAc)₂-catalyzed annulative coupling of β -ketothioamides with diazo compounds for the synthesis of highly functionalized thiazolidinones and thiazolines (Scheme 1).

Scheme 1. Synthesis of Thiazolidinones and Thiazolines

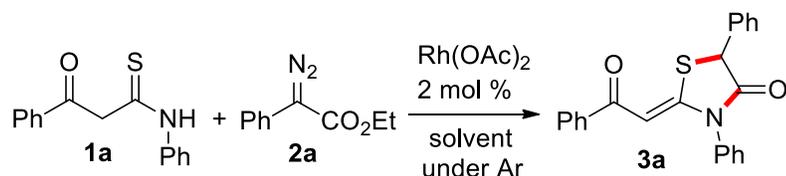
Annulation of β -ketothioamides with α -diazocompounds



RESULTS AND DISCUSSION

In general, most of the reactions in organic chemistry are reagent/catalyst-controlled, and their regio- and stereoselectivity are directed by the inherent nature of the reagent or catalyst. While in substrate-directed reactions, the selectivity is determined by the functional group on the substrate. In this perspective, one of the simple substrate is β -ketothioamide, which exerts amazing reactivity and emerged as novel multi-role-synthon for the construction of various heterocyclic motifs.^{17,18} In continuation of our ongoing research toward the development of powerful synthetic strategies to assemble different heterocyclic scaffolds employing β -ketothioamide as a reaction partner, we turned our attention to react β -ketothioamide with diazoalkane, which could generate hitherto unknown heterocyclic motif via one-pot cascade annulation. To validate our hypothesis, we commenced our study employing β -ketothioamide **1a** and diazoalkane **2a** as model substrates to examine various reaction parameters to optimize the reaction conditions. The results are summarized in Table 1.

Initially, the reaction of **1a** and **2a** in equimolar ratio in the presence of 2 mol % of $\text{Rh}(\text{OAc})_2$ in dichloromethane (DCM) at room temperature under inert atmosphere could provide only a trace of the desired product **3a** after 24 h (Table 1, entry 1). The above result encouraged us to continue our investigations for the formation of desired product **3a** in good yield. To this end, the test reaction was carried out at $50\text{ }^\circ\text{C}$, which afforded the desired product **3a** in 82% isolated yield (Table 1, entry 2). Further increasing the temperature not only decreased the yield, but also made the reaction messy (Table 1, entry 3).

Table 1. Optimization of Reaction Conditions^a

| entry | conditions | solvent | temp (°C) | yield (%) ^b 3a |
|-------|---|--------------------|--------------|-------------------------------------|
| 1 | as shown | DCM | 25 | Trace |
| 2 | as shown | DCM | 50 | 82 |
| 3 | as shown | DCM | 60 | 65 ^c |
| 4 | as shown | DCE | 50 | NR |
| 5 | as shown | CH ₃ CN | 50 | NR |
| 6 | as shown | DMSO | 50 | NR |
| 7 | as shown | DMF | 50 | NR |
| 8 | as shown | Dioxane | 50 | NR |
| 9 | [Rh(COD)Cl] ₂ | DCM | 50 | NR |
| 10 | [RhCl(PPh ₃) ₃] | DCM | 50 | 60 |
| 11 | CuCl | DCM | 50 | 65 |
| 12 | CuBr | DCM | 50 | 65 |
| 13 | Cu(OAc) ₂ | DCM | 50 | 60 |
| 14 | AgSbF ₆ | DCM | 50 | NR |
| 15 | No catalyst | DCM | 50 | NR |

^aUnless otherwise noted all the reaction were carried out using 0.5 mmol of each **1a** and **2a** in the presence of 2 mol % of catalyst in 5 mL of solvent. ^bIsolated yields. ^cComplex TLC pattern. NR = No reaction.

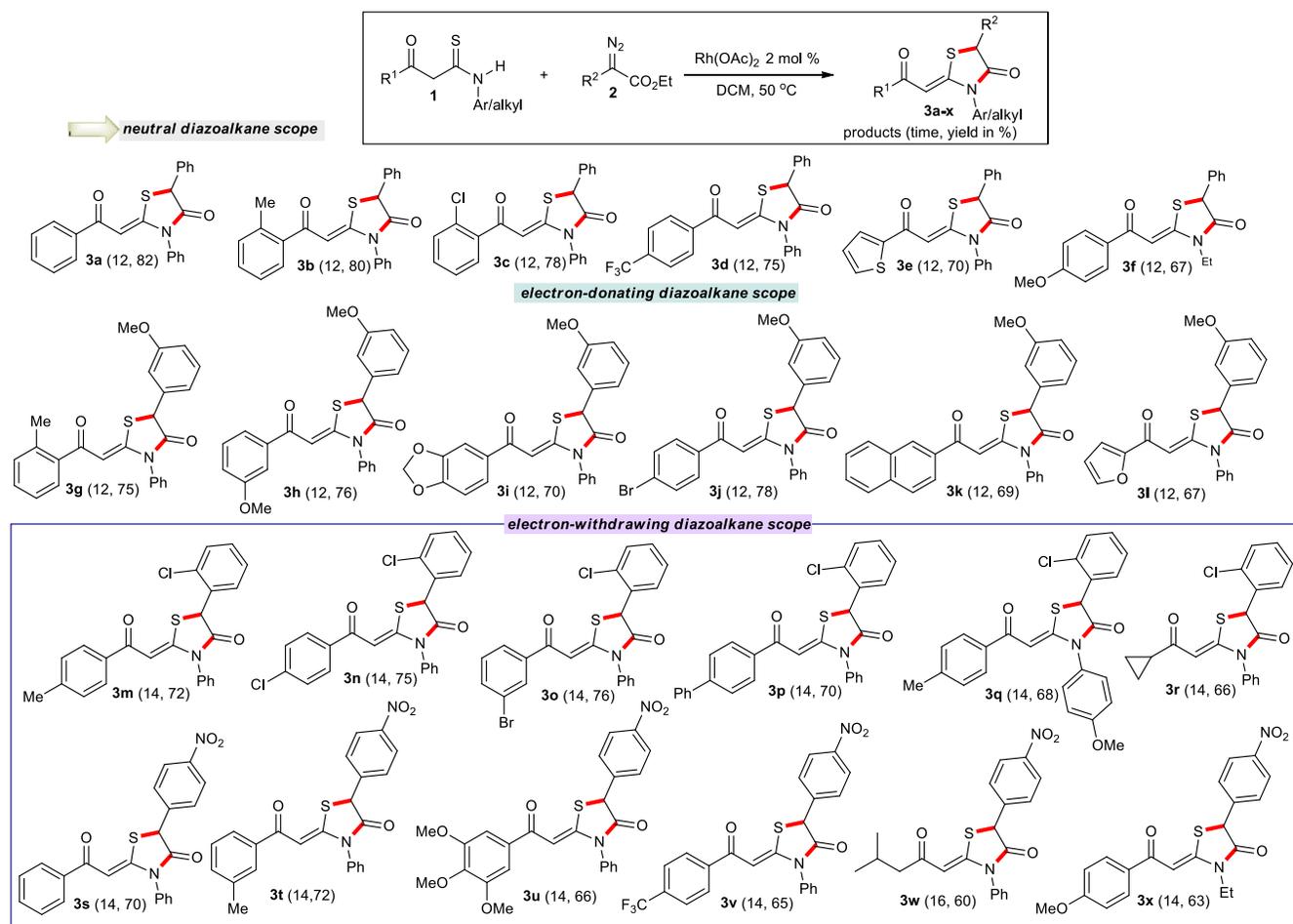
Inspired by the above result, we intended to see the effect of some other solvents. Thus, the test reaction is performed in various solvents such as dichloroethane (DCE), CH₃CN, DMSO, DMF and diox-

ane, but none of them did trigger the reaction (Table 1, entries 4-8). After optimizing DCM as a choice of solvent, next we screened some other rhodium catalysts such as Rh(COD)Cl]₂ and RhCl(PPh₃)₃. Notably, Rh(COD)Cl]₂ did not trigger the reaction, while RhCl(PPh₃)₃ triggered the reaction producing the desired product **3a** albeit in lower yield (Table 1, entries 9 & 10). Keeping in mind the use of relatively inexpensive and less toxic copper salts, we screened some copper catalysts such as CuCl, CuBr, and Cu(OAc)₂, which also could not provide better result than rhodium (Table 1, entries 11-13). To check the generality of the protocol, we also screened AgSbF₆, which did not trigger the reaction (Table 1, entry 14). In the absence of a catalyst, the reaction did not occur at all revealing the significance of catalyst to the success of the reaction (Table 1, entry 15). Thus, the optimum condition for the synthesis of **3a** was achieved by employing **1a** (0.5 mmol), **2a** (0.5 mmol), Rh(OAc)₂ (2 mol %), in 5 mL of DCM at 50 °C under inert atmosphere (Table 1, entry 2).

With the established optimal conditions in hand, we explored the scope and generality of the protocol. A wide range of β-ketothioamides (KTAs) with diverse substituents at R¹ moiety irrespective of their substitution pattern (*ortho*, *meta*, or *para*) are found to be compatible well under the optimized conditions showing no obvious electronic/steric hindrance (Table 2). The KTAs bearing substituents like Cl, Br, OMe, Me, CF₃, and -OCH₂O- at phenyl group of R¹ moiety participated well with neutral, electron-donating and electron-withdrawing diazoalkanes affording the corresponding thiazolidinones in good to excellent yields (Table 2). The introduction of halogen (e.g., chloro and bromo) substituents into target product is attractive because of their potential for further synthetic elaborations. Different thioamides (**1a-1f**, SI Table S4) were well tolerated with ethyl 2-diazo-2-phenylacetate furnishing the desired thiazolidinones up to 82% yield (Table 2, entries **3a-3f**). After effective implementation of the neutral diazoalkane, next we employed the electron-rich diazoalkane, which was found to be viable participant with different thioamides (**1g-1l**, SI Table S4) and afforded the corresponding products in good yields (Table 2, entries **3g-3l**). Further, we examined the scope of this transformation with electron-poor diazo-

alkanes with diverse KTAs (**1m-1x**, SI Table S4), which also delivered the corresponding thiazolidinones in excellent yield (Table 2, entries **3m-3x**).

Table 2. Scope of β -Ketothioamides and α -Diazoacetates to Access Diverse Thiazolidinones^a



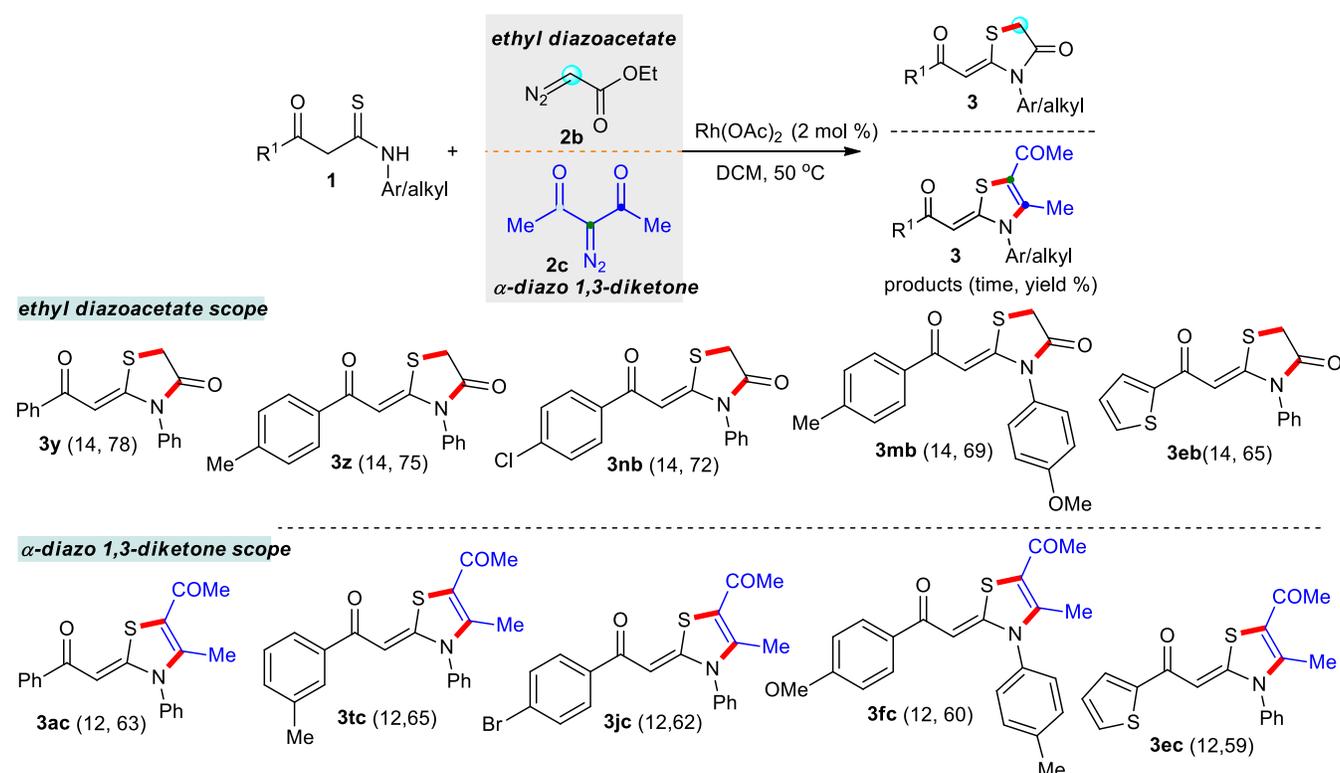
^aAll the reaction were performed with 0.5 mmol of each thioamides (**1**) and diazo compounds (**2**).

Remarkably, KTAs bearing not only aromatic, extended aromatic, and heteroaromatic entrants at R¹ moiety were found to be highly amenable to this protocol, but also alicyclic and aliphatic groups like cyclopropyl and *iso*-butyl at R¹ moiety were also tolerated well under the standard conditions affording the corresponding thiazolidinones in good yields (Table 2, entries **3r** and **3w**). Interestingly, thioamides bearing three strong electron-donating methoxy groups and strong electron-withdrawing CF₃ group at R¹ moiety were also coupled well with electron-poor diazoalkane furnishing the desired products in good yields (Table 2, entries **3u** and **3v**). After successful utilization of β -ketothioamides containing ar-

omatic moiety at nitrogen atom, next we employed the thioamides bearing alkyl substituents at nitrogen like ethyl group and found to be compatible well affording the corresponding thiazolidinones in good yield (Table 2, entries **3f** and **3x**).

Next, the protocol generality was investigated by carrying out the reaction of thioamides **1** with different diazo compounds with a view to add further diversity to thiazolidinone ring (Table 3). Notably, the α -diazocarbonyl (**2b**) and α -diazo 1,3-diketone (**2c**) also underwent the reaction smoothly with diverse β -ketothioamides bearing R^1 moiety as phenyl and aryl (substituted with both electron-withdrawing and electron-donating groups) substituents under optimized conditions affording the corresponding thiazolidinones and thiazolines in good yields (Table 3, entries **3y-3ec**).

Table 3. Scope of ethyl diazoacetate (2b**) and α -diazo 1,3-diketone (**2c**)^a**



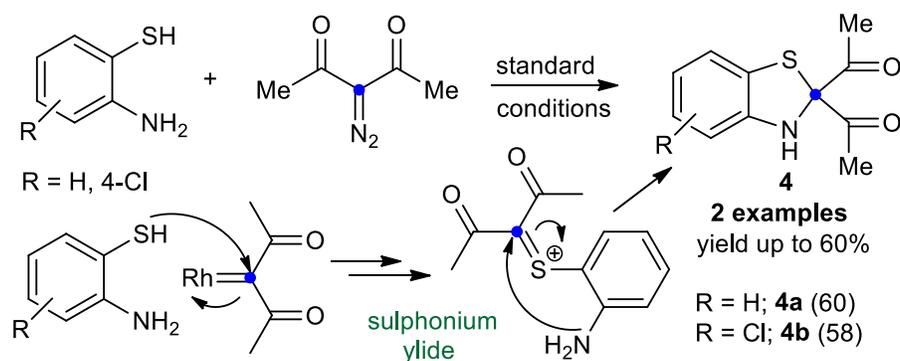
^aAll the reaction were performed with 0.5 mmol of each thioamides (**1**) and diazocompounds (**2**).

Importantly, thioamide bearing π -electron excessive group like 2-thienyl substituent at R^1 moiety was also well tolerated under standard conditions affording the corresponding thiazolidinone and thiazoline

in good yields (Table 3, **3eb** and **3ec**). It has been observed that thioamides with α -diazocarbonyl as coupling partner provided the desired products up to 78% yield, while with α -diazo 1,3-diketone as coupling partner afforded slightly decreased yield (up to 65%).

To validate the further synthetic utility of the present protocol, and in order to extend the substrate scope, we intended to employ 2-aminothiophenol instead of β -ketothioamide with different diazo compounds under above optimized conditions. It has been observed that only α -diazo 1,3-diketone gave the expected product with 2-amino thiophenol, while α -diazocarbonyl and α -diazoalkane could not provide even a trace of the anticipated product under standard conditions. In this regard, we employed 2-aminothiophenols, which underwent reaction efficiently with α -diazo 1,3-diketone under standard conditions providing the desired 2,3-dihydrobenzo[*d*]thiazoles in satisfactory yield (Scheme 2, **4a** and **4b**). As one of the most important exponents, 2-substituted benzothiazoles are of huge interest since they have been used as antimicrobial, antitumor, anticonvulsant, and neuroprotective agents.¹⁹

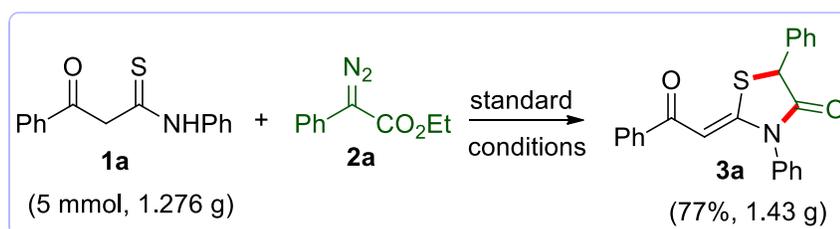
Scheme 2. Application towards Benzo[*d*]thiazoles



The structural determination of all the newly synthesized compounds was achieved by their satisfactory spectral (^1H , ^1H decoupled ^{13}C NMR, and HRMS) studies. The structure of one of the representative compound (Z)-2-(2-oxo-2-(thiophen-2-yl)ethylidene)-3,5-diphenylthiazolidin-4-one (**3e**) was further unambiguously established by the single crystal X-ray crystallography (see Supporting Information Figure S1 for details).²⁰

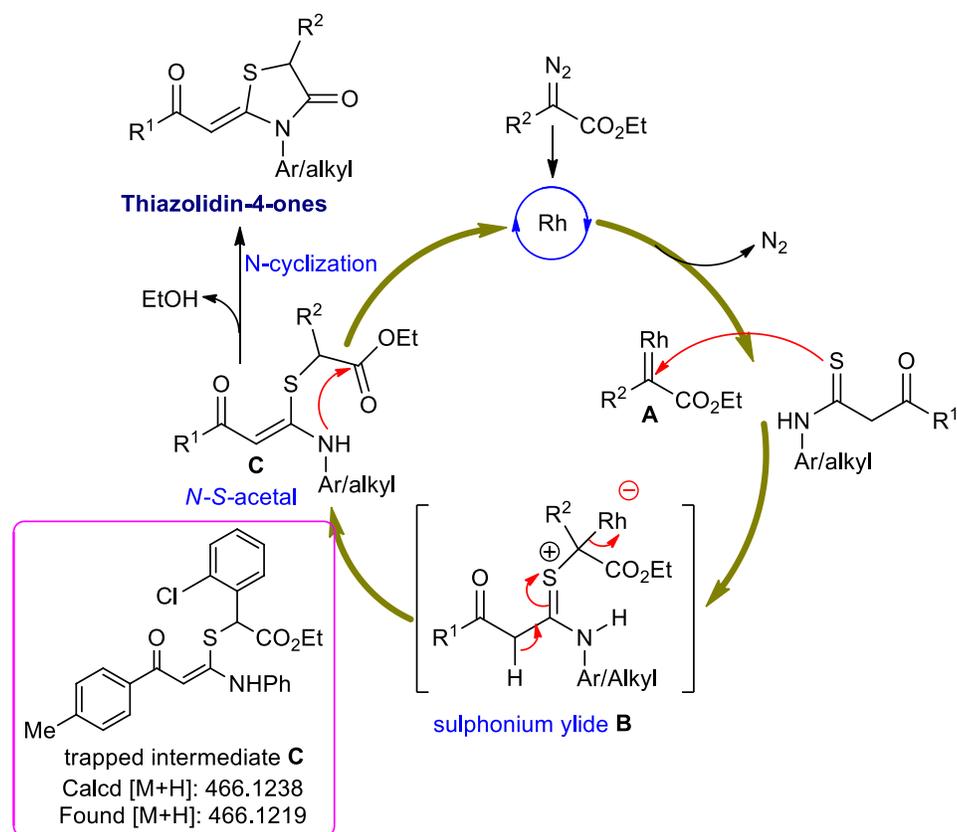
Further to check the practical efficacy of this protocol, we performed the large scale reaction employing β -ketothioamide **1a** (5 mmol) and α -diazocetate **2a** (5 mmol) under optimized reaction conditions (Scheme 3). The desired thiazolidinone **3a** was obtained in 77% yield (1.43 g), which was found to be comparable to the small scale reaction. This observation indicated that the present method could be easily implemented for a large scale preparation of thiazolidinones.

Scheme 3. Gram-scale Synthesis of Compound 3a



Based on our experimental observation, a plausible mechanism is proposed in Scheme 4. The first step is believed to be the reaction between $\text{Rh}(\text{OAc})_2$ and diazo compound to generate rhodium carbenoid species **A** via the extrusion of nitrogen. Next, the rhodium carbenoid species **A** reacts with thioamide sulfur to form the sulphonium ylide **B**, which in turn follows proton transfer to form *N,S*-acetal intermediate **C** by the regeneration of rhodium catalyst. Finally, in situ generated *N,S*-acetal intermediate **C** undergoes intramolecular *N*-cyclization by elimination of ethanol to give the desired thiazolidinones and thiazolines, respectively. To validate the proposed reaction mechanism, we recorded the mass spectrum of the reaction mixture, which shows the existence of intermediate **C** (See SI, Figure S2). The above observation supported that reaction proceeds via intermediate **C**.

Scheme 4. Proposed Mechanism for the Reaction



CONCLUSION

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In summary, we have developed a concise and efficient protocol for the synthesis of diverse thiazolidinones and thiazolines via Rh(II)-catalyzed annulative coupling of β-ketothioamides with diazo compounds for the first time. Moreover, the described methodology is not limited to only β-ketothioamides but was also applicable to 2-amino thiophenols, which enabled direct approach to substituted benzo[*d*]thiazoles. The clean reaction profile, flexible structural modification, broad substrate scope, good functional group tolerance, and alcohol being the only byproduct are additional attributes to this one-pot strategy. Our method enables the synthesis of a broad range of highly functionalized thiazolidinones and thiazolines in good to excellent yield under mild reaction conditions.

EXPERIMENTAL SECTION

Commercially available chemicals used in this manuscript were used without further purification and were purchased from TCI, Sigma-Aldrich and Alfa Aesar. Solvents used for reactions were p.A. grade, and solvents for column chromatography were technical grade and distilled before use; solvent mixtures are understood as volume/volume. All the reactions were carried out in a flame or oven dried glass wares with freshly distilled dry solvents under anhydrous conditions. All the reactions were monitored by analytical thin layer chromatography (TLC) using Merck precoated aluminium sheets and visualized by UV lamp. Column chromatography using silica gel (100-200 mesh) was performed for the purification of products. The ^1H and ^{13}C NMR spectra were recorded on JEOL 500 FT-NMR spectrometer operating at 500 and 125 MHz, respectively. Chemical shifts (δ) for ^1H and $^{13}\text{C}\{^1\text{H}\}$ NMR are given in parts per million (ppm) using the residual solvent peaks as reference relative to tetramethylsilane (TMS). Coupling constant (J) values are reported in Hz. Mass spectra were recorded on Sciex X500R Q-TOF instruments. Melting points are uncorrected.

Materials: All solvents were distilled under argon and dried before use. Catalysts used in this protocol $\text{Rh}(\text{OAc})_2$, $[\text{Rh}(\text{COD})\text{Cl}]_2$, $\text{RhCl}(\text{PPh}_3)_3$, CuCl , CuBr , $\text{Cu}(\text{OAc})_2$ and AgSbF_6 were purchased from Sigma-Aldrich Chemicals Pvt. Ltd. The β -ketoamides^{21a} and diazocompounds^{21b,c} were synthesized by reported procedures.

Important Safety Note. Diazo compounds should only be handled in a well-ventilated fume cupboard. An additional blast shield is recommended. Although we did not realise any problem in the handling of diazo compounds, extreme care should be taken when manipulating them due to their potentially explosive nature. However, the reader should be aware of hazards of the diazo compounds. Safety precautions should be followed and strict risk assessment and proper safety precautions are recommended.

General Procedures for the Synthesis of Thiazolidinones and Thiazolines. In a two neck round bottom flask 0.5 mmol of β -ketoamides **1**, 0.5 mmol of diazo compounds (**2a/2b/2c**) and 2 mol % of Rh(OAc)₂ were taken and sealed with double channel balloon adapter and flushed with argon for 5 to 6 times with the help of vacuum pump. Thereafter dichloromethane (5 mL) was added with the help of syringe. Now, the whole reaction mixture was put into pre-heated oil bath at 50 °C and stirred for stipulated period of time till completion. After the completion of the reaction (monitored by TLC), aqueous work up of reaction mixture was performed using ethyl acetate as organic phase. Organic phase was separated and dried over anhydrous Na₂SO₄ and solvent was evaporated under vacuum. Thus crude reaction mixture obtained was purified by column chromatography using hexane-ethyl acetate (8:2) as eluent to afford the pure final product.

The spectral and analytical data of all the compounds are given as follows:

(Z)-2-(2-oxo-2-phenylethylidene)-3,5-diphenylthiazolidin-4-one (**3a**): The product was obtained as brown solid (82%, 152 mg); mp 205-206 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.74 (d, *J* = 5.0 Hz, 2H), 7.61-7.53 (m, 3H), 7.51-7.46 (m, 3H), 7.43-7.34 (m, 7H), 6.40 (s, 1H), 5.16 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.9, 173.4, 160.7, 138.2, 135.6, 135.3, 132.3, 130.3, 130.0, 129.2, 128.8, 128.5, 128.4, 128.0, 127.6, 97.5, 50.3; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₃H₁₈NO₂S 372.1053; Found 372.1031.

(Z)-2-(2-oxo-2-(*o*-tolyl)ethylidene)-3,5-diphenylthiazolidin-4-one (**3b**): The product was obtained as brown solid (80%, 154 mg); mp 144-145 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.54 (t, *J* = 7.5 Hz, 2H), 7.49 (d, *J* = 5.0 Hz, 2H), 7.43-7.41 (m, 2H), 7.36 (d, *J* = 5.0 Hz, 1H), 7.31-7.26 (m, 4H), 7.18-7.13 (m, 3H), 6.07 (s, 1H), 5.16 (s, 1H), 2.44 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 193.1, 173.4, 159.7, 139.7, 137.2, 135.6, 135.3, 131.5, 130.8, 130.6, 130.3, 129.9, 129.3, 128.8, 128.5, 128.0, 127.8, 126.9, 125.7, 101.5, 101.2, 50.3, 31.7, 22.7, 20.7, 14.2; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₄H₂₀NO₂S 386.1209; Found 386.1183.

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(Z)-2-(2-(2-chlorophenyl)-2-oxoethylidene)-3,5-diphenylthiazolidin-4-one (**3c**): The product was obtained as white solid (78%, 158 mg); mp 149-150 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.55 (t, *J* = 7.5 Hz, 2H), 7.51-7.47 (m, 5H), 7.39-7.36 (m, 1H), 7.32-7.31 (m, 2H), 7.29-7.28 (m, 2H), 7.29-7.18 (m, 2H), 6.19 (s, 1H), 5.18 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 189.9, 173.4, 160.1, 139.5, 135.4, 135.1, 131.5, 131.3, 130.8, 130.4, 130.2, 130.0, 129.9, 129.3, 128.9, 128.5, 128.0, 127.0, 101.8, 50.5; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₃H₁₇ClNO₂S 406.0663; Found 406.0645.

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(Z)-2-(2-oxo-2-(4-(trifluoromethyl)phenyl)ethylidene)-3,5-diphenylthiazolidin-4-one (**3d**): The product was obtained as white solid (75%, 165 mg); mp 198-199 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.72 (d, *J* = 5.0 Hz, 2H), 7.55-7.45 (m, 5H), 7.39 (d, *J* = 10.0 Hz, 2H), 7.33-7.25 (m, 5H+1H of CDCl₃), 6.26 (s, 1H), 5.08 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 187.5, 173.3, 162.2, 141.1, 135.3, 135.1, 133.5 (q, ²*J*_{CF3} = 32.5 Hz), 130.4, 130.1, 129.5, 129.3, 128.9, 128.6, 128.4, 128.0, 127.9, 126.6, 125.6, 125.6, 123.7 (q, ¹*J*_{CF3} = 271.2 Hz), 97.2, 50.3; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₄H₁₇F₃NO₂S 440.0927; Found 440.0897.

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(Z)-2-(2-oxo-2-(thiophen-2-yl)ethylidene)-3,5-diphenylthiazolidin-4-one (**3e**): The product was obtained as yellow solid (70%, 132 mg); mp 204-205 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.50-7.39 (m, 6H), 7.33-7.17 (m, 6H+1H of CDCl₃), 6.93 (t, *J* = 5.0, 1H), 6.13 (s, 1H), 5.06 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 181.5, 173.2, 159.9, 145.5, 135.6, 135.2, 132.7, 130.3, 130.0, 129.9, 129.2, 128.8, 128.4, 128.1, 128.0, 97.6, 50.4; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₁H₁₆NO₂S₂ 378.0617; Found 378.0593.

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(Z)-3-ethyl-2-(2-(4-methoxyphenyl)-2-oxoethylidene)-5-phenylthiazolidin-4-one (**3f**): The product was obtained as white solid (67%, 118 mg); mp 142-143 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.87 (d, *J* = 10.0 Hz, 2H), 7.28-7.24 (m, 5H+1H of CDCl₃), 6.88 (d, *J* = 10.0 Hz, 2H), 6.66 (s, 1H), 4.86 (s, 1H), 3.81 (q, *J* = 6.6 Hz, 2H), 3.78 (s, 3H), 1.24 (t, *J* = 5.0 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ

187.6, 173.5, 163.0, 158.6, 135.7, 131.3, 131.1, 129.7, 129.1, 128.6, 128.4, 114.0, 113.8, 95.1, 55.5, 50.1, 39.1, 12.1; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₀H₂₀NO₃S 354.1158; Found 354.1132.

(*Z*)-5-(3-methoxyphenyl)-2-(2-oxo-2-(*o*-tolyl)ethylidene)-3-phenylthiazolidin-4-one (**3g**): The product was obtained as yellow solid (75%, 156 mg); mp 134-135 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.53 (t, *J* = 7.5 Hz, 2H), 7.47 (t, *J* = 7.5 Hz, 1H), 7.33-7.24 (m, 5H+1H of CDCl₃), 7.17-7.12 (m, 2H), 7.07 (d, *J* = 5.0 Hz, 1H), 7.02 (s, 1H), 6.90-6.88 (m, 1H), 6.06 (s, 1H), 5.12 (s, 1H), 3.81 (s, 3H), 2.44 (s, 3H); ¹³C {¹H} NMR (125 MHz, CDCl₃) δ 193.0, 173.1, 160.1, 159.6, 139.6, 137.1, 137.0, 135.2, 131.5, 130.5, 130.2, 130.2, 129.9, 127.9, 127.8, 125.6, 120.6, 114.3, 114.1, 101.2, 55.4, 50.1, 20.6; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₅H₂₂NO₃S 416.1315; Found 416.1287.

(*Z*)-5-(3-methoxyphenyl)-2-(2-(3-methoxyphenyl)-2-oxoethylidene)-3-phenylthiazolidin-4-one (**3h**): The product was obtained as yellow solid (76%, 164 mg); mp 166-167 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.58-7.52 (m, 3H), 7.37 (s, 1H), 7.33-7.21 (m, 5H+1H of CDCl₃), 7.07 (d, *J* = 5.0 Hz, 1H), 7.02 (s, 2H), 6.90-6.88 (m, 1H), 6.36 (s, 1H), 5.11 (s, 1H), 3.81 (s, 3H), 3.80 (s, 3H); ¹³C {¹H} NMR (125 MHz, CDCl₃) δ 188.6, 173.2, 160.8, 160.2, 159.8, 139.7, 137.0, 135.3, 130.3, 130.3, 130.0, 129.5, 128.0, 120.6, 119.9, 118.7, 114.3, 114.2, 112.2, 97.7, 55.4, 50.2; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₅H₂₂NO₄S 432.1264; Found 432.1235.

(*Z*)-2-(2-(benzo[*d*][1,3]dioxol-5-yl)-2-oxoethylidene)-5-(3-methoxyphenyl)-3-phenylthiazolidin-4-one (**3i**): The product was obtained as brown solid (70%, 156 mg); mp 194-195 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.59-7.52 (m, 3H), 7.33-7.25 (m, 5H+1H of CDCl₃), 7.07 (d, *J* = 10.0 Hz, 1H), 7.01 (s, 1H), 6.89-6.88 (m, 1H), 6.75 (d, *J* = 5.0 Hz, 1H), 6.28 (s, 1H), 5.99 (s, 2H), 5.10 (s, 1H), 3.81 (s, 3H); ¹³C {¹H} NMR (125 MHz, CDCl₃) δ 187.1, 173.2, 160.2, 160.1, 151.2, 148.1, 137.1, 135.3, 133.0, 130.3, 130.3, 130.0, 128.1, 123.3, 120.7, 114.3, 114.2, 107.9, 107.8, 101.8, 97.4, 55.4, 50.2; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₂₅H₂₀NO₅S 446.1057; Found 446.1057.

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(Z)-2-(2-(4-bromophenyl)-2-oxoethylidene)-5-(3-methoxyphenyl)-3-phenylthiazolidin-4-one (**3j**): The product was obtained as yellow solid (78%, 187 mg); mp 182-183 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.59-7.50 (m, 7H), 7.32 (t, *J* = 7.5 Hz, 3H), 7.0 (d, *J* = 5.0 Hz, 1H), 7.0 (s, 1H), 6.90-6.88 (m, 1H), 6.30 (s, 1H), 5.11 (s, 1H), 3.81 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 187.6, 173.2, 161.4, 160.2, 137.0, 136.8, 135.2, 131.8, 130.4, 130.3, 130.1, 129.1, 128.0, 127.2, 120.6, 114.3, 114.2, 97.1, 55.4, 50.2; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₄H₁₉BrNO₃S 480.0264; Found 480.0258.

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(Z)-5-(3-methoxyphenyl)-2-(2-(naphthalen-2-yl)-2-oxoethylidene)-3-phenylthiazolidin-4-one (**3k**): The product was obtained as light brown solid (69%, 156 mg); mp 194-195 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.19 (s, 1H), 7.87-7.82 (m, 4H), 7.62-7.47 (m, 5H), 7.36-7.31 (m, 3H), 7.08 (d, *J* = 10.0 Hz, 1H), 7.04 (s, 1H), 6.89 (d, *J* = 5.0 Hz, 1H), 6.53 (s, 1H), 5.13 (s, 1H), 3.82 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.8, 173.3, 160.7, 160.2, 137.0, 135.6, 135.4, 135.2, 132.6, 130.4, 130.3, 130.0, 129.5, 128.6, 128.4, 128.1, 127.8, 126.7, 123.9, 120.7, 114.3, 114.2, 97.8, 55.4, 50.3; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₈H₂₂NO₃S 452.1315; Found 452.1302.

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(Z)-2-(2-(furan-2-yl)-2-oxoethylidene)-5-(3-methoxyphenyl)-3-phenylthiazolidin-4-one (**3l**): The product was obtained as yellow solid (67%, 131 mg); mp 177-178 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.59-7.52 (m, 3H), 7.43 (s, 1H), 7.31 (t, *J* = 10.0 Hz, 3H), 7.07-06 (m, 2H), 7.01 (s, 1H), 6.89-6.88 (m, 1H), 6.45 (d, *J* = 5.0 Hz, 1H), 6.27 (s, 1H), 5.11 (s, 1H), 3.81 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 177.8, 173.1, 160.2, 160.0, 153.5, 145.4, 137.0, 135.2, 130.3, 129.9, 128.1, 120.6, 115.5, 114.2, 112.4, 97.4, 55.4, 50.3; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₂H₁₈NO₄S 392.0951; Found 392.0923.

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(Z)-5-(2-chlorophenyl)-2-(2-oxo-2-(*p*-tolyl)ethylidene)-3-phenylthiazolidin-4-one (**3m**): The product was obtained as yellow solid (72%, 151 mg); mp 203-204 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.61 (t, *J* = 10.0 Hz, 4H), 7.56 (t, *J* = 7.5 Hz, 1H), 7.45-7.43 (m, 1H), 7.41-7.36 (m, 3H), 7.31-7.28 (m, 2H), 7.16 (d, *J* = 10.0 Hz, 2H), 6.36 (s, 1H), 5.53 (s, 1H), 2.36 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.6,

172.7, 160.4, 143.0, 135.6, 135.4, 134.5, 133.7, 130.5, 130.3, 130.1, 130.0, 129.2, 128.0, 127.7, 127.6, 97.6, 48.6, 21.6; HRMS (ESI-TOF) m/z: [M + H]⁺Calcd for C₂₄H₁₉ClNO₂S 420.0820; Found 420.0801.

(*Z*)-5-(2-chlorophenyl)-2-(2-(4-chlorophenyl)-2-oxoethylidene)-3-phenylthiazolidin-4-one (**3n**): The product was obtained as white solid (75%, 165 mg); mp 209-210 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.65 (d, *J* = 10.0 Hz, 2H), 7.61 (d, *J* = 10.0 Hz, 2H), 7.58-7.55 (m, 1H), 7.46-7.44 (m, 1H), 7.40-7.36 (m, 3H), 7.34-7.32 (m, 2H), 7.31-7.29 (m, 2H), 6.30 (s, 1H), 5.53 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 187.4, 172.6, 161.6, 138.6, 136.5, 135.2, 134.5, 133.5, 130.6, 130.5, 130.4, 130.2, 130.1, 129.0, 128.8, 127.9, 127.6, 97.1, 48.6; HRMS (ESI-TOF) m/z: [M + H]⁺Calcd for C₂₃H₁₆Cl₂NO₂S 440.0273; Found 440.0255.

(*Z*)-2-(2-(3-bromophenyl)-2-oxoethylidene)-5-(2-chlorophenyl)-3-phenylthiazolidin-4-one (**3o**): The product was obtained as yellow solid (76%, 184 mg); mp 166-167 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.86 (s, 1H), 7.63-7.56 (m, 5H), 7.45-7.43 (m, 1H), 7.40-7.36 (m, 3H), 7.31-7.29 (m, 2H), 7.22 (t, *J* = 7.5 Hz, 1H), 6.28 (s, 1H), 5.33 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 187.1, 172.6, 161.9, 140.0, 135.1, 135.0, 134.4, 133.4, 130.6, 130.5, 130.4, 130.2, 130.2, 130.1, 127.8, 127.6, 126.0, 122.9, 97.0, 48.6; HRMS (ESI-TOF) m/z: [M + H]⁺Calcd for C₂₃H₁₆BrClNO₂S 483.9768; Found 483.9757.

(*Z*)-2-(2-([1,1'-biphenyl]-4-yl)-2-oxoethylidene)-5-(2-chlorophenyl)-3-phenylthiazolidin-4-one (**3p**): The product was obtained as yellow solid (70%, 169 mg); mp 207-208 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.79 (d, *J* = 10.0 Hz, 2H), 7.64-7.56 (m, 7H), 7.46-7.44 (m, 3H), 7.43-7.36 (m, 4H), 7.32-7.30 (m, 2H), 6.40 (s, 1H), 5.55 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.4, 172.7, 160.9, 145.0, 140.0, 136.9, 135.4, 134.5, 133.7, 130.5, 130.4, 130.2, 130.1, 129.0, 128.4, 128.2, 128.2, 128.0, 127.6, 127.4, 127.3, 127.2, 97.6, 48.6; HRMS (ESI-TOF) m/z: [M + H]⁺Calcd for C₂₉H₂₁ClNO₂S 482.0976; Found 482.0975.

(Z)-5-(2-chlorophenyl)-3-(4-methoxyphenyl)-2-(2-oxo-2-(*p*-tolyl)ethylidene)thiazolidin-4-one (**3q**):

The product was obtained as yellow solid (68%, 153 mg); mp 176-177 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.63 (d, *J* = 5.0 Hz, 2H), 7.45-7.43 (m, 1H), 7.40-7.38 (m, 1H), 7.31-7.27 (m, 4H), 7.17 (d, *J* = 5.0 Hz, 2H), 7.11-7.09 (m, 2H), 6.37 (s, 1H), 5.52 (s, 1H), 3.90 (s, 3H), 2.36 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.6, 172.9, 160.8, 160.5, 143.0, 135.7, 134.5, 133.9, 130.5, 130.1, 129.2, 127.8, 127.7, 127.6, 115.6, 97.6, 55.7, 48.5, 21.7; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₅H₂₁ClNO₃S 450.0925; Found 450.0896.

(Z)-5-(2-chlorophenyl)-2-(2-cyclopropyl-2-oxoethylidene)-3-phenylthiazolidin-4-one (**3r**): The prod-

uct was obtained as yellow solid (66%, 122 mg); mp 156-157 °C; ¹H NMR (500 MHz, CDCl₃) δ 6.69 (t, *J* = 7.5 Hz, 2H), 7.64 (t, *J* = 7.5 Hz, 1H), 7.54 (t, *J* = 5.0 Hz, 1H), 7.48-7.46 (m, 1H), 7.43 (d, *J* = 10.0 Hz, 2H), 7.40-7.38 (m, 2H), 5.40 (s, 1H), 5.60 (s, 1H), 1.84-1.80 (m, 1H), 1.05 (t, *J* = 5.0 Hz, 2H), 0.81 (t, *J* = 5.0 Hz, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 198.4, 172.7, 157.4, 135.3, 134.5, 133.9, 130.4, 130.3, 130.1, 129.9, 128.0, 127.6, 101.1, 48.5, 21.5, 10.9, 10.9; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₀H₁₇ClNO₂S 370.0663; Found 370.0638.

(Z)-5-(4-nitrophenyl)-2-(2-oxo-2-phenylethylidene)-3-phenylthiazolidin-4-one (**3s**): The product was

obtained as brown solid (70%, 146 mg); mp 178-179 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.27 (d, *J* = 10.0 Hz, 2H), 7.74-7.70 (m, 4H), 7.62-7.57 (m, 4H), 7.50 (t, *J* = 7.5 Hz, 1H), 7.40 (t, *J* = 7.5 Hz, 3H), 6.43 (s, 1H), 5.25 (s, 1H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 189.1, 172.2, 159.5, 148.1, 142.6, 137.9, 135.0, 132.6, 130.5, 130.3, 129.6, 128.7, 127.9, 127.7, 124.4, 98.3, 49.6; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₃H₁₇N₂O₄S 417.0904; Found 417.0887.

(Z)-5-(4-nitrophenyl)-2-(2-oxo-2-(*m*-tolyl)ethylidene)-3-phenylthiazolidin-4-one (**3t**): The product

was obtained as brown solid (72%, 155 mg); mp 139-140 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.28 (d, *J* = 10.0 Hz, 2H), 7.71 (d, *J* = 5.0 Hz, 2H), 7.66-7.56 (m, 5H), 7.47 (d, *J* = 5.0 Hz, 1H), 7.32-7.27 (m, 3H),

6.41 (s, 1H), 5.25 (s, 1H), 2.36 (s, 3H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3) δ 189.3, 172.2, 159.3, 148.2, 142.7, 138.6, 138.0, 135.0, 133.5, 130.5, 130.3, 129.6, 128.5, 128.3, 128.0, 124.8, 124.4, 98.5, 49.6, 21.5; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{24}\text{H}_{19}\text{N}_2\text{O}_4\text{S}$ 431.1060; Found 431.1033.

(Z)-5-(4-nitrophenyl)-2-(2-oxo-2-(3,4,5-trimethoxyphenyl)ethylidene)-3-phenylthiazolidin-4-one

(3u): The product was obtained as brown solid (66%, 167 mg); mp 141-142 °C; ^1H NMR (500 MHz, CDCl_3) δ 8.28 (d, $J = 10.0$ Hz, 2H), 7.72 (d, $J = 10.0$ Hz, 2H), 7.61-7.42 (m, 5H), 6.98 (s, 2H), 6.35 (s, 1H), 5.26 (s, 1H), 3.88 (s, 3H), 3.82 (s, 6H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3) δ 188.0, 172.1, 159.4, 153.2, 148.2, 142.5, 142.4, 135.0, 133.3, 130.4, 130.3, 129.6, 128.0, 124.4, 105.4, 98.1, 61.0, 56.4, 56.2, 49.7; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{26}\text{H}_{23}\text{N}_2\text{O}_7\text{S}$ 507.1220; Found 507.1192.

(Z)-5-(4-nitrophenyl)-2-(2-oxo-2-(4-(trifluoromethyl)phenyl)ethylidene)-3-phenylthiazolidin-4-one

(3v): The product was obtained as brown solid (65%, 157 mg); mp 171-172 °C; ^1H NMR (500 MHz, CDCl_3) δ 8.28 (d, $J = 10.0$ Hz, 2H), 7.82 (d, $J = 5.0$ Hz, 2H), 7.71 (d, $J = 5.0$ Hz, 2H), 7.67-7.59 (m, 5H), 7.36-7.31 (m, 2H), 6.40 (s, 1H), 5.28 (s, 1H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3) δ 187.84, 172.18, 161.00, 148.23, 142.28, 140.81, 138.12, 134.82, 133.8 (q, $^2J_{\text{CF}_3} J = 32.5$), 131.72, 130.62, 130.49, 130.40, 129.65, 128.02, 127.91, 125.78, 125.76, 124.49, 123.7 (q, $^1J_{\text{CF}_3} J = 270.8$ Hz), 97.90, 49.65; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{24}\text{H}_{16}\text{F}_3\text{N}_2\text{O}_4\text{S}$ 485.0777; Found 485.0767.

(Z)-2-(4-methyl-2-oxopentylidene)-5-(4-nitrophenyl)-3-phenylthiazolidin-4-one (3w): The product was obtained as dark brown solid (60%, 119 mg); mp 135-136 °C; ^1H NMR (500 MHz, CDCl_3) δ 8.27 (d, $J = 10.0$ Hz, 2H), 7.68 (d, $J = 10.0$ Hz, 2H), 7.58-7.55 (m, 1H), 7.41 (t, $J = 7.5$ Hz, 3H), 7.29 (t, $J = 7.5$ Hz, 1H), 5.67 (s, 1H), 5.19 (s, 1H), 2.24 (d, $J = 5.0$ Hz, 2H), 2.11-2.04 (m, 1H), 0.89 (d, $J = 5.0$ Hz, 6H); $^{13}\text{C}\{^1\text{H}\}$ NMR (125 MHz, CDCl_3) δ 172.2, 156.6, 142.7, 134.9, 130.4, 129.7, 129.6, 127.9, 127.3, 125.4, 124.4, 101.7, 52.3, 49.5, 25.5, 22.7; HRMS (ESI-TOF) m/z: $[\text{M} + \text{H}]^+$ Calcd for $\text{C}_{21}\text{H}_{21}\text{N}_2\text{O}_4\text{S}$ 397.1217; Found 397.1204.

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(Z)-3-ethyl-2-(2-(4-methoxyphenyl)-2-oxoethylidene)-5-(4-nitrophenyl)thiazolidin-4-one (**3x**): The product was obtained as white solid (63%, 125 mg); mp 183-184 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.23 (d, *J* = 5.0 Hz, 2H), 7.96 (d, *J* = 5.0 Hz, 2H), 7.59 (d, *J* = 5.0 Hz, 2H), 6.98 (d, *J* = 7.5 Hz, 2H), 6.79 (s, 1H), 5.06 (s, 1H), 3.93 (d, *J* = 3.33, 2H), 3.89 (s, 3H), 1.36 (t, *J* = 5.0 Hz, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 187.8, 172.3, 163.3, 157.4, 148.0, 142.9, 131.0, 129.9, 129.6, 124.3, 114.0, 95.8, 55.6, 49.5, 39.5, 12.2; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₀H₁₉N₂O₅S 399.1009; Found 399.0997.

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(Z)-2-(2-oxo-2-phenylethylidene)-3-phenylthiazolidin-4-one (**3y**): The product was obtained as brown solid (78%, 115 mg); mp 190-191 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.72-7.70 (m, 2H), 7.62-7.56 (m, 3H), 7.45 (d, *J* = 5.0 Hz, 1H), 7.37 (t, *J* = 7.5 Hz, 2H), 7.32-7.30 (m, 2H), 6.32 (s, 1H), 3.90 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.9, 172.6, 162.3, 138.3, 135.2, 132.3, 130.4, 130.1, 128.6, 128.0, 127.6, 97.8, 32.2; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₇H₁₄NO₂S 296.0740; Found 296.0725.

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(Z)-2-(2-oxo-2-(*p*-tolyl)ethylidene)-3-phenylthiazolidin-4-one (**3z**): The product was obtained as brown solid (75%, 116 mg); mp 215-216 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.62-7.56 (m, 5H), 7.30 (d, *J* = 5.0 Hz, 2H), 7.17 (d, *J* = 10.0 Hz, 2H), 6.30 (s, 1H), 3.90 (s, 2H), 2.36 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.6, 172.7, 161.9, 143.0, 135.7, 135.2, 130.4, 130.0, 129.3, 128.0, 127.7, 97.9, 32.2, 21.7; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₈H₁₆NO₂S 310.0896; Found 310.0883.

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(Z)-2-(2-(4-chlorophenyl)-2-oxoethylidene)-3-phenylthiazolidin-4-one (**3nb**): The product was obtained as brown solid (72%, 119 mg); mp 160-161 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.69 (s, 1H), 7.63-7.54 (m, 4H), 7.43 (d, *J* = 10.0 Hz, 1H), 7.32-7.29 (m, 3H), 6.24 (s, 1H), 3.91 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 169.2, 153.1, 139.3, 128.93, 128.6, 125.4, 123.2, 122.1, 121.3, 121.3, 121.2, 119.7, 119.7, 119.7, 29.8, 20.4; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₇H₁₃ClNO₂S 330.0350; Found 330.0330.

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(Z)-3-(4-methoxyphenyl)-2-(2-oxo-2-(*p*-tolyl)ethylidene)thiazolidin-4-one (**3mb**): The product was
obtained as brown solid (69%, 117 mg); mp 194-195 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.63 (d, *J* =
10.0 Hz, 2H), 7.22-7.17 (m, 4H), 7.08 (d, *J* = 5.0 Hz, 2H), 6.32 (s, 1H), 3.89 (s, 3H), 3.87 (s, 2H), 2.36
(s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 188.6, 172.9, 162.2, 160.5, 143.0, 135.8, 129.2, 129.1,
127.7, 127.6, 115.6, 97.8, 55.7, 32.1, 29.8, 21.7; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for
C₁₉H₁₈NO₃S 340.1002; Found 340.0985.

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(Z)-2-(2-oxo-2-(thiophen-2-yl)ethylidene)-3-phenylthiazolidin-4-one (**3eb**): The product was obtained
as dark brown solid (65%, 98 mg); mp 139-140 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.62-7.54 (m, 3H),
7.52 (d, *J* = 5.0 Hz, 1H), 7.35 (d, *J* = 5.0 Hz, 1H), 7.29 (d, *J* = 5.0 Hz, 2H), 7.01 (t, *J* = 5.0 Hz, 1H), 6.15
(s, 1H), 3.90 (s, 2H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 181.5, 172.5, 161.6, 145.5, 135.1, 132.7,
130.4, 130.1, 129.8, 128.0, 128.0, 97.8, 32.3; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₁₅H₁₂NO₂S₂
302.0304; Found 302.0296.

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(Z)-2-(5-acetyl-4-methyl-3-phenylthiazol-2(3H)-ylidene)-1-phenylethanone (**3ac**): The product was
obtained as yellow solid (63%, 106 mg); mp 229-230 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.73 (d, *J* =
10.0 Hz, 2H), 7.67-7.61 (m, 3H), 7.40-7.29 (m, 5H), 6.07 (s, 1H), 2.51 (s, 3H), 2.31 (s, 3H); ¹³C{¹H}
NMR (125 MHz, CDCl₃) δ 191.0, 184.3, 162.3, 145.6, 139.1, 136.7, 130.9, 130.9, 130.6, 128.3, 128.1,
127.1, 116.7, 90.1, 30.4, 14.5; HRMS (ESI-TOF) *m/z*: [M + H]⁺ Calcd for C₂₀H₁₈NO₂S 336.1053;
Found 336.1053.

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(Z)-2-(5-acetyl-4-methyl-3-phenylthiazol-2(3H)-ylidene)-1-(*m*-tolyl)ethanone (**3tc**): The product was
obtained as pale yellow solid (65%, 113 mg); mp 174-175 °C; ¹H NMR (500 MHz, CDCl₃) δ 7.68-7.63
(m, 3H), 7.59 (s, 1H), 7.48-7.46 (m, 1H), 7.31-7.30 (m, 2H), 7.22 (t, *J* = 7.5 Hz, 2H), 6.06 (s, 1H), 2.52
(s, 3H), 2.34 (s, 3H), 2.31 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 191.1, 184.5, 162.2, 145.6,

139.1, 138.0, 136.7, 131.7, 130.9, 130.5, 128.1, 128.1, 127.8, 124.1, 116.6, 90.2, 30.4, 21.5, 14.5;

HRMS (ESI-TOF) m/z : $[M + H]^+$ Calcd for $C_{21}H_{20}NO_2S$ 350.1209; Found 350.1186.

(Z)-2-(5-acetyl-4-methyl-3-phenylthiazol-2(3H)-ylidene)-1-(4-bromophenyl)ethanone (**3jc**): The product was obtained as pale yellow solid (62%, 128 mg); mp 215-216 °C; 1H NMR (500 MHz, $CDCl_3$) δ 7.68-7.64 (m, 3H), 7.59 (d, $J = 10.0$ Hz, 2H), 7.46 (d, $J = 10.0$ Hz, 2H), 7.30 (d, $J = 5.0$ Hz, 2H), 6.0 (s, 1H), 2.52 (s, 3H), 2.32 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$) δ 190.9, 182.8, 162.6, 145.7, 137.9, 136.5, 131.5, 130.9, 130.7, 128.7, 128.0, 125.5, 116.9, 89.7, 30.4, 14.5; HRMS (ESI-TOF) m/z : $[M + H]^+$ Calcd for $C_{20}H_{17}BrNO_2S$ 414.0158; Found 414.0159.

(Z)-2-(5-acetyl-4-methyl-3-(*p*-tolyl)thiazol-2(3H)-ylidene)-1-(4-methoxyphenyl)ethanone (**3fc**): The product was obtained as yellow solid (60%, 104 mg); mp 205-206 °C; 1H NMR (500 MHz, $CDCl_3$) δ 7.66 (d, $J = 5.0$ Hz, 2H), 7.22-7.19 (m, 2H), 7.16-7.11 (m, 4H), 6.08 (s, 1H), 3.93 (s, 3H), 2.51 (s, 3H), 2.36 (s, 3H), 2.31 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$) δ 191.1, 184.2, 162.4, 160.8, 146.2, 141.3, 136.5, 129.2, 129.1, 129.0, 127.2, 115.9, 90.0, 55.8, 30.4, 21.6, 14.5; HRMS (ESI-TOF) m/z : $[M + H]^+$ Calcd for $C_{22}H_{22}NO_3S$ 380.1315; Found 380.1303.

(Z)-2-(5-acetyl-4-methyl-3-phenylthiazol-2(3H)-ylidene)-1-(thiophen-2-yl)ethanone (**3ec**): The product was obtained as yellow solid (59%, 101 mg); mp 218-219 °C; 1H NMR (500 MHz, $CDCl_3$) δ 7.68-7.63 (m, 3H), 7.40 (d, $J = 5.0$ Hz, 1H), 7.32-7.29 (m, 3H), 6.98 (t, $J = 5.0$ Hz, 1H), 5.90 (s, 1H), 2.50 (s, 3H), 2.31 (s, 3H); $^{13}C\{^1H\}$ NMR (125 MHz, $CDCl_3$) δ 190.9, 177.6, 161.8, 145.6, 136.6, 130.9, 130.6, 130.3, 128.1, 127.7, 127.6, 117.1, 89.9, 30.4, 14.5; HRMS (ESI-TOF) m/z : $[M + H]^+$ Calcd for $C_{18}H_{16}NO_2S_2$ 342.0617; Found 342.0617.

1-(3-methyl-4H-benzo[*b*][1,4]thiazin-2-yl)ethanone (**4a**): The product was obtained as brown sticky solid (60%, 66 mg); 1H NMR (500 MHz, $CDCl_3$) δ 8.02 (d, $J = 10.0$ Hz, 1H), 7.88 (d, $J = 5.0$ Hz, 1H), 7.49 (t, $J = 7.5$ Hz, 1H), 7.39 (t, $J = 7.5$ Hz, 1H), 5.19 (s, 1H), 2.49 (s, 3H), 1.96 (s, 3H); $^{13}C\{^1H\}$ NMR

(125 MHz, CDCl₃) δ 206.7, 174.0, 153.8, 135.7, 132.0, 129.9, 127.8, 127.1, 126.2, 125.5, 125.3, 123.2, 122.5, 121.9, 81.3, 25.9, 23.5; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₁H₁₂NO₂S 222.0583; Found 222.0559.

1-(6-chloro-3-methyl-4H-benzo[b][1,4]thiazin-2-yl)ethanone (4b): The product was obtained as white solid (58%, 74 mg); mp 134-135 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.01 (s, 1H), 7.80 (d, *J* = 10.0 Hz, 1H), 7.37 (dd, *J* = 10.0 Hz, 5.0 Hz, 1H), 5.17 (s, 1H), 2.50 (s, 3H), 1.93 (s, 3H); ¹³C{¹H} NMR (125 MHz, CDCl₃) δ 206.4, 176.2, 154.7, 134.0, 132.3, 125.9, 123.1, 122.7, 81.4, 26.0, 23.5; HRMS (ESI-TOF) m/z: [M + H]⁺ Calcd for C₁₁H₁₁ClNO₂S 256.0194; Found 256.0181.

ASSOCIATED CONTENT

Supporting Information

Experimental procedure, full characterization of products, copies of ¹H, ¹³C NMR, HRMS spectra (PDF), and CIF information. The Supporting Information is available free of charge on the ACS Publications website.

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Notes

The authors declare no competing financial interest.

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REFERENCES

- (1) (a) Brown, F. C. 4-Thiazolidinones. *Chem. Rev.* **1961**, *61*, 463-521. (b) Singh, S. P.; Parmar, S. S.; Raman, K.; Stenberg, V. I. Chemistry and biological activity of thiazolidinones. *Chem. Rev.* **1981**, *81*, 175-203. (c) Jain, A. K.; Vaidya, A.; Ravichandran, V.; Kashaw, S. K.; Agrawal, R. K. Recent developments and biological activities of thiazolidinone derivatives: A review. *Bioorg. Med. Chem.* **2012**, *20*, 3378-3395. (d) Cunico, W.; Gomes, C. R. B.; Vellasco, W. T. Jr. Chemistry and Biological Activities of 1,3-Thiazolidin-4-ones. *Mini-Rev. Org. Chem.* **2008**, *5*, 336-344.
- (2) (a) Mahmoodia, N. O.; Zeydib, M. M.; Biazarc, E.; Kazeminejad, Z. Synthesis of novel thiazolidine-4-one derivatives and their anticancer activity. *Phosphorus, Sulfur, & Silicon and the Related Elements* **2017**, *192*, 344-350. (b) Vittoria Diurno, M.; Mazzoni, O.; Piscopo, E.; Calignano, A.; Giordano, F.; Bolognese, A. Synthesis and antihistaminic activity of some thiazolidin-4-ones. *J. Med. Chem.* **1992**, *35*, 2910-2912. (c) Rao, A.; Carbone, A.; Chimirri, A.; Clercq, E. De.; Monforte, A. M.; Monforte, P.; Pannecouque, C.; Zappala, M. Synthesis and anti-HIV activity of 2,3-diaryl-1,3-thiazolidin-4-(thi)one derivatives. *II Farmaco* **2002**, *57*, 747-751. (d) Zarghia, A.; Najafni, L.; Daraee, B.; G. Dadrass, O.; Heydayati, M. Synthesis of 2,3-diaryl-1,3-thiazolidine-4-one derivatives as selective cyclooxygenase (COX-2) inhibitors. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 5634-5637.
- (3) Udaykumar, D.; Bhoi, A. A complete review of thiazolidine-4-ones. *J. Pharm. Res.* **2011**, *7*, 2436-2440.

1 (4) (a) Oecal, N.; Aydogan, F.; Yolacan, C.; Turgut, Z. Synthesis of Some Furo-Thiazolidine Deriva-
2 tives Starting from Aldimines. *J. Heterocycl. Chem.* **2003**, *40*, 721-724. (b) Bolognese, A.; Correale, G.;
3 Manfra, M.; Lavecchia, A.; Novellino, E.; Barone, V. Thiazolidin-4-one formation: Mechanistic and syn-
4 thetic aspects of the reaction of imines and mercaptoacetic acid under microwave and conventional heating.
5 *Org. Biomol. Chem.* **2004**, *2*, 2809-2813. (c) Lingampalle, D.; Jawale, D.; Waghmare, R.; Mane, R. Ion-
6 ic Liquid-Mediated One-Pot Synthesis for 4-Thiazolidinones. *Synth. Commun.* **2010**, *40*, 2397-2401.
7
8

9 (5) (a) Jagodziński, T. S.; Wesołowska, A.; Sośnicki, J. G. Reactions of Secondary β -Ketoamides
10 with Ethyl Bromoacetate and Ethyl 2-Bromopropionate. The Synthesis of *N*-Substituted 2-
11 Acylmethylidene-1,3-thiazolidin-4-ones. *Polish J. Chem.* **2000**, *74*, 1101-1114. (b) Verma, G. K.;
12 Shukla, G.; Nagaraju, A.; Srivastava, A.; Singh, M. S. *Tetrahedron* **2014**, *70*, 6980-6984. (c) Verma, G.
13 K.; Shukla, G.; Nagaraju, A.; Srivastava, A.; Raghuvanshi, K.; Singh, M. S. DMAP-promoted domino
14 annulation of β -ketoamides with internal alkynes: a highly regioselective access to functionalized 1,3-
15 thiazolidin-4-ones at room temperature. *RSC Adv.* **2014**, *4*, 11640-11647.
16
17
18
19
20
21
22
23
24
25
26
27
28
29

30 (6) (a) Davies, H. M. L. Finding Opportunities from Surprises and Failures: Development of Rhodi-
31 um-Stabilized Donor/Acceptor Carbenes and Their Application to Catalyst-Controlled C–H Function-
32 alization. *J. Org. Chem.* **2019**, *84*, 12722-12745. (b) Ciszewski, L. W.; Rybicka-Jasinska, K.; Gryko, D.
33 Recent developments in photochemical reactions of diazo compounds. *Org. Biomol. Chem.* **2019**, *17*, 432-
34 448. (c) Xiang, Y.; Wang, C.; Ding, Q.; Peng, Y. Diazo Compounds: Versatile Synthons for the Synthesis
35 of Nitrogen Heterocycles *via* Transition Metal-Catalyzed Cascade C–H Activation/Carbene Inser-
36 tion/Annulation Reactions. *Adv. Synth. Catal.* **2019**, *361*, 919-944. (d) Xia, Y.; Qiu, D.; Wang, J. Transi-
37 tion-Metal-Catalyzed Cross-Couplings through Carbene Migratory Insertion. *Chem. Rev.* **2017**, *117*,
38 13810-13889. (e) Liu, L.; Zhang, J. Gold-catalyzed transformations of α -diazocarbonyl compounds: selec-
39 tivity and diversity. *Chem. Soc. Rev.* **2016**, *45*, 506-516. (f) Galkina, O. S.; Rodina, L. L. Photochemical
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

transformations of diazocarbonyl compounds: expected and novel reactions. *Russ. Chem. Rev.* **2016**, *85*, 537-555.

(7) Zhang, Z. K.; Yu, W. Z.; Wu, C. G.; Wang, C. P.; Zhang, Y.; Wang, J. B. Reaction of Diazo Compounds with Difluorocarbene: An Efficient Approach towards 1,1-Difluoroolefins. *Angew. Chem., Int. Ed.* **2016**, *55*, 273-277.

(8) (a) Coquerel, Y.; Rodriguez, J. The Wolff Rearrangement: Tactics, Strategies and Recent Applications in Organic Synthesis. *Molecular Rearrangements in Organic Synthesis*; Wiley, 2015; Chapter 3, pp 59-84. (b) Kirmse, W. 100 Years of the Wolff Rearrangement. *Eur. J. Org. Chem.* **2002**, 2193-2256. (c) Hu, X.; Chen, X.; Shao, Y.; Xie, H.; Deng, Y.; Ke, Z.; Jiang, H.; Zeng, W. Co(III)-Catalyzed Coupling-Cyclization of Aryl C-H Bonds with α -Diazoketones Involving Wolff Rearrangement. *ACS Catal.* **2018**, *8*, 1308-1312.

(9) (a) Allouche, E. M. D.; Charette, A. B. Cyclopropanation Reactions of Semi-stabilized and Non-stabilized Diazo Compounds. *Synthesis* **2019**, *51*, 3947-3963. (b) Ebner, C.; Carreira, E. M. Cyclopropanation Strategies in Recent Total Syntheses. *Chem. Rev.* **2017**, *117*, 11651-11679. (c) Gurmessa, T. G.; Singh, G. S. Recent progress in insertion and cyclopropanation reactions of metal carbenoids from α -diazocarbonyl compounds. *Res. Chem. Intermed.* **2017**, *43*, 6447-6504. (d) Nagode, S. B.; Kant, R.; Rastogi, N. Hantzsch Ester-Mediated Benzannulation of Diazo Compounds under Visible Light Irradiation. *Org. Lett.* **2019**, *21*, 6249-6254.

(10) (a) Marichev, K. O.; Doyle, M. P. Catalytic asymmetric cycloaddition reactions of enoldiazo compounds. *Org. Biomol. Chem.* **2019**, *17*, 4183-4195. (b) Cheng, Q.-Q.; Deng, Y.; Lankelma, M.; Doyle, M. P. Cycloaddition reactions of enoldiazo compounds. *Chem. Soc. Rev.* **2017**, *46*, 5425-5433. (c) Hashimoto, T.; Maruoka, K. Recent Advances of Catalytic Asymmetric 1,3-Dipolar Cycloadditions. *Chem. Rev.* **2015**, *115*, 5366-5412.

(11) (a) Chu, C. K. J.; Rovis, T. Complementary Strategies for Directed C(sp³)-H Functionalization: A Comparison of Transition-Metal-Catalyzed Activation, Hydrogen Atom Transfer, and Carbene/Nitrene Transfer. *Angew. Chem., Int. Ed.* **2018**, *57*, 62-101. (b) Ring, A.; Ford, A.; Maguire, A. R. Substrate and catalyst effects in C-H insertion reactions of α -diazacetamides. *Tetrahedron Lett.* **2016**, *57*, 5399-5406. (c) Santiago, J. V.; Machado, A. H. L. Enantioselective carbenoid insertion into C(sp³)-H bonds. *Beilstein J. Org. Chem.* **2016**, *12*, 882-902. (d) Hu, F.; Xia, Y.; Ma, C.; Zhang, Y.; Wang, J. C-H bond functionalization based on metal carbene migratory insertion. *Chem. Commun.* **2015**, *51*, 7986-7995. (e) Caballero, A.; Díaz-Requejo, M. M.; Fructos, M. R.; Olmos, A.; Urbano, J.; Perez, P. J. Catalytic functionalization of low reactive C(sp³)-H and C(sp²)-H bonds of alkanes and arenes by carbene transfer from diazo compounds. *Dalton Trans.* **2015**, *44*, 20295-20307. (f) Kaur, T.; Wadhwa, P.; Bagchi, S.; Sharma, A. Isocyanide based [4 + 1] cycloaddition reactions: an indispensable tool in multi-component reactions (MCRs). *Chem. Commun.* **2016**, *52*, 6958-6976. (g) Chen, J. R.; Hu, X. Q.; Lu, L. Q.; Xiao, W. J. Formal [4 + 1] Annulation Reactions in the Synthesis of Carbocyclic and Heterocyclic Systems. *Chem. Rev.* **2015**, *115*, 5301-5365. (h) Lam, H.-W.; Man, K. Y.; Chan, W.-W.; Zhou, Z.; Yu, W.-Y. Rhodium(III)-catalyzed formal oxidative [4 + 1] cycloaddition of benzohydroxamic acids and α -diazesters. A facile synthesis of functionalized benzolactams. *Org. Biomol. Chem.* **2014**, *12*, 4112-4116. (i) Empel, C.; Patureau, F. W.; Koenigs, R. M. Visible Light Induced Metal-Free Carbene N-Carbazolation. *J. Org. Chem.* **2019**, *84*, 11316-11322.

(12) (a) Liu, K.; Zhu, C.; Min, J.; Peng, S.; Xu, G.; Sun, J. Stereodivergent Synthesis of N-Heterocycles by Catalyst-Controlled, Activity-Directed Tandem Annulation of Diazo Compounds with Amino Alkynes. *Angew. Chem., Int. Ed.* **2015**, *54*, 12962-12967. (b) Sun, P.; Wu, Y.; Yang, T. Wu, X.; Xu, J.; Lin, A.; Yao, H. Synthesis of Heterocycle-fused Pyridine N-Oxides from Oximes and Diazo Compounds via RhIII-Catalyzed C-H Activation and Annulation. *Adv. Synth. Catal.* **2015**, *357*, 2469-2473. (c) Min, J.; Xu, G.; Sun, J. Synthesis of Six-Membered Carbo-/Heterocycles via Cascade Reac-

tion of Alkynes and Diazo Compounds. *J. Org. Chem.* **2017**, *82*, 5492-5498. (d) Yan, S.; Cao, S.; Sun, J. Synthesis of seven-membered heterocycles *via* copper-catalyzed cross-coupling of terminal alkynes with diazo compounds and sequential Michael addition. *Org. Biomol. Chem.* **2017**, *15*, 5272-5274. (e) Wang, C.; Ding, Q.; Zheng, Q.; Bao, P.; Peng, Y. An efficient route to quinoline-2-carboxylates *via* a rhodium-catalyzed oxidative [5+1] annulation of 2-vinylanilines with α -diazocarbonyl compounds. *Tetrahedron* **2018**, *74*, 348-353. (f) He, M.; Chen, N.; Zhou, T.; Li, Q.; Li, H.; Lang, M.; Wang, J.; Peng, S. Copper-Catalyzed Tandem Cross-Coupling/[2 + 2] Cycloaddition of 1,6-Allenynes with Diazo Compounds to 3-Azabicyclo[5.2.0] Ring Systems. *Org. Lett.* **2019**, *21*, 9559-9563. (g) He, Y.; Lou, J.; Wu, P.; Zhou, Y.-G.; Yu, Z. Copper-Catalyzed Annulative Coupling of S,S-Disubstituted Enones with Diazo Compounds to Access Highly Functionalized Thiophene Derivatives. *J. Org. Chem.* **2020**, *85*, 1044-1053.

(13) (a) Yan, K.; Kong, Y.; Li, B.; Wang, B. Sulfhydryl-Directed Iridium-Catalyzed C–H/Diazo Coupling and Tandem Annulation of Naphthalene-1-thiols. *Org. Lett.* **2019**, *17*, 7000-7003. (b) Empel, C.; Hock, K. J.; Koenigs, R. M. Dealkylative intercepted rearrangement reactions of sulfur ylides. *Chem. Commun.* **2019**, *55*, 338-341.

(14) Paulissen, R.; Reimlinger, H.; Hayez, A.; Hubert, A. J.; Teyssie, P. H. Transition-metal catalyzed reactions of diazocompounds-II insertion in the hydroxylic bond. *Tetrahedron Lett.* **1973**, *14*, 2233-2236.

(15) (a) Taber, D. F.; Amedio, J. C. Jr.; Raman, K. Enantioselective ring construction with control of side-chain stereochemistry. Synthesis of (+)-isoneonepetalactone. *J. Org. Chem.* **1988**, *53*, 2984-2990. (b) Doyle, M. P.; Westrum, L. J.; Wolthuis, N. E.; See, M. M.; Boone, W. P.; Bagheri, V.; Pearson, M. M. Electronic and steric control in carbon-hydrogen insertion reactions of diazoacetoacetates catalyzed by dirhodium(II) carboxylates and carboxamides. *J. Am. Chem. Soc.* **1993**, *115*, 958-964. (c) Taber, D. F.; M. Hennessy, J.; Louey, J.-P. Rhodium-mediated cyclopentane construction can compete with β -hydride elimination: synthesis of (+)-tochuinyl acetate. *J. Org. Chem.* **1992**, *57*, 436-441. (d) Sundberg,

1 R. J.; Baxter, E. W.; Pitts, W. J.; Schofield, R. A.; Nishiguchi, T. Synthesis of the left-hand ring of the
2 antitumor antibiotic CC-1065 by an intramolecular carbenoid addition route. Synthesis and reactivity of
3 4-diazo-4,7-dihydroindol-7-ones and related compounds. *J. Org. Chem.* **1988**, *53*, 5097-5107. (e) Pad-
4 wa, A.; Austin, D. J.; Price, A. T.; Semones, M. A.; Doyle, M. P.; Protopopova, M. N.; Winchester, W.
5 R. Ligand effects on dirhodium(II) carbene reactivities. Highly effective switching between competitive
6 carbenoid transformations. *J. Am. Chem. Soc.* **1993**, *115*, 8669-8680.

7 (16) (a) Chen, Z.; Hu, X.; Huang, J.; Zeng, W. Rhodium(I)-Catalyzed Coupling-Cyclization of C=O
8 Bonds with α -Diazoketones. *Org. Lett.* **2018**, *20*, 3980-3983. (b) Shi, B.; Blake, A. J.; Lewis, W.; Camp-
9 bell, Ian B.; Judkins, B. D.; Moody, C. J. Rhodium Carbene Routes to Oxazoles and Thiazoles. Catalyst
10 Effects in the Synthesis of Oxazole and Thiazole Carboxylates, Phosphonates, and Sulfones. *J. Org.*
11 *Chem.* **2010**, *75*, 152-161. (c) Kim, H.-S.; Kwon, C.; Kim, H. Synthesis of 2-Substituted-4-
12 carbethoxythiazoles. *J. Heterocycl. Chem.* **1995**, *32*, 937-939. (d) King, L. C.; Miller, F. M. The Reac-
13 tion of Diazoketones with Thioamide Derivatives. *J. Am. Chem. Soc.* **1949**, *71*, 367-368.

14 (17) (a) Jagodziński, T. S. Thioamides as Useful Synthons in the Synthesis of Heterocycles. *Chem.*
15 *Rev.* **2003**, *103*, 197-228. (b) Wen, L.-R.; Men, L.-B.; He, T.; Ji, G.-J.; Li, M. Switching Regioselectivi-
16 ty of β -Ketothioamides by Means of Iodine Catalysis: Synthesis of Thiazolyliidenes and 1,4-Dithiines.
17 *Chem. Eur. J.* **2014**, *20*, 5028-5033. (c) Luo, X.; Ge, L.-S.; An, X.-L.; Jin, J.-H.; Wang, Y.; Sun, P.-P.;
18 Deng, W.-P. Regioselective Metal-Free One-Pot Synthesis of Functionalized 2-Aminothiophene Deriva-
19 tives. *J. Org. Chem.* **2015**, *80*, 4611-4617. (d) Li, M.; Kong, X.-J.; Wen, L.-R. Yb(OTf)₃-Mediated Ac-
20 cess to Furans from β -Ketothioamides via Eschenmoser Sulfide Contraction Reaction. *J. Org. Chem.*
21 **2015**, *80*, 11999-12005. (e) Li, C.-X.; Liu, R.-J.; Yin, K.; Wen, L.-R.; Li, M. Synthesis of disulfides teth-
22 ered pyrroles from β -ketothioamides via a bicyclization/ring-opening/oxidative coupling reaction *Org. Bio-*
23 *mol. Chem.* **2017**, *15*, 5820-5823. (f) Man, N.-N.; Wang, J.-Q.; Zhang, L.-M.; Wen, L.-R.; Li, M. Cop-
24 per-Catalyzed Coupling of 2-Siloxy-1-alkenes and Diazocarbonyl Compounds: Approach to Multisub-
25

stituted Furans, Pyrroles, and Thiophenes. *J. Org. Chem.* **2017**, *82*, 5566-5573. (g) Guo, W.-S.; Wen, L.-R.; Li, M. β -Ketothioamides: efficient reagents in the synthesis of heterocycles *Org. Biomol. Chem.* **2015**, *13*, 1942-1953. (h) Zeng, X.-M.; Meng, C.-Yu.; Bao, J.-X.; Xu, D.-C.; Xie, J.-W.; Zhu, W.-D. Enantioselective Construction of Polyfunctionalized Spiroannulated Dihydrothiophenes via a Formal Thio [3 + 2] Cyclization. *J. Org. Chem.* **2015**, *80*, 11521-11528.

(18) (a) Verma, G. K.; Shukla, G.; Nagaraju, A.; Srivastava, A.; Singh, M. S. In(OTf)₃-mediated dehydrative annulation of β -ketothioamides with phenylglyoxal: one-pot access to diversely functionalized pyrrol-2-thiones. *Tetrahedron Lett.* **2014**, *55*, 5182-5185. (b) Nandi, G. C.; Singh, M. S. *p*-TSA/Base-Promoted Propargylation/Cyclization of β -Ketothioamides for the Regioselective Synthesis of Highly Substituted (Hydro)thiophenes. *J. Org. Chem.* **2016**, *81*, 5824-5836. (c) Ansari, M. A.; Yadav, D.; Soni, S.; Srivastava, A.; Singh, M. S. Visible-Light-Mediated Synthesis of 1,2,4-Dithiazolidines from β -Ketothioamides through a Hydrogen-Atom-Transfer Photocatalytic Approach of Eosin Y. *J. Org. Chem.* **2019**, *84*, 5404-5412. (d) Ansari, M. A.; Yadav, D.; Soni, S.; Singh, M. S. Phosphonium ylide catalysis: a divergent diastereoselective approach to synthesize cyclic ketene acetals [thia(zolidines/zinanes)] from β -ketothioamides and dihaloalkanes. *Org. Biomol. Chem.* **2019**, *17*, 9151-9162.

(19) Keri, R. S.; Patil, M. R.; Patil, S. A.; Budagumpi, S. A comprehensive review in current developments of benzothiazole-based molecules in medicinal chemistry. *Eur. J. Med. Chem.* **2015**, *89*, 207-251.

(20) The crystallographic coordinates have been deposited with the Cambridge Crystallographic Data Centre; deposition nos. CCDC 1977512 (**3e**). These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

(21) (a) Zeng, X.-M.; Meng, C.-Yu.; Bao, J.-X.; Xu, D.-C.; Xie, J.-W.; Zhu, W.-D. Enantioselective Construction of Polyfunctionalized Spiroannulated Dihydrothiophenes via a Formal Thio [3 + 2] Cyclization. *J. Org. Chem.* **2015**, *80*, 11521-11528. (b) Boddy, A. J.; Affron, D. P.; Cordier, C. J.; Rivers,

1 E. L.; Spivey, C. A.; Bull, J. A. Rapid Assembly of Saturated Nitrogen Heterocycles in One-Pot: Di-
2 azo-Heterocycle "Stitching" by N-H Insertion and Cyclization. *Angew. Chem., Int. Ed.* **2019**, *58*, 1458-
3 1462. (c) Abed, H. B.; Mammoliti, O.; Bande, O.; Lommen, G. V.; Herdewijn, P. Strategy for the Syn-
4 thesis of Pyridazine Heterocycles and Their Derivatives. *J. Org. Chem.* **2013**, *78*, 7845-7858.
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