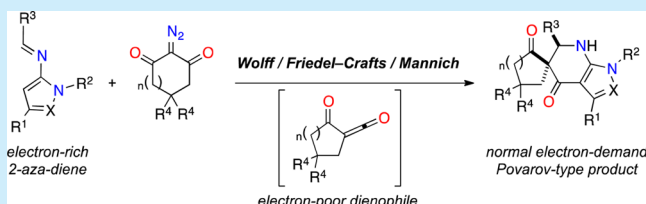


Divergent Chemo-, Regio-, and Diastereoselective Normal Electron-Demand Povarov-Type Reactions with α -Oxo-ketene DienophilesJaime Galvez,[†] Juan-Carlos Castillo,[‡] Jairo Quiroga,[†] Michel Rajzmann,[‡] Jean Rodriguez,^{*,‡} and Yoann Coquerel^{*,‡}[†]Departamento de Química, Universidad del Valle, A.A. 25360, Cali, Colombia[‡]Aix Marseille Université, Centrale Marseille, CNRS, iSm2 UMR 7313, 13397 Marseille, France

Supporting Information

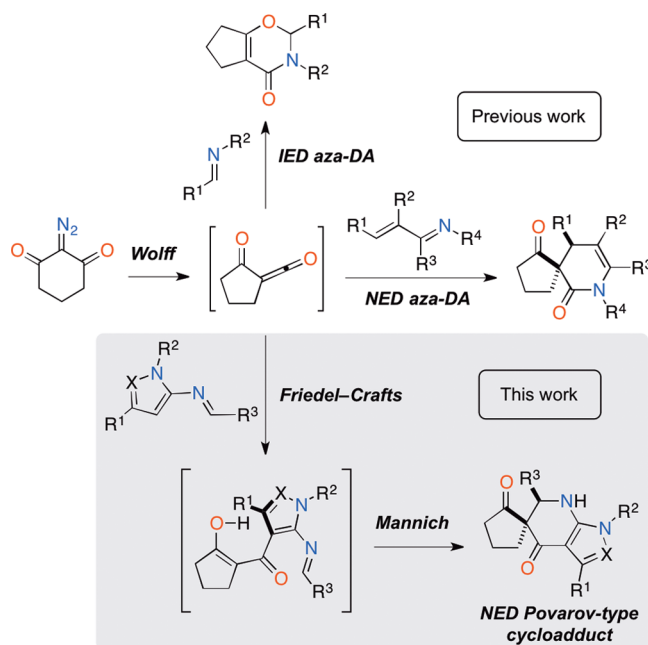
ABSTRACT: The reactions between electron-rich 2-azadienes and α -oxo-ketenes derived from the Wolff rearrangement of 2-diazocycloalkane-1,3-diones chemo- and regioselectively produced spiro hydropyrid-4-ones with good to excellent diastereoselectivities. These reactions are likely to proceed via a domino Wolff/Friedel–Crafts/intramolecular Mannich process. Prolonged domino sequences also allowed the expeditious preparation of a series of pyrazolopyridine and pyridopyrimidine heterocycles.



The stereocontrolled formation of carbon–carbon bonds is at the core of the science of organic synthesis. In the context of economies in synthesis,¹ multiple bond-forming transformations (MBFTs)² allowing for the sequential creation of two or more C–C bonds in a single chemical operation are particularly valuable. As to the formation of six-membered azacycles, the aza-Diels–Alder (aza-DA) strategy, often involving stepwise nonconcerted processes, counts among the most reliable MBFT-based approaches.³ Formal aza-DA reactions generally fall into two main categories according to the electronic properties of the reaction partners: normal electron-demand (NED) aza-DA with electron-rich dienes and electron-impoorished imines and inverse electron-demand (IED) aza-DA with electron-deficient 1- or 2-azadienes and electron-rich olefins.³ It should be noted that only aza-DA cycloadditions with 2-aza-dienes allow for the formation of two C–C bonds. The Povarov reaction, i.e., the IED aza-DA reaction between *N*-aryl imines and electron-rich olefins to give tetrahydroquinolines, is an archetypal example of the reactivity of 2-aza-dienes.^{3c,f} The realization of regio- and stereoselective NED Povarov-type reactions with electron-enriched *N*-aryl imines and electron-poor olefins would greatly expand the scope and possible applications of this strategy. However, this complementary reactivity of *N*-aryl imines has remained unstudied.

α -Oxo-ketenes are electrophilic reactive intermediates with a rich chemistry.⁴ The microwave-assisted Wolff rearrangement of 2-diazo-1,3-diketones is an extremely efficient and convenient source of α -oxo-ketenes,^{5,6} and this technology has recently allowed revisiting their fundamental reactivity. α -Oxo-ketenes normally react with imines as 4π reaction partners in formal IED aza-DA cycloadditions to afford the corresponding oxazinones (Scheme 1, top).⁷ It was recently uncovered that their reactions with electron-rich 1-aza-dienes were following an alternative path, involving formal NED aza-DA cycloadditions in which the

Scheme 1. Known and Proposed Aza-DA Cycloadditions with α -Oxo-ketenes^a



^aIED = inverse electron-demand; aza-DA = aza-Diels–Alder; NED = normal electron-demand.

C=C bond of the α -oxoketene is this time the 2π reaction partner, leading to stereodefined δ -lactam products with successive C–N and C–C bond-forming events (Scheme 1,

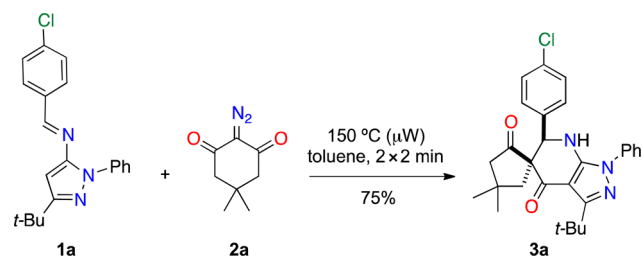
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right).⁸ A similar reactivity was also observed in the 1,3-dipolar cycloadditions of electron-rich hydrazones with α -oxoketenes.⁹ Lately, α -oxoketenes were found to be valuable electrophilic reaction partners in Friedel–Crafts α -ketoacylations of heteroaromatic compounds.¹⁰ Herein, we report regioselective and diastereoselective NED Povarov-type reactions with α -oxoketene dienophiles for the preparation of stereodefined spiro compounds (Scheme 1, gray highlight). This idea was based on the mechanistic hypothesis that a catalyst-free domino Wolff rearrangement/Friedel–Crafts α -ketoacylation/intramolecular Mannich sequence would be possible.

Because of its excellent reactivity in Friedel–Crafts α -ketoacylations¹⁰ and the availability of the corresponding 5-amino derivatives,¹¹ the pyrazolyl group was selected as the prototypical *N*-aryl moiety to prepare electron-rich *N*-aryl imines amenable to the planned transformation.¹² It can be noted that the first Povarov reactions (IED aza-DA) with aminopyrroles and aminopyrazoles were reported only recently.¹³ The present study was initiated with the reaction of the *N*-aryl imine **1a** with the α -oxoketene derived from diazodimedone (**2a**, 1.5 equiv) at 150 °C under microwave irradiation for 6 min. Gratifyingly, the expected spiro hydropyrid-4-one product **3a** resulting from a formal regio- and stereoselective NED Povarov-type reaction was obtained in 65% yield as a single diastereomer. In a modified protocol, the addition of 2.4 equiv of the diazo compound **2a** in two portions afforded the same product **3a** in an increased 75% yield (Scheme 2) together with some dimerization product of the α -oxoketene.^{4a,b}

Scheme 2. Regio- and Diastereoselective NED Povarov-Type Reaction



The scope of the reaction was then investigated with several *N*-pyrazolyl imines and a few cyclic 2-diazo-1,3-diketones (Table 1). The reaction was found to be general, and diversely substituted and functionalized NED Povarov-type spiro products **3** could be stereoselectively prepared. Depending on the substrate, reaction temperatures ranging from 150 to 250 °C were required with reaction times varying from 2 × 2 min to 3 × 15 min (see Supporting Information for experimental details). It was found that the pyrazolyl group well tolerated either alkyl or aryl *R*¹ and *R*² substituents, as well as electron-donating and electron-withdrawing functional groups. In some cases, especially those involving substrates bearing an electron-withdrawing group, minor amounts of the other possible diastereomer were also formed (Table 1, entries 8, 9, and 13).

As a general tendency, more electron-donating substituents on the 2-aza-diene substrate led to higher yield and diastereoselectivity (compare for example Table 1, entries 7, 8, and 9). The reaction appeared more difficult with the α -oxo-ketene derived from the seven-membered diazo compound **2c** (*n* = 2, *R*⁴ = H), requiring the highest temperature and longest reaction time of

Table 1. Generalization of the Reaction^a

entry	3; X, <i>R</i> ¹ , <i>R</i> ² , <i>R</i> ³ , <i>R</i> ⁴ , <i>n</i>	dr ^b	yield (%) ^c
1	3b ; N, <i>t</i> -Bu, Ph, 4-Me-C ₆ H ₄ , Me, 1	>25:1	81
2	3c ; N, <i>t</i> -Bu, Ph, 4-OMe-C ₆ H ₄ , Me, 1	>25:1	70
3	3d ; N, <i>t</i> -Bu, Ph, 3,4-(OCH ₂ O)-C ₆ H ₃ , H, 1	>25:1	79
4	3e ; N, <i>t</i> -Bu, Ph, 4-Me-C ₆ H ₄ , H, 1	>25:1	72
5	3f ; N, <i>t</i> -Bu, Ph, 2-OMe-C ₆ H ₄ , H, 1	>25:1	51
6	3g ; N, Me, Ph, 4-OMe-C ₆ H ₄ , Me, 1	>25:1	76
7	3h ; N, <i>t</i> -Bu, Me, 4-OMe-C ₆ H ₄ , Me, 1	>25:1	80
8	3i ; N, <i>t</i> -Bu, Me, 4-CF ₃ -C ₆ H ₄ , Me, 1	11:1	57
9	3j ; N, <i>t</i> -Bu, Me, 3,5-(CF ₃) ₂ -C ₆ H ₃ , Me, 1	4:1	36
10	3k ; N, <i>t</i> -Bu, Me, 4-OMe-C ₆ H ₄ , H, 2	>25:1	30
11	3l ; N, Ph, Me, 4-OMe-C ₆ H ₄ , Me, 1	>25:1	81
12	3m ; N, 4-Me-C ₆ H ₄ , Me, 4-OMe-C ₆ H ₄ , Me, 1	>25:1	73
13	3n ; N, 4-Cl-C ₆ H ₄ , Me, 4-OMe-C ₆ H ₄ , Me, 1	17:1	63
14	3o ; N, Ph, Ph, 4-OMe-C ₆ H ₄ , Me, 1	>25:1	41
15	3p ; N, 4-F-C ₆ H ₄ , Ph, 4-OMe-C ₆ H ₄ , H, 1	>25:1	31
16	3q ; CH, CN, <i>t</i> -Bu, 4-OMe-C ₆ H ₄ , Me, 1	1.4:1	58

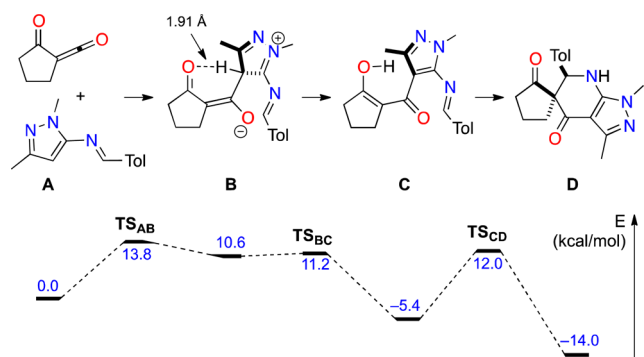
^aReaction conditions: **1b–o** (0.25 mmol), **2a–c** (0.30–0.90 mmol) added in 1–3 portions, anhydrous toluene (2 mL); see Supporting Information for details. ^bDetermined by ¹H NMR analysis of the crude reaction mixture. ^cBased on isolated product after silica gel chromatography.

the study to obtain the desired NED Povarov-type spiro product **3k** in 30% yield (Table 1, entry 10).

The structure of the spiro hydropyrid-4-one **3b** was resolved by X-ray diffraction techniques, which confirmed the chemo-, regio-, and stereochemical outcome of the reaction.¹⁴ Remarkably, the NED Povarov-type reactions described herein allows for the regio- and stereocontrolled formation of two C–C bonds and two contiguous stereogenic carbon atoms, including a challenging “all-carbon” quaternary center.¹⁵ Very interestingly, the reaction could be extended to the pyrrole series in the case of product **3q** (Table 1, entry 16).¹⁶

The postulated mechanism of the reaction was examined by DFT theoretical calculations using the B3LYP functional with the extended base set 6-311++G** to account for long-range interactions (Scheme 3 and Supporting Information). Our preliminary results¹⁷ indicate that the proposed Friedel–Crafts/Mannich sequence is very plausible, with reasonable transition state energies (activation barrier of 13.8 kcal/mol). In the calculated reaction path, the Friedel–Crafts α -ketoacylation step (**A** → **B** → **C**) occurs through a nucleophilic addition/1,5-proton shift involving the ketone oxygen atom of the α -oxoketene.¹⁸ The intramolecular Mannich step (**C** → **D**) would then ensue as a rare example of concerted asynchronous proton transfer/6-enol exo-endo trig cyclization, which dictates the stereochemical outcome of the reaction (Figure 1). It is believed that with electron-impooverished substrates (e.g., in Table 1, entries 8 and 9) having a less basic nitrogen atom of the imine moiety looser transition states of type **TS_{CD}** with weaker stabilizing O–H–N interactions are produced, leading to an erosion of the diastereoselectivity.

When the reactions presented in Table 1 were attempted with the five-membered diazo compound **2d**, the pyrazolopyridines

Scheme 3. Theoretical Study of the Proposed Mechanism^a

^aEnergy profile computed at the B3LYP/6-311++G** level including ZPE correction using the IEFPCM solvation model for toluene.

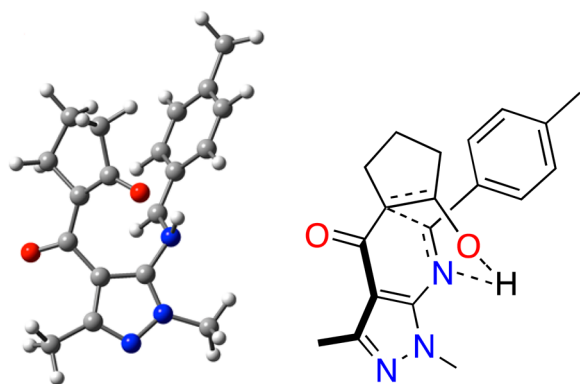
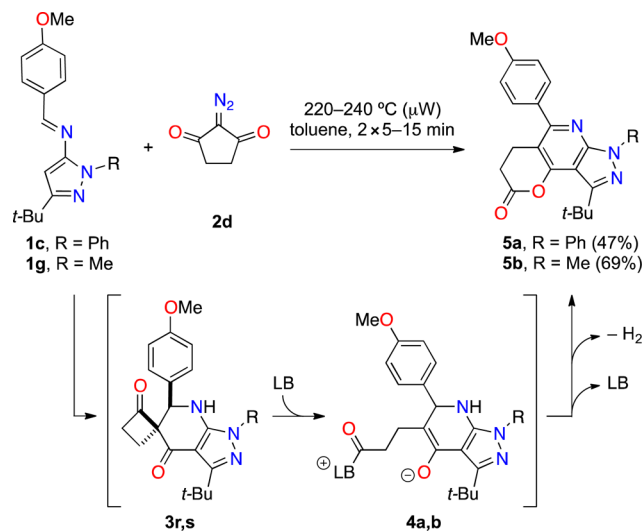


Figure 1. Calculated (left) and schematized (right) transition state TS_{CD}.

products **5a** and **5b** were obtained instead of the expected spiro cyclobutanone products **3r** and **3s** (Scheme 4). Because of the intrinsic ring strain of cyclobutanones, the formation of products **5a** and **5b** is believed to result from a Lewis base catalyzed ring-rearrangement/oxidation domino sequence via the zwitterionic intermediates **4a** and **4b**, respectively. Although no external Lewis base or oxidant was added to the reaction mixtures, the nucleophilic species responsible for the prolonged domino

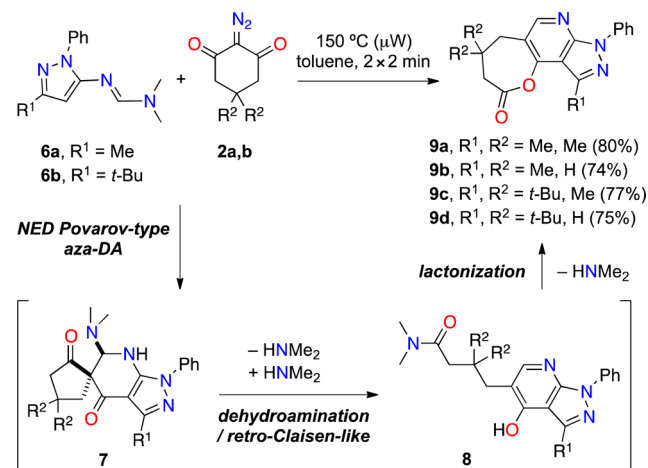
Scheme 4. Prolonged NED Povarov-Type Aza-DA



transformation might be either the starting imines **1** or the intermediates **3** themselves, and oxidation is probably occurring with dissolved oxygen gas.

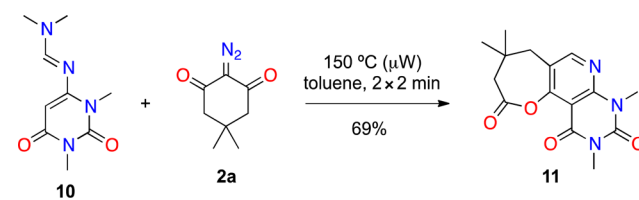
The pyrazolopyridine ring system found in products **5a,b** is a heterocyclic core amenable to applications in medicinal chemistry.¹⁹ It was reasoned that a straightforward access to the related pyrazolopyridine derivatives **9a–d** could rely on the utilization of formimidamides **6a** and **6b** bearing a noninnocent dimethylamino group. This could allow for an original domino sequence initiated by the NED Povarov-type reaction described above to give the intermediate cycloadducts **7**, followed by a dehydroamination/retro-Claisen-like/lactonization cascade (Scheme 5). The *N,N*-dimethyl-*N'*-pyrazolylformimidamides

Scheme 5. Domino Approach to Functionalized Pyrazolopyridine Derivatives



6a and **6b** suitable for the designed MBFT were prepared from the corresponding aminopyrazoles and *N,N*-dimethylformamide dimethyl acetal.^{12e} Satisfactorily, their reactions with the diazo compounds **2a** and **2b** afforded the tricyclic products **9a–d**, probably via the anticipated domino sequence. The structure of product **9c** was secured by X-ray diffraction analysis.¹⁴ The scope of the domino reaction could be successfully extended to the formimidamide **10** derived from 6-aminouracil to provide the functionalized pyridopyrimidine **11** (Scheme 6).

Scheme 6. Domino Approach to a Functionalized Pyridopyrimidine



In summary, normal electron-demand Povarov-type cycloadditions proved possible between some electron-rich *N*-aryl imines and cyclic α -oxo-ketenes. The reaction chemo- and regioselectively produced spiro hydropyrazolopyrid-4-ones with good to excellent diastereoselectivities, and is probably occurring via a domino Wolff/Friedel–Crafts/intramolecular Mannich process forming consecutively two carbon–carbon bonds. In a series of divergent reactions, the ring recombination of the Povarov-type products via prolonged domino processes has led

to fused tricyclic pyrazolopyridine and pyridopyrimidine heterocycles. The elaborate MBFTs described herein (up to 6 steps in the domino process) were built on the functional group density and rich chemistry of α -oxo-ketene reactive intermediates and are expected to open new opportunities for heterocycle and alkaloid synthesis.

■ ASSOCIATED CONTENT

■ Supporting Information

Details for the mechanistic computational study presented in Scheme 3 and Figure 1, detailed experimental procedures, full characterization data, CIFs for compounds **3b** and **9c**, and copies of ^1H and ^{13}C NMR spectra for all compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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(17) A comprehensive theoretical study on the cycloaddition reactions of α -oxo-ketenes with imines and aza-dienes is ongoing and will be reported in due course.

(18) Mono- and bimolecular paths involving the ketene oxygen atom of the α -oxo-ketene were also examined for the proton transfer step (1,3-proton shift) and were found less favorable.

(19) Pyrazolopyridines form a class of nonbenzodiazepine anxiolytic drugs acting as positive allosteric modulators of the GABA_A receptor at the barbiturate binding site; known examples include Cartazolate, Etazolate, ICI-190,622, and Tracazolate. See: Shi, D.; Padgett, W. L.; Hutchinson, K. D.; Moore, S. P.; Daly, J. W. *Drug Dev. Res.* **1997**, *42*, 41.