

Synthesis of Tetracyclic Quinazolinones Using a Visible-Light-Promoted Radical Cascade Approach

Yue-Yue Han, Heng Jiang, Ruzhi Wang, and Shouyun Yu

J. Org. Chem., **Just Accepted Manuscript** • DOI: 10.1021/acs.joc.6b00869 • Publication Date (Web): 03 Jun 2016

Downloaded from <http://pubs.acs.org> on June 3, 2016

Just Accepted

“Just Accepted” manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides “Just Accepted” as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. “Just Accepted” manuscripts appear in full in PDF format accompanied by an HTML abstract. “Just Accepted” manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). “Just Accepted” is an optional service offered to authors. Therefore, the “Just Accepted” Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the “Just Accepted” Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these “Just Accepted” manuscripts.

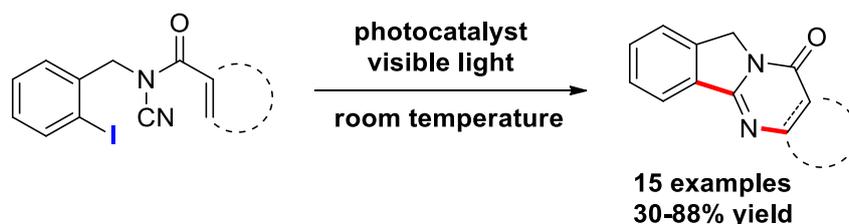
Synthesis of Tetracyclic Quinazolinones Using a Visible-Light-Promoted Radical Cascade Approach

Yue-Yue Han, Heng Jiang, Ruzhi Wang and Shouyun Yu*

State Key Laboratory of Analytical Chemistry for Life Science, School of Chemistry and

Chemical Engineering, Nanjing University, Nanjing 210023, China.

*E-mail: yushouyun@nju.edu.cn

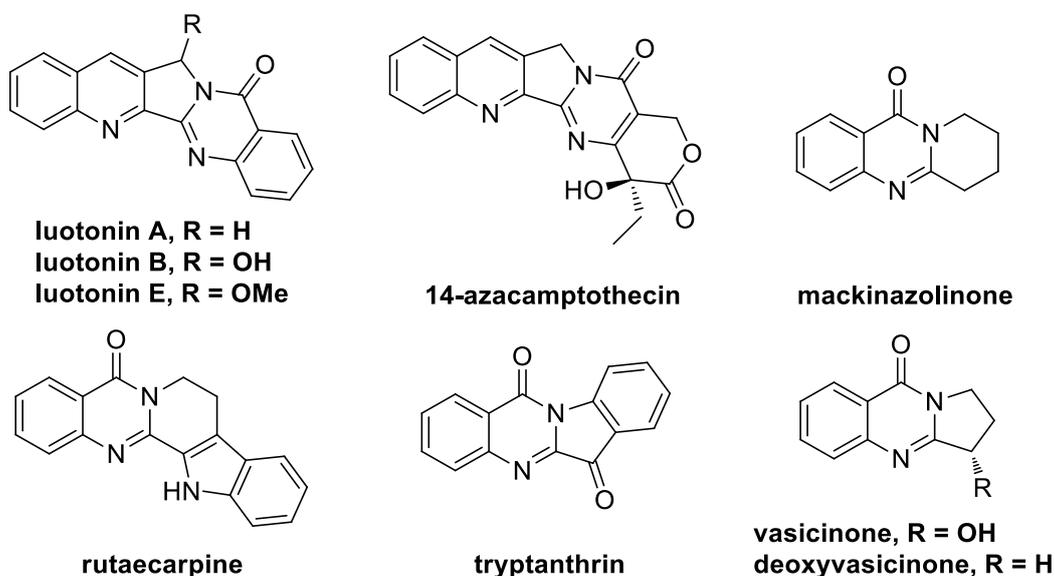


ABSTRACT: A practical approach for the synthesis of tetracyclic pyrroloquinazolines using photoredox strategy has been developed. The visible-light-promoted intramolecular single-electron-transfer (SET) process between photocatalyst and *N*-(2-iodobenzyl)-*N*-acylcyanamides is considered to be involved in this transformation. Targeted pyrroloquinazoline derivatives (15 examples) are presented in good isolated yields (30%-88%).

Tetracyclic pyrroloquinazoline occurs as a key structural entity in dozens of natural products, such as luotonins A and E, mackinazolinone, rutaecarpine and deoxyvasicinone (Scheme 1).¹ This type of natural products disclose a wide range of biological activities including the specific cytotoxicity and potent inhibition to proteases, which would be applicable in the discovery of pharmaceutical candidates.²

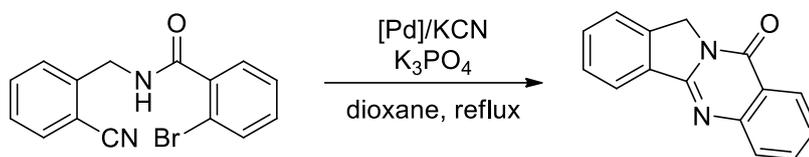
1
2
3
4 Therefore, the synthesis of heterocyclic compounds bearing the tetracyclic
5
6 pyrroloquinazoline scaffold has drawn intensive attention to synthetic chemists.
7
8 Previously, the construction of tetracyclic pyrroloquinazoline scaffold commonly
9
10 focused on the Pd-catalyzed cyclizations.³ For example, Li and co-workers reported a
11
12 Pd-catalyzed approach to prepare tetracyclic pyrroloquinazolines from
13
14 2-bromo-*N*-(2-iodobenzyl)benzamides, which underwent
15
16 cyanation/*N*-addition/*N*-arylation reaction sequence in a two-stage and one-pot
17
18 manner (Scheme 2a).^{3a} The Wu group also reported a decent method to prepare
19
20 tetracyclic pyrroloquinazolines through palladium-catalyzed carbonylation of
21
22 2-bromobenzylamines with 2-bromoanilines (Scheme 2b).^{3b} In addition, tetracyclic
23
24 pyrroloquinazolines could also be prepared through radical approach.^{4,5} Malacria and
25
26 co-workers described the AIBN/*n*-Bu₃SnH-mediated radical cascade cyclization of
27
28 *N*-(2-iodobenzyl)-*N*-acylcyanamides to prepare pyrroloquinazoline-type polycyclic
29
30 compounds (Scheme 2c).^{4a} A concise total synthesis of luotonin A was achieved using
31
32 this strategy. The aforementioned methods are normally conducted at elevated
33
34 temperature or proceed in the presence of toxic reagents (e. g. *n*-Bu₃SnH). These
35
36 limitations considerably restrict not only the operational simplicity towards the
37
38 preparation of this pyrroloquinazoline-type tetracyclic compounds, but also the
39
40 pharmaceutical availability of compounds bearing this characteristic scaffold.
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Scheme 1. Quinazolinone-Containing Biologically Active Natural Products

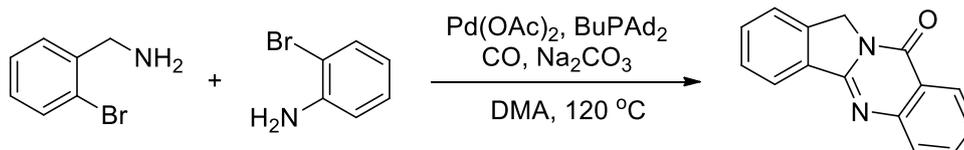
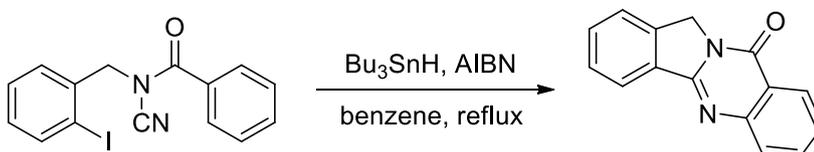
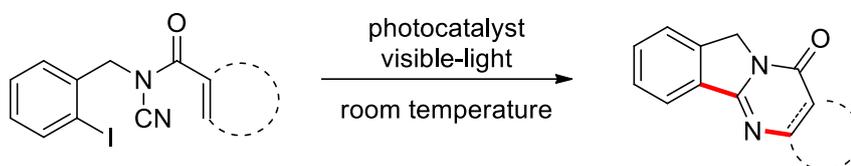


Inspired by these beautiful works, as well as our previous work with respect to the photoredox-catalyzed radical triple bond insertions to construct various heterocycles,⁶ we envisioned that the tetracyclic pyrroloquinazolinanes could be prepared by photoredox-catalyzed intramolecular radical cyanide insertion of *N*-(2-iodobenzyl)-*N*-acylcyanamides (Scheme 2d). Several challenges associated with this approach have to be pointed out, including (1) the inertia of aryl iodides⁷ for the generation of aryl radical species upon photoredox conditions compared with other active aryl radical precursors such as aryl diazonium salts⁸ and diaryliodonium salts,⁹ (2) the inertia of the cyano group as the radical acceptor in photoredox catalysis.¹⁰

Scheme 2. Synthesis of Tetracyclic Quinazolinones

a) Pd-catalyzed sequential cyanation/*N*-addition/*N*-arylation

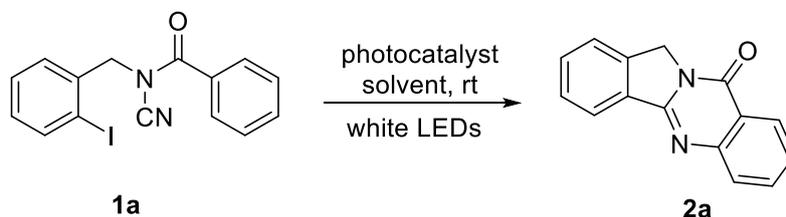
b) Palladium-catalyzed carbonylation

c) AIBN/Bu₃SnH-mediated radical cascade cyclizationd) Visible-light-promoted radical cascade cyclization: *this work*

With these considerations in mind, we initially used *N*-acyl-*N*-(2-iodobenzyl)-cyanamide (**1a**)^{4a} as the model substrate to explore this transformation. To our delight, 30% yield of the desired pyrroloquinazoline **2a** based on ¹H NMR analysis was obtained when a solution of **1a**, TEA and *fac*-Ir(ppy)₃ (2.0 mol%) in MeCN was irradiated by white LED strips for 12 h at room temperature (Table 1, entry 1). The use of inorganic bases such as Na₂HPO₄ and KO^tBu instead of TEA couldn't give any desired product (entries 2-3). This phenomenon suggests that TEA was not only used as a base, but also played as a quencher of photo-excited catalyst (*fac*-Ir(ppy)₃*) in the catalytic cycle. We further screened a range of organic

1
2
3
4 bases including Ph₃N, DIPEA, DBU, DMAP, NMM and TMEDA (entries 4-9).
5
6
7 DIPEA was proved to be the optimal organic base affording **2a** in 69% yield (entry 5).
8
9
10 The employment of other commonly used photocatalysts Ir(ppy)₂(dtbbpy)PF₆,
11
12 Ir(dFCF₃ppy)₂(dtbbpy)PF₆, Ru(bpy)₃Cl₂ and Ru(Phen)₃(PF₆)₂ couldn't give any
13
14 improvement of the yield (entries 10-13). Other solvents, such as DMF, MeOH,
15
16 DMSO and CH₂Cl₂, were not superior to MeCN in this transformation (entries 14-17).
17
18
19 The use of 3.0 equivalent of DIPEA gave comparable yield of **2a** (entry 18).
20
21
22 Satisfactorily, the NMR yield of **2a** could be increased to 87% (83% isolated yield) by
23
24 diluting the reaction system to 0.02 M (entry 19). The reaction couldn't proceed
25
26 without photocatalyst or visible light irradiation (entries 20-21).
27
28
29
30
31
32

33 **Table 1. Reaction Condition Optimization^a**



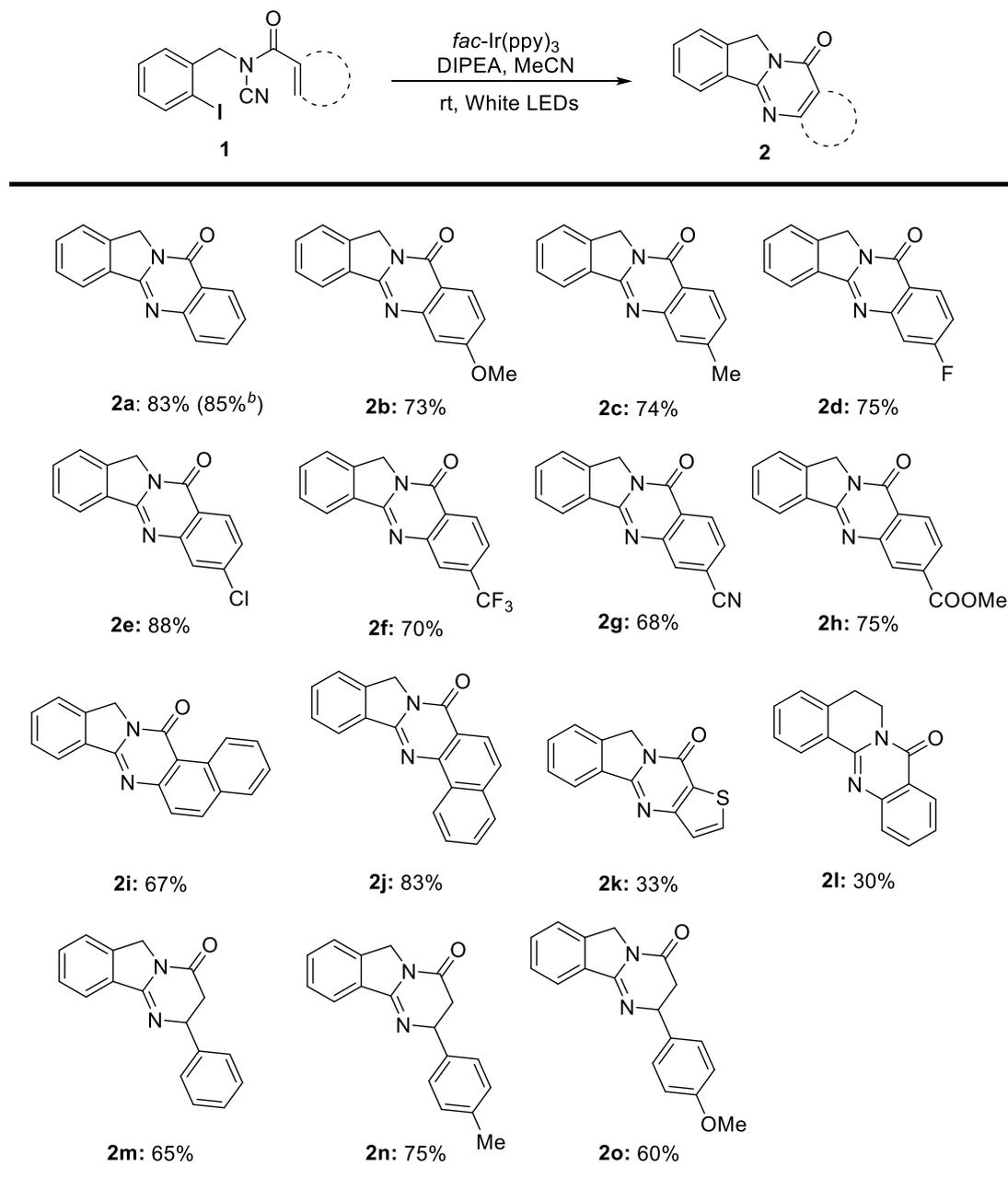
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

entry	photocatalyst	base	solvent	2a / ^b %
1	Ir(ppy) ₃	TEA	MeCN	30
2	Ir(ppy) ₃	Na ₂ HPO ₄	MeCN	0
3	Ir(ppy) ₃	KO ^t Bu	MeCN	0
4	Ir(ppy) ₃	Ph ₃ N	MeCN	0
5	Ir(ppy) ₃	DIPEA	MeCN	69

6	Ir(ppy) ₃	DBU	MeCN	trace
7	Ir(ppy) ₃	DMAP	MeCN	trace
8	Ir(ppy) ₃	NMM	MeCN	9
9	Ir(ppy) ₃	TMEDA	MeCN	38
10	Ir(ppy) ₂ (dtbbpy)PF ₆	DIPEA	MeCN	66
11	Ir(dFCF ₃ ppy) ₂ (dtbbpy)PF ₆	DIPEA	MeCN	18
12	Ru(bpy) ₃ Cl ₂	DIPEA	MeCN	20
13	Ru(Phen)(PF ₆) ₂	DIPEA	MeCN	17
14	Ir(ppy) ₃	DIPEA	DMF	29
15	Ir(ppy) ₃	DIPEA	MeOH	0
16	Ir(ppy) ₃ Cl ₂	DIPEA	DMSO	64
17	Ir(ppy) ₃	DIPEA	CH ₂ Cl ₂	35
18 ^c	Ir(ppy) ₃	DIPEA	MeCN	70
19^d	Ir(ppy)₃	DIPEA	MeCN	87 (83^e)
20 ^f	Ir(ppy) ₃	DIPEA	MeCN	0
21	none	DIPEA	MeCN	0

^aReaction conditions: a solution of **1a** (0.1 mmol), base (0.2 mmol) and photocatalyst (0.002 mmol, 2.0 mol %) in the indicated solvent (1.0 mL) was irradiated by white LED strips at rt for 12 h. ^bThe yields are based on ¹HNMR analysis. ^c0.3 mmol of DIPEA was used. ^d5.0 mL of MeCN was used. ^eIsolated yield. ^fReaction was conducted in the dark.

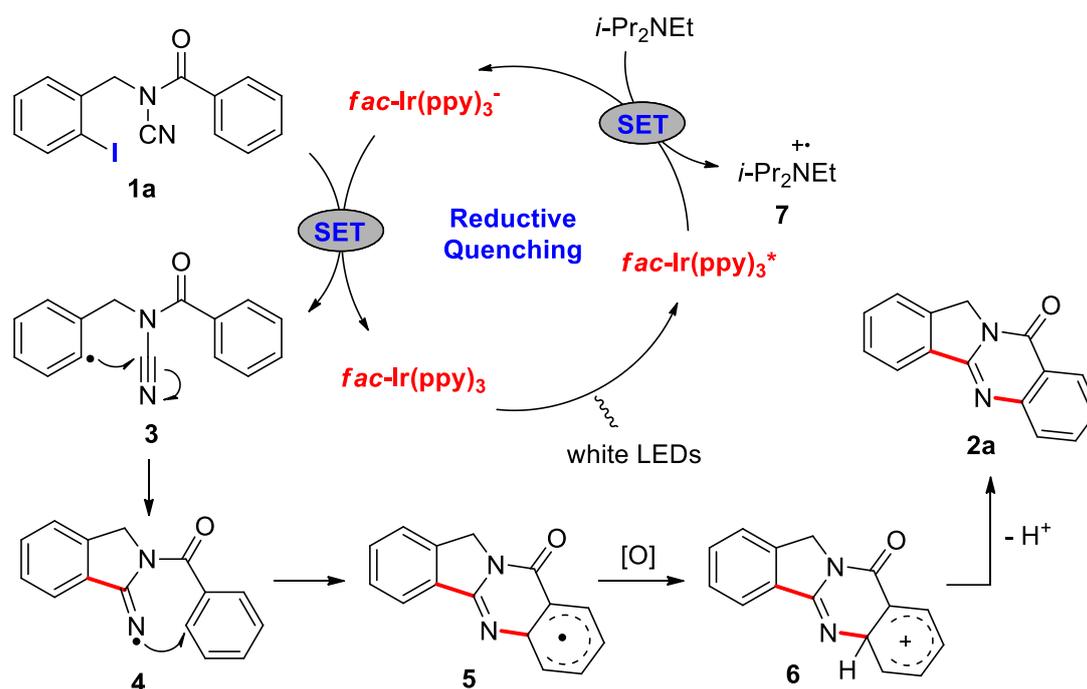
1
2
3
4 With the optimized conditions in hand, we then intended to explore the scope of
5
6 this transformation. In general, the desired cyclization products **2** with 6/5/6/6 ring
7
8 system were achieved in acceptable to excellent yields (Scheme 3). The substituents
9
10 on the aryl of benzoyl motif could be a range of electronic donating or withdrawing
11
12 functional groups (-OMe, -Me, -F, -Cl, -CF₃, -CN, CO₂Me), affording desired
13
14 products **2a-2h** in 68%-88% yields. The aryl of benzoyl motif could also be 1- or
15
16 2-naphthalene (**2i**, 67% and **2j**, 83%) and 1-thiophene (**2k**, 33%). In addition,
17
18 *N*-acylcyanamide derived from 2-iodophenylethanamine was also compatible in this
19
20 transformation to give **2l** with 6/6/6/6 ring system in 30% yield. Cinnamic
21
22 acid-derived cyanamides were also be suitable in this transformation to give tricyclic
23
24 products **2m-2o** in 50%-86% yields, which were not achieved in
25
26 AIBN/*n*-Bu₃SnH-mediated radical cascade cyclization.⁴ The reaction could be scaled
27
28 up to gram scale. When 3 mmol (1.09 g) of **1a** was subjected to standard conditions
29
30 with 1.0 mol% of photocatalyst, the product **2a** could be isolated in 85% yield.
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Scheme 3. Substrate Scope^a

^aReaction conditions: a solution of **1a** (0.2 mmol), DIPEA (0.4 mmol) and *fac*-Ir(ppy)₃ (0.004 mmol, 2.0 mol %) in MeCN (10.0 mL) was irradiated by white LED strips at rt for 12 h. Yields are for the isolated products. ^bThe reaction was running in 5 mmol scale for 24 h, and the loading of *fac*-Ir(ppy)₃ was reduced from 2.0 mol % to 1.0 mol %.

1
2
3
4 To investigate the mechanism of this reaction, a series of emission quenching
5
6
7 experiments were performed to acquire further insight into the photoredox catalytic
8
9
10 cycle. The experiments revealed that it was DIPEA to quench the excited state of
11
12 *fac*-Ir(ppy)₃ (for details, see the Supporting Information). This result suggests that the
13
14 reaction might undergo a reductive quenching mechanism. On the basis of the
15
16 experimental observations, a plausible mechanism is proposed in Scheme 4. The
17
18 catalytic cycle starts from the photo-excitation of *fac*-Ir(ppy)₃ upon visible light
19
20 irradiation to generate the excited-state photocatalyst (*fac*-Ir(ppy)₃^{*}). The reductive
21
22 quenching of *fac*-Ir(ppy)₃^{*} by DIPEA gives *fac*-Ir(ppy)₃⁻ and *N*-centered radical cation
23
24 species **7**.¹¹ Afterwards, single electron transfer (SET) between *N*-acylcyanamide **1a**
25
26 and *fac*-Ir(ppy)₃⁻ can afford an aryl radical intermediate **3**,⁷ which undergoes
27
28 intramolecular radical cyclization to give radical intermediate **5**.¹² Subsequently, one
29
30 electron oxidation of **5** can generate aryl cation intermediate **6**, following the
31
32 aromatization to give tetracyclic pyrroloquinazoline **2a**.⁶
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Scheme 4. Proposed Mechanism



In summary, we have described a practical approach to construct polycyclic pyrroloquinazolines using photoredox-catalyzed intramolecular radical cyanide insertion. This photoredox neutral strategy proceeds under visible light irradiation at room temperature and the experiments are easy to carry out. The hazardous radical initiator AIBN and toxic *n*-Bu₃SnH can be avoided, which renders this protocol particularly valuable in terms of green synthetic chemistry aspect.

Experimental Section

General Information. All reagents and solvents were used without further purification. Thin layer chromatography (TLC) was performed on EMD precoated plates (silica gel 60 F254, Art 5715) and visualized by fluorescence quenching under UV light and by staining with phosphomolybdic acid or potassium permanganate, respectively. Column chromatography was performed on EMD Silica Gel 60 (300–

1
2
3
4 400 Mesh) using a forced flow of 0.5–1.0 bar. ^1H NMR (400 MHz), ^{13}C NMR
5
6
7 (100MHz) and ^{19}F (376MHz) were measured on a 400 M NMR spectrometer.
8
9
10 Chemical shifts are expressed in parts per million (ppm) with respect to the residual
11
12 solvent peak. Coupling constants are reported as Hertz (Hz), signal shapes and
13
14 splitting patterns are indicated as follows: s, singlet; d, doublet; t, triplet; q, quartet; m,
15
16 multiplet. Infrared (IR) spectra were recorded on a IR spectrophotometer and are
17
18 reported as wavenumber (cm^{-1}).
19
20
21

22 23 **General Procedure for Preparation of *N*-acyl-*N*-(2-iodobenzyl)-cyanamides.^{4a}**

24
25 KOH (326 mg, 5.8 mmol) was added to a solution of *N*-(2-iodobenzyl)-cyanamide
26
27 (1.5 g, 5.8 mmol)^{4a} in THF/H₂O (15 mL, v:v = 1:1). The mixture was stirred at
28
29 ambient temperature for 30 min and concentrated under reduced pressure. Toluene (7
30
31 mL) was added to the residue. After cooling to 0 °C, acyl chloride (11 mmol) was
32
33 added slowly. The mixture was then warmed to 25 °C and stirred for another 2 h.
34
35
36 After the reaction was complete (as judged by TLC analysis), the reaction mixture
37
38 was poured into a separatory funnel containing 20 mL of H₂O and 20 mL of CH₂Cl₂.
39
40
41 The layers were separated and the organic layers were extracted with H₂O (2 × 20
42
43 mL). The organic layers were dried over Na₂SO₄ and concentrated under reduced
44
45 pressure after filtration. The crude product was purified by flash chromatography on
46
47 silica gel to afford the desired product.
48
49
50
51
52
53

54
55 The *N*-acyl-*N*-(2-iodobenzyl)-cyanamides derivatives 1a, 1b, 1f, 1g, 1h and 1k are
56
57 known compounds and were synthesized according to the literature.^{4a}
58

59
60 *N*-cyano-*N*-(2-iodobenzyl)-4-methylbenzamide (**1c**). Purification by
chromatography (petroleum ether/EtOAc = 15:1) afforded **1c** as a white solid (1.7 g,

1
2
3
4 79% yield). m.p.: 84—86 ° C; IR (film, cm⁻¹): 2989, 2896, 2230, 1702, 1608, 1567,
5
6 1510, 1453, 1435, 835, 756, 742, 725 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.95–7.90
7
8 (m, 1H), 7.78 (d, *J* = 8.2 Hz, 2H), 7.45–7.37 (m, 2H), 7.29 (d, *J* = 8.0 Hz, 2H), 7.12–
9
10 7.05 (m, 1H), 4.93 (s, 2H), 2.42 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 168.2, 144.4,
11
12 140.2, 135.9, 130.9, 130.7, 129.3, 129.0, 128.8, 127.8, 110.7, 99.3, 55.3, 21.7;
13
14
15
16
17 HRMS-DART *m/z* calcd for C₁₆H₁₄ON₂I⁺ [M+H⁺] 377.0145, found 377.0143.

18
19
20 *N*-cyano-4-fluoro-*N*-(2-iodobenzyl)benzamide (**1d**): Purification by
21
22 chromatography (petroleum ether/EtOAc = 15:1) afforded **1d** as a white solid (1.5 g,
23
24 67% yield). m.p.: 87—89 °C; IR (film, cm⁻¹): 2989, 2895, 2226, 1707, 1603, 1507,
25
26 1440, 844, 752, 745 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.96–7.88 (m, 3H), 7.45–
27
28 7.37 (m, 2H), 7.21–7.13 (m, 2H), 7.13–7.07 (m, 1H), 4.94 (s, 2H); ¹³C NMR (100
29
30 MHz, CDCl₃) δ 167.1, 165.7 (d, *J* = 256.8 Hz) , 140.2, 135.7, 131.7 (d, *J* = 9.4 Hz),
31
32 131.1, 130.9, 128.9, 126.8 (d, *J* = 3.3 Hz), 116.1 (d, *J* = 22.3 Hz), 110.4, 99.4, 55.5;
33
34 ¹⁹F NMR (376 MHz, CDCl₃) δ -103.55; HRMS-DART *m/z* calcd for C₁₅H₁₁ON₂FI⁺
35
36 [M+H⁺] 380.9895, found 380.9893.
37
38
39
40
41
42

43
44 *4*-chloro-*N*-cyano-*N*-(2-iodobenzyl)benzamide (**1e**): Purification by
45
46 chromatography (petroleum ether/EtOAc = 15:1) afforded **1e** as a white solid (1.7 g,
47
48 74% yield). m.p.: 88—89 °C; IR (film, cm⁻¹): 3090, 2975, 2231, 1702, 1590, 1569,
49
50 1489, 845, 744 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.96–7.90 (m, 1H), 7.84–7.79 (m,
51
52 2H), 7.49–7.45 (m, 2H), 7.44–7.39 (m, 2H), 7.14–7.06 (m, 1H), 4.93 (s, 2H); ¹³C
53
54 NMR (100 MHz, CDCl₃) δ 167.2, 140.3, 140.0, 135.6, 131.2, 130.9, 130.3, 129.1,
55
56 129.0, 128.9, 110.2, 99.4, 55.4; HRMS-DART *m/z* calcd for C₁₅H₁₁ON₂ClI⁺ [M+H⁺]
57
58
59
60

1
2
3
4 396.9599, found 396.9599.
5

6
7 *N*-cyano-*N*-(2-iodobenzyl)-1-naphthamide (**1i**): Purification by chromatography
8
9 (petroleum ether/EtOAc = 20:1) afforded **1i** as a white solid (1.7 g, 72% yield). m.p.:
10
11 132–134 °C; IR (film, cm⁻¹): 2989, 2898, 2231, 1723, 1698, 1688, 1558, 1467, 801,
12
13 779, 765, 742 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.08 (d, *J* = 8.4 Hz, 1H), 8.03 (d, *J*
14
15 = 8.3 Hz, 1H), 7.95 (dd, *J* = 7.9, 0.8 Hz, 1H), 7.92–7.88 (m, 1H), 7.86 (dd, *J* = 7.1, 0.8
16
17 Hz, 1H), 7.64–7.58 (m, 1H), 7.58–7.55 (m, 1H), 7.55–7.50 (m, 1H), 7.49 (dd, *J* = 7.7,
18
19 1.6 Hz, 1H), 7.42 (td, *J* = 7.5, 0.9 Hz, 1H), 7.10 (td, *J* = 7.7, 1.7 Hz, 1H), 5.06 (s, 2H);
20
21 ¹³C NMR (100 MHz, CDCl₃) δ 168.4, 140.2, 135.8, 133.6, 132.8, 130.84, 130.82,
22
23 129.98, 128.9, 128.7, 128.5, 128.0, 127.0, 126.9, 124.4, 124.3, 109.7, 99.4, 54.9;
24
25 HRMS-DART *m/z* calcd for C₁₉H₁₄ON₂I⁺ [M+H⁺] 413.0145, found 413.0141.
26
27

28
29
30
31
32
33 *N*-cyano-*N*-(2-iodobenzyl)-2-naphthamide (**1j**): Purification by chromatography
34
35 (petroleum ether/EtOAc = 20:1) afforded **1j** as a white solid (1.8 g, 75% yield). m.p.:
36
37 82–84 °C; IR (film, cm⁻¹): 3055, 2986, 2901, 2235, 1703, 1684, 1628, 1558, 1466,
38
39 833, 752, 734 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 8.43 (s, 1H), 7.92 (t, *J* = 7.4 Hz,
40
41 3H), 7.85 (d, *J* = 8.4 Hz, 2H), 7.62–7.50 (m, 2H), 7.45 (dd, *J* = 7.6, 1.6 Hz, 1H), 7.39
42
43 (t, *J* = 7.5 Hz, 1H), 7.08 (t, *J* = 7.6 Hz, 1H), 4.98 (s, 2H); ¹³C NMR (100 MHz, CDCl₃)
44
45 δ 168.3, 140.2, 135.9, 135.5, 132.1, 131.0, 130.8, 130.4, 129.4, 128.88, 128.89, 128.8,
46
47 127.9, 127.8, 127.3, 124.4, 110.6, 99.5, 55.5; HRMS-DART *m/z* calcd for
48
49 C₁₉H₁₄ON₂I⁺ [M+H⁺] 413.0145, found: 413.0145.
50
51
52
53
54
55

56
57 *N*-cyano-*N*-(2-iodophenethyl)benzamide (**1l**): Purification by chromatography
58
59 (petroleum ether/EtOAc = 20:1) afforded **1l** as a white solid (1.6 g, 76% yield). IR
60

(film, cm^{-1}): 3085, 3055, 2980, 2230, 1702, 1591, 1568, 1489, 1472, 845, 744 cm^{-1} ;
 ^1H NMR (400 MHz, CDCl_3) δ 7.86 (d, $J = 7.8$ Hz, 1H), 7.76–7.70 (m, 2H) 7.61–7.54
(m, 1H), 7.50–7.42 (m, 2H), 7.36–7.28 (m, 2H), 7.70–6.93 (m, 1H), 4.02 (t, $J = 7.4$
Hz, 2H), 3.24 (t, $J = 7.4$ Hz, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 168.2, 140.0, 139.4,
133.2, 130.9, 130.3, 129.1, 128.7, 128.62, 128.59, 110.9, 100.4, 47.5, 38.6;
HRMS-DART m/z calcd for $\text{C}_{16}\text{H}_{14}\text{ON}_2\text{I}^+$ [$\text{M}+\text{H}^+$] 377.0145, found: 377.0145.

N-cyano-*N*-(2-iodobenzyl)cinnamamide (**1m**): Purification by chromatography
(petroleum ether/EtOAc = 40:1) afforded **1m** as a white solid (1.2 g, 53% yield). m.p.:
112–114 °C; IR (film, cm^{-1}): 2975, 2930, 2227, 1680, 1617, 1598, 1570, 1468, 745,
748, 687 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.98–7.99 (m, 2H), 7.64–7.58 (m, 2H),
7.47–7.36 (m, 5H), 7.16 (d, $J = 15.4$ Hz, 1H), 7.10–7.05 (m, 1H), 4.91 (s, 2H); ^{13}C
NMR (100 MHz, CDCl_3) δ 164.5, 149.1, 140.1, 135.9, 133.5, 131.5, 130.6, 130.2,
129.1, 128.83, 128.79, 113.6, 110.1, 99.2, 54.2; HRMS-DART m/z calcd for
 $\text{C}_{17}\text{H}_{14}\text{ON}_2\text{I}^+$ [$\text{M}+\text{H}^+$] 389.0145, found: 389.0145.

(*E*)-*N*-cyano-*N*-(2-iodobenzyl)-3-(*p*-tolyl)acrylamide (**1n**): Purification by
chromatography (petroleum ether/EtOAc = 40:1) afforded **1n** as a white solid (1.4 g,
60% yield). m.p.: 129–131 °C; IR (film, cm^{-1}): 2968, 2920, 2228, 1695, 1619, 1604,
1568, 1456, 1330, 813, 747, 718 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.91 (dd, $J =$
11.6, 3.6 Hz, 2H), 7.50 (d, $J = 8.0$ Hz, 2H), 7.42 – 7.32 (m, 2H), 7.23 (d, $J = 7.9$ Hz,
2H), 7.15 – 7.02 (m, 2H), 4.91 (s, 2H), 2.40 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ
164.7, 149.2, 142.3, 140.1, 136.0, 130.9, 130.6, 130.1, 129.8, 128.9, 128.8, 112.5,
110.2, 99.9, 54.1, 21.6; HRMS-DART m/z calcd for $\text{C}_{18}\text{H}_{16}\text{ON}_2\text{I}^+$ [$\text{M}+\text{H}^+$] 403.0302,

found: 403.0302.

(*E*)-*N*-cyano-*N*-(2-iodobenzyl)-3-(4-methoxyphenyl)acrylamide (**1o**): Purification by chromatography (petroleum ether/EtOAc = 20:1) afforded **1o** as a white solid (1.6 g, 67% yield). m.p.: 135—137 °C; IR (film, cm⁻¹): 2996, 2928, 2228, 1688, 1616, 1599, 1571, 1508, 1458, 1344, 1289, 821, 745 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.93–7.84 (m, 2H), 7.56 (d, *J* = 8.7 Hz, 2H), 7.43–7.31 (m, 2H), 7.09–7.04 (m, 1H), 7.01 (d, *J* = 15.3 Hz, 1H), 6.93 (d, *J* = 8.7 Hz, 2H), 4.89 (s, 2H), 3.85 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.8, 162.4, 148.8, 140.1, 136.1, 130.7, 130.5, 130.0, 128.8, 126.3, 114.6, 110.9, 110.3, 99.2, 55.5, 54.1; HRMS-DART *m/z* calcd for C₁₈H₁₆O₂N₂I⁺ [M+H⁺] 419.0251, found 419.0251.

General Procedure for the Synthesis of Tetracycle Quinazolinones. A 10 mL round bottom flask equipped with a rubber septum and magnetic stir bar was charged with *N*-cyano-*N*-(2-iodobenzyl)benzamide **1a** (0.2 mmol, 1.0 equiv), DIPEA (0.4 mmol, 2.0 equiv) and Ir(ppy)₃ (0.004 mmol, 0.02 equiv). The flask was evacuated and backfilled with nitrogen for 3 times. MeCN (10.0 mL) was added with a syringe under nitrogen. The mixture was then irradiated by white LED strips. After the reaction was complete (as judged by TLC analysis), the mixture was concentrated under reduced pressure after filtration. The crude product was purified by flash chromatography on silica gel (petroleum ether/ethyl acetate = 10:1).

Isoindolo[1,2-*b*]quinazolin-10(12*H*)-one (**2a**).^{4a} Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2a** as a yellow solid (38.8 mg, 83% yield). ¹H NMR (400 MHz, CDCl₃) δ 8.37 (d, *J* = 7.8 Hz, 1H), 8.17 (d, *J* = 7.5 Hz, 1H),

1
2
3
4 7.86–7.74 (m, 2H), 7.65–7.54 (m, 3H), 7.49 (t, $J = 7.4$ Hz, 1H), 5.14 (s, 2H); ^{13}C
5
6
7 NMR (100 MHz, CDCl_3) δ 160.7, 155.0, 149.5, 139.6, 134.3, 132.7, 132.4, 128.9,
8
9
10 127.4, 126.5, 126.4, 123.54, 123.51, 120.6, 49.8.

11
12 *7-methoxyisoindolo[1,2-b]quinazolin-10(12H)-one* (**2b**).^{4c} Purification by
13
14 chromatography (petroleum ether/EtOAc = 10:1) afforded **2b** as a yellow solid (38.6
15
16 mg, 73% yield). ^1H NMR (400 MHz, CDCl_3) δ 8.27 (d, $J = 8.8$ Hz, 1H), 8.17 (d, $J =$
17
18 7.6 Hz, 1H), 7.66–7.11 (m, 2H), 7.61–7.55 (m, 1H), 7.22 (d, $J = 2.4$ Hz, 1H), 7.06 (dd,
19
20 $J = 8.8, 2.4$ Hz, 1H), 5.13 (s, 2H), 3.95 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 164.6,
21
22 160.3, 155.6, 151.8, 139.9, 132.7, 132.3, 128.9, 127.9, 123.5, 123.4, 116.6, 114.2,
23
24 108.0, 55.7, 49.7.

25
26
27
28
29
30
31 *7-methylisoindolo[1,2-b]quinazolin-10(12H)-one* (**2c**).^{3b} Purification by
32
33 chromatography (petroleum ether/EtOAc = 10:1) afforded **2c** as a yellow solid (36.7
34
35 mg, 74%). ^1H NMR (400 MHz, CDCl_3) δ 8.25 (d, $J = 8.1$ Hz, 1H), 8.20 (d, $J = 7.6$ Hz,
36
37 1H), 7.66–7.60 (m, 3H), 7.59–7.54 (m, 1H), 7.31 (dd, $J = 8.1, 0.9$ Hz, 1H), 5.13 (s,
38
39 2H), 2.52 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 160.5, 155.1, 149.3, 145.4, 139.7,
40
41 132.6, 132.4, 128.9, 128.1, 127.0, 126.3, 123.7, 123.5, 118.1, 49.8, 21.9.

42
43
44
45
46
47 *7-fluoroisoindolo[1,2-b]quinazolin-10(12H)-one* (**2d**).^{3b} Purification by
48
49 chromatography (petroleum ether/EtOAc = 10:1) afforded **2d** as a yellow solid (37.8
50
51 mg, 75%). ^1H NMR (400 MHz, CDCl_3) δ 8.39 (t, $J = 7.3$ Hz, 1H), 8.18 (d, $J = 7.3$ Hz,
52
53 1H), 7.73–7.56 (m, 3H), 7.48 (d, $J = 9.5$ Hz, 1H), 7.21 (t, $J = 7.7$ Hz, 1H), 5.16 (s,
54
55 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 166.5 (d, $J = 253.5$ Hz), 160.0, 156.2, 151.7 (d, J
56
57 = 13.3 Hz), 139.8, 132.7, 132.4, 129.1 (d, $J = 11.0$ Hz), 129.0, 123.8, 123.6, 117.3,
58
59
60

1
2
3
4 115.1 (d, $J = 23.5$ Hz), 112.7 (d, $J = 22.2$ Hz), 49.9. ^{19}F NMR (376 MHz, CDCl_3) δ –
5
6
7 103.62;

8
9 *7-chloroisoindolo[1,2-*b*]quinazolin-10(12*H*)-one* (**2e**). Purification by
10 chromatography (petroleum ether/EtOAc = 10:1) afforded **2e** as a yellow solid (47.3
11 mg, 88%). m.p.: 236–238 °C; IR (film, cm^{-1}): 2974, 2920, 1668, 1619, 1597, 1549,
12 1454, 878, 773 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 8.29 (d, $J = 8.5$ Hz, 1H), 8.16 (d,
13 $J = 7.6$ Hz, 1H), 7.81 (d, $J = 1.9$ Hz, 1H), 7.69–7.56 (m, 3H), 7.43 (dd, $J = 8.5, 1.9$ Hz,
14 1H), 5.14 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3) δ 160.1, 156.1, 150.5, 140.5, 139.8,
15 132.7, 132.4, 129.0, 127.9, 127.0, 126.9, 123.7, 123.5, 119.0, 49.9; HRMS-DART m/z
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
calcd for $\text{C}_{15}\text{H}_{10}\text{ON}_2\text{Cl}^+$ [$\text{M}+\text{H}^+$] 269.0476, found: 269.0476.

*7-(trifluoromethyl)isoindolo[1,2-*b*]quinazolin-10(12*H*)-one* (**2f**).^{3b} Purification by
chromatography (petroleum ether/EtOAc = 10:1) afforded **2f** as a yellow solid (42.3
mg, 70%). ^1H NMR (400 MHz, CDCl_3) δ 8.47 (d, $J = 8.3$ Hz, 1H), 8.18 (d, $J = 7.6$ Hz,
1H), 8.11 (s, 1H), 7.74–7.56 (m, 4H), 5.17 (s, 2H); ^{13}C NMR (100 MHz, CDCl_3)
 δ 159.8, 156.2, 149.5, 139.7, 135.8 (q, $J = 32.9$ Hz), 132.9, 132.2, 129.2, 127.6, 124.9
(q, $J = 4.0$ Hz), 123.8, 123.6, 123.5 (q, $J = 273.7$ Hz), 122.8, 122.3 (q, $J = 4.1$ Hz),
50.0. ^{19}F NMR (376 MHz, CDCl_3) δ –63.11.

*10-oxo-10,12-dihydroisoindolo[1,2-*b*]quinazoline-7-carbonitrile* (**2g**).^{4a}
Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2g** as a
yellow solid (35.3 mg, 68%). ^1H NMR (400 MHz, CDCl_3) δ 8.47 (d, $J = 8.1$ Hz, 1H),
8.20 (d, $J = 7.5$ Hz, 1H), 8.15 (s, 1H), 7.72–7.63 (m, 4H), 5.19 (s, 2H); ^{13}C NMR (100
MHz, CDCl_3) δ 159.6, 156.6, 149.6, 139.8, 133.2, 132.3, 132.2, 129.3, 128.1, 127.8,

1
2
3
4 124.0, 123.6, 123.5, 117.9, 117.6, 50.1.
5

6 *Methyl 10-oxo-10,12-dihydroisoindolo[1,2-b]quinazoline-7-carboxylate (2h).*^{4a}
7

8
9 Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2h** as a
10 yellow solid (43.8 mg, 75%). ¹H NMR (400 MHz, CDCl₃) δ 8.51 (s, 1H), 8.43 (d, *J* =
11 8.3 Hz, 1H), 8.20 (d, *J* = 7.5 Hz, 1H), 8.09 (d, *J* = 8.3 Hz, 1H), 7.72–7.56 (m, 3H),
12 5.18 (s, 2H), 4.00 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 166.2, 160.2, 155.7, 149.5,
13 139.7, 135.4, 132.7, 132.5, 129.3, 129.1, 126.8, 126.4, 123.7, 123.6, 123.5, 52.6, 49.9.
14
15
16
17
18
19
20
21

22 *Benzo[f]isoindolo[1,2-b]quinazolin-14(12H)-one (2i).* Purification by
23 chromatography (petroleum ether/EtOAc = 10:1) afforded **2i** as a yellow solid (38.1
24 mg, 67%). m.p.: 247–249 °C; IR (film, cm⁻¹): 2989, 2920, 1656, 1621, 1609, 1592,
25 1552, 1470, 836, 784, 724 cm⁻¹; ¹H NMR (400 MHz, DMSO) δ 9.99 (d, *J* = 8.6 Hz,
26 1H), 8.34 (d, *J* = 8.9 Hz, 1H), 8.16 (d, *J* = 7.6 Hz, 1H), 8.10 (d, *J* = 7.7 Hz, 1H), 7.86
27 (dd, *J* = 8.2, 5.1 Hz, 2H), 7.83–7.74 (m, 2H), 7.71–7.64 (m, 2H), 5.31 (s, 2H); ¹³C
28 NMR (100 MHz, DMSO) δ 160.6, 156.1, 152.0, 141.4, 135.9, 132.9, 132.7, 131.8,
29 131.2, 129.2, 129.01, 128.96, 126.9, 126.8, 126.7, 124.7, 123.3, 113.6, 51.1;
30 HRMS-DART *m/z* calcd for C₁₉H₁₃ON₂⁺ [M+H⁺] 285.1022, found: 285.1022.
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45

46 *Benzo[h]isoindolo[1,2-b]quinazolin-7(9H)-one (2j).* Purification by
47 chromatography (petroleum ether/EtOAc = 10:1) afforded **2j** as a yellow solid (47.2
48 mg, 83%). m.p.: 223–225 °C; IR (film, cm⁻¹): 2919, 2850, 1654, 1651, 1610, 1556,
49 1469, 763, 735, 721 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 9.51–8.94 (m, 1H), 8.33–
50 8.21 (m, 2H), 7.92–7.86 (m, 1H), 7.81 (d, *J* = 8.7 Hz, 1H), 7.73–7.66 (m, 2H), 7.65–
51 7.53 (m, 3H), 5.15 (s, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 160.7, 154.9, 148.4, 139.7,
52
53
54
55
56
57
58
59
60

1
2
3
4 136.2, 133.1, 132.2, 130.2, 129.1, 128.8, 127.8, 126.7, 126.6, 125.3, 123.6, 123.4,
5
6
7 121.8, 116.8, 50.0; HRMS-DART m/z calcd for $C_{19}H_{13}ON_2^+$ $[M+H^+]$ 285.1022, found:
8
9 285.1023.

10
11 *Thieno[3',2':4,5]pyrimido[2,1-a]isoindol-11(9H)-one* (**2k**).^{4a} Purification by
12
13 chromatography (petroleum ether/EtOAc = 10:1) afforded **2k** as a yellow solid (15.8
14
15 mg, 33%). ¹H NMR (400 MHz, CDCl₃) δ 8.21 (d, J = 7.6 Hz, 1H), 7.85 (d, J = 5.3 Hz,
16
17 1H), 7.67–7.63 (m, 2H), 7.62–7.56 (m, 1H), 7.48 (d, J = 5.3 Hz, 1H), 5.16 (s, 2H); ¹³C
18
19 NMR (100 MHz, CDCl₃) δ 157.3, 157.0, 156.6, 139.7, 134.6, 132.6, 131.8, 129.1,
20
21 124.7, 123.7, 123.6, 121.3, 50.0.

22
23 *5,6-dihydro-8H-isoquinolino[1,2-b]quinazolin-8-one* (**2l**).^{4b} Purification by
24
25 chromatography (petroleum ether/EtOAc = 10:1) afforded **2l** as a yellow solid (14.9
26
27 mg, 30%). ¹H NMR (400 MHz, CDCl₃) δ 8.54–8.46 (m, 1H), 8.36–8.27 (m, 1H),
28
29 7.94–7.72 (m, 2H), 7.55–7.39 (m, 3H), 7.29 (d, J = 7.2 Hz, 1H), 4.42 (t, J = 6.5 Hz,
30
31 2H), 3.11 (t, J = 6.5 Hz, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 161.7, 149.5, 147.6,
32
33 137.1, 134.3, 131.8, 129.4, 128.1, 127.7, 127.53, 127.50, 126.9, 126.6, 120.7, 39.6,
34
35 27.4.

36
37 *2-phenyl-2,6-dihydropyrimido[2,1-a]isoindol-4(3H)-one* (**2m**).¹³ Purification by
38
39 chromatography (petroleum ether/EtOAc = 10:1) afforded **2m** as a yellow solid (34.1
40
41 mg, 65%). ¹H NMR (400 MHz, CDCl₃) δ 7.85 (dd, J = 7.9, 1.1 Hz, 1H), 7.44–7.35 (m,
42
43 3H), 7.35–7.27 (m, 2H), 7.26–7.22 (m, 1H), 7.04–6.92 (m, 2H), 4.92 (s, 2H), 4.18 (dd,
44
45 J = 9.8, 6.0 Hz, 1H), 3.25 (dd, J = 18.2, 9.8 Hz, 1H), 2.84 (dd, J = 18.2, 5.9 Hz, 1H);
46
47 ¹³C NMR (100 MHz, CDCl₃) δ 175.0, 169.6, 139.7, 138.2, 137.3, 129.5, 129.0, 128.3,
48
49
50
51
52
53
54
55
56
57
58
59
60

1
2
3
4 128.2, 127.8, 126.4, 97.9, 48.3, 45.1, 38.2.
5
6

7 *2-(p-tolyl)-2,6-dihydropyrimido[2,1-a]isoindol-4(3H)-one (2n)*. Purification by
8
9 chromatography (petroleum ether/EtOAc = 10:1) afforded **2n** as a yellow solid (47.5
10 mg, 86%). IR (film, cm^{-1}): 3288, 2973, 2920, 1743, 1648, 1514, 1430, 1399, 1346,
11 818, 733 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.84 (dd, $J = 7.9, 1.0$ Hz, 1H), 7.29 (td,
12 $J = 7.7, 1.0$ Hz, 1H), 7.20 (d, $J = 7.9$ Hz, 2H), 7.12 (d, $J = 8.1$ Hz, 2H), 7.03–6.91 (m,
13 2H), 4.87 (s, 2H), 4.07 (dd, $J = 9.8, 6.2$ Hz, 1H), 3.20 (dd, $J = 18.2, 9.9$ Hz, 1H), 2.80
14 (dd, $J = 18.2, 6.2$ Hz, 1H), 2.36 (s, 3H); ^{13}C NMR (100 MHz, CDCl_3) δ 175.1, 147.1,
15 139.7, 138.1, 137.5, 130.1, 129.9, 129.0, 128.3, 127.7, 126.3, 97.9, 48.2, 44.9, 38.3,
16 21.1; HRMS-DART m/z calcd for $\text{C}_{18}\text{H}_{17}\text{ON}_2^+$ [$\text{M}+\text{H}^+$] 277.1335, found: 277.1335.
17
18
19
20
21
22
23
24
25
26
27
28
29

30 *2-(4-methoxyphenyl)-2,6-dihydropyrimido[2,1-a]isoindol-4(3H)-one (2o)*.
31

32 Purification by chromatography (petroleum ether/EtOAc = 10:1) afforded **2o** as a
33 yellow solid (29.2 mg, 50%). IR (film, cm^{-1}): 3289, 2966, 2918, 1734, 1648, 1609,
34 1583, 1512, 1430, 1348, 1249, 832, 745 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.85 (d,
35 $J = 7.7$ Hz, 1H), 7.29 (t, $J = 7.3$ Hz, 1H), 7.15 (d, $J = 8.6$ Hz, 2H), 6.98 (t, $J = 7.2$ Hz,
36 2H), 6.92 (d, $J = 8.7$ Hz, 2H), 4.87 (s, 2H), 4.07 (dd, $J = 9.9, 6.4$ Hz, 1H), 3.82 (s, 3H),
37 3.20 (dd, $J = 18.2, 9.9$ Hz, 1H), 2.79 (dd, $J = 18.2, 6.4$ Hz, 1H); ^{13}C NMR (101 MHz,
38 CDCl_3) δ 174.9, 170.6, 159.5, 139.7, 137.1, 129.8, 129.1, 129.0, 128.4, 126.6, 114.8,
39 97.9, 55.4, 48.6, 44.1, 38.2. HRMS-DART m/z calcd for $\text{C}_{18}\text{H}_{17}\text{O}_2\text{N}_2^+$ [$\text{M}+\text{H}^+$]
40 293.1285, found: 293.1284.
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60

Acknowledgments

1
2
3
4 Financial support from the National Natural Science Foundation of China (21472084
5
6 and 81421091), the Qing Lan Project of Jiangsu Province and Shanghai Institute of
7
8 Organic Chemistry, CAS is acknowledged.
9
10

11 12 13 **Supporting Information** 14

15
16 The Supporting Information is available free of charge on the ACS Publication
17
18 website at DOI:
19

20
21 Characterization of products (copies of ^1H , ^{13}C , and ^{19}F NMR spectra)
22
23
24

25 26 **References** 27

28 (1) (a) Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. *Heterocycles* **1997**, *46*, 541. (b)
29
30 Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. *Heterocycles* **1999**, *51*, 1883. (c) Ma,
31
32 Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. *Phytochemistry* **2000**, *53*, 1075. (d) Cagir, A.;
33
34 Jones, S. H.; Gao, R.; Eisenhauer, B. M.; Hecht, S. M. *J. Am. Chem. Soc.* **2003**, *125*,
35
36 13628. (e) Cheng, K.; Rahier, N. J.; Eisenhauer, B. M.; Thomas, S. J.; Gao, R.; Hecht,
37
38 S. M. *J. Am. Chem. Soc.* **2005**, *127*, 838. (f) Rahier, N. J.; Cheng, K.; Gao, R.;
39
40 Eisenhauer, B. M.; Hecht, S. M. *Org. Lett.* **2005**, *7*, 835. (g) Elban, M. A.; Sun, W.;
41
42 Eisenhauer, B. M.; Gao, R.; Hecht, S. M. *Org. Lett.* **2006**, *8*, 3513.
43
44
45
46
47
48

49 (2) (a) Thomas, C. J.; Rahier, N. J.; Hecht, S. M. *Bioorg. Med. Chem.* **2004**, *12*, 1585.
50
51 (b) Ma, Z.-Z.; Hano, Y.; Nomura, T.; Chen, Y.-J. *Bioorg. Med. Chem. Lett.* **2004**, *14*,
52
53 1193. (c) Michael, J. P. *Nat. Prod. Rep.* **2004**, *21*, 650. (d) Lee, E. S.; Park, J. G.; Kim,
54
55 S. I.; Jahng, Y. *Heterocycles* **2006**, *68*, 151.
56
57
58
59

60 (3) For Pd-catalyzed cyclization to prepare tetracyclic quinazolinones, see: (a) Ju,

1
2
3
4 Y.; Liu, F.; Li, C. *Org. Lett.* **2009**, *11*, 3582. (b) Shen, C.; Man, N. Y. T.; Stewart, S.;
5
6
7 Wu, X.-F. *Org. Biomol. Chem.* **2015**, *13*, 4422.

8
9 (4) For radical approaches to prepare tetracyclic quinazolinones, see: (a) Servais, A.;
10
11
12 Azzouz, M.; Lopes, D.; Courillon, C.; Malacria, M. *Angew. Chem., Int. Ed.* **2007**, *46*,
13
14
15 576. (b) Bowman, W. R.; M. Elsegood, R. J.; Stein, T.; Weaver, G. W. *Org. Biomol.*
16
17
18 *Chem.* **2007**, *5*, 103. (c) Beaume, A.; Courillon, C.; Derat, E.; Malacria, M. *Chem. Eur.*
19
20
21 *J.* **2008**, *14*, 1238.

22
23 (5) For reviews on HAS, see: (a) Studer, A.; Curran, D. P. *Nat. Chem.* **2014**, *6*, 765;
24
25
26 (b) Studer, A.; Curran, D. P. *Angew. Chem., Int. Ed.* **2011**, *50*, 5018; (c) Bowman, W. R.;
27
28
29 Storey, J. M. D. *Chem. Soc. Rev.* **2007**, *36*, 1803.

30
31 (6) For a review on photoredox-catalyzed radical triple bond insertion to construct
32
33
34 heterocycles, see: (a) Sun, X.; Yu, S. *Chin. J. Org. Chem.* **2016**, *36*, 239. For seminal
35
36
37 reports, see: (b) Tong, K.; Zheng, T.; Zhang, Y.; Yu, S. *Adv. Synth. Catal.* **2015**, *357*,
38
39
40 3681. (c) Sun, X.; Li, J.; Ni, Y.; Ren, D.; Hu, Z.; Yu, S. *Asian J. Org. Chem.* **2014**, *3*,
41
42
43 1317. (d) Cheng, Y.; Yuan, X.; Jiang, H.; Wang, R.; Ma, J.; Zhang, Y.; Yu, S. *Adv.*
44
45
46 *Synth. Catal.* **2014**, *356*, 2859. (e) Sun, X.; Yu, S. *Org. Lett.* **2014**, *16*, 2938. (f) Jiang,
47
48
49 H.; Cheng, Y.; Wang, R.; Zheng, M.; Zhang, Y.; Yu, S. *Angew. Chem., Int. Ed.* **2013**,
50
51
52 *52*, 13289. (g) Cheng, Y.; Jiang, H.; Zhang, Y.; Yu, S. *Org. Lett.* **2013**, *15*, 5520.

53
54 (7) For examples on photoredox activation of iodides, see: (a) Ghosh, I.; Ghosh, T.;
55
56
57 Bardagi, J. I.; König, B. *Science*, **2014**, *346*, 725. (b) Cheng, Y.; Gu, X.; Li, P. *Org.*
58
59
60 *Lett.* **2013**, *15*, 2664. (c) Nguyen, J. D.; D'Amato, E. M.; Narayanam, J. M. R.;
Stephenson, C. R. J. *Nat. Chem.* **2012**, *4*, 854. (d) Kim, H.; Lee, C. *Angew. Chem., Int.*

1
2
3
4 *Ed.* **2012**, *51*, 12303.
5
6

7 (8) For examples on photoredox activation of diazonium salts, see: (a) Wang, H.;
8 Yu, S. *Org. Lett.* **2015**, *17*, 4272. (b) Hari, D. P.; König, B. *Angew. Chem., Int. Ed.*
9 **2013**, *52*, 4734. (c) Hari, D. P.; Schroll, P.; König, B. *J. Am. Chem. Soc.* **2012**, *134*,
10 2958. (d) Pratsch, G.; Anger, C. A.; Ritter, K.; Heinrich, M. R. *Chem. Eur. J.* **2011**, *17*,
11 4104. (e) Kalyani, D.; McMurtrey, K. B.; Neufeldt, S. R.; Sanford, M. S. *J. Am. Chem.*
12 *Soc.* **2011**, *133*, 18566. (f) Wetzel, A.; Pratsch, G.; Kolb, R.; Heinrich, M. R. *Chem.*
13 *Eur. J.* **2010**, *16*, 2547. (g) Wetzel, A.; Ehrhardt, V.; Heinrich, M. R. *Angew. Chem.,*
14 *Int. Ed.* **2008**, *47*, 9130. (h) Heinrich, M. R.; Wetzel, A.; Kirschstein, M. *Org. Lett.*
15 **2007**, *9*, 3833.
16
17
18
19
20
21
22
23
24
25
26
27
28
29

30 (9) For examples on photoredox activation of diaryliodonium salts, see: (a) Jiang,
31 H.; Cheng, Y.; Wang, R.; Zhang, Y.; Yu, S. *Chem. Commun.* **2014**, *50*, 6164. (b)
32 Fumagalli, G.; Boyd, S.; Greaney, M. F. *Org. Lett.* **2013**, *15*, 4398; (c) Neufeldt, S. R.;
33 Sanford, M. S. *Adv. Synth. Catal.* **2012**, *354*, 3517;
34
35
36
37
38
39
40

41 (10) To the best of our knowledge, there is only one example on the radical cyanide
42 insertion using photoredox catalysis, see ref 6c.
43
44
45
46

47 (11) (a) Nicewicz, D. A.; MacMillan, D. W. C. *Science* **2008**, *322*, 77. (b) Ischay,
48 M. A.; Anzovino, M. E.; Du, J.; Yoon, T. P. *J. Am. Chem. Soc.* **2008**, *130*, 12886. (c)
49 Narayanam, J. M. R.; Tucker, J. W.; Stephenson, C. R. J. *J. Am. Chem. Soc.* **2009**,
50 *131*, 8756. (d) Furst, L.; Matsuura, B. S.; Narayanam, J. M. R.; Tucker, J. W.;
51 Stephenson, C. R. J. *Org. Lett.* **2010**, *12*, 3104.
52
53
54
55
56
57
58
59

60 (12) For intramolecular radical cyclization of cyanides, see: (a) Vervisch, K.;

- 1
2
3
4 D'hooghe, M.; Törnroos, K. W.; Kimpe, N. D. *Org. Biomol. Chem.* **2012**, *10*, 3308. (b)
5
6
7 Larraufie, M.-H.; Ollivier, C.; Fensterbank, L.; Malacria, M.; Lacôte, E. *Angew.*
8
9
10 *Chem., Int. Ed.* **2010**, *49*, 2178. (c) Fernandez-Mateos, A.; Teijón, P. H.; Burón, L. M.;
11
12 Clemente, R. R.; González, R. R. *J. Org. Chem.* **2007**, *72*, 9973. (d) Benati, L.;
13
14 Bencivenni, G.; Leardini, R.; Minozzi, M.; Nanni, D.; Scialpi, R.; Spagnolo, P.;
15
16 Zanardi, G.; Rizzoli, C. *Org. Lett.* **2004**, *6*, 417. (e) Camaggi, C. M.; Leardini, R.;
17
18 Nanni, D.; Zanardi, G. *Tetrahedron* **1998**, *54*, 5587. (f) Curran, D. P.; Liu, H.; Josien,
19
20 H.; Ko, S.-B. *Tetrahedron* **1996**, *52*, 11385. (g) Nanni, D.; Pareschi, P.; Rizzoli, C.;
21
22 Sgarabotto, P.; Tundo, A. *Tetrahedron* **1995**, *51*, 9045.
23
24
25
26
27
28 (13) Liu, F.; Li, C. *J. Org. Chem.* **2009**, *74*, 5699.
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60