

Ag/Brønsted Acid Co-Catalyzed Spiroketalization of β -Alkynyl Ketones toward Spiro[chromane-2,1'-isochromene] Derivatives

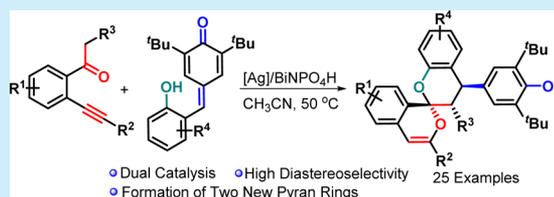
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S Supporting Information

ABSTRACT: A new Ag/Brønsted acid co-catalyzed spiroketalization of β -alkynyl ketones with *para*-quinone methides (*p*-QMs) has been established, enabling multiple C–C and C–O bond-forming events to access densely functionalized spiro[chromane-2,1'-isochromene] derivatives with generally excellent diastereoselectivity and good yields. A reasonable mechanism for forming these 6,6-dibenzannulated spiroketals involving 6-*endo*-dig *oxo*-cyclization and 1,6-addition-cyclization cascades is proposed.



Spiroketal appears as a privileged structural motif that exists in a broad array of both natural products and synthetic bioactive molecules. Compounds incorporating a spiroketal core structure have been found to exhibit a wide spectrum of biological activities.¹ Among the family of spiroketals, the 6,6-benzannulated spiroketals behave as a key pharmacophore in many bioactive natural products, such as cynandione B,² chaetoquadrin B,³ and virgatolide B⁴ (Figure 1). In particular,

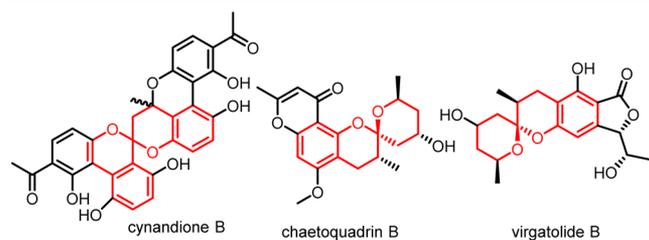


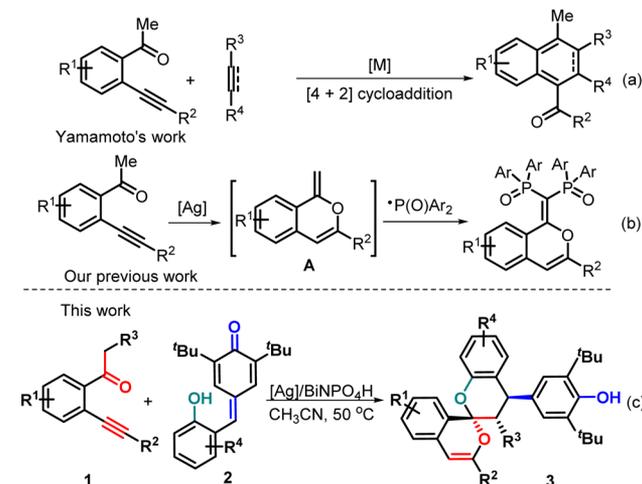
Figure 1. 6,6-Benzannulated spiroketal-containing natural products.

cynandione B displays significant *in vitro* cytotoxic activity against both T-24 cell lines and PLC/PRF/5 cell lines⁵ and anti-inflammatory activity.⁶ Consequently, much effort has been made toward identifying general methods for the synthesis of these 6,6-benzannulated spiroketals. Generally, strategies for 6,6-benzannulated spiroketal syntheses include acid-promoted cyclization of protected diol with ketone precursors⁷ or dehydration-cyclization of dihydroxy ketones,⁸ intramolecular Michael reaction,⁹ Ti-mediated spirocyclization,¹⁰ and oxidative cyclization,¹¹ hetero Diels–Alder reaction,¹² reductive cyclization reaction,¹³ and Me₃SiI-promoted spiroketalization of salicylic aldehydes.¹⁴ In spite of all these significant achievements, Ag/Brønsted acid co-catalysis toward 6,6-benzannulated spiroketals via bicyclization cascade of β -alkynyl ketones, to the best of our knowledge, is virtually unexplored so far.

Catalytic cyclization of β -alkynyl ketones offers a flexible and reliable method for the synthesis of biologically interesting

cyclic structures in a convergent manner.^{15,16} Specifically, the group of Yamamoto pioneered the seminal work on metal-catalyzed [4 + 2] cycloaddition between β -alkynyl ketones and alkenes or alkynes (Scheme 1a).^{15a,b} Very recently, we have

Scheme 1. Profile Application of Cycloadditions



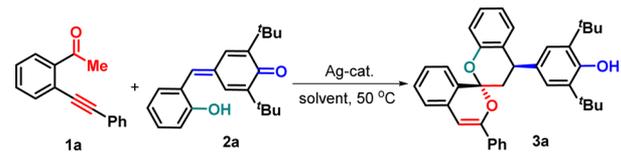
reported silver-mediated *oxo*-cyclization and C(sp³)–H biphosphinylation of β -alkynyl ketones toward functionalized isochromenes, in which the *in situ* generation of methyleneisochromenes A is a key step to capture diarylphosphine oxide radicals (Scheme 1b).¹⁷ In view of these successful transformations and our recent findings on dual synergistic catalysis,¹⁸ we envisaged that under Ag-catalyzed conditions, β -alkynyl ketones could be transformed into methyleneisochromenes A with a nucleophilic site, enabling their 1,6-addition

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reactions with *para*-quinone methides (*p*-QMs)¹⁹ catalyzed by Brønsted acid through [4 + 2] cycloaddition toward spiro-[chromane-2,1'-isochromene] derivatives. As expected, the combination of silver trifluoroacetate (AgTFA) and 1,1'-binaphthyl-2,2'-diyl hydrogen phosphate (BiNPO₄H) as a dual catalyst system makes these transformations work well, accessing the desired 6,6-dibenzannulated spiroketals (Scheme 1c). Interestingly, the diastereoselectivity of 6,6-dibenzannulated spiroketals could be controlled well by using a Ag/Brønsted acid co-catalytic system, together with realization of C(sp³)-H functionalization adjacent to the carbonyl group on the β -alkynyl ketone unit. Herein, we report these interesting observations in which new and selective 6-*endo-dig* oxo-cyclization and 1,6-addition-cyclization cascades were first achieved.

Initially, β -alkynyl ketone **1a** was selected as a model substrate and subjected to the reaction with *p*-QM **2a** in acetonitrile (CH₃CN) using silver trifluoroacetate (AgTFA, 10 mol %) as a catalyst at 50 °C under air conditions (Table 1,

Table 1. Optimization of Reaction Conditions^a



entry	co-catalyst (mol %)	solvent	yield (%) ^b
1	AgTFA (10)	CH ₃ CN	trace
2	AgTFA (10)/BiNPO ₄ H (20)	CH ₃ CN	66 ^c
3	AgOAc (10)/BiNPO ₄ H (20)	CH ₃ CN	NR ^d
4	AgOTf (10)/BiNPO ₄ H (20)	CH ₃ CN	35.
5	AgNO ₃ (10)/BiNPO ₄ H (20)	CH ₃ CN	NR
6	AgTFA (10)/PivOH (20)	CH ₃ CN	trace
7	AgTFA (10)/ <i>p</i> -TsOH (20)	CH ₃ CN	20
8	AgTFA (10)/TFA (20)	CH ₃ CN	46
9	AgTFA (10)/BiNPO ₄ H (10)	CH ₃ CN	30
10	AgTFA (5)/BiNPO ₄ H (20)	CH ₃ CN	43
11	AgTFA (10)/BiNPO ₄ H (20)	toluene	55
12	AgTFA (10)/BiNPO ₄ H (20)	1,4-dioxane	trace
13	AgTFA (10)/BiNPO ₄ H (20)	THF	trace
14	AgTFA (10)/BiNPO ₄ H (20)	CH ₃ CN	53 ^e
15	AgTFA (10)/BiNPO ₄ H (20)	CH ₃ CN	60 ^f

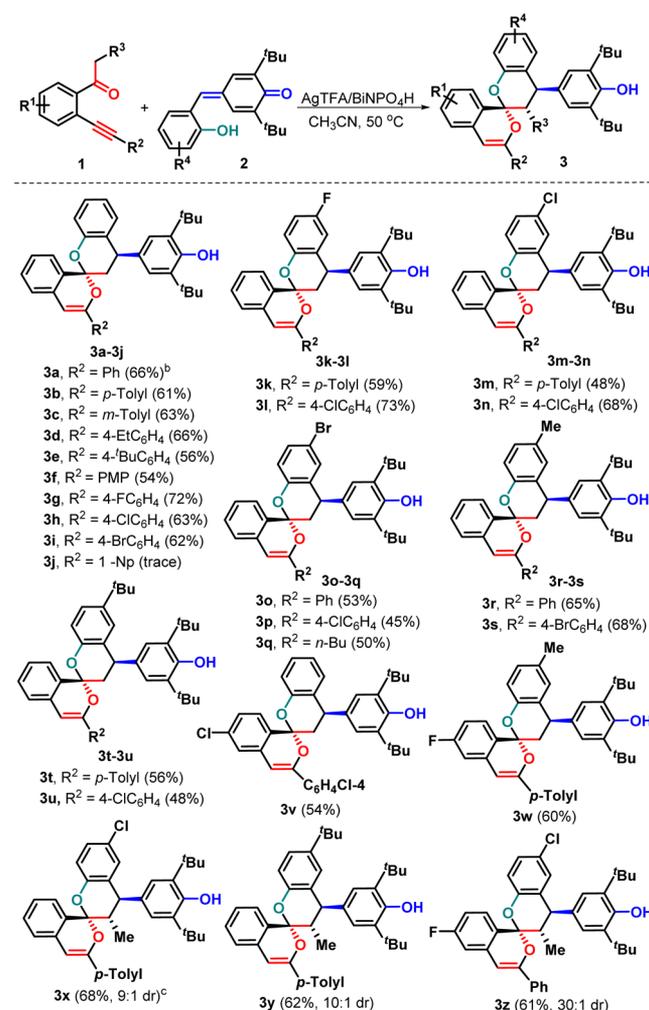
^aReaction conditions: **1a** (0.3 mmol), **2a** (0.2 mmol), Ag-catalyst (*x* mol %), Brønsted acid catalyst (*y* mol %), solvent (3.0 mL), under air conditions. ^bIsolated yield based on substrate **2a**. ^cBiNPO₄H (1,1'-binaphthyl-2,2'-diyl hydrogen phosphate). ^dNo reaction (NR). ^eThe reaction was carried out at room temperature. ^fThe reaction was carried out at 60 °C.

entry 1). This reaction gave a trace amount of product **3a** which failed to be isolated, and substrate **2a** was recovered. To our delight, when BiNPO₄H (20 mol %) as a Brønsted acid co-catalyst was added into the above reaction system, the desired 6,6-benzannulated spiroketal **3a** was generated in 66% yield (entry 2), indicating that Brønsted acid catalyst may play a key role in accelerating the intermolecular 1,6-addition reaction.¹⁹ The following screening of several other silver salts such as AgOAc, AgOTf and AgNO₃ showed that these silver catalysts did not show any improvements in the yield of product **3a** (entries 3–5). Exchanging BiNPO₄H for pivalic acid (PivOH) as a Brønsted acid catalyst completely suppressed the reaction process (entry 6). Employment of both *p*-toluenesulfonic acid

(*p*-TsOH) and trifluoroacetate acid (TFA) had no positive effect on the yield of **3a** as compared with BiNPO₄H (entry 2 vs entries 7–8). Lowering the BiNPO₄H or AgTFA loading is not beneficial for this transformation (entries 9–10). We next investigated the solvent effect by employing different solvents including toluene, 1,4-dioxane, and tetrahydrofuran (THF), but <55% yield was observed (entries 11–13). Moreover, the reaction could proceed at either room temperature or 60 °C, but provided the slightly lower conversion into **3a** as compared with the reaction temperature being 50 °C (entry 2 vs entries 14–15).

With these acceptable reaction conditions in hand, we then systematically investigated the generality of this Ag/BiNPO₄H co-catalyzed spiroketalization cascades for accessing spiro-[chromane-2,1'-isochromene] derivatives **3** by examining β -alkynyl ketone and *p*-QM components (Scheme 2). At first, β -alkynyl ketones with diverse functionalities were evaluated in combination with *p*-QM **2a**. Various substituents with electronically poor and rich nature at different positions of arylalkynyl (R²) moiety were proven not to hamper this Ag/BiNPO₄H co-catalysis, affording the corresponding 6,6-benzannulated spiroketals **3b–3i** with excellent diastereoselec-

Scheme 2. Substrate Scope for Synthesis of 3^{a,b,c}



^aReaction conditions: **1** (0.3 mmol), **2** (0.2 mmol), AgTFA (10 mol %), BiNPO₄H (20 mol %), CH₃CN (3.0 mL), air conditions. ^bIsolated yield based on substrate **2**. ^cdr Value based on the analysis of ¹H NMR.

tivity (single diastereoisomer) and yields ranging from 54% to 72% (see SI). Functional groups like methyl (**1b** and **1c**), ethyl (**1d**), *tert*-butyl (**1e**), methoxy (*p*-methoxyphenyl = PMP, **1f**), fluoride (**1g**), chloride (**1h**), and bromide (**1i**) were compatible with the catalytic conditions. Among them, a slight increase in the yield was obtained (**3g**, 72%) as the *p*-fluorophenyl counterpart (**1g**) was served as a reaction partner, whereas the methoxy group resulted in a decreased yield (**3f**). β -Alkynyl ketone **1j** having a 1-naphthyl (1-Np) group on the alkynyl moiety was not an appropriate candidate, as only a trace amount of product **3j** was detected under the standard conditions (Scheme 2). Next, *p*-QMs **2** carrying fluoro, chloro, bromo, methyl, and *t*-butyl groups at 4-position of the phenol ring were successfully engaged in the current Ag/BiNPO₄H co-catalyzed spiroketalization, enabling 6-*endo-dig* oxo-cyclization/[4 + 2] cycloaddition toward the sole diastereoisomeric 6,6-dibenzannulated spiroketals **3k–3u** in 48%–73% yields. Notably, substrate **1k** with an *n*-butyl (*n*-Bu) group on the alkynyl moiety was proven to be a suitable reaction partner, accessing the corresponding product **3q** as a sole diastereoisomer in 50% yield. Moreover, representative chloro (**1l**) and fluoro (**1m**) substituents at 4-position of the internal arene ring of substrates **1** would be accommodated, confirming the success of transformations, as the corresponding 6,6-dibenzannulated spiroketals **3v** and **3w** were afforded with 54% and 60% yields, respectively. In view of these results, we considered changing the substituents (R³) located at α -position of carbonyl group of substrates **1** to explore the feasibility of this Ag/BiNPO₄H co-catalysis. Replacing the methyl group with an ethyl functionality at α -position of carbonyl group, β -alkynyl ketones **1n** and **1o** were tolerated well under the standard catalytic conditions, furnishing the corresponding inseparably diastereoisomeric spiroketals **3x–3z** with 61%–68% yields and 9:1 to 30:1 dr (Scheme 2). It is noteworthy that the current protocol represents a new and practical pathway for assembly of richly decorated 6,6-dibenzannulated spiroketals **3** with generally excellent diastereoselectivity through Ag/BiNPO₄H co-catalyzed bicyclization involving C(sp³)-H functionalization of β -alkynyl ketones. The structures of these spiroketals **3** were characterized by their NMR spectroscopy and HRMS. In the case of **3i**, its structure was unequivocally confirmed by carrying out single crystal X-ray diffraction (Figure 2).

To expand potential application of this method, a gram-scale reaction of **1b** (6 mmol) with **2a** (4 mmol) was conducted under the standard conditions, offering **3b** in 57% yield (Scheme 3).

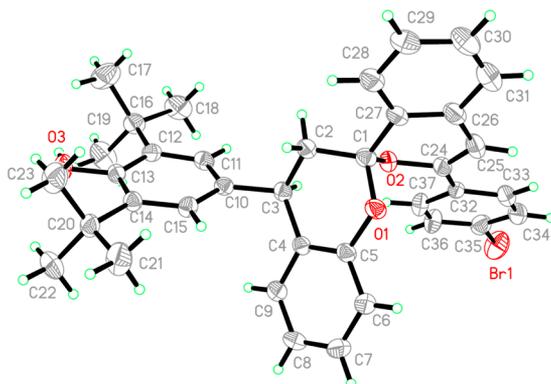
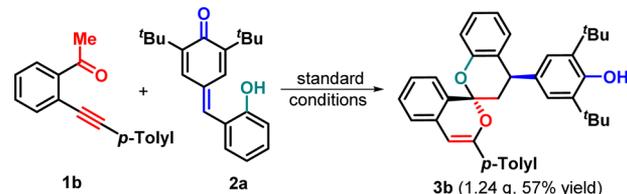


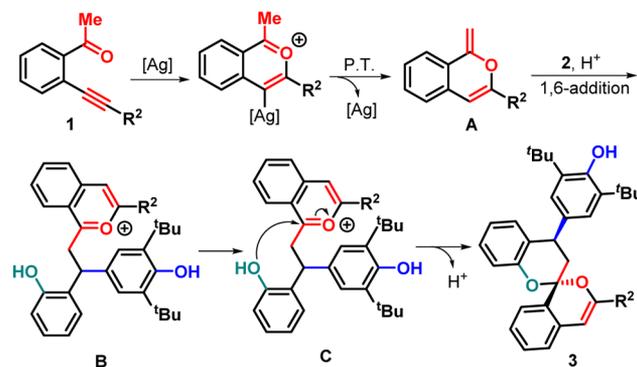
Figure 2. ORTEP drawing of **3i**.

Scheme 3. Gram-Scale Experiment



Combining the above observations and the literature precedents of silver-catalyzed oxo-cyclization,^{16,20} a reasonable mechanism for forming products **3** was proposed as shown in Scheme 4. First, Ag-catalyzed 6-*endo-dig* oxo-cyclization of β -

Scheme 4. Plausible Reaction Pathway



alkynyl ketones **1** generates isobenzopyrylium intermediate,¹⁶ followed by proton transfer (PT) to afford isochromene **A**. Next, BiNPO₄H-catalyzed 1,6-addition between **A** and **2** yields intermediate **B**,¹⁹ which undergoes intramolecular oxo-nucleophilic addition and subsequent deprotonation to offer products **3**.

In conclusion, starting from β -alkynyl ketones and *p*-QMs, we have established Ag/BiNPO₄H co-catalyzed 6-*endo-dig* oxo-cyclization/[4 + 2] cycloaddition cascades for highly diastereoselective synthesis of structurally diverse spiro[chromane-2,1'-isochromene] derivatives with generally good yields. The present dual catalytic system provides an efficient and practical pathway for constructing 6,6-dibenzannulated spiroketals, in which β -alkynyl ketones with a methyl group adjacent to a carbonyl group were converted into the sole diastereoisomeric spiro compounds containing an isochromene unit via C(sp³)-H functionalization. Further assessment of the bioactivity of these spiroketal compounds is in progress in our laboratory.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.7b01705.

Experimental procedures and spectroscopic data for all new compounds **3a–3z** (PDF)

X-ray crystal data (CIF) for **3i** (CIF)

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Notes

The authors declare no competing financial interest.

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