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Discovery of Hydrolysis-resistant Isoindoline *N*-Acyl Amino Acid Analogs that Stimulate Mitochondrial Respiration

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ABSTRACT: *N*-acyl amino acids directly bind mitochondria and function as endogenous uncouplers of UCP1-independent respiration. We found that administration of *N*-acyl amino acids to mice improves glucose homeostasis and increases energy expenditure indicating that this pathway might be useful for treating obesity and associated disorders. We report the full account of the synthesis and mitochondrial uncoupling bioactivity of lipidated *N*-acyl amino acids and their unnatural analogs. Unsaturated fatty acid chains of medium length and neutral amino acid head groups are required for optimal uncoupling activity on mammalian cells. A class of unnatural *N*-acyl amino acid analogs, characterized by isoindoline-1-carboxylate head groups (**37**), were resistant to enzymatic degradation by PM20D1 and maintained uncoupling bioactivity in cells and in mice.

INTRODUCTION

The *N*-acyl amino acids are a class of endogenous lipid metabolites with pleotropic bioactivities¹. Specific members of this metabolite class include *N*-oleoyl serine, *N*-arachidonoyl glycine, and *N*-oleoyl glycine, which have been shown to regulate bone remodelin², pain sensation³, and food intake⁴, respectively. At a molecular level, these lipids can act as ligands for ion channels⁵ (e.g., TRPV1) or G-protein coupled receptors⁶ (e.g., GPR92). To date, over fifty distinct species of *N*acyl amino acids have been detected in mammalian tissues⁷. However, the full spectrum of biological functions for this class of endogenous lipids remain incompletely characterized.

We have recently identified a novel function for several Nacyl amino acids, including N-oleoyl leucine, N-oleoyl phenylalanine, and N-arachidonoyl glycine. These compounds can uncouple mitochondrial respiration, stimulating respiration on isolated mitochondria in both cells and mice⁸. Uncoupling oxidative respiration makes ATP production less efficient resulting in increased metabolic rate, and loss of energy through heat production^{9,10}. Key to this discovery was the annotation of the circulating factor peptidase M20 domain containing 1 (PM20D1) as a dominant enzyme that regulates N-acyl amino acid levels in vivo. Genetically increased circulating PM20D1 by adeno-associated virus (AAV) delivery to the liver augments circulating N-acyl amino acids, leading to increased respiration and blunted weight gain on high fat diet in mice. Direct administration of N-acyl amino acids to mice 57 by intraperitoneal (IP) injection also produces increased whole 58 body energy expenditure, with weight loss and improved glu-59

cose homeostasis. Mechanistically, *N*-acyl amino acids directly act on isolated mitochondria to increase respiration, potentially by interacting with the SLC25 family of inner mitochondrial transporters¹¹.

To more fully understand how *N*-acyl amino acids stimulate respiration, we herein disclose a full account of our structureactivity relationship (SAR) studies of *N*-acyl amino acids and their uncoupling bioactivity. The uncoupling activity of this metabolite class is largely restricted to those with neutral amino acid head groups and unsaturated fatty acyl chains of medium length. By exploring unnatural analogs of these metabolites, we identified the proline derivative N-oleoyl-isoindoline-1-carboxylate **37** as a semisynthetic *N*-acyl amino acid analog with exceptional uncoupling activity. **37** possessed uncoupling activity in cells and in mice, but unlike the natural metabolites, was entirely resistant to enzymatic hydrolysis by PM20D1.

RESULTS AND DISCUSSION

To determine the uncoupling activity of the *N*-acyl amino acid analogs, we tested their effect on cellular respiration of C2C12 mouse myoblast cells. Because these compounds show a timedependent effect on changes in respiration, we standardized our assay results by reporting the maximal stimulation of respiration as compared to DMSO-control. Further standardizing our respiration assays, all compounds were assayed at 50 μ M (see also **Supplemental Figure 1**). This is below the critical micelle concentration for these analogs (see **Supplemental Figure 2**)¹². Initially, we restricted the scope to oleic acidconjugated amino acids. As previously reported, C18:1-Phe Environment.

possess potent uncoupling bioactivity⁸, stimulating respiration to $\sim 160\%$ of baseline levels (**Table 1**, compound 1). Amino acid stereospecificity was not observed in this uncoupling bioactivity, as both C18:1-D-Phe and C18:1-L-Phe stimulated cellular respiration equally well (Table 1, compound 2). Consistent with our previous report, the amino acid head group carboxylate was absolutely required for activity: removal of the carboxylate moiety entirely abolished uncoupling activity (Table 1, compound 3). Next, by varying the amino acid head group, we found that amino acids containing neutral side chains (e.g., Leu, Ile, Gln, Pro, Trp, compounds 4 - 8) were potent uncouplers, but those compounds with charged side chains (e.g., Lys, Tyr, Glu, compounds 9 - 11) were unable to induce respiration (Table 1). Some tolerance for backbone steric modifications were acceptable on the amino acid side. For instance, inclusion of homoglycine or the dipeptide Gly-Gly head group preserved uncoupling bioactivity (Table 1, compounds 12 and 13).

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Table 1. Structure and uncoupling bioactivity of compounds with different amino acid head groups.^{*a*}

| Ŭ O | ~ ~ ~ | | |
|--------|---------------|--|----------------|
| Cmpd | Head group | R | % ^b |
| 1 | L-Phe | HN ₅ ^{CO} 2 ^H | 161 ± 12 |
| 2 | D-Phe | | 173 ± 9 |
| 3 | n/a | H ₃ s ⁴ | 104 ± 4 |
| 4 | L-Leu | HN ₃ ⁴ | 178 ± 13 |
| 5 | L-Ile | | 167 ± 14 |
| 6 | L-Gln | H ₂ N CO ₂ H | 164 ± 4 |
| 7 | L-Pro | N M→ CO ₂ H | 182 ± 7 |
| 8 | L-Trp | HN HN CO ₂ H | 147 ± 9 |
| 9 | L-Lys | H ₂ N CO ₂ H | 108 ± 4 |



"Respiration in C2C12 cells is shown as maximal increases versus basal oligomycin-treated respiration, which is normalized to 100%. Data are shown as means \pm SEM, n=3-6/group; ^bstimulation of respiration (baseline = 100%).

For the fatty acid side chain, we observed striking structural requirements regarding length and type of unsaturation for uncoupling bioactivity. By sequentially stepping through increasing fatty acid chain lengths, we found that fatty acyl chains that were too short (e.g., < C12:0) or too long (e.g., >C20:0) possessed no uncoupling activity (Table 2, compounds 14 - 18). Within the "medium" fatty acyl chain range, specific types of unsaturation potently improve the respiration response. For instance, we observed a step-wise increase in uncoupling activity from C18:0-Phe < C18:1-Phe < C18:2-Phe (compound 16 vs. 1 vs. 19) that, surprisingly, drops off entirely with C18:3-Phe (compound 20). Other specific olefin perturbations also underscored the importance of specific types and locations of unsaturation. For instance, compared to cis- Δ^9 -C18:1-Phe (compound 1), trans- Δ^9 -C18:1-Phe (Table 2, compound 21) olefin shows essentially no uncoupling bioactivity. Similarly, $cis \Delta^6$ -C18:1-Phe and $cis \Delta^{11}$ -C18:1-Phe (Table 3, compound 22 and 23) were all inferior to the parent compound 1.

Given the strong influence of unsaturation on the acyl chain length, we systematically increased unsaturation of C20:0-Phe (compound 17), a parent compound that initially does not possess uncoupling activity. Again, increasing unsaturation augments uncoupling bioactivity with an apparent optimal unsaturation at C20:4 (Table 2, compounds 24 - 26). A similar strategy to increase unsaturation also converted inactive C22:0-Phe (compound 18) into a potent uncoupler (C22:6-Phe, compound 27). The specific requirement for unsaturation is reminiscent of cannabinoid receptor ligands, in which only arachidonoylbut not other fatty acyl-containing lipids can agonize the receptor13. The unsaturation specificity observed in this N-acyl amino acid series (Table 2) is consistent with the shape complementarity imposed by interactions with proteinaceous factors that mediate uncoupling. Taken together, our data map out the optimal requirements for uncoupling responses within the *N*-acyl amino acid class: an uncharged carboxylate-containing head group, a medium fatty acid chain length, and specific sites of unsaturation along the acyl chain.

Table 2. Structure and uncoupling bioactivity of compounds with different fatty acid side chains.^a

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| Cm | pd Fatty acid | R | % ^{<i>b</i>} |
|----|------------------|----------------|-----------------------|
| 14 | C12:0 |) O | 96 ± 4 |
| 15 | 5 C16:0 | ¥~~~~~~ | 180 ± 20 |
| 16 | 6 C18:0 | | 148 ± 12 |
| 17 | C20:0 | | 95 ± 8 |
| 18 | B C22:0 | ¥ | 100 ± 3 |
| 19 | C18:2 | | 218 ± 10 |
| 20 | C18:3 | | 109 ± 12 |
| 21 | trans- C18:1 | j ⁴ | 107 ± 3 |
| 22 | Δ6- C18:1 | ř. | 110 ± 12 |
| 23 | Δ11- C18:1 | Å. | 115 ± 8 |
| 24 | C20:1 | | 103 ± 3 |
| 25 | 5 C20:4 | × | 200 ± 12 |
| 26 | 6 C20:5 | ř | 259 ± 43 |
| 27 | C22:6 | × | 166 ± 11 |

^{*a*}Respiration in C2C12 cells is shown as maximal increases versus basal oligomycin-treated respiration, which is normalized to 100%. Data are shown as means \pm SEM, n=3-6/group; ^{*b*} stimulation of respiration (baseline = 100%).

Given the structure activity relationships we observed by simply varying the two sides of the amide bond, we next explored an alternative bond linkage between the fatty acid and amino acid to identify those that may possess superior properties as compared to the naturally occurring metabolites, with the ultimate goal of developing improved mitochondriauncoupling analogs that might be useful for the treatment of obesity and metabolic disease^{14,15}. Towards this end, we synthesized two urea analogs of C18:1-Gly and C18:1-Phe that we hypothesized might be less metabolically labile compared to the natural amides, and tested their effects on cellular respiration. Both urea analogs failed to induce respiration on cells, demonstrating a potential preference for the presence of an amide bond in the bioactivity of this class of compounds (**Table 3**).

| Table 3. | Structure | and | uncoupling | bioactivity | of | unnatural |
|-----------|------------|-------|--------------|---------------------|----|-----------|
| urea-cont | aining N-a | cyl a | mino acid an | alogs. ^a | | |

| Cmpd | Structure | % ^{<i>b</i>} |
|------|-----------|-----------------------|
| | | |



"Respiration in C2C12 cells is shown as maximal increases versus basal oligomycin-treated respiration, which is normalized to 100%. Data are shown as means \pm SEM, n=3-6/group; ^bstimulation of respiration (baseline = 100%).

We also surveyed unnatural amino acid head groups. We were intrigued by the potent uncoupling observed with C18:1-Pro (compound 7) and noted that this derivative with a cyclic head group is the most structurally distinct from the other N-acyl amino acids tested. We therefore synthesized and tested a variety of unnatural proline and homoproline N-acyl amino acid derivatives with oleate as a fatty acyl chain. Testing these unnatural analogs in cellular respiration assays revealed that they all stimulated uncoupling to a maximum of ~150-240% of DMSO control (Table 4, compounds 30 – 35). Derivatization of these isoindoline head groups with linoleoyl instead of oleoyl acyl groups also produced potent uncouplers (Table 4, compounds 36 and 37). These analogs 30 - 37 demonstrate that the uncoupling bioactivity of N-acyl amino acids can extend beyond structures containing only the proteinogenic amino acids.

Table 4. Structure and uncoupling bioactivity of compounds with unnatural proline- and homoproline-drived head groups.^{*a*}

| _ | C18:1 | _ | :18:2 |
|--------|---------------|--|--------------|
| R J | ~~ <u> </u> | or R | |
| Cmpd | Fatty acid | R | % |
| 30 | C18:1 | | 174 ± 3 |
| 31 | C18:1 | N ₅ ² | 187 ± 19 |
| 32 | C18:1 | N ₅ ² CO ₂ H | 216 ± 17 |
| 33 | C18:1 | N-≹- ĈO₂H | 240 ± 24 |
| 34 | C18:1 | N ₹ CO ₂ H | 190 ± 12 |
| 35 | C18:1 | | 146 ± 5 |

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"Respiration in C2C12 cells is shown as maximal increases versus basal oligomycin-treated respiration, which is normalized to 100%. Data are shown as means \pm SEM, n=3-6/group; ^bstimulation of respiration (baseline = 100%).

N-acyl amino acids can be hydrolytically inactivated through cleaveage by PM20D1, a bidirectional amidase that cleaves fatty acid amides to liberate free fatty acids and amino acids⁸. Given the unusual cyclic structures of 30 - 37, we hypothesized that these analogs might have lower rates of hydrolysis, and therefore higher stability. We therefore incubated purified, recombinant murine PM20D1 with a canonical N-acyl amino acid substrate, N-oleoyl phenylalanine, or one of the unnatural N-acvl amino acids. We observed >90% hydrolysis of Noleoyl phenylalanine to oleic acid in our assay conditions, demonstrating that this enzyme efficiently cleaves naturally occurring N-acyl amino acids. Remarkably, all unnatural Nacyl amino acids tested were completely resistant to PM20D1mediated hydrolysis (Fig. 1a). Consistent with these observations, when endogenously occurring N-acyl amino acids or unnatural analogs were incubated with mouse liver microsomes, the unnatural analogs also maintained exceptionally long half-lives (Fig. 1b): as a group, the average half-life of the natural N-acyl amino acids was 8 min, whereas the average for N-acyl analogs with unnatural head groups was 64 min. The tertiary amide structures in **30-37** may also contribute to improved metabolic stability relative to the secondary amides tested. These data therefore demonstrate that changing the amide head group can alter the downstream metabolism of Nacyl amino acid analogs.

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Fig. 1. (A) Amount of the indicated compound remaining in the presence of PBS or recombinant, murine PM20D1 after incubation at 37°C for 1 h. Data are shown as means \pm SEM, n = 3/group. *** P < 0.001. (B) Half-life (min) of the indicated compound after incubation in mouse liver microsomes. "Natural" indicates endogenously present *N*-acyl amino acid whereas "unnatural" indicates an *N*-acyl amino acid analog with a non-proteinogenic head group.

Lastly, we tested the uncoupling bioactivity of unnatural Nacyl amino acid analogs in mice. Towards this end, we adopted a previously used assay in which mice rendered obese by high fat diet feeding (diet-induced obesity, DIO) are given Nacyl amino acids daily by IP administration. Synthetic chemical uncouplers such as 2,4-dinitrophenol have been shown to blunt weight gain in this model¹⁵, and administration of N-acyl amino acid led to weight loss⁸. For these experiments, we selected compound 37 because this compound showed excellent uncoupling bioactivity (Table 4) and resistance to hydrolysis (Fig. 1). DIO mice (initial weight, 46.8 ± 0.6 g) were treated with increasing doses of compound 37 (5, 10, and 25 mg/kg/day IP) daily for 7 consecutive days (Fig. 2a). Though no differences in weight were observed at 5 mg/kg vs. vehicletreated mice, both 10 and 25 mg/kg doses of compound 37 led to dose-dependent blunting of body weight gain (net weight change, 0.2 ± 0.2 g and -1.2 ± 0.6 g, respectively, vs. 1.1 ± 0.1 g for vehicle-treated mice, P < 0.05 for each comparison). Food intake was not statistically significantly reduced in mice at these higher doses (Fig. 2b). However, ad lib blood glucose 2 h after the last compound injection was significantly lower in mice treated with 25 mg/kg compound 37 versus vehicletreated mice $(172 \pm 7 \text{ mg/dl} \text{ versus } 204 \pm 7 \text{ mg/dl}, \text{ respective$ ly, Fig. 2c). These data therefore indicate that unnatural *N*-acyl amino acid 37 maintains uncoupling bioactivity in vivo.

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Fig. 2. (A-C) Total change in body weight (A), food intake (B), and ad lib blood glucose (C) of diet-induced obese mice (19 weeks high fat diet) after 7 days daily treatment with the indicated dose (IP) of compound **37**. Prior to compound administration, mouse weights were not different between the groups (average weight across all groups, 46.8 ± 0.6 g). Data are shown as means \pm SEM, n=5/group. *, P < 0.05; **, P < 0.01 for the indicated dose versus vehicle-treated mice.

CHEMISTRY

Compounds were synthesized as depicted in Scheme 1 following well-known procedures. Commercially available acid chlorides **39** (using General procedure A), or those generated from acid **38** with oxalyl chloride (using General procedure B), were coupled with the corresponding natural or unnatural amino acids **40** to provide *N*-acyl amino acid analogs **1-27** and **30-37** (Scheme 1). Oleoyl chloride reacted with ammonia hydroxide to give primary amide **41**, which was reduced with LiAlH₄ to afford primary amine **42**. Imidazole intermediate **43** was obtained by treatment of amine **42** with CDI, which then reacted with amino esters **44** in the presence of DIPEA. The resulting urea esters **45** were hydrolyzed with LiOH to provide the desired urea analogs **28** or **29** in good yields (using General procedure C) (Scheme 2).

Scheme 1. Synthesis of Natural and Unnatural *N*-Acyl Amino Acid Analogs^{*a*}



Scheme 2. Synthesis of Urea Analogs^a



^{*a*}Reaction conditions: (a) oxalyl chloride, DMF, DCM; (b) K_2CO_3 , acetone/H₂O or DIPEA, DCM; (c) NH₄OH, THF; (d) LiAlH₄, THF; (e) CDI, DCM; (f) DIPEA, DCM; (g) LiOH, then HCl.

CONCLUSIONS

Here we provide a detailed account of the synthesis and structure activity relationships of the N-acyl amino acids and their uncoupling activity on mammalian cells. Our data reveal the structural requirements for the uncoupling bioactivity of this class of endogenous metabolites, which include a strict requirement for neutral carboxylate-containing head groups, medium chain fatty acids with specific unsaturation, and an amide bond connecting the two. Through exploration of unnatural N-acyl amino acid analogs, we identify N-acyl isoindoline 37 as a potent, hydrolysis-resistant compound that displays uncoupling bioactivity in cells and in vivo. Projecting forward, compound 37 or other hydrolysis-resistant N-acyl amino acid analogs may serve as useful probes for understanding the full spectrum of bioactivities associated with the Nacyl amino acid family of lipids. Such compounds may also serve as leads for alternative structural classes that may have widened therapeutic windows or altered beneficial and adverse profiles compared with the more "classical" chemical uncouplers such as 2,4-dinitrophenol.

EXPERIMENTAL SECTION

All solvents and chemicals were reagent grade. Unless otherwise mentioned, all reagents and solvents were purchased from commercial vendors and used as received. Flash column chromatography was carried out on a Teledyne ISCO CombiFlash Rf system using prepacked columns. Solvents used include hexane, ethyl acetate (EtOAc), dichloromethane and

methanol. Purity and characterization of compounds were established by a combination of HPLC, TLC, mass spectrometry, and NMR analyses. ¹H and ¹³C NMR spectra were recorded on a Bruker Avance DPX-400 (400 MHz) spectrometer and were determined in chloroform-d or DMSO- d^{6} with solvent peaks as the internal reference. Chemical shifts are reported in ppm relative to the reference signal, and coupling constant (J) values are reported in hertz (Hz). Thin layer chromatography (TLC) was performed on EMD precoated silica gel 60 F254 plates, and spots were visualized with UV light or iodine staining. Low resolution mass spectra were obtained using a Thermo Scientific ultimate 3000/ LCQ Fleet system (ESI). High resolution mass spectra were obtained using a Thermo Scientific EXACTIVE system (ESI). All test compounds were greater than 95% pure as determined by qNMR on a Bruker Avance DPX-400 (400 MHz) spectrometer.

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General procedure A. To the amino acid (2 eq.) in acetone and water (0.1 M, 1:1, v/v) was added K_2CO_3 (3 eq.) and acyl chloride (1 eq.) at 0 °C. The reaction mixture was stirred at room temperature overnight and was then acidified with HCl (1 M) until pH<3. The mixture was extracted with ethyl acetate, washed with brine, dried (Na₂SO₄) and filtered. The filtrate was concentrated in vacuo, and the residue was purified by column chromatography (silica gel, ethyl acetate/hexanes) to give the desired product.

General procedure B. To a solution of fatty acid (1 eq.) in DCM was added oxalyl chloride (1.2 eq.) and one drop of DMF at 0 °C. The reaction was aged at room temperature for 2 hours and then concentrated *in vacuo*. The crude residue was dissolved in DCM and added to a suspension of amino acid (1.5 eq.) and DIPEA (2 eq.) and DCM. The reaction mixture was stirred at room temperature overnight and was then acidified with 1.0M HCl to pH<3. The result mixture was extracted with DCM, washed with brine, dried (Na₂SO₄), and concentrated. The crude residue was purified by flash chromatography on silica gel to give the product.

General procedure C. Step 1: To a 0 °C solution of oleoyl chloride (1 g) in THF (20 mL) was added ammonia hydroxide (10 eq.). The mixture was allowed to warm to room temperature and stirred for 3 hours. The mixture was filtered to give the desired oleamide 41 as a colorless solid (796 mg, 86%). Step 2: To a suspension of LiAlH₄ (2 eq.) in THF was added oleamide in one portion at 0 °C. The mixture was heated to reflux overnight. Then the mixture was quenched with water (3 drops) at 0 °C, followed by 1 M NaOH solution and then stirred at room temperature for 1 hour. The suspension was filtered through celite. The filtrate was diluted with ethyl acetate, washed with water and brine, dried (Na₂SO₄) and concentrated to give (Z)-octadec-9-en-1-amine 42 (726 mg, 95%) which was used for the next step without further purification. Step 3: To a solution of (Z)-octadec-9-en-1-amine 42 in DMF was added DIEA (2 eq.) and CDI (1.5 eq.) at 0 °C. The mixture was stirred at room temperature overnight. The mixture was then diluted with ethyl acetate, washed with NaHCO₃, brine, dried (Na₂SO₄) and concentrated. Purification of the crude residue on silica gel afforded (Z)-N-(octadec-9-en-1-yl)-1H-imidazole-1-carboxamide 43 (795 mg, 81%). Step 4: To a of (Z)-N-(octadec-9-en-1-yl)-1H-imidazole-1solution carboxamide 43 in DMF was added DIEA (2 eq.) and amino ester 44. The mixture was stirred at room temperature overnight. Then the mixture was diluted with ethyl acetate, washed with NaHCO₃, brine, dried (Na₂SO₄) and concentrated in vacuo. The crude residue was purified on silica gel to give intermediate **45** (75-80%). Step 5: To a solution of intermediate **45** in THF and H_2O (1:1) was added LiOH (5 eq.). The mixture was stirred at room temperature for 3 hours. The mixture was then acidified with HCl (1M) until pH<3. Then the mixture was extracted with ethyl acetate, washed with brine, dried (Na₂SO₄) and concentrated to give the desired compound **28** or **29** (94-96%).

PM20D1 hydrolysis assays. 10 nmol of the indicated compound was incubated in 100 µl PBS (100 µM initial substrate). Reactions were initiated by the addition of PBS or mPM20D1 (5 µl). After 1 h at 37°C, reactions were quenched with 600 µl of a 2:1 v/v mixture of chloroform and methanol with 10 nmol d_{31} -palmitate as an internal standard. The reactions were vortexed and the organic layer was transferred to a sample vial for analysis by LC-MS. For separation of polar metabolites, normal-phase chromatography was performed with a Luna-5 mm NH₂ column (50 mm \times 4.60 mm, Phenomenex). Mobile phases were as follows: Buffer A, acetonitrile: Buffer B, 95:5 water/ acetonitrile with 0.1% formic acid or 0.2% ammonium hydroxide with 50 mM ammonium acetate for positive and negative ionization mode, respectively. The flow rate for each run started at 0.2 ml/min for 2 min, followed by a gradient starting at 0% B and increasing linearly to 100% B over the course of 15 min with a flow rate of 0.7 ml/min, followed by an isocratic gradient of 100% B for 10 min at 0.7 ml/min before equilibrating for 5 min at 0% B with a flow rate of 0.7 ml/min. MS analysis was performed with an electrospray ionization (ESI) source on an Agilent 6430 QQQ LC-MS/MS. The capillary voltage was set to 3.5 kV, and the fragmentor voltage was set to 100 V. The drying gas temperature was 325 °C, the drying gas flow rate was 10 l/min, and the nebulizer pressure was 45 psi. Monitoring of hydrolysis starting materials and products was performed by scanning a mass range of m/z 50-1200. Peaks corresponding to the liberated fatty acids (products) or the intact starting material was integrated.

ASSOCIATED CONTENT

Supporting Information.

The Supporting information is available free of charge on the ACS Publications website at DOI:

Full characterization data on all analogs including Molecular Formula Strings, cell respiration and microsome stability assays, PM20D1 generation, Seahorse experimentals, critical micelle concentration measurements, and *in vivo* details.

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Author Contributions

The manuscript was written through contributions of all authors. / All authors have given approval to the final version of the manuscript.

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS USED

UCP1, uncoupling protein 1; TRPV1, transient receptor potential cation channel subfamily V member 1; GPR92, G-protein coupled receptor 92; SLC25, solute carrier family 25; DMSO, Dimethyl sulfoxide; CDI, Carbonyldiimidazole; DIPEA, N,N-Diisopropylethylamine; DMF, Dimethylformamide; DCM, Dichloromethane; THF, Tetrahydrofuran.

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