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Tandem double acylation/[3,3]-rearrangement of aliphatic nitro compounds: route to α -oxygenated oxime derivatives

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A new tandem double acylation/rearrangement reaction of nitro compounds is described. It has broad substrate scope allowing the mild and efficient synthesis of α -acyloxy oxime esters in high yields and regioselectivity. According to the obtained data, the mechanism for the transformation was proposed. Utility of the obtained α -hydroxy oxime esters was demonstrated.

Introduction

Rearrangement reactions constitute an important class of organic transformations,1 being particularly attractive the cases assisted by the cleavage of weak N–O bond.² Its average bond energy of 240 kJ/mol compared to the energies of other common C-X bonds (290-380 kJ/mol for X=C,N,O) allows the formation of carbon-carbon or carbon-heteroatom bond in a thermodynamically predictable fashion. Representative examples include Beckmann³ and Neber⁴ rearrangements as well as multiple variants of hetero-Claisen rearrangements.⁵ In a tandem process, rearrangement reactions are considered as invaluable tools for the quick modification of organic molecules. Some noticeable cases that involve the N-O bond cleavage are Trofimov pyrrole synthesis (tandem oxime vinylation, [3,3]-rearrangement, ring closure) (Scheme 1, eq. 1),⁶ exhaustive acylation of oximes (eq. 2),⁷ Bartoli indole synthesis (tandem vinylation, [3,3]-rearrangement, ring closure)⁸ (eq. 3), Fischer-like benzofuran synthesis (tandem oxime formation, [3,3]-rearrangement, ring closure)⁹ (eq. 4) and α -oxygenation of ketone (tandem enamine formation, [3,3]-rearrangement) developed by Tomkinson *et al.* (eq. 5).¹⁰ The majority of the mentioned reactions rely on such starting materials as hydroxylamines or oximes, i.e. compounds that already possess single N-O bond. However, nitro compounds could be useful precursors for the synthesis of compounds possessing N–O-bond, including those with a reactive C=C–N– O-moiety.¹¹ Thus, the possibility to combine the generation of C=C–N–O-group with N–O-cleavage assisted rearrangement looks very promising. This would allow the functionalization of a carbon backbone of the parent nitro compound and the synthesis of polyfunctional products.¹² Herein we report tandem double acylation/rearrangement of nitro compounds leading to acylated α -hydroxyoxime derivatives (Scheme 1, eq. 6).



Results and discussion

For the optimization studies, nitro compound **1a** and pivaloyl chloride were used (Table 1). Among the used solvents (Entries 1-4) the highest yield and regioselectivity was achieved in MeCN. Pyridine and DIPEA resulted in strong deceleration of reaction (Entries 6-9). With NEt₃ reaction was complete in 2 days, while use of DMAP as an additive allowed to shorten the

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reaction time to 1 day (*cf.* Entries 4 and 10). Decreasing the reaction temperature somewhat increased the regioselectivity, while significantly decreased the reaction rate (Entries 5, 11). Overall an optimal combination of yield/reaction rate/selectivity was achieved using NEt₃ as a base with catalytic amounts of DMAP in MeCN at 0°C (Entry 10).



NՉ	Base	Solv.	т,	time,	Recovery	Total yield
	(equiv.)		°C	h	1a , %ª	2a+2'a, %,
						(ratio 2a:2'a)ª
1	NEt₃ (2.5)	CH_2Cl_2	0	24	38	46 (4:1)
2	NEt₃ (2.5)	THF	0	24	87	0
3	NEt₃ (2.5)	DMF	0	48	29	27 (4:1)
4	NEt₃ (2.5)	MeCN	0	48	0	91 (15:1)
5	NEt ₃ (2.5)	MeCN	-20	72	25	57 (only 2a)
6	DIPEA (2.5)	MeCN	0	48	87	2
7	DIPEA (2.5)	MeCN	25	96	73	23 (1:2)
8	NEt ₃ (1) +	MeCN	0	24	40	50 (10:1)
	DIPEA (1.5)					
9	Py (2.5)	MeCN	0	24	100	0
10	NEt₃ (2.5) +	MeCN	0	24	0	93 (10:1)
	DMAP (0.1)					
11	NEt ₃ (2.5) +	MeCN	-20	96	20	65 (20:1)
	DMAP (0.1)					
^a Determined by ¹ H NMR with an internal standard (1,4-dinitrobenzene).						

With the optimized conditions in hand, we proceeded to apply them on differently substituted nitro compounds **1** to test the substrate scope (Scheme 2). As can be seen a rather high variety of α -acyloxyoxime derivatives **2** was obtained in high yields. Among tested acyl chlorides pivaloyl chloride, benzoyl chloride and 1-adamantanecarbonyl chloride gave target products **2b,c** smoothly. Acyl chlorides possessing α -protons (acetyl chloride, isobutyryl chloride) failed to give target products presumably to the ease of ketene formation. Use of more electrophilic reagent, *e.g. p*-nitrobenzoyl chloride, gave intractable product mixtures. This can be attributed to the instability of intermediate acyl nitronates **3** and their rearrangement into nitroso compounds (see proposed mechanism, Scheme 3), that is in accordance with previous results on the acylation of silyl nitronates.^{5a}



Scheme 2 Synthesis of oxime esters 2 *via* tandem double acylation-rearrangement of nitro compounds 1. Reagent ratio: RC(O)Cl (2.2 equiv.), NEt₃ (2.5 equiv.), DMAP (0.1 equiv.). MeCN: 0.5 mL / 1 mmol of 1. Scale: 0.4-5 mmol of starting nitro compound 1. An = 4-methoxyphenyl; Ad = 1-adamantyl. ^a Average yield for two diastereomers of starting compound 1 (see Experimental/ESI). ^b Starting compound 1r with unprotected OH-group, 3.4 equiv. of PivCl, 3.8 equiv. of NEt₃ were used.

Reactivity of nitro compounds 1 showed strong dependence substitution pattern. Moderately on their sterically encumbered nitro compounds 1 possessing CH₃ and CH₂groups at the β -position usually showed a full conversion of starting materials after 1 day at 0 °C. Bulkier substrates, e.g. those possessing a tertiary β -carbon, reacted slower thus requiring higher temperatures (25 °C) and prolonged reaction times (2 d). The regiochemistry of the process for nonsymmetric nitro compounds depended on the relative bulkiness of β - and β '-positions of the substrate 1. Thus, products 2a and 2f were obtained with moderate

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regiochemistry in favor of CH₃-functionalization (**2a**,**f**) vs. CH₂-functionalization (**2'a**,**2'f**), while oxime **2g** was obtained as a sole product. Expectedly, low regioselectivity was observed for secondary β - and β' -carbons (**2d**:**2'd** = 3:1, **2e**:**2'e** = 1:1). The competition between primary and tertiary β -positions resulted in the complete selectivity toward CH₃-group (products **2j**-**2r**). The described tandem acylation/rearrangement process tolerated functional groups such as ester (products **2f**,**g**,**j**), acetal (**2o**,**p**), and ether including silyl ones (**21**,**m**,**n**). The free hydroxyl_{ie}group was converted into pivalate ester (**2r**) but did Not interfere with the desired tandem process. The carbonyl moiety was tolerated for cyclic product **2k**, while for open-chain cases (**20**,**p**,**q**) the protection was needed.¹³

The structures of obtained products were supported by ¹H and ¹³C NMR spectra (including 2D) as well as by HRMS data. For the oxime ester **2q** single crystal X-ray analysis was also performed (Figure 1).¹⁴





Figure 1 General view of the compound **2q** in representation of atoms *via* thermal ellipsoids at 50% probability level.¹⁴

The proposed mechanism for the process is shown in the Scheme 3. At the first step the nitro compound **1** is acylated to give acyl nitronate **3**, thus consuming one equivalent of acyl chloride. At the second step intermediate **3** attacks second equivalent of acyl chloride forming iminium cation **A** (step 2), that is further β -deprotonated¹⁵ by a base giving rise to enamine **4** (step 3). The [3,3]-rearrangement of **4** ultimately gives target oxime ether **2** (step 4). The effect of catalytic amounts of DMAP may be explained by two reasons. Firstly,

DMAP is a known Lewis base additive for the acylation reactions via N-acyl iminium cations. Secondly, it may accelerate the deprotonation of nitro compound 1 as a less sterically hindered base compared to NEt₃. Next the stability/reactivity of intermediate acyl nitronates 3 should be mentioned. Occasional literature examples describe such compounds as rather unstable even at low temperatures.¹⁶ Thus, they might have been expected to suffer some side reactions like [2,3]-rearrangement or Nef reaction.¹⁷ Nevertheless the high yields of the target oximes 2 suggested the fast step 2, *i.e.* the nitronate **3** is acylated at a much faster rate than other side-processes.¹⁸ The intermediacy of cation A is supported by the isolation of nitroso acetal 5 alongside the product 20. The formation of 5 should be explained by the electrophilic ring closure and methoxy-group transfer.¹⁹ Sideprocess associated with the reactivity of enamines 4 results in the formation of enoxime 6. Here proposed reaction pathway involves the heterolysis of N–O bond and proton abstraction (Scheme 3). Yet rather synchronous nature of the [3,3]-rearrangement is corroborated by the configuration of the formed oxime C=N double bond. The predominant migration of the acyloxy-group syn- to the double C=C bond in enamine 4 should be expected. Indeed, usually oximes 2 were isolated as E-isomers.^{20,21} As exemplified by the product 2g, the E,Z-isomerization occurred at elevated temperatures (Scheme 3).

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from Scheme 2 the As can be seen tandem acylation/rearrangement smoothly proceeded with secondary nitro compounds 1. Use of primary nitro compound 1s did not give any acyloxy-oxime derivatives, that can be ascribed to the decomposition of intermediate nitronate 3s into nitrile oxide **B**.²² Performing the reaction in the presence of styrene allowed the isolation of the corresponding [3+2]-cycloaddition product - isoxazoline 7 (Scheme 4).



Presented tandem reaction nicely complements asymmetric The addition of nitroethane organocatalysis. to give cinnamaldehyde is known to corresponding γ-nitroaldehydes with high asymmetric induction at CH*-Ph, albeit completely non-diastereoselective.²³ Thus preparation of nitro compound 1p gave the mixture of syn- and antiequimolar ratio. isomers in nearly The tandem acylation/rearrangement sequence CH*-NO₂ removed stereocenter giving rise to one enantiomer of oxime (S)-2p in perfect yields (Scheme 5).



Acylated oximes are useful intermediates for organic synthesis.²⁴ Thus, to demonstrate the synthetic utility of obtained products some further transformations were performed (Scheme 6). Hydrogenation was accomplished to give either *O*-protected (9) or *N*-protected (10) amino alcohol derivatives.²⁵ Oxime ester was selectively cleaved by the action of methanolic ammonia (product 11), while the deoximation²⁶ gave ketone 12. Finally, tri-substituted pyridine was prepared starting from the oxime ester 2h.²⁷



i: NaBH₄, NiCl₂·6H₂O, EtOH, r.t.,1h ; ii: DMAP (10 mol%), toluene, 100°C, 8 h; iii: 7M NH₃ in MeOH, r.t., 45 min; iv: Fe, TMSCI, AcOH, THF, r.t., 30 min; then H₂O, 15 min; v: AnCH=CHCHO, I₂, NEt₃, toluene; 120 °C, 10 h An = 4-methoxyphenyl

Scheme 6 Synthetic transformations of obtained oxime esters 2.

Experimental

Determination of configuration of obtained compounds

Determination of *syn*- or *anti*- relative configuration in nitro compounds **1o**-**r** was made using coupling constants between C<u>H</u>-Ph and C<u>H</u>-NO₂ protons: *J* (anti-isomer) > *J* (*syn*-isomer) (see ESI for more details). Absolute configuration of CH*-Ph stereocenter in *anti*-**1p**, *syn*-**1p** and **2p** was assigned on the basis of the literature data.²³

General procedure for the synthesis of oxime esters 2 (GP).

The solution of nitro compound 1 in MeCN (2 mL/1 mmol of nitro compound 1) was cooled to 0 °C in argon atmosphere. NEt₃ (2.5 equiv), DMAP (0.1 equiv), and, after additional 10 min, acyl chloride (2.2 equiv) were consequently added with stirring. The reaction mixture was either left at the same temperature (for 1a-c,f-i) or warmed to r.t. (for 1d,e,j-r) and then maintained for 1 d (for 1a-c,f-h,j) or 2 d (for 1d,e,i,k-r). After that it was transferred into EtOAc (25 mL)/ H₂O(20 mL). The organic layer was washed with NaHSO₄ (0.5 M in H₂O, 20 mL), brine (20 mL), dried (Na₂SO₄) and evaporated. The residue was preadsorbed on Celite and subjected to column chromatography on silica (eluent: PE/EtOAc) to give target oxime esters 2. E,Z-configuration of oxime group was determined based on ¹³C NMR (anti-arrangement towards oxime OR-group results in higher chemical shift of α-carbon as compared with syn-arrangement).²¹ Note: prolonged exposure of oximes 2 at r.t. and higher temperatures (e.g, during evaporation on a rotary evaporator) may result in partial E,Zisomerization).

Characterization data for selected products **2** is presented below. For other products **2** see ESI.

3-(4-Methoxyphenyl)-2-(pivaloyloxyimino)propyl pivalate 2a and 1-(4-Methoxyphenyl)-2-(pivaloyloxyimino)propyl pivalate 2'a

Oxime esters **2a** and **2'a** were obtained from nitro compound **1a** (121 mg, 0.62 mmol) and pivaloyl chloride (0.17 mL, 0.17 g, 1.36 mmol) according to GP (0 °C, 1 d). Column

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chromatography (eluent: PE/EtOAc, 10:1) afforded 191 mg (82 %, 2a:2'a = 10:1 (¹H NMR)) of target oxime esters as colorless oil, that solidifies in a fridge. R_f = 0.38 (PE/EtOAc, 5:1, anisaldehyde). Pure 2a was obtained by crystallization from PE/EtOAc, 20:1. mp = 33-35 °C (PE/EtOAc, 20:1). (E)-2a:(Z)-2a = 10:1. ¹H NMR (300 MHz, CDCl₃): (E)-2a: δ 1.19 (s, 9H, t-Bu), 1.28 (s, 9H, t-Bu), 3.78 (s, 2H, CH₂Ar), 3.79 (s, 3H, OMe), 4.72 (s, 2H, CH₂-O), 6.85 (d, J = 8.6 Hz, 2H, CH_{Ar}), 7.09 (d, J = 8.5 Hz, 2H, CH_{Ar}). (Z)-2a (characteristic signals): δ 3.75 (s, 2H, CH₂Ar), 4.85 (s, 2 H, CH_2 –O), 7.20 (d, J = 8.6 Hz, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃): (E)-2a: δ 27.1 (Me₃C), 27.2 (Me₃C), 32.6 (CH₂Ar), 38.7 (Me₃C), 38.8 (Me₃C), 55.3 (OMe), 63.1 (CH₂-O), 114.4 (CH_{Ar}), 125.9 (C_{Ar}), 129.7 (CH_{Ar}), 158.8 (C_{Ar}-OMe), 163.0 (C=N), 174.7 (C=O), 177.6 (C=O). (Z)-2a (characteristic signals): δ 36.6 (CH₂Ar), 58.2 (CH₂-O), 129.9 (CH_{Ar}), 164.0 (C=N). 2'a (characteristic signals): ¹H NMR (300 MHz, CDCl₃): δ 1.91 (s, 3H, CH₃), 6.41 (s, 1H, CH–O), 7.36 (d, J = 8.6, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃): δ 11.7 (C(=N)-Me), 75.2 (CH-O), 127.7 (CH_{Ar}). HRMS (ESI): m/z calcd. for [C₂₀H₂₉NO₅+Na⁺]: 386.1938, found: 386.1927.

2-(Pivaloyloxyimino)cyclopentyl pivalate 2ba

Oxime ester 2ba was obtained from nitro compound 1b (188 mg, 1.63 mmol) and pivaloyl chloride (0.45 mL, 0.43 g, 3.6 mmol) according to GP (0 °C, 1 d). Column chromatography (eluent: PE/EtOAc, 10:1) afforded 414 mg (90 %) of target oxime ester 2ba as white solid. Rf = 0.38 (PE/EtOAc, 5:1, anisaldehyde). mp = 56-58 °C (PE). ¹H NMR (300 MHz, CDCl₃): δ 1.21 (s, 9H, t-Bu), 1.28 (s, 9H, t-Bu), 1.77-2.01 (m, 3H) and 2.03-2.13 (m, 1H) (2 × CH₂), 2.52-2.77 (m, 2H, CH₂), 5.59 (t, J = 5.4 Hz, 1H, CH–O). 13 C NMR (75 MHz, DEPT, HSQC, CDCl₃): δ 20.6 (CH₂), 27.1 (Me₃C), 27.3 (Me₃C), 27.5 (CH₂), 32.0 (CH₂), 38.7 (Me₃C), 38.8 (Me₃C), 73.5 (CH-O), 171.1 (C=N), 174.8 177.3 (C=O). HRMS (ESI): m/z calcd. (C=O). for [C₁₅H₂₅NO₄+Na⁺]: 306.1671, found: 306.1676.

2-(Pivaloyloxyimino)propyl pivalate 2ca

Oxime ester 2ca was obtained from nitro compound 1c (0.45 mL, 0.45 g, 5 mmol) and pivaloyl chloride (1.37 mL, 1.33 g, 11 mmol) according to GP (0 °C, 1 d). Column chromatography (eluent: PE/EtOAc, 7:1) afforded 1.02 g (79 %) of target oxime ester 2ca as slightly yellow oil. R_f = 0.33 (PE/EtOAc, 5:1, anisaldehyde). NMR matches previously reported data.^{5a}

1-(4-Methoxyphenyl)-2-(pivaloyloxyimino)butyl pivalate 2d and 4-(4-Methoxyphenyl)-3-(pivaloyloxyimino)butan-2-yl pivalate 2'd

Oxime esters 2d and 2'd were obtained from nitro compound 1d (139 mg, 0.67 mmol) and pivaloyl chloride (0.19 mL, 0.18 g, 1.5 mmol) according to GP (r.t., 2 d). Column chromatography (eluent: PE/EtOAc, 15:1, then 10:1) afforded 172 mg (68 %, 2d:2'd = 3:1 (¹H NMR)) of target oxime esters as colorless oil, that solidifies upon storage. R_f = 0.19 (PE/EtOAc, 9:1, UV, anisaldehyde). mp (2d+2'd) = 120-122 °C (PE/EtOAc, 10:1). ¹H NMR (300 MHz, COSY, CDCl₃): 2d: δ 0.98 (t, J = 7.6 Hz, 3H, CH₂Me), 1.27 (s, 18H, 2 × t-Bu), 2.22-2.38 (m, 2H, CH₂Me), 3.78 (s, 3H, OMe), 6.39 (s, 1H, CH–O), 6.89 (d, J = 8.7 Hz, 2H, CH_{Ar}), 7.34 (d, J = 8.7 Hz, 2H, CH_{Ar}). 2'd: δ 1.12 (s, 9H, t-Bu), 1.18 (s, 9H, t-Bu), 1.38 (d, J = 6.6 Hz, CHMe), 3.69 (d, J = 14.5 Hz, 1H, CH_{2a}Ar), 3.76 (s, 2H, CH₂Ar), 3.83 (d, J = 14.5 Hz, 1H, CH_{2b}Ar), 5.58 (q, J = 6.6 Hz, 1 H, CH–O), 6.82 (d, J = 8.5 Hz, 2H, CH_{Ar}), 7.06 (d, J = 8.5 Hz, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPATic HSQC, HMBC, CDCl₃): 2d: δ 10.9 (CH₂Me), 20!6¹⁰ (CH₂Me), B029.966 (Me₃C), 27.20 (Me₃C), 38.7 (Me₃C), 38.9 (Me₃C), 55.2 (OMe), 75.0 (CH–O), 114.4 (CH_{Ar}), 127.7 (C_{Ar}), 127.9 (CH_{Ar}), 159.8 (C_{Ar}– OMe), 167.7 (C=N), 174.6 (C=O), 176.6 (C=O). 2'd: δ 18.0 (CHMe), 27.0 (Me₃C), 27.1 (Me₃C), 32.0 (CH₂Ar), 38.6 (Me₃C), 38.7 (Me₃C), 55.2 (OMe), 70.2 (CH–O), 114.4 (CH_{Ar}), 126.5 (C_{Ar}), 129.3 (CH_{Ar}), 158.5 (<u>C</u>_{Ar}-OMe), 165.9 (C=N), 174.6 (C=O), 177.1 (C=O). HRMS (ESI): m/z calcd. for $[C_{21}H_{31}NO_5+Na^+]$: 400.2094, found: 400.2096.

Methyl 2-methyl-5-(pivaloyloxy)-4-(pivaloyloxyimino)pentanoate 2g

Oxime ester 2g was obtained from nitro compound 1g (302 mg, 1.7 mmol) and pivaloyl chloride (0.47 mL, 0.46 g, 3.8 mmol) according to GP (0 °C, 1 d). Column chromatography (eluent: PE/EtOAc, 7:1) afforded 409 mg (69 %) of target oxime ester 2g as colorless oil. R_f = 0.36 (PE/EtOAc, 5:1, anisaldehyde). Maintaining NMR sample of pure (E)-2g in $CDCl_3$ at 50 °C (oil bath) for 7 h resulted in ratio (E)-2g:(Z)-2g = 1:1. ¹H NMR (300 MHz, CDCl₃): (E)-2g: δ 1.24 (s, 9H, t-Bu), 1.25 (d, overlapped, 3H, CH-Me), 1.30 (s, 9H, t-Bu), 2.60 (dd, J = 13.3, 7.9 Hz, 1H, CH_{2a}), 2.76 (dd, J = 13.3, 7.3 Hz, 1H, CH_{2b}), 2.89 (app sex, J = 7.1 Hz, 1H, CH), 3.69 (s, 3H, CO₂Me), 4.74-4.84 (m, 2H, CH₂-O). (Z)-2g (characteristic signals): δ 1.25 (s, 9H, t-Bu), 1.27 (s, 9H, t-Bu), 2.46 (dd, J = 15.4, 6.6 Hz, 1H, CH_{2a}), 2.80-2.98 (m, 2H, CH and CH_{2b}), 3.70 (s, 3H, CO₂Me), 4.88-5.04 (m, 2H, CH₂–O). ¹³C NMR (75 MHz, DEPT, CDCl₃):(E)-**2g**: δ 17.2 (CH₃), 27.1 (Me₃C), 27.2 (Me₃C), 30.8 (CH₂), 36.4 (CH), 38.6 (Me₃C), 38.8 (Me₃C), 52.0 (CO₂Me), 63.5 (CH₂-O), 162.6 (C=N), 174.4 (C=O), 174.9 (C=O), 177.5 (C=O). (Z)-2g (characteristic signals): δ 17.0 (CH₃), 27.1 (Me₃C), 27.2 (Me₃C), 34.4 (CH₂), 36.2 (CH), 38.7 (Me₃C), 38.8 (Me₃C), 51.9 (CO₂Me), 59.4 (CH₂-O), 163.6 (C=N), 174.1 (C=O), 175.6 (C=O), 177.6 (C=O). HRMS (ESI): m/z calcd. for $[C_{17}H_{29}NO_6+Na^+]$: 366.1887, found: 366.1883.

2-Phenyl-2-(pivaloyloxyimino)ethyl pivalate 2h

Oxime ester 2h was obtained from nitro compound 1h (150 mg, 1.0 mmol) and pivaloyl chloride (0.27 mL, 0.27 g, 2.2 mmol) according to GP (0 °C, 1 d). Column chromatography (eluent: PE/EtOAc, 15:1) afforded 196 mg (61 %) of target oxime ester 2h as colorless oil, that solidifies in a fridge. R_f = 0.49 (PE/EtOAc, 5:1, anisaldehyde). mp = 85-87 °C (PE). (*E*)-**2h**:(*Z*)-**2h** = 1.7:1. ¹H NMR (300 MHz, CDCl₃): (*E*)-**2h**: δ 1.01 (s, 9H, t-Bu), 1.09 (s, 9H, t-Bu), 5.11 (s, 2H, CH₂-O), 7.35-7.44 (m, 5H, CH_{Ph}). (Z)-2h: δ 1.01 (s, 9H, t-Bu), 1.33 (s, 9H, t-Bu), 5.29 (m, 2 H, CH₂–O), 7.35-7.44 (m, 3H, CH_{Ph}), 7.61 (d, J = 7.8 Hz, 2H, CH_{Ph}). ¹³C NMR (75 MHz, DEPT, CDCl₃): (E)-**2h**: δ 26.8 (Me₃C), 26.9 (Me₃C), 38.4 (Me₃C), 38.7 (Me₃C), 63.5 (CH₂-O), 127.7 (CH_{Ph}), 128.2 (CH_{Ph}), 130.0 (CH_{Ph}), 130.3 (C_{Ph}), 162.5 (C=N), 174.8 (C=O), 177.4 (C=O). (Z)-2h (characteristic signals): δ 26.8 (Me₃C), 27.2 (Me₃C), 38.6 (Me₃C), 38.8 (Me₃C), 57.3 (CH₂-O), 128.1 (CH_{Ph}), 128.3 (CH_{Ph}), 130.5 (CH_{Ph}), 131.7 (C_{Ph}), 162.8 (C=N), 174.4 (C=O), 177.5 (C=O). HRMS (ESI): m/z calcd. for [C₁₈H₂₅NO₄+Na⁺]: 342.1676, found: 342.1674.

1-Phenyl-1-(pivaloyloxyimino)propan-2-yl pivalate 2i

Oxime ester 2i was obtained from nitro compound 1i (77 mg, 0.47 mmol) and pivaloyl chloride (0.13 mL, 0.13 g, 1.03 mmol)

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according to GP (0 °C, 2 d). Column chromatography (eluent: PE/EtOAc, 30:1, then 15:1) afforded 103 mg (66 %) of target oxime ester 2i as colorless oil, that solidifies in a fridge. R_f = 0.56 (PE/EtOAc, 9:1, anisaldehyde). mp = 115-118 °C (PE/EtOAc, 10:1). (E)-2i:(Z)-2i = 1:1. ¹H NMR (300 MHz, COSY, for (*E*,*Z*)-mixture, CDCl₃): δ 0.93 (s, 9H, *t*-Bu), 1.05 (s, 9H, *t*-Bu), 1.08 (s, 9H, t-Bu), 1.35 (s, 9H, t-Bu), 1.55 (d, J = 6.6 Hz, 3H, (E)isomer, Me), 1.73 (d, J = 7.0 Hz, 3H, (Z)-isomer, Me), 5.83 (q, J = 6.6 Hz, 3H, (E)-isomer, CH-O), 6.13 (q, J = 7.0 Hz, 3H, (Z)isomer, CH–O), 7.28-7.43 (m, 5H ((E)-isomer) + 3H ((Z)-isomer), CH_{Ph}), 7.59 (d, J = 7.9 Hz, 2H, (Z)-isomer, CH_{Ph}). ¹³C NMR (75 MHz, DEPT, HSQC, for (E,Z)-mixture, CDCl₃): δ 17.8 (Me, (E)isomer), 18.2 (Me, (Z)-isomer), 26.7 (Me₃C), 26.9 (2 × Me₃C), 27.3 (Me₃C), 38.4 (2 × Me₃C), 38.6 (Me₃C), 38.8 (Me₃C), 66.3 (CH–O, (Z)-isomer), 70.3 (CH–O, (E)-isomer), 127.4 (CH_{Ph}), 128.0 (CH_{Ph}),128.1 (CH_{Ph}), 128.6 (CH_{Ph}), 129.4 (CH_{Ph}), 130.1 (CH_{Ph}), 131.1 (C_{Ph}), 131.2 (C_{Ph}), 166.4 (C=N), 167.4 (C=N), 174.3 (C=O), 174.7 (C=O), 177.3 (2 × C=O). HRMS (ESI): m/z calcd. for [C₁₉H₂₇NO₄+Na⁺]: 356.1832, found: 356.1838.

3-(*tert*-Butyldimethylsilyloxy)-2-(pivaloyloxyimino)butyl pivalate 2l

Oxime ester 2I was obtained from nitro compound 1I (111 mg, 0.48 mmol) and pivaloyl chloride (0.13 mL, 0.13 g, 1.06 mmol) according to GP (r.t., 2 d). Column chromatography (eluent: PE/EtOAc, 20:1) afforded 127 mg (66 %) of target oxime ester 21 as colorless oil. R_f = 0.67 (PE/EtOAc, 5:1, anisaldehyde). (E)-2I:(Z)-2I = 1:4. ¹H NMR (300 MHz, COSY, CDCl₃): (Z)-2I: δ 0.07 (s, 3 H, MeSi), 0.09 (s, 3 H, MeSi), 0.90 (s, 9H, t-BuSi), 1.25 (s, 9H, *t*-Bu-C), 1.29 (s, 9H, *t*-Bu-C), 1.43 (d, *J* = 6.5 Hz, 3H, CH₃), 4.67 (d, J = 13.7 Hz, 1H, CH_{2a}–OPiv), 5.12 (d, J = 13.7 Hz, 1H, CH_{2b}-OPiv), 5.12 (q, J = 6.5 Hz, 1H, CH-OTBS). (E)-2I (characteristic signals): δ 4.74 (q, J = 6.6 Hz, 1H, CH–OTBS), 4.82-4.92 (m, 2H, CH2-OPiv). ¹³C NMR (75 MHz, DEPT, HSQC, CDCl₃): (Z)-2I: δ -5.3 (MeSi), -4.9 (MeSi), 18.0 (<u>C</u>-Si), 21.6 (CH₃), 25.6 (t-BuSi), 27.2 (2 × Me₃C-C), 38.7 (Me₃C), 38.8 (Me₃C), 60.9 (CH₂-O), 64.8 (CH-O), 167.4 (C=N), 174.0 (C=O), 177.6 (C=O). (E)-2I (characteristic signals): δ 22.0 (CH₃), 54.8 (CH₂-O), 68.8 (CH–O). HRMS (ESI): m/z calcd. for $[C_{20}H_{39}NO_5Si+Na^+]$: 424.2490, found: 424.2485.

5,5-Dimethoxy-3-phenyl-2-(pivaloyloxyimino)pentyl pivalate 20

Oxime ester 20 was obtained from nitro compound syn-10 (125 mg, 0.49 mmol) and pivaloyl chloride (0.13 mL, 0.13 g, 1.1 mmol) according to GP (r.t., 2 d). Column chromatography (eluent: PE/EtOAc, 15:1, then 9:1) afforded 15 mg of nitroso acetal 5, 71 mg of mixture 2o+5 (2o:5 = 6:1 (¹H NMR)) and 71 mg of oxime ester 2o as colorless oils. Total yields: 2o: 64 %, 5: 12 %. Similar procedure for anti-1o (177 mg, 0.7 mmol) gave 182 mg (62%) of target oxime ester 20. Oxime ester 20: $R_f =$ 0.36 (PE/EtOAc, 5:1, anisaldehyde). Only (E)-20. ¹H NMR (300 MHz, COSY, CDCl₃): δ 1.15 (s, 9H, *t*-Bu), 1.28 (s, 9H, *t*-Bu), 2.32 (dd, J = 7.9, 5.7 Hz, 1H, CH₂), 3.30 (s, 3H, OMe), 3.31 (s, 3H, OMe), 4.34 (t, J = 5.7 Hz, 1H, CH(OMe)₂), 4.62 (t, J = 7.9 Hz, 1H, CHPh), 4.68 (s, 2 H, CH2-O), 7.24-7.35 (m, 5H, Ph). ¹³C NMR (75 MHz, DEPT, HSQC, CDCl₃): δ 27.0 (Me₃C), 27.1 (Me₃C), 33.9 (CH₂), 38.6 (Me₃C), 38.7 (Me₃C), 40.3 (CHPh), 52.4 (OMe), 53.7 (OMe), 62.8 (CH₂-O), 102.3 (<u>C</u>H(OMe)₂), 127.5 (CH_{Ph}), 127.7 (CH_{Ph}), 128.9 (CH_{Ph}), 137.7 (C_{Ph}), 164.1 (C=N), 174.6 (C=O),

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 177.5 (C=O). HRMS (ESI): m/z calcd. for [C₂₃H_{∂5}NQ₆₅NQ₆NQ₆₅NQ

Oxime ester 2p was obtained from nitro compound anti-1p (86 mg, 0.35 mmol) and pivaloyl chloride (0.10 mL, 0.10 g, 0.84 mmol) according to GP (r.t., 2 d). Column chromatography (eluent: PE/EtOAc, 6:1) afforded 131 mg (89 %) of target oxime ester **2p** as colorless oil. R_f = 0.28 (PE/EtOAc, 5:1, anisaldehyde). Similar procedure for syn-1p (98 mg, 0.39 mmol) gave 140 mg (86%) of target oxime ester 2p. Similar procedure for (+)-anti-1p (97 mg, 0.39 mmol) gave 145 mg (90%, ee = 95%) of target oxime ester (-)-2p. Similar procedure for (+)-syn-1p (90 mg, 0.36 mmol) gave 133 mg (89%, ee = 96%) of target oxime ester (–)-**2p**. (–)-(*S*)-**2p**: [*a*]²⁴_D = -93.5 (*c* = 1.0, MeOH, 96% ee, E-isomer). ¹H NMR (300 MHz, COSY, CDCl₃): (E)-2p: δ 1.11 (s, 9H, t-Bu), 1.26 (s, 9H, t-Bu), 2.26-2.48 (m, 2H, CH₂), 3.77-3.84 and 3.88-3.97 (m, 4H, OCH₂CH₂O), 4.66 (d, J = 13.1 Hz, 1H, CH_{2a}-OPiv), 4.73 (d, J = 13.1 Hz, 1H, CH_{2b}-OPiv), 4.76 (t, J = 7.7 Hz, 1H, CHPh), 4.87 (t, J = 4.6 Hz, 1H, CH(OCH₂)₂), 7.18-7.35 (m, 5H, Ph). (Z)-2p: δ 1.13 (s, 9H, t-Bu), 1.28 (s, 9H, t-Bu), 2.17 (ddd, J = 13.9, 7.6, 4.3 Hz, 1H, CH_{2a}), 2.51 (ddd, J = 13.9, 7.6, 5.8 Hz, 1H, CH_{2a}), 3.75-3.85 and 3.87-3.96 (m, 4H, OCH₂CH₂O), 4.13 (t, J = 7.6 Hz, 1H, CHPh), 4.52 (d, J = 14.7 Hz, 1H, CH_{2a}-OPiv), 4.84 (dd, J = 5.8, 4.3 Hz, 1H, C<u>H(OCH₂)₂), 4.94</u> (d, J = 14.7 Hz, 1H, C<u>H_{2b}</u>–OPiv), 7.22-7.34 (m, 5H, Ph). ¹³C NMR (75 MHz, DEPT, HSQC, CDCl₃): (E)-**2p**: δ 27.0 (Me₃C), 27.1 (Me₃C), 34.9 (CH₂), 38.6 (Me₃C), 38.7 (Me₃C), 39.5 (CHPh), 62.8 (CH2-OPiv), 64.9 and 65.0 (OCH2CH2O), 102.2 (CH(OCH₂)₂), 127.4 (CH_{Ph}), 127.7 (CH_{Ph}), 128.8 (CH_{Ph}), 137.8 (C_{Ph}), 164.2 (C=N), 174.6 (C=O), 177.5 (C=O). (Z)-2p: δ 27.0 (Me₃C), 27.2 (Me₃C), 37.0 (CH₂), 38.7 (Me₃C), 38.8 (Me₃C), 43.6 (<u>C</u>HPh), 58.2 (<u>C</u>H₂–OPiv), 64.8 (OCH₂CH₂O), 102.4 (<u>C</u>H(OCH₂)₂), 127.4 (CH_{Ph}), 128.3 (CH_{Ph}), 128.8 (CH_{Ph}), 138.7 (C_{Ph}), 165.4 (C=N), 174.5 (C=O), 177.5 (C=O). HRMS (ESI): m/z calcd. for $[C_{23}H_{33}NO_6+H^+]$: 420.2381, found: 420.2374. HPLC separation conditions: column: Chiralpak AD-3, 250 x 4.6 mm, temp. 25 °C, eluent: Hexane-i-PrOH, 90:10, 1 mL/min, detection at 207 nm. t_R = 4.9 min (Z-isomer, minor), 5.2 min (Z-isomer, major), 6.3 min (E-isomer, major), 7.1 (E-isomer, minor). 3-Phenyl-2-(pivaloyloxyimino)pentane-1,5-diyl bis(2,2-

dimethylpropanoate) 2r

Oxime ester **2r** was obtained from nitro compound *syn*-**1r** (86 mg, 0.42 mmol) according to GP (r.t., 2 d) with the change: increasing amounts of pivaloyl chloride (3.4 equiv., 0.18 mL, 0.17 g, 1.4 mmol) and NEt₃ (3.8 equiv., 0.22 mL, 0.16 g, 1.6 mmol) were taken. Column chromatography (eluent: PE/EtOAc, 10:1) afforded 156 mg (82 %) of target oxime ester **2r** as colorless oil, that solidifies in a fridge. R_f = 0.44 (PE/EtOAc, 5:1, anisaldehyde). mp = 73-75 °C (PE/EtOAc, 10:1). Similar procedure for *anti*-**1r** (94 mg, 0.45 mmol) gave 160 mg (77 %) of target oxime ester **2r**. Only (*E*)-**2r**. ¹H NMR (300 MHz, CDCl₃): δ 1.16 (s, 18H, 2 × *t*-Bu), 1.27 (s, 9H, *t*-Bu), 2.37 (app q, *J* = 7.0 Hz, CH₂(4)), 4.00-4.18 (m, 2H, CH₂(5)–OPiv), 4.55 (app t, *J*= 7.8 Hz, CHPh), 4.62-4.72 (m, 2H, CH₂(1)–OPiv), 7.23-7.33 (m, 5H, Ph). ¹³C NMR (75 MHz, DEPT, HSQC, CDCl₃): δ 27.07 (Me₃C), 27.12 (Me₃C), 27.18 (Me₃C), 29.8 (CH₂(4)), 38.63

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 $\begin{array}{l} (Me_3\underline{C}), 38.68 \; (Me_3\underline{C}), 38.74 \; (Me_3\underline{C}), 41.4 \; (\underline{C}HPh), 62.2 \; (\underline{C}H_2(5)-OPiv), \; 62.7 \; (\underline{C}H_2(1)-OPiv), \; 127.7 \; (CH_{Ph}), \; 127.8 \; (CH_{Ph}), \; 129.0 \\ (CH_{Ph}), \; 137.2 \; (C_{Ph}), \; 163.4 \; (C=N), \; 174.3 \; (C=O), \; 177.4 \; (C=O), \\ 178.2 \; (C=O). \; HRMS \; (ESI): \; m/z \; calcd. \; for \; [C_{26}H_{39}NO_6+Na^+]: \\ 484.2670, \; found: \; 484.2659. \end{array}$

O-Pivaloyl-*N*-(pivaloyloxy)-*N*-((3*E*)-1,4-dimethoxy-1-methylbut-3en-1-yl)hydroxylamine 5

Obtained as a side-product during acylation of *syn*-**10**. $R_f = 0.44$ (PE/EtOAc, 5:1, anisaldehyde). Relative configuration of stereocenters was not determined. ¹H NMR (300 MHz, CDCl₃): δ 1.13 (s, 9H, *t*-Bu), 1.25 (s, 9H, *t*-Bu), 1.28 (s, 3H, Me), 3.51 (s, 3H, N–C–OMe), 3.52 (s, 3H, =CHO<u>Me</u>), 3.84 (d, *J* = 9.2 Hz, 1H, C<u>H</u>Ph), 5.36 (dd, *J* = 12.7, 9.2 Hz, =C<u>H</u>–CH), 6.21 (d, *J* = 12.7 Hz, 1H, =CH–O), 7.19-7.35 (m, 5H, Ph). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃): δ 17.9 (Me), 26.8 (<u>Me₃C</u>), 27.0 (<u>Me₃C</u>), 38.4 (Me₃C), 38.5 (Me₃C), 50.6 (<u>C</u>HPh), 52.2 (N–C–O<u>Me</u>), 56.2 (=CHO<u>Me</u>), 98.7 (C–N), 104.1 (=<u>C</u>H–CH), 126.4 (CH_{Ph}), 128.3 (CH_{Ph}), 129.3 (CH_{Ph}), 142.7 (C_{Ph}), 148.1 (=CH–O), 174.2 (C=O), 174.5 (C=O). HRMS (ESI): m/z calcd. for [C₂₃H₃₅NO₆+Na⁺]: 444.2357, found: 444.2374.

Methyl 3-(5-phenyl-4,5-dihydroisoxazol-3-yl)propanoate 7

The solution of nitro compound **1s** (115 mg, 0.78 mmol) in MeCN (1.55 mL) was cooled to 0 °C in argon atmosphere. Styrene (0.27 mL, 0.24 g, 2.3 mmol), NEt₃ (0.27 mL, 0.19 g, 2.0 mmol), DMAP (9.5 mg, 0.08 mmol), and, after additional 10 min, pivaloyl chloride (0.21 ml, 0.21 g, 1.7 mmol) were consequently added with stirring. The reaction mixture was maintained at the same temperature for 1 d, transferred into EtOAc (50 mL)/ H₂O(40 mL). The organic layer was washed with NaHSO₄ (0.5 M in H₂O, 40 mL), brine (40 mL), dried (Na₂SO₄) and evaporated. The residue was subjected to column chromatography (eluent: PE/EtOAc, 6:1, then 3:1) to give 82 mg (45%) of isoxazoline **7** as colorless oil. R_f = 0.42 (PE/EtOAc, 1:1, anisaldehyde). NMR matches previously reported data.²⁸

2-Amino-3-(4-methoxyphenyl)propyl pivalate 9

To the solution of oxime ester 2a (98 mg, 0.27 mmol) and NiCl₂·6H₂O (130 mg, 0.55 mmol) in EtOH (2.8 mL) NaBH₄ (102 mg, 2.7 mmol) was added portionwise during 30 min at 0 °C. The reaction mixture was stirred at r.t. for 1 h and transferred into EtOAc (100 mL)/ H₂O(75 mL). The water layer was extracted with EtOAc (3 × 20 mL). The combined organic layer was washed with brine (150 mL), dried (Na₂SO₄) and evaporated to give crude amine 9. Column chromatography (eluent: PE/EtOAc, 1:1, then PE/EtOAc/MeOH, 1:1:0.2) afforded 33 mg (46 %) of target amine 9 as colorless oil. $R_f =$ 0.13 (EtOAc, ninhydrine). ¹H NMR (300 MHz, CDCl₃): δ 1.25 (s, 9H, t-Bu), 1.90 (br s, 2H, NH₂), 2.57 (dd, J = 13.6, 8.0 Hz, 1H, CH_{2a}-Ar), 2.78 (dd, J = 13.6, 5.5 Hz, 1H, CH_{2b}-Ar), 3.29 (app br quint, J = 6.2 Hz, 1H, CH-N), 3.80 (s, 3H, OMe), 3.92 (dd, J = 10.9, 6.8 Hz, 1H, CH_{2a}–O), 4.05 (dd, J = 10.9, 4.5 Hz, 1H, CH_{2b}– O), 6.86 (d, J = 8.5 Hz, 2H, CH_{Ar}), 7.13 (d, J = 8.5 Hz, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPT, HMBC, CDCl₃): δ 27.3 (Me₃C), 38.9 (Me₃<u>C</u>), 39.5 (<u>C</u>H₂-Ar), 51.7 (CH-N), 55.3 (OMe), 68.4 (CH₂-O), 114.0 (CH_{Ar}), 130.0 (C_{Ar}), 130.2 (CH_{Ar}), 158.3 (<u>C</u>_{Ar}-OMe), 178.3 (C=O). HRMS (ESI): m/z calcd. for $[C_{15}H_{23}NO_3+H_{VE^{+}}^+]_{W^{2}} \times 266.1751_{W^{2}}$ found: 266.1743. DOI: 10.1039/C9OB01005J

N-(1-Hydroxy-3-(4-methoxyphenyl)propan-2-yl)pivalamide 10

Oxime ester 2a (94 mg, 0.26 mmol) was subjected to hydrogenation as described for the synthesis of amine 9. Then the solution of crude amine 9 and DMAP (3.1 mg, 0.03 mmol) in toluene (0.52 mL) was heated at 100 °C (oil bath) for 8 h. The reaction mixture was evaporated and preadsorbed on Celite[®]. Column chromatography (eluent: PE/EtOAc, 1:1, then PE/EtOAc/MeOH, 1:1:0.1) afforded 30 mg (44 %) of target amide **10** as colorless oil. R_f = 0.13 (PE/EtOAc, 1:1, Hanessian stain). ¹H NMR (300 MHz, COSY, CDCl₃): δ 1.13(s, 9H, *t*-Bu), 2.77 (dd, J = 13.9, 7.6 Hz, 1H, CH_{2a}–Ar), 2.87 (dd, J = 13.9, 6.9 Hz, 1H, CH_{2b}-Ar), 3.44 (br s, 1H, OH), 3.60 (dd, J = 11.0, 5.3 Hz, 1H, CH_{2a}–O), 3.68 (dd, J = 11.0, 3.6 Hz, 1H, CH_{2b}–O), 3.79 (s, 3H, OMe), 4.03-4.15 (m, 1H, CH-N), 5.89 (br d, J = 6.6 Hz, 1H, NH), 6.85 (d, J = 8.5 Hz, 2H, CH_{Ar}), 7.14 (d, J = 8.5 Hz, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃): δ 27.5 (Me₃C), 36.0 (<u>C</u>H₂-Ar), 38.7 (Me₃<u>C</u>), 52.9 (CH-N), 55.3 (OMe), 64.6 (CH₂-O), 114.0 (CH_{Ar}), 129.6 (C_{Ar}), 130.2 (CH_{Ar}), 158.4 (<u>C</u>_{Ar}–OMe), 179.4 (C=O). HRMS (ESI): m/z calcd. for [C₁₅H₂₃NO₃+H⁺]: 266.1751, found: 266.1748.

2-(Hydroxyimino)-3-(4-methoxyphenyl)propyl pivalate 11 and 2-(Hydroxyimino)-1-(4-methoxyphenyl)propyl pivalate 11'

Mixture of oxime esters 2a+2'a (121 mg, 0.33 mmol) was dissolved in the solution of NH₃ in MeOH (7M, 1.43 mL). Reaction mixture was stirred for 45 min at r.t., transferred into EtOAc (50 mL) / $H_2O(40$ mL). The organic layer was washed with NaHSO₄ (0.5 M in H_2O , 30 mL), brine (40 mL), dried (Na₂SO₄) and evaporated. Column chromatography (eluent: PE/EtOAc, 6:1) afforded 80 mg (87 %) of target oxime 11+11' (11:11' = 12:1 (¹H NMR)) as colorless oil, that solidifies in a fridge. R_f = 0.33 (PE/EtOAc, 5:1, UV, anisaldehyde). mp = 58-60 °C (PE/EtOAc, 20:1). (E)-11:(Z)-11 = 10:1. ¹H NMR (300 MHz, CDCl₃): (E)-11: δ 1.21 (s, 9H, t-Bu), 3.76 (s, 2H, CH₂Ar), 3.80 (s, 3H, OMe), 4.59 (s, 2H, CH₂–O), 6.85 (d, J = 8.6 Hz, 2H, CH_{Ar}), 7.19 (d, J = 8.6 Hz, 2H, CH_{Ar}), 9.22 (br s, 1H, NOH). (Z)-11 (characteristic signals): δ 3.59 (s, 2H, CH₂Ar), 4.95 (s, 2 H, CH₂-O). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃): (E)-**11**: δ 27.1 (Me₃C), 30.8 (CH₂Ar), 38.8 (Me₃C), 55.3 (OMe), 63.6 (CH₂-O), 114.1 (CH_{Ar}), 127.5 (C_{Ar}), 130.1 (CH_{Ar}), 155.4 (C=N), 158.4 (CAr-OMe), 178.0 (C=O). (Ζ)-11 (characteristic signals): δ 36.3 (CH₂Ar), 58.4 (CH₂–O), 129.8 (CH_{Ar}). **11'** (characteristic signals): ¹H NMR (300 MHz, CDCl₃): δ 1.86 (s, 3H, CH₃), 6.30 (s, 1H, CH– O), 7.31 (d, J = 8.7, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃, characteristic signals): δ 10.4 (C(=N)–<u>Me</u>), 75.6 (CH–O), 127.8 (CH_{Ar}). HRMS (ESI): m/z calcd. for [C₁₅H₂₁NO₄+Na⁺]: 302.1363, found: 302.1362.

3-(4-Methoxyphenyl)-2-oxopropyl pivalate 12 and 3-(4methoxyphenyl)-2-oxopropyl pivalate 12'

Ketone **12** was obtained similar to the reported procedure.²⁶ To the solution of oxime esters **2a+2'a** (145 mg, 0.4 mmol) in THF (4 mL) Fe powder (223 mg, 4 mmol), AcOH (2 drops) and TMSCI (2 drops) were consequently added. The reaction mixture was stirred at r.t. for 0.5 h. Water (4 mL) was added. The mixture was stirred for additional 15 min, transferred into

EtOAc (55 mL)/ H₂O (15 mL). The organic layer was washed with NaHCO₃ (sat. aq., 50 mL), brine (50 mL), dried (Na₂SO₄) and evaporated. The residue was preadsorbed on Celite® and subjected to column chromatography (eluent: PE/EtOAc, 10:1) to give 50 mg (47 %) of target ketones 12+12' (12:12'=11:1) as colorless oil, that solidifies in a fridge. $R_f = 0.47$ (PE/EtOAc, 5:1, anisaldehyde). mp = 29-31 °C (PE). 12: ¹H NMR (300 MHz, CDCl₃): δ 1.28 (s, 9H, *t*-Bu), 3.68 (s, 2H, CH₂-Ar), 3.81 (s, 3H, OMe), 4.69 (s, 2H, CH₂-O), 6.89 (d, J = 8.5 Hz, 2H, CH_{Ar}), 7.15 (d, J = 8.8 Hz, 2H, CH_{Ar}). ¹³C NMR (75 MHz, DEPT, CDCl₃): δ 27.2 (Me₃C), 38.7 (Me₃C), 45.4 (CH₂-Ar), 55.3 (OMe), 67.4 (CH₂-O), 114.3 (CH_{Ar}), 124.8 (C_{Ar}), 130.5 (C_{Ar}), 158.9 (C_{Ar}-OMe), 177.8 (COO), 201.7 (C=O). 12': ¹H NMR (300 MHz, CDCl₃): δ 1.30 (s, 9H, t-Bu), 2.12 (s, 3H, Me), 3.83 (s, 3H, OMe), 5.89 (s, 1H, CH-O), 6.94 (d, J = 8.7 Hz, 2H, CH_{Ar}), 7.35 (d, J = 8.7 Hz, 2H, CH_{Ar}). HRMS (ESI): m/z calcd. for $[C_{15}H_{20}O_4+Na^+]$: 287.1254, found: 287.1255.

4-(4-Methoxyphenyl)-2-phenylpyridin-3-yl pivalate 13

Substituted pyridine 13 was obtained similar to the reported procedure.²⁷ The solution of oxime ester 2h (82.3 mg, 0.26 mmol), p-methoxycinnamaldehyde (63 mg, 0.39 mmol), NEt₃ (18 $\mu L,$ 13 mg, 0.13 mmol) and I_2 (33 mg, 0.13 mmol) in toluene (1.3 mL) was heated at 120 °C (oil bath) for 9 h. The reaction mixture was transferred into EtOAc (50 mL)/ H₂O (20 mL). The organic layer was washed with brine (30 mL), dried (Na₂SO₄) and evaporated. Column chromatography (eluent: PE/EtOAc, 9:1, then 7:1) afforded 43.7 mg (47 %) of target pyridine **13** as yellow oil. R_f = 0.18 (PE/EtOAc, 5:1, UV). ¹H NMR (300 MHz, COSY, CDCl₃): δ 0.93 (s, 9H, *t*-Bu), 3.86 (s, 3H, OMe), 6.98 (d, J = 8.6 Hz, 2H, CH_{Ar}), 7.30 (d, J = 5.0 Hz, 1H, CH(5)), 7.29-7.47 (m, 5H, CH_{Ar} and CH_{Ph}), 7.67 (d, J = 8.0 Hz, 2H, CH_{Ph}), 8.60 (d, J = 5.0 Hz, 1H, CH(6)). ¹³C NMR (75 MHz, DEPT, HSQC, HMBC, CDCl₃): δ 26.8 (Me₃C), 38.7 (Me₃C), 55.4 (OMe), 113.8 (CH_{Ar}), 124.1 (CH(5)), 127.7 (C_{Ar}), 128.0, 128.5, 129.1 and 130.2 (3×CH_{Ph} and CH_{Ar}), 137.3 (C_{Ph}), 142.7 (C(3)), 144.0 (C(4)), 146.9 (CH(6)), 153.7 (C(2)), 159.9 (CAr-OMe), 175.6 (C=O). HRMS (ESI): m/z calcd. for [C₂₃H₂₃O₃+H⁺]: 362.1751, found: 362.1752.

Conclusions

In summary, a new tandem double acylation/rearrangement of nitro compounds was investigated. The overall process represents the C-H functionalization of β -position of starting nitro compound with the concomitant conversion of nitro to oxime moiety. Considering ready availability of starting nitro compounds by simple means (Henry, Michael reactions) this allows a quick increase in molecular complexity in a few steps. The reaction has rather broad substrate scope, proceeds with high regioselectivity and can be used in asymmetric synthesis. Based on the obtained data, the mechanism for the possible including transformation side-reactions was proposed. Synthetic utility of the obtained α -hydroxy oxime esters was demonstrated.

Conflicts of interest

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There are no conflicts to declare.

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Novel efficient regioselective method for the carbon chain activation of aliphatic nitro compounds is described using tandem double acylation/[3,3]-rearrangement sequence.