

# Synthesis of Naphthyl-, Quinolin- and Anthracenyl Analogues of Clofibric Acid as PPARα Agonists

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PPAR $\alpha$  is a ligand activated transcription factor belonging to the nuclear receptor subfamily, involved in fatty acid metabolism in tissues with high oxidative rates such as muscle, heart and liver. PPARa activation is important in steatosis, inflammation and fibrosis in preclinical models of non-alcoholic fatty liver disease identifying a new potential therapeutic area. In this work, three series of clofibric acid analogues conjugated with naphthyl, quinolin, chloroquinolin and anthracenyl scaffolds were synthesized. In an effort to obtain new compounds active as PPARa agonists, these molecules were evaluated for PPARa transactivation activity. Naphthyl and quinolin derivatives showed a good activation of PPARa; noteworthy, optically active naphthyl derivatives activated PPARa better than corresponding parent compound.

Key words: anthracenyl, clofibric acid, naphthyl, peroxisome proliferator-activated receptor agonist, quinolin, transactivation assay

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Peroxisome proliferator-activated receptors (PPARs) are a subfamily of ligand-activated nuclear hormone receptors that are highly expressed in metabolically active tissues, where regulate genes encoding lipid and glucose metabolism, and overall energy homeostasis. Three PPAR subtypes, commonly designated as PPAR $\alpha$ , PPAR $\gamma$  and PPAR $\delta(\beta)$ , have been identified; these receptors differ in their tissue distributions, selectivity and responsiveness to ligands, thus leading to the regulation of different sets of genes involved in lipid metabolism and energy balance (1,2). PPAR $\alpha$  is highly expressed in metabolically active tissues including liver, muscle, intestine and brown adipose

tissue; it plays a pivotal role on lipid metabolism by decreasing both serum triglycerides and free fatty acid levels, and increasing high-density lipoprotein level (HDL) (3). PPAR $\alpha$  is activated by the fibrate antilipidemic drugs (4). The PPAR $\gamma$  controls fatty acid storage and glucose metabolism by coordinating the expression of genes involved in lipid metabolism, adipogenesis and inflammation (5). This receptor is activated by the thiazolidinedione (TZD) antidiabetic drugs (6). PPAR $\beta/\delta$  regulates fatty acid catabolism, insulin sensitivity and energy homeostasis in muscle and adipose tissue (7). Some phenoxyacetic acid derivatives, L-165041, GW501516, GW0742 and MBX-8025 are very potent and selective activators of PPAR $\beta$ both *in vitro* and *in vivo*. Although PPAR $\beta$  agonists are not yet in clinical use, human studies have been performed providing very encouraging findings for the treatment of metabolic disorders in dyslipidemic obese individuals (8-10).

The structures of all PPARs subtypes are similar: an amino terminal activation domain (AF-1), a DNA binding domain (DBD), a ligand binding domain (LBD) and a second carboxy terminal activation domain (AF-2). When a ligand interacts with PPAR, there is a binding to PPAR responsive elements (PPREs) in the DNA after dimerization with another nuclear receptor, the retinoid-X receptor (RXR). The release of corepressors and recruitment of coactivators regulate the transcription of specific genes (1).

In the last years, the development of new compounds that activate PPAR $\alpha$  has been an important target to better understand structure-activity relationships (SARs) for obtaining new hypolipidemic drugs with lower adverse events. A lot of PPAR $\alpha$ -selective agonists have been reported previously (11,12). The common structure is composed of an aromatic ring and carboxylic acid with or without various spacers. The aromatic ring of the ligands interacts with the hydrophobic binding pocket and the acidic moiety forms hydrogen bonds with amino acids on the AF-2 region in a roughly Y-shaped ligand binding site (13,14).

In the past years, we have synthesized different series of fibrate derivatives with the aim of obtaining new hypolipidemic compounds active as PPAR $\alpha$  agonists (15–17). In this field, stilbenes and chalcones and some of their synthetic derivatives have shown to activate PPAR $\alpha$  or to



Figure 1: Chemical modifications of clofibric acid.

lower plasma lipid levels (18,19). Therefore, our research group has previously synthesized chiral analogues of clofibrate with different biological activity related to stereo-chemistry (20,21).

The present work describes the synthesis of methylated and demethylated derivatives of clofibric acid (the active metabolite of clofibrate), where the aromatic centre is substituted by naphthalene, quinoline, chloroquinoline or anthracene scaffolds (Figure 1). The new compounds were *in vitro* evaluated for their PPAR agonistic activity using the reporter gene assay.

All compounds (Figure 2) were easily obtained in good yields by standard synthetic procedures. Esters **13–16** and



*rac*-**17–20** were obtained by  $S_N2$  reaction of phenols **9–12** with ethyl 2-bromo-2-methylpropanoate or ethyl 2-bromopropanoate, in the presence of sodium metal in absolute EtOH, under nitrogen atmosphere at reflux. The basic hydrolysis of **13–16** and *rac*-**17–20** with 1% alcoholic solution of KOH gave the acids **1**, **3**, **5**, **7** and *rac*-**2**, *rac*-**4**, *rac*-**6**, *rac*-**8** (Scheme 1). The phenols 1-naphthol (**9**), quinolin-2-ol (**10**) and 6-chloroquinolin-2-ol (**11**) are commercially available; conversely, 1-anthrol (**12**) was synthesized by the reaction of 1-anthrylamine in the presence of NaHSO<sub>3</sub> and EtOH at reflux (Scheme 2).

The clofibric acid chiral analogues (R)-2 and (S)-2 were synthesized by using (S)-4-isopropyl-2-oxazolidinone as chiral auxiliary because of its well-established capability of optical resolution of the  $\alpha$ -aryloxyacetic acids (22). The acid rac-2 was chlorinated with SOCI2 to obtain the acyl chloride rac-21 that was used to N-acylate the chiral auxiliary. The equimolar diastereomeric mixture of chiral imides 22 were obtained in good yield and separated on silica gel. The cleavage with lithium hydroperoxyde led to both dextrorotatory and levorotatory enantiomers of optical active acid 2 (Scheme 3). The enantiomeric purity of (R)-2 and (S)-2 was evaluated by a capillary electrophoretic run, by using  $\beta$ -cyclodextrin as a chiral selector. Analyses showed enantiomeric excesses up to 98% for the two tested compounds. As reported elsewhere, the compounds (R)-2 and (S)-2 are levorotatory and dextrorotatory



Scheme 1: Reagents and conditions: (a) ethyl 2-bromo-2-methylpropanoate or ethyl 2-bromopropanoate, Na, EtOH, N<sub>2</sub>, reflux; (b) KOH, EtOH, reflux.



Scheme 2: Reagents and conditions: (a) NaHSO<sub>3</sub>, H<sub>2</sub>O, EtOH, reflux.



Scheme 3: Reagents and conditions: (a) SOCI<sub>2</sub>, CH<sub>2</sub>CI<sub>2</sub>, 40 °C; (b) BuLi, (S)-4-isopropyl-2-oxazolidinone, THF, -78 °C, (c) LiOOH, THF/H<sub>2</sub>O, rt.

$Ar \xrightarrow{O} COOH R_1 R_2$							
Compound	Ar	R <sub>1</sub>	R <sub>2</sub>	clogP <sup>a</sup>	PPARα (E%) <sup>b</sup>	PPARγ (E%) <sup>b</sup>	PPAR $\delta$ (E%) <sup>b</sup>
<b>1</b> rac- <b>2</b> ( <i>R</i> )- <b>2</b> ( <i>S</i> )- <b>2</b>		CH <sub>3</sub> H H H	$\begin{array}{c} CH_3\\ CH_3\\ CH_3\\ CH_3\\ CH_3 \end{array}$	$\begin{array}{c} 3.269 \pm 0.262 \\ 2.929 \pm 0.255 \\ 2.929 \pm 0.255 \\ 2.929 \pm 0.255 \\ 2.929 \pm 0.255 \end{array}$	37 50 106 56	23 36 35 12	31 29 12 33
<b>3</b> rac- <b>4</b>	N	CH <sub>3</sub> H	CH <sub>3</sub> CH <sub>3</sub>	$\begin{array}{c} 2.458 \pm 0.305 \\ 2.108 \pm 0.293 \end{array}$	43 31	32 23	32 23
<b>5</b> rac- <b>6</b>	Cl N	CH <sub>3</sub> H	CH <sub>3</sub> CH <sub>3</sub>	$\begin{array}{c} 3.104 \pm 0.334 \\ 2.754 \pm 0.324 \end{array}$	37 50	22 35	11 22
7 rac- <b>8</b>		CH <sub>3</sub> H	CH <sub>3</sub> CH <sub>3</sub>	$\begin{array}{l} 4.499 \pm 0.263 \\ 4.150 \pm 0.256 \end{array}$	12 12	14 24	34 15
Clofibric acid	Cl	CH <sub>3</sub>	CH <sub>3</sub>	2.724 ± 0.272	100	-	_
Rosiglitazone L-165 041	-			_	_	100	_ 100

**Table 1:** In vitro transactivation activity of new clofibric acid analogues

<sup>a</sup>clogP were calculated by using ACD/Labs Extension for CS ChemDraw (version 5.0).

<sup>b</sup>Compounds were tested in at least three separate experiments at 150  $\mu$ M. Efficacy values were calculated as percentage of the maximum obtained fold induction with the clofibric acid for PPAR $\alpha$ , rosiglitazone for PPAR $\gamma$  and L-165 041 for PPAR $\delta$ .

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respectively (23). All compounds were purified by column chromatography on silica gel or by crystallization and fully characterized by IR, <sup>1</sup>H and <sup>13</sup>C NMR spectroscopy.

To investigate PPAR $\alpha$  agonistic potency and subtype selectivity, all compounds were evaluated for human PPAR $\alpha$ , PPAR $\gamma$  and PPAR $\delta$  functional activity by a cell-based transactivation assay in eukaryotic cells (24). In this method, we utilized firefly luciferase reporter gene that provides a good assay sensitivity, dynamic range when quantifying nuclear receptor activity and optimal correlation with *in vivo* activity. In this study, we used clofibric acid, rosiglitazone and L-165 041, as reference compounds for PPAR $\alpha$ , PPAR $\gamma$  and PPAR $\delta$  respectively; results are expressed as efficacy (E%) relative to positive control. These results and the clogP calculated with ACD/Labs Extension for CS ChemDraw (version 5.0) were reported in Table 1.

All naphthyl derivatives (1, rac-2, (R)-2 and (S)-2) showed an activation of PPAR $\alpha$  higher than anthracenyl analogues (7 and rac-8) but similar to guinolin and 6-chloroguinolin derivatives (3, rac-4, 5 and rac-6). Probably, the activity is influenced by the lipophilicity; indeed, these compounds have clogP values similar to clofibric acid (1 clogP =  $3.269 \pm 0.262$ ; rac-2, (R)-2 and (S)-2 clogP =  $2.929 \pm 0.255$ ; 3 clogP =  $2.458 \pm 0.305$ ; *rac*-4 clogP =  $2.108 \pm 0.293$ ; 5 clogP =  $3.104 \pm 0.334$ ; rac-6 clogP = 2.754  $\pm$  0.324; clofibric acid clogP = 2.724  $\pm$ 0.272). Furthermore, naphthyl, guinolin and 6-chloroguinolin analogues possess a less hindered aromatic scaffold that probably fit better the hydrophobic pocket of the PPARa receptor, whereas the anthracenyl scaffold of compounds 7 and rac-8 probably accesses in an alternative hydrophobic site with displacement of carboxylic head. When one of the methyls in  $\alpha$  to the carboxylic group was deleted, generating a chiral center (rac-2, rac-4, rac-6 and rac-8), it resulted in a different agonistic activity as compared to ramificated compound (1, 3, 5 and 7), showing that the  $\alpha$ -C to the carboxylic function plays an important role in the modulation of activity of the test compounds.

To better understand how the chirality influences the PPAR $\alpha$  activity, we have selected the acid rac-2 and we have synthesized its optically active derivatives (R)-2 and (S)-2; they showed a good activation of PPAR $\alpha$ , with efficacies of 106% and 56% respectively; (R)-2 and (S)-2, the more efficacious of our series of compounds, were also selected for the determination of EC<sub>50</sub> values and tested in at least three separate experiments at five concentrations ranging from 1 to 150  $\mu$ M. Both compounds, with  $EC_{50} = 6.9 \pm 0.6 \ \mu M$  for (R)-2 and  $EC_{50} = 11.6 \pm 0.9 \ \mu M$ for (S)-2 showed a better activation of the PPARa than clofibric acid (EC\_{50} = 55  $\pm$  3.9  $\mu\text{M}).$  On the basis these results, we hypothesize that compounds with (R)-stereochemistry fit very well into the receptor Y-shaped pocket; however, a rationalization of the results will require further investigations based on molecular modelling studies. At last, none of tested compounds was able to exhibit full agonist activity on PPAR  $\!\gamma$  and PPAR  $\!\delta,$  showing lower potency than rosiglitazone and L-165 041.

In conclusion, three series of clofibric acid analogues were synthesized and *in vitro* evaluated for human PPARs transactivation assay. Compounds with naphthyl, quinolin and 6-chloroquinolin scaffold showed a better increase in the transcriptional activity of receptor than the anthracenyl derivatives. The best results were obtained with compounds (*R*)-2 and (S)-2 that showed EC<sub>50</sub> values of 6.9 ± 0.6  $\mu$ M and 11.6 ± 0.9  $\mu$ M respectively.

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## **Conflict of interest**

The authors declare no conflict of interest.

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## **Supporting Information**

Additional Supporting Information may be found in the online version of this article:

Appendix S1. Materials and methods.