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Crystal structure and molecular docking studies of new pyrazole-4-carboxamides

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Abstract: Two pyrazol-4-carboxamides, 3-(difluoromethyl)-N-(mesitylcarbamoyl)-1-methyl-1*H*-pyrazole-4-carboxamide (**7a**) and 3-(difluoromethyl)-N-((3,5-dimethylphenyl) carbamoyl)-1-methyl-1*H*-pyrazole-4-carboxamide (**7b**) were synthesized and their structures were confirmed by the aid of ¹H NMR and HRMS analyses. The structure of the pyrazole-4-carboxamide, **7a** was also determined by X-ray diffraction. The preliminary activity results demonstrate that these two compounds exhibit good inhibitory activity against *Botrytis cinerea*. Further docking results indicated that the key active group is difluoromethyl pyrazole moiety.

Keywords: pyrazole; urea; synthesis; crystal structure; docking

Introduction

Heterocyclic compounds received important attention due to their wide range of biological activities [1-5]. Many pyrazole carboxamide compounds had been developed as commercial fungicides targeting succinate dehydrogenase inhibitors (SDHIs), including penthiopyrad, sedaxane, pydiflumetofen, bixafen, fluxapyroxad, isopyrazam and benzovindiflupyr. In addition, pyrazole derivatives displayed diversity activities, such as antimicrobial [6], DPPH radical scavenging [7], anticancer activity [8], antifungal [9, 10], anti-inflammatory [11], nematicidal [12-19]

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Li Qiao, Peng-Peng Cai, Zhong-Hua Shen, Hong-Ke Wu, Cheng-Xia Tan and Jian-Quan Weng, College of Chemical Engineering, Zhejiang University of Technology, Hangzhou, 310014, Zhejiang, China and cholinesterases inhibitory activity [20]. On the other hand, urea group is always important key building block in many drugs or pesticides. The compounds with urea group exhibited diversity bioactivities, such as mosquito [21], antimicrobial [22], antiviral [23], antifungal [24-26] and antinociceptive activity [27].

In view of these facts mentioned above, and also as a part of our work on the synthesis of bioactive lead compounds for drug discovery [28-46], two new pyrazole-4-carboxamides, 3-(difluoromethyl)-N-(mesitylcarbamoyl)-1-methyl-1H-pyrazole-4-carboxamide 3-(difluoromethyl)-N-((3,5-dimethylphenyl) (**7a**) and carbamoyl)-1-methyl-1H-pyrazole-4-carboxamide(7b)were designed and synthesized. The structures were characterized by the aid of ¹H NMR and HRMS analysis. The single crystal structure of compound 7a was determined by X-ray diffraction. The fungicidal activity of these compounds was tested and the docking studies were also carried out to study the mode of action.

Results and discussion

Synthesis

The synthetic route of 3-(difluoromethyl)-N-(mesitylcarbamoyl)-1-methyl-1*H*-pyrazole-4-carboxamide is outlined in Scheme 1. In this paper, ethyl-2-(ethoxymethylene)-4,4-difluoro-3-oxobutanoate and triethyl orthoformate took place smoothly in the presence of acetic anhydride resulted in the formation of intermediate 1[47]. Then the pyrazole ring was prepared by reacting intermediate 1 with methylhydrazine[48]. The pyrazole ester hydrosis under NaOH condition, then acid by HCl[49]. The pyrazole acyl chloride was given using the SOCl, as chlorinate reagent[50]. Then the pyrazole acyl chloride reacted with NH₂•H₂O to give pyrazole amide[50]. The key intermediate pyrazole isocyanate was prepared by pyrazole amide and triphosgene[50]. Finally, the key intermediate pyrazole isocyanate reacted with 2,4,6-trimethylaniline and

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Scheme 1 The synthetic route of title compounds

3,5-dimethylaniline at room temperature. The reaction condition is mild.

The structures of the pure compounds **7a** and **7b** were confirmed by ¹H NMR, and HR MS. From the ¹H NMR data of compound **7a**, the signals at 2.28 and 2.31 ppm were recognized as three methyl group of benzene ring. The appearance of signal at 3.92 ppm belongs to the methyl protons of pyrazole ring. The two NH protons of urea bridge are found at 10.04 and 10.86 ppm as single peak. The CHF₂ proton signals were triple peak with the coupling constant 54 Hz due to the influence of F atom. The high resolution mass spectroscopies of the two compounds are in agreement with their molecular formula $C_{16}H_{18}F_2N_4O_2$ (**7a**) and $C_{16}H_{16}F_2N_4O_2$ (**7b**).

Crystal structure

The crystal belongs to triclinic system with space group *P-1*. The molecular structure of compound **7a** is shown in Fig. **1**, and the packing diagram in Fig. **2**. The selected bond lengths and torsion angles are listed in Table **1**.

Fig.1 View of the title compound **7a**, with displacement ellipsoids drawn at the 30% probability level.

Generally, the average bond lengths and bond angles of pyrazole ring and phenyl ring were normal ranges. The N2-C4 [1.346(3) Å], N1-C2 [1.336(3) Å] bond were

longer than the general C=N double bond length of 1.27 Å, which indicated significant electron delocalization in the fused ring system. The torsion angle of C7-N3-C6-C3 and C8-N4-C7-N3 was -175.7(2)° and -179.4(2)° respectively, which indicated the two carbonyl groups are opposite. The angle between the pyrazole ring and benzene ring is 68.2° .

In the intermolecular edge-to-face π - π stacking pattern of the title compound, it is worth mention that the two molecules of each stacking unit are cetrosymmetric, which can be proved by the relative position of the phenyl rings (C8, C9, C10, C11, C12, C13) and methyl of pyrazole: the centroid separation of them is 3.131 Å. These interactions are estimated to play a role in stabilizing the crystal structure. The title compound 7a has an extensive network of hydrogen bonding. The parameters of intramolecular and intermolecular bonds are given in Table 2. From Fig. 1, the N(4)-H(4)...O(1) hydrogen bond formed a six member ring in the molecule. In the ac plane, they are linked together by N(3)-H(3)...O(2) hydrogen bonds. This hydrogen-bonding sequence is repeated to form a ring. The ring is shaped like a decagon and has two N(4) and two O(1) atoms at the vertices, leading to a hydrogen-bond network defining cyclic motifs denoted R_2^2 (8). The hydrogen bonds and weak π - π interactions strengthen the integration of the 3D networks.



Figure 1 View of the title compound 7a, with displacement ellipsoids drawn at the 30% probability level.



Figure 2 A view of pack molecule 7a.

Evaluation of fungicidal activity

Fungicidal activity of compounds **7a** and **7b** against *Fusarium oxysporum, Corynespora mazei, Pseudomonas syringae* and *Botrytis cinerea* was evaluated at 50 μ g/mL according to our previous work [9, 42], Fluxapyroxad was used as controls and the results are listed in Table 3. The primary bioassay showed the two compounds exhibits good inhibiting activity (60.00% and 77.27%) towards *Botrytis cinere,* which is the same as control (67.27%). For

the other three fungals *Fusarium oxysporum*, *Corynespora mazei*, *Pseudomonas syringae*, they exhibited weak activity with inhibitory -49.20% and -55.59%, 19.13% and 0.43%, -14.77% and 6.09% at 50μ g/mL, respectively.

Docking study

In order to study the action mode of high active compound and the target, the binding modes between SDH (PDB:2FBW) and the active compound **7a** were selected as exemplified in the case of representative compound by using the Discovery studio. The compound **7a** can tightly occupy binding site of SDH, the docking results are shown in Figure **3**. From the docking results, the compound **7a** held two weak interactions: π -cation and π -sigma. The π -cation bond was formed between the pyrazole ring and Arg 43 amino acid residue with the distance of 4.1821 Å. The other π -sigma bond was formed between the pyrazole ring and Ile 218 amino acid residue with the distance of 2.82714 Å. The results indicated that the key active group is difluoromethyl pyrazole moiety.

Experimental

Instruments

Melting points were determined by an X-4 apparatus and uncorrected. ¹H NMR spectra were measured on a Bruker AV-400 or 500 MHz instrument using TMS as an internal standard and CDCl_3 or $\text{DMSO-}d_6$ as the solvent. High

Table 1	Selected B	Bond length	s (A), Sel	ected Bon	d angles	(°) for
Compou	und 7a.					

Bond	Dist.	Angle	(°)
F(1)-C(1)	1.376(3)	C(2)-N(1)-N(2)	104.4(2)
F(2)-C(1)	1.363(3)	C(4)-N(2)-N(1)	112.6(2)
N(1)-N(2)	1.360(3)	C(4)-N(2)-C(5)	127.2(2)
O(1)-C(6)	1.232(3)	N(1)-N(2)-C(5)	120.2(2)
O(2)-C(7)	1.242(3)	C(6)-N(3)-C(7)	127.8(2)
N(1)-C(2)	1.336(3)	C(7)-N(4)-C(8)	121.7(2)
N(3)-C(7)	1.391(3)	F(2)-C(1)-F(1)	105.7(2)
N(2)-C(4)	1.346(3)	N(1)-C(2)-C(1)	120.0(2)
N(4)-C(7)	1.345(3)	C(3)-C(2)-C(1)	128.3(2)
C(1)-C(2)	1.494(4)	N(3)-C(6)-C(3)	115.2(2)
C(2)-C(3)	1.424(4)	N(4)-C(7)-N(3)	119.1(2)

Table 2	Hydrogen-bond	Parameters ((Å)	of Com	pound	7a
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D-HA	d(D-H)	d(HA)	d(DA)	∠(DHA)
N(3)-H(3)O(2)#	0.88	2.03	2.693(5)	130.8
N(4)-H(4)O(1)	0.88	1.98	2.830(6)	161.6

Symmetry transformations used to generate equivalent atoms: #-1-X,-Y,-Z

resolution mass spectra were recorded on an Agilent LC-Q-TOF-MS 6520 instrument. Crystallographic data of the compound were collected on a Bruker APEX-II CCD diffractometer. All the reagents are of analytical grade or freshly prepared before use.

General procedure

The intermediates **1**, **2**, **3**, **4**, **5** and **6** were synthesized according to our previous work [42]. A 10 mL round bottom flask was charged with 3-(difluoromethyl)-1-methyl-1*H*-py-razole-4-carbonyl isocyanate **6** (201 mg, 1 mmol) and 2,4,6-trimethylaniline (135 mg, 1 mmol) in CH₂Cl₂ (4 mL). The mixture was stirred at room temperature for overnight and the correspondence product was given. The target compounds were filtered and crude solids were recrystallized from ethanol to give the compound **7a.** White solid, yield 66.7%, 224.2 mg, m.p. 247~248 °C, ¹H NMR (500 MHz, CDCl₃) & 2.28 (s, 6H, 2CH₃), 2.31 (s, 3H, CH₃), 3.62 (s, 3H, CH₃), 6.85 (t, *J* = 54 Hz, 1H, CHF₂), 6.97 (m, 2H, Ar), 8.35 (s, 1H, CH), 10.04 (s, 1H, NH), 10.86 (s, 1H, NH); HRMS (ESI) for C₁₆H₁₈F,N₆O₂ *m/z*: Calculated, 337.1471, Found,



Figure 3 The docking mode of compound 7a and the SDH

Table 3 The fungicidal activity of compounds 7a and 7b against four fungus at 50 mg/L

No	R	Corynespora mazei	Pseudomonas syringae	Fusarium oxysporum	Botrytis cinerea
7a	2,4,6-3Me	-49.20	19.13	-14.77	60.00
7b	3,5-2Me	-55.59	0.43	6.09	77.27
Control	Fluxapyroxad	74.09	52.55	71.301	67.27
СК	water	0	0	0	0

337.1470 [M+H]⁺. The compound **7b** was synthesized according to this method. 3-(difluoromethyl)-*N*-((3,5-dimethyl-phenyl)carbamoyl)-1-methyl-1*H*-pyrazole-4-carboxamide **7b** :White solid, yield 53.5%, 172.3 mg, m.p. 207~208 °C, ¹H NMR (500 MHz, DMSO-*d*₆) & 2.35 (s, 6H, 2CH₃), 3.85 (s, 3H, CH₃), 6.85 (s, 1H, Ar), 7.14 (t, *J* = 54 Hz, 1H, CHF₂), 7.19 (s, 2H, Ar), 8.39 (s, 1H, CH), 9.89 (s, 1H, NH), 10.69 (s, 1H, NH); HRMS (ESI) for C₁₅H₁₆F₂N₄O₂ *m/z*: Calculated, 323.1314, Found, 323.1314 [M+H]⁺.

Structure determination

The cube-shaped single crystal of compound 7a was obtained by recrystallization from EtOH. The crystal with dimensions of 0.20mm × 0.18mm × 0.08mm was mounted on a Rigaku Saturn diffractometer with a graphite-monochromated Mo*K* α radiation (λ = 0.71073Å) by using a Phi scan mode at 110(2) K in the range of $3.646 \le \theta \le 55.758^\circ$. A total of 9797 reflections were collected, of which 3892 were independent ($R_{int} = 0.0813$) and 2075 were observed with $I > 2\sigma(I)$. The calculations were performed with SHELXS-97 program [51] and the empirical absorption corrections were applied to all intensity data. The nonhydrogen atoms were refined anisotropically. The hydrogen atoms were determined with theoretical calculations and refined isotropically. Crystal data for title compound: triclinic system, *P-1* space group with a = 8.426(9) Å, b =9.149(9) Å, c = 12.406(16) Å, $\alpha = 68.374(11)^{\circ}$, $\beta = 71.393(15)^{\circ}$, $y = 74.816(11)^{\circ}, V = 831.3(16) \text{ Å}^3, Z = 2, T = 110(2) \text{ K},$ μ (MoK α) = 0.106 mm⁻¹, *Dcalc* = 1.344 g/cm³, μ (MoKa) = 0.106 mm⁻¹, GOOF = 0.973. The final full-matrix leastsquares refinement gave R = 0.0654 and wR = 0.1429.

Docking studies

Molecular docking studies were done using Discovery Studio 2.5 software. The binding sites were generated from the SDH structure (PDB code: 2FBW) [12-19]. At first, water molecules were removed from the complex. Then, the crystallographic disorders and unfilled valence atoms were corrected using protein report and utility and clean protein options. Protein energy was minimized by applying CHARMM and MMFF94 force fields. The rigid structure of protein was obtained by applying fixed atom constraint. The protein binding site was defined and prepared for docking process. The structure of compound **7a** was drawn using Discovery Studio 2.5 and minimized by applying CHARMM force field. Then, the minimized structures were prepared for docking using prepare ligand protocol. Docking process was carried out using CDOCKER protocol. CDOCKER is a grid-based molecular docking method that employs CHARMM-based molecular dynamics (MD) scheme to dock ligands into a receptor binding site. The receptor was held rigid while the ligands were allowed to be flexible during the refinement. Each molecule was allowed to produce ten different interaction poses with the protein. Then, docking scores (-CDOCKER interaction energy) of the best-fitted poses with the active site at the SDH structure were recorded.

Conclusion

In summary, two new pyrazole-4-carboxamides, had been synthesized by multi-step reaction and characterized by ¹H NMR, HRMS and single-crystal X-ray structure determination. The results show that the crystal structure exhibits intermolecular and intramolecular hydrogen bonds. The fungicidal activity results showed that it possessed moderate activity against *Botrytis cinerea*. The docking results indicated the key active group is difluoromethyl pyrazole moiety.

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