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Tweaking Subtype Selectivity and Agonist Efficacy at (S)-2-Amino-3-(3-hydroxy-5-methyl-isoxazol-4-yl)propionic acid (AMPA) Receptors in a Small Series of BnTetAMPA Analogues

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(5) Supporting Information



ABSTRACT: A series of analogues of the (S)-2-Amino-3-(3-hydroxy-5-methyl-isoxazol-4-yl)propionic acid (AMPA) receptor agonist BnTetAMPA (**5b**) were synthesized and characterized pharmacologically in radioligand binding assays at native and cloned AMPA receptors and functionally by two-electrode voltage clamp electrophysiology at the four homomeric AMPA receptors expressed in *Xenopus laevis* oocytes. The analogues **6** and **7** exhibit very different pharmacological profiles with binding affinity preference for the subtypes GluA1 and GluA3, respectively. X-ray crystal structures of three ligands (**6**, **7**, and **8**) in complex with the agonist binding domain (ABD) of GluA2 show that they induce full domain closure despite their low agonist efficacies. Trp767 in GluA2 ABD could be an important determinant for partial agonism of this compound series at AMPA receptors, since agonist efficacy also correlated with the location of the Trp767 side chain.

INTRODUCTION

(S)-Glutamic acid (glutamate, Glu, 1, Chart 1) is the major excitatory neurotransmitter in the brain and has a central role for the normal functions of the brain. On the other hand, the neurotransmitter is also involved in a number of central nervous system disorders.¹ The physiological and pathophysiological effects of Glu are mediated by a highly heterogeneous receptor population comprising G-protein coupled metabotropic Glu receptors and ionotropic Glu receptors (iGluRs).¹ The iGluRs are Glu-gated cation channels that can be further divided into three major classes according to their respective activation by the agonists N-methyl-D-aspartic acid (NMDA), (S)-2-amino-3-(3-hydroxy-5-methyl-isoxazol-4-yl)propionic acid ((S)-AMPA, 2), and kainic acid (kainate, KA).^{2,3} The iGluRs are homo- or heteromeric assemblies of various subunits, NMDA receptors being made up of the GluN1, GluN2A-D, or GluN3A,B subunits, AMPA receptors of the

GluA1–4 subunits, and KA receptors of the GluK1–5 subunits. Recent X-ray crystallographic studies of full-length AMPA and NMDA receptors have confirmed that the iGluR is a tetrameric assembly^{4–8} consisting of three structural domains, the transmembrane domain forming the ion channel, an agonist binding domain (ABD), and a large N-terminal domain. Prior to these structures, a large number of structures of soluble constructs of the iGluR ABDs had disclosed important information on the molecular basis for orthosteric ligand recognition, and the mechanisms underlying activation, desensitization, and allosteric modulation of the receptors.⁹ The ABD is a clamshell-like structure with two lobes closing around the agonist upon receptor activation, and the degree of receptor activation has been proposed to be closely linked to

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Chart 1. Chemical Structures of Glutamic Acid and Some Selective iGluR Agonists 1–5a,b and the New Compounds 6–11



the degree of domain closure around the orthosteric ligand.^{10,11} The details on how domain closure translates into channel opening are unknown, but H-D exchange NMR studies of various full and partial agonists have provided details on the dynamics of domain closure that are not observed in the X-ray crystal structures.^{12–14} The studies showed that domain closure induces a number of H-bonds between the upper and lower lobes in the clamshell that are weaker for partial agonists than for full agonists and that efficacy can be correlated with the stability of these hydrogen bonds. In particular, large differences in the NMR signal from the indole N-H of Trp767 was observed for a series of substituted willardiines.¹² It is currently unknown whether the ABD X-ray structures reflect a partially activated state of the receptor or rather are slightly stabilized intermediate states in a range of different conformations in a more dynamic system. More recently, full-length X-ray crystal

structures containing partial agonists have been achieved giving more insight on the activation mechanisms^{7,8} but with little information on the dynamics, and as yet structures of the open channel state of GluA2 have not been reported.

The medicinal chemistry efforts in the iGluR field have resulted in the development of several agonists with selective activity at the receptors comprised of one of the three major iGluR classes, that is, the NMDA, AMPA, or KA receptors.¹ In contrast, agonists (and antagonists) discriminating between different subtypes within these three classes are very few and far between due to the high degree of sequence identity for the orthosteric sites of the respective subunits within each class. In the AMPA receptor field, the compounds 3 and 4 have been found to be selective agonists of GluA1 and GluA2 over the GluA3 and GluA4 subtypes.^{15,16} A single residue in the orthosteric site of the AMPA receptors has been found to be the key molecular determinant: a tyrosine in the GluA1/2 ABD that is a phenylalanine in GluA3/4. In another series of analogues, the potent but completely nonselective AMPA receptor agonist 2-amino-3-[3-hydroxy-5-(2-methyl-2H-5-tetrazolyl)-4-isoxazolyl]propionic acid (MeTetAMPA, 5a) was employed as lead structure in the development of 2-amino-3-[3-hydroxy-5-(2-benzyl-2H-5-tetrazolyl)-4-isoxazolyl]propionic acid (BnTetAMPA, **5b**),^{17,18} which selectively activates GluA2, GluA3, and GluA4 over GluA1. This subtype preference was shown to not arise from the above-mentioned tyrosine/ phenylalanine difference in the Glu binding site of the receptors but rather to be rooted in interactions of the benzyl group of BnTetAMPA with residues in a less conserved region in the proximity of the Glu binding site.¹⁹ Trauner and co-workers recently developed 5b into a photoswitchable AMPA receptor agonist, but the subtype selectivity profile of this compound was not established.^{20,21} Furthermore, an X-ray crystal structure of 5b in the GluA2 ABD demonstrated that 5b induces a high degree of domain closure (21.5°) but is nevertheless a partial agonist with an efficacy of 0.83 compared with Glu.¹⁸ There is considerable potential in subtype-selective AMPA receptor ligands as pharmacological tools for the delineation of the physiological functions governed by the respective subtypes. The interesting pharmacological properties of 5b and the apparent importance of the side pocket adjacent to the Glu binding site targeted by the benzyl group of 5b for subtype selectivity spurred our interest in this compound as a lead for



Figure 1. Sequence differences in AMPA receptor agonist binding pockets emphasizing important residues for the pharmacological profile of BnTetAMPA (5b). GluA1 residues are displayed in salmon and GluA2 residues in gray. Compound 5b is shown in stick representation and GluA2 ABD as cartoon (PDB code 2P2A, molB).

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further medicinal chemistry development. In the present study, we report the design, synthesis, and characterization of a series of BnTetAMPA analogues comprising various modifications to the benzyl group.

RESULTS

Design. In the design of the new compounds, we studied the side pocket to the Glu binding site targeted by the benzyl group of **5b** according to the GluA2 ABD-**5b** crystal structure. A mutation study¹⁹ has suggested that the altered selectivity profile of 5b compared with 5a arises from the benzyl group in 5b protruding into a less conserved region in the AMPA receptor ABD. Four residues in this region are different in GluA1 compared with GluA2-4, and three of these residues were identified as determinants of the subtype selectivity profile of 5b. Two of the residues are negatively charged acidic residues in GluA1 (Asp399 and Glu683), whereas the corresponding residues in GluA2-4 are neutral residues (Figure 1). We therefore speculated that an amino group protruding from the meta position (6) of the benzyl group toward Asp399 and Glu683 would reverse the observed selectivity of 5b. Furthermore, the opposing meta position was close to a lipophilic area in the binding pocket that potentially could accommodate a chlorine (7) or methyl (8)group. Along with these analogues, we decided to probe the region further with more variation by introducing $CF_3(9)$ and methylsulfonyl (10) groups and by exchanging the benzyl group with a 4-pyridinemethyl group (11). Finally, N-amino alkylated analogues of 6 (compounds 6a-c) were also included in the SAR study.

Synthesis. The synthesis of the analogues followed the original procedure,¹⁸ but employing the optimization by Trauner et al.²⁰ that uses an O-allyl protecting group. As shown in Scheme 1, it starts with the tetrazole 12 as a common intermediate for alkylation with 3-substituted benzyl bromides (for 6-10) or 4-(bromomethyl)pyridine (for 11) in the presence of potassium carbonate in acetone at room temperature. This gave N-2 substituted tetrazoles (13-19) as the major regioisomer (the ratio of N-1 and N-2 regioisomers was varying around 1:4) in high yield. The N-1 alkylated regioisomer could not be separated from the N-2 alkylated regioisomer at this stage. Various conditions were tested for deprotection of the alkylated tetrazoles, and hydrogen chloride was found to enable removal of all three protecting groups in one step without affecting the N-benzyl group. Subsequent purification by reverse phase HPLC gave the final compound as a mixture of the 1- and 2-isomers, which were assigned based on a comparison of NMR signals in accordance with previously reported values. In the case of 6 and 8, we were able to obtain small amounts of the 1-isomer ((N-1)-6 and (N-1)-8).

Pharmacology. Initially, all compounds were screened in HEK293 cell lines stably expressing the rat GluA1, GluA2, GluA3, and GluA4 receptors in the Fluo-4/Ca²⁺ assay (see Supporting Information Table S1). In this assay, compound **6** displayed significant agonist activity at GluA1 at concentrations of 30 μ M and higher (a complete concentration–response curve could not be generated). In contrast, **6** exhibited no significant activity at the other three subtypes at concentrations up to 1 mM. This interesting profile prompted the synthesis of the *N*-alkylated analogues (**6a**–**c**) that also exhibited agonist activity in the GluA1-HEK293 cell line in the Fluo-4/Ca²⁺ assay, although the three analogues were observed to be significantly weaker agonists at GluA1 than **6**. Furthermore, **7**



"Reagents and conditions: (a) $BrCH_2Ar$, K_2CO_3 , acetone, rt; (b) from 17, N-methyl-2-nitrobenzenesulfonamide, K_2CO_3 , acetone, rt; (c) PhSH, K_2CO_3 , DMF; (d) HCl (37% aq), 90 °C.

displayed agonist activity for all four subtypes, whereas 8 displayed weak antagonism of the Glu-induced response for all of the receptors. All other analogues did not display measurable agonist or antagonist activity at any of the four receptors.

Compounds **6–9** and **11** were selected for characterization of their AMPA, KA, and NMDA receptor binding affinity in a radioligand displacement assay using membranes prepared from rat cortical brain tissue (Table 1). Only **6**, 7, and **8** showed weak affinity in the AMPA assay compared with compounds **2– 5**. The observed affinity for the 1-isomer of **8** ((*N*-1)-**8**) is due to the presence of 3.5% of **8** as determined from NMR and HPLC.

Next we determined the binding affinities of selected compounds to cloned rat AMPA receptor subtypes expressed in sf9 insect cells (Table 2). Here, the intention with compound **6** to achieve a ligand with inverted selectivity profile compared with **5b** was found to be successful. The affinity of **6** in this assay increased 20-fold at GluA1 compared with **5b**, whereas affinities at other subtypes decreased 4–10-fold. The *N*-alkylated analogues **6a**–**c** also showed altered selectivity profile although with less pronounced GluA1 preference.

In contrast, the chloro-substituted compound 7 displayed increased binding affinity to all four subtypes compared with **5b**. This was most pronounced at GluA3 where 7 exhibited a 50-fold higher binding affinity than **5b**, thus having a slightly higher affinity to this receptor than AMPA (2). At the other three subtypes, the binding affinities of 7 show 3-10-fold increase compared with **5b**. Thus, 7 displayed a significant

Table 1. Binding Affinities of Compounds at Native Rat Brain iGluRs

		iGlu		
	compd	AMPA-R	KA-R	NMDA-R
	(RS)-2 ^a	0.040	>100	>100
	3 ^{<i>a</i>}	0.22	>100	18
	4 ^b	0.37	>100	>100
	(S)-5a ^c	0.009	11	>100
	$5b^d$	2.7	>100	87
	6	23.1 ± 3.5	>100	>100
	6a	>100	>100	>100
	6b	>100	>100	>100
	6c	>100	>100	>100
	6 (N-1)	>1000	>100	>100
	7	0.80 ± 0.09	>100	>100
	8	2.30 ± 0.75	>100	>100
	8 $(N-1)^{e}$	58 ± 6.3	g	g
	9	>100	>100	g
	11	>100	>100	>100
an	c	ha c ca c	- d-	

^{*a*}Reference 15. ^{*b*}Reference 16. ^{*c*}Reference 17. ^{*d*}Reference 18. ^{*e*}Contains 3.5% of the 2-isomer of **8**. ^{*f*}Shown are means \pm SEM from three to four individual experiments. ^{*g*}Not determined.

preference in binding for GluA3 and GluA4 over GluA1 and GluA2 (GluA3 \approx GluA4 > GluA2 \approx GluA1). The methyl analogue 8 also displayed increased affinity compared with 5b at all four subtypes and a similar selectivity profile as 7. All analogues showed little or no affinity at KA receptor subtypes GluK1-3, which is similar to 5b. Compound 9 showed little or no affinity at all subtypes tested, whereas 11 was a GluK1 selective ligand. Compound 7 was also characterized on a mutant of GluA1_o where Tyr716 is exchanged for a phenylalanine that is present in the same position in GluA3 and GluA4, resulting in an affinity ($K_i = 4.76 \ \mu M$) that was not statistically significantly different from that at wild-type GluA1 (*t* test, p = 0.056). This observation indicates that this tyrosine/ phenylalanine residue is not responsible for increased activity of 7 at GluA3 and GluA4 compared with GluA1 and GluA2 as has been seen with other ligands.²²

Next we determined the functional properties of the most potent compounds (6-8) by TEVC electrophysiology at recombinant rat AMPA receptor subtypes expressed in Xenopus laevis oocytes (Table 3). In this assay, compound 6 displayed the highest potency (EC₅₀ = 19.5 μ M) at GluA1_i compared with the other subtypes, although differences were less pronounced compared with the binding affinities (Figure 2A, Table 3). There was no statistically significant difference in the potency of 6 between GluA1, and GluA1_o (p > 0.05, Kruskal-Wallis one-way ANOVA on Ranks), indicating that the flip/flop region of the receptor is not a determining factor for its subtype selectivity. Compounds 7 and 8 displayed some preference for GluA3; vs GluA1; (p < 0.05, Kruskal-Wallis one-way ANOVAon Ranks with Dunn's post-test). Compounds 6-8 showed low efficacy at the cloned AMPA receptor subtypes, indicating that they are partial agonists (Table 3). Compound 7 was a particularly low efficacy agonist at GluA3_i (Figure 2B), suggesting that it might be possible to convert it into a subtype-selective antagonist with some structural modifications.

In order to test our hypothesis that introduction of an electrostatic interaction between the positively charged benzylamino group of 6 and the acidic groups Asp399 and Glu683 in GluA1 was responsible for the inversed selectivity, the compound was characterized pharmacologically at three mutant receptors. We therefore introduced the two acidic residues of GluA1, into in the corresponding positions of $GluA2(Q)_i$ (Ser403Asp and Ala687Glu). The potencies of Glu at wild-type $GluA2(Q)_i$ and the mutants were not statistically significantly different (p > 0.05, Kruskal–Wallis one-way ANOVA on Ranks with Dunn's post-test). Furthermore, the potencies of **6** at wild-type $GluA2(Q)_i$ and the mutants were not statistically significantly different (p > 0.05, Kruskal–Wallis one-way ANOVA on Ranks with Dunn's post-test) (Figure 2). However, a trend toward increased relative potency of 6 compared with Glu at GluA2 with introduction of acidic residues can be observed.

Structure. The interesting activity profiles of compounds 6-8 spurred us to cocrystallize the compounds with the soluble GluA2 ABD to further understand the binding modes as well as basis of the selectivity of the compounds and their agonist efficacies. The X-ray crystal structures reveal that all three

Table 2.	Binding	Affinities	of Com	pounds at	t Cloned	Rat iGluR	Subtype	es Ex	pressed i	n sf	9 Cells	а

	$K_{ m i}~(\mu{ m M})$							
compd	GluA1 _o	$\operatorname{GluA2}(R)_{o}$	GluA3 _o	GluA4 _o	GluK1	GluK2	GluK3	
2 ^b	0.022	0.017 ^c	0.021	0.040	1.15	>100	42.7	
3 ^b	0.13	0.37	27.5	11.9	6.8	>100	90.5	
4 ^{<i>c</i>}	0.180	0.348	4.52	13.5	4.48	>100	29.8	
(S)-5a ^d	0.0052	0.0039	0.0022	0.0027	0.081	5.5	>100	
5b ^e	7.70	0.75	0.39	0.20	58	39	>100	
6	0.434 ± 0.129	8.15 ± 0.92	1.53 ± 0.38	2.75 ± 0.76	16.7 ± 4.8	>100	>100	
6a	10.8 ± 3.7	84.4 ± 27.6	8.47 ± 1.21	h	>100	>100	>100	
6b	5.29 ± 1.30	84.5 ± 28.3	22.3 ± 2.8	h	>100	>100	>100	
6c	3.2^{f}	55 ^f	16 ^f	h	>100	>100	>100	
7^g	0.864 ± 0.060	0.213 ± 0.035	0.00786 ± 0.00126	0.0210 ± 0.0089	1.27 ± 0.21	>100	>100	
8	4.61 ± 0.96	1.27 ± 0.35	0.147 ± 0.005	0.522 ± 0.045	6.36 ± 1.34	>100	>100	
9	>100	69.1 ± 21.6	4.22 ± 0.96	h	h	>100	>100	
11	>100	>100	19.1 ± 4.6	55.6 ± 4.9	1.05 ± 0.37	>100	>100	

^{*a*}Recombinant rat AMPA-R in *sf9* cell membranes. Shown are means \pm SEM from $n \ge 3$ experiments conducted in triplicate at 12–16 ligand concentrations. Hill coefficients were close to unity. ^{*b*}Reference 15. ^{*c*}Reference 16. ^{*d*}Reference 17. ^{*e*}Reference 18. ^{*f*}From one experiment only. ^{*g*}K_i at (Y716F)GluA1₀ is 4.76 \pm 1.46 μ M. ^{*h*}Not determined

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Table 3. TEVC Functional Characterization of C	Compounds 6–8"
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		6		7		8	
	Glu EC ₅₀ (μ M)	EC ₅₀ (µM)	efficacy ^b	EC ₅₀ (µM)	efficacy ^b	EC ₅₀ (µM)	efficacy ^b
GluA1 _i	12.5 ± 1.4	19.5 ± 5.0	0.729 ± 0.019	25.7 ± 9.1	0.207 ± 0.012	61.5 ± 8.3	С
GluA1 _o	3.4 ± 0.6	25.1 ± 5.9	С	С	С	с	С
$GluA2(Q)_i$	14.7 ± 1.4	432 ± 33	с	3.66 ± 0.33	0.461 ± 0.020	32.7 ± 2.7	0.346 ± 0.049
GluA3 _i	21.2 ± 1.4	49.7 ± 11.8	с	1.32 ± 0.06	0.0526 ± 0.0105	18.8 ± 2.7	С
GluA4 _i	19.6 ± 2.7	129 ± 34	С	4.71 ± 0.58	0.244 ± 0.031	с	С
$(S403D)GluA2(Q)_i$	13.9 ± 5.4	287 ± 21	с	С	с	с	С
$(A687E)GluA2(Q)_i$	35.3 ± 4.6	433 ± 40	с	С	с	с	С
(S403D,A687E)GluA2(Q) _i	26.9 ± 4.7	212 ± 27	С	С	С	С	с

^aShown are mean \pm SEM ($n \ge 3$ eggs). ^bEfficacy measurements are performed at saturating concentrations of compound in the presence of 100 μ M cyclothiazide. Under these conditions, 1 mM (S)-Glu is defined to produce an efficacy = 1.000. ^cNot determined.



Figure 2. TEVC responses of 6 and 7. Shown are means \pm SEM of normalized, pooled data from 5–9 oocytes. (A) Concentration–response curves of 6 at GluA1_i and wild-type and mutant GluA2(Q)_i. Inset shows traces from one GluA1_i expressing oocyte ($V_h = -70 \text{ mV}$), conducted in duplicate stimulations at the indicated concentrations (μ M) of 6. Scale bars = 50 nA and 5 min. (B) Concentration–response curves of 7 at GluA1_i-4_i scaled to indicate the relative efficacies. Inset shows traces from one GluA1_i expressing oocyte ($V_h = -50 \text{ mV}$), conducted in duplicate stimulations at the indicated concentrations (μ M) of 7. Scale bars = 50 nA and 5 min.

compounds bind at GluA2 in a manner very similar to **5b** (see Supporting Information, Figure S1), but with notable differences (Figures 3 and 4).

To accommodate the chlorine in the meta-position of 7, a displacement of the benzyl group occurs compared with **5b** (Figure 4A), resulting in a lobe 2 movement relative to lobe 1 away from the ligands. A two-dimensional order parameter (ξ_1 , ξ_2) has been introduced to describe the large-scale conforma-



Figure 3. Overlay of X-ray crystal structures of GluA2 ABD with 5b, 6, 7, and 8. The structures are shown in ribbon representation and overlaid on lobe 1 residues (top). Differences in lobe 2 (lower) movement relative to lobe 1 are seen. Sb, gray (PDB code 2P2A, molB), 6, light blue, 7, orange, and 8, light purple.

tional transitions of the GluA2 ABD.²³ The lobe 2 movement induced by 7 is reflected in a larger ξ^2 (defined as the distance between the center of masses of residues 401–403 in lobe 1 and 686–687 in lobe 2), being 8.9 ± 0.2 Å (A 9.1 Å, B 8.9 Å, C 8.7 Å, D 8.9 Å) and 8.3 Å in the structure with 7 and **5b**, respectively (Figure 5). Negligible differences are seen in ξ^1 (9.4–9.6 Å; defined as the distance between the center of masses of residues 479–481 in lobe 1 and residues 654–655 in lobe 2). The chlorine atom is accommodated in a cavity created by Tyr405, Thr707, Tyr711, and Trp767 (Figure 4A). Likewise, the benzyl group of **8** is oriented with the methyl group protruding into the same cavity (Figure 4A).

For compound 6, the benzyl group containing a *meta*aminomethyl group is flipped in comparison to 5b, 7, and 8 in order to enable the amino group to make a salt bridge with Glu402 and a charge-assisted hydrogen bond with Ser403 as



Figure 4. Zoom on ligand binding site of the GluA2 ABD. (A) Comparison of binding mode of 5b, 7, and 8. Residues forming the cavity accommodating the chlorine and methyl group are shown in stick representation. (B) Hydrogen bonding interactions of the aminomethyl group of 6 with GluA2 residues as well as one water molecule and a putative ethylene glycol. The benzyl ring is flipped compared with that in 7. (C) The conformation of Met708 undergoes induced fit to accommodate the ligands. Color coding as in Figure 3 has been used.

well as contacts to a putative ethylene glycol and a water molecule (Figure 4B). Thereby, the aminomethyl group is interfering with the important interlobe contact between Glu402 and Thr686.²⁴

Finally, the side chain conformation of Met708 is affected by binding of the ligands. Met708 has previously been shown to undergo induced fit upon binding of ligands.²⁵ Whereas Met708 in the structure of **5b** moves away from the binding site compared with other agonist structures like that of **5a**, Met708 is pushed even further in the structures of **6**, 7, and **8** (Figure 4C).

DISCUSSION AND CONCLUSIONS

In the present study, we present the design, synthesis, and characterization of a series of analogues of the AMPA receptor agonist BnTetAMPA 5b where variation in the benzyl moiety is introduced. Compound 5b was developed from the most potent AMPA receptor agonist reported, MeTetAMPA 5a, and displays a quite unusual activity profile with lower affinity and agonist potency at GluA1 compared with GluA2-4.17,18 Previously, a few agonists had been developed that discriminated GluA1/GluA2 from GluA3/GluA4 due to the only nonconserved residue (Tyr/Phe) in the Glu binding site of the AMPA receptors, 15,22,26 but the selectivity profile of **5b** (Table 2) could not be explained by this difference. However, the GluA2 ABD crystal structure with 5b¹⁸ and a modeling study suggested that the benzyl group of 5b induced movement of Met708 in GluA2 into a nonconserved region of the receptor, where several residues were unique for GluA1.¹⁹ A subsequent mutation study substantiated this hypothesis, since the activity of 5b at GluA2 could be established in the GluA1 triple mutant Asp399Ser/Met686Val/Ile687Ala.¹⁹ This part of the receptor is generally less conserved in the AMPA receptor subtypes compared with the Glu binding pocket, and this might enable the development of ligands with preference for a single AMPA receptor subtype. We speculate that further variation in moieties of AMPA receptor agonists protruding into this part of the receptor could change the pharmacological profile at the AMPA receptor subtypes and perhaps lead to compounds with preference for individual subtypes.

Introduction of an amino methyl group on the benzyl group (6) potentially might favorably interact with two acidic residues (Asp399 and Glu683) in GluA1. Indeed, the affinity at GluA1 increases 18-fold compared with 5b whereas its affinity at GluA2-4 is decreased 4-14-fold compared with 5b. Thus, 6 exhibits 3.5-, 6.3-, and 19-fold higher binding affinity at GluA1 than at GluA3, GluA4, and GluA2, respectively. A similar selectivity profile is observed in the TEVC recordings, where 6 displays 2.5-, 6.6-, and 22-fold higher agonist potency at GluA1 than at GluA3, GluA4, and GluA2. This makes 6 the first agonist capable of discriminating between GluA1 and GluA2 both in binding and in functional assays. We therefore probed whether an increased activity of 6 would be observed at GluA2 mutants containing the two acidic amino acids of GluA1 (Asp399 and Glu683 in GluA1; Ser403Asp and Ala687Glu in GluA2). However, there was not any significant change in the potency of 6 at these mutants (nor at the double mutant)



Figure 5. Agonist efficacy of glutamate, 5b, 7, and 8, correlation with the distance in Å (blue) between the indole-N of Trp767 and backbone carbonyl oxygen of Thr707 in GluA2, and correlation with distance between the center of masses of residues 401-403 in lobe 1 and 686-687 in lobe 2 (ξ_2 , red).

compared with wild-type GluA2, and probably other residues are involved in the observed selectivity of 6.

In order to increase the general affinity of the ligands, we introduced chlorine or a methyl group in the meta position of the benzyl group of 5b that would possibly point into a lipophilic cavity in GluA2. The binding affinity profile of the methyl substituted 8 was very similar to 5b, but introduction of chlorine yielding 7 led to a 50-fold increase in affinity at GluA3 and only 3-10-fold increases at the other three subtypes. The racemate of 7 is more potent than (RS)-AMPA at GluA3 and close to equipotent with the most potent AMPA receptor agonist 5a assuming that the R-form and N-1 isomer of 7 is inactive. This assumption is supported by the fact that the N-1 regioisomers of 5a, 17,19 5b, 18 6, and 8 did not show any significant binding affinity and that the agonist binding domain crystallized from the mixture only with the N-1 isomer with (S)-configuration in this compound class. We speculated whether the weak affinity of 7 at GluA1 was partly due to Tyr716 in the Glu binding pocket, which corresponds to Phe728 in GluA3 and has been shown to be the determinant for the affinity difference between GluA1/2 and GluA3/4 of 3 and 4. However, the affinity of 7 at (Tyr716Phe)GluA1, was unchanged from wild-type, indicating that the selectivity likely comes from the region around the benzyl group and not the bioisosteric isoxazolyl group.

The X-ray structures of the GluA2 ABD with 6, 7, and 8 show that the clamshell-like structure is tightly closed around the ligand. It is notable how much the conformation of Trp767 is affected by the chlorine and methyl substituent that seem to push lobe 2 and thus Trp767 away. The degree of receptor activation has been linked to the degree of domain closure based upon the notion that partial agonists induce partial domain closure and that there is a correlation between the degree of domain closure and agonist efficacy as seen in a series of halogen substituted willardiines with varying agonist efficacy.^{10,11} This would suggest that the partial domain closure partially shifts the equilibrium between the closed and open channel compared with full agonists. However, an NMR study¹² of the same willardiines showed that a range of dynamic changes was observed that are not visible in the X-ray crystal structures. In particular, the indole N-H signal from Trp767 showed large variation in the NMR study. The rate of HD exchange for the indole N-H of Trp767 is much more rapid when full agonists are bound than when partial agonists are bound. A correlation between large-scale conformational transitions of the GluA2 ABD expressed by the parameter ξ_2 (see Results for definition) and the agonist can be seen (Figure 5). However, it is also interesting to note in the X-ray crystal structures of 5b, 7, and 8 that the interaction between the indole nitrogen of Trp767 and the carbonyl oxygen of Thr707 shows variation. Thus, a linear correlation can also be seen between the agonist efficacy and the distance between the indole nitrogen in Trp767 and backbone carbonyl oxygen of Thr707 in the structures of Glu, 5b, 7, and 8 (Figure 5). Thus, Trp767 might be an important determinant for the agonist efficacy of 5b analogues. Trp767 is part of helix K that has a disulfide bridge at Cys773-Cys718, so movements in this part can be coupled to other regions. Recent full-length structures with partial agonists^{7,8} have provided more insight into the domain movements that might occur upon receptor activation; however, the channel is closed and does not represent a conducting state conformation. One study⁷ suggests Ile633 in GluA2 to be an important switch by forming interactions with

Ile504, Leu639, Ile645, and Val723. The position of Val723 is close to the disulfide bridge, and it is therefore easy to imagine that changes in the dynamics of helix K could translate to this region. Furthermore, helix K is also linked directly to TM4 and thus could affect ion channel function.

In conclusion, this study shows that the pharmacological activity of ligands at AMPA receptors responds to small changes in the region around the benzyl group of 5b and that various AMPA receptor subtypes respond differently to these changes. We find that subtle substitutions to the benzyl of 5b dramatically change the selectivity profile of the compounds and present two novel ligands with interesting selectivity profiles. Compound 6 is the first published agonist capable of discriminating between GluA1 and GluA2 with preference for GluA1, and compound 7 exhibits pronounced selectivity for GluA3 and GluA4 in binding. Changes in both potency and agonist efficacy are observed, and X-ray crystal structures suggest that the agonist efficacy is determined by the bulk introduced at Trp767, whereas residues surrounding the substituted benzyl group determine selectivity. In this sense, the compounds can be important tools for understanding receptor activation mechanisms. Examination of the residues in vicinity of the benzyl group shows that several of them are nonconserved residues, and these differences might be exploited even further. We believe that the biostructural information presented is important for future design of subtype selective AMPA receptor ligands with control of the intrinsic activity. It is currently unknown whether the TetAMPA scaffold is unique or other Glu analogues carrying substituents that can interact with the region around the benzyl group can show similar variation in the pharmacological profile over the AMPA scaffold. Nevertheless, we have clearly demonstrated that the pharmacological profile is highly influenced by subtle changes in this region.

EXPERIMENTAL SECTION

Chemistry. General Procedures. All reactions requiring inert conditions were performed in flame-dried glassware under an inert atmosphere of N2 using standard Schlenk and syringe-septum technique. Petroleum ether (PE, 40-65 °C) was obtained in technical grade and used without further purification. Tetrahydrofuran (THF) was dried using a glass contour solvent system (SG Water, USA LCC). All other solvents for reaction, extraction, and purification were purchased in HPLC grade and used as received. Normal phase column chromatography was performed on silica gel 60 (Merck, 0.015-0.040 mm). Reactions and chromatography fractions were monitored by analytical thin layer chromatography (TLC), which was carried out using silica gel 60 F₂₅₄ aluminum plates (Merck), and the compounds were visualized by using UV light (254 or 365 nm or both) or by spraying with diluted solutions of KMnO₄ or ninhydrin. *n*-Butyllithium (n-BuLi) was titrated using N-pivaloyl-o-toluidine prior to use. Intermediate tert-butyl 3-(3-(allyloxy)-5-(2H-tetrazol-5-yl)isoxazol-4yl)-2-(bis(tert-butoxycarbonyl)amino)propanoate (S5) was synthesized according to the published procedures, and the analytical data matched those reported previously.²¹ All other starting materials as well as the reagents were obtained from commercial sources and used without further purification. Proton (¹H) and carbon (¹³C) NMR spectra were recorded at 300 K in the solvent indicated. Bruker 400 and Bruker 600 spectrometers were operated at 400 and 600 MHz for proton and at 100 and 150 MHz for carbon nuclei. Chemical shifts (δ) are given in parts per million (ppm) using signals of residual nondeuterated solvent as internal standards. The following abbreviations are used for the proton spectra multiplicities: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; br, broad. Coupling constants (J) are given in hertz. For regioisomeric mixtures, fractions of protons are reported for integrals to illustrate the ratio of isomers.

LC/MS was performed on an Agilent 1200 system with a Zorbax Eclipse XBD-C18 column (4.6 mm \times 50 mm) and UV detector using a linear gradient of the binary solvent system of H₂O/MeCN/formic acid (A = 95/5/0.1 and B = 5/95/0.1) with a flow rate of 1 mL/min (0–90% B over 4 min, then 0.5 min 100% B) equipped with an API ion source (HP 1100 MSD).

Analytical HPLC was performed on a system consisting of an Ultimate 3000 pump and PDA detector, and a TSP AS-3000 autosampler with a Gemini-NX C_{18} column (4.6 mm × 250 mm) or on an Agilent 1260 system with Innoval C_{18} column (4.6 mm × 250 mm) equipped with a 254 nm UV detector using a linear gradient elution of the binary solvent system of H₂O/MeCN/TFA (v/v/v; A = 95/5/0.1 and B = 5/95/0.1) with the concentration of B from 0% to 100% over 20 min with a flow rate of 1.0 mL/min. Preparative HPLC for the purification of final compounds was performed on an Ultimate 3000 system with a Gemini-NX C_{18} column (21 mm × 250 mm) or on a Waters instrument with an Innoval C_{18} column (21.2 mm × 250 mm) equipped with a 254 nm UV detector with the same binary solvent system with a flow rate of 20 mL/min. Purities of the tested compounds were determined by analytical HPLC to be >95%.

Representative synthetic procedures for a few of the compounds are provided.

(*RS*)-2-*Amino-3-(5-(2-(3-(aminomethyl)benzyl)-2H-tetrazol-5-yl)-3-hydroxyisoxazol-4-yl)propanoic Acid (6). Hydrochloric acid (10 mL, 10 M) was added to 18 (180 mg, 0.21 mmol), and the resulting suspension was stirred for 40 min at 90 °C as the mixture turned clear. The solution was cooled, concentrated, and re-evaporated with water (5 mL × 3). Purification by reverse phase HPLC gave the pure <i>N*-2-alkylated isomer of **6** (41 mg, 54%) as a white solid. A small amount of pure *N*-1 isomer was also isolated. ¹H NMR (400 MHz, D₂O) δ 7.52–7.38 (m, 4.0H), 5.97 (s, 2.0H), 4.19 (dd, *J* = 7.3, 6.1 Hz, 1.0H), 4.14 (s, 2.0H), 3.34 (dd, *J* = 15.3, 6.2 Hz, 1.0H), 3.26 (dd, *J* = 15.3, 7.3 Hz, 1.0H). ¹³C NMR (101 MHz, D₂O) δ 171.66, 169.85, 155.59, 155.28, 133.59, 133.54, 129.99, 129.64, 129.53, 129.17, 105.44, 56.98, 52.66, 42.75, 22.49. LC/MS (API) Calcd. for C₁₅H₁₇N₇O₄: 359.1. Found: 360.1 [M + H]⁺.

(RS)-2-Amino-3-(5-(2-(3-chlorobenzyl)-2H-tetrazol-5-yl)-3-hydroxyisoxazol-4-yl)propanoic Acid (7). Hydrochloric acid (10 mL, 10 M) was added to 13 (150 mg, 0.23 mmol), and the resulting suspension was stirred for 40 min at 90 °C as the mixture turned clear. The solution was cooled, concentrated, and re-evaporated with water $(5 \text{ mL} \times 3)$. Purification by reverse phase HPLC gave 7 (50 mg, 60%) as a white solid. The N-1 and N-2 alkylated isomers could not be separated completely, ¹H NMR showed the ratio of the two isomers was 1:3. ¹H NMR (600 MHz, DMSO-d₆; 1:3 mixture of regioisomers) δ 7.46 (s, 1.0H), 7.43–7.26 (m, 4.1H), 7.15–7.11 (m, 0.4H), 6.00 (d, J = 2.3 Hz, 2.1H), 5.83 (s, 0.7H), 3.67 (ddd, J = 9.3, 6.9, 3.9 Hz, 1.5H), 3.25–3.19 (m, 1.5H), 3.00 (dd, J = 16.0, 7.2 Hz, 1.5H). ¹³C NMR (151 MHz, DMSO- d_6 ; 1:3 mixture of regio-isomers) δ 172.75, 172.12, 171.65, 171.45, 156.47, 153.69, 150.59, 145.39, 137.17, 136.39, 133.84, 133.80, 131.33, 131.25, 129.25, 128.92, 128.30, 127.74, 126.99, 111.37, 108.26, 55.98, 52.93, 52.64, 51.63, 40.54, 40.40, 39.91, 39.77, 24.87, 24.62. LC/MS (API) Calcd for C14H13ClN6O4: 364.1. Found: $365.0 [M + H]^+$

(*RS*)-2-Amino-3-(3-hydroxy-5-(2-(3-methylbenzyl)-2H-tetrazol-5yl)isoxazol-4-yl)propanoic Acid (**8**). Hydrochloric acid (10 mL, 10 M) was added to 14 (140 mg, 0.22 mmol), and the resulting suspension was stirred for 40 min at 90 °C as the mixture turned clear. The solution was cooled, concentrated, and re-evaporated with water (5 mL × 3). Purification by reverse phase HPLC gave **8** (21 mg, 28%) as white solid. The N-1 and N-2 alkylated isomers could not be separated completely; ¹H NMR showed the ratio of the two isomers was 1:3.6. ¹H NMR (600 MHz, D₂O; 1:3.6 mixture of regio-isomers) δ 7.20 (dd, J = 13.6, 6.2 Hz, 2.2H), 7.14 (t, J = 6.4 Hz, 2.0H), 7.08 (m, 0.5H), 7.03 (s, 0.3H), 6.99 (m, 0.5H), 5.82 (s, 2.0H), 5.79 (s, 0.6H), 4.24 (t, J = 7.0 Hz, 1.4H), 3.32 (dd, J = 15.2, 6.7 Hz, 1.4H), 3.22 (dd, J = 15.2, 7.3 Hz, 1.4H), 2.19 (s, 3.0H), 2.15 (s, 0.8H). ¹³C NMR (151 MHz, D₂O; 1:3.6 mixture of regio-isomers) δ 176.50, 173.56, 156.15, 153.70, 139.50, 132.93, 129.86, 129.18, 129.13, 125.66, 109.26, 57.18, 54.41, 23.73, 20.34. LC/MS (API) Calcd for $C_{15}H_{16}N_6O_4{:}$ 344.1. Found: 345.1 $[M\,+\,H]^+.$

In Vitro Pharmacology. Native Receptor Binding Assays. Affinities for native AMPA, KA, and NMDA receptors in rat cortical synaptosomes were determined using 5 nM $[^{3}H]AMPA$ (55.5 Ci/mmol),²⁷ 5 nM $[^{3}H]KA$ (58.0 Ci/mmol),²⁸ and 2 nM $[^{3}H]CGP$ 39653 (47.3 Ci/mmol),²⁹ respectively, with minor modifications as previously described.³⁰ Rat brain membrane preparations used in these receptor binding experiments were prepared according to a method previously described.³¹

Recombinant Receptor Binding Assays. Sf9 cells were cultured and infected with recombinant baculovirus of rat AMPA receptors (GluA1_o-4_o) or rat KA receptors (GluK1-3), and membranes were prepared and used for binding as previously detailed.^{17,32} The affinities of the compounds for GluA1_o, GluA2(R)_o, GluA3_o, and GluA4_o were determined from competition experiments with 2–5 nM (*RS*)-[³H]AMPA and at GluK1(Q)_{1b}, GluK2(V,C,R)_a, and GluK3_a using 5–10 nM [³H]KA (47.2 Ci/mmol). Italic letters in parentheses indicate the RNA-edited isoforms of the subunits. Mutant receptors are numbered according to their full-length protein sequence. Ligand concentration–response curves were analyzed using GraphPad Prism v6 (Graphpad Software Inc., San Diego, CA) to determine the IC₅₀, $n_{\rm H}$, and K_i .

TEVC Pharmacology. Surgical procedures were conducted under the approval of the Danish Ministry of Justice Animal Experiments Inspectorate. Mature female Xenopus laevis were anaesthetized using 0.1% ethyl 3-aminobenzoate methanesulfonate (tricaine), and ovaries were surgically removed. The ovarian tissue was dissected and treated with 2 mg/mL collagenase in nominally Ca²⁺-free Barth's medium (in mM: 88 NaCl, 1 KČl, 0.33 Ca(NO₃)₂, 0.41 CaCl₂, 0.82 MgSO₄, 2.4 NaHCO₃, 10 HEPES, pH 7.4) for 2 h at room temperature. On the second day, oocytes were injected with 25–50 nL of ~ 1 mg/mL rat $GluA2(Q)_i$ cRNA and incubated in Barth's medium with 0.10 mg/mL gentamicin (Sigma Chemical) and 1% penicillin-streptomycin (Life Technologies, Paisley, UK) at 17 °C. Oocytes were typically used for recordings from 3 to 10 days postinjection and were voltage-clamped with the use of a two-electrode voltage clamp (GeneClamp 500B, Axon Instruments, Union City, CA) with both microelectrodes filled with 3 M KCl. Recordings were made while the oocytes were continuously superfused with frog Ringer's solution (in mM: 115 NaCl, 2 KCl, 1.8 BaCl₂, 5 HEPES, pH 7.6). Compounds were dissolved in frog Ringer's solution and added by bath application. Recordings were made at room temperature. Efficacy measurements were made in the presence of 100 μ M cyclothiazide in order to block receptor desensitization. To determine the maximum response, oocytes were stimulated with 1 mM (S)-glutamate plus 100 μ M cyclothiazide. Agonist concentration-response curves were analyzed using GraphPad Prism v6 (Graphpad Software Inc., San Diego, CA) to determine the EC_{50} using a four-parameter logistic equation.

Fluo-4/C a^{2+} Assay. The functional properties of (S)-glutamate and compounds 6-11 were characterized in stable GluA1_i, GluA2Q_i, $GluA3_{i}$ and $GluA4_{i}$ -HEK293 cell lines in the $Fluo-4/Ca^{2+}$ assay essentially as previously described.³³ The cells were split into poly(Dlysine)-coated black 96-well plates with clear bottom (BD Biosciences, Bedford, MA). The medium was aspirated 16-24 h later, and the cells were incubated in 50 μ L of assay buffer [Hanks buffered saline solution (HBSS) containing 20 mM HEPES, 1 mM CaCl₂, 1 mM MgCl₂, and 2.5 mM probenecid, pH 7.4] supplemented with 6 μ M Fluo4/AM (Molecular Probes, Eugene, OR) at 37 °C for 1 h. The buffer was aspirated, the cells were washed with 100 μ L of assay buffer, and then 100 μ L of assay buffer supplemented with 100 μ M cyclothiazide was added to the wells (in the antagonist experiments, this buffer was supplemented with various antagonist concentrations). Then the 96-well plate was assayed in a FlexStation3 benchtop multimode microplate reader (Molecular Devices) measuring emission at 520 nm [in fluorescence units (FU)] caused by excitation at 485 nm a total of 90 s before and after addition of 33 μ L of agonist solution (the agonists were dissolved in assay buffer). The experiments were performed in duplicate at least three times for each compound at each cell line.

X-ray Structures. Crystallization. Rat GluA2 ABD (GluR2- $S1S2J^{24}$), comprising segment S1 residues 392–506, a GT linker, and segment S2 residues 631–776 (numbering without signal peptide), was expressed and purified in the presence of L-aspartate as previously reported.^{34,35}

Crystallization experiments were performed by the hanging drop vapor diffusion method using drops containing 1 μ L of protein solution plus 1 μ L of reservoir solution. The drops were equilibrated against 0.5 mL of reservoir solution. All experiments were setup at 6 °C. For the complex of GluA2 ABD with 6, 4.6 mg/mL GluA2 ABD and 7.8 mM 6 in buffer containing 10 mM HEPES, 20 mM NaCl, and 1 mM EDTA, pH 7.0, was used to crystallize the complex. The crystal used for data collection was obtained from reservoir solution containing 20% PEG4000, 0.3 M lithium sulfate, and 0.1 M sodium acetate buffer, pH 5.5. For the complex of GluA2 LBD with 7, 5.2 mg/ mL GluA2 ABD and 5.3 mM 7 were used, and the crystal used for data collection was obtained from reservoir solution containing 24.4% PEG4000, 0.1 M ammonium sulfate, and 0.1 M acetate buffer, pH 5.5. For the complex of GluA2 ABD with 8, 5.2 mg/mL GluA2 ABD and 5.3 mM 8 were used, and the crystal used for data collection was obtained from reservoir solution containing 18% PEG4000, 0.1 M ammonium sulfate, and 0.1 M phosphate-citrate buffer, pH 4.5. Before data collection, the crystals were flash cooled in liquid nitrogen after briefly soaking in reservoir solution with 20% glycerol.

Structure Determination. X-ray diffraction data were collected on beamline 1911-3 at MAX-lab, Lund, Sweden.³⁶ The data were processed with XDS^{37} (6, 7) or mosfilm³⁸ (8) and scaled using SCALA³⁹ in CCP4i.⁴⁰ All three structures were solved by molecular replacement using PHASER⁴¹ within CCP4i. The structure of GluA2 ABD with **5b** (PDB code 2P2A, molB, protein atoms only)¹⁸ was used as search model. Visual inspection of the structures in COOT⁴² revealed unambiguous density corresponding to (S)-6, (S)-7, and (S)-8, respectively. Ligand coordinates were generated in Maestro [Maestro, version 9.2, Schrödinger, LLC, New York, NY, 2011] and fitted into the electron density. Topology and parameter files for the ligands were obtained using eLBOW.43 The structures were refined in PHENIX.44 The structure of GluA2 ABD with 6 was refined with isotropic B factors, TLS, and riding hydrogen atoms. The structure of GluA2 ABD with 7 was refined with isotropic B factors and riding hydrogen atoms. The structure of GluA2 ABD with 8 was refined with isotropic B factors, TLS, and riding hydrogen atoms. The structures were validated using tools in COOT and PHENIX, as well as the PDB ADIT validation server. Figures were prepared using PyMOL (Schrödinger, LLC, New York, NY). Statistics of data collection and refinement can be found in Supporting Information Table S2.

ASSOCIATED CONTENT

S Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.jmed-chem.5b01982.

Synthetic procedures for compounds 9-22, pharmacological data from Fluo-4/Ca²⁺ assay (Table S1), crystal data, data collection, and refinement statistics of GluA2 ABD with 6, 7, and 8 (Table S2), and electron density and hydrogen bonding interactions for 6, 7, and 8 (Figure S1) (PDF)

SMILES compound representations with activity data (CSV)

Accession Codes

The structure coordinates and corresponding structure factor file of GluA2 ABD with 6, 7, and 8 have been deposited in the Protein Data Bank under the accession codes SFHM, SFHO, and SFHN, respectively.

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

ABD, agonist binding domain; AMPA, (*S*)-2-amino-3-(3-hydroxy-5-methyl-isoxazol-4-yl)propionic acid; MeTetAMPA, 2-amino-3-[3-hydroxy-5-(2-methyl-2*H*-5-tetrazolyl)-4-isoxazolyl]propionic acid; BnTetAMPA, 2-amino-3-[3-hydroxy-5-(2-benzyl-2*H*-5-tetrazolyl)-4-isoxazolyl]propionic acid; Glu, (*S*)-glutamic acid; iGluRs, ionotropic Glu receptors; KA, kainic acid; NMDA, *N*-methyl-D-aspartic acid; TEVC, two-electrode voltage clamp; HEPES, 4-(2-hydroxyethyl)-1-piperazineethane-sulfonic acid

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