

Accepted Article

Title: The proteome-wide potential for reversible covalency at cysteine

Authors: Kristine Senkane, Ekaterina Vinogradova, Radu Suciu, Vincent Crowley, Balyn Zaro, Michael Bradshaw, Ken Brameld, and Benjamin Cravatt

This manuscript has been accepted after peer review and appears as an Accepted Article online prior to editing, proofing, and formal publication of the final Version of Record (VoR). This work is currently citable by using the Digital Object Identifier (DOI) given below. The VoR will be published online in Early View as soon as possible and may be different to this Accepted Article as a result of editing. Readers should obtain the VoR from the journal website shown below when it is published to ensure accuracy of information. The authors are responsible for the content of this Accepted Article.

To be cited as: *Angew. Chem. Int. Ed.* 10.1002/anie.201905829
Angew. Chem. 10.1002/ange.201905829

Link to VoR: <http://dx.doi.org/10.1002/anie.201905829>
<http://dx.doi.org/10.1002/ange.201905829>

The proteome-wide potential for reversible covalency at cysteine

Kristine Senkane,^[a] Ekaterina V. Vinogradova,^{*[a]} Radu M. Suciuc,^[a] Vincent M. Crowley,^[a] Balyn W. Zaro,^[a] J. Michael Bradshaw,^[b] Ken A. Brameld,^[b] and Benjamin F. Cravatt^{*[a]}

Abstract: Reversible covalency, achieved with, for instance, highly electron-deficient olefins, offers a compelling strategy to design chemical probes and drugs that benefit from the sustained target engagement afforded by irreversible compounds, while avoiding permanent protein modification that persists following unfolding and/or proteolytic processing. So far, reversible covalency has mainly been evaluated for cysteine residues in individual kinases and the broader potential for this strategy to engage cysteines across the proteome remains unexplored. Here we describe a mass-spectrometry-based platform that integrates gel filtration (GF) with activity-based protein profiling (ABPP) to assess cysteine residues across the human proteome for both irreversible and reversible interactions with small-molecule electrophiles. Using this method, we identify numerous cysteine residues from diverse protein classes that are reversibly engaged by cyanoacrylamide fragment electrophiles, revealing the broad potential for reversible covalency as a strategy for chemical probe discovery.

Chemical probes and drugs that operate by a covalent irreversible mechanism have several potentially advantageous properties, including increased duration of action, reduced pharmacokinetic sensitivity, and the potential for improved potency at otherwise shallow small-molecule binding pockets.^[1-4] A number of FDA-approved drugs act by a covalent irreversible mechanism, including multiple recently approved kinase inhibitors used to treat diverse cancers.^[5-7] These compounds react with non-catalytic cysteine residues in the active sites of target kinases like EGFR and BTK. Despite the remarkable success of drugs that act by a covalent irreversible mechanism, concerns remain about the potential safety and immunogenicity risks associated with the chemical modification of proteins *in vivo*, especially for drugs that require higher doses for efficacy, which may increase the adduction of off-target proteins.^[8-9]

Advanced chemical proteomic methods have emerged to facilitate the characterization and optimization of target selectivity for covalent, irreversible drugs *in vitro*^[10-12] and *in vivo*.^[13] These methods, combined with additional strategies – including the design of reactive groups with i) tempered intrinsic electrophilicity,^[14-16] ii) metabolic vulnerabilities that attenuate reactivity,^[17] and iii) covalent reversible mechanisms of action^[18-23] have expanded the optionality for design of advanced chemical probes and drugs that covalently bind to proteins.^[24-25] The third strategy, which has a rich history of success for targeting catalytic serines/threonines in the active sites of

hydrolases/proteases (e.g., α -ketoamides (serine),^[26-28] boronic acids (serine and threonine),^[29] cyanamides^[30-32]) has more recently been extended to cysteine (e.g., α -cyanoacrylamide,^[18-22] reversible formation of Meisenheimer complexes^[33]), and lysine (e.g., 2-acetyl arylboronic acids^[34]) residues. Optimized covalent reversible electrophiles have potential advantages of preserving the pharmacological benefits of extended on-target residence time associated with irreversibly acting compounds, while possibly also i) achieving greater selectivity through avoidance of weaker-binding (and, consequently, rapidly disassociating) off-targets, and ii) minimizing risk for idiosyncratic toxicity that may be caused by permanent modification of proteins.

Most of the methods described to date for characterizing covalent reversible electrophiles are target-specific, often employing recombinantly expressed proteins, and, to our knowledge, strategies to evaluate reversible covalency on a proteome-wide scale have not yet been described. Establishing a robust method to profile the landscape of protein targets of covalent reversible electrophiles in native biological systems would enable the optimization of compound selectivity, as well as the discovery of additional proteins amenable to this form of pharmacological perturbation. Here, we describe a quantitative method that combines gel filtration (GF) with activity-based protein profiling (ABPP) to evaluate the proteome-wide target landscape of α -cyanoacrylamide fragments as a prototype cysteine-directed covalent reversible electrophile.

We adapted a competitive isoTOP-ABPP (isotopic tandem orthogonal proteolysis-ABPP) method, which has been used to quantify the interactions of cysteine^[11] and lysine^[10] residues with covalent *irreversible* electrophilic fragments, to evaluate the covalent *reversible* interactions of α -cyanoacrylamide fragments with cysteine residues in the human proteome (**Fig. 1A**). We hypothesized that introducing a GF step after fragment treatment could distinguish fragments that reversibly versus irreversibly bind to cysteines, as the former, but not latter events should show substantially reduced competitive isoTOP-ABPP ratios, or R values (DMSO-treated/fragment-treated), following GF (**Fig. 1B**).

The human Ramos B cell line proteome was prepared and treated with DMSO, α -chloroacetamide fragment **1**, or one of two α -cyanoacrylamides (**2** or **3**) (**Fig. 2A**). α -Chloroacetamide **1** was chosen because this electrophilic fragment has been found to show broad reactivity with cysteines in the human proteome, enabling its deployment as a “scout” fragment to discover druggable cysteines at protein-protein interfaces^[11, 35] and that support E3 ligase-mediated protein degradation.^[36] The electron-withdrawing nitrile group on the α -cyanoacrylamide of the corresponding 6-methoxy-tetrahydroquinoline fragments **2** and **3** elevates the reactivity of the Michael acceptor towards nucleophilic addition at the β -carbon compared to the corresponding acrylamide group and also increases the acidity of the C α -H bond due to stabilization of the α -carbanion, rendering the reaction reversible.^[18-20] α -Cyanoacrylamides have been used to create potent and selective kinase inhibitors that act by a covalent reversible mechanism.^[18-22] In most of these cases, however, α -cyanoacrylamides were appended to high-

[a] K. Senkane, Dr. E. V. Vinogradova,* Dr. R. M. Suciuc, Dr. V. M. Crowley, Dr. B. W. Zaro, and Prof. Dr. B. F. Cravatt*

Department of Chemistry
The Scripps Research Institute
La Jolla, CA 92037

*E-mail: vinograd@scripps.edu, cravatt@scripps.edu

[b] Dr. J. M. Bradshaw and Dr. K. A. Brameld
Principia Biopharma
220 E. Grand Avenue, South San Francisco, CA 94080

Supporting information for this article is given via a link at the end of the document.

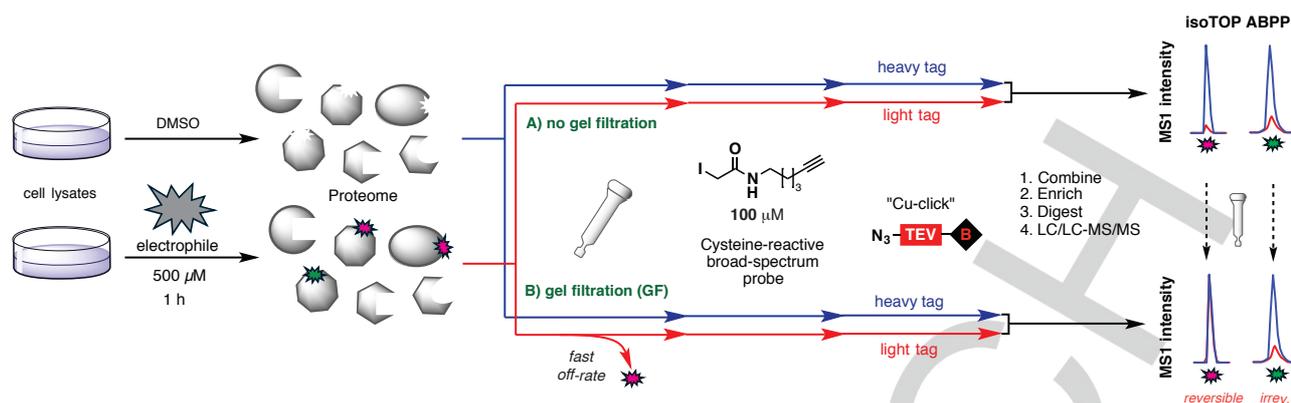


Figure 1. isoTOP-ABPP (A) and GF-isoTOP-ABPP (B) for proteome-wide evaluation of reactivity and reversibility of cysteine-directed electrophilic compounds.

affinity binding elements targeting the kinase ATP pocket. The extent to which the hyper-electrophilic α -cyanoacrylamide group can reversibly bind to cysteine residues in other proteins across the human proteome remains unknown.

Following treatment with compounds (500 μ M each, 1h) or DMSO, Ramos cell proteome samples were split in half, with one portion undergoing GF on a Zeba Spin Desalting Column (7K MWCO, 2 mL) to remove compounds. Both gel-filtered and unfiltered samples were then treated separately with an iodoacetamide (IA)-alkyne probe (100 μ M, 1h), which broadly reacts with cysteine residues, and analyzed by isoTOP-ABPP to identify compound-sensitive cysteines. In total, more than 5000 cysteines were quantified on 2499 proteins (**Supplementary Table 1**) and individual sites were considered: 1) liganded, if they displayed R values ≥ 4 ($\geq 75\%$ reduction in IA-alkyne labeling) before GF, and 2) reversibly liganded, if the reduction in R value (ΔR) following GF was ≥ 2 fold ($\geq 50\%$).

Both chloroacetamide **1** and α -cyanoacrylamide **2** showed broad reactivity profiles, with each electrophilic fragment liganding more than 100 cysteines in the Ramos cell proteome (**Fig. 2B**, **Supplementary Fig. 1A and 1B**, and **Supplementary Table 2**). α -Cyanoacrylamide **3**, on the other hand, was much less reactive with the cysteine proteome, likely reflecting the sterically obstructive impact of the larger *tert*-butyl capping group (**Fig. 2B**, **Supplementary Fig. 1C**, and **Supplementary Table 2**). The vast majority of cysteines liganded by **2** and **3** were found to be reversible, while a much smaller fraction of apparently reversible interactions was observed for **1** (**Fig. 2B and C**).

A comparison of the target landscape of **1** and **2** revealed a striking number of cysteines that were preferentially liganded by one of the two fragments (**Fig. 3A–D**, and **Supplementary Table 3**). However, this difference in target interactions is unlikely to contribute to the distinct reversibility profiles displayed by **1** and **2**,

as cysteines liganded by both fragments generally showed reversible interactions exclusively with fragment **2** (e.g., see REEP5_C18 in **Fig. 3E** and other examples in **Supplementary Fig. 2**). We also note that most of the cysteines preferentially liganded by **2** did not interact with the analogous acrylamide fragment **SI-1** (**Supplementary Fig. 1**), indicating that the greater intrinsic electrophilicity of **2** contributed to its broader reactivity profile with the cysteine proteome (**Supplementary Fig. 1B**). We confirmed the respective reactivity profiles of REEP5_C18 with **1** and **2**, and the selective reversibility of the latter interaction by gel-based ABPP, using recombinantly expressed wild type and C18A mutant forms of this protein (**Fig. 3F** and **Supplementary Fig. 3**).

The cysteines liganded by **2** were broadly distributed across different protein classes, including proteins such as transcriptional regulators and adapters that have historically

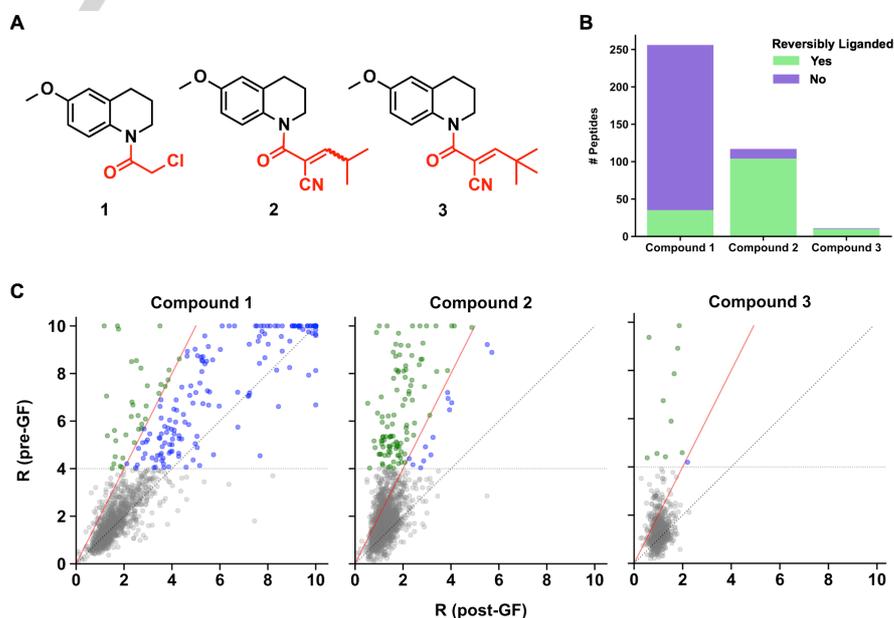


Figure 2. Proteome-wide assessment of reversibility of cysteine-electrophilic compound interactions by GF-isoTOP-ABPP. (A) Structures of covalent irreversible (**1**) and covalent reversible (**2** and **3**) electrophiles used in the study. (B) Bar graph showing cysteines that are liganded irreversibly (purple) or reversibly (green) by compounds **1-3**. (C) Scatter plot comparisons of isoTOP-ABPP R values for cysteines before and after GF. The color-coding matches that used in part B to designate cysteines that are reversibly or non-reversibly liganded by compounds **1-3**. Red line denotes limit of reversibility ($R \geq 4$ pre-GF and $\Delta R \geq 50\%$ post-GF). Identity line ($R_{\text{pre-GF}} = R_{\text{post-GF}}$) is dotted grey. Cysteines that were not liganded ($R < 4$ pre-GF) are depicted in grey.

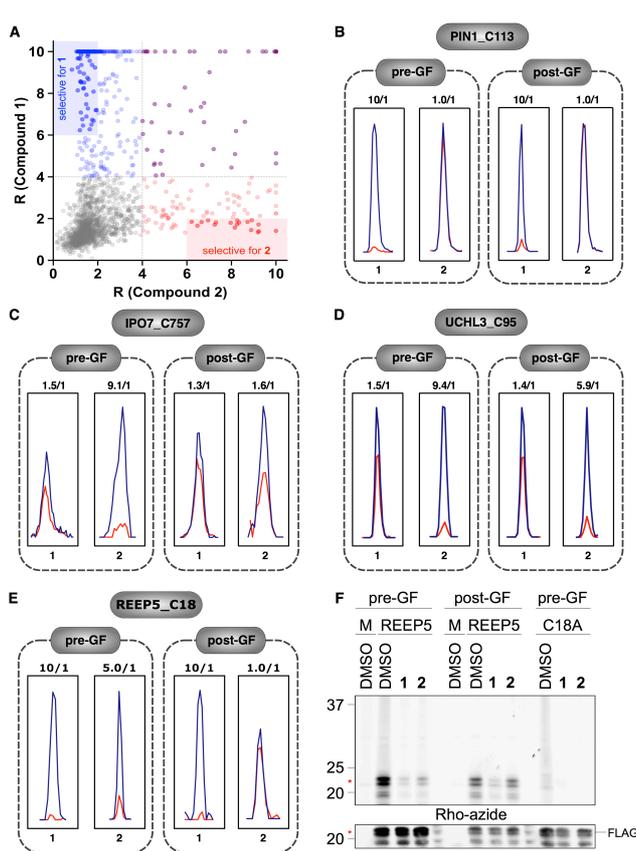


Figure 3. Comparison of protein targets of chloroacetamide **1** and α -cyanoacrylamide **2**. (A) Scatter plot showing pre-GF targets of **1** (blue) and **2** (red), with overlapping targets shown in purple. Areas of high selectivity for individual compounds (> 3 -fold) are shaded. (B–E) Representative MS1 spectra showing examples of cysteines that were preferentially liganded by compounds **1** or **2** – (B) C113 of PIN1, (C) C757 of IPO7, (D) C95 of UCHL3 – or generally liganded by both – (E) C18 of REEP5. Examples of reversible (C, E) and non-reversible (D) liganding with **2** are shown. (F) Fluorescent gel and Western blot confirmation of non-reversible and reversible interactions of C18 of REEP5 with **1** and **2**, respectively. Top, gel-based ABPP of HEK293T cells expressing recombinant REEP5, REEP5_C18A or empty vector (mock, M) treated with DMSO, **1**, or **2** with and without GF and then subsequently labeled with an alkyne analogue of **1** (**1-alkyne**) and conjugated to an azide-rhodamine tag by copper-catalyzed azide-alkyne cycloaddition chemistry for visualization (see SI for details). Bottom, recombinant protein expression was confirmed by anti-FLAG Western blotting.

represented challenging targets for chemical probe development (Fig. 4 and Supplementary Table 4). Interestingly, while most cysteines interacted with **2** in a reversible manner, there were compelling examples of cysteines that maintained engagement with **2** post-GF, including some cysteines that were not targeted by **1** (despite its greater overall cysteine reactivity profile across the proteome). A prominent example was the catalytic cysteine (C95) in the ubiquitin hydrolase UCHL3 (Fig. 3D). We speculate that these cases reflect a binding interaction that is sufficiently strong to preserve **2**-cysteine interactions following removal of excess free compound. Consistent with this hypothesis, we confirmed that PRN629, an optimized α -cyanoacrylamide inhibitor of BTK,^[22] maintained target engagement post-GF (Supplementary Fig. 4).

In summary, we have developed a chemical proteomic platform to globally evaluate reversible covalency of cysteine-reactive electrophilic compounds. Building upon our experience in mapping reactive cysteines on a proteome-wide scale,^[11, 35, 37]

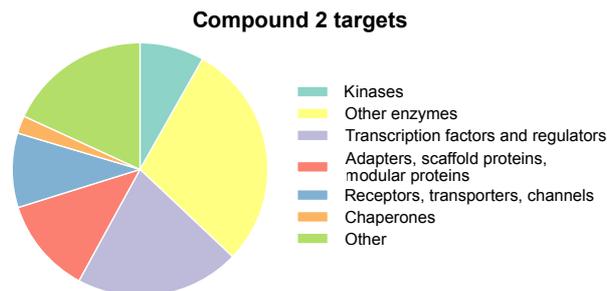


Figure 4. Functional classes of proteins with cysteines that are liganded by compound **2**.

we have shown that introducing a GF step after electrophilic compound treatment and prior to IA-alkyne exposure and chemical proteomic workup can illuminate cysteines that interact with compounds in a reversible manner. We used the described platform to evaluate the proteomic reactivity of the hyper-electrophilic α -cyanoacrylamide group, revealing a strikingly broad potential to engage cysteines across diverse protein classes, in many cases with selectivity over a structurally related α -chloroacetamide. These data indicate that even the presumably modest degree of binding affinity afforded by the 6-methoxy-tetrahydroquinoline fragment recognition group is sufficient to stabilize a large number of cysteine- α -cyanoacrylamide interactions in native proteomes. That most of these interactions are reversed following GF, unlike the PRN629-BTK interaction, indicates future studies could use the persistent blockade of IA-reactivity following GF as a convenient assay to evaluate analogue compounds for improved potency of binding to specific targets of interest. As one qualification to the approach, we should note that some proteins, such as those that are part of dynamic complexes or that require small molecule/metal cofactors for stability, may unfold following GF and produce profiles that are accordingly challenging to interpret for ligand interactions. We found, for instance, that several cysteines showing apparently reversible engagement by α -chloroacetamide **1** are in ribosomal proteins (Supplementary Table 2), and it is possible that these proteins undergo complex disassembly (or unfolding) following GF to expose a greater fraction of cysteines for labeling by the IA-alkyne probe. This caveat notwithstanding, we envision the application of the chemical proteomic platform described herein to additional cell types and electrophilic chemotypes to create a comprehensive map of cysteines amenable to reversible covalency for chemical probe and drug development, as well as to other nucleophilic amino acid residues and corresponding reversible covalent chemistries.^[23]

Acknowledgements

We gratefully acknowledge NIH (CA231991). This work was supported by the National Cancer Institute (CA212467 to V.M.C.) and by the American Cancer Society (PF-15-142-01-CDD to B.W.Z.). We thank Yan Lou and Leonard Sung for help with fragment synthesis and characterization. E.V.V. was supported by the Life Sciences Research Foundation (Pfizer Fellow). We thank Chris Joslyn for help with Gateway cloning.

Keywords: proteomics • reversible covalency • ABPP • reactive cysteines • α -cyanoacrylamides

- [1] J. Singh, R. C. Petter, T. A. Baillie, A. Whitty, *Nat. Rev. Drug Discov.* **2011**, *10*, 307-317.
- [2] Q. S. Liu, Y. Sabnis, Z. Zhao, T. H. Zhang, S. J. Buhrlage, L. H. Jones, N. S. Gray, *Chem. Biol.* **2013**, *20*, 146-159.
- [3] R. A. Bauer, *Drug Discov. Today* **2015**, *20*, 1061-1073.
- [4] D. S. Johnson, E. Weerapana, B. F. Cravatt, *Future Med Chem* **2010**, *2*, 949-964.
- [5] L. A. Honigberg, A. M. Smith, M. Sirisawad, E. Verner, D. Loury, B. Chang, S. Li, Z. Pan, D. H. Thamm, R. A. Miller, J. J. Buggy, *Proc Natl Acad Sci U S A* **2010**, *107*, 13075-13080.
- [6] P. Maione, A. Rossi, M. Bareschino, P. C. Sacco, C. Schettino, F. Casalupe, A. Sgambato, C. Gridelli, *Current pharmaceutical design* **2014**, *20*, 3894-3900.
- [7] J. C. Byrd, B. Harrington, S. O'Brien, J. A. Jones, A. Schuh, S. Devereux, J. Chaves, W. G. Wierda, F. T. Awan, J. R. Brown, P. Hillmen, D. M. Stephens, P. Ghia, J. C. Barrientos, J. M. Pagel, J. Woyach, D. Johnson, J. Huang, X. Wang, A. Kaptein, B. J. Lannutti, T. Covey, M. Fardis, J. McGreivoy, A. Hamdy, W. Rothbaum, R. Izumi, T. G. Diacovo, A. J. Johnson, R. R. Furman, *The New England Journal of medicine* **2016**, *374*, 323-332.
- [8] R. Mah, J. R. Thomas, C. M. Shafer, *Bioorg. Med. Chem. Lett.* **2014**, *24*, 33-39.
- [9] S. Nakayama, R. Atsumi, H. Takakusa, Y. Kobayashi, A. Kurihara, Y. Nagai, D. Nakai, O. Okazaki, *Drug metabolism and disposition: the biological fate of chemicals* **2009**, *37*, 1970-1977.
- [10] S. M. Hacker, K. M. Backus, M. R. Lazear, S. Forli, B. E. Correia, B. F. Cravatt, *Nature chemistry* **2017**, *9*, 1181-1190.
- [11] K. M. Backus, B. E. Correia, K. M. Lum, S. Forli, B. D. Horning, G. E. Gonzalez-Paez, S. Chatterjee, B. R. Lanning, J. R. Tejjaro, A. J. Olson, D. W. Wolan, B. F. Cravatt, *Nature* **2016**, *534*, 570-574.
- [12] Y. C. Chen, C. Zhang, *Genes & cancer* **2016**, *7*, 148-153.
- [13] S. Niessen, M. M. Dix, S. Barbas, Z. E. Potter, S. Lu, O. Brodsky, S. Planken, D. Behenna, C. Almaden, K. S. Gajiwala, K. Ryan, R. Ferre, M. R. Lazear, M. M. Hayward, J. C. Kath, B. F. Cravatt, *Cell chemical biology* **2017**, *24*, 1388-1400.
- [14] J. W. Chang, A. B. Cognetta, 3rd, M. J. Niphakis, B. F. Cravatt, *ACS chemical biology* **2013**, *8*, 1590-1599.
- [15] T. Barf, T. Covey, R. Izumi, B. van de Kar, M. Gulrajani, B. van Lith, M. van Hoek, E. de Zwart, D. Mittag, D. Demont, S. Verkaik, F. Krantz, P. G. Pearson, R. Ulrich, A. Kaptein, *J. Pharmacol. Exp. Ther.* **2017**, *363*, 240-252.
- [16] N. Shindo, H. Fuchida, M. Sato, K. Watari, T. Shibata, K. Kuwata, C. Miura, K. Okamoto, Y. Hatsuyama, K. Tokunaga, S. Sakamoto, S. Morimoto, Y. Abe, M. Shiroishi, J. M. M. Caaveiro, T. Ueda, T. Tamura, N. Matsunaga, T. Nakao, S. Koyanagi, S. Ohdo, Y. Yamaguchi, I. Hamachi, M. Ono, A. Ojida, *Nat Chem Biol* **2019**, *15*, 250-258.
- [17] B. W. Zaro, L. R. Whitby, K. M. Lum, B. F. Cravatt, *J. Am. Chem. Soc.* **2016**, *138*, 15841-15844.
- [18] I. M. Serafimova, M. A. Pufall, S. Krishnan, K. Duda, M. S. Cohen, R. L. Maglathlin, J. M. McFarland, R. M. Miller, M. Frodin, J. Taunton, *Nat. Chem. Biol.* **2012**, *8*, 471-476.
- [19] R. M. Miller, V. O. Paavilainen, S. Krishnan, I. M. Serafimova, J. Taunton, *J. Am. Chem. Soc.* **2013**, *135*, 5298-5301.
- [20] S. Krishnan, R. M. Miller, B. X. Tian, R. D. Mullins, M. P. Jacobson, J. Taunton, *J. Am. Chem. Soc.* **2014**, *136*, 12624-12630.
- [21] N. London, R. M. Miller, S. Krishnan, K. Uchida, J. J. Irwin, O. Eidam, L. Gibold, P. Cimermancic, R. Bonnet, B. K. Shoichet, J. Taunton, *Nat. Chem. Biol.* **2014**, *10*, 1066-1072.
- [22] J. M. Bradshaw, J. M. McFarland, V. O. Paavilainen, A. Bisconte, D. Tam, V. T. Phan, S. Romanov, D. Finkle, J. Shu, V. Patel, T. Ton, X. Y. Li, D. G. Loughhead, P. A. Nunn, D. E. Karr, M. E. Gerritsen, J. O. Funk, T. D. Owens, E. Verner, K. A. Brameld, R. J. Hill, D. M. Goldstein, J. Taunton, *Nat. Chem. Biol.* **2015**, *11*, 525-531.
- [23] A. Bandyopadhyay, J. M. Gao, *Curr. Opin. Chem. Biol.* **2016**, *34*, 110-116.
- [24] M. H. Potashman, M. E. Duggan, *J Med Chem* **2009**, *52*, 1231-1246.
- [25] T. A. Baillie, *Angew. Chem.-Int. Edit.* **2016**, *55*, 13408-13421.
- [26] R. B. Pemi, S. J. Almqvist, R. A. Byrn, G. Chandorkar, P. R. Chaturvedi, L. F. Courtney, C. J. Decker, K. Dinehart, C. A. Gates, S. L. Harbeson, A. Heiser, G. Kalker, E. Kolaczowski, K. Lin, Y. P. Luong, B. G. Rao, W. P. Taylor, J. A. Thomson, R. D. Tung, Y. Y. Wei, A. D. Kwong, C. Lin, *Antimicrob. Agents Chemother.* **2006**, *50*, 899-909.
- [27] S. Venkatraman, S. L. Bogen, A. Arasappan, F. Bennett, K. Chen, E. Jao, Y. T. Liu, R. Lovey, S. Hendrata, Y. H. Huang, W. D. Pan, T. Parekh, P. Pinto, V. Popov, R. Pike, S. Ruan, B. Santhanam, B. Vibulbhan, W. L. Wu, W. Y. Yang, J. S. Kong, X. Liang, J. Wong, R. Liu, N. Butkiewicz, R. Chase, A. Hart, S. Agrawal, P. Ingravallo, J. Pichardo, R. Kong, B. Baroudy, B. Malcolm, Z. Y. Guo, A. Prongay, V. Madison, L. Broske, X. M. Cui, K. C. Cheng, Y. S. Hsieh, J. M. Brisson, D. Prelusky, W. Korfmacher, R. White, S. Bogdanowich-Knipp, A. Pavlovsky, P. Bradley, A. K. Saksena, A. Ganguly, J. Piwinski, V. Girijavallabhan, F. G. Njoroge, *J. Med. Chem.* **2006**, *49*, 6074-6086.
- [28] D. L. Boger, H. Sato, A. E. Lerner, M. P. Hedrick, R. A. Fecik, H. Miyachi, G. D. Wilkie, B. J. Austin, M. P. Patricelli, B. F. Cravatt, *Proc Natl Acad Sci U S A* **2000**, *97*, 5044-5049.
- [29] J. Adams, M. Kauffman, *Cancer Invest* **2004**, *22*, 304-311.
- [30] A. Berteotti, F. Vacondio, A. Lodola, M. Bassi, C. Silva, M. Mor, A. Cavalli, *ACS medicinal chemistry letters* **2014**, *5*, 501-505.
- [31] C. Mathieu, E. Degrande, *Vascular health and risk management* **2008**, *4*, 1349-1360.
- [32] D. J. Augeri, J. A. Robl, D. A. Betebenner, D. R. Magnin, A. Khanna, J. G. Robertson, A. Wang, L. M. Simpkins, P. Taunk, Q. Huang, S.-P. Han, B. Abboa-Offei, M. Cap, L. Xin, L. Tao, E. Tozzo, G. E. Welzel, D. M. Egan, J. Marcinkeviciene, S. Y. Chang, S. A. Biller, M. S. Kirby, R. A. Parker, L. G. Hamann, *J. Med. Chem.* **2005**, *48*, 5025-5037.
- [33] R. J. Pearson, D. G. Blake, M. Mezna, P. M. Fischer, N. J. Westwood, C. McInnes, *Cell chemical biology* **2018**, *25*, 1107-1116.
- [34] G. Akcay, M. A. Belmonte, B. Aquila, C. Chuaqui, A. W. Hird, M. L. Lamb, P. B. Rawlins, N. Su, S. Tentarelli, N. P. Grimster, Q. Su, *Nat Chem Biol* **2016**, *12*, 931-936.
- [35] L. Bar-Peled, E. K. Kemper, R. M. Suci, E. V. Vinogradova, K. M. Backus, B. D. Horning, T. A. Paul, T. A. Ichu, R. U. Svensson, J. Olucha, M. W. Chang, B. P. Kok, Z. Zhu, N. T. Ihle, M. M. Dix, P. Jiang, M. M. Hayward, E. Saez, R. J. Shaw, B. F. Cravatt, *Cell* **2017**, *171*, 696-709.
- [36] X. Zhang, V. M. Crowley, T. G. Wucherpfennig, M. M. Dix, B. F. Cravatt, **2018**, 443804.
- [37] E. Weerapana, C. Wang, G. M. Simon, F. Richter, S. Khare, M. B. Dillon, D. A. Bachovchin, K. Mowen, D. Baker, B. F. Cravatt, *Nature* **2010**, *468*, 790-795.