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The SAR of 4-substituted (6,6-bicyclic) piperidine cathepsin S inhibitors

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Abstract—A series of competitive, reversible cathepsin S (CatS) inhibitors was investigated. An earlier disclosure detailed the discovery of the 4-(2-keto-1-benzimidazolinyl)-piperidin-1-yl moiety as an effective replacement for the 4-arylpiperazin-1-yl group found in our screening hit. Continued investigation into replacements for the 4-aryl piperazine resulted in the identification of potentially useful CatS inhibitors with enzymatic and cellular activity similar to that of JNJ 10329670 as disclosed in a previous publication.

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Cathepsin S (CatS), a cysteine protease found in the lysosome of hematopoietic cells, is integrally involved in antigen presentation of major histocompatibility complex class II (MHC-II) molecules. These molecules bind antigens and transport them to the cell surface for display to various cells of the immune system. The invariant chain (Ii), a component of MHC-II complex, prevents premature binding of non-antigenic peptides by acting as a chaperone. CatS mediates the cleavage of the Ii p10 fragment, prior to cell surface antigen presentation to CD4⁺ T cells.¹⁻³ Inhibition of CatS would block the necessary degradation of the Ii, preventing antigen presentation, resulting in immunosuppression with specificity for $CD4^+$ T cells. In CatS -/- mice, the flow of MHC-II molecules to the cell surface is significantly reduced.³ It is anticipated that selective inhibition of CatS would be therapeutically useful in diseases that are characterized by hyperimmune responses.

Recently, we reported on our efforts to identify novel noncovalent inhibitors of CatS.^{4,5} Our initial lead compound I, as previously disclosed, was identified through virtual screening of a subset of the J&J PRD library using DOCK.⁴ Subsequent development of the SAR

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led to the identification of compounds II and III (JNJ 10329670).^{4,5} Compounds in the latter series have improved selectivity profiles, cellular activity, and pharmacokinetics, with suitable physicochemical properties for further development. Both series share several common structural motifs; the aryl substituted pyrazole group, a saturated linker three carbons in length, and a 1,4-substituted basic nitrogen containing ring. Increasing the lipophilicity of the tetrahydropyrazolopyridine aryl, substituent was previously noted to improve enzymatic CatS activity.^{5,6}

In this report, we wish to detail our continued investigation into replacements for the aryl piperazine portion of compound II. The analogs included here maintain the structural commonality detailed above for both series II and III. The headgroup replacements of interest are shown in Figure 1.

Amines 1 and 2 were prepared according to Scheme 1 starting with readily available intermediates, 13a and 13b.⁷ Reductive amination⁸ of the anilines with *tert*-bu-tyl-4-oxo-1-piperidinecarboxylate followed by reduction of the double bond using H_2 in the presence of Pd/C or PtO₂ afforded the desired amines 14a and 14b. Subsequent hydrolysis, cyclization, and deprotection using standard conditions resulted in the preparation of the desired amines.⁹

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Figure 1. Aryl piperazine replacements.



Scheme 1. Synthesis of compounds 1 and 2. Reagents and conditions: (a) Boc-piperidone, NaBH(OAc)₃, CH₂Cl₂, AcOH, rt, R = H, 71%, R = Cl, 66%; (b) R = H: H₂, 10% Pd/C, EtOAc, rt, 95%; R = Cl: PtO₂, H₂, EtOAc, rt, 46%; (c) 1.0 M NaOH, 4:1 MeOH/H₂O, rt, R = H, 93%, R = Cl, 74%; (d) EDCI, CH₂Cl₂, rt, R = H, 95%, R = Cl, 52%; (e) 1:1 TFA/CH₂Cl₂, rt, R = H, 95%, R = Cl, 93%.

Piperidines 3–8 and piperazine 12 were prepared according to literature procedures.^{10–12} Amines 9 and 10 were prepared via reductive amination using readily available starting materials.⁸ Amine 11 was prepared according to Scheme 2. The hydroxyl group of compound 15, prepared according to a literature procedure,¹³ was alkylated with ethyl bromoacetate to yield ester 16. Hydrolysis of the ester to the acid, followed by cyclization and deprotection, gave piperidine 11.



Scheme 3. Preparation of compounds 20–23, 28, 31, 32, 34, and 37–57. Reagents and conditions: (a) epichlorohydrin (4–6 equiv), Cs_2CO_3 (2 equiv), DMF, rt; (b) amine, EtOH, reflux.



Scheme 4. Preparation of compounds 24–27, 29, 30, 33, 35, and 36. Reagents and conditions: (a) 3-bromopropanol, Cs₂CO₃, DMF, rt; (b) amine, NaBH(OAc)₃, AcOH, CH₂Cl₂.

formation of a secondary alcohol in the linker region as depicted by structure **II** above. These analogs were prepared according to Scheme 3. The pyrazole **17** was converted to racemic epoxide intermediate **18** using epichlorohydrin. Epoxide ring opening was accomplished using the amines shown in Figure 1 in refluxing EtOH to afford the desired compounds in 50–90% yield.

The second method, Scheme 4, provides access to the unsubstituted linker analogs as depicted by structure III above.⁵ Common intermediate 17, was regioselectively alkylated with 3-bromopropanol. The alcohol was oxidized to the corresponding aldehyde using Dess–Martin periodinane. The amines from Figure 1 were coupled to the aldehyde under reductive amination conditions to afford the desired deshydroxy analogs in 40–80% yield.

Variation of R⁴ (SO₂Me, Ac, Boc, and H), as shown in Table 1, indicates a preference for methyl sulfonamide. A notable loss of in vitro enzymatic activity is seen with R^4 = Boc and H. Substitution of H for OH at R^5 has a minimal effect on in vitro enzymatic activity or invariant chain degradation, with comparable analogs being equipotent (22 vs 26 and 23 vs 27). Both Tables 1 and 2 list the affinity for the alpha-1 $(\alpha 1_a)$ adrenergic receptor as percent inhibition at 1 µM for select compounds. Several analogs described below, offer insight into substitutions that can be incorporated to decrease alpha-1 cross-reactivity. As previously reported in the literature, substitution at R^2 with a chloro substituent is expected to reduce potentially unwanted cross reactivity with the αl_a receptor.⁵ The data obtained for entries 38 and 42 support this trend.

As shown in Table 2, modifications of the C3-aryl moiety of the pyrazole confirmed both 4-trifluoromethyl and 4-bromo substituents as optimal for in vitro enzymatic activity. Comparison of compounds containing $R^5 = OH$ and $R^5 = H$, again, demonstrates that this change has no effect on in vitro enzymatic activity. The data suggest a modest preference with respect to cellular potency for the C3 4-trifluoromethyl substitution as measured by the invariant chain degradation assay.

Table 1. Effect of N-substitution on cathepsin S activity



Entry	\mathbb{R}^4	R ⁵	CatS IC ₅₀ ^a (nM)	Ii IC ₅₀ (µM)	$\alpha l_a^{\ b}$
20	Н	OH	1925	_	_
21	Boc	OH	826	_	
22	Ac	OH	120	1.1	
23	MeSO ₂	OH	28	0.41	60%
24	Н	Н	1800	_	
25	Boc	Н	900		
26	Ac	Н	100	0.81	
27	MeSO ₂	Н	23	0.95	72%

^a Average of 4-6 determinations.

^bαl_a% inhibition values at 1 μM were determined by Cerep, SA.

Table 2. Effect of aryl substitution on CatS activity



Entry	Х	R ²	R ³	R ⁵	CatS	Ii	$\alpha l_a^{\ b}$
					$IC_{50}{}^{a}(nM)$	$IC_{50} \ (\mu M)$	
23	С	Н	A4-CF ₃	OH	16	0.41	60%
27	С	Н	A4-CF ₃	Н	23	0.95	72%
28	С	6'-Cl	A4-CF ₃	OH	16	1.3	
29	С	6'-Cl	A4-CF ₃	Н	130		
30	С	5'-Me	A4-CF ₃	Н	145	2.0	
31	Ν	Н	A4-CF ₃	OH	80	1.3	
32	С	Н	A4-Br	OH	32	1.8	
33	С	Н	A4-Br	Н	60	1.6	
34	С	6'-Cl	A4-Br	OH	48	4.4	
35	С	6'-Cl	A4-Br	Н	130	2.6	
36	С	5'-Me	A4-Br	Н	170	1.6	
37	Ν	Н	A4-Br	OH	68	0.9	

^a Average of 4–6 determinations.

 ${}^{b}\alpha 1_{a}$ % inhibition values at 1 μ M were determined by Cerep, SA.

The data in Table 3 indicate that structural changes to Y, R^2 , and R^3 failed to confer improved cellular activity but were well tolerated in terms of enzyme inhibition.

Additional molecules incorporating modifications to the left-hand side utilizing amines 8, 9, 10, 11, and 12 are included in Table 4. Substitution at the anilinic position (tetrahydroquinoline, 48) versus the homolog (tetrahydroisoquinoline, 49) is preferred for CatS enzymatic inhibition. However, the potency of 48 is an order of magnitude lower than 41, suggesting that the presence of a second basic nitrogen in this position is unfavorable. Analogs 50 and 51 also substituted in the anilinic

Table 3. Effect of varying the headgroup on cathepsin S activity

$\begin{array}{c} 0 \\ Y \\ W \\ 8' \\ 7' \\ R^2 \\ 6' \\ \end{array} \begin{array}{c} N \\ OH \\ N \\ SO_2 Me \\ SO_2 Me \\ \end{array}$						
Entry	Y	\mathbb{R}^2	\mathbb{R}^3	CatS	Ii IC ₅₀ (μM)	$\alpha l_a^{\ b}$
				$IC_{50}^{a}(nM)$		
38	CH_2	7'-Cl	CF_3	13	0.98	<10%
39	CH_2	7'-Cl	Br	18	0.75	
40	CH_2	Η	CF_3	23	1.8	
41	CH_2	Η	Br	33	2.0	
42	NH	7'-Cl	CF_3	18	1.9	40%
43	NH	7'-Cl	Br	33	1.5	
44	NH	Η	CF_3	48	1.4 ^c	
45	NH	Н	Br	65	1.1°	
46	0	Н	CF_3	88	1.4 ^c	
47	0	Н	Br	133	_	

^a Average of 4–6 determinations.

 $^b\alpha 1_a$ % inhibition values at 1 μM were determined by Cerep, SA. c Single determination.

Table 4. Effect of varying the headgroup on cathepsin S activity



			=	
Entry	Amines (Fig. 1)	R ³	CatS IC ₅₀ ^a (nM)	$Ii \; IC_{50} \; (\mu M)$
48	9	Br	322	_
49	10	Br	1775	_
50	8	CF_3	30	1.3
51	8	Br	75	2.7 ^b
52	11	CF_3	163	0.93 ^b
53	12	CF_3	1700	

^a Average of four determinations.

^b Single determination.

position are equipotent to analogs 42 and 43, further supporting this hypothesis. Ring expansion of the oxazine ring, as illustrated by compound 52, resulted in a 10-fold loss in enzymatic activity over compound 23. Interestingly, these compounds show similar potency in the cellular assay. Maintaining the benzylic substitution but increasing the aromaticity of the pendant group with concomitant incorporation of a piperazine ring resulted in a significant drop in enzymatic activity as illustrated by compound 53.

In conclusion, replacement of the *N*-arylpiperazine in our original series of inhibitors with 4-(6,6-bicyclic) piperidines resulted in the identification of several potent, noncovalent CatS inhibitors. The incorporation of substituents on the remote aryl ring of the bicyclic head-groups provided improved selectivity over the αl_a adrenergic receptor while maintaining CatS activity. Several of these analogs have been selected for further evaluation.

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