

Synthesis and Evaluation of Sulfonylnitrophenylthiazoles (SNPTs) as Thyroid Hormone Receptor–Coactivator Interaction Inhibitors

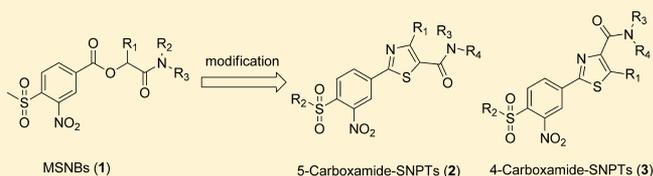
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S Supporting Information

ABSTRACT: We previously identified a series of methylsulfonylnitrobenzoates (MSNBs) that block the interaction of the thyroid hormone receptor with its coactivators. MSNBs inhibit coactivator binding through irreversible modification of cysteine 298 of the thyroid hormone receptor (TR). Although MSNBs have better pharmacological features than our first generation inhibitors (β -aminoketones), they contain a potentially unstable ester linkage. Here we report the bioisosteric replacement of the ester linkage with a thiazole moiety, yielding sulfonylnitrophenylthiazoles (SNPTs). An array of SNPTs representing optimal side chains from the MSNB series was constructed using parallel chemistry and evaluated to test their antagonism of the TR-coactivator interaction. Selected active compounds were evaluated in secondary confirmatory assays including regulation of thyroid response element driven transcription in reporter constructs and native genes. In addition the selected SNPTs were shown to be selective for TR relative to other nuclear hormone receptors (NRs).



INTRODUCTION

The nuclear hormone receptors (NRs) are transcription factors that are therapeutic targets for metabolic disease, immunology, reproductive health, and cancer.^{1–3} The NR superfamily includes the thyroid hormone receptors (TRs), TR α and TR β , that regulate development, growth, and metabolism.^{4,5} Although the TR isoforms are widely expressed, they follow tissue specific patterns that vary with developmental stage.^{6,7} The TR isoforms have distinct regulatory roles.^{8,9}

Thyroid hormone (T3) regulates transcriptional responses mediated by TR,⁹ which contains an amino terminal transcription activation domain (AF-1), a central DNA binding domain (DBD), and a carboxyl terminal ligand binding domain (LBD) that contains a T3-inducible coactivator binding domain, AF-2.¹⁰ TR usually functions as a heterodimer with the retinoid X receptor (RXR). At low levels of T3, TR binds corepressors using the AF-2 domain and suppresses basal transcription at thyroid-responsive elements (TREs). In response to increasing concentrations of T3, TR undergoes a conformational change, releasing corepressor proteins and binding coactivator proteins, thus activating gene transcription.^{11,12} The dominant family of coactivators is the SRCs, which include SRC1 (NcoA1), SRC2 (GRIP1/TIF2), and SRC3 (AIB1/TRAM1/RAC3/ACTR).¹³ The SRCs include both nuclear receptor interaction (NID) and activation domains. The SRC's NID includes a variable number of a conserved NR box motif, containing the LXXLL sequence, that binds to the TR's AF-2 domain.^{14,15} This interaction is mediated by a small, well-defined binding pocket¹⁶ that

makes the AF-2 domain an ideal target for developing inhibitors of TR–SRC interactions. Although a number of small molecule modulators of TR have been developed recently, including agonists such as GC-1,^{17–19} TRIAC,²⁰ and KB-141^{21,22} and antagonists such as NH-3,^{23–25} most target the ligand binding pocket in the LBD.

We have previously reported a series β -aminoketones that disrupt the TR–coactivator interaction without affecting T3 binding.^{26–28} Unfortunately these compounds suffered from multiple liabilities in vivo, thus requiring development of a new scaffold. The second generation TR β –SRC2 inhibitors, methylsulfonylnitrobenzoates (**1**, MSNBs), were identified in a quantitative high throughput screen (qHTS).²⁹ Both the β -aminoketones and MSNBs have a similar inhibition mechanism, irreversibly modifying Cys298 within the AF-2 domain of TR.³⁰ However, the MSNBs have two major advantages for the development of TR-coactivator inhibitors for use in vivo. First, MSNB members are predicted to lack the cardiac activity exhibited by the β -aminoketones because they lack the basic tertiary amines that lead to ion channel binding. Second, the MSNBs are more stable than the β -aminoketones at physiological pH. The MSNBs have two distinct structural features: the methylsulfonyl group that acts as a leaving group and the ester-linked acetamide group that appears to target the compound to the AF-2 domain.

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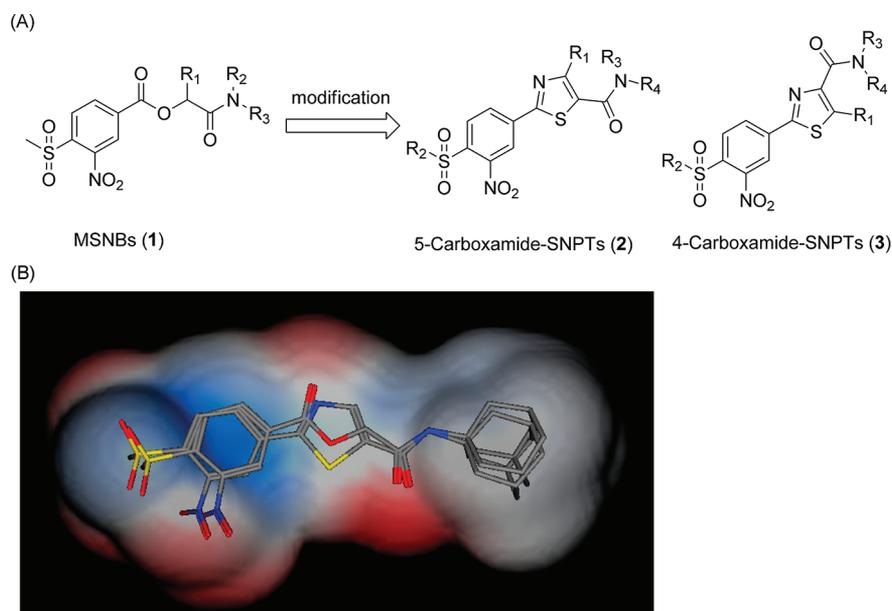


Figure 1. (A) Structural modification of MSNBs leading to SNPTs. (B) The translucent shape is the van der Waals surface of MSNBs and SNPTs. The colors of the translucent portion represent electrostatics of both molecules: red (negative), blue (positive), and white (neutral). Overall there is good alignment between the MSNBs and the SNPTs, thus indicating their theoretical viability as more stable bioisosteres.

Carboxylic esters are often metabolically unstable *in vivo* because of facile hydrolysis by esterases in multiple compartments and intrinsic chemical instability in the stomach. A common strategy to replace esters is to use heterocyclic bioisosteres with increased stability to degradation.^{31,32} A structural analysis indicated that thiazole-linked MSNBs, called sulfonylnitrophenylthiazoles (SNPTs), gave good alignments between the requisite aromatic and side chain groups of the MSNBs (Figure 1). For this reason, we modified the MSNB structure to produce SNPTs. Here we report an efficient method of parallel synthesis of SNPTs and their evaluation as thyroid hormone receptor–coactivator inhibitors.

RESULTS AND DISCUSSION

Chemistry. The bioisostere hypothesis was tested by constructing a compound array containing all of the variations of the warhead and side chains that were reasonably potent in the MSNB background, with replacement of the ester by one of two thiazole linkages. This allowed any synergistic interactions affecting potency to adjust to the new core. The SNPT array was constructed using a parallel chemistry method with three diversification steps (Figure 2). First, six α -haloketones were employed to selectively give either the 4- or 5-carboxamides (x variation). Second, three thiols were used to provide sulfonyl group diversity (y variation). Finally, 24 amines were employed as the third building block (z variation). The amine series were chosen to systematically vary the size, electrostatics, and hydrophobicity at this position. All three sets of building blocks selected for this compound array are shown in Figure 3.

The synthetic route to the SNPTs is depicted in Scheme 1. Commercially available chloronitrobenzamide **4** was converted to benzamide **5** through oxidation with hydrogen peroxide. Benzamide **5** was then converted to thiobenzamide **6** with high yield by treatment with Lawesson's reagent. Intermediate **6** was treated with various 2-chlorooxoacetates **7** and 3-bromooxoacetate **12** to give the arrays of 5-carboxyesters **8** and 4-carboxyesters **13**{ x }, respectively. Next, compound

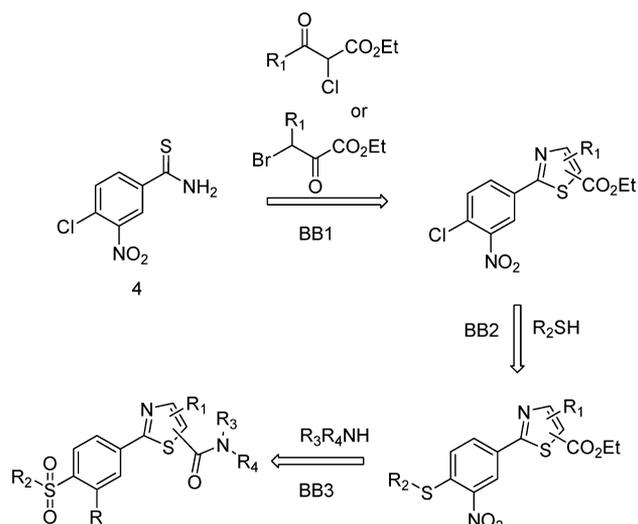


Figure 2. Plan for the construction of SNPT compound array.

arrays **8** and **13** were reacted with three thiolates (sodium methanethiolate, *n*-butanethiol/ K_2CO_3 , and benzylthiol/ K_2CO_3), yielding sulfide compound arrays **9**{ x,y } and **14**{ x,y }, respectively. Oxidation with *m*-CPBA gave the sulfonyl compound sets **10** and **15**, followed by hydrolysis to give 5-carboxylic acid SNPT array **11**{ x,y } and 4-carboxylic acid SNPT array **16**{ x,y }, respectively. This chemistry was performed cleanly enough to allow going from starting material **4** to the penultimate step without column purification, with all intermediates being purified by crystallization. In the last step carboxylic acids arrays **11** and **16** were treated with 24 amines, PyBOP, and DIEA at room temperature to give the final SNPT arrays (**2**{ x,y,z } and **3**{ x,y,z }). The final products were purified using flash chromatography followed by automated preparative HPLC. The identity of all compounds was established using NMR and MS. The purity was confirmed by LC/MS/UV/

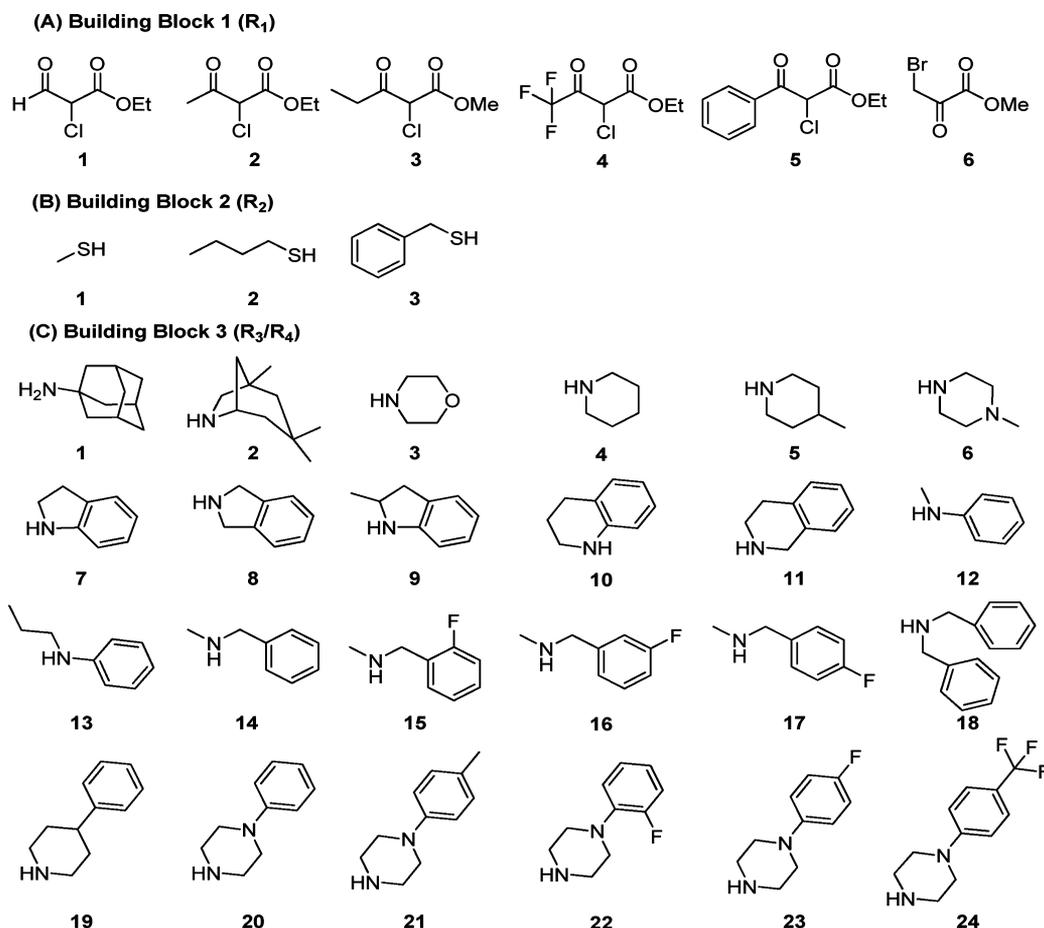


Figure 3. Building blocks (R_1 – R_3/R_4).

ELSD/CLND. The yield and purity of the SNPTs are available in Supporting Information.

Biology. All of the SNPTs were evaluated to test their TR-coactivator (COA) antagonism using a previously reported fluorescence polarization (FP) in replicate dose–response experiments using TR β -LBD and Texas Red labeled SRC2-2 peptide (Tx-SRC2-2).^{26,34} Compounds were serially diluted in 10 3-fold steps from a 10 000 μ M DMSO stock. The resulting concentration series of each SNPT were transferred to the assay wells using hydrodynamic pins with a final concentration of 0.1% DMSO. All assays were run in triplicate and the entire experiment was replicated twice, for a total of six replicates; the data are reported as average values across all assays as IC_{50} with 95% confidential range (Supporting Information). Fifty-two out of 291 SNPTs tested showed detectable inhibitory activity ($EC_{50} < 60 \mu$ M). Among the active analogues, 19 had IC_{50} values below 10 μ M (Table 1).

Comparing potency trends between classes of substituent at R_1 , R_2 , and R_3/R_4 allowed an initial analysis of structure–activity relationships (Figure 4). In general, 5-carboxamide-SNPTs $2\{x,y,z\}$ gave better inhibitory potency compared to 4-carboxamide-SNPTs $3\{x,y,z\}$. Among the 4-carboxamide-SNPTs $3\{x,y,z\}$ the most active compound was compound $3\{6,1,13\}$ with 8 μ M IC_{50} value. Among the 5-carboxamide-SNPTs $2\{x,y,z\}$, a wide range of substituents were tolerated at R_1 , but the phenyl substitutions $2\{5,y,z\}$ were mostly inactive, with the exception of $2\{5,1,4\}$ ($IC_{50} = 1 \mu$ M). The coactivator-binding site on TR is shallow; thus, it is likely that large phenyl groups are not favorable for binding. Remarkably CF_3

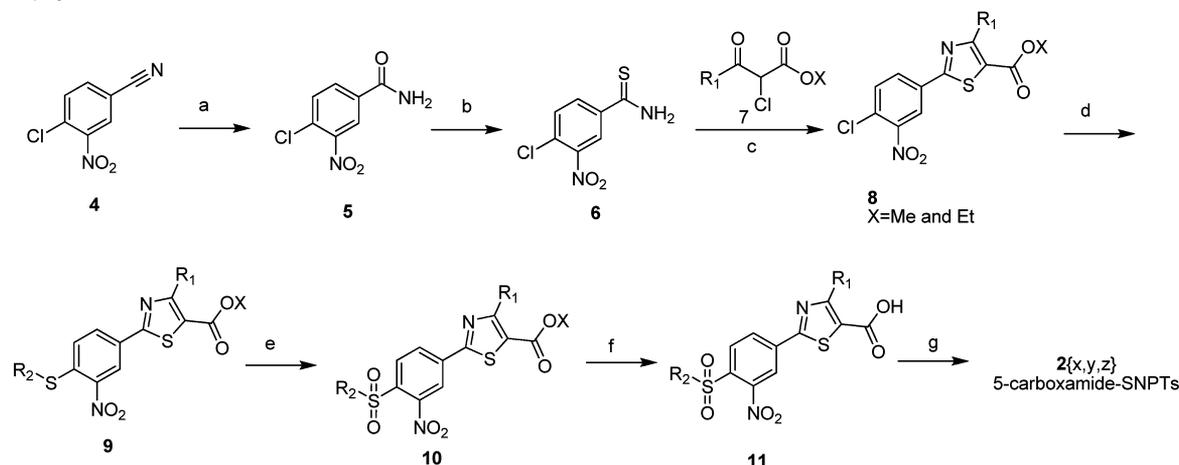
substituted compounds $2\{4,y,z\}$ showed slightly better TR–SRC2 antagonism (Figure 4A). Two compounds in this subseries, $2\{4,1,5\}$ and $2\{4,1,4\}$, showed the most potent activity (IC_{50} values of 0.3 and 0.6 μ M, respectively).

The potency was significantly less tolerant of variation in R_2 with most *n*-butyl- and benzyl-SNPTs exhibiting no inhibitory activity regardless of substitution on R_1 and R_3/R_4 . However, 55 out of 136 methyl-SNPTs ($R_2 = Me$) gave detectable inhibition with potency depending on the substitution pattern at R_1 and R_3/R_4 (Figure 4B). Only two compounds from the *n*-butyl and benzyl-SNPT subseries showed any activity (17 μ M for $2\{2,2,4\}$, 20 μ M for $2\{2,3,4\}$, 20 μ M for $2\{3,2,4\}$, and 19 μ M for $2\{3,3,4\}$ in Supporting Information).

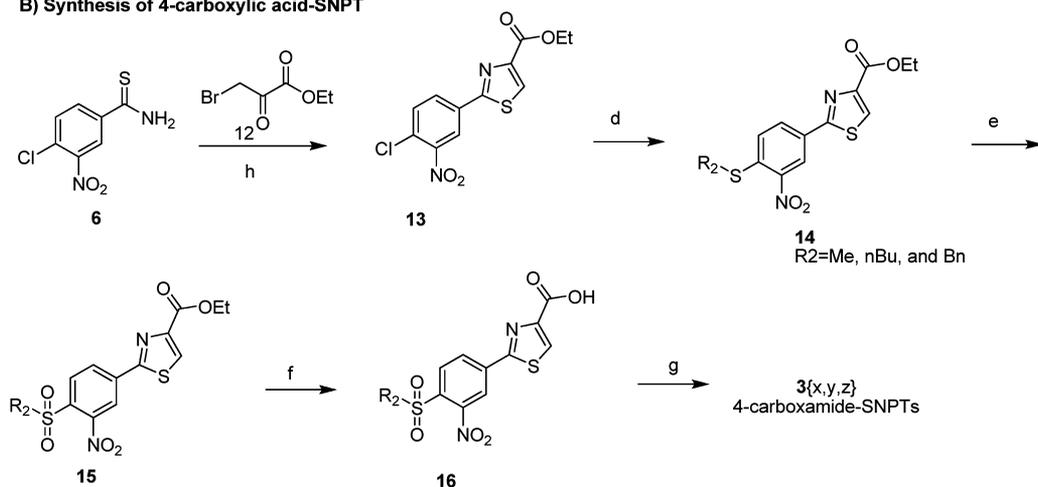
Finally, the binding pocket seems to be fairly tolerant of a wide range of substituents at R_3 , with both aliphatic amines and nonaliphatic amines (anilines, benzylamines, and piperazines) giving reasonable potency (Figure 4C). It is apparent that small amines are more favorable than large amines. For instance, the piperidine ($z = 4$) and 4-methylpiperidine ($z = 5$) substituted series showed generally good inhibitory activity while those series containing larger amines such as anilines, benzylamines, and phenylpiperazines did not. However, activity is not simply described by sterics, as hydrophilic aliphatic amines such as morpholines ($z = 3$) and 1-methylpiperazine ($z = 6$) showed weak or no inhibitory activity. This indicates that both electrostatic and hydrophobic interactions are important in this portion of the binding site. These results strongly support our previous finding that hydrophilic atoms on the amide group

Scheme 1. Synthesis of SNPT Analogues^a

A) Synthesis of 5-carboxamide-SNPT



B) Synthesis of 4-carboxylic acid-SNPT



^aReagents and conditions: (a) H₂O₂, K₂CO₃, DMSO, 60 °C, 0.5 h; (b) Lawesson's reagent, 1,4-dioxane, 110 °C, 2 h; (c) 2-chloro-2-ketoacetate 7, EtOH, reflux, 24–36 h; (d) NaSMe or RSH/K₂CO₃, THF, 50 °C, 18 h; (e) *m*-CPBA, DCM, 24–36 h, rt; (f) LiOH, H₂O/THF, rt, 3–5 h; (g) amines, PyBOP, DIEA, DMF, rt, 24 h; (h) bromooxobutanoate 12, EtOH, reflux, 24 h.

of the MSNB series and on side chain of β -aminoketone series were not favorable in this portion of the binding pocket.^{28,30}

MSNBs inhibit the TR–coactivator interaction through alkylation of the Cys 298 residue located on the AF-2 cleft by nucleophilic replacement with methylsulfonyl group.³⁰ Our results indicate that a bulky sulfonyl group hinders nucleophilic attack by this cysteine residue. In addition, larger alkyl groups do not appear to fit within the relatively shallow and small binding pocket on TR β surface, thus blocking reaction.

Next, the SNPT compound array was tested for the compounds' ability to inhibit T3-mediated transcription of a luciferase reporter gene assay. This was done using a single concentration of inhibitor (5 μ M), and the data are summarized in Table 1. Strikingly, biochemically highly potent compounds 2{4,1,5} and 2{4,1,4} had weak transcriptional inhibitory activity at 5 μ M (8.1% and 2.3% inhibition, respectively). In addition 2{5,1,4} and 2{1,1,1}, which exhibited good biochemical inhibitory activity (1.4 and 1.7 μ M IC₅₀), completely failed to inhibit T3-response luciferase expression. Instead biochemically moderately active com-

pounds 2{3,1,2} and 2{2,1,2} significantly inhibited T3-response luciferase activity.

To further validate transcriptional inhibition of T3-driven genes by SNPT compounds, we performed RT-PCR experiments using two well-accepted thyroid-responsive genes, PEPCK and MMP11, which are known to be T3-responsive in HepG2 cells (Figure 5).^{30,35} Cells were co-treated with T3 (100 nM) and compound (10 μ M). Controls included NH₃, a ligand antagonist of T3, and a representative MSNB (1), a known T3 antagonist. mRNA was isolated, and real-time PCR experiments were carried out on the diluted cDNA prepared from each mRNA sample. Both genes were inhibited by SNPTs, with efficacy matching that of the ligand antagonist NH₃. Compound 2{3,1,2} had slightly better inhibitory potency for both genes in comparison to other inhibitors. These data provide support that SNPT analogues effectively block TR-mediated gene transcription at native response elements in live cells.

Solubility and permeability of the compounds in the SNPT array were evaluated to elucidate likely relationships between biochemical assay and cellular assays. Compound solubility was

Table 1. Summary of Pharmacological Properties of SNPTs^c

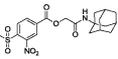
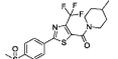
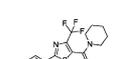
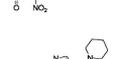
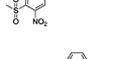
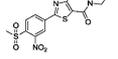
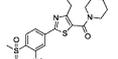
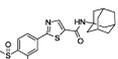
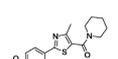
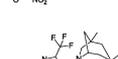
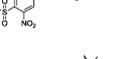
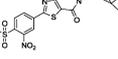
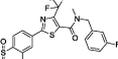
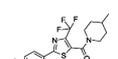
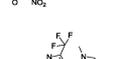
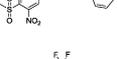
No	Structure	TRβ-SRC2-2 Inhibition, IC ₅₀ (μM, FP assay) ^a	TRE Response Inhibition (%, at 5 μM) ^a	Permeability ^b (×10 ⁻⁶ cm/s)	Solubility ^b (μM)	Cytotoxicity, HepG 2, EC ₅₀ (μM) ^b
1 (MSNB)		4.1±0.2	6.2±5.0	167±44	2.9±0.8	16.4±0.4
2{4,1,5}		0.31±0.17	8.1±0.6	1322±159	6.4±0.3	>27
2{4,1,4}		0.66±0.57	2.3±2.7	791±101	1.6±0.9	>27
2{1,1,4}		1.3±0.6	11.9±6.2	287±17	5.0±0.7	>27
2{5,1,4}		1.4±0.3	no inhibition	971±210	2.3±0.8	>27
2{3,1,4}		1.6±0.9	11.6±1.3	657±43	50.1±1.1	11.2
2{1,1,1}		1.7±0.9	no inhibition	379±39	0.7±0.1	>27
2{2,1,4}		1.8±0.6	9.8±6.0	280±42	18.0±0.9	>27
2{4,1,2}		2.1±1.6	15.4±5.2	8±1	1.0±0.8	>27
2{3,1,2}		2.4±1.1	42.1±5.9	949±195	2.8±0.5	10.1
2{4,1,16}		2.8±1.8	26.0±2.0	271±79	1.2±0.9	>27
2{4,1,5}		3.3±1.3	5.8±7.1	702±34	38.8±1.1	>27
2{4,1,14}		3.3±1.4	17.9±4.7	1499±536	1.2±0.7	>27
2{4,1,15}		3.4±2.5	24.7±3.3	221±21	0.8±0.1	>27
2{4,1,11}		4.2±2.9	33.4±1.4	84±17	0.2±0.5	>27
2{4,1,9}		5.2±2.7	17.5±6.8	43±6	0.4±0.4	>27

Table 1. continued

No	Structure	TR β -SRC2-2 Inhibition, IC ₅₀ (μ M, FP assay) ^a	TRE Response Inhibition (%, at 5 μ M) ^a	Permeability ^b ($\times 10^{-6}$ cm/s)	Solubility ^b (μ M)	Cytotoxicity, HepG 2, EC ₅₀ (μ M) ^b
2{2,1,2}		7.1 \pm 1.9	42.1 \pm 5.6	1416 \pm 41	5.7 \pm 1.0	>27
2{2,1,5}		7.4 \pm 3.4	8.1 \pm 5.2	912 \pm 205	6.7 \pm 0.4	>27
3{6,1,13}		8.2 \pm 2.0	14.6 \pm 2.8	1619 \pm 260	2.4 \pm 0.1	>27
2{1,1,16}		8.5 \pm 4.4	26.7 \pm 2.6	1031 \pm 288	2.4 \pm 0.4	>27

^aValues are the mean of two independent experiments in triplicate. ^bValues are the mean of a single triplicate experiment. ^cThe IC₅₀ values are for the inhibition of coregulatory peptide binding (SRC2-2) to the TR-LBD using a fluorescence polarization assay. The EC₅₀ values are for cellular proliferation inhibition (HepG2) using the measurement of total ATP content with the CellTiter-Glo (Promega) method. T3-response TRE-luciferase inhibitory activity was evaluated in HEK293 cells transfected with both CMV-TR β and a DR4-TRE driven luciferase expression vector. The data were normalized to co-transfected constitutive *Renilla* luciferase activity. Solubility was measured using the Millipore method at pH 7.4 in PBS. Permeability was measured using the parallel artificial membrane permeation assay (PAMPA) at pH 7.4. Compounds are ordered by potency of TR β and SRC2-2 inhibition.

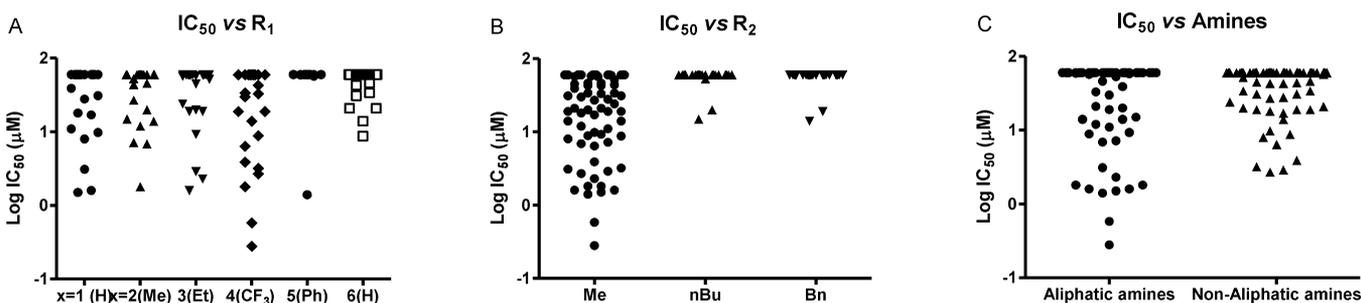


Figure 4. Analysis of SNPTs based on substitution position and class.

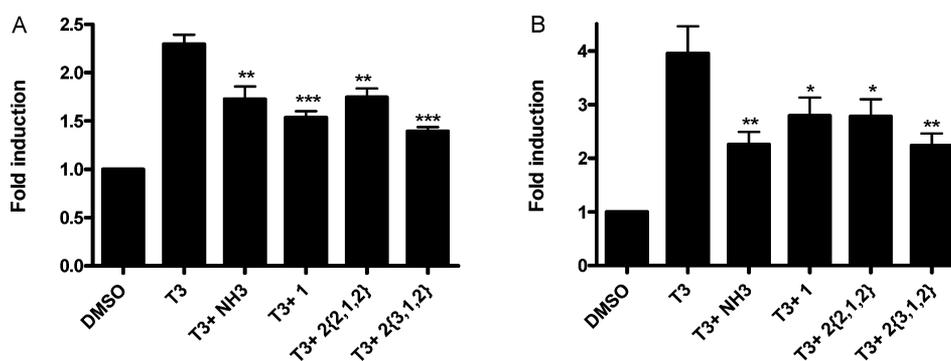


Figure 5. Regulation of native T3-controlled genes in HepG2 cells by treatment with SNPTs. The cells were exposed to compounds at a fixed concentration in the presence of T3 for 24 h. RT-PCR was carried out to determine transcription levels of the PEPCK (A) and MMP11 (B) genes. The $\Delta\Delta$ Ct method was used to calculate fold induction of expression. Error bars represent the standard errors of two independent experiments performed in triplicate: *, $P < 0.05$; **, $P < 0.01$; ***, $P < 0.005$.

determined in PBS buffer containing 1% DMSO, reflecting the conditions of the biochemical assays. SNPTs generally possessed relatively poor solubility (0.4–39 μ M), but most of the potent compounds showed reasonable solubility, being freely soluble at concentrations well above their biochemical potency. Permeability of compounds was measured using a

PAMPA at pH 7.4. All of the compounds showed acceptable to good permeability ($>40 \times 10^{-6}$ cm/s) except 2{4,1,2} (8×10^{-6} cm/s). No correlation was observed between cellular activity and solubility or permeability.

We also measured the cytotoxicity of selected inhibitors in HepG2, a hepatocellular carcinoma derived line. Most of

SNPTs showed no cytotoxic effects in the concentration range being evaluated ($EC_{50} > 27 \mu\text{M}$). Only three compounds, 2{3,1,4}, 2{3,1,2}, and 2{2,1,1}, showed weak cytotoxicity potency (11.2, 10.1, and 6.4 μM , respectively)

The association of TR β with SRC2-2 is ligand dependent.³⁶ Previously, we reported that MSNBs could not block the binding of T3 to TR β .³⁰ In order to confirm the suspected mechanism of action, we tested the antagonism of T3 binding by SNPTs using a ($[^{125}\text{I}]\text{T3}$) competition assay.³⁷ None of SNPTs blocked T3 binding at concentrations up to 60 μM (data not shown). Thus, SNPTs do not appear to act as T3 competitive antagonists.

We examined the NR specificity of the SNPT series by testing the following NR-coactivator interactions in FP assays: TR α with SRC2-2 and peroxisome proliferator-activated receptor γ (PPAR γ) with vitamin D receptor interacting protein 2 (DRIP-2). The coactivators were selected on the basis of previously published work mapping the preferred interaction partners.^{38,39} Additionally, PPAR γ was chosen because it contains a reactive cysteine in the ligand binding pocket that can be alkylated by electrophilic reagents and might potentially give poor selectivity. A set of 27 compounds was surveyed (Figure 6). In general, most the compounds were selective for the TRs but exhibited similar antagonistic potency toward both TR α and TR β . However, 2{4,1,4}, 2{4,1,2}, 2{4,1,3}, and 2{4,1,15}, which possess a 4- CF_3 at the 4-position, are more potent against TR α by at least 5-fold. Compounds 2{3,1,1}, 2{2,1,1}, and 2{1,1,2} were more potent against TR β by 4- to 5-fold, although they are only modestly potent (IC_{50} ranging from 11 to 15 μM). This set of compounds had almost no effect on PPAR γ .

When the aggregate of all the data is considered, the best-balanced two compounds are 2{2,1,2} and 2{3,1,2}. These two compounds showed both good thyroid hormone receptor coactivator interaction inhibitory potency and reasonable cellular toxicity. In addition these molecules significantly inhibited T3-mediated target gene expression in both luciferase based transcription assay and real-time PCR experiments. While their physicochemical properties are not ideal, they are both class II compounds⁴⁰ with high permeability and low solubility. Reasonable formulation strategies exist for handling class II compounds.⁴¹ Further optimization will likely focus on careful studies of the amine substitution pattern where the SAR is apparently subtle.

CONCLUSION

This paper describes the structural modification of the previously reported MSNB series of thyroid receptor antagonists using a bioisosteric approach to remove a potentially labile ester group linking the two critical pharmacophore elements for the inhibitors. In order to identify the best combination of linker and optimal versions of each portion of the molecule, the candidate array of inhibitors (SNPTs) was constructed using a parallel chemistry method with three diversification steps. Antagonism of SNPTs toward TR-coactivator binding was evaluated, revealing the most potent biochemical inhibitors of this interaction reported to date. Among 291 SNPT analogues tested, 60 compounds inhibited the interaction between TR β and SRC2-2 peptide. The cellular activity of SNPT analogues was explored using a TRE responsive luciferase reporter gene assay, demonstrating that a number of compounds were potent inhibitors of T3 induction of this element. This antagonism of TR-mediated T3

No	TR β	vs TR α	vs PPAR γ
	SRC2-2 (IC_{50} , μM)	SRC2-2 (IC_{50} , μM)	vs DRIP-2 (IC_{50} , μM)
2{4,1,5}	0.31 \pm 0.17	0.19 \pm 0.08	>60
2{4,1,4}	0.66 \pm 0.57	0.1 \pm 0.01	>60
2{1,1,4}	1.3 \pm 0.6	2.4 \pm 1.3	>60
2{5,1,4}	1.9 \pm 0.3	1.6 \pm 0.4	>60
2{3,1,4}	1.6 \pm 0.9	2.6 \pm 2.1	>60
2{1,1,1}	1.7 \pm 0.9	1.1 \pm 0.6	>60
2{2,1,4}	1.8 \pm 0.6	2.5 \pm 1.0	>60
2{4,1,2}	2.1 \pm 1.6	0.43 \pm 0.14	>60
2{3,1,2}	2.4 \pm 1.1	2.5 \pm 1.2	>60
2{4,1,16}	2.8 \pm 1.8	1.0 \pm 0.6	>60
2{4,1,5}	3.3 \pm 1.3	1.7 \pm 0.7	>60
2{4,1,14}	3.3 \pm 1.4	0.46 \pm 0.27	>60
2{4,1,15}	3.4 \pm 2.5	0.3 \pm 0.1	>60
2{4,1,11}	4.2 \pm 2.9	1.1 \pm 1.2	>60
2{4,1,9}	5.2 \pm 2.7	1.1 \pm 1.2	>60
2{2,1,2}	7.1 \pm 1.9	3.5 \pm 0.2	>60
2{2,1,5}	7.4 \pm 3.4	5.6 \pm 2.7	>60
3{6,1,13}	8.2 \pm 2.0	4.7 \pm 1.7	>60
2{1,1,16}	8.5 \pm 4.4	3.2 \pm 1.1	30.2 \pm 9.6

Figure 6. Biochemical selectivity of SNPT analogues in inhibiting coregulator binding to other NR family members. Compounds are ordered by potency against TR β . The coactivator and NR interactions tested were SRC2-2 with TR α , and DRIP-2 with PPAR γ . All values are the mean of two independent fluorescence polarization experiments, each carried out in triplicate.

signaling was confirmed on native response elements using RT-PCR. Interestingly, moderately active compounds in the FP assay significantly inhibited in the cellular level, whereas biochemically highly potent compounds showed weak inhibitory activity. While a number of potent cellular antagonists were identified, they tended not to be among the most potent from the biochemical assay, which is attributed to overly high reactivity of some of the compound array members leading to poor selectivity. The best two compounds 2{2,1,2} and 2{3,1,2} showed moderate inhibitory activity in biochemical assay but exhibited stronger inhibitory activity in cellular level. We successfully changed the metabolically unfavorable ester group of MSNBs to a thiazole group without any loss of activity in both biochemical assay and cellular assays. This result indicated that the SNPTs can be used as new tools for use in further TR biology studies. Currently we are studying their activity in *in vivo* models.

EXPERIMENTAL SECTION

Chemistry. All materials were obtained from commercial suppliers and used without further purification. All solvents used were dried using an aluminum oxide column. Thin layer chromatography was

performed on precoated silica gel 60 F254 plates. Purification of intermediates was carried out by normal phase column chromatography (SP1 [Biotage], silica gel 230–400 mesh). Chromatographic separation was performed using a UPLC–MS (BEH C18 1.7 μ m, 2.1 mm \times 50 mm column, Waters Corp.). Data were acquired using Masslynx, version 4.1, and analyzed using the Openlynx software suite. The flow was then split to an evaporative light scattering detector (ELSD) and an SQ mass spectrometer. The total flow rate was 1.0 mL/min. The gradient program started at 90% A (0.1% formic acid in H₂O), changed to 95% B (0.1% formic acid in ACN), then to 90% A. The mass spectrometer was operated in positive-ion mode with electrospray ionization. NMR spectra were recorded on a Bruker 400 MHz instrument and NMR peaks were assigned by MestReNova (version 5.2.2). The identity of all final compounds was confirmed by proton NMR and by mass spectrometry. The purity of all final compounds was assessed using LC/MS/UV/ELSD, with the purity (>95%) being assigned as the average determined by UV/ELSD (see Supporting Information for details).

4-Chloro-3-nitrobenzamide (5). To a solution of 4-chloro-3-nitrobenzitrile **4** (30 g, 164 mmol) and potassium carbonate (27.3 g, 197 mmol) in DMSO (400 mL) was cautiously added hydrogen peroxide (27.9 mL, 30% aqueous solution). The reaction mixture was heated for 15 min at 60 °C, then cooled to room temperature. The reaction mixture was poured to ice–water to afford a precipitate. Then the precipitate was washed with water to give the desired product **5** (25 g, 76%). ¹H NMR (400 MHz, DMSO) δ 8.52 (d, J = 2.1 Hz, 1H), 8.27 (d, J = 24.5 Hz, 1H), 8.16 (dd, J = 8.4, 2.1 Hz, 1H), 7.92–7.87 (m, 1H), 7.78 (s, 1H); ¹³C NMR (101 MHz, DMSO) δ 164.81, 147.31, 134.37, 132.52, 131.90, 127.85, 124.65.

4-Chloro-3-nitrobenzothioamide (6). To a solution of compound **5** (25 g, 125 mmol) in 1,4-dioxane (300 mL) was added Lawesson's reagent (25.2 g, 62.3 mmol). The reaction mixture was stirred for 2 h at 110 °C. The reaction mixture was concentrated in vacuo, then diluted with ethyl acetate, and washed with water. The organic solution was concentrated and the resulting solid product crystallized from dichloromethane to afford a yellow solid product (25 g, 93%). ¹H NMR (400 MHz, CDCl₃) δ 8.35 (d, J = 2.2 Hz, 1H), 8.04 (dd, J = 8.4, 2.2 Hz, 1H), 7.66 (br, 1H), 7.61 (d, J = 8.4 Hz, 1H), 7.19 (br, 1H); ¹³C NMR (101 MHz, CDCl₃) δ 198.17, 138.54, 132.14, 131.31, 130.50, 123.56.

Ethyl 2-(4-Chloro-3-nitrophenyl)-4-methylthiazole-5-carboxylate (8{2}). To a solution of compound **6** (3 g, 13.9 mmol) in EtOH (40 mL) was added ethyl 2-chloro-3-oxobutanoate (2.9 mL, 20.8 mmol). The reaction mixture was stirred overnight at 110 °C and then cooled to room temperature. The resulting precipitate was isolated by filtration and washed with MeOH to give a pure, white solid product (3.5 g, 77%). ¹H NMR (400 MHz, CDCl₃) δ 8.48 (d, J = 2.1 Hz, 1H), 8.07 (dd, J = 8.4, 2.1 Hz, 1H), 7.66–7.56 (m, 1H), 4.38 (q, J = 7.1 Hz, 2H), 2.79 (s, 3H), 1.40 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 165.26, 161.77, 161.40, 148.40, 132.88, 132.63, 130.53, 129.00, 123.54, 123.44, 61.65, 17.48, 14.33.

Ethyl 4-Methyl-2-(4-(methylthio)-3-nitrophenyl)thiazole-5-carboxylate (9{2,1}). To a solution of compound **8{2}** (0.5 g, 1.5 mmol) in THF (5 mL) was added sodium methane thiolate (0.13 g, 1.8 mmol). The reaction mixture stirred for 1 h at room temperature. The resulting precipitate was filtered and washed with THF and ethyl acetate to afford a yellow solid product (0.5 g, 97%). ¹H NMR (400 MHz, CDCl₃) δ 8.83 (d, J = 2.0 Hz, 1H), 8.16 (dd, J = 8.5, 2.0 Hz, 1H), 7.45 (d, J = 8.6 Hz, 1H), 4.37 (q, J = 7.1 Hz, 2H), 2.79 (s, 3H), 2.56 (s, 3H), 1.40 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 166.45, 161.97, 161.33, 145.54, 142.27, 130.91, 129.48, 126.22, 124.11, 122.61, 61.51, 17.53, 16.16, 14.36.

Ethyl 4-Methyl-2-(4-(methylsulfonyl)-3-nitrophenyl)thiazole-5-carboxylate (10{2,1}). To a solution of compound **9{2,1}** (0.27 g, 0.8 mmol) in DCM was added *m*-chloroperoxybenzoic acid (0.49 g, 2.0 mmol, <70% maximum activity). The reaction mixture was stirred for 6 h at room temperature and washed with water and saturated sodium bicarbonate. The organic solution was dried over MgSO₄ and concentrated in vacuo. The crude product was purified by recrystallization with DCM to give a white solid product

(0.27 g, 91%). ¹H NMR (400 MHz, CDCl₃) δ 8.39 (s, 1H), 8.24–8.18 (m, 2H), 4.32 (q, J = 7.1 Hz, 2H), 3.40 (s, 3H), 2.74 (s, 3H), 1.34 (t, J = 7.1 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 163.91, 161.77, 161.52, 149.71, 139.21, 134.99, 132.37, 129.87, 124.90, 122.70, 61.87, 45.18, 17.50, 14.32.

4-Methyl-2-(4-(methylsulfonyl)-3-nitrophenyl)oxazole-5-carboxylic Acid (11{1,1}). To a solution of compound **10{1,1}** (40 mg, 0.11 mmol) in H₂O/THF (4 mL, 1:3 v/v) was added lithium hydroxide (3.2 mg, 0.14 mmol). The reaction mixture was stirred for 3 h at room temperature and neutralized by 1 N HCl to pH 5–6, resulting in formation of a solid precipitate. The solid product was washed with water to give a pure, white solid product (20 mg, 54%). ¹H NMR (400 MHz, DMSO) δ 8.67 (d, J = 1.8 Hz, 1H), 8.58 (s, 1H), 8.55 (dd, J = 8.3, 1.8 Hz, 1H), 8.28–8.25 (m, 1H), 3.54 (s, 3H); ¹³C NMR (101 MHz, DMSO) δ 167.26, 161.67, 149.08, 148.86, 138.18, 133.88, 133.17, 132.34, 130.46, 122.44, 44.43.

Amides 2{x,y,z} and 3{x,y,z}. To a solution of carboxylic acid (50 mmol) in DMF (0.25 mL) in a 48-position Mettler Toledo XT reaction block were added PyBOP (50 mmol, 0.2 mL of 0.3 M solution in DMF) and TEA (75 mmol, 0.05 mL of 1.5 M solution in DMF) followed by the appropriate amine (55 mmol, 0.55 mL of 1 M solution in DMF). The mixtures were stirred at room temperature for 24 h and concentrated using a GeneVac HT-4. The crude product mixtures were dissolved in EtOAc (1 mL), filtered through a silica-packed column, and washed with EtOAc (2 \times 3 mL). The organic solutions were concentrated using a GeneVac HT-4 and dissolved in DMSO (1 mL). The crude products were purified by automated HPLC/MS. NMR and MS results of all final compounds are available in Supporting Information.

Compound Evaluation. All biological and pharmacological methods have been previously published and followed the established procedures. For details, see the Supporting Information.

■ ASSOCIATED CONTENT

● Supporting Information

Experimental procedures, tabulated activity data, and characterization data for intermediates and final compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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■ ABBREVIATIONS USED

MSNB, methylsulfonylnitrobenzoate; SNPT, sulfonylnitrophenylthiazole; NR, nuclear hormone receptor; TR, thyroid hormone receptor; RXR, retinoid X receptor; PPAR, peroxisome proliferator-activated receptor; AF, transcription activation; NID, nuclear receptor interaction domain; LBD, ligand binding domain; TRE, thyroid-responsive element; T₃, triiodothyronine; SRC, steroid receptor coactivator; COA, coactivator; PAMPA, parallel artificial membrane permeability assay; FP, fluorescence polarization; PDK, pyruvate dehydrogenase kinase; PEPCK, phosphoenolpyruvate carboxykinase; MMP11, matrix metalloproteinase 11

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