

The Asymmetric Syntheses of Methyl D-Digitoxoside, L-Oleandrose and L-Cymarose from Methyl Sorbate, an Achiral Precursor

Machiko Ono,^a Xi Ying Zhao,^b Keisuke Kato,^b and Hiroyuki Akita^{*b†}

^aSchool of Pharmaceutical Sciences, International University of Health and Welfare; 2600–1 Kitakanemaru, Ohtawara, Tochigi 324–8501, Japan; and ^bFaculty of Pharmaceutical Sciences, Toho University; 2–2–1 Miyama, Funabashi, Chiba 274–8510, Japan. Received April 5, 2012; accepted May 28, 2012

The addition of 4 eq of chloral to osmundalactone (4*S*,5*R*)-4 gave quantitative formation of the hemiacetal derivative (4*S*,5*R*)-8, which was treated with methane sulfonic acid to afford the intramolecular Michael addition product (+)-(3*S*,4*S*,5*R*)-9 possessing a 3,4-*cis*-dihydroxy- δ -lactone in 78% overall yield from (4*S*,5*R*)-4. The obtained (+)-(3*S*,4*S*,5*R*)-9 was subsequently converted to methyl D-digitoxoside (pyranoside) (12) in 13% overall yield and methyl D-digitoxoside (furanoside) (12) in 20% overall yield. The reaction of benzyl-osmundalactone (4*R*,5*S*)-3 and MeOH in the presence of Amberlyst A-26 as a basic catalyst gave 3,4-*trans*- δ -lactone (–)-(3*S*,4*R*,5*S*)-20 in 28% yield and 3,4-*cis*- δ -lactone (–)-(3*R*,4*R*,5*S*)-21 in 45% yield. Dibal-H reduction of (–)-(3*S*,4*R*,5*S*)-20 followed by catalytic hydrogenation gave L-oleandrose (6) in 86% overall yield, while Dibal-H reduction of (–)-(3*R*,4*R*,5*S*)-21 followed by catalytic hydrogenation provided L-cymarose (7) in 85% overall yield.

Key words osmundalactone; methyl D-digitoxoside; L-oleandrose; L-cymarose

We previously reported the syntheses of (4*S*,5*R*)- and (4*R*,5*S*)-4-benzyloxy-5-hydroxyhexen-2(*E*)-oates (2) based on a chemoenzymatic method from methyl sorbate (1).¹⁾ Hydrolysis of (4*S*,5*R*)- and (4*R*,5*S*)-2 gave the corresponding δ -hydroxy-*trans*- α,β -unsaturated carboxylic acids, which were converted to osmundalactones (4*S*,5*R*)- and (4*R*,5*S*)-4 via the formation of δ -lactones (4*S*,5*R*)- and (4*R*,5*S*)-3 accompanied by *trans*-*cis* isomerization, respectively.²⁾

Introduction of an oxygen functional group at C-3 position of (4*S*,5*R*)- and (4*R*,5*S*)-4 may enable the synthesis of D-gigitoxose (5) and L-oleandrose (6), L-cymarose (7), respectively,

possessing three contiguous chiral centers as shown in Chart 1. D-Digitoxose (5) is an important structural component of cardiac glycoside, digoxin,^{3–5)} while L-oleandrose (6) is a component of several antibiotics such as oleandomycin^{6–8)} and the avermectin series.^{9–11)} L-Cymarose (7) is a glycosidic component of a number of cardiac glycosides. Several syntheses of methyl D-digitoxoside (12) (Chart 2) and L-oleandrose (6) have been reported,¹²⁾ but the synthesis of L-cymarose (7) has scarcely been reported. Herein we report the asymmetric syntheses of methyl D-digitoxoside (12), L-oleandrose (6) and L-cymarose (7) from methyl sorbate, an achiral precursor.

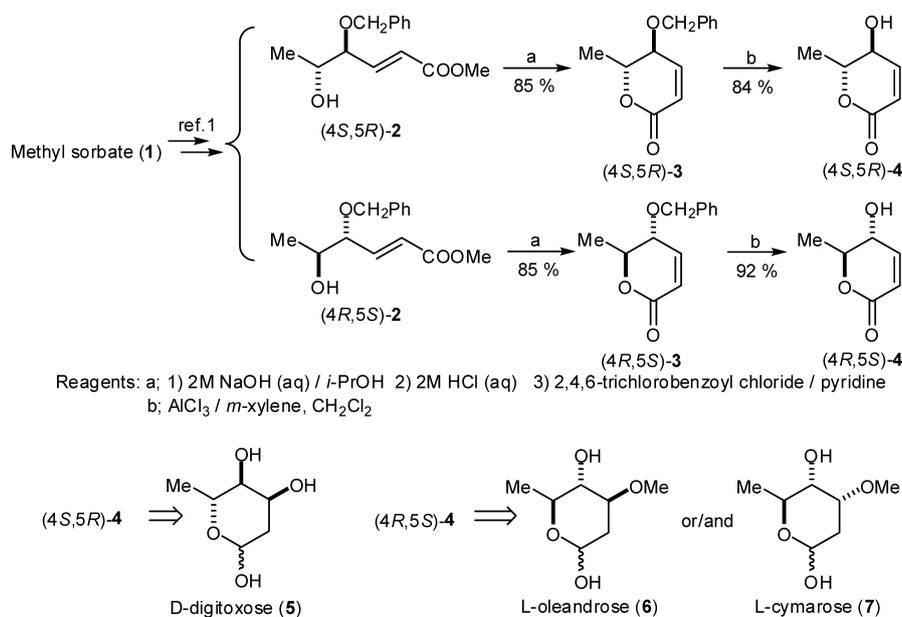


Chart 1

The authors declare no conflict of interest.

[†]Present address: Nihon Pharmaceutical University; 10281 Komuro, Inamachi, Kitaadachigun, Saitama 362–0806, Japan.

* To whom correspondence should be addressed. e-mail: hiroakita@nichiyaku.ac.jp

© 2012 The Pharmaceutical Society of Japan

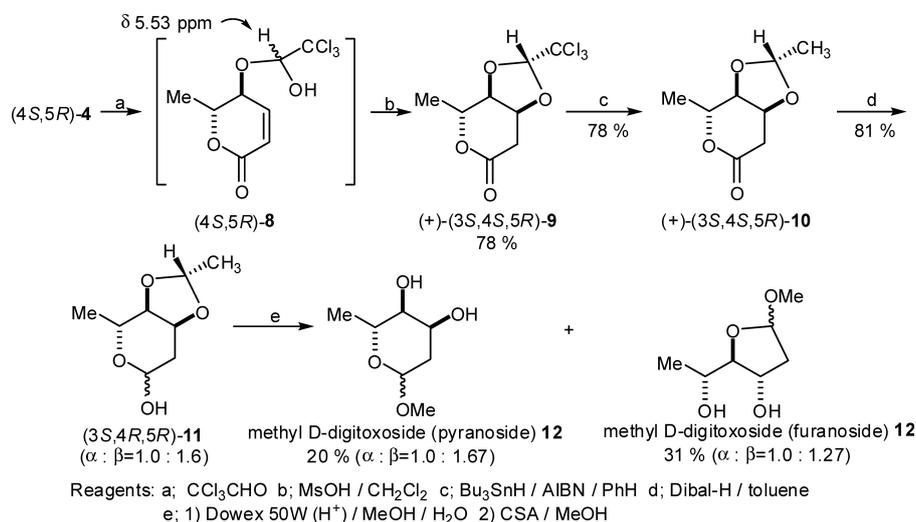
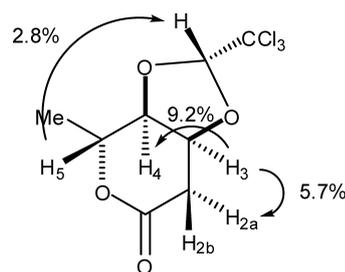


Chart 2

Synthesis of Methyl D-Digitoxoside (12) For the synthesis methyl D-digitoxoside (**12**), the construction of a 3,4-*cis*-dihydroxy- δ -lactone moiety is required from (4*S*,5*R*)-**4**. The oxymercuration–demercuration reaction of a trichloroacetaldehyde hemiacetal derivative derived from an unsaturated cyclic alcohol possessing an allylic alcohol moiety for a neighboring carbon–carbon double bond was reported to give an acetal possessing a 1,2-*cis*-dihydroxy-structure.¹³ Owing to its high tendency for formation of hemiacetal, trichloroacetaldehyde (chloral) was selected for this study. The addition of 4 eq of chloral to (4*S*,5*R*)-**4** resulted in quantitative formation of the hemiacetal derivative (4*S*,5*R*)-**8** as shown in Chart 2. Hemiacetal formation was easily monitored by observing the appearance of the characteristic singlet (δ 5.53 ppm) of the methine hydrogen attached to the trichloromethyl-bearing carbene in nuclear magnetic resonance (NMR) spectrum. To promote the Micheal addition of the hemiacetal hydroxyl group to the neighboring α,β -unsaturated double bond, methane sulfonic acid (MsOH) was added and the intramolecular Micheal addition product (+)-(3*S*,4*S*,5*R*)-**9** possessing the 3,4-*cis*-dihydroxy-structure was obtained in 78% overall yield from (4*S*,5*R*)-**4**. The structure of (+)-(3*S*,4*S*,5*R*)-**9** was confirmed by nuclear Overhauser effect (NOE) spectroscopy as shown in Fig. 1.

Treatment of (+)-(3*S*,4*S*,5*R*)-**9** with tributyltin hydride (Bu_3SnH) in the presence of azoisobutyronitrile (AIBN) gave an acetal (+)-(3*S*,4*S*,5*R*)-**10** in 78% yield. Treatment of (+)-(3*S*,4*S*,5*R*)-**10** with diisobutylaluminum hydride (Dibal-H) gave a 1.0:1.6 mixture of α - and β -epimers of lactol (3*S*,4*R*,5*R*)-**11** in 81% yield. Acid treatment of this mixture with Dowex 50W (H^+) in MeOH followed by addition of camphor sulfonic acid (CSA) in MeOH afforded a 1.0:1.67 mixture of α - and β -epimers of methyl D-digitoxoside (pyranoside)-**12** (20% yield) and a 1.0:1.27 mixture of α - and β -epimers of methyl D-digitoxoside (furanoside)-**12** (31% yield). $^1\text{H-NMR}$ data of both epimers of (pyranoside)-**12** and both epimers of (furanoside)-**12** were identical with those of the reported compounds.¹⁴ The $[\alpha]_D$ value of the 1.0:1.67 mixture of α - and β -epimers (pyranoside)-**12** $\{[\alpha]_D^{20} +35.4$ ($c=0.85$, CHCl_3) $\}$ were consistent with the calculated value $\{[\alpha]_D +41.7$ (CHCl_3) $\}$. The calculated value was obtained based on the

Fig. 1. NOE Correlation of (+)-(3*S*,4*S*,5*R*)-**9**

reported pure α -epimer (pyranoside)-**12** $\{[\alpha]_D^{20} +174$ ($c=1.0$, CHCl_3) $\}$ ¹⁴ and β -epimer (pyranoside)-**12** $\{[\alpha]_D^{20} -36$ ($c=1.0$, CHCl_3) $\}$ ¹⁴ using the following equation.

$$\begin{aligned} \text{calculated value } ([\alpha]_D) \\ = +174 \times 0.37 + (-36) \times 0.63 = +41.7 \end{aligned}$$

The $[\alpha]_D$ value of the 1.0:1.27 mixture of α - and β -epimers (furanoside)-**12** $\{[\alpha]_D^{25} +1.4$ ($c=0.44$, CHCl_3) $\}$ was consistent with the calculated one $\{[\alpha]_D +2.24$ (CHCl_3) $\}$. The calculated value was obtained based on the reported pure α -isomer (furanoside)-**12** $\{[\alpha]_D^{20} +140$ ($c=1.0$, CHCl_3) $\}$ ¹⁴ and β -isomer (furanoside)-**12** $\{[\alpha]_D^{20} -106$ ($c=1.0$, CHCl_3) $\}$ ¹⁴ as follows.

$$\begin{aligned} \text{calculated value } ([\alpha]_D) \\ = +140 \times 0.44 + (-106) \times 0.56 = +2.24 \end{aligned}$$

Thus the structure of the synthetic methyl D-digitoxoside (**12**) was unequivocally confirmed.

Synthesis of L-Oleandrose (6) and L-Cymarose (7) For the synthesis of L-oleandrose (**6**) and L-cymarose (**7**), the construction of a 3,4-*trans*-dihydroxy- δ -lactone moiety is required from (4*R*,5*S*)-**4**. The reaction of (\pm)-**4** and mercury(II) trifluoroacetate [$\text{Hg}(\text{OCOCF}_3)_2$] followed by addition of an oxygen nucleophile such as water or methanol could give compound (\pm)-**14**, which could be reduced with sodium borohydride (NaBH_4) to afford compound (\pm)-**15** possessing the 3,4-*trans*-dihydroxy- δ -lactone moiety as shown in Chart 3. In this case, the structure of intermediary (\pm)-**13** could be either 2,3-*cis*- and 3,4-*cis* because of the chelation between hydroxyl group and mercury ion.

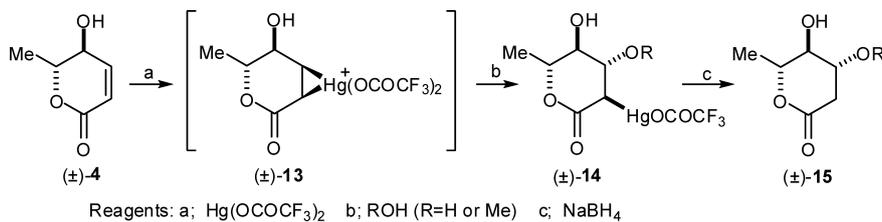


Chart 3

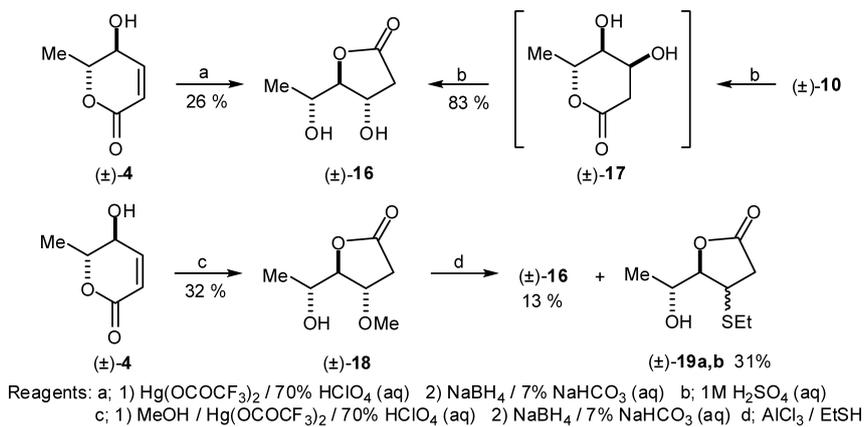


Chart 4

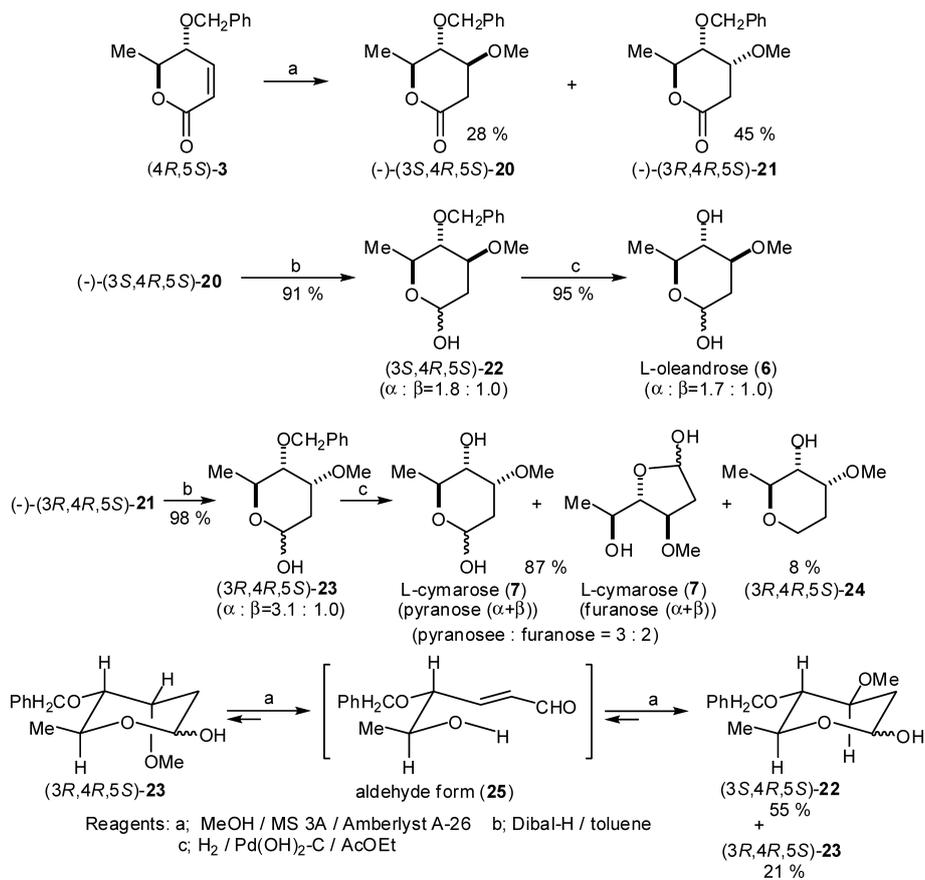


Chart 5

In order to examine this working hypothesis, the reaction of (\pm)-**4** and $\text{Hg}(\text{OCOCF}_3)_2$ in the presence of 70% HClO_4 solution followed by the reduction with alkaline NaBH_4 was carried out to afford compound (\pm)-**16** in 26% overall yield as shown in Chart 4. In this case, no other products were obtained.

The synthetic (\pm)-**16** was identical with the authentic (\pm)-**16**, which was obtained by treatment of previously mentioned (\pm)-**10** (Chart 2) with aqueous 1 M H_2SO_4 in 83% overall yield. The reaction of (\pm)-**4**, $\text{Hg}(\text{OCOCF}_3)_2$ and MeOH in the presence of 70% HClO_4 solution followed by the reduction with alkaline NaBH_4 was carried out to give compound (\pm)-**18** in 32% overall yield (Chart 4). In this case, no other products were obtained. AlCl_3 -assisted demethylation of (\pm)-**18** in ethane thiol (EtSH) afforded (\pm)-**16** (13%) and a mixture of α - and β -isomers of (\pm)-**19a, b** (31%). Consequently, the structure of (\pm)-**18** was confirmed. From these fact, the structure of intermediary (\pm)-**13** could be either 2,3-*cis*- and 3,4-*trans* because steric repulsion of between hydroxyl group and mercury ion could be larger than the above mentioned chelation effect. Consequently, water could attack at C(3)-position from β -side to give 3,4-*cis* (\pm)-**17**, which could be immediately converted to (\pm)-**16**.

Due to the low yield of (\pm)-**16** and (\pm)-**18**, the introduction of methoxyl group at the 3-position of the benzyl ether (4*R*,5*S*)-**3** via the Michael addition was examined. The reaction of (4*R*,5*S*)-**3** and MeOH in the presence of Amberlyst A-26 as a basic catalyst gave 3,4-*trans*- δ -lactone (-)-(3*S*,4*R*,5*S*)-**20** $\{[\alpha]_D^{22} -112$ ($c=1.73$, CHCl_3) $\}$ in 28% yield and 3,4-*cis*- δ -lactone (-)-(3*R*,4*R*,5*S*)-**21** $\{[\alpha]_D^{24} -72$ ($c=1.5$, CHCl_3) $\}$ in 45% yield. The structures of both products were confirmed by the fact that (-)-(3*S*,4*R*,5*S*)-**20** and (-)-(3*R*,4*R*,5*S*)-**21** were converted to natural products L-oleandrose (**6**) and L-cymarose (**7**), respectively, as shown in Chart 5.

Dibal-H reduction of (-)-(3*S*,4*R*,5*S*)-**20** provided a 1.8:1.0 mixture of α - and β -epimers (3*S*,4*R*,5*S*)-**22** $\{[\alpha]_D^{23} -73$ ($c=0.55$, CHCl_3) $\}$ in 91% yield, which was subjected to catalytic hydrogenation to give a 1.7:1.0 mixture of α - and β -L-oleandroses (**6**) in 95% yield. The specific rotation of the synthetic L-oleandrose (**6**) $\{[\alpha]_D^{23} +10$ ($c=0.51$, H_2O) $\}$ was consistent with the reported data $\{[\alpha]_D^{23} +11.2$ ($c=1.0$, H_2O) $\}$.⁹ $^1\text{H-NMR}$ data and $^{13}\text{C-NMR}$ data of the synthetic L-oleandrose (**6**) were identical with those of the reported data.¹⁵

Dibal-H reduction of (-)-(3*R*,4*R*,5*S*)-**21** provided a 3.1:1.0 mixture of α - and β -epimers (3*R*,4*R*,5*S*)-**23** $\{[\alpha]_D^{24} -87.7$ ($c=1.46$, CHCl_3) $\}$ in 98% yield, which was subjected to catalytic hydrogenation to give a 3:2 mixture of L-cymarose (pyranose) (**7**) and L-cymarose (furanose) (**7**) in 87% yield and (3*R*,4*R*,5*S*)-**24** in 8% yield. The specific rotation of the synthetic mixture L-cymarose (**7**) $\{[\alpha]_D^{24} -52$ ($c=0.42$, H_2O) $\}$ was consistent with the reported data $\{[\alpha]_D^{23} -50$ ($c=1.0$, H_2O) $\}$.¹⁶ $^1\text{H-NMR}$ data and $^{13}\text{C-NMR}$ data of the synthetic L-cymarose (**7**) were identical with those of the reported data.¹⁶

The reaction of the 3.1:1.0 mixture of α - and β -epimers (3*R*,4*R*,5*S*)-**23** and MeOH in the presence of Amberlyst A-26 gave a 1.8:1.0 mixture of α - and β -epimers (3*S*,4*R*,5*S*)-**22** in 55% yield and the recovery of (3*R*,4*R*,5*S*)-**23** in 21% yield. It was found that an equilibrium relationship was established between (3*R*,4*R*,5*S*)-**23** and (3*S*,4*R*,5*S*)-**22** via the aldehyde intermediate (**25**). This process could be explained in the following manner. Acid-promoted *trans* double bond and

aldehyde group formations based on MeOH elimination could afford α,β -unsaturated aldehyde **25**, which could be accompanied by Micheal addition of MeOH and hemiacetal formation to afford (3*S*,4*R*,5*S*)-**22** possessing more stable all equatorial configurations.

Conclusion

The addition of 4 eq of chloral to osmundalactone (4*S*,5*R*)-**4** gave quantitative formation of the hemiacetal derivative (4*S*,5*R*)-**8**, which was treated with methane sulfonic acid (MsOH) to afford the intramolecular Micheal addition product (+)-(3*S*,4*S*,5*R*)-**9** possessing a 3,4-*cis*-dihydroxy- δ -lactone moiety in 78% overall yield from (4*S*,5*R*)-**4**. Thus obtained (+)-(3*S*,4*S*,5*R*)-**9** was converted to acetal- δ -lactone (+)-(3*S*,4*S*,5*R*)-**10**, which was reduced with Dibal-H to provide lactol (3*S*,4*R*,5*R*)-**11** in 81% yield. Deprotection of the acetal in (3*S*,4*R*,5*R*)-**11** followed by acid treatment in MeOH gave methyl D-digitoxoside (pyranoside) (**12**) in 20% yield and methyl D-digitoxoside (furanoside) (**12**) in 31% yield.

The reaction of benzyl-osmundalactone (4*R*,5*S*)-**3** and MeOH in the presence of Amberlyst A-26 as a basic catalyst gave 3,4-*trans*- δ -lactone (-)-(3*S*,4*R*,5*S*)-**20** in 28% yield and 3,4-*cis*- δ -lactone (-)-(3*R*,4*R*,5*S*)-**21** in 45% yield. Dibal-H reduction of (-)-(3*S*,4*R*,5*S*)-**20** followed by catalytic hydrogenation gave L-oleandrose (**6**) in 86% overall yield, while Dibal-H reduction of (-)-(3*R*,4*R*,5*S*)-**21** followed by catalytic hydrogenation provided L-cymarose (**7**) in 85% overall yield.

Experimental

^1H - and ^{13}C -NMR spectra were recorded on JEOL AL 400 spectrometer in CDCl_3 . Carbon substitution degrees were established by distortionless enhancement by polarization transfer (DEPT) pulse sequence. The fast atom bombardment-mass spectra (FAB-MS) were obtained with a JEOL JMS 600H spectrometer. IR spectra were recorded with a JASCO FT/IR-300 spectrometer. Optical rotations were measured with a JASCO DIP-370 digital polarimeter. All evaporations were performed under reduced pressure. For column chromatography, silica gel (Kieselgel 60) was employed.

Synthesis of Methyl D-Digitoxoside (11) i) A mixture of (4*S*,5*R*)-**4** (0.601 g, 4.7 mmol) and CCl_3CHO (3.46 g 23.5 mmol) was stood for 24 h at rt and a solution of MsOH (1.36 g, 14.1 mmol) in CH_2Cl_2 (10 mL) was added to the above reaction mixture. All reaction mixture was stirred for 8 h at rt. The reaction mixture was diluted with Et_2O and the organic layer was washed with brine. The organic layer was dried over MgSO_4 and evaporated to give a crude oil, which was chromatographed on silica gel (35 g, *n*-hexane–AcOEt=1:3) to give (+)-(3*S*,4*S*,5*R*)-**9** (1.010 g, 78%) as a colorless oil. (+)-(3*S*,4*S*,5*R*)-**9**: IR (CHCl_3): 1755 cm^{-1} ; $[\alpha]_D^{30} +58.7$ ($c=0.34$, CHCl_3), $^1\text{H-NMR}$ δ : 1.51 (3H, d, $J=6$ Hz), 2.81 (1H, dd, $J=16$, 6 Hz), 3.02 (1H, dd, $J=16$, 6 Hz), 4.42 (1H, qd, $J=6$, 6 Hz), 4.52 (1H, dd, $J=6$, 6 Hz), 5.07 (1H, ddd, $J=6$, 6, 6 Hz), 5.55 (1H, s). *Anal.* Calcd for $\text{C}_8\text{H}_9\text{O}_4\text{Cl}_3$: C, 34.88; H, 3.27. Found: C, 34.59; H, 3.26. MS (FAB) m/z : 276 (M^++1).

ii) To a solution of (+)-(3*S*,4*S*,5*R*)-**9** (0.453 g, 1.64 mmol) and AIBN (70 mg, 0.43 mmol) in benzene (40 mL) under argon atmosphere were added *n*- Bu_3SnH (2.16 g, 7.42 mmol) and the reaction mixture was stirred for 2.5 h at 90°C. The reaction mixture was evaporated to give a residue, which was chromatographed on silica gel (55 g, *n*-hexane–AcOEt=1:3)

to give (+)-(3*S*,4*S*,5*R*)-**10** (0.222 g, 78%) as a colorless oil. (+)-(3*S*,4*S*,5*R*)-**10**: IR (neat): 1756 cm⁻¹; [α]_D²⁶ +105.9 (*c*=0.43, CHCl₃), ¹H-NMR δ : 1.34 (3H, d, *J*=5 Hz), 1.47 (3H, d, *J*=6 Hz), 2.72 (1H, dd, *J*=16, 8 Hz), 2.93 (1H, dd, *J*=16, 6 Hz), 4.08 (1H, dd, *J*=8.4, 8 Hz), 4.33 (1H, qd, *J*=8.4, 6 Hz), 4.59 (1H, ddd, *J*=8.4, 8, 6 Hz), 5.30 (1H, q, *J*=5 Hz). *Anal.* Calcd for C₈H₁₂O₄·0.25H₂O: C, 54.37; H, 7.08. Found: C, 54.49; H, 7.25. MS (FAB) *m/z*: 173 (M⁺+1).

iii) To a solution of (+)-(3*S*,4*S*,5*R*)-**10** (0.173 g, 1.0 mmol) in toluene (15 mL) was added 1 M Dibal-H toluene solution (2.7 mL, 2.7 mmol) under argon atmosphere at -20°C and the reaction mixture was stirred for 1 h at -20°C. The reaction mixture was diluted with 2 M HCl solution and extracted with Et₂O. The organic layer was washed with brine and dried over MgSO₄. Evaporation of the organic solvent gave a crude oil, which was chromatographed on silica gel (10 g, *n*-hexane-AcOEt=5:1) to give a 1:1.6 mixture of α - and β -epimers of (+)-(3*S*,4*R*,5*R*)-**11** (0.141 g, 81%) as a colorless oil. (+)-(3*S*,4*R*,5*R*)-**11**: IR (neat): 3432 cm⁻¹; [α]_D²⁴ +82.03 (*c*=0.79, CHCl₃), *Anal.* Calcd for C₈H₁₄O₄·0.25H₂O: C, 53.75; H, 8.12. Found: C, 53.76; H, 8.19. MS (FAB) *m/z*: 175 (M⁺+1). ¹H-NMR of each epimer was analyzed based on NOE and proton-proton decoupling (homo decoupling) analysis technique. α -Epimer δ : 1.26 (3H, d, *J*=6 Hz), 1.32 (3H, d, *J*=6 Hz), 1.80 (1H, ddd, *J*=14, 9, 4 Hz), 2.32 (1H, ddd, *J*=14, 4, 3 Hz), 3.15 (1H, d, *J*=5.8 Hz), 3.53 (1H, qd, *J*=9, 6 Hz), 3.78 (1H, dd, *J*=10, 4 Hz), 4.25-4.35 (1H, m), 5.03 (1H, ddd, *J*=9, 5.8, 2 Hz), 5.38 (1H, q, *J*=5.2 Hz). β -epimer δ : 1.26 (3H, d, *J*=6 Hz), 1.32 (3H, d, *J*=6 Hz), 2.06-2.20 (2H, m), 3.75 (1H, dd, *J*=10, 5 Hz), 3.97 (1H, qd, *J*=9, 6 Hz), 4.25-4.35 (1H, m), 5.15 (1H, ddd, *J*=8, 4, 4 Hz), 5.42 (1H, q, *J*=5.2 Hz).

iv) To a solution of a 1:1.6 mixture of α - and β -epimers of (+)-(3*S*,4*R*,5*R*)-**11** (0.107 g, 0.61 mmol) in MeOH (0.5 mL) was added H₂O (3.6 mL) and Dowex 50W(H+) (0.4 g), and the reaction mixture was stirred for 1.5 h at 70°C. The reaction mixture was filtered and the filtrate was evaporated to give a crude oil. To a solution of the above oil in MeOH (2 mL) was added CSA (0.12 g, 0.52 mmol) and the reaction mixture was stood for 1 h at rt. The reaction mixture was condensed to give a crude oil, which was chromatographed on silica gel (10 g, *n*-hexane-AcOEt=1:1) to give a 1:1.67 mixture of α - and β -epimers of (+)-methyl D-digitoxoside (**12**) (pyranoside, 0.020 g, 20%) as a colorless oil and a 1:1.27 mixture of α - and β -epimers of (+)-methyl D-digitoxoside (**12**) (furanoside, 0.031 g, 31%) as a colorless oil in elution order: D-digitoxoside (**12**) (pyranoside): IR (neat): 3445 cm⁻¹; [α]_D²⁴ +35.4 (*c*=0.85, CHCl₃), *Anal.* Calcd for C₇H₁₄O₄·0.25H₂O: C, 50.44; H, 8.77. Found: C, 50.65; H, 8.84. MS (FAB) *m/z*: 163 (M⁺+1). ¹H- and ¹³C-NMR of each epimer were assigned by comparison of the reported data. α -Epimer, ¹H-NMR δ : 1.31 (3H, d, *J*=6 Hz), 1.89 (1H, ddd, *J*=14.5, 3.4, 3.4 Hz), 2.15 (1H, ddd, *J*=14.5, 5.3, 1 Hz), 3.12 (1H, d, *J*=7 Hz), 3.35 (3H, s), 3.69 (1H, qd, *J*=10, 6 Hz), 3.92 (1H, brs), 4.75 (1H, d, *J*=3.4 Hz). ¹³C-NMR δ : 98.21 (d), 72.51 (d), 67.54 (d), 64.29 (d), 55.20 (s), 35.20 (t), 17.90 (s). β -Epimer δ : 1.29 (3H, d, *J*=6 Hz), 1.68 (1H, ddd, *J*=14, 10, 3 Hz), 2.08 (1H, ddd, *J*=14, 4, 2 Hz), 3.28 (1H, d, *J*=9 Hz), 3.45 (3H, s), 3.71 (1H, qd, *J*=9, 6 Hz), 4.08 (1H, dd, *J*=5, 3 Hz), 4.69 (1H, dd, *J*=10, 2 Hz). ¹³C-NMR δ : 98.71 (d), 73.00 (d), 69.52 (d), 68.00 (d), 56.49 (s), 37.70 (t), 18.22 (s). D-digitoxoside (**12**) (furanoside): IR (neat): 3445 cm⁻¹; [α]_D²⁴ +1.37 (*c*=0.44, CHCl₃), MS (FAB) *m/z*: 163 (M⁺+1). ¹H-NMR

and ¹³C-NMR of each epimer were assigned by comparison of the reported data. α -Epimer, ¹H-NMR δ : 1.22 (3H, d, *J*=6 Hz), 1.96 (1H, dd, *J*=14, 2 Hz), 2.07 (1H, ddd, *J*=14, 7, 2 Hz), 3.35 (3H, s), 3.80-3.90 (2H, m), 4.22 (1H, brs), 5.06 (1H, d, *J*=4 Hz). ¹³C-NMR δ : 105.20 (d), 90.80 (d), 71.16 (d), 67.64 (d), 54.80 (s), 42.05 (t), 18.72 (s). β -Epimer δ : 1.21 (3H, d, *J*=6 Hz), 2.04-2.14 (1H, m), 2.24 (1H, ddd, *J*=14, 7, 2 Hz), 3.34 (3H, s), 3.75-3.85 (2H, m), 4.56 (1H, brs), 5.04 (1H, d, *J*=6 Hz). ¹³C-NMR δ : 105.40 (d), 91.00 (d), 70.97 (d), 68.58 (d), 55.40 (s), 42.74 (t), 18.98 (s).

(±)-(3*S*',4*R*',5*R*')-3,5-Dihydroxyhexano-4-lactone (**16**)

To a solution of (±)-**4** (0.069 g, 0.54 mmol) in tetrahydrofuran (THF) (3 mL) at 0°C were added Hg(OCOCF₃)₂ (0.69 g, 1.62 mmol) and 70% HClO₄ solution (0.15 mL, 0.97 mmol), and the reaction mixture was stirred for 2 h at 0°C. To the reaction mixture were added 7% NaHCO₃ solution (7 mL) and NaBH₄ (0.06 g, 1.59 mmol) at 0°C, and the reaction mixture was stirred for 1 h. The reaction mixture was filtered and the filtrate was acidified with AcOH. The reaction mixture was condensed to afford a residue, which was chromatographed on silica gel (25 g, *n*-hexane-AcOEt=1:1) to give (±)-**16** (0.020 g, 26%) as a colorless oil. (±)-**16**: IR (neat): 1732, 3444 cm⁻¹; ¹H-NMR (CD₃OD) δ : 1.22 (3H, d, *J*=6 Hz), 2.34 (1H, dd, *J*=18, 2 Hz), 2.89 (1H, dd, *J*=18, 8 Hz), 3.89 (1H, qd, *J*=6, 4 Hz), 4.18 (1H, dd, *J*=4, 2 Hz), 4.50 (1H, dt, *J*=8, 2, 2 Hz). ¹³C-NMR (CD₃OD) δ : 178.33 (s), 93.27 (d), 67.89 (d), 67.63 (d), 39.27 (t), 19.02 (q). *Anal.* Calcd for C₆H₁₀O₄·0.25H₂O: C, 47.84; H, 7.03. Found: C, 47.56; H, 7.08. MS (FAB) *m/z*: 147 (M⁺+1).

Conversion of (±)-(3*S*',4*R*',5*R*')-**10** to (±)-(3*S*',4*R*',5*R*')-**16**

To a solution of (±)-**10** (0.166 g, 0.96 mmol) in THF (4 mL) was added 1 M H₂SO₄ solution (2.5 mL) and the reaction mixture was stirred for 2 h at 80°C. The reaction mixture was diluted with 7% NaHCO₃ solution (7 mL) and extracted with AcOEt. The organic layer was dried over Na₂SO₄ and evaporated to afford a residue which was chromatographed on silica gel (20 g, *n*-hexane-AcOEt=1:2) to give (±)-**16** (0.117 g, 83%) as a colorless oil. ¹H- and ¹³C-NMR data of (±)-**12** were identical with those of the previous (±)-**16**.

(±)-(3*S*',4*R*',5*R*')-5-Hydroxy-3-methoxyhexano-4-lactone (**18**)

To a solution of (±)-**4** (0.081 g, 0.63 mmol) in MeOH (5 mL) was added Hg(OCOCF₃)₂ (0.82 g, 1.92 mmol) and the reaction mixture was stirred for 2 h at rt. To the reaction mixture were added 7% NaHCO₃ solution (7 mL) and NaBH₄ (0.07 g, 1.85 mmol) at 0°C, and the reaction mixture was stirred for 1 h. The reaction mixture was worked up in the same way of (±)-**12** to give (±)-**18** (0.033 g, 32%) as a colorless oil. (±)-**18**: IR (neat): 1775, 3448 cm⁻¹; ¹H-NMR δ : 1.24 (3H, d, *J*=6 Hz), 2.50 (1H, dd, *J*=18, 2 Hz), 2.82 (1H, dd, *J*=18, 7 Hz), 3.30 (3H, s), 4.04 (1H, qd, *J*=6, 4 Hz), 4.12 (1H, dt, *J*=7, 3 Hz), 4.26 (1H, dd, *J*=4, 3 Hz). ¹³C-NMR δ : 175.85 (s), 88.56 (d), 75.76 (d), 66.74 (d), 56.51 (q), 35.40 (t), 18.34 (q). *Anal.* Calcd for C₇H₁₂O₄: C, 52.49; H, 7.55. Found: C, 52.51; H, 7.29. MS (FAB) *m/z*: 161 (M⁺+1).

Conversion of (±)-(3*S*',4*R*',5*R*')-**18** to (±)-(3*S*',4*R*',5*R*')-**16**

To a mixture of AlCl₃ (0.120 g, 0.9 mmol) in EtSH (2 mL, 0.03 mmol) was added a solution of (±)-**18** (0.047 g, 0.29 mmol) in EtSH (1 mL) at 0°C and the reaction mixture was stirred for 4 h at rt. The reaction mixture was diluted with H₂O and extracted with Et₂O. The organic layer was dried over Na₂SO₄ and evaporated to afford a residue which was chromatographed on silica gel (10 g) to give a 3:2 mixture

of (\pm)-**19a, b** (0.017 g, 31%) as a colorless oil from *n*-hexane–AcOEt=2:1 elution, recovery of (\pm)-**18** (0.020 g, 42%) from *n*-hexane–AcOEt=1:1 elution and (\pm)-**16** (0.006 g, 13%) from *n*-hexane–AcOEt=1:3 elution. ^1H - and ^{13}C -NMR data of (\pm)-**16** were identical with those of the previous (\pm)-**16**. (\pm)-**19a, b**: IR (neat): 1732, 3414 cm^{-1} ; Anal. Calcd for $\text{C}_8\text{H}_{14}\text{O}_3$: C, 50.50; H, 7.42. Found: C, 50.26; H, 7.35. MS (FAB) m/z : 191 ($\text{M}^+ + 1$). ^1H -NMR of each isomer was analyzed based on the proton–proton decoupling (homo decoupling) analysis technique. **19a**, ^1H -NMR (CD_3OD) δ : 1.21 (3H, d, $J=6$ Hz), 1.28 (3H, d, $J=7$ Hz), 2.44 (1H, dd, $J=18, 4$ Hz), 2.50–2.74 (2H, m), 3.09–3.19 (1H, m), 3.68 (1H, ddd, $J=9, 4, 4$ Hz), 3.94 (1H, qd, $J=6, 4$ Hz), 4.22 (1H, dd, $J=7, 3$ Hz). **19b**, ^1H -NMR (CD_3OD) δ : 1.25 (3H, d, $J=7$ Hz), 1.40 (3H, d, $J=6$ Hz), 2.50–2.74 (2H, m), 3.05–3.18 (3H, m), 3.35 (1H, dd, $J=9, 7$ Hz), 4.19 (1H, qd, $J=6, 5$ Hz).

Synthesis of L-Oleandrose (6) i) To a mixture of (4*R*,5*S*)-**3** (0.866 g, 3.97 mmol) and molecular sieve (MS) 3A (1.0 g) in MeOH (20 mL) was added Amberlyst A-26 (hydroxide form) (0.6 g) and the reaction mixture was stirred for 3 h at rt. The reaction mixture was filtered and the filtrate was condensed to give a crude oil. To a solution of the above oil in THF (20 mL) was added aqueous 2 M NaOH (6 mL) and the reaction mixture was stirred for 1 h at rt. The reaction mixture was acidified with aqueous 2 M HCl and condensed to give a crude product, which was diluted with AcOEt. The AcOEt layer was filtered and the filtrate was condensed to give a crude oil, which was chromatographed on silica gel (30 g) to give (–)-(3*S*,4*R*,5*S*)-**20** (0.284 g, 28%) as a colorless oil from *n*-hexane–AcOEt=5:1 eluent and (–)-(3*R*,4*R*,5*S*)-**21** (0.453 g, 45%) as a colorless oil from *n*-hexane–AcOEt=2:1 eluent. (–)-(3*S*,4*R*,5*S*)-**20**: $[\alpha]_{\text{D}}^{22} -111.5$ ($c=1.73$, CHCl_3), IR (neat): 1758 cm^{-1} ; ^1H -NMR δ : 1.41 (3H, d, $J=6$ Hz), 2.74 (1H, ddd, $J=16, 4, 1$ Hz), 2.79 (1H, dd, $J=16, 5$ Hz), 3.36 (3H, s), 3.39 (1H, ddd, $J=8, 3, 1$ Hz), 3.74 (1H, dd, $J=8, 4$ Hz), 4.19 (1H, qd, $J=6, 4$ Hz), 4.58, 4.72 (each 1H, d, $J=12$ Hz), 7.28–7.38 (5H, m). Anal. Calcd for $\text{C}_{14}\text{H}_{18}\text{O}_4$: C, 67.18; H, 7.25. Found: C, 66.98; H, 7.41. MS (FAB) m/z : 251 ($\text{M}^+ + 1$). (–)-(3*R*,4*R*,5*S*)-**21**: $[\alpha]_{\text{D}}^{24} -71.9$ ($c=1.5$, CHCl_3), IR (neat): 1733 cm^{-1} ; ^1H -NMR δ : 1.37 (3H, d, $J=6$ Hz), 2.58 (1H, dd, $J=18, 4$ Hz), 2.90 (1H, dd, $J=18, 5.6$ Hz), 3.41 (3H, s), 3.46 (1H, dd, $J=7, 2$ Hz), 3.75 (1H, ddd, $J=5.8, 4, 2$ Hz), 4.70 (1H, qd, $J=7, 6$ Hz), 4.61, 4.72 (each 1H, d, $J=12$ Hz), 7.26–7.38 (5H, m). Anal. Calcd for $\text{C}_{14}\text{H}_{18}\text{O}_4$: C, 67.18; H, 7.25. Found: C, 66.89; H, 7.31. MS (FAB) m/z : 251 ($\text{M}^+ + 1$).

ii) To a solution of (–)-(3*S*,4*R*,5*S*)-**20** (0.246 g, 0.98 mmol) in toluene (12 mL) was added 1 M Dibal-H toluene solution (2 mL, 2 mmol) under argon atmosphere at -20°C and the reaction mixture was stirred for 1 h at -20°C . The reaction mixture was diluted with 2 M HCl solution (2 mL) and extracted with Et_2O . The organic layer was washed with brine and dried over MgSO_4 . Evaporation of the organic solvent gave a crude oil, which was chromatographed on silica gel (10 g, *n*-hexane–AcOEt=2:1) to give a 1.8:1 mixture of α - and β -epimers of (–)-(3*S*,4*R*,5*S*)-**22** (0.226 g, 91%) as a colorless oil. (–)-(3*S*,4*R*,5*S*)-**22**: IR (neat): 3440 cm^{-1} ; $[\alpha]_{\text{D}}^{23} -72.5$ ($c=0.55$, CHCl_3), Anal. Calcd for $\text{C}_{14}\text{H}_{20}\text{O}_4$: C, 66.65; H, 7.99. Found: C, 66.57; H, 8.25. MS (FAB) m/z : 253 ($\text{M}^+ + 1$). ^1H -NMR of each epimer was analyzed based on NOE and proton–proton decoupling (homo decoupling) analysis technique. α -Epimer: ^1H -NMR δ : 1.25 (3H, d, $J=6$ Hz), 1.54 (1H, ddd, $J=14, 14, 3$ Hz), 2.27 (1H, ddd, $J=14, 5, 2$ Hz), 3.02 (1H, dd,

$J=10, 9$ Hz), 3.44 (3H, s), 3.72 (1H, ddd, $J=14, 9, 5$ Hz), 3.94 (1H, qd, $J=10, 6$ Hz), 4.63, 4.89 (each 1H, d, $J=11$ Hz), 5.31 (1H, brs), 7.23–7.40 (5H, m). β -Epimer: ^1H -NMR δ : 1.30 (3H, d, $J=6$ Hz), 1.43 (1H, ddd, $J=12, 12, 10$ Hz), 2.40 (1H, ddd, $J=12, 5, 2$ Hz), 3.00 (1H, dd, $J=6, 3$ Hz), 3.33–3.43 (2H, m), 3.42 (3H, s), 4.62, 4.88 (each 1H, d, $J=11$ Hz), 5.05 (1H, brs), 7.23–7.40 (5H, m).

iii) A mixture of (–)-(3*S*,4*R*,5*S*)-**22** (1.8:1 mixture of α - and β -epimers) (0.207 g, 0.82 mmol) and 20% $\text{Pd}(\text{OH})_2\text{-C}$ (0.16 g) in AcOEt (20 mL) was subjected to a catalytic hydrogenation under ordinary pressure for 12 h at rt. The reaction mixture was filtered and the filtrate was condensed to give a crude oil, which was chromatographed on silica gel (10 g, *n*-hexane–AcOEt=1:2) to give a 1.7:1 mixture of α - and β -L-oleandrose (**6**) (0.153 g, 95%) as a colorless oil. L-Oleandrose (**6**): IR (neat): 3443 cm^{-1} ; $[\alpha]_{\text{D}}^{23} +10.0$ ($c=0.51$, H_2O), Anal. Calcd for $\text{C}_7\text{H}_{14}\text{O}_4$: C, 51.84; H, 8.70. Found: C, 51.58; H, 8.98. MS (FAB) m/z : 145 ($\text{M}^+ + 1 - \text{H}_2\text{O}$). ^1H -NMR of each epimer was analyzed based on NOE and proton–proton decoupling (homo decoupling) analysis technique. α -Pyranose (**6**): ^1H -NMR δ : 1.26 (3H, d, $J=6$ Hz), 1.47 (1H, ddd, $J=14, 12, 4$ Hz), 2.28 (1H, ddd, $J=14, 4.4, 1.6$ Hz), 3.13 (1H, dd, $J=9, 9$ Hz), 3.37 (3H, s), 3.55 (1H, ddd, $J=14, 9, 5$ Hz), 3.91 (1H, qd, $J=9, 6$ Hz), 5.32 (1H, brs, $J=3$ Hz). ^{13}C -NMR δ : 91.89 (d), 77.85 (d), 76.10 (d), 67.65 (d), 56.45 (q), 34.15 (t), 18.00 (q). β -Pyranose (**6**): ^1H -NMR δ : 1.32 (3H, d, $J=6$ Hz), 1.35–1.37 (1H, m), 2.39 (1H, ddd, $J=12, 4, 2$ Hz), 3.08–3.44 (3H, m), 3.35 (3H, s), 4.79 (1H, dd, $J=10, 2$ Hz). ^{13}C -NMR δ : 93.90 (d), 80.50 (d), 75.15 (d), 71.75 (d), 56.38 (q), 36.50 (t), 17.95 (q).

Synthesis of L-Cymarose (7) i) To a solution of (–)-(3*R*,4*R*,5*S*)-**21** (0.313 g, 1.25 mmol) in toluene (13 mL) was added 1 M Dibal-H toluene solution (2.5 mL, 2.5 mmol) under argon atmosphere at -20°C and the reaction mixture was stirred for 1 h at -20°C . The reaction mixture was diluted with 2 M HCl solution (2.5 mL) and extracted with Et_2O . The organic layer was washed with brine and dried over MgSO_4 . Evaporation of the organic solvent gave a crude oil, which was chromatographed on silica gel (15 g, *n*-hexane–AcOEt=2:1) to give a 3.1:1 mixture of α - and β -epimers of (–)-(3*R*,4*R*,5*S*)-**23** (0.308 g, 98%) as a colorless oil. (–)-(3*R*,4*R*,5*S*)-**23**: IR (neat): 3420 cm^{-1} ; $[\alpha]_{\text{D}}^{24} -87.7$ ($c=1.46$, CHCl_3), Anal. Calcd for $\text{C}_{14}\text{H}_{20}\text{O}_4 \cdot 0.25\text{H}_2\text{O}$: C, 65.42; H, 7.98. Found: C, 65.22; H, 7.94. MS (FAB) m/z : 253 ($\text{M}^+ + 1$). ^1H -NMR of each epimer was analyzed based on NOE and proton–proton decoupling (homo decoupling) analysis technique. α -Epimer: ^1H -NMR δ : 1.29 (3H, d, $J=6$ Hz), 1.70 (1H, ddd, $J=14, 6, 2$ Hz), 2.19 (1H, ddd, $J=14, 4, 1.8$ Hz), 3.10 (1H, dd, $J=10, 3$ Hz), 3.52 (3H, s), 3.83 (1H, brs), 4.21 (1H, qd, $J=10, 6$ Hz), 4.55, 4.65 (each 1H, d, $J=12$ Hz), 5.17 (1H, brs), 7.24–7.40 (5H, m). β -Epimer: ^1H -NMR δ : 1.26 (3H, d, $J=6$ Hz), 1.42 (1H, ddd, $J=14, 10, 2$ Hz), 2.27 (1H, ddd, $J=14, 4, 2$ Hz), 3.10 (1H, dd, $J=10, 3$ Hz), 3.42 (3H, s), 3.72 (1H, brs), 3.96 (1H, qd, $J=10, 6$ Hz), 4.51, 4.63 (each 1H, d, $J=12$ Hz), 5.05 (1H, brs), 7.24–7.40 (5H, m).

ii) A mixture of (–)-(3*R*,4*R*,5*S*)-**23** (3.1:1 mixture of α - and β -epimers) (0.142 g, 0.56 mmol) and 20% $\text{Pd}(\text{OH})_2\text{-C}$ (0.16 g) in AcOEt (20 mL) was subjected to a catalytic hydrogenation under ordinary pressure for 12 h at rt. The reaction mixture was filtered and the filtrate was condensed to give a crude oil, which was chromatographed on silica gel (10 g) to give (3*R*,4*R*,5*S*)-**24** (0.007 g, 8%) as a colorless oil from *n*-hexane–AcOEt=1:1 eluent and a 3:2 mixture of pyranose type and

furanose type of L-cymarose (**7**) (0.079 g, 87%) as a colorless oil from *n*-hexane–AcOEt=1:2 eluent. (3*R*,4*R*,5*S*)-**24**: IR (neat): 3450 cm⁻¹; *Anal.* Calcd for C₇H₁₄O₃: C, 57.51; H, 9.65. Found: C, 57.23; H, 9.73. MS (FAB) *m/z*: 147 (M⁺+1). ¹H-NMR δ: 1.18 (3H, d, *J*=6 Hz), 1.80–1.97 (2H, m), 2.22 (1H, brs), 3.29 (3H, s), 3.58 (1H, dd, *J*=4, 4 Hz), 3.78–3.98 (3H, m). ¹³C-NMR δ: 88.09 (d), 80.85 (d), 67.50 (d), 67.35 (t), 56.80 (q), 32.65 (t), 18.60 (q). L-Cymarose (**7**) IR (neat): 3418 cm⁻¹; [α]_D²³ –51.6 (*c*=0.42, H₂O), *Anal.* Calcd for C₇H₁₄O₄: C, 51.84; H, 8.70. Found: C, 51.54; H, 8.87. MS (FAB) *m/z*: 145 (M⁺+1–H₂O). ¹H-NMR of each epimer was analyzed based on NOE and proton–proton decoupling (homo decoupling) analysis technique. α-Pyranose (**7**): ¹H-NMR δ: 1.17 (3H, d, *J*=6 Hz), 1.73 (1H, ddd, *J*=15, 4, 3 Hz), 2.02–2.17 (1H, m), 3.32 (3H, s), 3.33–3.43 (1H, m), 3.57–3.66 (1H, m), 3.89–3.90 (1H, m), 5.06 (1H, brs). β-Pyranose (**7**): ¹H-NMR δ: 1.25 (3H, d, *J*=6 Hz), 1.49 (1H, ddd, *J*=14, 10, 3 Hz), 2.26 (1H, ddd, *J*=14, 4, 2 Hz), 3.19 (1H, ddd, *J*=10, 10, 3 Hz), 3.39 (3H, s), 3.57–3.66 (1H, m), 3.86–3.98 (1H, m), 4.97 (1H, dd, *J*=10, 2 Hz). A mixture of furanose (**7**): ¹H-NMR δ: 1.19, 1.25 (each 3H, d, *J*=6 Hz), 1.98, 2.30 (each 1H, ddd, *J*=15, 4, 3 Hz and ddd, *J*=14, 4, 2 Hz), 2.18, 2.30 (each 1H, ddd, *J*=14, 7, 2 Hz and ddd, *J*=14, 4, 2 Hz), 3.26, 3.49 (each 3H, s), 4.04, 4.13 (each 1H, dd, *J*=4, 2 Hz and ddd, *J*=8, 5, 4 Hz), 5.44, 5.56 (each 1H, d, *J*=5 Hz and dd, *J*=6, 2.8 Hz).

Conversion of (3*R*,4*R*,5*S*)-23** to (3*S*,4*R*,5*S*)-**22**** To a solution of a mixture of (–)-(3*R*,4*R*,5*S*)-**23** (3.1:1 mixture of α- and β-epimers) (0.035 g, 0.14 mmol) in MeOH (2 mL) was added Amberlyst A-26 (hydroxide form) (0.1 g) and the reaction mixture was stirred for 2 h at 35°C. The reaction mixture was filtered and the filtrate was condensed to give a crude oil, which was chromatographed on silica gel (10 g, *n*-hexane–AcOEt=2:1) to give a 1.7:1 mixture of α- and β-epimers of (3*S*,4*R*,5*S*)-**22** (0.019 g, 55%) as a colorless oil and starting

(3*R*,4*R*,5*S*)-**23** (0.007 g, 21%) as a colorless oil. ¹H-NMR data of both compounds were identical with those of the previous (3*S*,4*R*,5*S*)-**22** and (3*R*,4*R*,5*S*)-**23**, respectively.

References

- 1) Ono M., Saotome C., Akita H., *Tetrahedron Asymmetry*, **7**, 2595–2602 (1996).
- 2) Ono M., Zhao X. Y., Shida Y., Akita H., *Tetrahedron*, **63**, 10140–10148 (2007).
- 3) Reichstein T., *Angew. Chem.*, **63**, 412–421 (1951).
- 4) Tschesche R., Wirtz S., Snatzke G., *Chem. Ber.*, **88**, 1619–1624 (1955).
- 5) Lichti H., Tamm C., Reichstein T., *Helv. Chim. Acta*, **39**, 1914–1932 (1956).
- 6) Jäger H., Schindler O., Reichstein T., *Helv. Chim. Acta*, **42**, 977–1013 (1959).
- 7) Korte F., Rippahhn J., *Liebigs Ann. Chem.*, **621**, 58–71 (1959).
- 8) Hochstein F. A., Els H., Celmer W. D., Shapiro B. L., Woodward R. B., *J. Am. Chem. Soc.*, **82**, 3225–3227 (1960).
- 9) Tsukamoto S., Hayashi K., Mitsunashi H., *Chem. Pharm. Bull.*, **33**, 2294–2304 (1985).
- 10) Tsukamoto S., Hayashi K., Mitsunashi H., Snykers F. O., Fourie T. G., *Chem. Pharm. Bull.*, **33**, 4807–4814 (1985).
- 11) Tsukamoto S., Hayashi K., Mitsunashi H., *Tetrahedron*, **41**, 927–934 (1985).
- 12) Ley S. V., Armstrong A., Diez-Martin D., Ford M. J., Grice P., Knight J. G., Kolb H. C., Madin A., Marby C. A., Mukherjee S., Shaw A. N., Slawin A. M. Z., Vile S., White A. D., Williams D., Woods M., *J. Chem. Soc., Perkin Trans. I*, **1991**, 667–692 (1991).
- 13) Overman L. E., Campbell C. B., *J. Org. Chem.*, **39**, 1474–1481 (1974).
- 14) Zeeck A., *Liebigs Ann. Chem.*, **11**, 2079–2088 (1975).
- 15) Bredenkamp M. W., Holzapfel C. W., Toerien F., *Synth. Commun.*, **22**, 2459–2477 (1992).
- 16) Nakagawa T., Hayashi K., Wada K., Mitsunashi H., *Tetrahedron*, **39**, 607–612 (1983).