



Cite this: *Org. Biomol. Chem.*, 2019, **17**, 10228

Received 17th October 2019,
Accepted 20th November 2019

DOI: 10.1039/c9ob02253h

rsc.li/obc

A trimethyllysine-containing trityl tag for solubilizing hydrophobic peptides†

Shun Masuda, Shugo Tsuda and Taku Yoshiya *

Hydrophobic membrane peptides/proteins having low water solubility are often difficult to prepare. To overcome this issue, temporal introduction of solubilizing tags has been demonstrated to be beneficial. Following our recent work on the solubilization of a difficult target by using a hydrophilic oligo-Lys tag bearing a trityl linker (Trt-K method), this paper describes a comparative study of the solubilizing abilities of several peptidic trityl tags containing Lys, Arg, Glu, Asn, *N*^ε-tri-Me-Lys or Cys-sulfonate using two hydrophobic model peptides. Among the tags evaluated, that containing *N*^ε-tri-Me-Lys exhibits superior solubilizing ability.

Introduction

Chemical peptide/protein synthesis mainly consists of two steps: peptide chain construction and purification using high performance liquid chromatography (HPLC). The first step is well established and based on methods such as solid-phase peptide synthesis (SPPS),¹ native chemical ligation (NCL),² and related techniques.³ In contrast, there is still room for improvement in the second step, since hydrophobic membrane peptides/proteins having low water solubility are difficult to purify by HPLC.⁴ Additionally, the low solubility of these compounds also hampers further development of NCL processes in which unprotected peptide segments are ligated in an aqueous solvent.⁵ To solve this problem, many researchers developed solubilizing tags such as a phenylacetamidomethyl linker method,^{6g} a removable backbone modification method,^{6h} and a canaline tag.^{6m} In this context, we recently developed a simple solubilizing trityl-type tag system (Trt-K, where Trt and K denote trityl and Lys, respectively) (Fig. 1),⁷ which is charac-

terized by the following features: (1) an introducing reagent for the trityl alcohol-type tag, “Trt(OH)-K_n”, can be readily prepared using commercially available 4-(diphenylhydroxymethyl) benzoic acid; (2) the tag can be attached to the thiol group in the unprotected peptide by simply adding the introducing reagent under acidic conditions such as neat 1,1,1,3,3,3-hexafluoroisopropanol (HFIP), which can dissolve hydrophobic and practically insoluble peptides,⁸ or an acidified thiol-additive-free NCL reaction mixture in a one-pot manner; (3) the tag can be quickly and cleanly detached by a standard deprotection approach using trifluoroacetic acid (TFA) with a cation scavenger such as triisopropylsilane (TIS). Unlike the other known solubilizing tag systems,⁶ the solubilizing tag in our method can be attached “on demand”, without the need for starting a new preparation when the target peptide segment is found insoluble. This is a great advantage because it is generally difficult to estimate the solubility of each protein intermediate prior to the synthesis. To broaden the scope of our method and facilitate its further application to challenging targets, we herein report a comparative study of the solubilizing abilities of ionic/nonionic peptide tags having different structures. We compare several peptide tags composed of basic residues (Lys/Arg), an acidic residue (Glu), or a nonionic residue (Asn). Furthermore, peptide tags composed of *N*^ε-tri-Me-Lys or Cys-sulfonate as pH-independent analogs are evaluated. As a result, the *N*^ε-tri-Me-Lys-containing tag is demonstrated to exhibit the best solubilizing ability in solution.

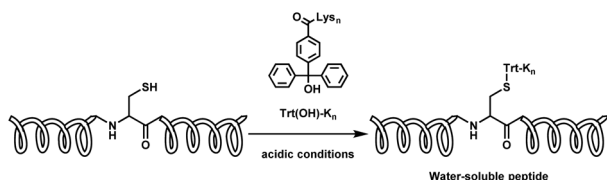


Fig. 1 Solubilizing trityl-type tag system.

Peptide Institute, Inc., Ibaraki, Osaka 567-0085, Japan.

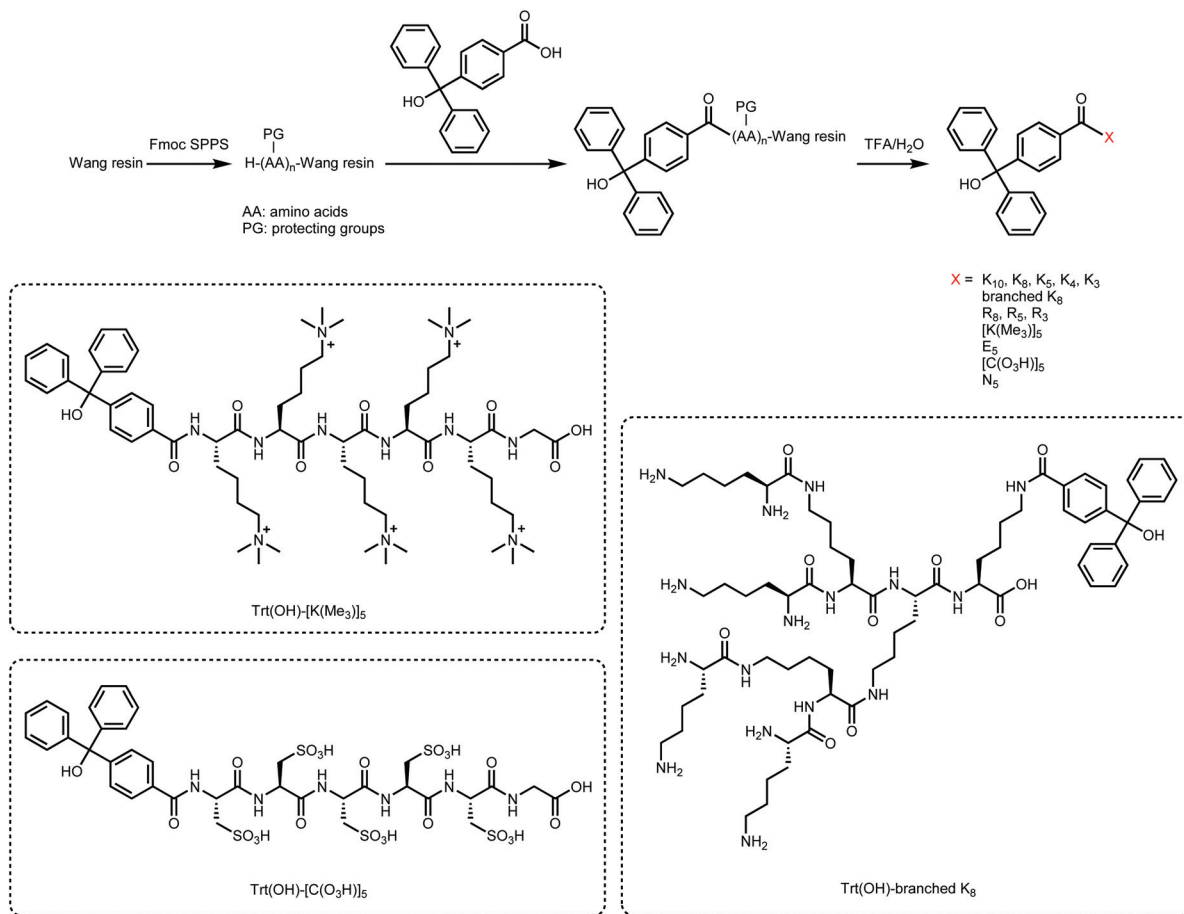
E-mail: t.yoshiya@peptide.co.jp; <https://www.peptide.co.jp/en>

† Electronic supplementary information (ESI) available: Solubility of tagged SPP4 in different solvents for various ligation techniques, and characterization data such as NMR spectra, HPLC traces, and MS spectra. See DOI: 10.1039/c9ob02253h

Results

Preparation of tag-introducing reagents

To compare the solubilizing abilities of several Trt-X tags, we designed several examples: Trt-K_n, where K denotes Lys and *n* = 10, 8, 5, 4, 3, branched 8;⁹ Trt-R_n, where R denotes Arg and



Scheme 1 Synthesis of Trt-X tags and structures of selected solubilizing tags.

$n = 8, 5, 3$; Trt-E₅, where E denotes Glu; Trt-N₅, where N denotes Asn;¹⁰ Trt-[K(Me₃)]₅, where K(Me₃) denotes N^ε-tri-Me-Lys; and Trt-[C(O₃H)]₅, where C(O₃H) denotes Cys-sulfonate. Each tag-introducing reagent was prepared straightforwardly by a standard Fmoc SPPS using Wang resin (Scheme 1). After final acylation using commercially available 4-(diphenylhydroxymethyl)benzoic acid, TFA treatment without a cation scavenger such as TIS followed by RP-HPLC purification afforded each desired trityl alcohol-type tag-introducing reagent successfully.

Solubilization of Ac-Val-Val-Cys-Val-Val-NH₂

Having several tag-introducing reagents in hand, we proceeded with their attachment under standard HFIP conditions to the first solubilizing target Ac-Val-Val-Cys-Val-Val-NH₂ (**1**), which is a highly hydrophobic model peptide.¹¹ The different tags were directly attached to crude **1** after Fmoc SPPS using Rink amide resin, and the respective tagged peptides were purified by RP-HPLC. The resulting HPLC profiles are summarized in Fig. 2. As we expected, the longer Lys/Arg-tagged peptide was eluted faster than its shorter counterpart. Moreover, the branched tagged peptide (branched K₈) was eluted later than the linear derivative. The sulfonate-containing tagged peptide¹² was eluted very fast because it remains

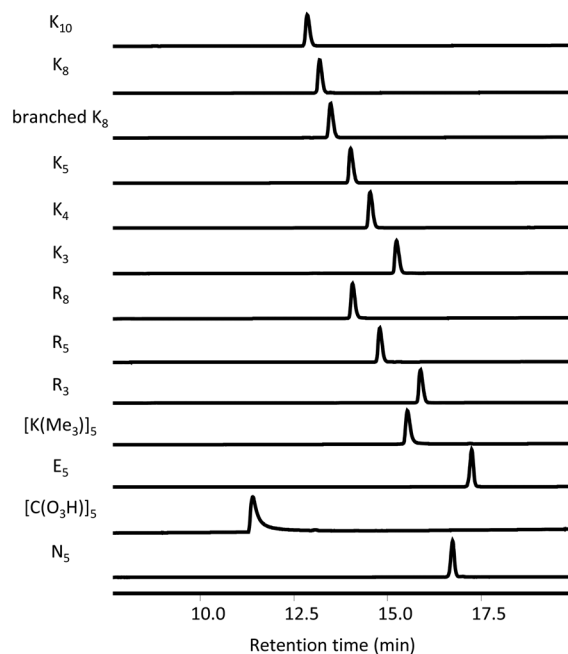


Fig. 2 HPLC profiles of tagged Ac-VVVCVV-NH₂. HPLC conditions: column, YMC-Pack ODS-A (4.6 × 150 mm); elution, 10%–60% CH₃CN in 0.1% TFA (25 min) at 40 °C; flow rate, 1.0 mL min^{−1}; detection, 220 nm.

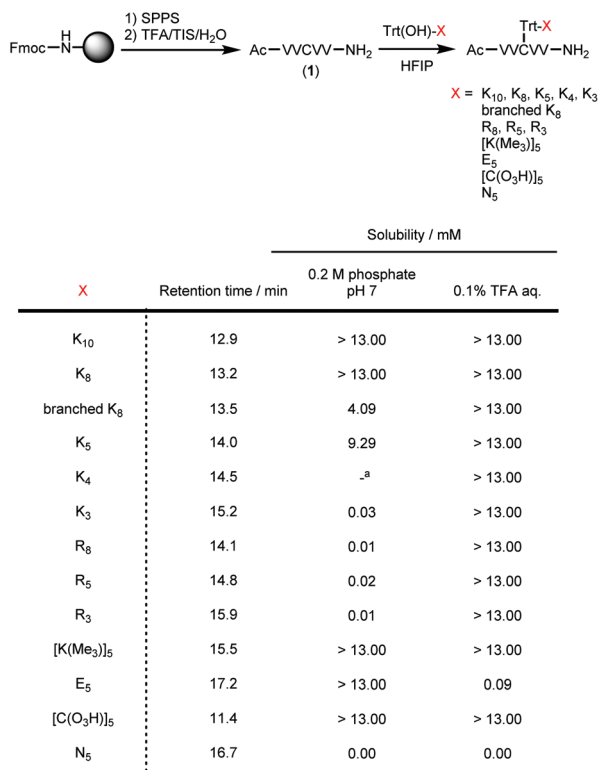


Fig. 3 Solubilization ability of tagged Ac-VVCVV-NH₂. Ac-VVCVV-NH₂ (1) was not soluble (0.00 mM) without a tag. ^a Gel was formed to prevent solubility evaluation.

in anionic form under 0.1% TFA aq conditions and it does not form TFA salt during HPLC. In contrast, the Glu-containing peptide was eluted very late, since it is not anionic under the HPLC conditions. The Asn-containing peptide was also eluted late. With respect to the solubilization ability under acidic conditions (Fig. 3), all basic tags [K, R, K(Me₃)] and the strongly acidic tag C(O₃H) solubilized 1 well. In contrast, the weakly acidic tag E and the nonionic tag N did not solubilize 1 efficiently under these acidic conditions. Under neutral conditions, we observed a dependence of the solubilizing ability on the peptide length of linear K_n and R_n, and linear K₈ solubilized 1 better than branched K₈. The K_n tags exhibited higher solubilizing ability than the corresponding R_n tags.¹³ Furthermore, the [K(Me₃)]₅ tag solubilized 1 more than the corresponding K₅/R₅ tags. Additionally, we observed that both strongly and weakly acidic tags [C(O₃H)] and E solubilized 1 well, whereas the nonionic tag N did not work under neutral conditions.

Solubilization of SPP4

Next, we evaluated the solubilizing ability using as a second target peptide SPP4 (2), which is a known hydrophobic peptide found in membrane proteins.^{6d,14} In a similar manner to that described for model peptide 1, several basic tags were attached to crude 2, and the tagged series 2 was purified by RP-HPLC. In this case, we could not prepare a sulfonate-containing

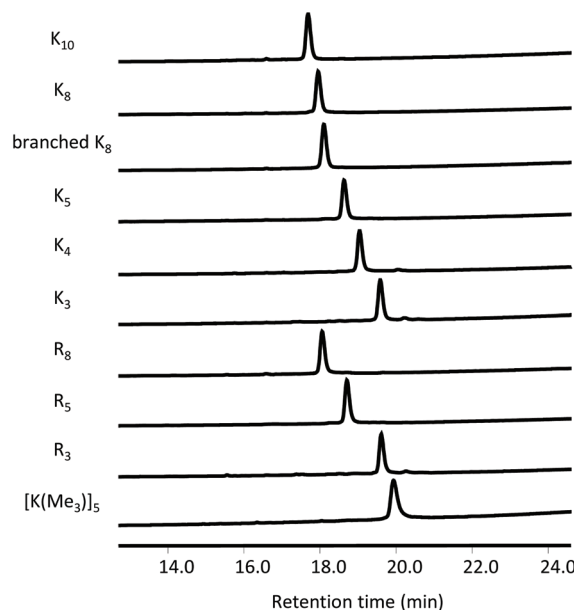


Fig. 4 HPLC profiles of tagged SPP4. HPLC conditions: column, YMC-Pack ODS-A (4.6 × 150 mm); elution, 10%–98% CH₃CN in 0.1% TFA (25 min) at 40 °C; flow rate, 1.0 mL min⁻¹; detection, 220 nm.

tagged 2 because its purification by RP-HPLC was difficult.¹² As shown in Fig. 4, the elution profile obtained by HPLC was similar to that of 1. Thus, the longer Trt-K/R-tagged peptides eluted faster than the shorter Trt-K/R-tagged peptides, the branched tagged peptide eluted later than the linear one, and the Trt-K(Me₃)-tagged peptide eluted later than the corresponding nonmethylated derivative. With respect to the solubilization ability (Fig. 5), a similar tendency to that of peptide 1 was observed. The Trt-K/R-tagged peptides were found to dissolve well in solutions of low pH but not under neutral conditions, and the longer Trt-K/R tags solubilized 2 more than their shorter counterparts. Furthermore, the Trt-K tags showed better solubilizing ability than the Trt-R tags. Additionally, the Trt-[K(Me₃)]₅-tagged peptide was more soluble than the Trt-K₅/R₅-tagged peptides.

Discussion

In this paper, we compare the solubilizing abilities of a series of Trt-based tags, *i.e.*, Trt-K/R/K(Me₃)/C(O₃H)/E/N, which is expected to depend on the electrical properties derived from their different pK_a values.¹⁵ Accordingly, basic tags [Trt-K/R/K(Me₃)] are cationic under both neutral and acidic conditions; the strongly acidic tag Trt-C(O₃H) is anionic under both neutral and cationic conditions; the weakly acidic tag Trt-E is anionic under neutral conditions but not under acidic conditions; and the neutral tag Trt-N is nonionic regardless of the solution pH. The results obtained for the retention times and solubilization abilities of the two hydrophobic peptides evaluated in this study are basically in agreement with that theory; longer/ionized tags showed good solubilization ability com-

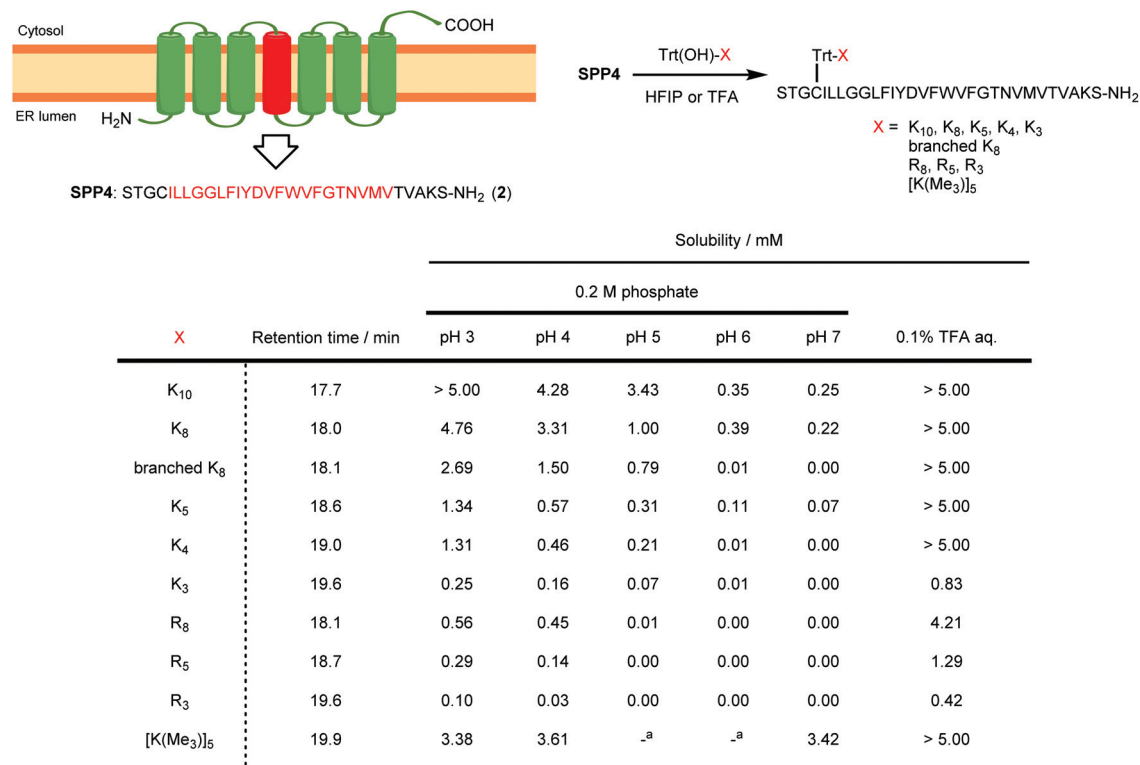


Fig. 5 Solubilization ability of tagged SPP4. SPP4 (2) was not soluble (0.00 mM) without a tag. ^a Not tested.

pared with shorter/nonionized tags. However, to our surprise, the Trt-K/R tag did not work well in solution under neutral conditions. In fact, the Trt-K(Me₃) tag showed higher solubilization ability than the Trt-K/R tags having the same length under neutral conditions. These results cannot be explained simply by the pK_a theory. On the basis of the pK_a of Lys and Arg, the Trt-K/R tags would be protonated consistently under neutral and acidic conditions, and should have similar solubilization abilities regardless of the solution pH. This contradiction can be rationalized in terms of the crowd effect (Fig. 6); under crowded conditions such as those found in the Trt-X tags, the interaction between functionalities would affect the protonation state. Similar protonation inhibitions are well known in peptide chemistry as a neighboring effect; for example, N-terminal Asn(Trt) deprotection is slow during final TFA treatment in Fmoc SPPS,¹⁶ and N^α-Boc deprotection from N-terminal Boc-His(Bom) on the resin is slow during Boc SPPS.¹⁷ In these cases, a protonated base prevents the protonation of a neighboring base, which is essential for the reaction progress. In our case, especially under neutral conditions, the crowd effect would decrease the solubilizing ability of Trt-K/R. In addition, the side-chain guanidine groups of Arg would tend to aggregate *via* π-π stacking under neutral conditions,¹⁸ thereby mitigating the solubilizing ability. Such stacked structures have been crystallographically observed.¹⁹ This phenomenon can also be invoked to explain the lower solubilizing ability of the branched K₈ tag compared to that of linear K₈. In the case of the branched tag, α-amino groups possessing rela-

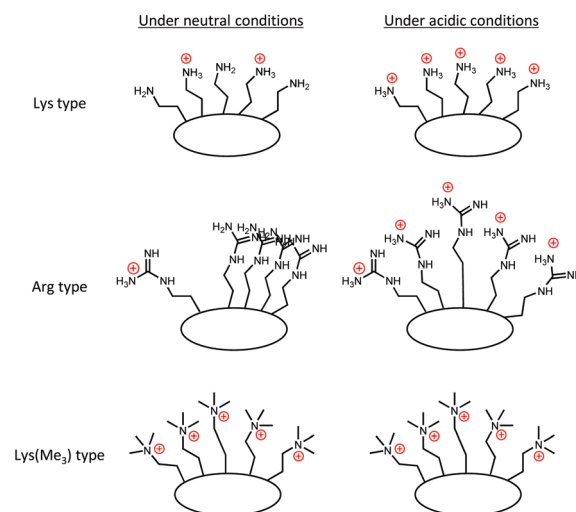


Fig. 6 Proposed structures of tags in the protonation state.

tively weaker basicity than ε-amino groups are present on the surface, and their protonation would be inhibited by the crowd effect. Although the anionic Trt-C(O₃H) tag has good solubilizing ability, its purification is difficult. The Trt-E tag does not solubilize the peptide under acidic conditions, and the Trt-N tag has no obvious solubilizing ability. From these results, it can be concluded that Trt-K(Me₃) is the best tag among those evaluated in this study.

Conclusions

In this study, we compared the solubilizing abilities of several Trt-X tags using two hydrophobic peptides. The results obtained indicate that the Trt-K(Me₃) tag is superior with respect to solubilizing ability among the tags examined. Additionally, the sulfonate-containing tag Trt-C(O₃H) solubilizes the hydrophobic peptide well; however, its HPLC purification is difficult. These results contribute to further development of chemical hydrophobic peptide/protein preparation.

Experimental

General information

Materials. All reagents and solvents were obtained from Peptide Institute, Inc. (Osaka, Japan), FUJIFILM Wako Pure Chemical Corporation (Osaka, Japan), Tokyo Chemical Industry Co., Ltd (Tokyo, Japan), Nacalai Tesque, Inc. (Kyoto, Japan), Watanabe Chemical Industries, Ltd (Hiroshima, Japan), Merck KGaA (Darmstadt, Germany) and Sigma-Aldrich Co. LLC (St Louis, MO).

HPLC, MS and NMR. Preparative HPLC was carried out on a Shimadzu liquid chromatograph Model LC-8A (Kyoto, Japan) with a YMC-Pack ODS-A (30 × 250 mm) or a YMC-Actus Triart C18 (30 × 250 mm) and the following solvent systems: 0.1% TFA in H₂O and 0.1% TFA in CH₃CN or 0.1 M NH₄OAc buffer (pH 7.0) and 60% CH₃CN/0.1 M NH₄OAc buffer (pH 7.0) at a flow rate of 20 mL min⁻¹ with detection at 220 nm. Analytical HPLC was performed on a Shimadzu liquid chromatograph Model LC-10A (Kyoto, Japan) with a YMC-Pack ODS-A (4.6 × 150 mm) or a YMC-Triart C18 (4.6 × 150 mm) and the following solvent systems: 0.1% TFA in H₂O and 0.1% TFA in CH₃CN or 0.1 M NH₄OAc buffer (pH 7.0) and 60% CH₃CN/0.1 M NH₄OAc buffer (pH 7.0) at a flow rate of 1 mL min⁻¹ (40 °C) with detection at 220 nm. Low resolution mass spectra (LRMS) were observed with an Agilent G1956B LC/MSD detector using an Agilent 1100 series HPLC system; for deconvolution, the observed masses (most abundant masses) were derived from the experimental *m/z* values for each protonation state of a target peptide. ¹H-/¹³C-NMR spectra were recorded on a JEOL-ECX400 spectrometer (Tokyo, Japan), as solutions in deuterated solvents as specified. Chemical shift values (δ) are given in parts per million (ppm) using the residual solvent as the internal standard.

SPPS. Automated Fmoc SPPS was performed on an ABI 433A peptide synthesizer (Applied Biosystems, USA). The peptide chains except Trt(OH)-[K(Me₃)]₅ and Trt(OH)-[C(O₃H)]₅ were elongated using the coupling protocol of Fmoc-amino acid/DIC/OxymaPure.²⁰ Trt(OH)-[K(Me₃)]₅ and Trt(OH)-[C(O₃H)]₅ were manually elongated using the coupling protocol of HCTU/DIEA.²¹ The following side-chain-protected amino acids and pseudoproline unit were employed: Arg(Pbf), Asn(Trt), Asp(OtBu), Glu(OtBu), Ser(*t*Bu), Cys(Trt), Lys(Boc), Lys(Fmoc), Ser(*t*Bu), Thr(*t*Bu), Trp(Boc), Tyr(*t*Bu), and Val-Thr(Ψ^{Me,Me}pro).

Experimental section

Synthesis of Fmoc-amino acid and peptides

Fmoc-Cys(O₃H)-OH. To a solution of cysteic acid (1.0 g, 5.91 mmol) and Na₂CO₃ (0.626 g, 5.91 mmol) in H₂O (25 mL) was added Fmoc-OSu solution (2.39 g, 7.09 mmol in 25 mL acetone). The reaction mixture was stirred at room temperature for 2 h, and then acetone was removed under reduced pressure. The aqueous residue was washed with AcOEt/hexane (v/v, 1/1). The aqueous layer was acidified with 1 M HCl to pH 2 and then subjected to preparative HPLC to yield the title compound (1.57 g, 80%). ¹H NMR (400 MHz, DMSO-*d*₆) δ 2.86 (dd, *J* = 13.7 and 5.0 Hz, 1H), 2.90 (dd, *J* = 13.7 and 6.4 Hz, 1H), 4.10–4.33 (m, 4H), 7.28–7.37 (m, 2H), 7.42 (t, *J* = 7.3 Hz, 2H), 7.47 (d, *J* = 6.9 Hz, 1H) 7.70 (dd, *J* = 7.3 and 2.7 Hz, 2H), 7.89 (d, *J* = 7.3 Hz, 1H); ¹³C NMR (100 MHz, DMSO-*d*₆) δ 46.7, 51.0, 51.4, 65.9, 120.2, 125.3, 125.4, 127.2, 127.7, 140.7, 140.8, 143.8, 143.9, 155.7, 172.3; LRMS (*M* – *H*) calcd for C₁₈H₁₆NO₇S 390.1, found 390.1.

Ac-VVCVV-NH₂ (1). The peptide was assembled on Rink amide resin (0.40 mmol) using an automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/TIS/H₂O (v/v, 95/2.5/2.5) for 1.5 h to give a crude product. The peptide was used to the next reaction without further purification. Analytical HPLC: *t*_R = 14.5 min (1–60% CH₃CN/0.1% TFA for 25 min); LRMS (*M* + *H*) calcd for C₂₅H₄₇N₆O₆S 559.3, found 559.3.

STGCILLGLFIYDVFVFGTNTVMVAKS-NH₂ = SPP4 (2). The peptide was assembled on Rink amide resin (0.25 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa and Fmoc-Val-Thr(Ψ^{Me,Me}pro): 4 equiv.). The subsequent deprotection of the resin was carried out with TFA/TIS/H₂O/DMB (v/v, 92.5/2.5/2.5/2.5) for 1.5 h to give a crude product. The peptide was used for the next reaction without further purification. Analytical HPLC: *t*_R = 22.3 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₁₅₃H₂₃₄N₃₄O₃₉S₂ 3236.7, found 3237.6.

Trt(OH)-K₁₀. The peptide was assembled on Fmoc-Lys(Boc)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Lys(Boc): 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 30 min to give a crude product, which was purified by preparative HPLC to yield the title compound (641 mg, 59%). Analytical HPLC: *t*_R = 14.4 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (*M* + *H*) calcd for C₈₀H₁₃₇N₂₀O₁₃ 1586.1, found 1586.0.

Trt(OH)-K₈. The peptide was assembled on Fmoc-Lys(Boc)-Wang resin (0.25 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 4 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 40 min to give a crude product, which was purified by preparative HPLC to yield the title compound (361 mg, 64%). Analytical HPLC: *t*_R = 14.8 min (1–40% CH₃CN/

0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{68}H_{113}N_{16}O_{11}$ 1329.9, found 1329.8.

Trt(OH)-K₅. The peptide was assembled on Fmoc-Lys(Boc)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Lys(Boc): 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 40 min to give a crude product, which was purified by preparative HPLC to yield the title compound (395 mg, 65%). Analytical HPLC: t_R = 15.7 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{50}H_{77}N_{10}O_8$ 945.6, found 945.5.

Trt(OH)-K₄. The peptide was assembled on Fmoc-Lys(Boc)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 1 h to give a crude product, which was purified by preparative HPLC to yield the title compound (328 mg, 64%). Analytical HPLC: t_R = 16.3 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{44}H_{65}N_8O_7$ 817.5, found 817.4.

Trt(OH)-K₃. The peptide was assembled on Fmoc-Lys(Boc)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 40 min to give a crude product, which was purified by preparative HPLC to yield the title compound (202 mg, 49%). Analytical HPLC: t_R = 17.4 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{38}H_{53}N_6O_6$ 689.4, found 689.4.

Trt(OH)-R₈. The peptide was assembled on the Fmoc-Arg (Pbf)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O/thioanisole (v/v/v, 92.5/2.5/5) for 25 min to give a crude product, which was purified by preparative HPLC to yield the title compound (202 mg, 20%). Analytical HPLC: t_R = 16.8 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{68}H_{113}N_{32}O_{11}$ 1553.9, found 1553.9.

Trt(OH)-R₅. The peptide was assembled on the Fmoc-Arg (Pbf)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O/thioanisole (v/v/v, 95/2.5/2.5) for 25 min to give a crude product, which was purified by preparative HPLC to yield the title compound (248 mg, 37%). Analytical HPLC: t_R = 17.4 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{50}H_{77}N_{20}O_8$ 1085.6, found 1085.5.

Trt(OH)-R₃. The peptide was assembled on the Fmoc-Arg (Pbf)-Wang resin (0.40 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.).

The subsequent deprotection of the resin was carried out with TFA/H₂O/thioanisole (v/v/v, 95/2.5/2.5) for 25 min to give a crude product, which was purified by preparative HPLC to yield the title compound (265 mg, 59%). Analytical HPLC: t_R = 18.8 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{38}H_{53}N_{12}O_6$ 773.4, found 773.4.

Trt(OH)-E₅. The peptide was assembled on the Fmoc-Glu (OtBu)-Wang resin (0.25 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 4 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 1 h to give a crude product, which was purified by preparative HPLC to yield the title compound (144 mg, 61%). Analytical HPLC: t_R = 14.3 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M – H) calcd for $C_{45}H_{50}N_5O_{18}$ 948.3, found 948.2.

Trt(OH)-N₅. The peptide was assembled on Rink amide resin (0.4 mmol) using the automated Fmoc SPPS procedure as described in General information (Fmoc-Xaa: 2.5 equiv., Fmoc-Asp-OtBu is used as C-terminal Xaa, 4-(diphenylhydroxymethyl)benzoic acid: 2.5 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 1 h to give a crude product, which was purified by preparative HPLC to yield the title compound (234 mg, 67%). Analytical HPLC: t_R = 12.9 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M – H) calcd for $C_{40}H_{45}N_{10}O_{13}$ 873.3, found 873.3.

Trt(OH)-branched K₈. The peptide was assembled on the Fmoc-Lys(Aloc)-Wang resin (0.25 mmol) using a manual Fmoc SPPS procedure (Fmoc-Lys(Fmoc): 4 equiv. to amino group) without final Fmoc deprotection. The resin was treated with Pd(PPh₃)₄ (72.2 mg, 0.0625 mmol) and PhSiH₃ (1.53 mL, 12.5 mmol) in CH₂Cl₂ for 1 h. Then, 4-(diphenylhydroxymethyl)benzoic acid was coupled using the DIC/Oxyma method, followed by deprotection of the Fmoc group with 20% piperidine/DMF. The obtained resin was treated with TFA/H₂O (v/v, 95/5) for 1 h to give a crude product, which was purified by preparative HPLC to yield the title compound (77.8 mg, 14%). Analytical HPLC: t_R = 15.9 min (1–40% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for $C_{68}H_{113}N_{16}O_{11}$ 1329.9, found 1329.8.

Trt(OH)-[K(Me₃)]₅. The peptide was assembled on the Fmoc-Gly-Wang resin (0.1 mmol) using the manual Fmoc SPPS procedure (Fmoc-Lys(Me₃)-Cl: 2 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 1 h to give a crude product, which was purified by preparative HPLC to yield the title compound (72.6 mg, 41%). Analytical HPLC: t_R = 12.1 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M) calcd for $C_{67}H_{114}N_{11}O_9^{5+}$ 243.4, found 243.5.

Trt(OH)-[C(O₃H)]₅. The peptide was assembled on the Fmoc-Gly-Wang resin (0.25 mmol) using the manual Fmoc SPPS procedure (Fmoc-Cys(O₃H): 2 equiv., 4-(diphenylhydroxymethyl)benzoic acid: 2 equiv.). The subsequent deprotection of the resin was carried out with TFA/H₂O (v/v, 95/5) for 1 h to give a crude product, which was purified by preparative HPLC to yield the title compound (96.9 mg, 35%). Analytical HPLC: t_R =

9.2 min (1–60% CH₃CN/0.1% TFA for 25 min); LRMS (M – H) calcd for C₃₇H₄₃N₆O₂₄S₅ 1115.1, found 1115.0.

Attachment of Trt(OH)-X to Ac-VVCVV-NH₂

General procedure. Ac-VVCVV-NH₂ (1 equiv., 10 mM) and Trt(OH)-X (1.1 equiv.) were dissolved in HFIP.⁶ After stirring for 1 h at room temperature, the reaction mixture was concentrated and subjected to preparative HPLC.

Ac-VVC(Trt-K₁₀)VV-NH₂. 27.8 mg, 85% isolated yield. Analytical HPLC: *t*_R = 12.9 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₁₀₅H₁₈₁N₂₆O₁₈S 2126.4, found 2126.3.

Ac-VVC(Trt-K₈)VV-NH₂. 25.6 mg, 92% isolated yield. Analytical HPLC: *t*_R = 13.2 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₉₃H₁₅₇N₂₂O₁₆S 1870.2, found 1870.1.

Ac-VVC(Trt-branched K₈)VV-NH₂. 25.1 mg, 90% isolated yield. Analytical HPLC: *t*_R = 13.5 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₉₃H₁₅₇N₂₂O₁₆S 1870.2, found 1870.1.

Ac-VVC(Trt-K₅)VV-NH₂. 26.1 mg, 85% isolated yield. Analytical HPLC: *t*_R = 14.0 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₇₅H₁₂₁N₁₆O₁₃S 1485.9, found 1485.8.

Ac-VVC(Trt-K₄)VV-NH₂. 24.0 mg, 88% isolated yield. Analytical HPLC: *t*_R = 14.5 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₆₉H₁₀₉N₁₄O₁₂S 1357.8, found 1357.8.

Ac-VVC(Trt-K₃)VV-NH₂. 26.3 mg, 84% isolated yield. Analytical HPLC: *t*_R = 15.2 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₆₃H₉₇N₁₂O₁₁S 1229.7, found 1229.6.

Ac-VVC(Trt-R₈)VV-NH₂. 22.1 mg, 73% isolated yield. Analytical HPLC: *t*_R = 14.1 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₉₃H₁₅₇N₃₈O₁₆S 2094.2, found 2094.2.

Ac-VVC(Trt-R₅)VV-NH₂. 29.1 mg, 88% isolated yield. Analytical HPLC: *t*_R = 14.8 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₇₅H₁₂₁N₂₆O₁₃S 1625.9, found 1625.9.

Ac-VVC(Trt-R₃)VV-NH₂. 27.6 mg, 83% isolated yield. Analytical HPLC: *t*_R = 15.9 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₆₃H₉₇N₁₈O₁₁S 1313.7, found 1313.7.

Ac-VVC(Trt-E₅)VV-NH₂. 16.5 mg, 74% isolated yield. Analytical HPLC: *t*_R = 17.2 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M – H) calcd for C₇₀H₉₄N₁₁O₂₃S 1488.6, found 1488.5.

Ac-VVC(Trt-N₅)VV-NH₂. 7.1 mg, 70% isolated yield. Analytical HPLC: *t*_R = 16.7 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M + H) calcd for C₆₅H₉₁N₁₆O₁₈S 1415.6, found 1415.6.

Ac-VVC{Trt-[K(Me₃)]₅}VV-NH₂. 21.3 mg, 81% isolated yield. Analytical HPLC: *t*_R = 15.5 min (10–60% CH₃CN/0.1% TFA for 25 min); LRMS (M) calcd for C₉₂H₁₅₈N₁₇O₁₄S⁵⁺ 351.4, found 351.6.

Ac-VVC{Trt-[C(O₃H)]₅}VV-NH₂. The title peptide was isolated using a 0.1 M NH₄OAc buffer (pH 7) solvent system as a single

peak (8.8 mg, 35% isolated yield). Analytical HPLC: *t*_R = 14.2 min (10–60% CH₃CN/0.1 M NH₄OAc buffer (pH 7) for 25 min); LRMS (M – H) calcd for C₆₂H₈₇N₁₂O₂₉S₆ 1655.4, found 1655.3.

Attachment of Trt(OH)-X to SPP4

General procedure. SPP4 (1 equiv., 10 mM) and Trt(OH)-X (1.1 equiv.) were dissolved in HFIP or TFA. After stirring for 1 h at room temperature, the reaction mixture was concentrated and subjected to preparative HPLC.

STGC(Trt-K₁₀)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-K₁₀. 46.1 mg, 35% isolated yield. Analytical HPLC: *t*_R = 17.7 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₂₃₃H₃₆₈N₅₄O₅₁S₂ 4804.7, found 4805.4.

STGC(Trt-K₈)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-K₈. 37.3 mg, 33% isolated yield. Analytical HPLC: *t*_R = 18.0 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₂₂₁H₃₄₄N₅₀O₄₉S₂ 4548.5, found 4549.0.

STGC(Trt-branched K₈)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-branched K₈. 31.6 mg, 34% isolated yield. Analytical HPLC: *t*_R = 18.0 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₂₂₁H₃₄₄N₅₀O₄₉S₂ 4548.5, found 4549.1.

STGC(Trt-K₅)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-K₅. 27.9 mg, 28% isolated yield. Analytical HPLC: *t*_R = 18.6 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₂₀₃H₃₀₈N₄₄O₄₆S₂ 4164.3, found 4164.5.

STGC(Trt-K₄)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-K₄. 12.8 mg, 14% isolated yield. Analytical HPLC: *t*_R = 19.0 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₁₉₇H₂₉₆N₄₂O₄₅S₂ 4036.2, found 4036.6.

STGC(Trt-K₃)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-K₃. 15.8 mg, 18% isolated yield. Analytical HPLC: *t*_R = 19.6 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₁₉₁H₂₈₄N₄₀O₄₄S₂ 3908.1, found 3908.1.

STGC(Trt-R₈)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-R₈. 32.8 mg, 28% isolated yield. Analytical HPLC: *t*_R = 18.1 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₂₂₁H₃₄₄N₆₆O₄₉S₂ 4772.6, found 4773.2.

STGC(Trt-R₅)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-R₅. 27.5 mg, 27% isolated yield. Analytical HPLC: *t*_R = 18.7 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₂₀₃H₃₀₈N₅₄O₄₆S₂ 4304.3, found 4304.6.

STGC(Trt-R₃)ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-R₃. 15.9 mg, 17% isolated yield. Analytical HPLC: *t*_R = 19.6 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (ESI) calcd for C₁₉₁H₂₈₄N₄₆O₄₄S₂ 3992.1, found 3992.2.

STGC{Trt-[K(Me₃)]₅}ILLGGLFIYDVFWVFGTNVMVTVAKS-NH₂ = SPP4-Trt-K[(Me₃)]₅. 11.5 mg, 13% isolated yield. Analytical HPLC: *t*_R = 19.9 min (10–98% CH₃CN/0.1% TFA for 25 min); LRMS (M) calcd for C₂₂₀H₃₄₆N₄₅O₄₇S₂⁵⁺ 887.3, found 887.3.

Evaluation of the solubility of Trt-X tagged peptides. The solubility of Trt-X tagged peptides was determined by calculation of peak areas of standard solutions and the supernatant of saturated solutions using analytical HPLC. The standard solutions were prepared by dissolving the peptides at 1 mg mL^{−1} in DMSO. The saturated solutions were prepared by adding small aliquots of buffers into the peptides. The satu-

rated solutions were centrifuged and the supernatants were diluted with 50% AcOH/H₂O or 3 M Gn-HCl in 50% AcOH/H₂O, respectively. The standard solutions and the diluted supernatant solutions were compared by peak areas using analytical HPLC.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- (a) R. B. Merrifield, *J. Am. Chem. Soc.*, 1963, **85**, 2149–2154; (b) L. A. Carpino and G. Y. Han, *J. Am. Chem. Soc.*, 1970, **92**, 5748–5749; (c) M. Amblard, J.-A. Fehrentz, J. Martinez and G. Subra, *Mol. Biotechnol.*, 2006, **33**, 239–254; (d) I. Coin, M. Beyermann and M. Bienert, *Nat. Protoc.*, 2007, **2**, 3247–3256.
- (a) P. E. Dawson, T. W. Muir, I. Clark-Lewis and S. B. H. Kent, *Science*, 1994, **266**, 776–779; (b) P. E. Dawson and S. B. H. Kent, *Annu. Rev. Biochem.*, 2000, **69**, 923–960; (c) S. B. H. Kent, *Chem. Soc. Rev.*, 2009, **38**, 338–351; (d) V. Agouridas, O. El Mahdi, V. Diemer, M. Cargoët, J.-C. M. Monbaliu and O. Melnyk, *Chem. Rev.*, 2019, **119**, 7328–7443.
- (a) S. Aimoto, *Biopolymers*, 1999, **51**, 247–265; (b) S. Sakakibara, *Biopolymers*, 1999, **51**, 279–296; (c) J. W. Bode, *Acc. Chem. Res.*, 2017, **50**, 2104–2115; (d) H. Liu and X. Li, *Acc. Chem. Res.*, 2018, **51**, 1643–1655.
- (a) A. K. Tickler, A. B. Clippingdale and J. D. Wade, *Protein Pept. Lett.*, 2004, **11**, 377–384; (b) M. Paradís-Bas, J. Tulla-Puche and F. Albericio, *Chem. Soc. Rev.*, 2016, **45**, 631–654.
- (a) C. Zuo, S. Tang and J.-S. Zheng, *J. Pept. Sci.*, 2015, **21**, 540–549; (b) J.-B. Li, S. Tang, J.-S. Zheng, C.-L. Tian and L. Liu, *Acc. Chem. Res.*, 2017, **50**, 1143–1153.
- (a) P. W. R. Harris and M. A. Brimble, *Pept. Sci.*, 2010, **94**, 542–550; (b) Z. Tan, S. Shang and S. J. Danishefsky, *Proc. Natl. Acad. Sci. U. S. A.*, 2011, **108**, 4297–4302; (c) Y.-C. Huang, Y.-M. Li, Y. Chen, M. Pan, Y.-T. Li, L. Yu, Q.-X. Guo and L. Liu, *Angew. Chem., Int. Ed.*, 2013, **52**, 4858–4862, (*Angew. Chem.*, 2013, **125**, 4958–4962); (d) J.-S. Zheng, M. Yu, Y.-K. Qi, S. Tang, F. Shen, Z.-P. Wang, L. Xiao, L. Zhang, C.-L. Tian and L. Liu, *J. Am. Chem. Soc.*, 2014, **136**, 3695–3704; (e) C. Bello, S. Wang, L. Meng, K. W. Moremen and C. F. W. Becker, *Angew. Chem., Int. Ed.*, 2015, **54**, 7711–7715, (*Angew. Chem.*, 2015, **127**, 7823–7828); (f) Y. Asahina, S. Komiya, A. Ohagi, R. Fujimoto, H. Tamagaki, K. Nakagawa, T. Sato, S. Akira, T. Takao, A. Ishii, Y. Nakahara and H. Hojo, *Angew. Chem., Int. Ed.*, 2015, **54**, 8226–8230, (*Angew. Chem.*, 2015, **127**, 8344–8348); (g) S. K. Maity, G. Mann, M. Jbara, S. Laps, G. Kamnesky and A. Brik, *Org. Lett.*, 2016, **18**, 3026–3029; (h) J.-S. Zheng, Y. He, C. Zuo, X.-Y. Cai, S. Tang, Z. A. Wang, L.-H. Zhang, C.-L. Tian and L. Liu, *J. Am. Chem. Soc.*, 2016, **138**, 3553–3561; (i) C. Zuo, S. Tang, Y.-Y. Si, Z. A. Wang, C.-L. Tian and J.-S. Zheng, *Org. Biomol. Chem.*, 2016, **14**, 5012–5018; (j) M. T. Jacobsen, M. E. Petersen, X. Ye, M. Galibert, G. H. Lorimer, V. Aucagne and M. S. Kay, *J. Am. Chem. Soc.*, 2016, **138**, 11775–11782; (k) S. F. Loibl, Z. Harpaz, R. Zitterbart and O. Seitz, *Chem. Sci.*, 2016, **7**, 6753–6759; (l) S. Bondalapati, E. Eid, S. M. Mali, C. Wolberger and A. Brik, *Chem. Sci.*, 2017, **8**, 4027–4034; (m) S. Tsuda, H. Nishio and T. Yoshiya, *Chem. Commun.*, 2018, **54**, 8861; (n) B. Zhang, Q. Deng, C. Zuo, B. Yan, C. Zuo, X.-X. Cao, T. F. Zhu, J.-S. Zheng and L. Liu, *Angew. Chem., Int. Ed.*, 2019, **58**, 12231–12237, (*Angew. Chem.*, 2019, **131**, 12359–12365).
- (a) M. Mochizuki, H. Hibino and Y. Nishiuchi, *Org. Lett.*, 2014, **16**, 5740–5743; (b) S. Tsuda, M. Mochizuki, H. Ishiba, K. Yoshizawa-Kumagaye, H. Nishio, S. Oishi and T. Yoshiya, *Angew. Chem., Int. Ed.*, 2018, **57**, 2105–2109, (*Angew. Chem.*, 2018, **130**, 2127–2131); (c) S. Tsuda, S. Masuda and T. Yoshiya, *Org. Biomol. Chem.*, 2019, **17**, 1202–1205; (d) T. Yoshiya, S. Tsuda and S. Masuda, *ChemBioChem*, 2019, **20**, 1906–1913; (e) S. Tsuda, S. Masuda and T. Yoshiya, *ChemBioChem*, 2019, **20**, 2063–2069.
- M. Narita, S. Honda, H. Umeyama and S. Obana, *Bull. Chem. Soc. Jpn.*, 1988, **61**, 281–284.
- M. Paradís-Bas, M. Albert-Soriano, J. Tulla-Puche and F. Albericio, *Org. Biomol. Chem.*, 2014, **12**, 7194–7196.
- (a) M. M. Islam, M. A. Khan and Y. Kuroda, *Biochim. Biophys. Acta*, 2012, **1824**, 1144–1150; (b) M. A. Khan, M. M. Islam and Y. Kuroda, *Biochim. Biophys. Acta*, 2013, **1834**, 2107–2115.
- T. Yoshiya, N. Ito, T. Kimura and Y. Kiso, *J. Pept. Sci.*, 2008, **14**, 1203–1208.
- Sulfonate-tagged **1** showed a slightly broadened elution profile (Fig. 2), probably suggesting that it has interaction with the HPLC column. In the case of SPP4 (**2**), sulfonate-tagged **2** could not be eluted. Such an anionic hydrophobic peptide is often difficult to be purified by HPLC.
- The results did not change even after 18 h.
- E. C. B. Johnson and S. B. H. Kent, *J. Am. Chem. Soc.*, 2006, **128**, 7140–7141.
- pK_a (or pK_b of basic functionality) values of Lys/Arg/Glu/Cys (O₃H) are apparently as follows: 10.53/12.48/4.25/–2.6, where pK_a of side chain sulfonate is correspondingly adopted from that of methanesulfonic acid. pK_a data were compiled from following websites: (a) D. H. Ripin and D. A. Evans, http://evans.rc.fas.harvard.edu/pdf/evans_pKa_table.pdf (accessed October 2019) (b) R. Williams, https://www.chem.wisc.edu/areas/reich/pkatable/pKa_compilation-1-Williams.pdf (accessed October 2019).
- M. Friede, S. Denery, J. Neimark, S. Kieffer, H. Gausepohl and J.-P. Briand, *Pept. Res.*, 1992, **5**, 145–147.
- K. Yoshizawa-Kumagaye, Y. Nishiuchi, H. Nishio and T. Kimura, *J. Pept. Sci.*, 2005, **11**, 512–515.
- All solubilization experiments in this paper were performed under aqueous conditions without denaturing agents such as guanidine-HCl. Incidentally, to evaluate the applicability

- of our Trt-K/R method for NCL, α -ketoacid-hydroxylamine (KAHA) ligation,²² and serine/threonine ligation (STL),²³ the solubilities of tagged SPP4 in guanidine-containing neutral buffer, DMSO-H₂O (9 : 1), and AcOH-pyridine (1 : 1) were also monitored (Table S1†). As a result, even simple Trt-K₅/R₅ tags worked well under the former two conditions but not in the latter one, suggesting that the Trt-K/R method can be simply applied to NCL and KAHA ligation.
- 19 R. M. Vernon, P. A. Chong, B. Tsang, T. H. Kim, A. Bah, P. Farber, H. Lin and J. D. Forman-Kay, *eLife*, 2018, 7, e31486.
 - 20 R. Subirós-Funosas, R. Prohens, R. Barbas, A. El-Faham and F. Albericio, *Chem. – Eur. J.*, 2009, 15, 9394–9403.
 - 21 C. A. Hood, G. Fuentes, H. Patel, K. Page, M. Menakuru and J. H. Park, *J. Pept. Sci.*, 2008, 14, 97–101.
 - 22 (a) J. W. Bode, R. M. Fox and K. D. Baucom, *Angew. Chem., Int. Ed.*, 2006, 45, 1248–1252, (*Angew. Chem.*, 2006, 118, 1270–1274); (b) J. W. Bode, *Acc. Chem. Res.*, 2017, 50, 2104–2115.
 - 23 (a) X. Li, H. Y. Lam, Y. Zhang and C. K. Chan, *Org. Lett.*, 2010, 12, 1724–1727; (b) H. Liu and X. Li, *Acc. Chem. Res.*, 2018, 51, 1643–1655.