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GRAPHICAL ABSTRACT



Methyl and ethyl substituted acids are the most selective inhibitors of AKR1C3.

2,3-Diarylpropenoic acids as selective non-steroidal inhibitors of type-5 17βhydroxysteroid dehydrogenase (AKR1C3)

Martin Gazvoda^{a,&}, Nataša Beranič^{b,&}, Samo Turk^c, Bojan Burja^a, Marijan Kočevar^a, Tea Lanišnik Rižner,^b Stanislav Gobec^{**,c}, Slovenko Polanc^{*,a}

^aFaculty of Chemistry and Chemical Technology, University of Ljubljana, Aškerčeva 5, SI-1000, Ljubljana, Slovenia

^bInstitute of Biochemistry, Faculty of Medicine, University of Ljubljana, Vrazov trg 2, SI-1000, Ljubljana, Slovenia

^cFaculty of Pharmacy, University of Ljubljana, Aškerčeva 7, SI-1000, Ljubljana, Slovenia

GRAPHICAL ABSTRACT

CO₂H MeSO

Abbreviations: DMF, *N*,*N*-dimethylformamide; DMSO, dimethyl sulfoxide; ESI, electrospray ionization; HPLC, high-performance liquid chromatography; HRMS, high-resolution mass spectrometry; IR, infrared; LC-MS liquid chromatography-mass spectrometry; NADPH, reduced nicotinamide adenine dinucleotide phosphate; NMR, nuclear magnetic resonance; NSAID, non-steroidal anti-inflammatory drug.

[&] Both authors contributed equally to the manuscript.

* Corresponding author. Tel.: +386-1-24-19-236; fax: +386-1-24-19-220.

** Corresponding author. Tel.: +386-1-47-69-585; fax: +386-1-42-58-031.

E-mail addresses: slovenko.polanc@fkkt.uni-lj.si (S. Polanc), stanislav.gobec@ffa.uni-lj.si (S. Gobec).

ABSTRACT

The aldo-keto reductase AKR1C3 is an important target for the development of new drugs. Selective inhibitors of this enzyme are needed because they should not inhibit other, structurally closely related AKR1C isoforms. A comprehensive series of 2,3-diarylpropenoic acids was synthesized and evaluated for the inhibition of AKR1C1–AKR1C3. We found that the 4-methylsulfonylphenyl substituent at position 2 of these acids is required to exhibit the selective inhibition of AKR1C3. The best results were obtained for the compounds that fulfill the above requirement and possess a 4-bromophenyl, 4-methylthiophenyl, 4-methylphenyl or 4-ethylphenyl substituent at position 3 of the substituted propenoic acids (i.e., acids **28**, **29**, **37**, and **39**, respectively). These compounds represent an important step towards the development of drug candidates for a treatment of the hormone-dependent and hormone-independent forms of prostate and breast cancers.

Keywords: Aldo-keto reductase; AKR1C3; Selective inhibitors; 2,3-Diarylpropenoic acids; Cancer

1. Introduction

The enzymes AKR1C1–AKR1C4, members of the aldo-keto reductase superfamily, catalyze the interconversions of 3-, 17- and 20-ketosteroids with the corresponding $3\alpha/\beta$ -, 17 β -, 20 α -hydroxysteroids to varying extents, using NADPH as a cofactor [1,2]. In this way they can control the ligand occupancy and *trans*-activation of androgen, estrogen and progesterone receptors by modulating the concentrations of the active steroids [3]. The AKR1C enzymes are also involved in the prostaglandin and neurosteroid production and inactivation, and in the metabolism of xenobiotics [2].

Among the AKR1C enzymes, AKR1C3 preferentially acts as a 17-ketosteroid reductase and converts a weak androgen 4-androstene-3,17-dione to a potent androgen testosterone and estrone to a potent estrogen 17 β -estradiol. As a 20-ketosteroid reductase, AKR1C3 inactivates progesterone by forming its metabolite 20 α -hydroxyprogesterone, which has a lower affinity towards the progesterone receptor [1,4]. AKR1C3 also acts as a 3-ketosteroid reductase by converting 5 α -dihydrotestosterone into an estrogen receptor β ligand 3 β -androstandiol, and inactive androgen 3 α -androstandiol [1]. Catalyzing these reactions AKR1C3 represents an important target enzyme for the development of potential drugs for a treatment of the hormone-dependent forms of cancer. Finally, AKR1C3 catalyzes the reduction of prostaglandin H2 (PGH2) into PGF2 α , and PGD2 into 11 β -PGF2, thereby diverting the

biosynthesis of prostanoids away from the antiproliferative J-series [2]. This makes AKR1C3 an interesting target enzyme also for the development of the hormone-independent forms of cancer.

The products of AKR1C3-catalyzed reactions stimulate tumor growth, so an imbalance in the AKR1C3 expression and action can be related to the different hormone-dependent and hormone-independent cancers like breast [5], prostate [6], endometrial [7], and also other diseases, such as endometriosis [8], benign prostatic hyperplasia and dysmenorrhea [9,10]. AKR1C3 thus represents an interesting therapeutic target for the development of new drugs. Since AKR1C3 shares at least an 84% sequence homology and overlapping catalytic properties with other human members of the AKR1C superfamily (AKR1C1, AKR1C2 and AKR1C4) that are, with the exception of the liver-specific AKR1C4, expressed in different tissues [1], selective AKR1C3 inhibition is crucial and remains a great challenge. To date several structurally different classes of compounds have been reported to possess AKR1C3inhibitory activity. The structural analogues of AKR1C3 substrates with a steroidal backbone, such as progestins [11] and estrogen lactones [12], have been shown to inhibit AKR1C3 in the low-µM or even the low-nM range, respectively. Inhibitors of AKR1C3 also comprise other classes of compounds, like cyclopentane derivatives [13], benzodiazepines [14], flavonoids [15], cinnamic acid derivatives [16] jasmonates [17] or NSAIDs [18]. All these compounds inhibit AKR1C3 with μ M IC₅₀ values, but most of these inhibitors have low selectivity over AKR1C3. To date only a few compounds that act as selective AKR1C3 inhibitors have been reported (e.g., an indole derivative known as CBM, and an indomethacin analogue that belongs to the group of NSAIDs). Unfortunately, CBM possesses poor bioavailability relative to its potency [10], so the search for a selective AKR1C3 inhibitor continues.

From the above facts it is evident that the desired inhibitor of AKR1C3 should not inhibit the closely related isoforms AKR1C1 and AKR1C2, which play distinct roles in the metabolism of steroidal hormones. Recently described *N*-phenylaminobenzoates were found to fulfill this requirement and can serve as potent and selective inhibitors of AKR1C3 [19]. On the other hand, the already-mentioned cinnamic acid and derivatives are also promising inhibitors of AKR1C3 [16].

Here we report on the effectiveness and the selectivity of inhibition of AKR1C3 by the related 2,3-diarylpropenoic acids (α -arylcinnamic acids, Table 1). Although a half of the examined compounds have been previously described in the literature, their inhibition of AKR1C3 and other isoforms has not been studied so far. These acids have been used in our study, together

with a set of new compounds, to get the information concerning an effect of the substituents, attached to the aromatic rings, on AKR1C3 selectivity and inhibitory activity.

[Table 1]

2. Results and discussion

2.1. Design of new AKR1C3 inhibitors based on a cinnamic acid scaffold

In 2006, some of us discovered that cinnamic acid and several of its simple, substituted derivatives inhibited AKR1C3 with micromolar IC₅₀ values [16a]. Among them, α -methylcinnamic acid was the most promising inhibitor. This prompted us to design the synthesis of a series of cinnamic acids with bulky aryl substituents on the position α (α -arylcinnamic acids) and to evaluate their AKR1C1–AKR1C3 inhibitory activities (Figure 1).

[Fig. 1]

2.2. Chemistry

Several acids were synthesized as described before (1 [20], 2 [21], 3 [21], 5 [21], 6 [21], 7 [21], 8 [21], 9 [21], 10 [21], 11 [21], 12 [21], 16 [22] and 26 [22]). Most of them were recently applied by us as a precursor to pyrazolo-fused combretastatins [21]. Some acids have also been mentioned in the literature (13 [23], 15 [24], 19 [25], 20 [24], 30 [26], 31 [25], 36 [27], and 40 [28]), but the synthesis and characterization details were not available. For this reason they were prepared following our general synthesis and all the missing data are provided (see Supporting information). The latter acids as well as the new ones were available with the Perkin reaction between substituted benzaldehydes and properly functionalized arylacetic acids in acetic anhydride using either potassium acetate or triethylamine as a base that led to the desired 2,3-diarylpropenoic acids as the final products (Scheme 1). Reaction conditions were optimized in all cases. The experiments indicated that the yield of the isolated product depend mainly on the nature of an aromatic aldehyde employed in a particular reaction. Namely, benzaldehydes bearing electron withdrawing groups (e.g., halogen, trifluoromethyl, methylsulfonyl) led to 2,3-diarylpropenoic acids 17, 18, 22, 25, 27, 35-38 in higher yields (63–89%) than those possessing electron donating substituents (e.g., alkyl, methoxy, hydroxy, methylthio) that gave the acids 28-34 and 39 in 21-76% yield. Similar results were reported for the synthesis of various cinnamic acids from substituted benzaldehydes and acetic anhydride [29].

[Scheme 1]

The synthesis of acetoxy-substituted acid **4** required an acetylation of the initially formed (2E)-3-(3-hydroxyphenyl)-2-(3,4,5-trimethoxyphenyl)prop-2-enoic acid applying acetic anhydride in the presence of catalytic amounts of 4-dimethylaminopyridine in DMF as a solvent. The methylsulfonyl- substituted acids **21**, **24**, and **40** were prepared from their methylthio analogues **20**, **23** and **39** using the appropriate reaction conditions to keep an olefinic C=C bond intact. The best results for such selective oxidation of the sulfur atom were obtained employing hydrogen peroxide in acetic acid (Scheme 2). Furthermore, the acid **42** resulted from the Suzuki coupling of **37** with 4-(methylthio)phenylboronic acid (Scheme 3).

[Scheme 2]

[Scheme 3]

2.3. Biological Activity

The compounds shown in Table 1 were assayed for the inhibition of the enzymes AKR1C1– AKR1C3. These enzymes catalyze the oxidation of an artificial substrate 1-acenaphthenol with NAD⁺ as a coenzyme, in the presence and absence of individual compounds. The enzymatic activity was followed spectrophotometrically (Table 2). The results are given as a percentage of the inhibition of the enzyme in the presence of 100 μ M of each acid. For the compounds that showed more than 50% inhibition the IC₅₀ values were determined. Medroxyprogesterone acetate, previously described inhibitor of AKR1C enzymes [14], was assayed on our test as a reference and a positive control. The IC₅₀ values were 1.8 μ M (AKR1C1), 4.1 μ M (AKR1C2), and 0.3 μ M (AKR1C3), respectively.

[Table 2]

A non-substituted 2,3-diphenylpropenoic acid (1) inhibits AKR1C1. Several acids with two or more methoxy groups in the molecule are either not active (2–5, 7–12) or slightly active against AKR1C2 or AKR1C3 (6, 13, 14). We noticed that the introduction of the sulfur-containing substituent increased the inhibitory activity. Thus, out of the four acids 15–18, three of them inhibit at least two enzymes. The acid 19 having the MeSO₂ substituent on the

fragment Ar¹ (i.e., at the position adjacent to the carboxylic functionality) revealed the selective inhibition of AKR1C3. In addition, 20 and 21 inhibit all three enzymes, but the acid 22 is a selective inhibitor of AKR1C3. These results indicate the influence of the sulfurcontaining group. Comparing the activity of the acids 21 and 22 it is evident that the acid 22 containing the 4-MeSO₂ group is a selective inhibitor of AKR1C3 but **21**, possessing the 4chloro substituent, is not. So, they differ only in the presence of the chlorine or the MeSO₂ functionalities. Similar results were obtained in the case of the acids 23–27. Namely, the molecules with a sulfur substituent as a part of the Ar^2 molecules are not selective inhibitors (23, 24 and 26), while the acids possessing the MeSO₂ group at the *para* position of the Ar^1 inhibit AKR1C3 selectively (25 and 27). Having the above results in hand we focused exclusively on the activity of the 2-(4-methylsulfonyl)phenylpropenoic acids 28-39 that differ only in the Ar^2 moiety. Most of them are indeed selective inhibitors of AKR1C3, with the exception of the non-active **31**, **32**, and **34**, as well as **36** that is not a selective inhibitor. The most efficient and selective inhibitors are the 4-bromo-substituted acid 37 and 4-methylthio analogue 39. The structure of the latter compound was further modified. The introduction of the methylsulfonyl functionality in the place of the methylthio group leads to a considerable drop of activity (an acid 40), although the selectivity is retained. On the other hand, 41, bearing the acetylaminosulfonyl moiety at the *para* position of the Ar^2 , is not active at all. The same is true if the para-phenylen linker is introduced to connect the methylthio group and the benzene ring of Ar² (an acid 42). It is also interesting to note that acids 28 and 29, although they exhibit a lower activity against AKR1C3 compared to the above-mentioned 37 and 39, are the most selective inhibitors of this enzyme. Namely, they practically do not inhibit AKR1C1 and AKR1C2 at a 100 µM concentration.

Several important conclusions can be drawn from these results. Firstly, the 4-MeSO₂ group on Ar^{1} seems to be crucial for AKR1C3 selectivity as shown by the activities of the compounds **19**, **22**, **25**, **27–30**, **33** and **35–40** with only compound **36** possessing also a weak AKR1C2 activity. This is consistent with the docking study (see below) predicting that MeSO₂ functionality forms interactions with the SP1 sub-pocket, and with previous studies which reported that the binding to the SP1 sub-pocket was responsible for the isoform selectivity [10]. Secondly, a group on Ar^{2} seems to regulate the activity. The nature of the substituents and their positions are of course important. Here, the bulky hydrophobic substituents seem to be the most appropriate. This can be concluded from a series of halogenated analogs. Namely, the activity is increased with the size of the substituent and is also dependent on a position of the substituent on the aromatic ring. Thus, the fluorine bearing acids **27** and **35** have

inhibitory activities of 43.1 and 38.9 μ M, respectively. Furthermore, the chlorophenyl acids 22 and 25 possess the activities of 10.8 and 16.8 μ M, respectively, and finally with the bromo analogs 36 and 37, which are the most active and have activities of 13.6 and 4.9 μ M, respectively. An enzyme inhibitory potency is therefore increased with the bulk of the substituent and is generally higher in the case of the *para* substituted analogues. A positive influence of the *para* substituent is particularly pronounced with its size. Thus, both fluorinated analogs possess almost the same activities, the *para* substituted chloro analog has 50% better activity as the *meta* one, while the *para* substituted bromo acid has almost 3-fold better activity than the *meta* compound. In addition, the need for a lipophilic substituent is indicated by a good activity of the acids 28, 29, 38 and 39 containing 4-Me, 4-Et, 4-CF₃ or 4-MeS functionality attached to the Ar² ring. On the other hand, and the lack of activity obtained for the compounds 31, 32, and 34, or weak activity in the case of 40, seems to be attributed to the more hydrophilic substituents on the Ar² ring (4-OH, 3-OH, 3,4,5-OMe and MeSO₂).

2.4. Molecular docking

To gain an additional insight into the binding mode and selectivity of the 2,3-diarylpropenoic acids, the compounds with the best AKR1C1 (compound 1) or AKR1C3 (compounds 20, 22, 28, 29, 36, 37, 39) inhibitory activities were docked to the crystal structures of both isoforms. For the docking, the AKR1C1 crystal structure with co-crystallized 3,5-dichlorosalicylic acid (PDB code 3C3U) [30] and the AKR1C3 crystal structure with co-crystallized indomethacin (PDB code 1S2A) [31] were used.

The docking results are in good correlation with the results of the inhibition assays. In the case of the AKR1C1 docking, compound **1** has the highest score, closely followed by compound **20**. In both cases the Ar^1 is aligned with the co-crystallized 3,5-dichlorosalicylic acid, with Ar^2 extending into the steroid-binding channel. The Ar^1 containing a bulky MeSO₂ substituent are unfavorable since they prevent the binding of Ar^1 in the same way as the 3,5-dichlorosalicylic acid, explaining the lack of AKR1C1 activity in those compounds (see Supporting information).

In the case of AKR1C3, the docked poses of the compounds 22, 28, 29, 36, 37 and 39 occupy a similar position of the active site as the co-crystallized indomethacin, with the Ar^1 overlapping with the *p*-chlorobenzoyl moiety of the indomethacin and the Ar^2 overlapping with an indole part of the indomethacin. A typical docking pose is presented in Figure 2 (docking pose for the compound **37**). Due to the nature of AKR1C3, which has a large

number of aromatic residues in a steroid-binding channel, the dominating interactions are predicted to be pi interactions. Ar^1 forms pi interactions with the Phe306 of the SP1 sub-pocket and the Trp227 of the SP2 sub-pocket, while the Ar^2 forms pi interactions with the Phe306 of the SP1. Additional pi interactions are also possible between the propenoic acid double bond and Tyr24, a part of the SP3 sub-pocket. The MeSO₂ group on the *para* position of the Ar^1 seems to be crucial for selectivity and forms important H-bonds with the amino acid residues Ser118 and Asn167 of the SP1 sub-pocket. The differences in the SP1 sub-pockets between the AKR1C isoforms have previously been noted and suggested to be exploited in the design of AKR1C3 selective inhibitors [10].

[Fig. 2]

3. Conclusions

The aldo-keto reductases AKR1C1-AKR1C3 play an important role in the production and inactivation of neurosteroids and prostaglandins as well as in the metabolism of xenobiotics and lipid aldehydes. They can regulate the occupancy and trans-activation of the androgen, estrogen and progesterone receptors. These facts make them important drug targets for the treatment of the hormone-dependent and hormone-independent forms of cancer and other diseases. Several 2,3-diarylpropenoic acids were synthesized by the Perkin reaction between the substituted benzaldehydes and the properly functionalized arylacetic acids. They were evaluated as inhibitors of AKR1C1-AKR1C3. A modification of the structure revealed that the 4-methylsulfonylphenyl substituent at the position 2 of the 2,3-diarylpropenoic acid is an appropriate one to obtain a selective inhibition of AKR1C3. We found that the compounds 37 and **39** are the best inhibitors of AKR1C3 and also show very good selectivity. On the other hand, their 4-methylphenyl and 4-ethylphenyl analogues (28 and 29, respectively), which exhibit a slightly lower activity against AKR1C3, are the most selective ones. Namely, they practically do not inhibit AKR1C1 and AKR1C2 at a 100 µM concentration. Our data suggest that 2,3-diarylpropenoic acids represent a new class of selective inhibitors of AKR1C3 and may serve as a good starting point for the development of new antitumor agents for the treatment of the hormone-dependent and hormone-independent forms of prostate and breast cancers.

4. Experimental protocols

4.1. Chemistry

Starting materials for the synthesis of the examined compounds were used as obtained from the commercial source (Aldrich). Melting points were determined on a Kofler micro hot stage and are uncorrected. NMR spectra were recorded at 29 °C with a Bruker Avance DPX 300 spectrometer (¹H NMR spectra at 300 MHz; ¹³C NMR spectra at 75.5 MHz) and with a Bruker Avance III spectrometer (¹H NMR spectra at 500 MHz; ¹³C NMR spectra at 125.8 MHz). Proton spectra are referenced to TMS as an internal standard; the carbon shifts are given against the central line of the solvent signal (DMSO- d_6 at $\delta = 39.5$ ppm; CDCl₃ at $\delta = 77.1$ ppm). IR spectra were obtained with a Bio-Rad FTS 3000MX (KBr pellets) and a Perkin–Elmer Spectrum 100, equipped with a Specac Golden Gate Diamond ATR as a solid sample support. MS spectra were recorded with a VG-Analytical AutoSpec Q instrument and an Agilent 62224 Accurate Mass TOF LC/MS spectrometer. Elemental analyses (C, H, N) were performed with Perkin Elmer 2400 Series II CHNS/O Analyzer. TLC was carried out on Fluka silica-gel TLC-cards.

A purity of all tested compounds was checked by HPLC (see Supporting information), ¹H NMR and ¹³C NMR spectroscopy as well as with elemental microanalysis. The figures obtained for the C, H, N analyses were always within $\pm 0.35\%$ of the calculated values.

4.1.1. General procedure for the synthesis of acids

Potassium acetate (982 mg, 10 mmol) or triethylamine (1.5 mL, 10.75 mmol; for the synthesis of **13**, **14**, **15**, **18**, **20**, **22**, **23**, and **25**) was added to a stirred mixture of the appropriate arylacetic acid (7.5 mmol), aromatic aldehyde (7.5 mmol) and acetic anhydride (3 mL, 32 mmol). The reaction mixture was stirred at 100–140 °C for 3.5-70 h. Then it was cooled to rt, water was added (40 mL) and the mixture was treated with a solution of 10 M NaOH until pH 12 was reached, followed by conc. HCl (to pH 1). The resulting mixture was extracted with CH₂Cl₂ (3 x 80 mL). Organic phases were extracted with 2 M NaOH (3 x 80 mL) and water (80 mL). The combined aqueous solutions were washed with CH₂Cl₂ (100 mL) to remove the impurities and traces of starting materials, then acidified with conc. HCl (until pH was 1), and finally extracted with CH₂Cl₂ (4 x 50 mL). The methylene chloride phase was washed with brine (100 mL), dried over anhydrous sodium sulfate, evaporated to dryness and the residue was crystallized from a suitable solvent.

4.1.1.1 (2*E*)-3-(3,4-Dimetoxyphenyl)-2-(naphthalen-2-yl)prop-2-enoic acid (14). Reaction time: 48 h; temperature: 100 °C; yield 22%; mp 220–222 °C (EtOAc); IR (KBr) 2955, 1677, 1595, 1512, 1421, 1259, 1143, 1026 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.07 (3H, s), 3.66 (3H, s), 6.50 (1H, d, J = 1.5 Hz), 6.73–6.81 (2H, m), 7.36 (1H, dd, $J_1 = 1.5$ Hz, $J_2 = 8.4$ Hz),

7.47–7.55 (2H, m), 7.78–7.80 (1H, m), 7.82–7.99 (4H, m), 12.62 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 54.3, 55.4, 111.3, 112.9, 124.7, 126.2, 126.9, 127.5, 127.7, 128.00, 128.04, 128.2, 130.6, 132.2, 133.2, 134.5, 139.5, 147.9, 149.8, 168.6; MS (ESI–) m/z 333 ([M - H]⁻, 22); HRMS for C₂₁H₁₇O₄ ([M - H]⁻): calcd 333.1127; found 333.1130. Anal. for C₂₁H₁₈O₄: calcd C, 75.43; H, 5.43; found C, 75.31; H, 5.33.

4.1.1.2 (*2E*)-2-[1,3-Benzodixol-5-yl]-3-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (17). Reaction time: 19 h; temperature: 100 °C; yield 89%; mp 231–232 °C (EtOAc); IR 3021, 2925, 1677, 1618, 1592, 1493, 1441, 1308, 1239, 1146 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.19 (3H, s), 6.05 (2H, s), 6.58 (1H, dd, $J_1 = 7.9$ Hz, $J_2 = 1.6$ Hz), 6.77 (1H, d, J = 1.6 Hz), 6.89 (1H, d, J = 7.9 Hz), 7.35 (2H, d, J = 8.4 Hz), 7.74 (1H, s), 7.77 (2H, d, J = 8.4 Hz), 12.94 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.1, 101.1, 108.5, 109.9, 123.0, 126.8, 128.9, 130.6, 135.9, 136.8, 139.8, 140.3, 146.9, 147.4, 168.0; MS (ESI+) *m*/*z* 347 ([M + H]⁺); HRMS for C₁₇H₁₅O₆S ([M + H]⁺): calcd 347.0584; found 347.0576. Anal. for C₁₇H₁₄O₆S: calcd C, 58.95; H, 4.07; found C, 58.66; H, 3.96.

4.1.1.3 (2*E***)-3-[4-(Acetylsulfamoyl)phenyl]-2-phenylprop-2-enoic acid (18)**. Reaction time: 3.5 h; temperature: 100 °C; yield 63%; mp 238–241 °C (H₂O); IR 3259, 2988, 2901, 1731, 1722, 1659, 1594, 1423, 1348, 1273 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 1.90 (3H, s), 7.16–7.20 (2H, m), 7.25 (2H, d, *J* = 8.4 Hz), 7.36–7.42 (3H, m), 7.69 (2H, d, *J* = 8.4 Hz), 7.81 (1H, s), 12.07 (1H, broad s), 13.1 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 23.3, 127.3, 128.0, 128.6, 129.4, 130.4, 135.5, 136.4, 137.0, 138.9, 139.7, 168.0, 168.9; MS (ESI–) *m/z* 346 ([M + H]⁺, 100 %); Anal. for C₁₇H₁₅NO₅S: C, 59.12; H, 4.38; N, 4.06; found C, 59.36; H, 4.35; N, 3.97.

4.1.1.4 (*2E*)-**3**-(**4**-Chlorophenyl)-**2**-[**4**-(methylsulfonyl)phenyl]prop-**2**-enoic acid (**22**). Reaction time: 70 h; temperature: 140 °C; yield 63%; mp 263–264 °C (EtOAc); IR 2970, 2879, 1667, 1610, 1592, 1564, 1492, 1414, 1393, 1311 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.27 (3H, s), 7.07 (2H, d, *J* = 8.8 Hz), 7.32 (2H, d, *J* = 8.8 Hz), 7.46 (2H, d, *J* = 8.5 Hz), 7.86 (1H, s), 7.93 (2H, d, *J* = 8.5 Hz), 13.05 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 127.2, 128.6, 130.7, 131.8, 132.6, 132.9, 133.9, 138.9, 140.0, 141.4, 167.5; MS (ESI+) *m*/*z* 354 ([M + NH₄]⁺); HRMS for C₁₆H₁₄ClO₄S ([M + H]⁺): calcd 337.0296; found 337.0288. Anal. for C₁₆H₁₃ClO₄S: calcd C, 57.06; H, 3.89; found C, 56.96; H, 3.74.

4.1.1.5 (2*E*)-2-(3-Chlorophenyl)-3-[4-(methylsulfanyl)phenyl]prop-2-enoic acid (23). Reaction time: 6 h; temperature: 100 °C; yield 49%; mp 213–215 °C (EtOAc/*n*-heptane); IR (KBr) 3418, 1668, 1609, 1589, 1426, 1287, 1272, 1191 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.42 (3H, s), 7.00 and 7.09 (4H, AA'BB', J = 8.7 Hz), 7.11–7.18 (1H, m), 7.27 (1H, broad s), 7.38–7.46 (2H, m), 7.77 (1H, s), 12.77 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 14.0, 125.1, 127.6, 128.3, 129.3, 130.2, 130.4, 130.6, 130.8, 133.1, 138.7, 139.3, 140.6, 167.8; MS (ESI–) m/z 303.0 ($[M - H]^{-}$); HRMS for C₁₆H₁₂ClO₂S ($[M - H]^{-}$): calcd 303.0247; found 303.0253. Anal. for C₁₆H₁₃ClO₂S: calcd C, 63.05; H, 4.30; found C, 53.23; H, 4.23.

4.1.1.6 (*2E*)-**3**-(**3**-Chlorophenyl)-**2**-[**4**-(methylsulfonyl)phenyl]prop-**2**-enoic acid (**25**). Reaction time: 41 h; temperature: 100 °C; yield 68%; mp 210–212 °C (EtOAc); IR 2979, 2942, 1674, 1618, 1560, 1478, 1424, 1288, 1211, 1144 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.25 (3H, s), 6.98 (1H, d, *J* = 8.0 Hz), 7.08–7.10 (1H, m), 7.25 (1H, t, *J* = 7.8 Hz), 7.33 (1H, dd, *J*₁ = 8.0 Hz, *J*₂ = 1.5 Hz), 7.47 (2H, d, J = 8.3 Hz), 7.85 (1H, s), 7.94 (2H, d, *J* = 8.3 Hz), 13.11 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 127.2, 128.6, 129.0, 129.7, 130.2, 130.7, 133.1, 133.5, 136.2, 138.6, 140.1, 141.4, 167.3; MS (ESI+) *m*/*z* 354 ([M + NH]⁺); HRMS for C₁₆H₁₄ClO₄S ([M + H]⁺): calcd 337.0296; found 337.0285. Anal. for C₁₆H₁₃ClO₄S: calcd C, 57.06; H, 3.89; found C, 57.21; H, 3.96.

4.1.1.7 (*2E*)-3-(4-Fluorophenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (27). Reaction time: 48 h; temperature: 140 °C; yield 65%; mp 224–226 °C (EtOAc/*n*-hexane); IR 3024, 1667, 1599, 1586, 1508, 1414, 1392, 1304, 1264, 1147 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.26 (3H, s), 7.08–7.13 (4H, m), 7.46 (2H, d, *J* = 8.4 Hz), 7.86 (1H, s), 7.93 (2H, d, *J* = 8.4 Hz), 12.94 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 115.6 (d, *J* = 21.8 Hz), 127.2, 130.5 (d, *J* = 3.2 Hz), 130.7, 131.7, 132.5 (d, *J* = 8.6 Hz), 139.1, 140.0, 141.7, 162.3 (d, *J* = 249.0 Hz), 167.6; MS (ESI+) *m*/*z* 338 ([M + NH₄]⁺); HRMS for C₁₆H₁₄FO₄S ([M + H]⁺): calcd 321.0591; found 321.0596. Anal. for C₁₆H₁₃FO₄S: calcd C, 59.99; H, 4.09; found C, 59.74; H, 3.98.

4.1.1.8 (*2E*)-**3**-(**4**-**Methylphenyl**)-**2**-[**4**-(**methylsulfonyl**)**phenyl**]**prop**-**2**-**enoic acid** (**28**). Reaction time: 22 h; temperature: 120 °C; yield 59%; mp 256–258 °C (EtOAc); IR 2976, 2927, 1684, 1602, 1593, 1563, 1510, 1420, 1275, 1148 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.22 (3H, s), 3.26 (3H, s), 6.93 (2H, d, *J* = 8.3 Hz), 7.03 (2H, d, *J* = 8.3 Hz), 7.45 (2H, d, *J* = 8.3 Hz), 7.84 (1H, s), 7.93 (2H, d, *J* = 8.3 Hz), 12.92 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 20.9, 43.6, 127.2, 129.2, 130.3, 130.8, 130.9, 131.1, 139.4, 139.8, 140.3, 142.2, 167.8; MS (ESI+) *m*/*z* 317 ([M + H]⁺); HRMS for C₁₇H₁₆O₄S ([M + H]⁺): calcd 317.0842; found 317.0837. Anal. for C₁₇H₁₆O₄S: calcd C, 64.54; H, 5.10; found C, 64.52.; H, 5.13.

4.1.1.9 (2*E*)-3-(4-Ethylphenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (29). Reaction time: 20 h; temperature: 120 °C; yield 59%; mp 227–229 °C (EtOAc); IR 3026, 2961, 2868, 1687, 1617, 1603, 1592, 1419, 1274, 1141 cm⁻¹; ¹H NMR (DMSO- d_6) δ 1.10 (3H, t, *J* = 7.8 Hz), 2.52 (2H, q, *J* = 7.8 Hz), 6.96 (2H, d, *J* = 8.5 Hz), 7.08 (2H, d, *J* = 8.5 Hz), 7.46 (2H, d, *J* = 8.3 Hz), 7.83 (1H, s), 7.93 (2H, d, *J* = 8.3 Hz), 12.88 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 15.1, 27.9, 43.5, 127.2, 128.0, 130.4, 130.7, 130.9, 131.3, 139.9, 140.2, 142.2, 145.6, 167.8; MS (ESI+) m/z 331 ([M + H]⁺); HRMS for C₁₈H₁₉O₄S ([M + H]⁺): calcd 331.0999; found 331.0993. Anal. for C₁₈H₁₈O₄S: calcd C, 65.43; H, 5.49; found C, 65.24; H, 5.46.

4.1.1.10 (*2E*)-3-(3-Hydroxyphenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (32). Reaction time: 23 h; temperature: 120 °C; yield 64%; mp 236–238 °C (H₂O/EtOH); IR 3371, 2988, 2901, 1667, 1578, 1470, 1394, 1266, 1137, 1066 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.25 (3H, s), 6.42–6.44 (1H, m), 6.48 (1H, d, *J* = 7.5 Hz), 6.65–6.69 (1H, m), 7.02 (1H, t, *J* = 8.0 Hz), 7.45 (2H, d, *J* = 8.5 Hz), 7.75 (1H, s), 7.92 (2H, d, *J* = 8.5 Hz), 9.48 (1H, broad s), 12.94 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.6, 116.6, 116.8, 121.3, 127.1, 129.5, 130.7, 131.8, 135.1, 139.8, 140.4, 142.0, 157.2, 167.8; MS *m*/*z* 336 ([M + NH₄]⁺); HRMS for C₁₆H₁₅O₅S ([M + H]⁺): calcd 319.0635; found 319.0637. Anal. for C₁₆H₁₄O₅S: calcd C, 60.37; H, 4.43; found C, 60.12; H, 4.33.

4.1.1.11 (*2E*)-3-(3,4-Dihydroxyphenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (33). Reaction time: 50 h; temperature: 120 °C; yield 44%; mp 229–231 °C (H₂O/EtOH); IR 3429, 2937, 1663, 1598, 1524, 1443, 1425, 1262, 1172, 1133 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.26 (3H, s), 6.39 (1H, d, *J* = 2.0 Hz), 6.44 (1H, dd, *J*₁ = 8.0 Hz, *J*₂ = 2.0 Hz), 6.59 (1H, d, *J* = 8.0 Hz), 7.44 (2H, d, *J* = 8.5 Hz) 7.67 (1H, s), 7.93 (2H, d, *J* = 8.5 Hz), 8.98 (1H, broad s), 9.50 (1H, broad s), 12.61 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.6, 115.4, 117.4, 123.6, 125.1, 127.2, 127.9, 130.8, 139.6, 140.8, 142.6, 145.0, 147.5, 168.1; MS (ESI+) *m*/*z* 335 ([M + H]⁺); HRMS for C₁₆H₁₅O₆S ([M + H]⁺): calcd 335.0584; found 335.0579. Anal. for C₁₆H₁₄O₆S: calcd C, 57.48; H, 4.22; found C, 57.41; H, 4.08.

4.1.1.12 (*2E*)-2-[4-(Methylsulfonyl)phenyl]-3-(3,4,5-trimethoxyphenyl)prop-2-enoic acid (34). Reaction time: 19.5 h; temperature: 130 °C; yield 76%; mp 218–220 °C (EtOH); IR 2988, 2842, 1677, 1604, 1578, 1505, 1451, 1417, 1308, 1252 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.24 (3H, s), 3.44 (6H, s), 3.61 (3H, s), 6.35 (2H, s), 7.51 (2H, d, *J* = 8.5 Hz), 7.80 (1H, s), 7.99 (2H, d, *J* = 8.5 Hz), 12.88 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.3, 55.3, 60.1, 108.1, 127.4, 129.0, 130.8, 130.9, 138.5, 140.0, 140.2, 142.5, 152.4, 167.6; MS (ESI+) *m*/*z* 393 ([M + H]⁺); HRMS for C₁₉H₂₁O₇S ([M + H]⁺): calcd 393.1003; found 393.1004. Anal. for C₁₉H₂₀O₇S: calcd C, 58.15; H, 5.14; found C, 58.31; H, 5.21.

4.1.1.13 (2*E*)-3-(3-Fluorophenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (35). Reaction time: 19 h; temperature: 100 °C; yield 75%; mp 201–203 °C (EtOAc); IR 3187, 3002, 1705, 1624, 1582, 1481, 1447, 1393, 1275, 1134 cm⁻¹; ¹H NMR (DMSO- d_6) δ 3.26 (3H, s), 6.84–6.89 (2H, m), 7.12 (1H, dt, $J_1 = 8.5$ Hz, $J_2 = 2.5$ Hz), 7.25–7.30 (1H, m), 7.47 (2H, d, J = 8.5 Hz), 7.86 (1H, s), 7.94 (2H, d, J = 8.5 Hz), 13.10 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 116.1 (d, J = 21.3 Hz), 116.5 (d, J = 22.4 Hz), 126.3 (d, J = 2.6 Hz), 127.2, 130.5 (d, J = 8.4 Hz), 130.7, 133.4, 136.4 (d, J = 8.1 Hz), 138.7, 140.1, 141.5, 161.7 (d, J = 244.2 Hz), 167.4; MS ESI–) m/z 319 ([M - H]⁻); HRMS for C₁₆H₁₂FO₄S ([M - H]⁻): calcd 319.0446; found 319.0447. Anal. for C₁₆H₁₃FO₄S: calcd C, 59.99; H, 4.09; found C, 59.70; H, 3.82.

4.1.1.14 (2*E*)-3-(4-Bromophenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (37). Reaction time: 19.5 h; temperature: 120 °C; yield 68%; mp 261–262 °C (H₂O); IR 2971, 1672, 1609, 1583, 1489, 1406, 1311, 1274, 1259, 1147 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.27 (3H, s), 6.99 (2H, d, *J* = 8.5 Hz), 7.44–7.48 (4H, m), 7.46 (2H, d, *J* = 8.3 Hz), 7.83 (1H, s), 7.93 (2H, d, *J* = 8.3 Hz), 13.03 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 122.8, 127.2, 130.7, 131.5, 132.0, 132.7, 133.2, 139.0, 140.0, 131.4, 167.5; MS (ESI+) *m/z* 381 ([M + H]⁺); HRMS for C₁₆H₁₄BrO₄S ([M + H]⁺): calcd 380.9791; found 380.9791. Anal. for C₁₆H₁₃BrO₄S: calcd C, 50.41; H, 3.44; found C, 50.26; H, 3.19.

4.1.1.15 (*2E*)-2-[4-(Methylsulfonyl)phenyl]-3-[4-(trifluoromethyl)phenyl]prop-2-enoic acid (38). Reaction time: 18.5 h; temperature: 120 °C; yield 66%; mp 218–221 °C (EtOAc); IR 2988, 1682, 1620, 1595, 1416, 1394, 1323, 1307, 1271, 1148 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.26 (3H, s), 7.26 (2H, d, *J* = 8.5 Hz), 7.48 (2H, d, *J* = 8.3 Hz), 7.61 (2H, d, *J* = 8.5 Hz), 7.93– 7.95 (3H, m), 13.20 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 123.9 (q, *J* = 272.7 Hz), 125.3 (q, *J* = 3.7 Hz), 127.2, 129.0 (q, *J* = 32.0 Hz), 130.69, 130.71, 134.4, 138.2 (q, *J* = 1.0 Hz), 138.6, 140.1, 141.1, 167.3; MS (ESI+) *m/z* 388 ([M + NH₄]⁺); HRMS for C₁₇H₁₄F₃O₄S ([M + H]⁺): calcd 371.0559; found 371.0559. Anal. for C₁₇H₁₃F₃O₄S: calcd C, 55.13; H, 3.54; found C, 55.15; H, 3.43.

4.1.1.16 (2*E*)-3-[4-(Methylsulfanyl)phenyl]-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (39). Reaction time: 20 h; temperature: 120 °C; yield 53%; mp 238–240 °C (EtOAc/*n*-heptane); IR (KBr) 3416, 2927, 1665, 1586, 1410, 1309, 1251, 1190, 1146, 1088 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.42 (3H, s), 3.26 (3H, s), 6.98 (2H, d, *J* = 8.7 Hz), 7.10 (2H, d, *J* = 8.7 Hz), 7.47 (2H, d, *J* = 8.4 Hz), 7.81 (1H, s), 7.94 (2H, d, *J* = 8.4 Hz), 12.83 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 14.0, 43.5, 125.1, 127.2, 130.0, 130.68, 130.69, 130.72, 139.7, 139.9, 140.8, 142.1, 167.7; MS (ESI+) *m*/*z* 349 ([M + H]⁺); HRMS for C₁₇H₁₇O₄S₂ ([M + H]⁺): calcd 349.0568; found 349.0555. Anal. for C₁₇H₁₆O₄S₂: calcd C, 58.60; H, 4.63; found C, 58.83; H, 4.79.

4.1.1.17 (2*E*)-**3-[4-(Acetylsulfamoyl)phenyl]-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (41).** Reaction time: 4.5 h; temperature: 110 °C; yield 25% (column chromatography);

mp 177.5–179.0 °C (H₂O/EtOH); IR 3342, 2988, 2901, 1721, 1689, 1607, 1470, 1406, 1287, 1145 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 1.91 (3H, s), 3.27 (3H, s), 7.28 (2H, d, *J* = 8.5 Hz), 7.47 (7H, d, *J* = 8.5 Hz), 7.74 (2H, d, *J* = 8.5 Hz), 7.92 (1H, s) 7.93 (2H, d, *J* = 8.5 Hz), 12.12 (1H, broad s), 13.20 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 23.2, 43.5, 127.2, 127.5, 130.5, 130.6, 134.7, 138.3, 139.0, 139.2, 140.2, 140.9, 167.3, 168.9; MS (ESI+) *m/z* 424 ([M + H]⁺); HRMS for C₁₈H₁₈NO₇S₂ ([M + H]⁺): calcd 424.0519; found 424.0515. Anal. for C₁₈H₁₇NO₇S₂ x ²/₃ H₂O: calcd C, 49.65; H, 4.24; N, 3.22; found C, 49.34.; H, 4.08.; N, 3.18.

4.1.2. Synthesis of (2*E*)-3-(3-acetoxyphenyl)-2-(3,4,5-trimethoxyphenyl)prop-2-enoic acid (4). Applying a general method described above, (2*E*)-3-(3-hydroxyphenyl)-2-(3,4,5-trimethoxyphenyl)prop-2-enoic acid was obtained from 3-hydroxybenzaldehyde and (3,4,5-trimethoxyphenyl)acetic acid using triethylamine as a base; reaction time: 45 h; temperature: 100 °C; yield 42%; mp 220–225 °C (EtOAc); IR (KBr) 3447, 2941, 1678, 1587, 1413, 1310, 1271, 1127 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.68 (6H, s), 3.71 (3H, s), 6.46 (2H, s), 6.53–6.59 (2H, m), 6.67–6.70 (1H, m), 7.02–7.05 (1H, m), 7.64 (1H, s), 9.40 (1H, s), 12.60 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 56.0, 60.1, 106.9, 116.3, 116.9, 121.3, 129.2, 131.7, 133.0, 135.7, 137.1, 139.1, 153.0, 157.1, 168.4; MS (ESI–) *m/z* 329 ([M - H]⁻, 68); HRMS for C₁₈H₁₇O₆: calcd 329.1025; found 329.1028. Anal. for C₁₈H₁₈O₆: calcd C, 65.45; H, 5.49; found C, 65.33; H, 5.51.

To the solution of (2*E*)-3-(3-hydroxyphenyl)-2-(3,4,5-trimethoxyphenyl)prop-2-enoic acid (3.96 g, 12 mmol) in DMF (12 mL) were added 4-dimethylaminopyridine (30 mg, 0.245 mmol) and acetic anhydride (1.73 mL, 18 mmol). The reaction mixture was stirred at rt for 48 h, treated with water (36 mL) and extracted with EtOAc:THF 3:1 (3 x 60 mL). The combined organic phases were washed with 5% HCl, dried over anhydrous sodium sulfate and evaporated to dryness to give the acid **4**. Yield 64%; mp 186.9–189.7 °C (EtOAc); IR 2946, 1683, 1582, 1412, 1289, 1240, 1211, 1128 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.18 (3H, s), 3.67 (6H, s), 3.71 (3H, s), 6.46 (2H, s), 6.77–6.79 (1H, m), 7.01–7.05 (2H, m), 7.26–7.31 (1H, m), 7.72 (1H, s), 12.73 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 20.6, 55.9, 60.1, 106.8, 122.5, 123.1, 127.7, 129.3, 131.3, 134.3, 135.9, 137.3, 137.7, 150.2, 153.1, 168.1, 168.8; MS (ESI–) *m*/*z* 371 ([M - H]⁻, 10); HRMS for C₂₀H₁₉O₇ ([M - H]⁻): calcd 371.1131; found 371.1123.

4.1.3. Oxidation of the methylthio moiety into methylsulfonyl group

Hydrogen peroxide (30%, 30 mmol, 3.1 mL) was added to a stirred suspension of the propenoic acid **20** or **23** (6 mmol) in acetic acid (60 mL) at 0 °C and the reaction mixture was stirred at rt for 48 h. Then, it was poured into cold water (300 mL) and the precipitate was filtered off to give the product **21** or **24**.

4.1.3.1. (2*E*)-2-(4-Chlorophenyl)-3-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (21). Yield 73%; mp 231–234 °C (AcOH/H₂O); IR (KBr) 3414, 2928, 1688, 1433, 1312, 1303, 1271, 1152 cm⁻¹; ¹H NMR (DMSO- d_6) δ 3.19 (3H, s), 7.22 and 7.44 (4H, AA'XX', J = 8.4 Hz), 7.33 and 7.79 (4H, AA'XX', J = 8.4 Hz), 7.88 (1H, s), 13.03 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 43.2, 126.8, 128.6, 130.6, 131.5, 132.8, 134.3, 135.1, 137.6, 139.4, 140.5, 167.6; MS (ESI–) *m*/*z* 335.0 ([M - H]⁻, 30); HRMS for C₁₆H₁₂ClO₄S ([M - H]⁻): calcd 335.0145; found 335.0150. Anal. for C₁₆H₁₃ClO₄S: calcd C, 57.06; H, 3.89; found C, 57.32; H, 3.66.

4.1.3.2. (2*E*)-2-(3-Chlorophenyl)-3-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (24). Yield 74%; mp 205–207 °C (AcOH/H₂O); IR (KBr) 3418, 1668, 1609, 1589, 1426, 1287, 1272, 1191 cm⁻¹; ¹H NMR (DMSO- d_6) δ 3.19 (3H, s), 7.12–7.16 (1H, m), 7.30–7.47 (5H, m), 7.77–7.80 (2H, m), 7.89 (1H, s), 13.06 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 43.2, 126.8, 128.0, 128.3, 129.3, 130.4, 130.7, 133.1, 134.8, 137.6, 137.9, 139.3, 140.6, 167.4; MS (ESI–) m/z 335.0 ([M - H]⁻); HRMS for C₁₆H₁₂ClO₄S ([M - H]⁻): calcd 335.0145; found 335.0141. Anal. for C₁₆H₁₃ClO₄S: calcd C, 57.06; H, 3.89; found C, 57.17; H, 3.76.

4.1.4. Suzuki coupling to obtain (2*E*)-3-[4'-(methylsulfanyl)biphenyl-4-yl]-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (42). An aqueous solution of 2 M potassium carbonate (4.0 mL), triphenylphosphine (32 mg, 0.12 mmol) and Pd(OAc)₂ (7 mg, 0.03 mmol) were added to a solution of (2*E*)-3-(4-bromophenyl)-2-[4-(methylsulfonyl)-phenyl]prop-2-enoic acid (37; 381 mg, 1.0 mmol) and 4-(methylthio)phenylboronic acid (252 mg, 1.5 mmol) in dimethoxyethane (10 mL) under an argon atmosphere. The reaction mixture was heated at 100 °C for 17 h, cooled to rt and the solvent was removed under reduced pressure. Dichloromethane (30 mL) and water (20 mL) were added and the mixture was acidified with 1 M aqueous HCl to pH 3. Layers were separated and the water phase was additionally extracted with dichloromethane (3 x 20 mL). The combined organic layers were washed with brine (2 x 30 mL), dried over anhydrous sodium sulfate and evaporated to dryness giving the acid **42**. Yield 72%; mp 291–293 °C (EtOAc); IR 2988, 2923, 1666, 1590, 1486, 1425, 1396, 1307, 1275, 1150 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.49 (3H, s), 3.28 (3H, s), 7.13 (2H, d, *J* = 8.5 Hz), 7.30 (2H, d, *J* = 8.5 Hz), 7.51 (2H, d, *J* = 8.3 Hz), 7.56 (2H, d, *J* =

8.5 Hz), 7.61 (2H, d, J = 8.5), 7.90 (1H, s), 7.96 (2H, d, J = 8.3 Hz), 12.98 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 14.5, 43.5, 126.2, 127.0, 127.2, 130.7, 131.0, 131.6, 132.7, 135.2, 138.3, 139.7, 139.9, 140.1, 142.0, 167.7 (one signal missing); MS (ESI+) *m*/*z* 423 ([M + H]⁺); HRMS for C₂₃H₂₁O₄S₂ ([M + H]⁺): calcd 425.0876; found 425.0876. Anal. for C₂₃H₂₀O₄S₂ x 0.5 H₂O: calcd C, 63.72; H, 4.88; found C, 63.56; H, 4.66.

4.2. Molecular docking

A multi-conformer library of 3D structures of the compounds 1, 20, 22, 28, 29, 36, 37 and 39 was prepared with the program Omega (OpenEye Scientific Software Inc.). For the preparation of 3D conformations the default parameters were used, except that the maximum number of conformations per compound was increased to 300. Active sites of AKR1C1 (PDB code 3C3U) [30] and AKR1C3 (PDB code 1S2A) [31] were prepared with the program fred_receptor (OpenEye Scientific Software Inc.). Co-crystallized cofactors were retained in the structures. An active site was in both cases defined as a box around co-crystallized inhibitor, 3,5-dichlorosalicylic acid in the case of AKR1C1 and indomethacin in the case of AKR1C3. A docking protocol was validated by re-docking of co-crystallized inhibitors. Subsequently, the multi-conformer library of the compounds 1, 20, 22, 28, 29, 36, 37 and 39 was docked with FRED [32] in both enzymes using chemgauss3 as a scoring function.

4.3. Inhibition Assays

Recombinant enzymes AKR1C1-AKR1C3 were prepared as described before [33]. These enzymes *in vitro* catalyze the oxidation of the 1-acenaphthenol in the presence of the coenzyme NAD⁺ and this reaction was followed spectrophotometrically by measuring the increase in NADH absorbance ($\varepsilon_{\lambda 340} = 6220 \text{ M}^{-1} \text{ cm}^{-1}$) in the presence and absence of each of the compounds. The assays were carried out in a 0.3 mL volume that included a 100 mM phosphate buffer (pH 9.0), 0.005% triton X-114 and 5% DMSO as a co-solvent. A substrate concentration of 30 μ M (K_m), 50 μ M (K_m) and 100 μ M (< K_m) and an enzyme concentration of 0.3 μ M, 0.2 μ M, and 1.5 μ M were used for assays with AKR1C1, AKR1C2 and AKR1C3, respectively, in the presence of 2.3 mM coenzyme. Screening was performed for 100 μ M compounds. For the compounds that showed more than 50% inhibition, the IC₅₀ values were determined. The measurements were performed on Biotek PowerWave XS microplate readers with initial velocities calculated, and the IC₅₀ values were determined graphically from plots of log₁₀ [inhibitor concentration] versus % inhibition, using GraphPad Prism Version 4.00

(GraphPad Software, Inc.). Ki values were then calculated using the Cheng-Prusoff equation for competitive inhibition.

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Appendix A. Supporting information

Supplementary material associated with this article can be found, in the online version, at http://...... It includes the characterization data of the compounds 13, 15, 19, 20, 30, 31, 36, and 40 as well as the predicted binding pose of the compound 1 in the AKR1C1 active site and the data about a purity of all tested acids checked by HPLC.

References

- [1] T.M. Penning, M.E. Burczynski, J.M. Jez, C.F. Hung, H.K. Lin, H. Ma, M. Moore, N. Palackal, K. Ratnam, Human 3α-hydroxysteroid dehydrogenase isoforms (AKR1C1–AKR1C4) of the aldo-keto reductase superfamily: functional plasticity and tissue distribution reveals roles in the inactivation and formation of male and female sex hormones, Biochem. J. 351 (2000) 67–77.
- [2] T.M. Penning, M.C. Byrns, Steroid hormone transforming aldo-keto reductases and cancer, Ann. N. Y. Acad. Sci. 1155 (2009) 33–42.
- [3] T.M. Penning, J.E. Drury, Human aldo-keto reductases: function, gene regulation, and single nucleotide polymorphisms, Arch. Biochem. Biophys. 464 (2007) 241–250.

- [4] I. Dufort, P. Rheault, X.F. Huang, P. Soucy, V. Luu-The, Characteristics of a highly labile human type 5 17β-hydroxysteroid dehydrogenase, Endocrinology 140 (1999) 568– 574.
- [5] M.C. Byrns, L. Duan, S.H. Lee, I.A. Blair, T.M. Penning, Aldo-keto reductase 1C3 expression in MCF-7 cells reveals roles in steroid hormone and prostaglandin metabolism that may explain its over-expression in breast cancer, J. Steroid Biochem. Mol. Biol. 118 (2010) 177–187.
- [6] R.B. Montgomery, E.A. Mostaghel, R. Vessella, D.L. Hess, T.F. Kalhorn, C.S. Higano, L.D. True, P.S. Nelson, Maintenance of intratumoral androgens in metastatic prostate cancer: a mechanism for castration-resistant tumor growth, Cancer Res. 68 (2008) 4447– 4454.
- [7] T. Lanišnik-Rižner, T. Šmuc, R. Rupreht, J. Šinkovec, T.M. Penning, AKR1C1 and AKR1C3 may determine progesterone and estrogen ratios in endometrial cancer, Mol. Cell. Endocrinol. 248 (2006) 126–135.
- [8] T. Šmuc, N. Hevir, M. Ribič-Pucelj, B. Husen, H. Thole, T.L. Rižner, Disturbed estrogen and progesterone action in ovarian endometriosis, Mol. Cell. Endocrinol. 301 (2009) 59–64.
- [9] P. Brožič, S. Turk, T.L. Rižner, S. Gobec, Inhibitors of Aldo-Keto reductases AKR1C1– AKR1C4, Curr. Med. Chem. 18 (2011) 2554–2565.
- [10] M.C. Byrns, Y. Jin, M.P. Penning, Inhibitors of type 5 17β-hydroxysteroid dehydrogenase (AKR1C3): Overview and structural insights, J. Steroid Biochem. Mol. Biol. 125 (2011) 95–104.
- [11] N. Beranič, S. Gobec, T.L. Rižner, Progestins as inhibitors of the human 20-ketosteroid reductases, AKR1C1 and AKR1C3, Chem. Biol. Interact. 191 (2011) 227–233.
- [12] P. Bydal, V. Luu-The, F. Labrie, D. Poirier, Steroidal lactones as inhibitors of 17βhydroxysteroid dehydrogenase type 5: chemical synthesis, enzyme inhibitory activity, and assessment of estrogenic and androgenic activities, Eur. J. Med. Chem. 44 (2009) 632–644.
- [13] B. Štefane, P. Brožič, M. Vehovc, T.L. Rižner, S. Gobec, New cyclopentane derivatives as inhibitors of steroid metabolizing enzymes AKR1C1 and AKR1C3, Eur. J. Med. Chem. 44 (2009) 2563–2571.
- [14] Y. Higaki, N. Usami, S. Shintani, S. Ishikura, O. El-Kabbani, A. Hara, Selective and potent inhibitors of human 20α-hydroxysteroid dehydrogenase (AKR1C1) that

metabolizes neurosteroids derived from progesterone, Chem. Biol. Interact. 143–144 (2003) 503–513.

- [15] L. Skarydova, L. Zivna, G. Xiong, E. Maser, V. Wsol, AKR1C3 as a potential target for the inhibitory effect of dietary flavonoids, Chem. Biol. Interact. 178 (2009) 138–144.
- [16] (a) P. Brožič, B. Golob, N. Gomboc, T. Lanišnik-Rižner, S. Gobec, Cinnamic acids as new inhibitors of 17β-hydroxysteroid dehydrogenase type 5 (AKR1C3), Mol. Cell. Endocrinol. 248 (2006) 233–235;
 (b) S. Endo, T. Matsunaga, A. Kanamori, Y. Otsuji, H. Nagai, K. Sundaram, O. El-

Kabbani, N. Toyooka, S. Ohta, A. Hara, Selective inhibition of human type-5 17βhydroxysteroid dehydrogenase (AKR1C3) by baccharin, a component of brazilian propolis, J. Nat. Prod. 75 (2012) 716–721.

- [17] N.J. Davies, R.E. Hayden, P.J. Simpson, J. Birtwistle, K. Mayer, J.P. Ride, C.M. Bunce, AKR1C isoforms represent a novel cellular target for jasmonates alongside their mitochondrial-mediated effects, Cancer Res. 69 (2009) 4769–4775.
- [18] M.C. Byrns, S. Steckelbroeck, T.M. Penning, An indomethacin analogue, N-(4chlorobenzoyl)-melatonin, is a selective inhibitor of aldo-keto reductase 1C3 (type 2 3α -HSD, type 5 17β -HSD, and prostaglandin F synthase), a potential target for the treatment of hormone dependent and hormone independent malignancies, Biochem. Pharmacol. 75 (2008) 484–493.
- [19] (a) A.O. Adeniji, B.M. Twenter, M.C. Byrns, Y. Jin, J.D. Winkler, T.M. Penning, Discovery of substituted 3-(phenylamino)benzoic acids as potent and selective inhibitors of type 5 17β-hydroxysteroid dehydrogenase (AKR1C3), Bioorg. Med. Chem. Lett. 21 (2011) 1464–1468;

(b) A.O. Adeniji, B.M. Twenter, M.C. Byrns, Y. Jin, M. Chen, J.D. Winkler, T.M. Penning, Development of potent and selective inhibitors of aldo-keto reductase 1C3 (type 5 17β -hydroxysteroid dehydrogenase) based on N-phenyl-aminobenzoates and their structure-activity relationships, J. Med. Chem. 55 (2012) 2311–2323.

- [20] S. Li, W. Yuan, S. Ma, Highly regio- and stereoselective three-component nickelcatalyzed syn-hydrocarboxylation of alkynes with diethyl zinc and carbon dioxide, Angew. Chem. Int. Ed. 50 (2011) 2578–2582.
- [21] B. Burja, T. Cimbora-Zovko, S. Tomić, T. Jelušić, M. Kočevar, S. Polanc, M. Osmak, Pyrazolone-fused combretastatins and their precursors: synthesis, cytotoxicity, antitubulin activity and molecular modeling studies, Bioorg. Med. Chem. 18 (2010) 2375–2387.

- [22] A. Moreau, O.-H. Chen, P.N.P. Rao, E.E. Knaus, Design, synthesis, and biological evaluation of (E)-3-(4-methanesulfonylphenyl)-2-(aryl)acrylic acids as dual inhibitors of cyclooxygenases and lipoxygenases, Bioorg. Med. Chem. 14 (2006) 7716–7727.
- [23] C. Zhu, Y. Zhang, G. Cao, S. Peng, W. Wang, J. Zheng, Studies on the nonsteroidal antifertility agents. II. Synthesis and antifertility activity of some p-coumaric acid derivatives, Nanjing Yaoxueyuan Xuebao (1982) 50–56. Chem. Abstr. 99 (1983) 212230.
- [24] J. J. Talley, J. S. Carter, P. W. Collins, S.W. Kramer, T. D. Penning, D.J. Rogier Jr., R.S. Rogers, Substituted thiazoles for the treatment of inflammation, WO 9603392 A1; Chem. Abstr. 125 (1996) 33628.
- [25] J.-L. Mao, X.-K. Ran, J.-Z. Tian, B. Jiao, H.-L. Zhou, L. Chen, Z.-G. Wang, Design, synthesis and biological evaluation of novel 4-hydroxybenzeneacrylic acid derivatives, Bioorg. Med. Chem. Lett. 21 (2011) 1549–1553.
- [26] X. Ran, J. Mao, H. Pang, L. Chen, B. Jiao, Preparation of 3-phenyl-2-(4mesylphenyl)acrylates as cyclooxygenase-2 inhibitors for treatment of inflammation and pain, Faming Zhuanli Shenqing Gongkai Shuomingshu, Patent CN 1762994 (April 26, 2006); Chem. Abstr. 145 (2006) 62680.
- [27] D. Macdonald, A. Mastracchio, H. Perrier, D. Dube, M. Gallant, C. Li, L.A. Trimble, S. Day, N. Chauret, D.A. Nicoll-Griffith, J.M. Silva, Z. Huang, F. Laliberte, S. Liu, D. Ethier, D. Pon, E. Muise, L. Boulet, C.C. Chan, A. Styhler, S. Charleson, J. Mancini, P. Masson, D. Claveau, D. Nicholson, M. Turner, R.N. Young, Y. Girard, Discovery of a substituted 8-arylquinoline series of PDE4 inhibitors: structure-activity relationship, optimization, and identification of a highly potent, well tolerated, PDE4 inhibitor, Bioorg. Med. Chem. Lett. 15 (2005) 5241–5246.
- [28] F. Wang, S. Li, W. Feng, Anti-inflammatory structure-activity relationships of a new series of open-loop COX-2 inhibitors derived from rofecoxib, Beijing Huagong Daxue Xuebao, Ziran Kexueban 31 (2004) 84–87, Chem. Abstr. 142 (2005) 385056.
- [29] T. Rosen, The Perkin reaction, in: B. Trost, I. Fleming (Eds.), Comprehensive Organic Synthesis, vol. 2, Pergamon Press, Oxford, 1991, pp. 395–408.
- [30] U. Dhagat, S. Endo, R. Sumii, A. Hara, O. El-Kabbani, Selectivity determinants of inhibitor binding to human 20α-hydroxysteroid dehydrogenase: crystal structure of the enzyme in ternary complex with coenzyme and the potent inhibitor 3,5-dichlorosalicylic acid, J. Med. Chem. 51 (2008) 4844–4848.

- [31] A.L. Lovering, J.P. Ride, C.M. Bunce, J.C. Desmond, S.M. Cummings, S.A. White, Crystal structures of prostaglandin D2 11-ketoreductase (AKR1C3) in complex with the nonsteroidal anti-inflammatory drugs flufenamic acid and indomethacin, Cancer Res. 64 (2004) 1802–1810.
- [32] M. McGann, FRED pose prediction and virtual screening accuracy, J. Chem. Inf. Model. 51 (2011) 578–596.
- [33] P. Brožič, T. Šmuc, S. Gobec, T. Lanišnik-Rižner, Phytoestrogens as inhibitors of the human progesterone metabolizing enzyme AKR1C1, Mol. Cell. Endocrinol. 259 (2006) 30–42.

List of captions

Fig. 1. Design of 2,3-diarylpropenoic acids.

Fig. 2. Predicted binding pose of the compound 37 (blue) in the AKR1C3 active site (green).

Scheme 1. Synthesis of 2,3-diarylpropenoic acids employing the Perkin reaction.

Scheme 2. Selective oxidation of the methylthio groups of the acids 20, 23 and 39.

Scheme 3. Synthesis of the acid 42.

Table 1. Structures of 2,3-diarylpropenoic acids 1–42.

 Table 2. Effect of 2,3-diarylpropenoic acids 1–42 on the inhibition of AKR1C1–AKR1C3.



Fig. 1. Design of 2,3-diarylpropenoic acids.



Fig. 2. Predicted binding pose of the compound **37** (blue) in the AKR1C3 active site (green). Only relevant amino acid residues are shown and labeled. The co-crystallized indomethacin is shown as magenta sticks and the cofactor as orange sticks. The H-bonds are shown as yellow dashes.



Reagents and conditions: arylacetic acid (7.5 mmol), aromatic aldehyde (7.5 mmol), acetic anhydride (32 mmol), potassium acetate (10 mmol) or triethylamine (10.75 mmol), 100–140 °C, 3.5–70 h.

Scheme 1. Synthesis of 2,3-diarylpropenoic acids employing the Perkin reaction.



Reagents and conditions: 2,3-diarylpropenoic acid (6 mmol), hydrogen peroxide (30%, 30 mmol, 3.1 mL), acetic acid (60 mL), rt, 48 h.

Scheme 2. Selective oxidation of the methylthio groups of the acids 20, 23 and 39.



Reagents and conditions: acid **37** (1 mmol), 4-(methylthio)phenylboronic acid (1.5 mmol), potassium carbonate (4 mL of 2 M aqueous solution), Ph₃P (0.12 mmol), Pd(OAc)₂ (0.03 mmol), DME (10 mL), argon, 100 °C, 17 h.

Scheme 3. Synthesis of the acid 42.

Structures of 2,3-diarylpropenoic acids 1–42.

Acid	Ar ¹	Ar ²	Acid	Ar ¹	Ar ²	
1	C ₆ H ₅	C ₆ H ₅	22	4-MeSO ₂ -C ₆ H ₄	4-Cl-C ₆ H ₄	
2	3,4,5-MeO-C ₆ H ₂	3-OH,4-MeO-C ₆ H ₃	23	$3-Cl-C_6H_4$	4-MeS-C ₆ H ₄	
3	3,4,5-MeO-C ₆ H ₂	3-OAc,4-MeO-C ₆ H ₃	24	$3-Cl-C_6H_4$	4-MeSO ₂ -C ₆ H ₄	
4	3,4,5-MeO-C ₆ H ₂	3-OAc-C ₆ H ₄	25	4-MeSO ₂ -C ₆ H ₄	$3-Cl-C_6H_4$	
5	3,4,5-MeO-C ₆ H ₂	3-F,4-MeO-C ₆ H ₃	26	4-F-C ₆ H ₄	4-MeSO ₂ -C ₆ H ₄	
6	3,4,5-MeO-C ₆ H ₂	$3-F-C_6H_4$	27	4-MeSO ₂ -C ₆ H ₄	$4-F-C_6H_4$	
7	3,4,5-MeO-C ₆ H ₂	3,4-MeO-C ₆ H ₃	28	4-MeSO ₂ -C ₆ H ₄	$4-\text{Me-C}_6\text{H}_4$	
8	3,4,5-MeO-C ₆ H ₂	$4-\text{MeO-C}_6\text{H}_4$	29	4-MeSO ₂ -C ₆ H ₄	$4-\text{Et-}C_6H_4$	
9	3,4,5-MeO-C ₆ H ₂	1-naphthyl	30	4-MeSO ₂ -C ₆ H ₄	4-MeO-C ₆ H ₄	
10	3,4,5-MeO-C ₆ H ₂	2-naphthyl	31	4-MeSO ₂ -C ₆ H ₄	$4-HO-C_6H_4$	
11	3,4-MeO-C ₆ H ₃	3,4,5-MeO-C ₆ H ₂	32	4-MeSO ₂ -C ₆ H ₄	3-HO-C ₆ H ₄	
12	3,4-MeO-C ₆ H ₃	2,4,5-MeO-C ₆ H ₂	33	4-MeSO ₂ -C ₆ H ₄	3,4-HO-C ₆ H ₃	
13	1-naphthyl	3,4-MeO-C ₆ H ₃	34	4-MeSO ₂ -C ₆ H ₄	3,4,5-MeO-C ₆ H ₂	
14	2-naphthyl	3,4-MeO-C ₆ H ₃	35	4-MeSO ₂ -C ₆ H ₄	$3-F-C_6H_4$	
15	C ₆ H ₅	$4-\text{MeS-C}_6\text{H}_4$	36	$4-MeSO_2-C_6H_4$	$3-Br-C_6H_4$	
16	C ₆ H ₅	$4-MeSO_2-C_6H_4$	37	$4-MeSO_2-C_6H_4$	4-Br-C ₆ H ₄	
17	3,4-OCH ₂ O-C ₆ H ₃	$4-MeSO_2-C_6H_4$	38	$4-MeSO_2-C_6H_4$	$4-CF_3-C_6H_4$	
18	C ₆ H ₅	4-MeCONHSO ₂ -C ₆ H ₄	39	$4-MeSO_2-C_6H_4$	$4-MeS-C_6H_4$	
19	$4-MeSO_2-C_6H_4$	C ₆ H ₅	40	$4-MeSO_2-C_6H_4$	$4-MeSO_2-C_6H_4$	
20	4-Cl-C ₆ H ₄	$4-\text{MeS-C}_6\text{H}_4$	41	$4-MeSO_2-C_6H_4$	4-MeCONHSO ₂ -C ₆ H ₄	
21	4-Cl-C ₆ H ₄	$4-MeSO_2-C_6H_4$	42	$4-MeSO_2-C_6H_4$	$4-(4-MeS-C_6H_4)-C_6H_4$	
	A C					

	Inhibition of		Inhibition of		Inhibition of	
	AKR1C1 ^a	IC_{50}	$AKR1C2^{a}$	IC_{50}	AKR1C3 ^a	IC_{50}
Acid	%	μM	%	μM	%	μM
1	90.1 ± 1.3	7.9 ± 0.9	47.5 ± 4.0		29.1 ± 2.8	
2	14.4 ± 8.1		20.6 ± 4.5		51.7 ± 4.3	121.1 ± 0.3
3	8.0 ± 0.2		3.2 ± 2.2		8.8 ± 2.0	
4	40.0 ± 14.8		27.0 ± 22.3		42.6 ± 2.3	
5	-5.2 ± 6.4		9.9 ± 1.1		10.3 ± 0.4	
6	7.8 ± 3.7		55.9 ± 3.1	77.0 ± 0.3	12.2 ± 2.4	
7	6.1 ± 0.1		0.6 ± 4.4		4.2 ± 0	
8	15.1 ± 1.1		4.3 ± 0.3		10.3 ± 0.4	
9	6.4 ± 8.3		13.8 ± 0.8		14.8 ± 3.5	
10	9.6 ± 8.6		15.9 ± 2.2		30.8 ± 2.6	
11	7.1 ± 1.7		4.2 ± 5.6		16.1 ± 4.6	
12	22.2 ± 6.8		9.4 ± 5.6		7.0 ± 0	
13	29.4 ± 6.8		77.8 ± 0.4	24.2 ± 0.1	59.7 ± 6.2	127.8 ± 0.3
14	9.7 ± 1.5		39.6 ± 9.5		78.3 ± 1.3	38.7 ± 0.2
15	87.0 ± 8.5	12.3 ± 0.0	59.1 ± 3.4	41.2 ± 0.2	92.8 ± 4.9	30.2 ± 0.2
16	77.6 ± 2.0	$\textbf{30.8} \pm \textbf{0.2}$	61.4 ± 0.3	34.4 ± 0.2	52.3 ± 3.2	139.2 ± 0.3
17	45.2 ± 5.4		24.2 ± 9.4		35.8 ± 6.8	
18	97.8 ± 0.3	8.7 ± 0.0	87.3 ± 7.6	25.2 ± 0.1	34.8 ± 1.5	
19	26.3 ± 3.7		27.3 ± 3.5		55.5 ± 0.2	69.8 ± 0.3
20	88.1 ± 0.7	6.6 ± 0.1	75.2 ± 1.1	27.3 ± 0.2	86.0 ± 5.3	8.3 ±0.0
21	89.6 ± 0.6	15.7 ± 0.1	63.4 ± 7.4	26.9 ± 0.2	61.1 ± 3.2	86.6 ± 0.3
22	34.7 ± 16.2		$24,3 \pm 10,4$		$90,6 \pm 3,3$	10.8 ± 0.0
23	81.7 ± 10.8	19.1 ± 0.1	78.3 ± 3.0	20.8 ± 0.1	84.7 ± 0.3	17.0 ± 0.1
24	75.9 ± 2.5	19.3 ± 0.1	66.4 ± 0.0	23.1 ± 0.1	57.4 ± 0.4	76.7 ± 0.3
25	44.5 ± 1.9		43.0 ± 1.0		81.6 ± 0.4	16.8 ± 0.1
26	86.1 ± 7.2	$\textbf{23.3} \pm \textbf{0.1}$	69.7 ± 3.6	44.6 ± 0.2	43.8 ± 1.2	
27	21.0 ± 0.2		9.2 ± 6.6		86.2 ± 0.0	43.1 ± 0.2
28	n.i. ^b		1.1 ± 14.9		92.7 ± 4.0	13.4 ± 1.1
29	n.i.		n.i.		89.1 ± 2.3	13.6 ± 1.1
30	n.i.		n.i.		71.2 ± 1.5	36.9 ± 1.1
31	n.i.		n.i.		39.9 ± 0.5	
32	n.i.		n.i.		43.8 ± 3.0	
33	46.5 ± 11.6		27.2 ± 9.8		71.8 ± 0.4	33.1 ± 1.1
34	n.i.		n.i.		5.7 ± 0.1	
35	15.4 ± 5.8		27.0 ± 2.8		78.3 ± 0.0	38.9 ± 1.1
36	34.8 ± 3.2		58.3 ± 4.2	74.3 ± 1.1	93.3 ± 1.8	13.6 ± 1.1
37	28.7 ± 13.4	1	19.2 ± 5.2		93.2 ± 0.4	4.9 ± 1.1
38	3.5 ± 4.7	/	18.4 ± 16.6		90.6 ± 1.5	23.3 ± 1.1
39	43.8 ± 13.1		14.4 ± 2.1		93.5 ± 0.5	$\textbf{5.8} \pm \textbf{0.1}$
40	n.i.		n.i.		63.7 ± 1.3	51.0 ± 1.1
41	5.9 ± 8.8		10.6 ± 2.6		26.1 ± 5.5	
42	64.7 ± 3.8		57.4 ± 5.8		72.2 ± 8.1	

Table 2

Effect of 2,3-diarylpropenoic acids 1–42 on the inhibition of AKR1C1–AKR1C3.

^{*a*} The values represent % inhibition at 100 μ M inhibitor. ^{*b*} No inhibition.

Highlights

- A series of new 2,3-diarylpropenoic acids were prepared.
- The compounds were evaluated for their inhibition of aldo-keto reductases.
- Selective inhibitors of AKR1C3 were identified.
- Substituents on benzene rings play a crucial role in selective inhibition of AKR1C3.

Appendix A. Supporting information

2,3-Diaryl propenoic acids as selective non-steroidal inhibitors of type 5 17 β -hydroxysteroid dehydrogenase (AKR1C3)

Martin Gazvoda, Nataša Beranič, Samo Turk, Bojan Burja, Marijan Kočevar, Tea Lanišnik Rižner, Stanislav Gobec, Slovenko Polanc

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Applying a general procedure described in the Experimental protocols the following acids were prepared:

(2*E*)-3-(3,4-Dimetoxyphenyl)-2-(naphthalen-1-yl)prop-2-enoic acid (13). Reaction time: 16 h; temperature: 100 °C; yield 36%; mp 190.5–192.5 °C (EtOAc); IR (KBr) 2833, 1673, 1593, 1510, 1418, 1258, 1143, 1020 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.98 (3H, s), 3.59 (3H, s), 6.31 (1H, d, *J* = 1.2 Hz), 6.66 (1H, d, *J* = 8.4 Hz), 6.80 (1H, dd, *J*₁ = 1.2 Hz, *J*₂ = 8.4 Hz), 7.39–7.58 (4H, m), 7.81–7.83 (1H, m), 7.92–7.97 (2H, m), 8.15 (1H, s), 12.67 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 54.3, 55.4, 111.3, 112.4, 125.0, 125.1, 126.2, 126.3, 126.6, 127.0, 127.2, 127.9, 128.5, 128.8, 131.7, 133.6, 135.1, 140.9, 148.1, 150.1, 168.9; MS (ESI–) *m*/*z* 333 ([M – H]⁻, 30); HRMS for C₂₁H₁₇O₄ ([M – H]⁻): calcd 333.1127; found 333.1123. Anal. for C₂₁H₁₈O₄: calcd C, 75.43; H, 5.43; found C, 75.50; H, 5.33.

(2*E*)-3-[4-(Methylsulfanyl)phenyl]-2-phenylprop-2-enoic acid (15). Reaction time: 20 h; temperature: 100 °C; yield 62%; mp 165.0–168.0 °C (EtOAc/*n*-heptane); mp²⁴ 167–169 °C (*i*-PrOH); IR (KBr) 1673, 1590, 1493, 1422, 1286, 1267, 1095, 702 cm⁻¹; ¹H NMR (CDCl₃) δ 2.42 (3H, s), 6.95–7.02 (4H, m), 7.22–7.26 (2H, m), 7.35–7.42 (3H, m), 7.88 (1H, s); ¹³C NMR (CDCl₃) δ 14.9, 125.2, 128.0, 128.8, 129.7, 130.58, 130.61, 131.2, 135.4, 141.3, 141.9, 173.1; MS (ESI–) *m*/*z* 269.1 ([M - H]⁻, 100); Anal. for C₁₆H₁₄O₂S: calcd C, 71.08; H, 5.22; found C, 71.32; H, 5.08.

(2*E*)-2-[4-(Methylsulfonyl)phenyl]-3-phenylprop-2-enoic acid (19). Reaction time: 24 h; temperature: 130 °C; yield 49%; mp 228–230 °C (EtOAc); mp²⁵ 229–230 °C; IR 2971, 2901, 1667, 1611, 1595, 1448, 1422, 1273, 1142, 1089 cm⁻¹; ¹H NMR (DMSO- d_6) δ 3.26 (3H, s), 7.03–7.09 (2H, m), 7.21–7.31 (3H, m), 7.46 and 7.92 (4H, AA'XX', *J* = 8.4 Hz), 7.87 (1H, s), 12.92 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 43.5, 127.1, 128.5, 129.4, 130.2, 130.7, 131.9, 133.9, 139.9, 140.2, 141.9, 167.6; MS (ESI+) *m*/*z* 303 ([M + H]⁺); HRMS for C₁₆H₁₅O₄S ([M + H]⁺): calcd 303.0686; found 303.0684; Anal. for C₁₆H₁₄O₄S: calcd C, 63.56; H, 4.67; found C, 63.28; H,4.71.

(2*E*)-2-(4-Chlorophenyl)-3-[4-(methylsulfanyl)phenyl]prop-2-enoic acid (20). Reaction time: 6 h; temperature: 100 °C; yield 55%; mp 180.4–182.7 °C (EtOAc/*n*-heptane); mp²⁴ 185–187 °C (toluene); IR (KBr) 1674, 1614, 1589, 1490, 1418, 1281, 1262, 1090 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 2.42 (3H, s), 7.00 (2H, d, *J* = 8.5 Hz), 7.10 (2H, d, *J* = 8.5 Hz), 7.20 (2H, d, *J* = 8.5 Hz), 7.45 (2H, d, *J* = 8.5 Hz), 7.74 (1H, s), 12.77 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 14.0, 125.1, 128.6, 130.3,

130.6, 131.0, 131.5, 132.3, 135.4, 139.1, 140.5, 168.0; MS (ESI–) *m*/*z* 303 ([M - H]⁻, 38). Anal. for C₁₆H₁₃ClO₂S: calcd C, 63.05; H, 4.30; found C, 63.16; H, 4.25.

(2*E*)-3-(4-Methoxyphenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (30). Reaction time: 22 h; temperature: 140 °C; yield 21%; mp 234–235 °C (EtOAc); IR 3025, 2938, 1666, 1597, 1566, 1511, 1408, 1312, 1174, 1146 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.27 (3H, s), 3.71 (3H, s), 6.81 (2H, d, J = 9.0 Hz), 7.00 (2H, d, J = 9.0 Hz), 7.47 (2H, d, J = 8.5 Hz), 7.81 (1H, s), 7.95 (2H, d, J = 8.5Hz), 12.78 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 55.2, 114.1, 126.2, 127.2, 129.2, 130.7, 132.1, 139.8, 139.9, 142.4, 160.2, 167.8; MS (ESI+) *m*/*z* 333 ([M + H]⁺); HRMS for C₁₇H₁₇O₅S ([M + H]⁺): calcd 333.0791; found 333.0791. Anal. for C₁₇H₁₆O₅S: calcd C, 61.43; H, 4.85; found C, 61.26; H, 4.91.

(2*E*)-3-(4-Hydroxyphenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (31). Reaction time: 19 h; temperature: 120 °C; yield 54% (column chromatography); mp 213–215 °C (*n*-hexane/EtOAc); mp²⁵ 219–220 °C; IR 3346, 2927, 1661, 1602, 1581, 1512, 1429, 1383, 1310, 1140 cm⁻¹; ¹H NMR (DMSO-*d*₆) δ 3.27 (3H, s), 6.60 (2H, d, *J* = 8.5 Hz), 6.87 (2H, d, *J* = 8.5 Hz), 7.44 (2H, d, *J* = 8.3 Hz), 7.73 (1H, s), 7.93 (2H, d, *J* = 8.3 Hz), 9.95 (1H, broad s) 12.72 (1H, broad s); ¹³C NMR (DMSO-*d*₆) δ 43.5, 115.4, 124.6, 127.2, 130.8, 132.3, 139.7, 140.3, 142.6, 158.9, 168.0 (1 signal missing); MS (ESI+) *m*/*z* 319 ([M + H]⁺); HRMS for C₁₆H₁₅O₅S ([M + H]⁺): calcd 319.0635; found 319.0634. Anal. for C₁₆H₁₄O₅S x 0.1 H₂O: calcd C, 60.03; H, 4.47; found C, 59.93; H, 4.12.

(2*E*)-3-(3-Bromophenyl)-2-[4-(methylsulfonyl)phenyl]prop-2-enoic acid (36). Reaction time: 24 h; temperature: 120 °C; yield 73 %; mp 219–221 °C (EtOAc); IR 2972, 1672, 1596, 1559, 1477, 1424, 1287, 1264, 1143, 1090 cm⁻¹; ¹H NMR (DMSO- d_6) δ 3.25 (3H, s), 7.03 (1H, d), 7.17–7.22 (2H, m), 7.44–7.45 (1H, m), 7.47 (2H, d, *J* = 8.5 Hz), 7.84 (1H, s), 7.94 (2H, d, *J* = 8.5 Hz), 13.1 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 43.6, 121.6, 127.2, 129.0, 130.5, 130.7, 131.8, 132.6, 133.5, 136.4, 138.6, 140.1, 141.4, 167.3; MS (ESI+) *m*/*z* 380 ([M+H]⁺); HRMS for C₁₆H₁₄BrO₄S ([M + H]⁺): calcd 380.9791; found 380.9787. Anal. for C₁₆H₁₃BrO₄S: calcd C, 50.41; H, 3.44; found C, 50.22; H, 3.20.

Applying the procedure described in the Experimental protocols for the oxidation of the methylthio moiety into methylsulfonyl group the following acid was prepared:

(2*E*)-2,3-Bis[4-(methylsulfonyl)phenyl]prop-2-enoic acid (40). Yield 85 %; mp 266–268 °C (EtOAc); IR 2988, 2901, 1683, 1622, 1593, 1409, 1311, 1275, 1149, 1067 cm⁻¹; ¹H NMR (DMSO-

 d_6) δ 3.19 (3H, s), 3.27 (3H, s), 7.30 (2H, d, J = 8.5 Hz), 7.48 (2H, d), 7.78 (2H, d, J = 8.5 Hz), 7.93–7.96 (3H, m), 13.20 (1H, broad s); ¹³C NMR (DMSO- d_6) δ 43.1, 43.5, 126.9, 127.2, 130.66, 130.73, 134.7, 138.3, 139.0, 140.2, 140.7, 140.9, 167.3; MS (ESI+) m/z 381 ([M + H]⁺); Anal. for C₁₇H₁₆O₆S₂: calcd C, 53.67; H, 4.24; found C, 53.96; H, 4.52.





Fig. 1. Predicted binding pose of the compound **1** (blue) in the AKR1C1 active site (green). Only relevant amino acid residues are shown and labeled. Co-crystallized 3,5-dichlorosalicylic acid is shown as magenta sticks and cofactor as orange sticks.














Purity of acids 1-42 determined by HPLC

COOH

A purity of each tested compound was checked by HPLC. The HPLC measurements were performed with an Agilent Technologies 1260 Infinity High Performance Autosampler and the data were processed by the ChemStation for LC 3D systems software. The absorbances at 250 nm, 260 nm and 210 nm were used for the UV detection. A SUPELCOSIL LC-18-DB HPLC column (5 μ m particle size, 15 cm x 4.6 mm) was applied with an eluent CH₃CN:H₂O = 60:40, with a flow rate of 1.0 mL/min or CH₃CN:H₂O = 80:20 with a flow rate of 1.5 mL/min.



Injection volume: 20 μ L; Eluent: CH₃CN:H₂O = 60:40; Flow rate: 1.0 mL/min; Run time: 10 minutes;



Peak R #	etTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
-		-				
1	1.486	BB	0.1252	290.95987	31.66351	1.3805
2	2.629	BB	0.1281	2.07860e4	2 <mark>473.24</mark> 243	98. <mark>6</mark> 195
Totals	:			2.10769e4	2504.90594	





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	1.323	BV	0.1241	62.92740	6.91976	0.7444
2	2.012	VB	0.1014	8390.16211	1233.55859	99.2556

Totals :

8453.08951 1240.47835





```
Signal 1: DAD1 A, Sig=250,4 Ref=off
```

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	1.097	BV	0.1130	10.49679	1.40698	0.0983
2	1.273	vv	0.0637	10.86171	2.54747	0.1017
3	1.379	vv	0.0807	36.87602	6.647 <mark>0</mark> 5	0.3454
4	2.001	VB	0.2887	264.13330	11.46 <mark>9</mark> 31	2.4743
5	4.172	BB	0.1550	1.03529e4	966.44574	96.9802

Totals :

1.06752e4 988.51655







Peak	RetTime	туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	26
1	0.764	BB	0.0635	8.79783	2.07160	0.0443
2	1.182	BV	0.0655	18.47845	4.18784	0.0930
3	1.494	vv	0.0962	124.85480	18.63762	0.6282
4	1.864	vv	0.1039	1.89198e4	2764.31226	95.1906
5	2.127	vv	0.1407	377.15109	35.86217	1.8975
6	2.516	vv	0.1625	80.95168	7.0 <mark>21</mark> 88	0.4073
7	2.809	vv	0.1391	139.42810	14.64727	0.7015
8	2.962	vv	0.1200	66.65894	8.10102	0.3354
9	3.156	VB	0.1726	72.91757	5.88503	0.3669
10	3.554	BB	0.1965	2 <mark>4.694</mark> 91	1.73070	0.1242
11	5.779	BB	0.2037	41.97354	3.18591	0.2112

Signal 1: DAD1 A, Sig=250,4 Ref=off

Totals :

1.98757e4 2865.64330







Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
				[
1	1.273	vv	0.1248	69.99376	7.64469	0.5653
2	1.468	vv	0.1116	13.65792	1.70167	0.1103
3	1.996	VB	0.1157	1.22561e4	1559.98694	98.9928
4	3.679	BB	0.2122	41.04433	2.88096	0.3315

Totals :

1.23808e4 1572.21425









Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	1.188	BV	0.0890	31.48886	5.48057	0.5178
2	1.276	VV	0.0519	7.28952	2.13008	0.1199
3	1.429	VV	0.0577	27.80760	7.09872	0.4573
4	1.495	vv	0.0790	43.63830	8.07783	0.7176
5	1.734	VV	0.0878	5937.51514	992.97858	97.6328
6	2.312	VB	0.1412	33.73643	3.35698	0.5547

Totals :

6081.47584 1019.12276







Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	1.203	BV	0.0807	101.15851	17.70278	0.4826
2	1.567	vv	0.1046	2.06139e4	2910.02417	98.3388
3	2.038	vv	0.1413	191.24721	19.02917	0.9123
4	2.300	VB	0.1791	55.81504	4.25445	0.2663

Totals :

2.09622e4 2951.01057





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
		·				
1	1.166	BV	0.0685	8.79301	1.81711	0.0364
2	1.389	vv	0.0733	19.89313	3.91278	0.0825
3	1.489	VV	0.0977	38.32263	5.75447	0.1589
4	1.646	VB	0.1241	26.56158	3.03633	0.1101
5	2.662	BB	0.1335	2.40306e4	2707.99438	99.6121

Totals :

2.41242e4 2722.51507







Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	0.838	BB	0.0740	9.37613	2.02443	0.0775
2	1.210	VV	0.0609	48.47398	12.06451	0.4007
3	1.451	VV	0.1365	60.00398	7.10669	0.4961
4	1.693	VB	0.1006	31.69940	4.47796	0.2621
5	2.877	BB	0.1375	1.19465e4	1249.67114	98.7636



1.20961e4 1275.34474







Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	1,211	BV	0.0621	24.33888	5.90599	0.1285
2	1.273	vv	0.0745	28.26418	5.45154	0.1492
3	1.795	vv	0.1125	1.88134e4	2479. <mark>70874</mark>	99.2959
4	2.263	VB	0.1478	80.79733	7.37801	0.4264

Totals :

1.89468e4 2498.44428





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime	Type	Width	Area	Height	Area %
	[]				[
1	1.159	BV	0.1158	11.65776	1.33606	0.0482
2	1.375	vv	0.0824	69.46424	12.21263	0.2870
3	1.446	vv	0.0655	37.49666	8.48671	0.1549
4	2.014	vv	0.2679	305.57709	15.37611	1.2626
5	2.555	VB	0.1607	31.46740	2.80905	0.1300
6	3.098	BV	0.1912	15.85656	1.27403	0.0655
7	3.991	VB	0.1664	2.37298e4	2123.12549	98.0517

Totals :

2.42013e4 2164.62008





Injection volume: 20 μ L; Eluent: CH₃CN:H₂O = 60:40; Flow rate: 1.0 mL/min; Run time: 10 minutes; DAD1 A, Sig=250,4 Ref=off (MARTING\ZS000069.D)



Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	1.648	VB	0.1231	12.00586	1.35792	0.0668
2	2.544	BV	0.1147	1.78365e4	2345.78271	99.2516
3	3.149	VB	0.2146	122.48573	7.65635	0.6816

Totals :

1.79710e4 2354.79699











Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Туре	Width	Area	Height	Area
- 花	[min]		[min]	[mAU^s]	[mA0]	-8
1	1.096	BV	0.1389	17.90719	1.88448	0.0637
2	1.289	vv	0.0798	80.60739	15.22711	0.2868
3	1.834	vv	0.3134	610.63336	24.11922	2.1729
4	2.396	vv	0.1208	15.37680	1.97566	0.0547
5	3.234	VB	0.1670	2.73781e4	2515.29565	97.4219

Totals :

2.81026e4 2558.50213











S28





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	0.492	BB	0.1014	11.20597	1.73281	0.1124
2	0.951	VV	0.0682	9822.73438	2195.45728	98.5357
3	1.335	VB	0.1074	134.76776	17.20043	1.3519

Totals :

9968.70810 2214.39052



Acid 19: COOH MeO₂S



Signal	1:	DADI	A,	Sig=250,4	Rei=oii	

Peak	RetTime	туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	do
1	0.743	BV	0.1026	130.89302	17.23842	0.3577
2	1.135	vv	0.1300	482.35974	50.24570	1.3181
3	1.754	VB	0.196 <mark>8</mark>	3.59817e4	2711. <mark>53</mark> 174	98.32 <mark>4</mark> 2

Totals :

3.65950e4 2779.01586





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	olo
1	0.844	BV	0.0564	21.50703	5.92053	0.0606
2	0.951	vv	0.0935	22.70934	3.51072	0.0640
3	1.138	vv	0.1094	44.62593	5.57644	0.1257
4	1.442	vv	0.1079	513.52 <mark>4</mark> 05	68.13187	1.4466
5	2.104	vv	0.3406	753.22632	27.24002	2.1219
6	2.451	vv	0.1397	30.56843	3.08369	0.0861
7	2.743	vv	0.1525	23.05767	2.19626	0.0650
8	3.021	VB	0.1946	26.29200	1.95895	0.0741
9	4.211	BB	0.2045	3.40627e4	2506.89941	95.9561

Totals :

3.54982e4 2624.51788





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime [min]	туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	0.788	BB	0.0982	29.55809	4.19636	0.1995
2	1.191	vv	0.0992	7.68496	1.02825	0.0519
3	1.826	VB	0.0946	1.47256e4	2244.23730	99.3819
4	2.530	BB	0.2093	54.34188	3,49621	0.3667

Totals :

1.48172e4 2252.95812





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	90
1	0.763	BV	0.0677	8.20078	1.71758	0.0273
2	1.194	vv	0.1363	379.45554	38.06166	1.2621
3	2.078	VB	0.1407	2.96785e4	3072. <mark>4</mark> 5068	98.7107

Totals :

3.00661e4 3112.22992











Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	0.730	BB	0.0951	22.62497	3.25497	0.0986
2	1.804	VB	0.1236	2.29083e4	2738.59839	99.8669
3	2.723	BB	0.1185	7.89540	1.11638	0.0344

Totals :

2.29388e4 2742.96974







8780.69542 1998.06011









Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	0.503	BV	0.0801	12.31536	2.47426	0.1591
2	0.578	vv	0.0688	17.99948	3.83443	0.2326
3	0.682	vv	0.0679	9.05 <mark>15</mark> 5	1.95813	0.1170
4	0.862	vv	0.0782	177.02937	34.30334	2.2876
5	1.017	VB	0.0625	7522.28564	1885.69958	97.2037

Totals :

7738.68141 1928.26974







Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Are <mark>a</mark> %
1	0.820	BB	0.0555	11.32088	3.03392	0.0473
2	1.168	BV	0.0949	159.70201	23.01826	0.6667
3	1.365	vv	0.0905	13.14723	2.24031	0.0549
4	<mark>1.</mark> 978	VB	0.1460	2.37707e4	2 <mark>4</mark> 32.41333	99.2312

Totals :

2.39548e4 2460.70582







Peak F #	etTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
-		-				
1	0.728	BV	0.1191	22.86541	2.53797	0.1674
2	0.913	VV	0.0796	10.22548	1.81841	0.0749
3	1.221	VV	0.1363	224.03180	22.87101	1.6400
4	1.966	VB	0.1056	1.34030e4	1826.13403	98.1177

Totals :

1.36601e4 1853.36142







Peak <mark>#</mark>	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
						
1	0.769	BV	0.0999	245.14107	33.27809	1.1686
2	1 <mark>.1</mark> 53	vv	0.1081	335. <mark>4</mark> 9460	42.51787	1.5993
3	1.267	VV	0.0865	64.89405	11. <mark>3</mark> 9256	0.3094
4	1.764	vv	0.1224	2.00465e4	2425.52710	95.5619
5	3.007	VB	0.3438	285.47015	10.68165	1.3608

Signal 1: DAD1 A, Sig=250,4 Ref=off

Totals :

2.09775e4 2523.39727







Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	olo
1	0.453	BB	0.0974	23.05219	3.14875	0.4341
2	0.696	BV	0.0753	7.46322	1.37522	0.1405
3	0.858	vv	0.0706	103.21199	20.56252	1.94 <mark>36</mark>
4	1.009	VV	0.0655	5124.88330	1207.27100	96.5061
5	1.332	VB	0.1316	51.81536	5.23149	0.9757

Signal 1: DAD1 A, Sig=250,4 Ref=off

Totals :

5310.42607 1237.58898







Signal 1: DAD1 A, Sig=250,4 Ref=off

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Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
	<mark>-</mark>					
1	0.909	BV	0.0425	47.67751	18.24652	0.1852
2	1.129	vv	0.1527	1065.52319	96.59529	4.1390
3	1.458	VB	0.1400	2.45963e4	2561.23975	95.5427
4	3.013	BB	0.1201	34.26773	4.64091	0.1331

Totals :

2.57438e4 2680.72246




Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	0.726	BV	0.0891	112.50421	16.59386	0.3659
2	1.378	VB	0.1707	3.06351e4	2577.07178	99.6341

Totals :

3.07476e4 2593.66563

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Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	0.880	BV	0.1534	6 <mark>1.</mark> 99350	6.16054	0.3664
2	1.605	VB	0.1084	1.68393e4	2172.75659	99.5222
3	2.411	BB	0.1222	18.85623	2.28572	0.1114

Totals :

1.69202e4 2181.20286





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	*
						I
1	0.876	BV	0.1213	67.80187	8.12850	0.4898
2	1.111	vv	0.0877	33.50153	5.30237	0.2420
3	1.758	VB	0.1048	1.37202e4	1803.41064	99.1211
4	2.547	BB	0.1300	20.34628	2.28352	0.1470

Totals :

1.38418e4 1819.12505







Peak	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	0.828	BV	0.1038	19.11580	2.60016	0.4266
2	1.113	VB	0.0880	7.81407	1.23271	0.1744
3	1.919	BB	0.1654	4454.55469	367.63260	99.3991

Totals :

4481.48456 371.46547







Totals :

7445.85667 1754.74905





Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
The states of	l a contral i					
1	0.749	BV	0.1062	148.18922	19.17753	0.5024
2	0.914	VV	0.1411	99.58707	10.27574	0.3376
3	1.199	vv	0.1343	510.98929	57.17010	1.7322
4	2.161	VB	0.1617	2.874 <mark>01</mark> e4	2508.14990	97.4278

Totals :

2.94988e4 2594.77327







Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	1.004	BV	0.0655	5878.83154	1384.33582	98.6218
2	1.344	VB	0.1229	65.90869	7.19091	1.1057
3	1.706	BV	0.0868	6.51858	1.07353	0.1094
4	1.847	VB	0.0809	9.72588	1.80471	0.1632

Totals :

5960.98469 1394.40497







Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak	RetTime	Type	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	8
1	1.012	BV	0.0629	4685.00781	1117.69604	98.1790
2	1.337	VB	0.0877	73.22070	11.91837	1.5344
3	1.799	BB	0.0761	13.67798	2.84629	0.2866

Totals :

4771.90649 1132.46071







Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak #	RetTime [min]	Туре	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	0.720	VV	0.0882	33.64045	5.53393	0.7870
2	0.865	vv	0.0719	4203.66064	847.74023	98.3485
3	1.196	VB	0.1124	36.94837	4.47115	0.8644

4274.24947 857.74531







Signal 1: DAD1 A, Sig=250,4 Ref=off

Peak 1	RetTime	Туре	Width	Area	Height	Area
#	[min]		[min]	[mAU*s]	[mAU]	용
-		-				1
1	0.762	BB	0.0935	11.9 <mark>11</mark> 82	1.70601	0.1064
2	1.215	BV	0.1470	122.50547	11.8 <mark>0</mark> 780	1.0946
3	1.406	VV	0.0805	40.70861	7.36750	0.3638
4	1.552	VB	0.1230	68.96523	7.97008	0.6162
5	1.941	BV	0.1524	16.95156	1.49396	0.1515
6	3.215	VB	0.2022	1.08720e4	728.29901	97.1471
7	5.532	BB	0.2419	58.23187	3.65438	0.5203

Totals :

1.11913e4 762.29874