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Synthesis and activity of *N*-oxalylglycine and its derivatives as Jumonji C-domain-containing histone lysine demethylase inhibitors

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ABSTRACT

N-Oxalylglycine (NOG) derivatives were synthesized, and their inhibitory effect on histone lysine demethylase activity was evaluated. NOG and compound **1** inhibited histone lysine demethylases JMJD2A, 2C and 2D in enzyme assays, and their dimethyl ester prodrugs DMOG and **21** exerted histone lysine methylating activity in cellular assays.

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Methylation of lysine residues in the N-terminal region of the core histones plays a pivotal role in the regulation of gene expression.¹ It has been reported that methylation of histone lysine residues occurs on lysine 26 of histone 1 (H1K26), H3K4, H3K9, H3K27, H3K36, H3K79 and H4K20.² Methylated lysine can be in a mono-, di-, or tri-methylated form, and these different forms differentially affect chromatin structure and are responsible for transcriptional activation as well as silencing.³

In contrast to other histone modifications such as acetylation and phosphorylation, histone methylation had been regarded as irreversible because of the high thermodynamic stability of the N–C bond. However, two classes of histone lysine demethylases have recently been identified. One is lysine-specific demethylase 1, a flavin-dependent amine oxidase domain-containing enzyme.⁴ The other comprises the Jumonji C domain-containing histone demethylases (JHDMs) such as JMJD2A, 2B, 2C and 2D.⁵ JHDMs are Fe(II) and α -ketoglutarate dependent enzymes that oxygenate methylated histone lysine residues and thereby cause their demethylation after releasing formaldehyde, carbon dioxide and succinic acid. The identification of these enzymes established that histone methylation is reversibly regulated by histone lysine methyltransferases and histone lysine demethylases.⁶

JHDMs are implicated in cancer cell growth.² For example, it was shown that JMJD2C is associated with the growth of oesophagal squamous cancer, and JMJD2A, 2B and 2C are involved in pros-

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tate cancer.^{5,7} Therefore, JHDM inhibitors can be not only tools to study the function of these enzymes, but also serve as potential anticancer agents.

To date, only a few JHDM inhibitors have been identified (Fig. 1). *N*-Oxalylglycine (NOG), the amide analogue of α -ketoglutarate, has been reported to inhibit JMJD2C in an in vitro assay,⁷ but has not been tested in a cellular assay. Succinic acid has been suggested to inhibit JMJD2D by product inhibition.⁸ In addition, during the course of our study presented in this report, 2,4-lutidinic acid, which is known as an inhibitor of other Fe(II)/ α -ketoglutarate dependent oxygenases, was reported as an inhibitor of JMJD2A and 2E,⁹ although it was also not examined in a cellular assay. To the best of our knowledge, there is no report on selective JHDM inhibitors. Therefore, we initiated a search for novel JHDM inhibitors with the goal of drug discovery as well as finding new tools for biological research. In this letter, we describe the design, synthesis, JHDM inhibition activity and cellular activity of NOG and its derivatives.

We designed JHDM inhibitors based on the crystal structure of JMJD2A complexed with NOG and histone trimethylated lysine



Figure 1. Structures of previously reported JHDM inhibitors.

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peptide (PDB ID 2OQ6).¹⁰ The crystal structure showed that the oxalyl group of NOG interacts with Fe(II), and the other carboxyl group forms a hydrogen bond with Tyr 132 in the active centre of JMJD2A (Fig. 2, left). In addition, the trimethylamino group of histone trimethylated lysine peptide is surrounded by Gly 170, Tyr 175, Glu 190 and Ser 288. On the basis of this structure, we designed potential selective JHDM inhibitors **1–4** (Fig. 3) in which NOG is connected with a dimethylamino group through a linker (Fig. 2, right). We anticipated that the nitrogen atom of the dimethylamino group forms a hydrogen bond with the hydroxyl group of Tyr 175 or Ser 288,^{11,12} and the aryl linker interacts with the aromatic rings of Tyr 177, Phe 185 and Trp 208, which might lead to the selective inhibition of IHDMs.

The synthesis of compound **1** is outlined in Scheme 1. The primary amine of 4-dimethylaminobenzylamine **5** was alkylated with *tert*-butyl bromoacetate to give secondary amine **6**. Treatment of compound **6** with *tert*-butyl chloroglyoxylate afforded tertiary



Figure 2. Interaction of NOG and trimethylated lysine substrate with JMJD2A (left), and models for the binding of designed inhibitors (right).



Figure 3. Structures of compounds 1-4.



Scheme 1. Reagents and conditions: (a) *tert*-butyl bromoacetate, Et₃N, CH₂Cl₂, rt, 54%; (b) *tert*-butyl chloroglyoxylate, Et₃N, CH₂Cl₂, 0 °C, 58%; (c) HCl, AcOEt, CH₂Cl₂, 0 °C to rt, 59%.



Scheme 2. Reagents and conditions: (a) $(Boc)_2O$, Et₃N, THF, 0 °C to rt, 82%; (b) H₂, Pd/C, EtOH, rt, 90%; (c) MeI, K₂CO₃, DMF, rt, 34%; (d) HCI, AcOEt, CH₂Cl₂, 0 °C to rt, 94%; (e) *tert*-butyl bromoacetate, Et₃N, CH₂Cl₂, rt; (f) *tert*-butyl chloroglyoxylate, Et₃N, CH₂Cl₂, rt; (g) HCI, AcOEt, CHCl₃, 0 °C to rt, 26% (three steps).

amine **7**. Removal of the two *tert*-butyl groups of **7** under acidic conditions gave the desired dicarboxylic acid **1**.

Scheme 2 shows the preparation of compound **2**. Reaction of 4nitrophenethylamine **8** with (Boc)₂O gave Boc-protected compound **9**. Reduction of the nitro group of **9** gave aniline **10**. Aniline **10** was reacted with iodomethane to give dimethylamine **11**. Then, deprotection of **11** using hydrochloric acid yielded amine **12**. Compound **2** was prepared from amine **12** using the procedure described for the synthesis of **1**.

Preparation of compound **3** is shown in Scheme 3. Horner-Wadsworth–Emmons reaction was applied to the conversion of benzaldehyde **13** into cinnamonitrile **14**. The double bond of **14** was hydrogenated to give compound **15**. The cyano group of **15** was then reduced, and subsequent treatment with $(Boc)_2O$ gave compound **16**. *N*-Boc compound **16** was converted to compound **3** using the procedure described for the synthesis of **1** and **2**.

Scheme 4 illustrates the synthesis of compound **4**. Coupling between carboxylic acid **17** and NH₃ in the presence of EDCI and



Scheme 3. Reagents and conditions: (a) cyanomethylphosphoic acid diethyl ester, NaH, THF, 0 °C, 66%; (b) H₂, Pd/C, MeOH, 83%; (c) (i) BH₃?SMe₂, THF, reflux; (ii) (Boc)₂O, Et₃N, THF, rt, 58%; (d) HCl, AcOEt, CH₂Cl₂, 0 °C to rt, 80%; (e) *tert*-butyl bromoacetate, Et₃N, CH₂Cl₂, rt; (f) *tert*-butyl chloroglyoxylate, Et₃N, CH₂Cl₂, 0 °C, rt; (g) HCl, AcOEt, CH₂Cl₂, 0 °C to rt, 12% (three steps).



Scheme 4. Reagents and conditions: (a) NH₃, EDCI, HOBt·H₂O, DMF, H₂O, rt, 54%; (b) (i) LiAlH₄, THF, reflux; (ii) (Boc)₂O, Et₃N, THF, rt, 87%; (c) HCl, AcOEt, 0 °C to rt; (d) *tert*-butyl bromoacetate, Et₃N, CH₂Cl₂, rt; (e) *tert*-butyl chloroglyoxylate, Et₃N, CH₂Cl₂, 0 °C, rt; (f) HCl, AcOEt, CH₂Cl₂, 0 °C to rt, 23% (four steps).

HOBt afforded amide **18**. The reduction of the amide group of **18**, followed by the treatment with $(Boc)_2O$ gave compound **19**. Compound **4** was prepared from compound **19** using the procedure described for the synthesis of **1** and **2**.

For the evaluation of the enzyme inhibitory activity of NOG and its derivatives 1-4, we initially generated GST fusion proteins of the catalytic N-terminus of JMJD2A and 2C as well as of full-length JMJD2D.¹³⁻¹⁶ Because JMJD2A, 2C and 2D have been reported to demethylate H3K9me₃,¹⁴ we confirmed the histone demethylating activity of these three proteins in an in vitro assay and revealed demethylation with an antibody recognizing trimethylated H3K9 (H3K9me₃).^{17–20} As shown in Figure 4, all three GST-JMJD2 fusion proteins completely removed H3K9me₃, whereas no demethylating activity was observed with mutated GST fusion proteins¹³ in which homologous histidine residues in the catalytic centre were mutated to alanine. Using this system, we next assessed the impact of NOG and putative IMID2 inhibitors **1–4** on the in vitro demethylation activity of GST-IMID2 fusion proteins.¹⁷ We first utilized inhibitors at 1 mM concentration and observed that only NOG significantly curtailed the demethylation activity of JMJD2A and [MJD2C (Fig. 5, lane 4). However, at a higher inhibitor concentration of 3 mM, significant inhibitory effects were also noted with compounds 1, 2, 4 and weakest with 3. In contrast, JMJD2D enzymatic activity was not only strongly inhibited by 1 mM NOG, but also by 1 mM compounds 1, 2 and 3. Furthermore, compound 4 also displayed an inhibitory effect at 3 mM concentration. Thus, all compounds tested were able to inhibit the demethylation activity of JMJD2 proteins.

Next, we assessed if the inhibitors would display activity in cells.^{14,21–25} We focused on the strongest inhibitors, NOG and compound **1**, and prepared dimethylester prodrugs of these compounds to allow for uptake through the cell membrane. Scheme 5 shows the synthesis of compound **21**, the dimethylester prodrug of compound **1**. Compound **5** was allowed to react with methyl bromoacetate to give compound **20**. Then, compound **20** was reacted with methyl chloroglyoxylate to give compound **21**.

In this cellular assay, the accumulation of H3K9me₃ and H3K36me₃ was examined, because JMJD2C has been reported to demethylate both H3K9 and H3K36.¹⁴ As shown in Figure 6, overexpression of JMJD2C resulted in robust demethylation of H3K9me₃ and H3K36me₃ (compare lanes 1 and 5). Further, neither 100 mM dimethyl succinate (DMS), a previously reported JMJD2D



Figure 4. Demethylation ability of GST-JMJD2 fusion proteins (top) and Coomassie stained gel (bottom) indicating that comparable amounts of GST-JMJD2 fusion proteins were utilized.



Figure 5. In vitro JMJD2 inhibition assay of NOG and compounds 1–4. H4K20me₁ was assessed as a control and revealed to be unchanged.



Scheme 5. Reagents and conditions: (a) bromoacetic acid methyl ester, Et₃N, CH₂Cl₂, rt, 58%; (b) chloroglyoxylic acid methyl ester, Et₃N, CH₂Cl₂, 0 °C, 35%.



Figure 6. Demethylation in cells. 293T cells transfected with or without Flag-JMJD2C were treated with 100 mM dimethyl succinate (DMS), 2.5 mM oxalylglycine dimethylester (DMOG), 2.5 mM compound **21** or DMS0 as a control. Demethylation activity of JMJD2C was assessed by anti-H3K9me₃ and anti-H3K36me₃ blotting. As loading controls, total H3 and actin levels were assessed, and expression of JMJD2C revealed by anti-Flag blotting.

inhibitor,⁸ nor 2.5 mM compound **21** affected H3K9me₃ and H3K36me₃ in the presence of JMJD2C. However, 2.5 mM dimethylester of NOG (DMOG) resulted in enhanced H3K9me₃ and H3K36me₃ levels both in the presence of JMJD2C and in the absence of overexpressed JMJD2C, indicating that DMOG represses the demethylation activity of JHDMs including JMJD2C in cells. On the other hand, compound **21** slightly enhanced H3K9me₃ lev-

els in the absence of overexpressed [M]D2C as compared with DMSO and DMS, suggesting that 2.5 mM compound 21 selectively inhibited the demethylase activity of a JHDM(s) other than JMJD2C in cells.

In summary, we have designed and synthesized NOG derivatives 1-4²⁶ and evaluated the JMJD2-inhibitory activity of NOG and compounds 1-4 both in enzyme assays and in cellular assays that have been established for this study. Our data suggest that NOG and compound 1 show inhibitory activity against JHDMs and are more potent than succinic acid. As far as we could determine, this is the first report demonstrating that NOG and its derivative inhibit JHDMs both in enzyme assays and in cellular assays. The findings presented here should be valuable for further explorative studies uncovering more potent and selective JHDM inhibitors. Further investigations pertaining to NOG derivatives are progressing and will be reported in due course.

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freezing in liquid nitrogen, GST fusion proteins were dialyzed against 10 mM Hepes pH 7.4, 50 mM NaCl, 10% glycerol, 0.2 mM phenylmethylsulfonyl fluoride and 1 mM DTT (Ref. 16).

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- The purity of compounds 1-4 and 21 was certified by elemental analysis. 1 HCl: Anal. Calcd for C13H16N2O5 HCl-1/4H2O: C, 48.60; H, 5.49; N, 8.72. Found: C, 48.89; H, 5.80; N, 8.43. 2 HCl: Anal. Calcd for C14H18N2O5 HCl: C, 50.84; H, 5.79; N, 8.47. Found: C, 50.56; H, 5.67; N, 8.48. 3 HCl: Anal. Calcd for C15H21N2O5 HCl 1/10H20: C, 51.91; H, 6.17; N, 8.07. Found: C, 52.25; H, 6.14; N, 8.12. 4 HCI: Anal. Calcd for C13H16N2O5 HCI: C. 49.30; H. 5.41; N. 8.84. Found: C. 49.13; H, 5.59; N, 8.64. 21: Anal. Calcd for C15H20N2O5: C, 58.43; H, 6.54; N, 9.09. Found: C, 58.35; H, 6.52; N, 9.05.