



Synthesis and adenosine receptors binding studies of new fluorinated analogues of pyrido[2,3-*d*]pyrimidines and quinazolines

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Received: 31 July 2017 / Accepted: 12 October 2017
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Abstract A series of new fluorine containing pyrido[2,3-*d*]pyrimidines and imidazo[1,2-*c*]pyrido[3,2-*e*]pyrimidines along with a series of bioisosteric fluorinated quinazolines were synthesised following appropriate synthetic schemes and characterised by spectral analytical means. X-ray crystal structure of the key precursor **1** (2-amino-3-cyano-4-trifluoro-methyl-6-phenyl-pyridine) was also determined to gain insight into its reactivity. Binding affinity data of all the compounds for adenosine receptors (ARs) showed that pyrido[2,3-*d*]pyrimidine scaffold with free amino (NH₂) group at 2- and 4-position (**2a**) exhibited the maximum binding affinity for hA₃ AR with similar affinity for the hA₁

and somewhat lower affinity for hA_{2A} ARs resulting in a compound with no A₃ selectivity vs. A₁ and moderate selectivity vs. A_{2A} AR (*K_i* hA₁ = 0.62 μM, hA_{2A} = 3.59 μM and hA₃ = 0.42 μM). Interestingly, the replacement of both the amino groups with carbonyl (C=O) groups (compound **4**) resulted in significantly improved affinity for hA₁ AR but with moderate selectivity against hA_{2A} and hA₃ ARs (*K_i* hA₁ = 0.17 μM, hA_{2A} = 0.67 μM and hA₃ = 0.68 μM). In case of fluorinated quinazolines, only compound **18a** showed remarkable affinity for hA₁ AR with significant selectivity against hA_{2A} and hA₃ ARs (*K_i* hA₁ = 0.73 μM, hA_{2A} > 30 μM and hA₃ = 9.27 μM). The preliminary results of these compounds demonstrate that the fluorinated pyrido [2,3-*d*]pyrimidine and imidazo[1,2-*c*]pyrido[3,2-*e*]pyrimidine can be considered as promising scaffolds for further optimisation in search of potential antagonists with better affinity and selectivity towards hA₁ and hA₃ ARs.

Electronic supplementary material The online version of this article (<https://doi.org/10.1007/s00044-017-2099-z>) contains supplementary material, which is available to authorized users.

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Keywords Fluorinated pyrido[2,3-*d*]pyrimidines · Fluorinated quinazolines · Adenosine receptors binding · X-ray crystallography

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Introduction

Adenosine receptors (ARs) belong to the super family of G-protein-coupled receptors (GPCRs) and are classified into four subtypes A₁, A_{2A}, A_{2B}, and A₃ ARs, respectively, all of which exhibit distinct physiological functions (Fredholm et al. 2011). A₁ ARs are mainly present in the brain with minimum levels in the heart, lungs, kidney, adipose tissue, stomach, spleen, and liver (Ribeiro et al. 2002). A_{2A} ARs are highly distributed in blood platelets, striatum, nucleus

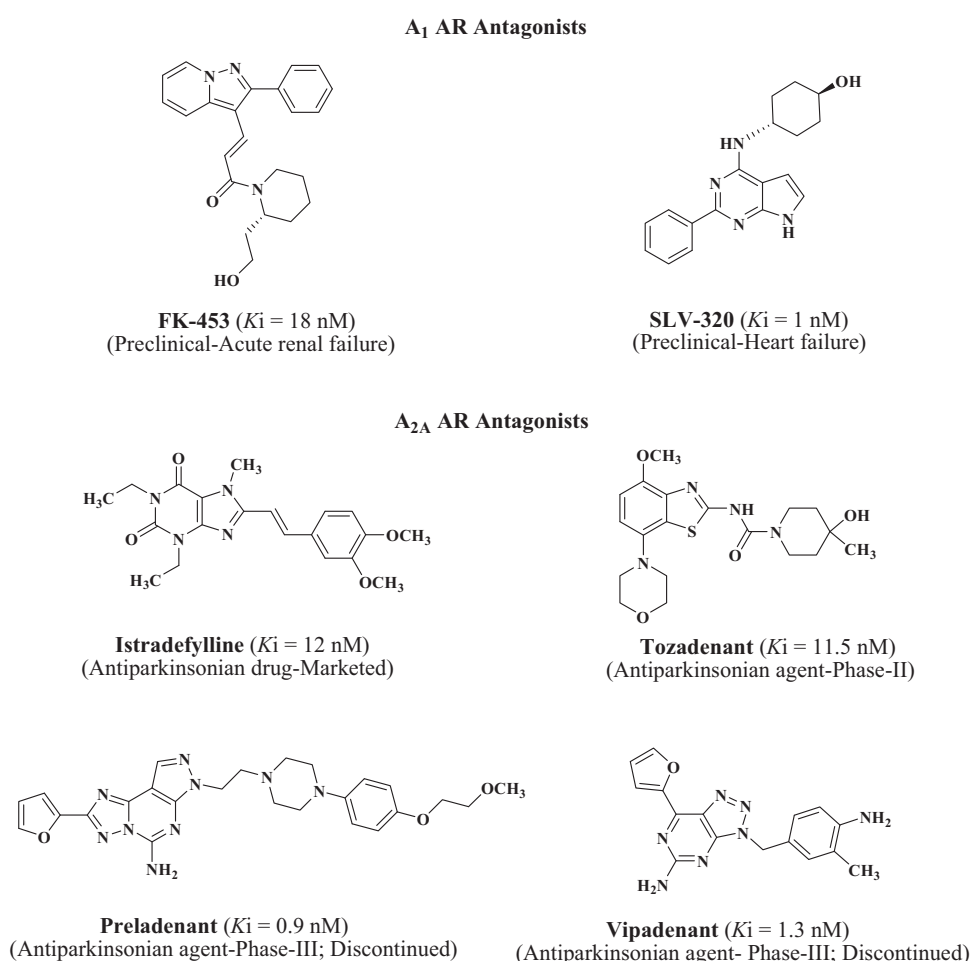
accumbens, and olfactory tubercle (Preti et al. 2015), while A_{2B} ARs regulate a varied number of physiological and pathological events that involve lungs, blood vessels, and bladder (Kalla and Zablocki 2009). A_3 ARs are highly expressed in immune cells, lung and liver and at lower densities in heart, aorta and brain (Borea et al. 2009).

Agonists selective for A_1 AR showed antinociceptive activity and are thought to be useful in the treatment of stroke, epilepsy, migraine, pain, cardiac ischemia, and arrhythmias (Cosimelli et al. 2016) while antagonists might be useful in conditions such as cognitive dysfunction associated with ageing, stroke-induced brain damage or neurodegenerative disorders like Alzheimer's disease (AD) (Ribeiro et al. 2002; Baraldi et al. 2008). A_{2A} AR antagonists are being considered as potential therapeutic alternative for the treatment of Parkinson's disease (PD) (Preti et al. 2015). Recent studies reported the use of highly potent and selective antagonists such as ZM241385, SCH58261 and some xanthine derivatives (istradefylline) as pharmacological tools for this receptor subtype (Bahreyni et al. 2017; Dong et al. 2017; Dungo and Deeks 2013; Kazemi et al. 2017; Shook et al. 2011; van Rhee et al. 1996; Wardas

et al. 2001). Recently, istradefylline has been approved for marketing in Japan as an anti-PD agent (Venkatesan et al. 2014). Selective A_{2B} antagonists were shown to decrease pain (inflammatory), and are promising candidates for the treatment of asthma (Brown et al. 2008) and diabetes (Kalla and Zablocki 2009; Baraldi et al. 2008). Agonists of the A_3 AR are thought to be useful for the treatment of stroke, lung injury (asthma and COPD), cardiac ischemia, rheumatoid arthritis, and cancer (Borea et al. 2009; Jacobson et al. 2009). The blockade of A_3 ARs could be beneficial for the treatment of glaucoma, stroke, asthma, and renal failure (Borea et al. 2009; Jacobson et al. 2009). During the past years a number of potent and selective AR antagonists have been developed, including xanthines and non-xanthine derivatives (Baraldi et al. 2008). Recently reported AR antagonists under different stages of clinical and pre-clinical trial are presented in Figs. 1 and 2 (Muller and Jacobson 2011).

The derivatives of pyrido[2,3-*d*]pyrimidine have shown remarkable biological and pharmacological activities, in particular for the treatment of various inflammatory conditions such as allergy, arthritis, and asthma (Suhagia et al.

Fig. 1 Recently reported potent hA_1 and hA_{2A} ARs antagonists under different stages of development

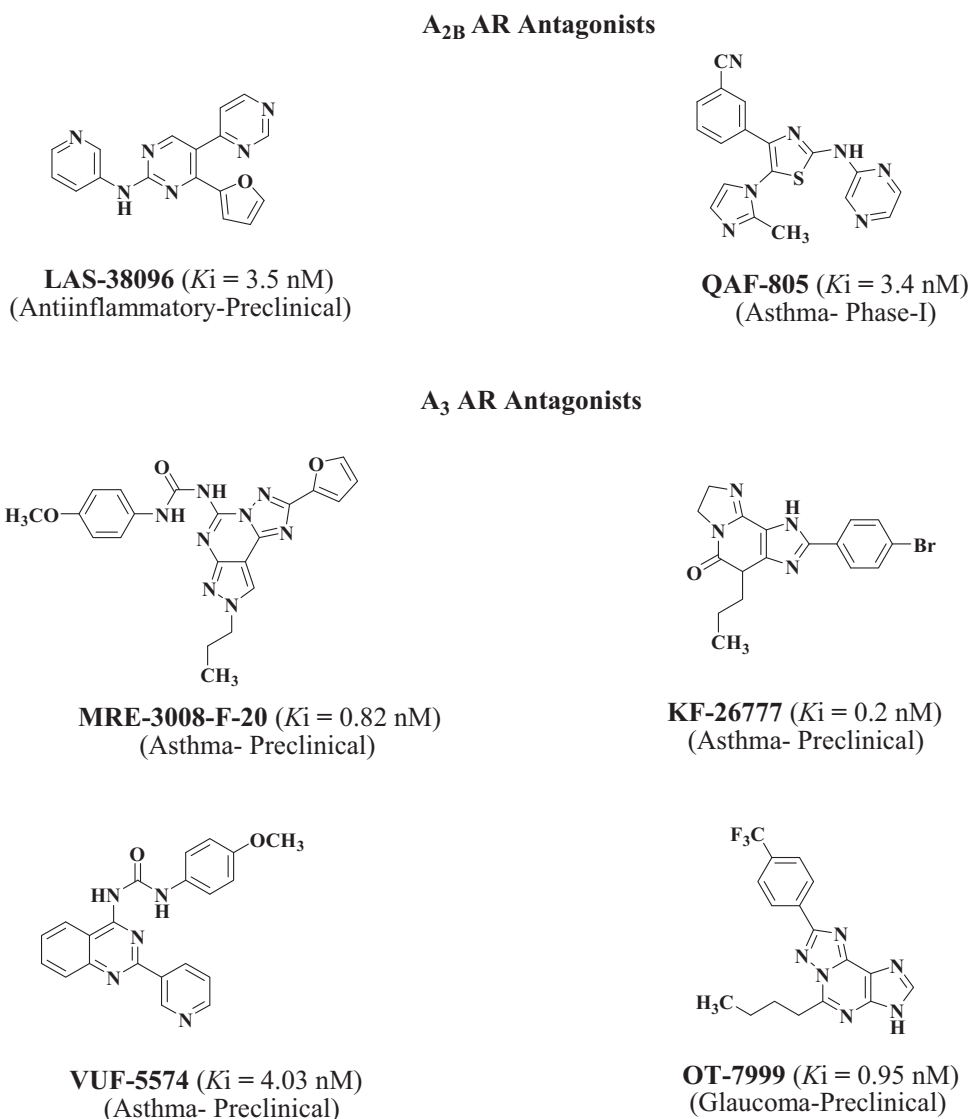


2006; Bulicz et al. 2006; Hafez et al. 2008; Nam et al. 2001). Recently, a series of pyrido[2,3-*d*]pyrimidine derivatives have also been reported as selective bicyclic A_{2B} AR antagonists (Eastwood et al. 2011). Thus, it was thought of interest to further explore this pyrido[2,3-*d*]pyrimidine scaffold and evaluate some novel analogues as potent AR antagonists.

Further, quinazolines are considered as bioisosters of pyrido[2,3-*d*]pyrimidines. Quinazoline derivatives exhibited various interesting pharmacological properties including adenosine receptor antagonism (Bertelli et al. 2000; Khan et al. 2014; Smutny et al. 2016). It has been observed that the presence of a fluoro or trifluoromethyl group at a strategic position on the heterocyclic nucleus enhances the activity of a molecule by increasing the lipophilicity (Balakumar et al. 2010). It is interesting to note that the pyrimidine core being part of the endogenous ligand of ARs

(adenosine), is a recurrent substructural motif of various heterofused bicyclic and tricyclic AR antagonists. Based on these observations, we earlier designed (Pran Kishore et al. 2011; Balakumar et al. 2017), synthesized and reported a series of triazolothienopyrimidines (Raghu Prasad et al. 2008), 4*H*-pyrimido[2,1-*b*]benzothiazole-2-aryl-amino-3-cyano-4-ones (Balakumar et al. 2012), pyrido[3,2-*e*][1,2,4]triazolo[1,5-*c*]pyrimidines (Veeraswamy et al. 2013), 2-amino[1,2,4]triazolo[1,5-*c*]quinazolines (Burbiel et al. 2016) as possible ARs antagonists. In continuation to our ongoing efforts, we are reporting a convenient method for the synthesis of trifluoromethyl substituted pyrido[2,3-*d*]pyrimidine analogues from a versatile precursor **1** (2-amino-3-cyano-4-trifluoro-methyl-6-phenyl-pyridine). In order to understand and illustrate the reactivity of this key precursor (**1**), we examined the X-ray crystal structure to gain insight into its structural features. All the newly synthesized

Fig. 2 Recently reported potent hA_{2B} and hA₃ ARs antagonists under different stages of development



compounds (**2a–c**, **4**, **6a–b**) along with the previously reported bioisosteric fluorinated fused quinazolines (**17a–c** and **18a–c**) have been evaluated for their binding affinity towards all the subtype of ARs.

Experimental

Reactions were routinely monitored by thin layer chromatography (TLC) on silica gel (*precoated F₂₅₄ Merck plates*) and the spots were visualized under UV light (254 nm). Melting points were recorded in open capillaries on *LABINDIA* melting point apparatus and were uncorrected. IR spectra were recorded on *Perkin Elmer FT-IR Spectrometer* (Spectrum RX I) using KBr pellet technique. ¹H and ¹³C NMR spectra were determined in CDCl₃ or DMSO-*d*₆ with a *Bruker Avance II* [400 MHz (¹H) and 100 MHz (¹³C)] spectrometer and signals were recorded in parts per million (δ) downfield from tetramethylsilane as an internal standard. Mass spectra (ESI) were recorded on *Waters Micromass Q-TOF Micro*. Silica gel (60–120 mesh) was used for column chromatography. The X-ray crystallographic study was performed for precursor **1**. All measurements were performed on an *Oxford X Calibur Mova* diffractometer using graphite-monochromated MoK_α radiation (λ = 0.71069 Å) and equipped with a CCD detector.

General procedure for the synthesis of compounds (2a–c)

Dried and pulverized guanidine (0.354 g)/thiourea (0.456 g)/urea (0.36 g) [6 mmol] was added to a solution prepared by dissolving sodium metal (0.138 g, 6 mmol) in absolute ethanol (20 mL) and the mixture was stirred vigorously at 25 °C for 30 min before adding the precursor **1** (0.789 g, 3 mmol). The resulting mixture was heated under reflux for 48–50 h. After completion of the reaction (monitored by TLC), the mixture was poured on to crushed ice, the solid separated was filtered, washed with water. The crude product was dried and purified through column of silica gel (60–120 mesh) using *n*-hexane/ethyl acetate (4:1) as eluent.

7-Phenyl-5-(trifluoromethyl)-pyrido[2,3-*d*]pyrimidin-2,4-diamine (**2a**)

White solid, yield: 68%, M.P.: 226–228 °C, IR: 3399, 3220, 2926, 1582, 1458, 1371, 1260, 1177, 1132 cm^{−1}; ¹H NMR (DMSO-*d*₆, 400 MHz) δ 6.27 (*s*, 2H, NH₂), 7.32 (*s*, 1H, Ar-H), 7.43–7.53 (*m*, 3H, Ar-H), 7.75 (*s*, 1H, NH), 8.04–8.12 (*m*, 2H, Ar-H and 1H, NH) ppm; ¹³C NMR (DMSO-*d*₆, 100 MHz) δ 103.78, 113.35, 124.36, 126.60, 128.83, 129.51, 130.46, 135.07, 137.48, 155.60, 156.41, 166.67

ppm; MS (ESI): *m/z* (%) = 306.1 (15) [*M* + H]⁺, 265.0 (100) [*M*⁺ − 42]. Elemental analysis: calcd. For C₁₄H₁₀F₃N₅: C, 55.08; H, 3.30; N, 22.94%. Found: C, 55.14; H, 3.36; N, 22.87%.

4-Amino-7-phenyl-5-(trifluoromethyl)-pyrido[2,3-*d*]pyrimidin-2(1H)-thione (**2b**)

Yellow solid, yield: 64%, M.P.: 236–237 °C, IR: 3468, 3395, 3301, 3180, 1637, 1581, 1371, 1261, 1198, 1131 cm^{−1}; ¹H NMR (DMSO-*d*₆, 400 MHz) δ 6.30 (*s*, 2H, NH₂), 7.46–7.51 (*m*, 3H, Ar-H), 7.77 (*s*, 1H, NH), 8.06–8.10 (*m*, 3H, Ar-H) ppm; ¹³C NMR (DMSO-*d*₆, 100 MHz) δ 104.8, 113.3, 121.4, 126.6, 128.8, 129.9, 135.4, 137.9, 155.6, 156.8, 166.6 ppm; MS (ESI): *m/z* (%) = 323.1 (15) [*M* + H]⁺, 280.1 (100) [*M*⁺ − 42]. Elemental analysis: calcd. For C₁₄H₉F₃N₄S: C, 52.17; H, 2.81; N, 17.38%. Found: C, 52.12; H, 2.87; N, 17.44%.

4-Amino-7-phenyl-5-(trifluoromethyl)-pyrido[2,3-*d*]pyrimidin-2(1H)-one (**2c**)

White solid, yield: 68%, M.P.: 267–269 °C, IR: 3467, 3395, 3302, 3178, 1637, 1370, 1198, 1133 cm^{−1}; ¹H NMR (DMSO-*d*₆, 400 MHz) δ 6.29 (*s*, 2H, NH₂), 7.45–7.51 (*m*, 3H, Ar-H), 7.77 (*s*, 1H, NH), 8.06–8.09 (*m*, 3H, Ar-H) ppm; ¹³C NMR (DMSO-*d*₆, 100 MHz) δ 103.6, 113.3, 121.6, 126.6, 128.6, 129.5, 135.1, 137.5, 155.6, 156.4, 166.6 ppm; MS (ESI): *m/z* (%) = 307.1 (15) [*M* + H]⁺, 264.1 (100) [*M*⁺ − 42]. Elemental analysis: calcd. For C₁₄H₉F₃N₄O: C, 54.91; H, 2.96; N, 18.29%. Found: C, 54.85; H, 2.99; N, 18.22%.

General procedure for the synthesis of 2-Amino-3-carboxamido-4-trifluoromethyl-6-phenyl pyridine (**3**)

It was prepared by following the reported method (Bhalerao and Krishnaiah 1995).

General procedure for the synthesis of 7-Phenyl-5-(trifluoromethyl)-pyrido[2,3-*d*]pyrimidin-2,4(1H, 3H)-dione (**4**)

A solution of triethylamine (0.218 g, 2.16 mmol) in anhydrous tetrahydrofuran (3 mL) was added dropwise to a mixture of **3** (0.258 g, 0.92 mmol) and triphosgene (0.106 g, 0.36 mmol) in anhydrous tetrahydrofuran (20 mL). The mixture was heated under reflux for 4 h, then cooled to room temperature and poured on to crushed ice. The solid separated was collected by filtration and washed with water. The crude product was dried and purified through column of silica gel (60–120 mesh) using *n*-hexane/ethyl acetate (4:1).

Pale yellow, yield: 68%, M.P.: > 300 °C, IR: 3267, 3132, 3053, 2847, 1739, 1694, 1595, 1570, 1402, 1368, 1130 cm^{-1} ; ^1H NMR ($\text{DMSO}-d_6$, 400 MHz) δ 7.53–7.55 (*m*, 3H, Ar-H), 7.89 (*s*, 1H, Ar-H), 8.13–8.16 (*m*, 2H, Ar-H), 11.55 (*s*, 1H, NH), 11.92 (*s*, 1H, NH) ppm; ^{13}C NMR ($\text{DMSO}-d_6$, 100 MHz) δ 104.62, 112.65, 123.49, 127.59, 128.98, 131.36, 135.76, 137.74, 149.87, 154.35, 159.4, 160.98 ppm; MS (ESI): m/z (%) = 308.1 (100) $[M + \text{H}]^+$, 264.1 (78) $[M - 43]^+$. Elemental analysis: calcd. For $\text{C}_{14}\text{H}_8\text{F}_3\text{N}_3\text{O}_2$: C, 54.73; H, 2.62; N, 13.68%. Found: C, 54.79; H, 2.68; N, 13.61%.

General procedure for the synthesis of 3-(4,5-Dihydro-1H-imidazol-2-yl)-6-phenyl-4-(trifluoromethyl)pyridin-2-amine (5)

A mixture of compound **1** (1.052 g, 4 mmol), ethylenediamine (0.96 g, 16 mmol) and sulfur (0.032 g, 1 mmol) was heated under reflux on an oil bath (120 °C) for 4 h. The progress of the reaction was monitored by TLC (EtOAc/MeOH, 4:1). After completion of the reaction, the mixture was cooled to room temperature and cold water was added and extracted with chloroform. The organic layer was dried over anhydrous Na_2SO_4 and evaporated under reduced pressure. The residue obtained was recrystallized using cyclohexane.

Yellow solids, yield: 74%, M.P.: 232–234 °C, IR: 3473, 3346, 3180, 2931, 2864, 1625, 1582, 1497, 1373, 1267, 1133 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 3.75 (*m*, 4H, 2CH_2), 5.77 (*s*, 2H, NH_2), 7.19 (*s*, 1H, NH), 7.24 (*s*, 1H, Ar-H), 7.41 (*m*, 3H, Ar-H), 7.90 (*m*, 2H, Ar-H) ppm; ^{13}C NMR (CDCl_3 , 100 MHz) δ 25.05, 28.12, 104.48, 112.61, 120.63, 126.20, 127.43, 128.79, 136.81, 137.13, 156.92, 157.02, 160.07 ppm; MS (ESI): m/z (%) = 307.1 (100) $[M + \text{H}]^+$. Elemental analysis: calcd. For $\text{C}_{15}\text{H}_{13}\text{F}_3\text{N}_4$: C, 58.82; H, 4.28; N, 18.29%. Found: C, 58.77; H, 4.35; N, 18.23%.

General procedure for the synthesis of compounds (6a–b)

A mixture of compound **5** (1.53 g, 5 mmol) and triethylorthoformate/triethylorthoacetate (15 mL) was heated under reflux for 3 h. After cooling, the reaction mixture was poured on to cold water. The solid product was isolated by filtration, washed with water, dried, and crystallized using cyclohexane.

8-Phenyl-10-(trifluoromethyl)-2,3-dihydroimidazo[1,2-c]pyrido[3,2-e]pyrimidine (6a)

White solid, yield: 61%, M.P.: 262–263 °C, IR: 1632, 1560, 1463, 1361, 1198, 1148, 1115 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 4.08–4.13 (*m*, 2H, CH_2), 4.24–4.29 (*m*, 2H, CH_2), 7.49–7.52 (*m*, 3H, Ar-H), 8.01–8.03 (*m*, 2H, Ar-H),

8.16–8.19 (*m*, 2H, Ar-H) ppm; MS (ESI): m/z (%) = 339.4 (100) $[M + \text{Na}]^+$, 317.4 (65) $[M + \text{H}]^+$. Elemental analysis: calcd. For $\text{C}_{16}\text{H}_{11}\text{F}_3\text{N}_4$: C, 60.76; H, 3.51; N, 17.71%. Found: C, 60.81; H, 3.58; N, 17.66%.

5-Methyl-8-phenyl-10-(trifluoromethyl)-2,3-dihydroimidazo[1,2-c]pyrido[3,2-e]pyrimidine (6b)

White solid, yield: 52%, M.P.: 282–283 °C, IR: 1633, 1561, 1467, 1304, 1265, 1129 cm^{-1} ; ^1H NMR (CDCl_3 , 400 MHz) δ 2.64 (*s*, 3H, CH_3), 4.14 (*t*, 2H, CH_2), 4.31 (*t*, 2H, CH_2), 7.47–7.51 (*m*, 3H, Ar-H), 7.83 (*s*, 1H, Ar-H), 8.11 (*m*, 2H, Ar-H) ppm; MS (ESI): m/z (%) = 353.1 (100) $[M + \text{Na}]^+$, 331.1 (65) $[M + \text{H}]^+$. Elemental analysis: calcd. For $\text{C}_{17}\text{H}_{13}\text{F}_3\text{N}_4$: C, 61.82; H, 3.97; N, 16.96%. Found: C, 61.89; H, 3.91; N, 16.88%.

X-ray crystallographic study of the precursor compound (1)

The intensities were corrected for Lorentz and polarization effects, and a multi-scan absorption correction was also performed. The structure was solved by direct methods using SHELXS97 (Sheldrick 1997), which revealed the positions of all non-hydrogen atoms. The non-hydrogen atoms were refined anisotropically. All of the hydrogen atoms were fixed at geometrically calculated positions and each was assigned a fixed isotropic displacement parameter with a value equal to 1.2 U_{eq} of its parent atom. They were riding with their bonded atoms. A weighting scheme of the form $w = 1/[\sigma^2(F_o^2) + (ap)^2 + bp]$ with $a = 0.0291$ and $b = 0.0$ was used. The refinement converged to a final R value of 0.0438 ($wR_2 = 0.0810$ for 2276 reflections, $[I > 2\sigma(I)]$). The final difference map was featureless. The data collection and refinement parameters are given in supporting information.

Biological activity

Binding at human A_1 , A_{2A} and A_3 ARs

Binding studies at hA_1 , hA_{2A} , and hA_3 ARs were carried out by employing our previously reported procedures (Klotz et al. 1998). Chinese hamster ovary (CHO) cells stably transfected with human (h) A_1 , A_{2A} , and A_3 ARs subtypes were used for the preparation of membranes for radioligand binding studies. Further, 1 nM [^3H]-2-chloro-6-cyclopentyl adenosine ([^3H]CCPA), 10 nM [^3H]-5'- N -ethylcarboxamido adenosine ([^3H]NECA) and 1 nM [^3H]-2-(1-hexynyl)- N^6 -methyl adenosine ([^3H]HEMADO) were used as radioligands at the hA_1 , hA_{2A} and hA_3 ARs, respectively. Nonspecific binding of [^3H]CCPA was determined in presence of 1 mM theophylline, while 100 μM (R)- N^6 -phenylisopropyladenosine (R-PIA) was used for [^3H]NECA and

interesting clue that the nucleophilicity of the 2-amino group could be the rate limiting factor in these reactions.

Pyrido[2,3-*d*]pyrimidine analogue **4** was obtained by the cyclization of an intermediate 2-amino-3-carboxamido-4-trifluoromethyl-6-phenylpyridine **3** (Bhalerao and Krishnaiah 1995) using triphosgene in the presence of basic medium (triethylamine). The carbonyl group showed a strong signal at 1694 cm^{-1} in its IR spectrum and the ^1H NMR spectrum exhibited two signals at δ 11.55 ppm and δ 11.92 ppm corresponding to two NH protons. Further, fused pyrido[2,3-*d*]pyrimidines (**6a** and **6b**) were obtained by the cyclization of 3-(4,5-dihydro-1*H*-imidazol-2-yl)-6-phenyl-4-(trifluoromethyl)pyridin-2-amine (**5**) a new intermediate with one carbon donor reagents like triethylorthoformate/triethylorthoacetate. The intermediate **5** was synthesized by the functional conversion of cyano group in **1** to an imidazoline group by using 1,2-ethylene-diamine in the presence of sulfur as a catalyst. The IR spectrum of compound **5** showed disappearance of $\text{C}\equiv\text{N}$ stretch and an absorption band at 3180 cm^{-1} related to the N–H stretch. Its ^1H NMR spectrum showed a multiplet signal at δ 3.75 ppm for methylene protons and a singlet at δ 7.19 ppm corresponding to the imidazoline –NH. The IR and ^1H NMR spectra of **6a–b** showed the absence of characteristic absorption bands for primary and secondary amino groups which authenticated the formation of cyclized compounds. The conversion of cyano to other groups (carboxamido or imidazolino) improved the nucleophilicity of the amino group of the compound **5** as compared to the precursor (**1**), facilitating subsequent convenient cyclization reaction under mild conditions to result in the formation of fused-pyrimidine analogues.

X-ray diffraction analysis of compound **1**

To explore the versatility of this precursor compound (**1**), a crystallization process with various solvents was attempted. Interestingly, distinct monoclinic crystals were obtained when absolute ethanol was used as a crystallizing solvent. It prompted us to investigate further its X-ray crystallographic data to understand the reactivity of this key intermediate towards a variety of reagents. Compound **1** crystallizes in a monoclinic crystal system (space group = $P2_1/c$) with four molecules in the unit cell. A perspective view of the molecule with numbering scheme is shown in Fig. 3.

All the bond lengths and angles observed in the structure are normal. The cyano group, as expected, is almost linear with a C2–C7–N3 angle of $177.8(2)^\circ$. The trifluoromethyl group orients with the pyridyl ring in such a fashion that there is a short contact of 2.391 \AA between one of the fluorine atoms and the adjacent hydrogen (F2...H4A = 2.391 \AA , C4–H4A... F2 = 100.08°) which can be considered as an intramolecular C–H...F interaction.

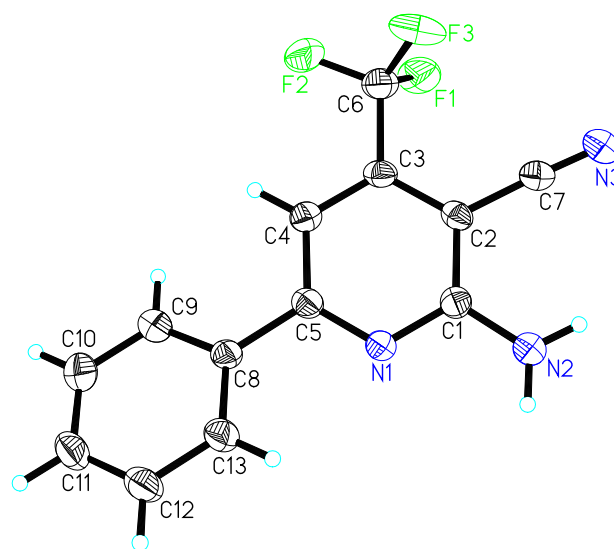


Fig. 3 An ORTEP view of compound **1** with atom numbering scheme employed

One of the noteworthy features in the molecular structure of **1** is the near planar orientation of the phenyl group with the pyridyl ring, the torsional angle N1–C5–C8–C13 is only 18.6° , even though there can be complete flexibility in rotation of C5–C8 bond. The cause for such an orientation is two-fold, the better resonance stabilization as the two aromatic rings become co-planar and the structure stabilization through an intramolecular C–H...N hydrogen bonding between N1 and one of the *ortho* hydrogen atoms (H13A) of the phenyl ring (N1...H13A = 2.509 \AA , C13–H13A... N1 = 99.6°). In the crystal lattice, molecules are packed in a herringbone manner as shown in Fig. 4. In the wedge shape region of the herringbone two molecules form a centrosymmetric dimer through a pair of N–H...N hydrogen bonds (N3...H2B = 2.212 \AA , N2–H2B...N3 = 160.01°) between the adjacent –NH₂ and –CN functionality (Fig. 5a). Further, these dimers stack one over the other through a sheared manner (Fig. 5b) in which the pyridyl and phenyl rings are 3.84 \AA apart that shows $\pi\cdots\pi$ stacking interactions between the aromatic moieties. There are no other short contacts observed in the crystal lattice.

The observed coplanarity of the two rings (phenyl and pyridyl) of the precursor might have contributed towards its improved reactivity of both amino and nitrile groups located on the pyridine ring. This could probably be the reason why several reactions on nitrile and amino groups individually as well as together in one step have been successfully conducted. Our observation of the quantitative yields with such condensations using appropriate reagents is noteworthy. Earlier, we reported reactions involving pyridopyrimidine system with low to moderate yields (Suma et al. 2000). Now, it can be understood that the strategically located

phenyl substituent at 6-position of pyridine nucleus could perhaps be responsible for these unexpectedly high yields of these type of reactions. The fluorinated fused quinazolines (**17a–c** and **18a–c**) were synthesized and characterized as per our earlier report (Balakumar et al. 2010) by following the synthetic route as shown in Scheme 2.

Biological activity

All the newly synthesised fluorinated pyrido[2,3-*d*]pyrimidines (**2a–c**, **4** and **6a–b**) and previously reported fluorinated quinazolines (**17a–c** and **18a–c**) were evaluated for their binding affinity and selectivity towards hA₁, A_{2A}, A_{2B}, and A₃ ARs which were expressed in CHO cells. Three different radioligands such as [³H]CCPA, [³H]NECA, and [³H]HEMADO were used for hA₁, hA_{2A}, and hA₃ ARs, respectively. The AR binding affinity and selectivity data of compounds **2a–c**, **4** and **6a–b** are provided in Table 1. A potential agonistic activity of these

compounds towards hA_{2B} AR was investigated by measuring adenylyl cyclase activity, while the amount of inhibition of NECA-stimulated adenylyl cyclase activity would provide a measurement of an antagonistic nature. Interestingly, none of these compounds showed detectable interaction with the hA_{2B} AR (data not shown in Table 1) in an agonistic or antagonistic fashion (EC₅₀ or IC₅₀ values > 90 μM).

In the series of fluorinated pyrido[2,3-*d*]pyrimidines (**2a–c** and **4**), compound **2a** exhibited the maximum binding affinity for hA₃ AR with similar affinity for the hA₁ and somewhat lower affinity for hA_{2A} ARs resulting in a compound with no A₃ selectivity vs. A₁ and moderate selectivity vs. A_{2A} AR (*K_i* hA₁ = 0.62 μM, hA_{2A} = 3.59 μM and hA₃ = 0.42 μM). This could be attributed to the presence of free amino group at 2-position of pyrido[2,3-*d*]pyrimidine nucleus in addition to amino group at 4-position of the scaffold. Replacement of the free amino (NH₂) group with sulphur or oxygen atom in 2-position of the pyrimidine moiety of compounds **2b** and **2c** drastically lowered the affinity for all the AR subtypes. Interestingly, in case of compound **4**, the presence of a carbonyl (C=O) group at 2-position and 4-position of the pyrimidine moiety improved the affinity for hA₁ AR but showed poor selectivity against hA_{2A} and hA₃ ARs (*K_i* hA₁ = 0.17 μM, hA_{2A} = 0.67 μM and hA₃ = 0.68 μM). The fusion of an imidazole ring with the pyrimido moiety of the scaffold in case of compounds **6a** and **6b** further showed detrimental effect on the affinity for all the ARs.

There are numerous examples of more potent antagonists for A₁, A_{2A}, and A₃AR (Fredholm et al. 2001, 2011), however, many structurally novel compounds show affinities in the low micromolar or high nanomolar range (Cagide et al. 2015). Although further development of such ligands is usually aiming at improved affinity this is not essential as many successfully used drugs show affinities in this range such as Ranitidine (0.20 μM) and Loratadine (0.16 μM) (Hill et al. 1997).

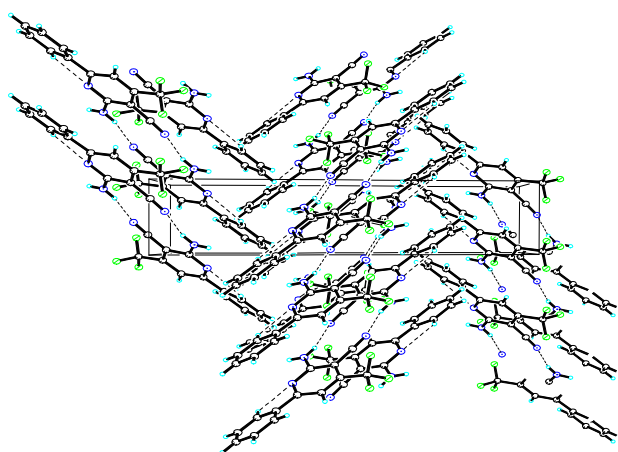


Fig. 4 Packing of **1** viewed down a axis showing the herring-bone arrangement of molecules in the crystal lattice. The dimer formation through N–H...N hydrogen bonds are shown as dashed lines

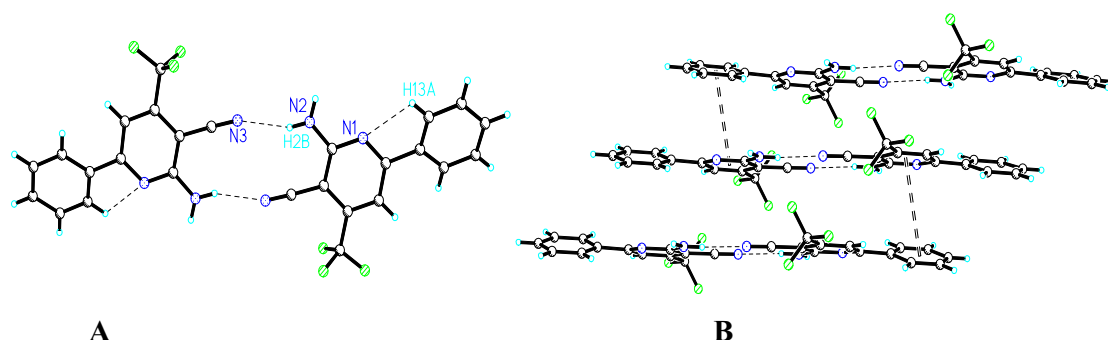
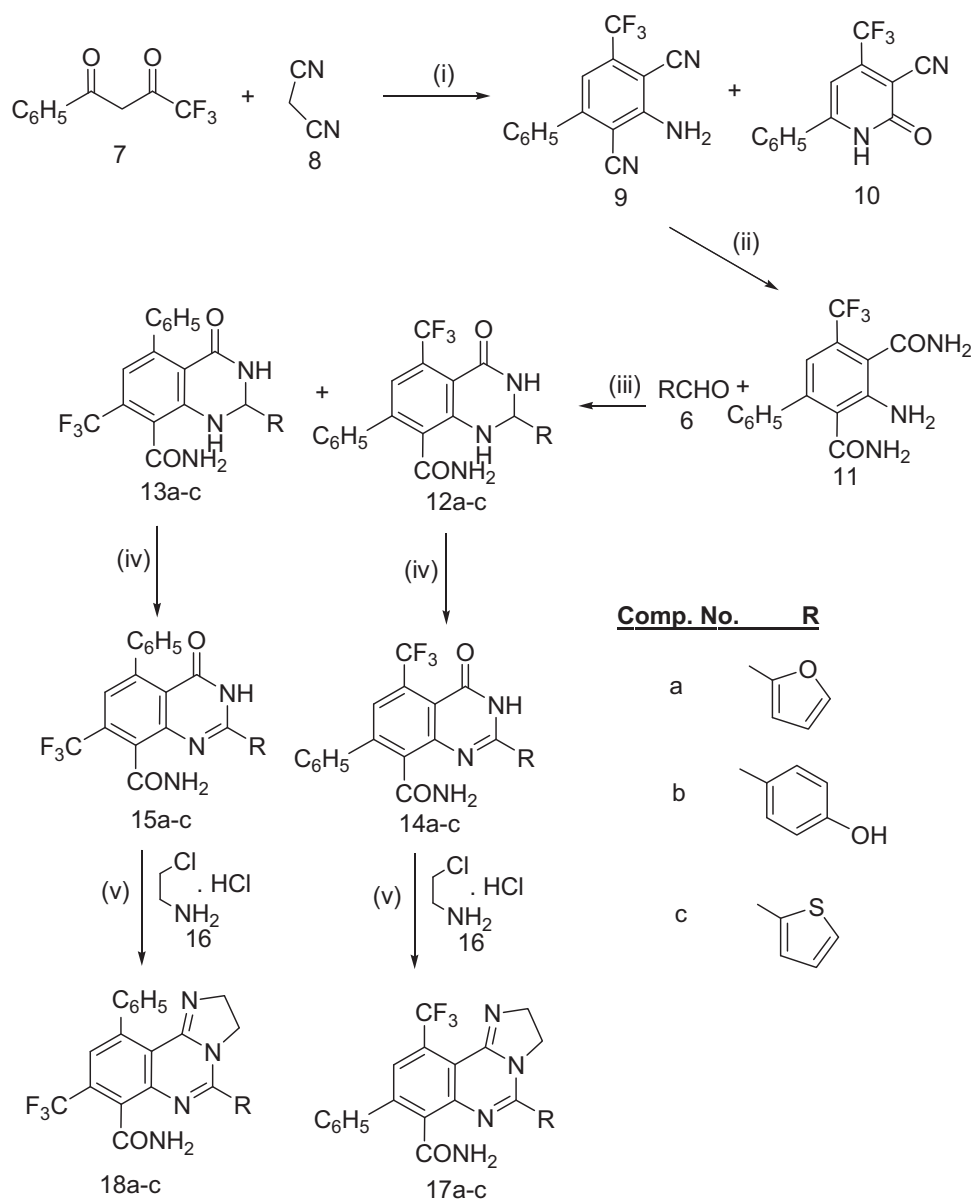


Fig. 5 **a** The dimer formation through N–H...N hydrogen bonding (dashed line) in **1**. The dashed solid line shows the π...π interaction between the pyridyl and phenyl moieties. **b** The stacking pattern of **1**. The dashed solid line shows π...π interaction between the pyridyl and phenyl moieties

Scheme 2 Synthesis of fluorinated quinazolines. Reagents and conditions: (i) methanol, reflux, 6 h; (ii) 20% KOH, reflux, 7 h; (iii) glacial acetic acid, r.t., 4 h; (iv) MnO_2 , dichloromethane, r.t., 2 h; (v) POCl_3 , reflux, 5 h (Balakumar et al. 2010)



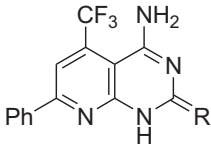
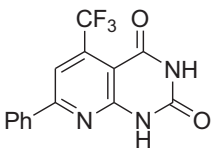
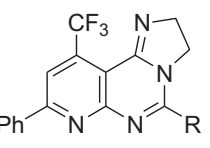
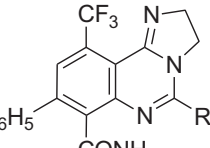
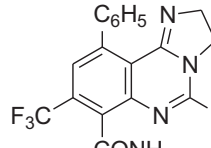


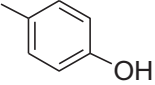
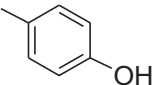


In the quinazoline series, compound **18a** was found to be the most active with substantial binding affinity for A_1 AR as well as significant selectivity against A_{2A} and A_3 ARs (K_i $hA_1 = 0.73 \mu\text{M}$, $hA_{2A} > 30 \mu\text{M}$ and $hA_3 = 9.27 \mu\text{M}$). Surprisingly, other quinazoline derivatives (**17a-c**, **18b** and **18c**) did not show remarkable affinity for any of the AR subtypes.

Conclusion

In conclusion, we have synthesized and characterised some new fluorinated pyrido[2,3-*d*]pyrimidines (**2a-c** and **4**) and fluorinated imidazo[1,2-*c*]pyrido[3,2-*e*]pyrimidines (**6a-b**)

in quantitative yields by using 2-amino-4-trifluoromethyl-6-phenyl nicotinonitrile (**1**) as a precursor following an appropriate synthetic Scheme 1. In order to gain insight into the structural features as well as to illustrate the reactivity of the key precursor (**1**), we also examined the X-ray crystal structure. Further, a series of fluorinated heterofused quinazolines (**17a-c** and **18a-c**) were also synthesized and characterized as they are bioisosteres to pyrido[2,3-*d*]pyrimidines based on our earlier report. Radioligand binding affinity studies of all these compounds for ARs showed that pyrido[2,3-*d*]pyrimidine scaffold with free amino (NH_2) group at 2-position and 4-position (compound **2a**) exhibited good binding affinity for hA_3 AR with no selectivity against hA_1 , but moderate hA_{2A} selectivity. Interestingly, the

Table 1 Binding affinity (K_i) of compounds (**2a–c**, **4**, **6a–b**, **17a–c** and **18a–c**) at hA₁, hA_{2A}, and hA₃ ARs and selectivity against hA₁ and hA_{2A} ARs

		    					
CN	R	K_i , μ M (95 % CI)			Selectivity		
		hA ₁ ^a	hA _{2A} ^b	hA ₃ ^c	hA ₁ /hA _{2A}	hA ₁ /hA ₃	hA _{2A} /hA ₃
2a	NH ₂	0.623 (0.561–0.692)	3.59 (1.76–7.33)	0.429 (0.310–0.595)	0.17	1.5	8.4
2b	S	12.9 (7.52–22.1)	>30	5.85 (3.45–9.91)	>0.43	2.2	>5.1
2c	O	11.1 (6.16–19.9)	>30	19.5 (10.3–37.2)	>0.37	0.57	>1.5
4	–	0.176 (0.161–0.192)	0.678 (0.469–0.980)	0.685 (0.444–1.06)	0.26	0.26	0.99
6a	H	3.08 (1.65–5.73)	7.03 (6.36–7.77)	3.19 (2.43–4.18)	0.44	0.96	2.2
6b	CH ₃	>30	27.3 (22.1–33.7)	13.40 (8.43–21.4)	>1	2.2	2.0
17a		5.84 (4.69–7.27)	22.6 (14.4–35.2)	2.34 (1.42–3.87)	0.25	2.5	9.7
18a		0.73 (0.65–0.82)	>30	9.27 (4.71–18.3)	0.02	0.07	3.2
17b		7.80 (6.81–8.94)	24.5 (17.0–35.3)	2.52 (1.25–5.09)	0.31	3.09	9.7
18b		2.57 (1.35–4.91)	8.62 (4.86–15.3)	3.96 (3.16–4.97)	0.29	0.64	2.2
17c		10.6 (7.72–14.4)	26.0 (22.5–30.1)	3.85 (3.09–4.78)	0.40	2.8	6.8
18c		1.97 (1.77–2.19)	6.62 (4.49–9.76)	1.87 (1.23–2.82)	0.29	1.05	3.5
							

Data are expressed as geometric means, with 95% confidence limits in parentheses

^a Displacement of specific [³H]CCPA binding at human A₁ AR expressed in CHO cells (n = 3–6)^b Displacement of specific [³H]NECA binding at human A_{2A} AR expressed in CHO cells (n = 3–6)^c Displacement of specific [³H]HEMADO binding at human A₃ AR expressed in CHO cells (n = 3–6)

replacement of both the amino groups with carbonyl (C=O) groups (compound **4**) resulted in significantly improved affinity for hA₁ AR but with moderate selectivity against hA_{2A} and hA₃ ARs. Further, fusion of an imidazole ring with the pyrimidine moiety of the scaffold (compounds **6a** and **6b**) showed detrimental effect in affinity for all the ARs. In case of fluorinated quinazolines, only compound **18a** showed good affinity for hA₁ AR with significant selectivity against hA_{2A} and hA₃ ARs. The preliminary result of these compounds demonstrate that the fluorinated pyrido[2,3-*d*]pyrimidine and imidazo[1,2-*c*]pyrido[3,2-*e*]pyrimidine can be considered as promising scaffolds for further optimisation in search of potential antagonists with better affinity and selectivity towards hA₁ and hA₃ ARs.

Acknowledgements We thank Mr. Raveesh Sankolli, IISc, Bangalore for providing the X-ray structural data. One of the authors (CB) is grateful to University Grants Commission (UGC), New Delhi for providing research fellowship [UGC-RFSMS; Grant No. F.4-1/2006 (XI Plan/BSR)]. This work was also supported by AICTE, New Delhi through a grant to ARR under Research Promotion Scheme (RPS-8023/2006-07).

Compliance with ethical standards

Conflict of interest The authors declare that they have no competing interests.

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