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Short Communication

Synthesis of new tricyclic melatoninergic ligands

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Abstract

We report the synthesis and biological evaluation of a series of new tricyclic analogs of the hormone melatonin, which act as probes of the constraints at the hormone's receptor site with regard to the lower N1-C2 region of the indole moiety of melatonin. Three of the new compounds, N-[2-(2-methoxy-6,7,8,9-tetrahydropyrido[1,2-*a*]indol-10-yl)ethyl]acetamide (9), and the respective propionamide 10 and butyramide 11, are as potent as melatonin in the *Xenopus laevis* melanophore model. © 2001 Elsevier Science S.A. All rights reserved.

Keywords: Tricyclic fused [1,2-a]indoles; Synthesis; Xenopus laevis melanophores; Melatonin agonists and antagonists

1. Introduction

The pineal hormone, melatonin, has been shown to have a physiological role in regulating seasonal breeding in photoperiodic species [1] and can entrain circadian rhythms in mammals [2]. In humans it has been suggested that melatonin might prove a valuable therapy for rhythm disorders, including seasonal affective disorders (SAD) [3], jet lag, and shift work problems [4]. Melatonin also inhibits dopamine release from amacrine cells within the retina [5], and can enhance vasoconstriction in the rat tail artery [6]. Many studies have also indicated an influence on immune function [7] and antioxidant actions [8].

A number of these effects are thought to be mediated through interaction with specific receptors and three distinct receptor subtype cDNAs have been isolated [9]. These encode high affinity receptors, which belong to the seven-transmembrane domain receptor family and which inhibit adenylyl cyclase *via* a pertussis toxin sensitive GTP-binding protein.

The understanding of the physiological and pathophysiological role(s) of melatonin in animals and man is hampered by the relatively small number of melatonin receptor agonists and antagonists available. In a series of studies during the last decade we have sought to understand how melatonin binds to and activates its receptor and to use the knowledge gained to design potent receptor agonists and antagonists, which will be useful tools for defining the full physiological and pathophysiological role of this hormone. Thus, several key interactions between ligand and receptor have been identified. The 5-methoxyl group and amide moiety, and their relative position, are critical to high affinity [10-17]. Positions 2 and 6 are also important for activity, and both are involved in the metabolism of melatonin [18,19].

In order to probe the constraints at the receptor site with regard to the lower N1-C2 region of the indole moiety of melatonin, we have recently reported the synthesis and biological activity of a number of 2phenyltryptamines annulated on the [a] face of the pyrrole moiety by the introduction of one, two or three methylene groups (1-3) [20]. In the *Xenopus* melanophore pigment aggregation model compounds 1a-j were found to exhibit agonistic activity while their

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congeners $3\mathbf{a}-\mathbf{j}$ were antagonists. Molecules $2\mathbf{a}-\mathbf{d}$ were antagonists and $2\mathbf{e}-\mathbf{h}$ had agonist acivity at low concentrations but antagonized responses at high concentrations (Fig. 1). These findings suggest that the *Xenopus* melatonin receptor cannot accommodate an *N*-*n*-alkyl chain attached to a phenyl ring substituent with n > 2 (Fig. 1, compounds $3\mathbf{a}-\mathbf{j}$) in the required orientation to induce or stabilize the active receptor conformation.

In the present study, we have extended this work to synthesize and evaluate a number of novel N-[2-(6,7,8,9-tetrahydropyrido[1,2-a]indol-10-yl)ethyl]alkanamides (4-8) and N-[2-(2-methoxy-6,7,8,9-tetrahydropyrido[1,2-*a*]indol-10-yl)ethyl]alkanamides (9-13)(Scheme 1). Similarly to compounds 1-3, these analogs probe the constraints at the receptor site with regard to the lower N1-C2 region of the indole moiety in the absence though of the annulated 2-phenyl ring. The results indicate that the non-methoxy substituted compounds, 4-8, are antagonists in the *Xenopus* melanophore pigment aggregation assay. In contrast, the new 2-methoxy analogs, 9-13, are full agonists, unlike their analogous congeners, 2e-h, which act as partial agonists (Fig. 1) [20].

2. Chemistry

The synthesis of 4-13 is shown in Scheme 1. The sulfone moiety at C-2 of indole (14) and 5-methoxyindole (15) was introduced by the method of Katritzky and Akutagava [21]. The derived compounds 16 and 17 were then N-alkylated to 18 and 19, respectively, using 1,4-dibromobutane in the presence of potassium hydroxide in DMF as first described by Dehaen [22] and later modified by Caddick et al. [23]. The tricyclic molecules 20 and 21 were formed upon treatment of 18 and 19 with tributyltin hydride under radical conditions [23]. Aldehydes 22 and 23 were obtained by applying the Vilsmeier-Haack reaction [20] on compounds 18 and 19 and the sequence of the Henry reaction [20] followed by reduction [20] gave the amines 26 and 27, which were not purified but were immediately acylated with the appropriate reagent [20] to give the desired N-[2-(6,7,8,9-tetrahydropyrido[1,2-*a*]indol-10-yl)ethyl]-alkanamides (4-8) [24] and N-[2-(2-methoxy-6,7,8,9-tetrahydropyrido[1,2-*a*]indol-10-yl)ethyl]alkanamides (9-13) [24].

3. Results and discussion

The agonist and antagonist potency of the new analogs was assessed in a well-established, specific model of melatonin action, the pigment aggregation response of Xenopus laevis melanophores [19,20,25]. All five non-methoxy substituted fused [1,2-a] indoles 4-8had no agonist activity, but displayed weak antagonistic activity like luzindole, the most-commonly used commercially available melatonin receptor antagonist (Table 1). The lower antagonistic potency of the new analogs 4–8 compared to that of 2a-d (cf. *p*IC₅₀ values of 2a-d in Fig. 1) may be due to the presence of the phenyl ring of the tetralin moiety. The phenyl ring in this position might enhance hydrophobic attachments. The 5-methoxy substituted (numbering with respect to the indole ring) molecules 9-13 were all full agonists and three of them, compounds 9-11, were almost equipotent to melatonin (Table 1). Furthermore, the new molecules 9-13, are much more potent agonists than 2e-h (cf. pEC_{50} values of 2a-d in Fig. 1) presumably because the extra aromatic ring in the latter reduces the ability of the receptor binding pocket to



Fig. 1. Tetracyclic analogs 1-3 [20]: compounds 1a-j are agonists in the X. *laevis* melanophore model; compounds 2a-d are antagonists; compounds 2e-h are partial agonists; compounds 3a-j are antagonists. ^a See footnotes, Table 1, for explanation.



Scheme 1. Synthetic methodology followed for the preparation of the new analogs 9–13.

recognize and bind these molecules. The relative decrease in agonist potency observed for compounds 12 and 13 compared to that of 9-11, seems reasonable as the former bear the *N*-cyclopropanoyl and *N*-cyclobutanoyl groups, respectively, in their side-chain. These moieties are known to exert a detrimental effect on agonistic activity [18,26]. In the case of 12 and 13,

however, a shift to antagonism at the expense of their agonistic activity was not observed.

The findings reported herein for analogs 4-13 suggest that the absence or presence of the methoxy substituent in their skeleton is the critical factor in determining agonist or antagonist action. The results are in agreement with earlier data, showing that the

absence of the 5-methoxyl group in tryptamine derivatives, as in luzindole [27], or in 4-phenyl substituted tetralines, as in 4-phenyl-2-acetylaminotetralin [28], leads to antagonism. We attribute the difference in agonist/antagonist behavior between analogs 4-8 and 9-13 to a combination of stereoelectronic effects exerted by both the methoxyl group, where present, and the cyclohexane ring, annulated between N1 and C2 of the indole nucleus. This argument is reinforced by our earlier observations on compounds 2, which, as mentioned previously, instead of a cyclohexane ring bear a tetralin moiety annulated between N1 and C2 [20].

In conclusion, the compounds reported herein are the first examples that utilize a fused cyclohexane at the lower N1-C2 region of the melatonin skeleton in order to probe the constraints at the receptor site. As these preliminary results merit further investigation we are currently engaged with the synthesis and biological evaluation of a new series of tricyclic compounds.

4. Materials and methods

Melanophore cells were grown in 96-well tissue culture plates, and growth medium [29,30] was replaced with $0.7 \times L-15$ culture medium 18 h before analogs were tested. Initial absorbance of cells (A_i , 630 nm) was measured in each well using a Bio-Tek microtiter plate reader (model EL3115, Anachem, UK), then cells were treated with the concentrations of the analogs indicated. All experiments used triplicate wells at six concentrations of analog. The final absorbance (A_f) was measured after 60 min, and the fractional change in absorbance ($1 - A_f/A_i$) was calculated. Vehicle did not alter pigment granule distribution itself or inhibit re-

Table 1

Agonistic and antagonistic activity of compounds 4-13 in the *X*. *laevis* melanophore assay

Comp.	R ₅	R	Agonist	Antagonist
			$(p EC_{50})$	(pIC_{50})
Melatonin			10.07	
Luzindole			NA ^a	5.61
4	Н	CH ₃	NA	3.75
5	Н	C_2H_5	NA	4.24
6	Н	C_3H_7	NA	3.43
7	Н	$c-C_3H_5$	NA	3.71
8	Н	$c-C_4H_7$	NA	3.22
9	OCH ₃	CH ₃	9.29	
10	OCH ₃	C_2H_5	9.78	
11	OCH_3	C_3H_7	9.91	
12	OCH ₃	$c-C_3H_5$	7.73	
13	OCH ₃	c-C ₄ H ₇	5.43	

 $^{\rm a}$ NA, no agonist or antagonist effect detected at 100 $\mu M.$ Agonist and antagonist data on melanophores are the mean of triplicate experiments.

sponses to melatonin. The concentration of analog producing 50% of the maximum agonist response (EC₅₀) was determined from concentration–response curves. For evaluation of antagonist potency, cells were treated with vehicle (1% DMSO or methanol) or varying concentrations (10^{-4} to 10^{-9} M) of the analogs for 60 min before melatonin (10^{-9} M) was added. The concentration of analog reducing melatonin-induced pigment aggregation by 50% (IC₅₀) was determined.

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