View Article Online / Journal Homepage / Table of Contents for this issue

Synthesis of Adenosine $5'[(R)\alpha-^{17}O]$ Triphosphate

Gordon Lowe,*a Gaynor Tansley,a and Paul M. Cullis^b

^a The Dyson Perrins Laboratory, Oxford University, South Parks Road, Oxford OX1 3QY, U.K.

^b Department of Chemistry, Leicester University, Leicester LE1 7RH, U.K.

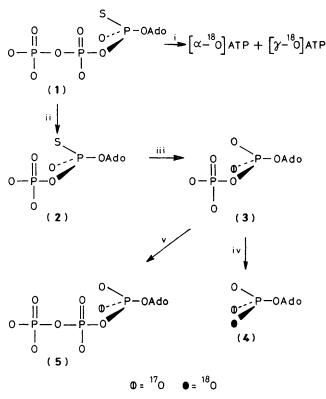
Adenosine $5'[(S)\alpha$ -thio]triphosphate is converted predominantly into adenosine $5'[\gamma^{-18}O]$ triphosphate by bromine in [¹⁸O]water, but adenosine $5'[(S)\alpha$ -thio]diphosphate gives adenosine $5'[(R)\alpha^{-17}O]$ diphosphate on treatment with bromine in [¹⁷O]water which can be converted enzymically into adenosine $5'[(R)\alpha^{-17}O]$ -triphosphate.

A method has been developed for the analysis of the chirality of [¹⁶O, ¹⁷O, ¹⁸O] phosphate esters,¹ which makes possible the determination of the stereochemical course of not only phosphokinases,² phosphomutases,³ phosphatases,⁴ and phospho-

diesterases,⁵ but also the nucleotidyl transferases, a group of enzymes which harness the chemical potential of ATP for the biosynthesis of all the major biopolymers. In order to undertake a stereochemical investigation of this class of enzymes, however, it is necessary to make ATP (or another nucleoside 5'-triphosphate) chiral at P_{α} , ideally by isotopic substitution. We report here a synthesis of adenosine 5' [(*R*) α -¹⁷O]triphosphate.

Adenosine 5'[(S) α -thio]triphosphate (1) can be prepared enzymically from adenosine 5'-phosphorothioate,⁶ and its chirality at P $_{\alpha}$ has been rigorously established.⁷ Although there appears to be no example in the literature of the conversion of an *O*,*O*-dialkyl phosphorothioate into a dialkyl phosphate, *S*,*O*,*O*-trialkyl phosphorothioates have been converted into dialkyl phosphates with bromine–water,⁸ and into trialkyl phosphates with bromine in an alcohol; in the latter case the reaction was shown to occur with inversion of configuration at phosphorus.⁸

Preliminary experiments showed that O,O-diethyl phosphorothioate was rapidly (<4 min) and quantitatively con-



Scheme 1. Reagents: i, Br₂, [¹⁸O]water; ii, hexokinase or myosin, Mg^{2+} ; iii, Br₂, [¹⁷O]water; iv, snake venom phosphodiesterase, [¹⁸O]water; v, phosphoenolpyruvate, pyruvate kinase, Mg^{2+} , K⁺.

J. CHEM. SOC., CHEM. COMMUN., 1982

verted into diethyl phosphate by bromine-water at ambient temperature. Since bromine-water is known to convert adenine nucleotides into 8-bromo-adenine nucleotides in buffer (pH 4.0 over several hours) at ambient temperature,¹⁰ it was necessary to establish that the reaction time was sufficiently short to avoid brominating the adenine ring. Adenosine 5'-phosphorothioate¹¹ (35 μ mol) in water (70 μ l) containing bromine (88 μ mol) was kept for 4 min at ambient temperature and then the excess of bromine destroyed with sodium hydrogen sulphite. Adenosine 5'-phosphate was obtained quantitatively as shown by ¹H and ³¹P n.m.r. spectroscopy and adenylate deaminase assay.

Adenosine 5'[(S) α -thio]triphosphate (1) was rapidly (<4 min) converted into ATP by bromine-water at ambient temperature, but when the reaction was run in [¹⁸O]water, the ¹⁸O incorporated into ATP was distributed between P_{α} and P_{γ} in the ratio of 1:4 as shown by ³¹P n.m.r. spectroscopy.¹² This was presumably due to the activated intermediate reacting directly with [¹⁸O]water to give [α -¹⁸O]ATP and intramolecularly with the γ -phosphate residue to give adenosine 5'-trimetaphosphate¹³ which would hydrolyse to [γ -¹⁸O]ATP (Scheme 1).

In order to circumvent this problem, adenosine $5'[(S)\alpha$ thio]triphosphate (1) was hydrolysed to adenosine $5'[(S)\alpha$ thio]diphosphate (2) with hexokinase¹⁴ (or myosin¹⁵). Adenosine 5'[(S) α -thio]diphosphate (2) (300 μ mol) in [¹⁷O]water (420 μ l; 4 atom % ¹⁶O, 43 atom % ¹⁷O, and 53 atom % ¹⁸O) containing bromine (1.5 mmol) was kept for 4 min at ambient temperature and then the excess of bromine destroyed by the addition of anhydrous sodium hydrogen sulphite. The [17O]water can be recovered by lyophilization on a vacuum line and the adenosine 5'[α -17O]diphosphate (3) isolated by ionexchange chromatography on DEAE-Sephadex A25, with elution with triethylammonium hydrogen carbonate buffer, pH 7.6. A portion of the adenosine $5'[\alpha^{-17}O]$ diphosphate (3) was hydrolysed in [18O]water by snake venom phosphodiesterase to adenosine 5' [18O, 17O, 18O] phosphate (4) which was shown to have the (S)-configuration at phosphorus by our established analytical procedure.¹ The ³¹P n.m.r. spectrum of the adenosine 5' [16O, 17O, 18O] phosphate after cyclization and methylation is shown in Figure 1. Since the hydrolysis of adenosine 5'triphosphate by snake venom phosphodiesterase has been shown to proceed with retention of configuration at phosphorus,^{5a} the adenosine $5'[\alpha^{-17}O]$ diphosphate must be the (R_p) -diastereoisomer (3), and hence the displacement of sulphur from adenosine $5'[(S)\alpha$ -thio]diphosphate with bromine-water proceeds with inversion of configuration. Comparison of the observed relative intensities of the ³¹P resonances from Figure 1 with those calculated for substitution with inversion of configuration, however, indicates that a small amount of racemization has occurred; the best fit of the data indicated that the reaction proceeds with 93% inversion and 7% retention of configuration (Table 1).

Table 1. Comparison of the observed relative peak intensities of the ³¹P resonances taken from Figure 1 with the calculated values expected for conversion of adenosine $5'[(S)\alpha$ -thio]diphosphate into adenosine $5'[\alpha^{-17}O]$ diphosphate with 100% and 93% inversion of configuration at P_{α}. The isotopically labelled diastereoisomeric triesters were derived by cyclization followed by methylation of the adenosine $5'[\alpha^{-17}O]$ diphosphate with snake venom phosphodiesterase in [¹⁸O]water (80 atom % ¹⁸O).

	Equatorial triester			Axial triester		
		Calcula			Calculated for	
MeO-P=O Me ¹⁸ O-P=O MeO-P= ¹⁸ O Me ¹⁸ O-P= ¹⁸ O	Observed 0.41 1.00 0.71 0.51	100% Inversion 0.40 1.00 0.67 0.44	93% Inversion 7% Retention 0.41 1.00 0.71 0.45	Observed 0.41 0.72 1.00 0.48	100% Inversion 0.40 0.67 1.00 0.44	93% Inversion 7% Retention 0.41 0.71 1.00 0.45

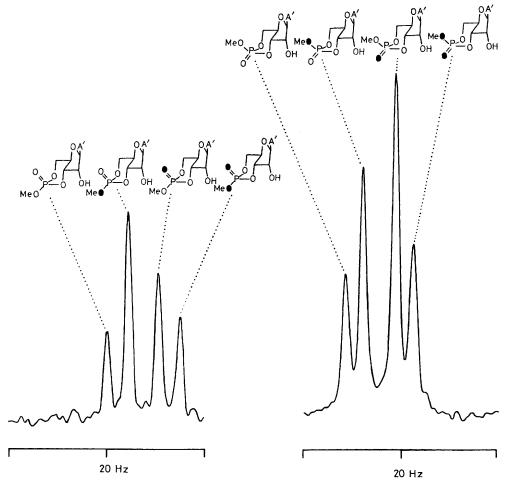


Figure 1. ³¹P N.m.r. spectrum (121.5 MHz) of the equatorial and axial triesters derived by the cyclization and methylation of the adenosine 5'[¹⁶O, ¹⁷O, ¹⁸O]phosphate obtained by snake venom phosphodiesterase-catalysed hydrolysis of adenosine 5'[α -¹⁷O]-diphosphate in [¹⁸O]water. The ratio of the [¹⁶O_{ax}, ¹⁸O_{eq}]-to [¹⁸O_{ax}, ¹⁶O_{eq}]-triesters shows that the adenosine 5'[¹⁶O, ¹⁷O, ¹⁸O]phosphate has the (S_p)-configuration and hence the adenosine 5'[α -¹⁷O]diphosphate has the (R_p)-configuration as shown in Scheme 1. $\bigcirc = ^{18}O$. A' = N¹-methyladenine. The solvent and ³¹P n.m.r. parameters are as reported previously.¹

Adenosine 5' [(R) α -17O]diphosphate (3) was then converted into adenosine $5'[(R)\alpha^{-17}O]$ triphosphate (5) with pyruvate kinase and phosphoenolpyruvate. The ³¹P n.m.r. spectrum of the adenosine $5'[(R)\alpha^{-17}O]$ triphosphate (5), isolated by ion exchange chromatography on DEAE-Sephadex A25 (with triethylammonium hydrogen carbonate buffer, pH 7.6) is compared with that of ATP in Figure 2. If the ¹⁷O site was fully enriched in adenosine $5'[(R)\alpha^{-17}O]$ triphosphate (5) only the P_β and P_γ resonances would be observed in the ^{31}P n.m.r. spectrum since ¹⁷O directly bonded to ³¹P causes nuclear electric quadrupolar relaxation of the ³¹P signal.¹⁶ In ATP (Figure 2b) the P_{α} resonance is more intense than those of P_{β} and P_{γ} in the proton-decoupled spectrum owing to the nuclear Overhauser effect; in the spectrum of adenosine $5'[(R)\alpha^{-17}O]$ triphosphate (Figure 2a) the P_{α} resonance is seen to be less intense than those of P_{β} and P_{γ} .

From the relative intensities it can be calculated that the ¹⁷O enrichment at P_{α} is about 40 atom %. The expanded P_{α} resonance in the spectrum of adenosine 5'[(R) α -¹⁷O]triphosphate (Figure 2a inset) is seen to be two doublets due to ATP (minor doublet) and adenosine 5'[(R) α -¹⁸O]triphosphate (major doublet) whereas the expanded P_{β} resonance is a single triplet as expected; the separation of the P_{α} doublets is caused by the upfield ¹⁸O isotope shift.¹² From the ³¹P n.m.r. spectrum the isotopic composition of the labelled site in adenosine 5'[(R) α -¹⁷O]triphosphate can be calculated to be

5 atom % ¹⁸O, 40 atom % ¹⁷O, and 55 atom % ¹⁸O, in good agreement with the isotopic composition of the [¹⁷O]water used in the synthesis (*vide supra*). This material is now being used to investigate the stereochemical course of a wide range of nucleotidyl transferases.

Added in proof: The substitution of S by ¹⁸O in nucleoside phosphorothioates has also been achieved with Nbromosuccinimide in dioxan/[18O]water, but the claim that the reaction is stereospecific is not justified by the spectroscopic evidence.¹⁷ When we treated thymidine 3', 5'(R)-cyclic phosphorothioate with N-bromosuccinimide in dioxan/[18O]water under the conditions used by Connolly et al.17 (we are grateful to Professor Eckstein for informing us of this work prior to publication), the thymidine 3',5'-cyclic[18O]phosphate obtained was shown by ³¹P n.m.r. spectroscopy (after methylation) to consist of 77% of the S_p and 23% of the $R_{\rm p}$ stereoisomers. In subsequent correspondence Professor Eckstein informed us that he has confirmed that the reaction of N-bromosuccinimide in dioxan/[18O]water is not stereospecific and has now found that replacement of S by ¹⁸O in both cyclic and acyclic phosphorothioates occurs with ca. 80% inversion and 20% retention of configuration at phosphorus (personal communication).

Cyanogen bromide in [¹⁸O]water has also been used to prepare the R_p and S_p stereoisomers of adenosine 5'[α -¹⁸O]diphosphate from the S_p and R_p diastereoisomers of adeno-

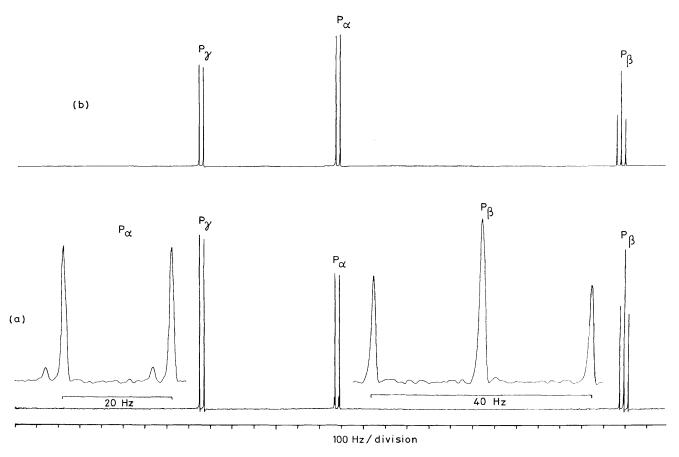


Figure 2. ³¹P N.m.r. spectrum (121.5 MHz) of (a) adenosine 5'[(R) α -¹⁷O]triphosphate with expansion of the P_{α} and P_{β} resonances, and (b) ATP, in 50% D₂O containing ethylenediaminetetra-acetic acid (10 mM) at pH 9.0. ³¹P N.m.r. parameters are: offset 1150 Hz, sweep width 3012 Hz, acquisition time 1.36 s, pulse width (angle) 15 μ s (75°), gaussian multiplication (line broadening-1.2 Hz, gaussian broadening 0.4) in 8K, and Fourier transform in 32K.

sine 5'[1-thio,2-cyanoethyl]diphosphate but the ³¹P n.m.r. spectra reported for determining the stereochemistry at P_{α} of the adenosine 5'[α -¹⁸O]diphosphates again do not permit an accurate assessment to be made of the stereoselectivity of the reaction.¹⁸ Moreover the synthetic route is significantly longer than the procedure reported here or by Connolly *et al.*¹⁷

We gratefully acknowledge a research studentship (to G. T.) and financial support from the S.E.R.C. This is a contribution from the Oxford Enzyme Group supported by the S.E.R.C. *Received*, 22nd February 1982; Com. 194

necessea, 22na reoraary 1902, 4

References

- 1 R. L. Jarvest, G. Lowe, and B. V. L. Potter, J. Chem. Soc., Perkin Trans. 1, 1981, 3186.
- 2 G. Lowe, P. M. Cullis, R. L. Jarvest, B. V. L. Potter, and B. S. Sproat, *Philos. Trans. R. Soc. London, Ser. B*, 1981, 293, 75; G. Lowe, P. M. Cullis, R. L. Jarvest, and B. V. L. Potter, 'Phosphorus Chemistry,' eds. L. D. Quin and J. G. Verkade, Am. Chem. Soc. Symposium Series, 1981, vol. 171, p. 103; G. Lowe and B. V. L. Potter, *Biochem. J.*, 1981, 199, 227; R. L. Jarvest and G. Lowe, *ibid.*, p. 273; R. L. Jarvest, G. Lowe, and B. V. L. Potter, P. M. Cullis, G. Lowe, and A. Cornish-Bowden, *ibid.*, 1982, 201, 421.
- 3 G. Lowe and B. V. L. Potter, Biochem. J., 1981, 199, 693.
- 4 G. Lowe and B. V. L. Potter, *Biochem. J.*, 1982, 201, 665.
- 5 (a) R. L. Jarvest and G. Lowe, *Biochem. J.*, 1981, 199, 447;
 (b) R. L. Jarvest, G. Lowe, J. Baraniak, and W. J. Stec, *Biochem. J.*, 1982, 203, 461.

- 6 K.-F. R. Sheu and P. A. Frey, J. Biol. Chem., 1977, 252, 4445;
 E. K. Jaffe and M. Cohn, Biochemistry, 1978, 17, 652.
- 7 R. L. Jarvest and G. Lowe, J. Chem. Soc., Chem. Commun., 1979, 364; P. M. J. Burgers and F. Eckstein, Proc. Natl. Acad. Sci. U.S.A., 1978, 75, 4798; F. R. Bryant and S. J. Benkovic, Biochemistry, 1979, 18, 2825.
- 8 C. J. M. Stirling, J. Chem. Soc., 1957, 3597.
- 9 D. B. Cooper, C. R. Hall, J. M. Harrison, and T. D. Inch, J. Chem. Soc., 1977, 1969.
- 10 M. Ikehara, S. Uesugi, and M. Kaneko, J. Chem. Soc., Chem. Commun., 1967, 17; M. Ikehara and S. Uesugi, Chem. Pharm. Bull., 1969, 17, 348; M. Ikehara, I. Tazawa, and T. Fukui, *ibid.*, p. 1019.
- 11 A. W. Murray and R. Atkinson, Biochemistry, 1968, 7, 4023.
- 12 M. Cohn and A. Hu, Proc. Natl. Acad. Sci., U.S.A., 1978, 75, 200; G. Lowe and B. S. Sproat, J. Chem. Soc., Chem. Commun., 1978, 565.
- 13 T. Glonek, R. A. Kleps, and T. C. Myers, *Science*, 1974, 185, 352; D. G. Knorre, V. A. Kurbatov, and V. V. Samukov, *FEBS Lett.*, 1976, 70, 105.
- 14 K. W. Stahl, E. Schlimme, and F. Eckstein, *FEBS Lett.*, 1974, 40, 241.
- 15 P. M. J. Burgers and F. Eckstein, Biochemistry, 1979, 18, 450.
- 16 G. Lowe, B. V. L. Potter, B. S. Sproat, and W. E. Hull, J. Chem. Soc., Chem. Commun., 1979, 733; M.-D. Tsai, Biochemistry, 1979, 18, 1468; M.-D. Tsai, S. L. Huang, J. F. Kozlowski, and C. C. Chang, *ibid.*, 1980, 19, 3531.
- 17 B. A. Connolly, F. Eckstein, and H. H. Füldner, J. Biol. Chem., 1982, 257, 3382.
- 18 R. D. Sammons and P. A. Frey, J. Biol. Chem., 1982, 257, 1138.