

L-Threonine-Derived Novel Bifunctional Phosphine–Sulfonamide Catalyst-Promoted Enantioselective Aza-Morita–Baylis–Hillman Reaction

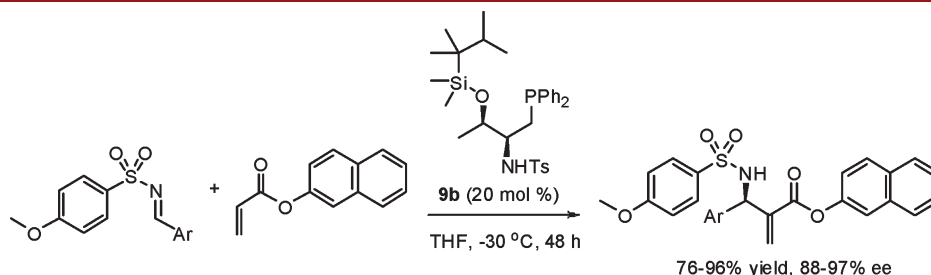
Fangrui Zhong,[†] Youqing Wang,[‡] Xiaoyu Han,[†] Kuo-Wei Huang,^{*,§} and Yixin Lu^{*,†}

Department of Chemistry & Medicinal Chemistry Program, Life Sciences Institute, National University of Singapore, 3 Science Drive 3, Singapore 117543, Provincial Key Laboratory of Natural Medicine and Immuno-Engineering, Henan University, Jinming Campus, Kaifeng, Henan, 475004, P. R. China, and KAUST Catalysis Center & Division of Chemical and Life Science and Engineering, King Abdullah University of Science and Technology, Kingdom of Saudi Arabia

chmlyx@nus.edu.sg; hkw@kaust.edu.sa

Received December 28, 2010

ABSTRACT



A series of novel bifunctional phosphine–sulfonamide organic catalysts were designed and readily prepared from natural amino acids, and they were utilized to promote enantioselective aza-Morita–Baylis–Hillman (MBH) reactions. L-Threonine-derived phosphine–sulfonamide **9b** was found to be the most efficient catalyst, affording the desired aza-MBH adducts in high yields and with excellent enantioselectivities.

Nucleophilic organic catalysts employing trivalent phosphines as Lewis bases are widely used in synthetic organic chemistry.¹ Compared with similarly substituted amines, phosphines are generally less basic and more nucleophilic; thus chiral phosphines often show unique catalytic activities in asymmetric synthesis.² Despite tremendous applications of phosphine-catalyzed reactions in organic synthesis, development of chiral phosphine catalysts is still at its early stage, and design of highly efficient and readily available chiral phosphines remains a huge challenge. Our group has been investigating primary amino acid based

asymmetric organocatalytic synthetic methods in the past few years,^{2p,3} and asymmetric amino catalysis *via* chiral primary amines has been established as a powerful and versatile tool in asymmetric synthesis.⁴

[†] National University of Singapore.

[‡] Henan University.

[§] King Abdullah University of Science and Technology.

(1) For reviews, see: (a) Lu, X.; Zhang, C.; Xu, Z. *Acc. Chem. Res.* **2001**, *34*, 535. (b) Methot, J. L.; Roush, W. R. *Adv. Synth. Catal.* **2004**, *346*, 1035. (c) Ye, L.-W.; Zhou, J.; Tang, Y. *Chem. Soc. Rev.* **2008**, *37*, 1140. (d) Kwong, C. K.-W.; Fu, M. Y.; Lam, C. S.-L.; Toy, P. H. *Synthesis* **2008**, 2307. (e) Cowen, B. J.; Miller, S. J. *Chem. Soc. Rev.* **2009**, *38*, 3102. (f) Marinetti, A.; Voituriez, A. *Synlett* **2010**, 174.

(2) For selected examples, see: (a) Zhu, G.; Chen, Z.; Jiang, Q.; Xiao, D.; Cao, P.; Zhang, X. *J. Am. Chem. Soc.* **1997**, *119*, 3836. (b) Vedejs, E.; Daugulis, O. *J. Am. Chem. Soc.* **1999**, *121*, 5813. (c) Zhu, X.-F.; Lan, J.; Kwon, O. *J. Am. Chem. Soc.* **2003**, *125*, 4716. (d) Wurz, R. P.; Fu, G. C. *J. Am. Chem. Soc.* **2005**, *127*, 12234. (e) Wilson, J. E.; Fu, G. C. *Angew. Chem., Int. Ed.* **2006**, *45*, 1426. (f) Cowen, B. J.; Miller, S. J. *J. Am. Chem. Soc.* **2007**, *129*, 10988. (g) Fang, Y. Q.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2008**, *130*, 5660. (h) Jiang, Y.-Q.; Shi, Y.-L.; Shi, M. *J. Am. Chem. Soc.* **2008**, *130*, 7202. (i) Voituriez, A.; Panossian, A.; Fleury-Brégeot, N.; Retailleau, P.; Marinetti, A. *J. Am. Chem. Soc.* **2008**, *130*, 14030. (j) Smith, S. W.; Fu, G. C. *J. Am. Chem. Soc.* **2009**, *131*, 14231. (k) Smith, S. W.; Fu, G. C. *J. Am. Chem. Soc.* **2009**, *131*, 14231. (l) Chung, Y. K.; Fu, G. C. *Angew. Chem., Int. Ed.* **2009**, *48*, 2225. (m) Sun, J.; C. Fu, G. C. *J. Am. Chem. Soc.* **2010**, *132*, 4568. (n) Sinisi, R.; Sun, J.; Fu, G. C. *Proc. Natl. Acad. Sci. U.S.A.* **2010**, *107*, 20652. (o) Xiao, H.; Chai, Z.; Zheng, C.-W.; Yang, Y.-Q.; Liu, W.; Zhang, J.-K.; Zhao, G. *Angew. Chem., Int. Ed.* **2010**, *49*, 4467. (p) Han, X.; Wang, Y.; Zhong, F.; Lu, Y. *J. Am. Chem. Soc.* **2011**, *133*, 1726.

To further expand amino acid based asymmetric catalysis, we were interested in establishing novel chiral phosphines based on simple amino acid scaffolds. Bifunctional catalytic systems have been widely used in organic chemistry, and the synergistic interactions of two functionalities have been proven to be extremely powerful in asymmetric catalysis. We envisioned that novel bifunctional phosphine catalysts can be easily derived from natural amino acids, and the designing principles are summarized in Figure 1.

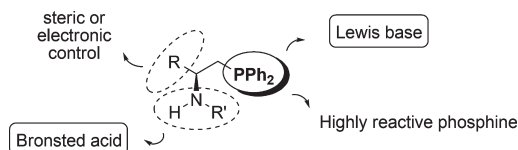


Figure 1. Novel bifunctional phosphine catalysts based on natural primary amino acids.

Simple functional group transformations convert the acid group into a phosphine, and the Brønsted acid site necessary for the bifunctional catalysts can be easily derived from the amine moiety. Proper selection of side chains then provides either steric or electronic tuning to the structures of the catalysts. Moreover, by connecting the phosphorus atom to a primary carbon, we anticipate higher nucleophilicity of the resulting chiral phosphines.

The Morita–Baylis–Hillman (MBH) reaction and its aza counterpart (Aza-MBH reaction) are among the most valuable reactions for the construction of densely functionalized products from simple precursors in a highly atom economic fashion.⁵ Considerable efforts have been devoted to the development of enantioselective versions of these reactions, and most successful examples are usually

based on chiral amine catalysts. Hatakeyama et al. reported highly enantioselective MBH and aza-MBH reactions with hexafluoroisopropyl acrylate (HFIPA) catalyzed by a tertiary amine catalyst β -isocupreidine (β -ICD).⁶ Shi and co-workers employed the same catalyst in the aza-MBH reactions with diverse activated olefins.⁷ Later, the group of Sasai applied BINOL-derived amine to promote the aza-MBH reaction.⁸ Raheem and Jacobsen also reported a thiourea-mediated aza-MBH reaction.⁹ More recently, Zhu et al. disclosed that a modified β -ICD bearing amide functionality (β -ICD amide) could be applied in the aza-MBH reaction in the presence of β -naphthol and excellent enantioselectivity was achieved with both aromatic and aliphatic sulfinyl imines.¹⁰ On the other hand, chiral phosphine-mediated enantioselective aza-MBH reactions are rather limited. In this context, elegant utilization of BINOL-derived phosphines by the groups of Shi, Sasai, and Ito clearly demonstrated the enormous potential of chiral phosphines in asymmetric MBH reactions.¹¹ Our research group recently showed that dipeptide-derived novel phosphines were powerful catalysts for the enantioselective allene–acrylate [3 + 2] cycloadditions;^{2p} herein, we demonstrate that amino acid based bifunctional phosphines could effectively catalyze aza-MBH of acrylates¹² with excellent enantiomeric control.

We began our investigation by preparing a number of L-valine-derived bifunctional chiral phosphines with different hydrogen bond donors (catalysts **4** to **7** in Figure 2). In particular, thioureas¹³ and sulfonamides¹⁴ are known to be remarkably useful in hydrogen-bonding (H-bond) interactions. The aza-MBH reaction between imines and acrylates was chosen as a model reaction, and the results

(3) For our work in this research field, see: (a) Jiang, Z.; Liang, Z.; Wu, X.; Lu, Y. *Chem. Commun.* **2006**, 2801. (b) Cheng, L.; Han, X.; Huang, H.; Wong, M. W.; Lu, Y. *Chem. Commun.* **2007**, 4143. (c) Cheng, L.; Wu, X.; Lu, Y. *Org. Biomol. Chem.* **2007**, *5*, 1018. (d) Wu, X.; Jiang, Z.; Shen, H.-M.; Lu, Y. *Adv. Synth. Catal.* **2007**, *349*, 812. (e) Zhu, Q.; Lu, Y. *Chem. Commun.* **2008**, 6315. (f) Han, X.; Kwiatkowski, J.; Xue, F.; Huang, K.-W.; Lu, Y. *Angew. Chem., Int. Ed.* **2009**, *48*, 7604. (g) Zhu, Q.; Lu, Y. *Chem. Commun.* **2010**, 2235. (h) Jiang, Z.; Yang, H.; Han, X.; Luo, J.; Wong, M. W.; Lu, Y. *Org. Biomol. Chem.* **2010**, *8*, 1368. (i) Jiang, Z.; Lu, Y. *Tetrahedron Lett.* **2010**, *51*, 1884. (j) Zhu, Q.; Lu, Y. *Angew. Chem., Int. Ed.* **2010**, *49*, 7753. (k) Luo, J.; Wang, H.; Han, X.; Xu, L.-W.; Kwiatkowski, J.; Huang, K.-W.; Lu, Y. *Angew. Chem., Int. Ed.* **2011**, *50*, 1861.

(4) For reviews on primary amino acids and primary amines in asymmetric catalysis, see: (a) Xu, L.-W.; Luo, J.; Lu, Y. *Chem. Commun.* **2009**, 1807. (b) Xu, L.-W.; Lu, Y. *Org. Biomol. Chem.* **2008**, *6*, 2047. (c) Peng, F.; Shao, Z. J. *Mol. Catal. A* **2008**, *285*, 1. (d) Chen, Y.-C. *Synlett* **2008**, 1919. (e) Bartoli, G.; Melchiorre, P. *Synlett* **2008**, 1759. For reviews on asymmetric catalysis mediated by synthetic peptides, see: (f) Davie, E. A. C.; Mennen, S. M.; Xu, Y.; Miller, S. J. *Chem. Rev.* **2007**, *107*, 5759. (g) Revell, J. D.; Wennemers, H. *Curr. Opin. Chem. Biol.* **2007**, *11*, 269.

(5) For reviews on the MBH reactions, see: (a) Basavaiah, D.; Reddy, B. S.; Badsara, S. S. *Chem. Rev.* **2010**, *110*, 5447. (b) Langer, P. *Angew. Chem., Int. Ed.* **2000**, *39*, 3049. (c) Basavaiah, D.; Rao, A. J.; Satyanarayana, T. *Chem. Rev.* **2003**, *103*, 811. (d) Masson, G.; Housseman, C.; Zhu, J. *Angew. Chem., Int. Ed.* **2007**, *46*, 4614. (e) Basavaiah, D.; Rao, K. V.; Reddy, R. J. *Chem. Soc. Rev.* **2007**, *36*, 1581. (f) Shi, Y.-L.; Shi, M. *Eur. J. Org. Chem.* **2007**, 2905. (g) Ma, G.-N.; Jiang, J.-J.; Shi, M.; Wei, Y. *Chem. Commun.* **2009**, 5496. (h) Declercq, V.; Martinez, J.; Lamaty, F. *Chem. Rev.* **2009**, *109*, 1. (i) Wei, Y.; Shi, M. *Acc. Chem. Res.* **2010**, *43*, 1005.

(6) (a) Iwabuchi, Y.; Nakatani, M.; Yokoyama, N.; Hatakeyama, S. *J. Am. Chem. Soc.* **1999**, *121*, 10219. (b) Kawahara, S.; Nakano, A.; Esumi, T.; Iwabuchi, Y.; Hatakeyama, S. *Org. Lett.* **2003**, *5*, 3103.

(7) (a) Shi, M.; Xu, Y.-M. *Angew. Chem., Int. Ed.* **2005**, *41*, 4507. (b) Shi, M.; Xu, Y.-M.; Shi, Y.-L. *Chem.—Eur. J.* **2005**, *11*, 1794. (c) Shi, M.; Qi, M.-J.; Liu, X.-G. *Chem. Commun.* **2008**, 6025.

(8) (a) Matsui, K.; Takizawa, S.; Sasai, H. *J. Am. Chem. Soc.* **2005**, *127*, 3680. (b) Matsui, K.; Tanaka, K.; Horii, A.; Takizawa, S.; Sasai, H. *Tetrahedron: Asymmetry* **2006**, *17*, 578.

(9) Raheem, I. T.; Jacobsen, E. N. *Adv. Synth. Catal.* **2005**, *347*, 1701.

(10) (a) Abermil, N.; Masson, G.; Zhu, J. *J. Am. Chem. Soc.* **2008**, *130*, 12596. (b) Abermil, N.; Masson, G.; Zhu, J. *Org. Lett.* **2009**, *11*, 4648.

(11) (a) Shi, M.; Chen, L.-H. *Chem. Commun.* **2003**, 1310. (b) Shi, M.; Chen, L.-H.; Teng, W.-D. *Adv. Synth. Catal.* **2005**, *347*, 1781. (c) Shi, M.; Chen, L.-H.; Li, C.-Q. *J. Am. Chem. Soc.* **2005**, *127*, 3790. (d) Liu, Y.-H.; Chen, L.-H.; Shi, M. *Adv. Synth. Catal.* **2006**, *348*, 973. (e) Matsui, K.; Takizawa, S.; Sasai, H. *Synlett* **2006**, 761. (f) Ito, K.; Nishida, K.; Gotanda, T. *Tetrahedron Lett.* **2007**, *48*, 6147.

(12) For a recent excellent example of aza-MBH reactions with metal catalysis, see: Yukawa, T.; Seelig, B.; Xu, Y.; Morimoto, H.; Matsunaga, S.; Berkessel, A.; Shibasaki, M. *J. Am. Chem. Soc.* **2010**, *132*, 11988.

(13) (a) Schreiner, P. R. *Chem. Soc. Rev.* **2003**, *32*, 289. (b) Taylor, M. S.; Jacobsen, E. N. *Angew. Chem., Int. Ed.* **2006**, *45*, 1520. (c) Doyle, A. G.; Jacobsen, E. N. *Chem. Rev.* **2007**, *107*, 5713. (d) Yu, X.; Wang, W. *Chem.—Asian J.* **2008**, *3*, 516. (e) Connon, S. J. *Chem. Commun.* **2008**, 2499.

(14) (a) Luo, J.; Xu, L.-W.; Hay, R. A. S.; Lu, Y. *Org. Lett.* **2009**, *11*, 437. (b) Wang, W.; Wang, J.; Li, H.; Liao, L. *Tetrahedron Lett.* **2004**, *45*, 7235. (c) Wang, W.; Wang, J.; Li, H. *Tetrahedron Lett.* **2004**, *45*, 7243. (d) Wang, W.; Wang, J.; Li, H. *Org. Lett.* **2004**, *6*, 2817. (e) Wang, W.; Wang, J.; Li, H. *Angew. Chem.* **2005**, *117*, 1393. *Angew. Chem., Int. Ed.* **2005**, *44*, 1369. (f) Wang, J.; Li, H.; Lou, B.; Zu, L.; Guo, H.; Wang, W. *Chem.—Eur. J.* **2006**, *12*, 4321. (g) Berkessel, A.; Koch, B.; Lex, J. *Adv. Synth. Catal.* **2004**, *346*, 1141. (h) Dahlin, N.; Bøgevig, A.; Adolffson, H. *Adv. Synth. Catal.* **2004**, *346*, 1101.

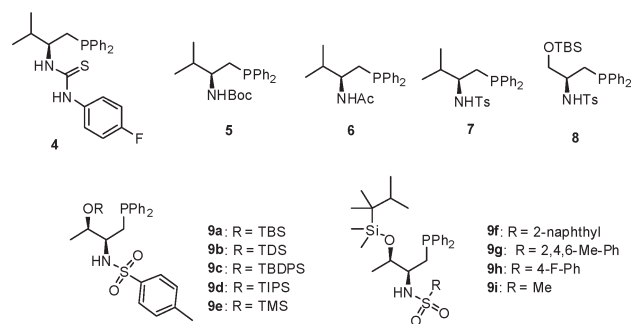


Figure 2. Structures of bifunctional phosphine catalysts.

are summarized in Table 1. Phosphine-thiourea **4** gave poor results (entry 1). While catalyst **5** with a Boc group was completely ineffective, phosphine-amide **6** was able to catalyze the reaction, although in very low ee (entries 2 and 3). Phosphine-sulfonamide catalyst **7** was found to display certain reactivity and selectivity (entry 4). We next varied sulfonamide protection on the imine nitrogen and also employed different acrylates (entries 5 to 7). To our delight, by combining *N*-(*p*-methoxybenzenesulfonyl)imine and

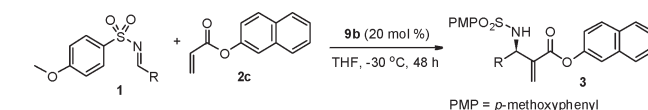
Table 1. Catalyst Screening and Optimization of Reaction Conditions

entry ^a	R ¹ /R ²	cat.	yield (%) ^b	ee (%) ^c
1	Me/Me	4	28	–13
2	Me/Me	5	–	–
3	Me/Me	6	49	23
4	Me/Me	7	42	40
5	Me/Ph	7	40	55
6	Me/2-naphthyl	7	63	57
7	OMe/2-naphthyl	7	90	57
8	OMe/2-naphthyl	8	90	64
9	OMe/2-naphthyl	9a	88	71
10	OMe/2-naphthyl	9b	90	74
11	OMe/2-naphthyl	9c	93	73
12	OMe/2-naphthyl	9d	94	73
13	OMe/2-naphthyl	9e	91	64
14	OMe/2-naphthyl	9f	87	74
15	OMe/2-naphthyl	9g	90	72
16	OMe/2-naphthyl	9h	43	67
17	OMe/2-naphthyl	9i	88	60
18 ^d	OMe/2-naphthyl	9b	95	89
19 ^e	OMe/2-naphthyl	9b	89	91

^a Reactions were carried out with **1** (0.05 mmol), **2** (0.1 mmol for **2a** and 0.06 mmol for **2b**, **2c**), and catalyst (0.005 mmol) in dry THF (0.1 mL at rt or 0.2 mL at –20 °C or 0.25 mL at –30 °C) under N₂. ^b Isolated yield. ^c Determined by HPLC analysis on a chiral stationary phase. ^d Reaction was performed with 20 mol % catalyst at –20 °C for 48 h. ^e Reaction was performed with 20 mol % catalyst at –30 °C for 48 h.

2-naphthyl acrylate, phosphine-sulfonamide **7** led to the formation of an aza-MBH adduct in excellent yield and with moderate enantioselectivity (entry 7). We reasoned that the isopropyl side chain from valine may not provide enough steric control in the asymmetric induction, and with our success in threonine-based catalytic systems,^{3b–d,g} we decided to synthesize serine-derived catalyst **8** and a number of threonine-based phosphine-sulfonamides **9a** to **9i**, in which various silyloxy groups¹⁵ were introduced. The threonine core proved to be superior to that of serine (entry 9 versus entry 8). Similar enantioselectivities were observed with bulky silyloxy groups, and the t-hexyldimethylsilyl (TDS) group was chosen as it gave marginally better results (entries 9–13). Exploration of different sulfonamides in the catalyst structures failed to improve the results (entries 14–17). The enantiomeric excess could be further improved by lowering the reaction temperature (entry 18). When the reaction was performed at –30 °C for 2 days, the desired aza-MBH adduct was obtained in 89% yield and with 91% ee (entry 19).

Table 2. Enantioselective Aza-MBH Reactions of Various Imines **1** and Acrylate **2c** Catalyzed by **9b**



entry ^a	product (R)	yield (%) ^b	ee (%) ^c
1	3a (Ph)	89	91
2	3b (4-Me-Ph)	91	90
3	3c (4-Et-Ph)	90	91
4	3d (4-F-Ph)	96	92
5	3e (4-Br-Ph)	95	92
6	3f (4-Cl-Ph)	84	93
7	3g (4-CF ₃ -Ph)	93	90
8	3h (4-NO ₂ -Ph)	84	92
9	3i (3-Me-Ph)	76	91
10	3j (3-Cl-Ph)	85	90
11	3k (2-CF ₃ -Ph)	95	97
12	3l (2-Cl-Ph)	95	90
13	3m (2-Br-Ph)	93	90
14	3n (2-F-Ph)	95	91
15	3o (2-furyl)	93	95
16	3p (2-thiophenyl)	95	96
17	3q (2-naphthyl)	76	88

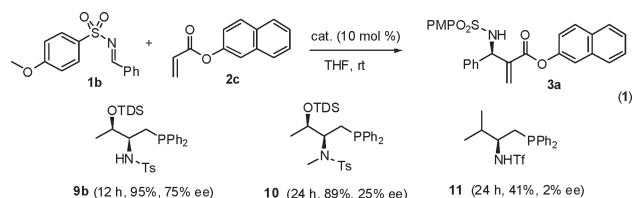
^a Reactions were performed with imine **1** (0.05 mmol), 2-naphthyl acrylate **2c** (0.06 mmol) and catalyst **9b** (0.01 mmol) in dry THF (0.25 mL) under N₂. ^b Isolated yield. ^c Determined by HPLC analysis on a chiral stationary phase.

With the optimized reaction conditions in hand, the scope of the aza-MBH reaction was evaluated (Table 2). The reaction is applicable to a wide range of aromatic imines; excellent chemical yields and very high enantioselectivities were attainable within 2 days. Notably, the *ortho*-substituted aromatic imines, which are well-known to be difficult substrates for the aza-MBH reaction, were found to be suitable,

(15) Xu, L.-W.; Li, L.; Shi, Z.-H. *Adv. Synth. Catal.* **2010**, *352*, 243.

and the products were obtained in nearly quantitative yields and with up to 97% ee (entries 11–14). These results represent by far the best enantioselectivities attainable for the *ortho*-substituted substrates in the aza-MBH reaction. In addition, imines with heterocyclic rings were also applicable; 95% and 96% ee were observed (entries 15–16). A less satisfactory result was obtained for a cyclohexyl imine.¹⁶

The Brønsted acidic sulfonamide and nucleophilic phosphine in our bifunctional catalysts are indispensable and work cooperatively in promoting enantioselective aza-MBH reactions. When *N*-methylated sulfonamide **10** was employed as the catalyst, the reaction became slower and enantioselectivity was dramatically decreased [eq 1]. This result suggested that the H-bond donor moiety in the catalyst contributes significantly to the reaction and is crucial for the asymmetric induction. Surprisingly, phosphine **11** containing *N*-trifluoromethanesulfonamide was found to be a very poor catalyst, giving the product in low yield and with virtually no enantioselectivity, which may be attributed to the overstabilization of the enolate intermediate by highly acidic N–H in the sulfonamide or formation of a protonated form of the enolate intermediate.



Despite their enormous applications, the mechanism of the MBH reactions is not entirely clear. Based on recent elegant mechanistic studies¹⁷ and our experimental observations, we propose the mechanism of the **9b**-promoted aza-MBH reaction as depicted in Figure 3. The reversible conjugate addition of phosphine to acrylate generates intermediate **A**. Since enolate **A** is more basic than the imine substrate, and thus is a better H-bond acceptor, it forms a H-bond with the Brønsted acid moiety of **9b**.¹⁸ We

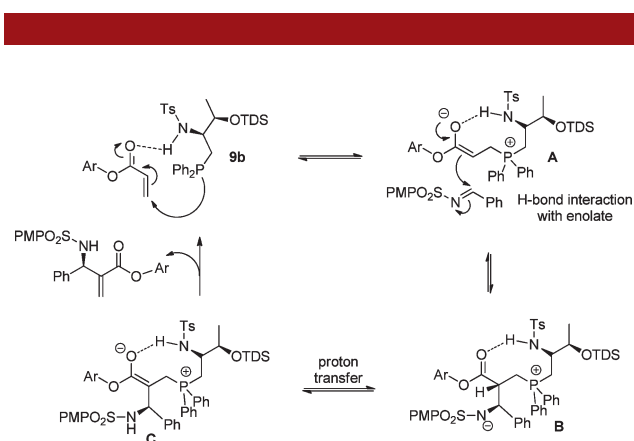


Figure 3. Proposed reaction mechanism.

believe such a H-bond, together with the bulky OTDS group, favors the formation of a structurally well-defined phosphonium enolate intermediate **A**, providing a confined spatial arrangement for the incoming imine. Had the OTDS be replaced by a less hindered OTMS group, the less rigid intermediate **A** is anticipated to provide less stereocontrol. This is consistent with experimental observations; the MBH adduct was obtained in 64% ee when OTMS-derived catalyst **9e** was employed, in contrast to a 74% ee attainable with OTDS-derived **9b** under otherwise identical reaction conditions (entry 13 versus 10, Table 1). A subsequent Mannich reaction of **A** with imine yields zwitterionic intermediate **B**. The proton transfer from the α -carbon atom then takes place to yield enolate **C**, which undergoes β -elimination to afford the MBH adduct and regenerate **9b** at the same time.

In conclusion, we have introduced a novel class of bifunctional phosphine catalysts. In particular, we have derived a series of phosphine–sulfonamide bifunctional catalysts from natural amino acids. The effectiveness of our catalysts has been demonstrated in highly enantioselective aza-MBH reactions. DFT calculations are ongoing toward to a deeper understanding of the reaction mechanism and will be reported in due course.

Acknowledgment. We are grateful for the generous financial support from the National University of Singapore and the Ministry of Education (MOE) of Singapore (R-143-000-362-112) to Y.L. and from King Abdullah University of Science and Technology to K.-W.H.

Supporting Information Available. Representative experimental procedures, HPLC chromatogram, and NMR spectra of compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

(16) The reaction was run with cyclohexyl tosylimine at $-10\text{ }^{\circ}\text{C}$ for 72 h, and the MBH product was obtained in 74% yield and with 72% ee.

(17) (a) Aggarwal, V. K.; Fulford, S. Y.; Lloyd-Jones, G. C. *Angew. Chem., Int. Ed.* **2005**, *44*, 1706. (b) Price, K. E.; Broadwater, S. J.; Jung, H. M.; McQuade, D. T. *Org. Lett.* **2005**, *7*, 147. (c) Price, K. E.; Broadwater, S. J.; Walker, B. J.; McQuade, D. T. *J. Org. Chem.* **2005**, *70*, 3980. (d) Buskens, P.; Klankermayer, J.; Leitner, W. *J. Am. Chem. Soc.* **2005**, *127*, 16762. (e) Kraft, M. E.; Haxell, T. F. N.; Seibert, K. A.; Abboud, K. A. *J. Am. Chem. Soc.* **2006**, *128*, 4174. (f) Amarante, G. W.; Milagre, H. M. S.; Vaz, B. G.; Vilacha Ferreira, B. R.; Eberlin, M. N.; Coelho, F. *J. Org. Chem.* **2009**, *74*, 3031. (g) Amarante, G. W.; Benassi, M.; Milagre, H. M. S.; Braga, A. A. C.; Maseras, F.; Eberlin, M. N.; Coelho, F. *Chem.—Eur. J.* **2009**, *15*, 12460. (h) Santos, L. S.; Pavam, C. H.; Almeida, W. P.; Coelho, F.; Eberlin, M. N. *Angew. Chem., Int. Ed.* **2004**, *43*, 4330.

(18) Such H-bonding interaction was supported by ^1H NMR analysis. The chemical shift corresponding to sulfonamide NH disappeared upon mixing **9b** with acrylate **2c**, and such resonance remained unchanged when **9b** and **1b** were mixed.