

Article

Subscriber access provided by UNIV OF WESTERN ONTARIO

Copper-Catalyzed Double Thiolation to Access Sulfur-bridged Imidazopyridines with Isothiocyanate

Lu-Lu Tian, Shuai Lu, Zhe-Hua Zhang, En-Ling Huang, Hua-Ting Yan, Xinju Zhu, Xin-Qi Hao, and Mao-Ping Song

J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.9b00186 • Publication Date (Web): 20 Mar 2019 Downloaded from http://pubs.acs.org on March 20, 2019

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

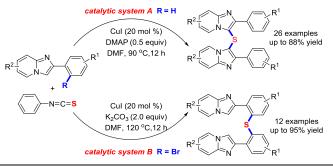
Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.

Copper-Catalyzed Double Thiolation to Access Sulfur-bridged Imidazopyridines with Isothiocyanate

Lu-Lu Tian, Shuai Lu, Zhe-Hua Zhang, En-Ling Huang, Hua-Ting Yan, Xinju Zhu,* Xin-Qi Hao, and Mao-Ping Song*

College of Chemistry and Molecular Engineering, Zhengzhou University, No. 100 of Science Road, Zhengzhou, Henan 450001, P. R. China

Supporting Information



ABSTRACT: A copper(I)-catalyzed sulfur-bridged dimerization of imidazopyridines has been developed using isothiocyanate as the sulfur source. This method enables a switchable synthesis of bis(imidazo[1,2-*a*]pyridin-3-yl)sulfanes or bis(2-(imidazo[1,2-*a*]pyridin-2-yl)phenyl)sulfanes in the presence of DMAP or K_2CO_3 when different imidazopyridines were employed. Under the optimized conditions, a variety of sulfur-bridged imidazopyridines were obtained in good yields. Moreover, thiourea was proved to be the key intermediate under catalytic system A.

INTRODUCTION

The construction of C-S bonds is of fundamental research interest due to their wide applications in pharmaceuticals, organic syntheses, and materials science.¹ Accordingly, a wide range of protocols have been developed to introduce sulfur moiety onto target molecules.² Traditionally, the formation of C-S bonds usually involves cross-coupling between organic halides and thiols.³ Recently, C-H functionalization has also emerged as a powerful strategy to construct C-S bonds, which is more atom economic and environmentally benign. In this context, various thiol surrogates, such as disulfide, thiols, sulfinic acids, arylsulfonyl chlorides, sulfonyl hydrazides, sodium sulfinates, and others, have been successfully utilized to access thioether derivatives.⁴

On the other hand, imidazo[1,2-*a*]pyridines have attracted considerable attention due to their significance in natural products, medicinal chemistry, and materials science.⁵ Meanwhile, imidazo[1,2-*a*]pyridines have been utilized as abnormal *N*-heterocyclic carbene and pincer-type ligand precursors, which exhibited application in catalysis after coordinated with transition metals.⁶ Consequently, great achievement has been made for the syntheses⁷ and functionalizations⁸ of imidazo[1,2-*a*]pyridine scaffold. Until now, direct C₃ arylation, alkylation, carbonylation, amination, alkoxyla

Scheme 1. Functionalizations of imidazopyridines

Previous work

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

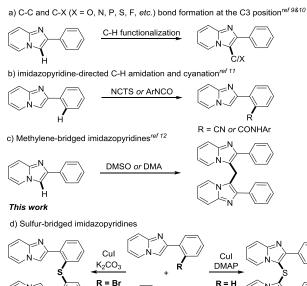
53

54

55 56

57 58 59

60



tion, halogenation, phosphonation, and thiolation have been well-developed to construct C-C⁹ and C-X¹⁰ (X = O, N, P, S, F, *etc.*) bonds (Scheme 1a). Meanwhile, imidazo[1,2-*a*]pyridine-directed C-H amidation and cyanation have also been developed (Scheme 1b).¹¹ Recently, DMA and DMSO have been utilized as the methylene sources to access bis(imidazo[1,2-*a*]pyridin-3-yl)methanes (Scheme 1c).¹² However, the methodology to access sulfurbridged imidazopyridines is rather limited.¹³ Considering the importance of sulfur-containing and bis(heteroaryl) compounds,^{1,14} it is higher desirable to develop an environmentally benign strategy to prepare sulfur-bridged imidazopyridines.

=C=S

As commercially available organic synthons, isothiocyanates have been utilized as thiocyanating agents¹⁵ and are versatile synthetic intermediates for various heterocycles.^{16,17} Meanwhile, the reaction involving cleavage of the C=S bond of isothiocyanates accompanied with formation of byproduct, such as CuS, Ag₂S, or H₂S, is reported.¹⁸⁻²⁰ As the continuation of our previous work in imidazopyridines,^{9c,10f,h,11b,21} we herein reported a unique approach to access sulfur-bridged imidazopyridines utilizing CuI as the catalyst.²¹ To the best of our knowledge, the employment of isothiocyanates as the sulfur source has not been reported.

RESULTS AND DISCUSSION

Our investigation was started with 2-phenylimidazo[1,2- α]pyridine 1a (0.2 mmol), isothiocyanatobenzene 2 (0.3 mmol), CuI (20 mol%), and 1,10-phen (3 equiv) in DMF (2.5 mL). To our delight, the desired sulfur-bridged product 3a was isolated in 29% yield at 100 °C for 12 h (Table 1, entry 1). Subsequently, various bases were screened, and DMAP was proved to be the best choice to afford 3a in 64% yield (Table 1, entry 3). Next, the effect of different

copper salts, such as CuCl, CuBr, and CuBr₂, were investigated, which all led to decreased yields (Table 1, entries 5-7). When DMF was replaced by other solvents, no improved result was obtained (Table 1, entries 8-10). In an attempt to improve reaction efficiency, reducing the usage amount of DMAP to 0.5 equivalent was beneficial, which provided 3a in 71% yields (Table 1, entry 11). To further improve the outcome, the temperature range was modulated. Product 3a was isolated in 88% yield at 90 °C (Table 1, entry 12), whereas decreased yield was observed at lower temperature (Table 1, entry 13). Moreover, when 10 mol% of copper salt was utilized, product 3a was isolated in 26% yield (Table 1, entry 14). And 3a could not be detected without CuI, indicating the necessity of copper salt (Table 1, entry 15). Finally, when the reaction was carried out under argon or oxygen atmosphere, decreased yields were observed. The structure of 3a was further confirmed by X-ray diffraction (See the sup info).

Table 1. Optimization of reaction conditions^a

+ PhNCS $\frac{Cu \text{ salt } (20 \text{ mol } \%)}{base, \text{ solvent, } 100 ^{\circ}\text{C}, 12 \text{ h}}$							
Entry	1a catalyst	base	solvent	yield (%) ^b			
Lifting	cuturyst		solvent	yicia (70)			
1	CuI	1,10-phen	DMF	29			
2	CuI	2,2'-bipyridine	DMF	59			
3	CuI	DMAP	DMF	64			
4	CuI	K ₂ CO ₃	DMF	-			
5	CuCl	DMAP	DMF	20			
6	CuBr	DMAP	DMF	36			
7	CuBr ₂	DMAP	DMF	62			
8	CuI	DMAP	DMA	44			
9	CuI	DMAP	dioxane	55			
10	CuI	DMAP	DMSO	20			
11	CuI	DMAP (o.5 equiv)	DMF	71, 71 ^c			
12 ^{c,d}	CuI	DMAP (o.5 equiv)	DMF	88			
13 ^{c,e}	CuI	DMAP (o.5 equiv)	DMF	75			
14 ^{cf}	CuI	DMAP (o.5 equiv)	DMF	26			
15 ^c	-	DMAP (o.5 equiv)	DMF	NR			

^aReaction Conditions: **1a** (0.2 mmol), **2** (0.3 mmol), Cu salt (20 mol %), base (3.0 equiv), solvent (2.5 mL), under air, T (°C), 12 h. ^bIsolated yields. ^cDMF (0.5 mL). ^d90 °C. ^e80 °C. ^fCuI (10 mol %).

With the optimized conditions in hand (Table 1, entry 12), the scope and generality of double C-H thiolation reaction was evaluated (Scheme 2). Initially, the substrate scope of 2-phenylimidazo[1,2- α]pyridines was investigated. It was found that Various substitutions at the *para*, *meta*, and *ortho* positions were well tolerated, affording the corresponding products **3a-m** in **37-88%** yields. However, substrate bearing strong electron-withdrawing groups (NO₂ and CF₃) at the *para* position failed to yield the tar-

2

3

4

5

6

7

8

9

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34 35

36

37 38

39

40

41

42 43

44 45 46

47

48 49

50 51

52

53

54 55 56

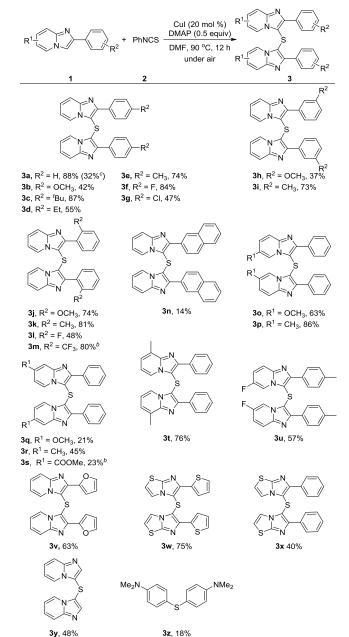
57 58 59

60

get products (not shown). When 2-naphthyl substituted substrates were employed, only 14% yield was achieved for **3n**. Subsequently, imidazopyridines bearing OCH₃, CH₃, F, or COOMe on the various positions of pyridine rings were also examined, which provided products 30-t in 21-86% yields. Also, disubstituted imidazopyridine were employed to afford products 3u in 57% yield. When 2-furanyl and 2-thienyl substituted imidazopyridines were utilized, **3v** and **3w** were isolated in 63% and 75% yields, respectively. Moreover, phenylimidazo[2,1-b] thiazole was also compatible to provide the sulfur-bridged product 3x 10 in 40% yield. Finally, when C2,C3-unsubstituted imidaz-11 opyridine was treated with PhNCS, only C3 sulfur-bridged 12 product 3y was obtained in 48% yield. Interestingly, when 13 electron-rich aromatics N,N-dimethylaniline was utilized, 14 the corresponding product 3z was isolated in 18% yield.²² 15 Finally, a gram-scale experiment was conducted to evalu-16 ate the practical utility of the current protocol as a syn-17 thetic tool. Unfortunately, product 3a was isolated in only 18 32% yield. 19

2-(2-bromophenyl)imidazo[1,2-Unexpected, when *a*]pyridine **4a** was utilized as the coupling partners, bis(2-(imidazo[1,2-a]pyridin-2-yl)phenyl)sulfane 5a was isolated in 19% yield (Table 2, entry 1). In the process to improve the yield of 5a, various base and solvents were examined (see the sup info), which revealed that K_2CO_2 (2 equiv) and DMF (2.0 mL) are the best choice (Table 2, entry 6). When the temperature was increased to 120 °C, product 5a was obtained in 89% yield (Table 2, entry 7). Further elevating temperature to 130 °C led to decreased yield (Table 2, entry 8). Moreover, when the Cu catalyst loading was decreased to 10 mol%, product 5a was isolated in 76% yield (Table 2, entry 9). The struc ture of 5a was further confirmed by X-ray diffraction (see the sup info).

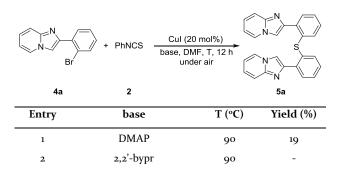
Scheme 2. Substrate scope of imidazopyridines^a



3z. 18%

^aReaction Conditions: 1 (0.2 mmol), 2 (0.3 mmol), CuI (20 mol %), DMAP (0.5 equiv), DMF (0.5 mL), 90 °C, 12 h, under air. ^b24 h. ^c1a (1.2 g, 6 mmol), 2 (9 mmol).

Table 2. Optimization of reaction conditions^a



ACS Paragon Plus Environment

Page 4	4 of 12
--------	---------

3	K ₂ CO ₃	90	24
4	Cs_2CO_3	90	17
5	K₂CO ₃ (2 equiv)	90	34
6 ^c	K₂CO ₃ (2 equiv)	90	40
7 ^c	K ₂ CO ₃ (2 equiv)	120	89
8 ^c	K₂CO₃(2 equiv)	130	83
$9^{c,d}$	K ₂ CO ₃ (2 equiv)	120	76

2

3

4

5

6

7 8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28 29

30

31

32

33 34

35 36

37

38

39

40 41

42 43

44

45

46

47

48

49

50

51

52

53

54

55 56

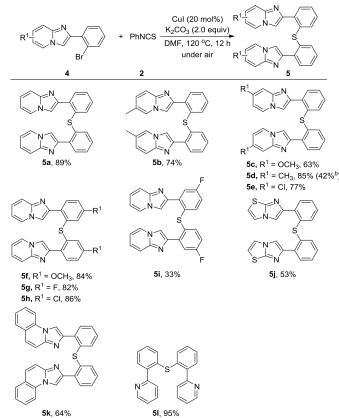
57 58 59

60

^aReaction Conditions: 1a (0.2 mmol), 2 (0.3 mmol), CuI (20 mol%), base (0.5 equiv), DMF (0.5 mL), under air, T (°C), 12 h. ^bIsolated yields. ^cDMF (2.0 mL). ^dCuI (10 mol %).

Next, the substrate scope of 2-(2bromophenyl)imidazo[1,2-a]pyridines 4 was investigated (Scheme 3). At first, imidazopyridines $4\mathbf{b}$ - \mathbf{e} bearing OCH₃, CH₃, or Cl substituent on the C6 and C7 positions gave products **5b-e** in 63-85% yields, while substrates with strong electron-withdrawing groups such as C6substituted CF3 and C7-substituted NO2 remained unreactive (not shown). Next, substitutions at the C2-benzene ring of imidazopyridines were examined, which gave para-substituted products 5f-h in 82-86% yields. However, meta-substituted product 5i was isolated in 33% yield. Also, imidazo-containing heterocycles pro

Scheme 3. Substrate scope of 2-(2-bromoaryl)imidazo[1,2-*a*]pyridines



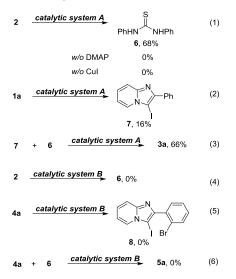
^aReaction Conditions: 4 (0.2 mmol), 2 (0.3 mmol), CuI (20

mol %), K₂CO₃ (2.0 equiv), DMF (2.0 mL), 120 °C, 12 h, under air. $^{c}4d$ (1.0 g, 4.6 mmol), 2 (0.90 g, 6.9 mmol).

ceeded smoothly to deliver product **5j** and **5k** in 53% and 64% yields. Finally, the catalytic system could be applied to 2-(2-bromophenyl)pyridine, which provide 5l in 95% yield. Other simple aryl bromides without a directing group, such as bromobenzene and 2-bromonaphthalene, failed to yield the desired products. We also tried to synthesize **5d** for a gram-scale production, which unfortunately gave the desired product in 42% yield.

To explore the reaction mechanism, a set of control experiments were carried out (Scheme 4). A symmetric thiourea 6 could be produced from isothiocyanatobenzene 2 under catalytic system A. Without DMAP or CuI, the thiourea intermediate 6 could not be detected (Scheme 4, eq 1). Meanwhile, iodo-substituted imidazopyridine 7 was isolated in 16% under catalytic system A (Scheme 4, eq 2). Moreover, coupling between 7 and 6 yielded the desired product 3a in 66% yield (Scheme 4, eq 3). These results indicate that 6 and 7 are probably the key intermediate in the catalytic system A. By contrast, compound 6 and 8 could not be detected under catalytic system B (Scheme 4, eqs 4 and 5). Also, reaction between 4a and 6 failed to give product **5a** under catalytic system B (Scheme 4, eq 6). These results indicate that formation of 5a does not proceed through intermediate 6 and 8.

Scheme 4. Mechanistic studies



On the basis of above discussion and related reports,^{18-20,23} a plausible mechanism was proposed for catalytic system A (Scheme 5). At first, Cu(I) underwent oxidative addition with iodo-substituted imidazopyridine 7 to yield Cu(III) intermediate I. Subsequently, the abstraction of a nitrogen proton from thiourea followed by nucleophilic substitution in the presence of DMAP gave Cu(III) intermediate II,²⁴ which underwent reductive elimination to provide CuI and intermediate III. Next, abstraction of a second proton generated diphenylmethanediimine and

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

21

22

23

24

25

26

27

28

29

30

31

32

33

34

35

36

37

38

39

40

41

42

43

44

45

46

47

48

49

50

51

52

53

54

55

56

57 58 59

60

intermediate **IV**,^{18c,19} which was confirmed by MS analysis (see the supporting information). Finally, nucleophilic attack of intermediate **I** by intermediate **IV** gave I⁻ and intermediate **V**, which underwent reductive elimination to deliver the final product **3a** and Cu(I) species.

On the other hand, catalytic system B also underwent similar reaction pathway (Scheme 6). Oxidative addition of Cu(I) with **4a** gave N1-chelated Cu(III) intermediate **VI**,^{II} which reacted with **VII** species to yield intermediate **VIII**, which underwent reductive elimination to generated intermediate **IX** and CuX. After release of PhNCO and CO_2 , intermediate **X** would be generated. Reaction between intermediates **X** and **VI** would afford Cu(III) intermediate **XI**, which underwent reductive elimination to deliver product **5a**, accompanied with the regeneration of Cu(I).

Scheme 5. Proposed catalytic cycles for catalytic system A

Scheme 6. Proposed catalytic cycles for catalytic system B

CONCLUSION

In conclusion, a novel CuI-catalyzed double thiolation of imidazopyridines using isothiocyanate as the sulfur source has been disclosed. This methodology could regioselective afford sulfur-bridged imidazopyridines via two proposed catalytic systems. Meanwhile, this protocol is featured with broad substrate scope, good functional group compatibility, and operational convenience. The discovery of isothiocyanate as sulfur source may promote the development of C-H thiolation.

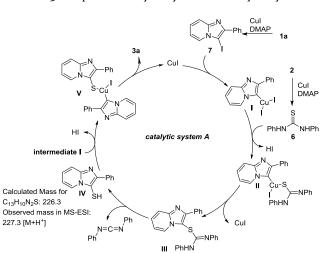
EXPERIMENTAL SECTION

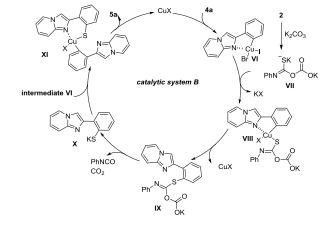
General Experimental Details. Unless otherwise mentioned, all materials were commercially obtained and used without further purification. All procedures were performed under the air atmosphere. Imidazo[1,2*a*]pyridines 1,²⁵ 2-(2-bromophenyl)imidazo[1,2-*a*]pyridines 4,26 and 2-(2-bromophenyl)pyridine27 are known compounds and synthesized according to previously described methods. 1,3-diphenylthiourea 6²⁸ and 3-iodo-2phenylimidazo[1,2-*a*]pyridine 7^{29} are known compounds. ¹H NMR, ¹³C NMR, and ¹⁹F NMR spectra were recorded at 400 MHz or 600 MHz, 100 MHz or 151 MHz, and 565 MHz respectively on a Bruker DPX instrument using Me₄Si as an internal standard. Chemical shift multiplicities are represented as follows: (s = singlet, d = doublet, t = triple, m = multiplet, dd = double doublet, td = triple doublet). Melting points were measured on a WC-1 instrument and uncorrected. New compounds for HRMS were tested on a Waters Q-Tof Micro MS/MS System ESI spectrometer.

Procedure for bis(2-phenylimidazo[1,2-*a***]pyridin-3yl)sulfane (3a) synthesis.** To a 10 mL sealed tube was added 2-phenylimidazo[1,2-*a*]pyridine **1a** (38.8 mg, 0.2 mmol), isothiocyanatobenzene **2** (40.5 mg, 0.3 mmol), Cul (7.6 mg, 20 mol %), and DMAP (12.2 mg, 0.5 equiv) in DMF (0.5 mL) under air. The reaction was heated at 90 °C for 12 h, and then cooled down to room temperature. Ethyl acetate (20 mL) was added and the mixture was extracted with water (2 × 10 mL). The organic layer was dried over anhydrous sodium sulfate and concentrated. The residue was purified by preparative TLC on silica gel plates using petroleum ether/EtOAc = 2/1 as the eluent to give the corresponding product **3a**.

Procedure for bis(2-(imidazo[1,2-*a*]pyridin-2yl)phenyl)sulfane (5a) synthesis. To a 10 mL sealed tube was added 2-(2-bromophenyl)imidazo[1,2-*a*]pyridine 4a (54.4 mg, 0.2 mmol), isothiocyanatobenzene 2 (40.5 mg, 0.3 mmol), CuI (7.6 mg, 20 mol %), and K₂CO₃ (55.2 mg, 2.0 equiv) in DMF (2.0 mL) under air. The reaction was heated at 120 °C for 12 h, and then cooled down to room temperature. Ethyl acetate (20 mL) was added and the mixture was extracted with water (2 × 10 mL). The organic layer was dried over anhydrous sodium sulfate and concentrated. The residue was purified by preparative TLC on silica gel plates using petroleum ether/EtOAc = 1/1 as the eluent to give the corresponding product 5a.

Procedure for 1,3-diphenylthiourea (6) synthesis. To a 10 mL sealed tube was added isothiocyanatobenzene **2** (40.5 mg, 0.3 mmol), Cul (7.6 mg, 20 mol %), and DMAP (12.2 mg, 0.5 equiv) in DMF (0.5 mL) under air. The reaction was heated at 90 °C for 12 h, and then cooled down to





room temperature. Ethyl acetate (20 mL) was added and the mixture was extracted with water (2 × 10 mL). The organic layer was dried over anhydrous sodium sulfate and concentrated. The residue was purified by preparative TLC on silica gel plates using petroleum ether/EtOAc = 2/1 as the eluent to give the corresponding product **6**.

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

56

57 58 59

60

Procedure for 3-iodo-2-phenylimidazo[1,2-a]pyridine (7) synthesis. To a 10 mL sealed tube was added 2phenylimidazo[1,2-*a*]pyridine 1a (38.8 mg, 0.2 mmol), Cul (7.6 mg, 20 mol %), and DMAP (0.5 equiv) in DMF (0.5 mL) under air. The reaction was heated at 90 °C for 12 h, and then cooled down to room temperature. Ethyl acetate (20 mL) was added and the mixture was extracted with water (2 × 10 mL). The organic layer was dried over anhydrous sodium sulfate and concentrated. The residue was purified by preparative TLC on silica gel plates using petroleum ether/EtOAc = 2/1 as the eluent to give the corresponding product 7.

17 Bis(2-phenylimidazo[1,2-a]pyridin-3-yl)sulfane (3a) puri-18 fied by preparative TLC on silica gel with petroleum 19 ether/EtOAc (2:1) as an eluent ($R_f = 0.34$); Light yellow 20 solid (36.8 mg, 88% yield; 0.40 g, 32% yield for scale-up 21 synthesis); mp 251 – 252 °C. ¹H NMR (400 MHz, CDCl₃) δ 22 8.12 - 8.10 (m, 4H), 7.63 - 7.60 (m, 4H), 7.58 - 7.55 (m, 23 4H), 7.50 (d, J = 8.9 Hz, 2H), 7.11 - 7.07 (m, 2H), 6.34 (td, J 24 = 6.9, 1.0 Hz, 2H). ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 150.9, 25 146.6, 133.8, 129.6, 128.9, 128.6, 126.3, 125.4, 117.4, 112.6, 107.5. 26 HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{26}H_{18}N_4S^+$: 27 419.1325, Found: 419.1324.

28 Bis(2-(4-methoxyphenyl)imidazo[1,2-a]pyridin-3-yl)sulfane 29 (3b). purified by preparative TLC on silica gel with petro-30 leum ether/EtOAc (3:1) as an eluent ($R_f = 0.23$); Brown 31 solid (20.0 mg, 42% yield); mp 224 - 225 °C. 1H NMR (400 32 MHz, CDCl₃) δ 8.08 (d, J = 9.0 Hz, 4H), 7.62 (d, J = 7.0 Hz, 33 2H), 7.48 (d, J = 9.0 Hz, 2H), 7.16 - 7.11 (m, 4H), 7.11 - 7.07 34 (m, 2H), 6.40 (td, J = 6.8, 0.9 Hz, 2H), 3.96 (s, 6H). ¹³C{¹H} 35 NMR (101 MHz, CDCl₃) δ 160.2, 150.6, 146.5, 130.8, 126.3, 36 125.4, 117.2, 114.0, 112.5, 106.5, 55.5. HRMS (positive ESI): [M 37 $+ H^{+}$ Calcd. For C₂₈H₂₂N₄O₂S⁺: 479.1536, Found: 479.1542. 38 Bis(2-(4-(tert-butyl)phenyl)imidazo[1,2-a]pyridin-3-

yl)sulfane (3c). purified by preparative TLC on silica gel 39 with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.55$); 40 Light yellow solid (46.0 mg, 87% yield); mp 267 - 268 °C. 41 ¹H NMR (600 MHz, CDCl₃) δ 8.03 (d, I = 7.6 Hz, 4H), 7.63 42 (d, J = 7.8 Hz, 4H), 7.54 (d, J = 6.5 Hz, 2H), 7.48 (d, J = 8.843 Hz, 2H), 7.07 (t, J = 7.8 Hz, 2H), 6.27 (t, J = 6.5 Hz, 2H), 44 1.46 (s, 18H). ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 152.1, 150.9, 45 146.5, 130.9, 129.3, 126.2, 125.5, 125.4, 117.3, 112.3, 107.7, 34.9, 46 31.5. HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{34}H_{34}N_4S^+$: 47 531.2577, Found: 531.2582. 48

49 Bis(2-(4-ethylphenyl)imidazo[1,2-a]pyridin-3-yl)sulfane

(3d). purified by preparative TLC on silica gel with petro-
leum ether/EtOAc (1:1) as an eluent ($R_f = 0.25$); Light yel-
low solid (26.0 mg, 55% yield); mp 209 – 210 °C. 'H NMR
(400 MHz, CDCl₃) δ 8.03 (d, J = 8.3 Hz, 4H), 7.58 (d, J =
6.9 Hz, 2H), 7.48 (d, J = 8.8 Hz, 2H), 7.44 (d, J = 8.0 Hz,
4H), 7.10 – 7.06 (m, 2H), 6.32 (td, J = 6.8, 1.1 Hz, 2H), 2.82

(q, *J* = 7.7 Hz, 4H), 1.37 (t, *J* = 7.6 Hz, 6H). ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 151.0, 146.5, 145.2, 131.2, 129.5, 128.1, 126.2, 125.46, 117.3, 112.4, 107.3, 28.9, 15.8. HRMS (positive ESI): [M + H]⁺ Calcd. For C₃₀H₂₆N₄S⁺: 475.1951, Found: 475.1953. *Bis*(2-(*p*-tolyl)*imidazo*[*1*,2-*a*]*pyridin*-3-*y*]*sulfane* (**3e**). purified by preparative TLC on silica gel with petroleum ether/EtOAc (2:1) as an eluent (R_f = 0.22); Light yellow solid (33.1 mg, 74% yield); mp 263 – 264 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.02 (d, *J* = 8.2 Hz, 4H), 7.60 (d, *J* = 6.9 Hz, 2H), 7.48 (t, *J* = 8.9 Hz, 2H), 7.41 (d, *J* = 8.0 Hz, 4H), 7.10 – 7.06 (m, 2H), 6.35 (td, *J* = 6.8, 0.9 Hz, 2H), 2.52 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 150.9, 146.5, 138.8, 130.9, 129.4, 129.3, 126.2, 125.4, 117.3, 112.5, 107.0, 21.5. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₈H₂₂N₄S⁺: 447.1638, Found: 447.1642.

Bis(2-(4-fluorophenyl)imidazo[1,2-a]pyridin-3-yl)sulfane

(*3f*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.27$); Light yellow solid (38 mg, 84% yield); mp 254 – 255 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.13 – 8.08 (m, 4H), 7.58 – 7.59 (m, 2H), 7.51 (d, J = 8.9 Hz, 2H), 7.34 – 7.28 (m, 4H), 7.16 – 7.12 (m, 2H), 6.44 (td, J = 7.8, 1.0 Hz , 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 163.3 ($J_{C-F} = 247.6$ Hz), 149.9, 146.5, 131.3 ($J_{C-F} = 247.6$ Hz), 129.9 ($J_{C-F} = 3.3$ Hz), 126.6, 125.1, 117.5, 115.7 ($J_{C-F} = 21.4$ Hz), 112.8, 107.0. ¹⁹F NMR (376 MHz, CDCl₃) δ -112.14. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₆H₁₆F₂N₄S⁺: 455.1137, Found: 455.1138.

Bis(2-(4-chlorophenyl)imidazo[1,2-a]pyridin-3-yl)sulfane.

(**3***g*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.28$); Light yellow solid (23.1 mg, 47% yield); mp 296 – 297 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.10 – 8.06 (m, 4H), 7.62 – 7.57 (m, 6H), 7.51 (d, *J* = 9.0 Hz, 2H), 7.17 – 7.13 (m, 2H), 6.47 (td, *J* = 6.8, 1.0 Hz, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 149.8, 146.6, 135.1, 132.3, 130.7, 129.0, 126.7, 125.1, 117.6, 113.0, 107.1. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₆H₁₆Cl₂N₄S: 487.0545, Found: 487.0548.

Bis(2-(3-*methoxyphenyl*)*imidazo*[1,2-*a*]*pyridin*-3-*y*]*sulfane*. (**3***h*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.25$); Light yellow solid (17.9 mg, 37% yield); mp 153 - 154 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.74 - 7.72 (m, 2H), 7.64 - 7.63 (m, 4H), 7.55 - 7.49 (m, 4H), 7.14 - 7.09 (m, 4H), 6.40 (td, J = 6.9, 1.0 Hz, 2H), 3.95 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 159.7, 150.7, 146.5, 135.08, 129.7, 126.5, 125.5, 122.0, 117.4, 115.3, 114.3, 112.7, 107.5, 55.5. HRMS (positive ESI): [M + H] + Calcd. For C₂₈H₂₂N₄O₂S⁺: 479.1536, Found: 479.1538. Bis(2-(m-tolyl)imidazo[1,2-a]pyridin-3-yl)sulfane (3i). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.30$); Light yellow solid (32.7 mg, 73% yield); mp 172 - 173 °C. 1H NMR (400 MHz, CDCl₃) δ 7.93 (d, J = 7.6 Hz, 2H), 7.89 (s, 2H), 7.58 (d, J = 6.8 Hz, 2H), 7.50 (t, J = 7.5 Hz, 2H), 7.36 (d, J = 7.6Hz, 2H), 7.12 - 7.07 (m, 2H), 6.36 (td, J = 6.9, 1.0 Hz, 2H), 2.52 (s, 6H). ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 151.0, 146.5, 138.3, 133.7, 130.1, 129.7, 128.5, 126.7, 126.3, 125.5, 117.3, 112.5,

56

57 58 59

60

107.4, 21.6. HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{28}H_{22}N_4S^+$: 447.1638, Found: 447.1641.

2 Bis(2-(2-methoxyphenyl)imidazo[1,2-a]pyridin-3-yl)sulfane 3 (3j). purified by preparative TLC on silica gel with petro-4 leum ether/EtOAc (3:1) as an eluent ($R_f = 0.53$); Light yel-5 low solid (35.3 mg, 74% yield); mp 195 - 196 °C. 1H NMR 6 (400 MHz, CDCl₃) δ 7.56 – 7.49 (m, 8H), 7.19 – 7.08 (m, 7 6H), 6.45 – 6.41 (m, 2H), 3.84 (s, 6H). ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 157.5, 148.4, 146.4, 132.6, 130.6, 125.7, 124.9, 8 123.1, 120.6, 117.4, 112.2, 111.3, 109.9, 55.6. HRMS (positive 9 ESI): $[M + H]^+$ Calcd. For $C_{28}H_{22}N_4O_2S^+$: 479.1536, Found: 10

479.1541. 11 Bis(2-(o-tolyl)imidazo[1,2-a]pyridin-3-yl)sulfane (3k). puri-12 fied by preparative TLC on silica gel with petroleum 13 ether/EtOAc (1:1) as an eluent ($R_f = 0.40$); Light yellow 14 solid (36 mg, 81% yield); mp 202 - 203 °C. 1H NMR (600 15 MHz, CDCl₃) δ 7.50 (d, J = 8.6 Hz, 2H), 7.47 – 7.44 (m, 4H), 16 7.40 - 7.35 (m, 6H), 7.16 (t, J = 7.5 Hz, 2H), 6.45 (t, J = 6.717 Hz, 2H), 2.25 (s, 6H). ${}^{13}C{}^{1}H{}$ NMR (101 MHz, CDCl₃) δ 18 151.9, 146.2, 137.9, 133.4, 131.8, 130.5, 129.0, 126.1, 125.5, 124.9, 19 117.2, 112.5, 109.5, 20.3. HRMS (positive ESI): [M + H]+ 20 Calcd. For C₂₈H₂₂N₄S⁺: 447.1638, Found: 447.1642. 21

Bis(2-(2-fluorophenyl)imidazo[1,2-a]pyridin-3-yl)sulfane

22 (31). purified by preparative TLC on silica gel with petro-23 leum ether/EtOAc (2:1) as an eluent ($R_f = 0.33$); Light yel-24 low solid (22.0 mg, 48% yield); mp 243 - 244 °C. ¹H NMR 25 (400 MHz, CDCl₃) δ 7.70 (td, J = 7.4, 1.7 Hz, 2H), 7.60 -26 7.51 (m, 6H), 7.40 - 7.33 (m, 4H), 7.17 - 7.13 (m, 2H), 6.47 27 (td, J = 6.9, 1.0 Hz, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 28 160.2 (J_{C-F} = 250.1 Hz), 146.8, 146.2, 132.7, (J_{C-F} = 2.4 Hz), 29 131.0 (J_{C-F} = 8.2 Hz), 126.4, 124.6, 124.3, (J_{C-F} = 3.6 Hz), 122.0 30 $(J_{C-F} = 13.9 \text{ Hz}), 117.7, 116.3 (J_{C-F} = 22.0 \text{ Hz}), 112.8, 109.6.$ ¹⁹F 31 NMR (376 MHz, CDCl₃) δ -113.04. HRMS (positive ESI): [M 32 + H]+Calcd. For $C_{26}H_{16}F_2N_4S^+$: 455.1137, Found: 455.1142.

33 Bis(2-(2-(trifluoromethyl)phenyl)imidazo[1,2-a]pyridin-3-34 *yl)sulfane* (**3***m*). purified by preparative TLC on silica gel 35 with petroleum ether/EtOAc (2:1) as an eluent ($R_f = 0.25$); 36 Yellow solid (36.4 mg, 80% yield); mp 232 - 233 °C. 1H 37 NMR (400 MHz, CDCl₃) δ 7.79 (d, J = 7.7 Hz, 2H), 7.62 (t, J= 7.5 Hz, 2H), 7.57 - 7.53 (m, 4H), 7.43 (d, J = 6.9 Hz, 2H), 38 7.32 (d, J = 7.5 Hz, 2H), 7.20 - 7.16 (m, 2H), 6.46 (td, J = 39 6.8, 0.9 Hz, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 149.2, 40 145.6, 133.2, 132.9, 131.2,130.4 ($J_{C-F} = 30.0 \text{ Hz}$), 129.0, 41 $126.6(J_{C-F} = 5.2 \text{ Hz}), 126.0, 124.1, 123.6 (J_{C-F} = 274.7 \text{ Hz}),$ 42 117.6, 112.7, 110.9. ¹⁹F NMR (376 MHz, CDCl₃) δ -58.70. 43 HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{28}H_{16}F_6N_4S^+$: 44 555.1073, Found: 555.1076.

45 555.1073, Found: 555.1076. Bis(2-(naphthalen-2-yl)imidazo[1,2-a]pyridin-3-yl)sulfane

46 (3n). purified by preparative TLC on silica gel with petro-47 leum ether/EtOAc (1:1) as an eluent ($R_f = 0.30$); yellow 48 solid (7.2 mg, 14% yield); mp 264 - 265 °C. 1H NMR (400 49 MHz, CDCl₃) δ 8.64 (s, 2H), 8.30 (dd, J = 8.5, 1.7 Hz, 2H), 50 8.10 (d, J = 8.5 Hz, 2H), 8.06 - 8.03 (m, 2H), 8.01 - 7.98 (m, 51 2 H), 7.62 – 7.57 (m, 6H), 7.51 (d, J = 8.9 Hz, 2H), 7.08 – 52 7.03 (m, 2H), 6.16 (td, J = 6.9, 1.0 Hz, 2H). ¹³C{¹H} NMR (101 53 MHz, CDCl₃) δ 150.8, 146.7, 133.5, 133.3, 131.2, 128.9, 128.7, 54 128.3, 127.9, 126.9, 126.8, 126.6, 126.5, 125.3, 117.4, 112.7, 107.3. 55

HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{34}H_{22}N_4S^+$: 519.1638, Found: 519.1643.

Bis(6-methoxy-2-phenylimidazo[1,2-a]pyridin-3-yl)sulfane (**30**) purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent (R_f = 0.42); Brown solid (30.0 mg, 63% yield); mp 264 – 265 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.50 (dd, J = 8.5, 1.3 Hz, 4H), 7.59 (t, J = 7.5, 4H), 7.47 – 7.41 (m, 6H), 6.90 (dd, J = 9.7, 2.5 Hz, 2H), 2.88 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 149.9, 149.1, 143.5, 134.0, 128.9, 128.8, 128.8, 122.1, 117.6, 106.6, 106.1, 54.8. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₈H₂₂N₄O₂S⁺: 479.1536, Found: 479.1537.

Bis(6-methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)sulfane (**3***p*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.33$); Light yellow solid (38.1 mg, 86% yield); mp 285 – 286 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.18 – 8.16 (m, 4H), 7.65 – 7.61 (m, 4H), 7.57 – 7.53 (m, 2H), 7.41 – 7.37 (m, 4H), 6.93 (dd, J =9.1, 1.6 Hz, 2H), 1.74 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 150.3, 145.5, 134.1, 129.5, 129.4, 128.8, 123.7, 122.5, 116.5, 106.8, 17.8. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₈H₂₂N₄S+: 447.1638, Found: 447.1640.

Bis(7-methoxy-2-phenylimidazo[1,2-a]pyridin-3-yl)sulfane (3**q**). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent (R_f = 0.39); Light yellow solid (27.3 mg, 57% yield); mp 272 - 273 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.52 - 8.49 (m, 4H), 7.60 - 7.57 (m, 4H), 7.47 - 7.41 (m, 6H), 6.89 (dd, J = 9.5, 2.4 Hz, 2H), 2.88 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 149.9, 149.1, 143.5, 134.0, 128.9, 128.8, 128.8, 122.1, 117.6, 106.6, 106.1, 54.8. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₈H₂₂N₄O₂S⁺: 479.1536, Found: 479.1537.

Bis(7-*methyl*-2-*phenylimidazo*[1,2-*a*]*pyridin*-3-*y*]*sulfane*

(**3***r*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.43$); Yellow solid (20.0 mg, 45% yield); mp 197 – 198 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.11-8.09 (m, 4H), 7.63 – 7.59 (m, 4H), 7.56 – 7.52 (m, 2H), 7.40 (d, *J* = 7.0 Hz, 2H), 7.24 (s, 2H), 6.18 (dd, *J* = 7.0, 1.6 Hz, 2H), 2.26 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 150.7, 146.9, 137.7, 133.9, 129.5, 128.8, 128.6, 124.6, 115.9, 115.2, 106.8, 21.3. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₈H₂₂N₄S⁺: 447.1638, Found: 447.1641.

Dimethyl 3,3'-thiobis(2-phenylimidazo[1,2-a]pyridine-7carboxylate) (**3s**). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.60$); Light yellow solid (12.0 mg, 23% yield); mp 208 – 209 °C. 'H NMR (400 MHz, CDCl₃) δ 8.20 – 8.21 (m, 2H), 8.09 (dd, J = 8.2, 1.7 Hz, 4H), 7.67 – 7.62 (m, 6H), 7.54 (dd, J = 7.2,0.7 Hz, 2H), 6.94 (dd, J = 7.1, 1.6 Hz, 2H), 3.90 (s, 6H). '³C{¹H} NMR (101 MHz, CDCl₃) δ 165.2, 153.0, 145.6, 133.1, 129.6, 129.5, 128.9, 127.9, 124.9, 119.9, 111.9, 109.0, 52.7 HRMS (positive ESI): [M + H]⁺ Calcd. For C₃₀H₂₂N₄O₄S⁺: 535.1435, Found: 535.1440.

Bis(8-methyl-2-phenylimidazo[1,2-a]pyridin-3-yl)sulfane

(*3t*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.50$); Brown solid (34 mg, 76% yield); mp 216 – 217 °C. ¹H NMR (400

MHz, CDCl₃) δ 8.09 – 8.07 (m, 4H), 7.62 – 7.58 (m, 4H), 7.55 – 7.51 (m, 2H), 7.41 (d, J = 6.7 Hz, 2H), 6.86 (d, J = 6.9 2 Hz, 2H), 6.25 (t, J = 6.9 Hz, 2H), 2.51 (s, 6H). ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 150.5, 146.8, 134.2, 129.8, 128.7, 128.5, 127.3, 125.1, 123.2, 112.5, 107.9, 16.7. HRMS (positive ESI): [M + H] + Calcd. For C₂₈H₂₂N₄S⁺: 447.1638, Found: 447.1642. 6 Bis(6-fluoro-2-(p-tolyl)imidazo[1,2-a]pyridin-3-yl)sulfane (3*u*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.51$); Light yel-8 low solid (27.3 mg, 57% yield); mp 272 - 273 °C. ¹H NMR 9 $(600 \text{ MHz}, \text{CDCl}_3) \delta 7.91 \text{ (d, } J = 7.4 \text{ Hz}, 4\text{H}), 7.46 - 7.39 \text{ (m,}$ 10 8H), 7.00 (t, J = 7.9 Hz, 2H), 2.52 (s, 6H). ¹³C{¹H} NMR (101 11 MHz, CDCl₃) δ 153.1 (*J*_{C-F} = 238.9 Hz), 152.4 (*J*_{C-F} = 2.4 Hz), 12 144.1, 139.4, 130.4, 129.4 (J_{C-F} = 21.9 Hz), 118.2 (J_{C-F} = 25.2 Hz), 13 117.6 ($J_{C-F} = 8.4 \text{ Hz}$), 112.8, 112.4, 108.9, 21.4.¹⁹F NMR (376) 14 MHz, CDCl₃) δ -138.14. HRMS (positive ESI): [M + H]⁺ 15 Calcd. For C₂₈H₂₀F₂N₄S⁺: 483.1450, Found: 483.1451. 16 Bis(2-(furan-2-yl)imidazo[1,2-a]pyridin-3-yl)sulfane (3v). 17 purified by preparative TLC on silica gel with petroleum 18 ether/EtOAc (3:1) as an eluent ($R_f = 0.34$); Light Brown 19 solid (25.2 mg, 63% yield); mp 235 - 236 °C. 'H NMR (600 20 MHz, CDCl₃) δ 8.18 (d, J = 6.9 Hz, 2H), 7.72 (s, 2H), 7.54 (d, 21 *J* = 9.1 Hz, 2H), 7.38 (d, *J* = 2.9 Hz, 2H), 7.16 (t, *J* = 7.6 Hz, 22 2H), 6.69 (s, 2H), 6.64 (t, J = 6.9 Hz, 2H). ¹³C{¹H} NMR (101 23 MHz, CDCl₃) δ 148.4, 147.0, 143.4, 141.8, 126.7, 125.0, 117.5, 24 113.3, 111.8, 110.7, 104.6. HRMS (positive ESI): [M + H]+ 25 Calcd. For C₂₂H₁₄N₄O₂S⁺: 399.0910, Found: 399.0913. 26 Bis(2-(thiophen-2-yl)imidazo[1,2-a]pyridin-3-yl)sulfane 27 (**3***w*). purified by preparative TLC on silica gel with petro-28 leum ether/EtOAc (1:1) as an eluent ($R_f = 0.38$); Light yel-29 low solid (32.1 mg, 75% yield); mp 217 - 218 °C. 1H NMR 30 $(600 \text{ MHz}, \text{CDCl}_3) \delta 8.15 \text{ (d, } J = 1.5 \text{ Hz}, 2\text{H}), 7.93 \text{ (d, } J = 6.7$ 31 Hz, 2H), 7.55 (d, J = 4.8 Hz, 2H), 7.52 (d, J = 8.9 Hz, 2H), 32 7.29 (s, 2H), 7.13 (t, J = 7.8 Hz, 2H), 6.54 (t, J = 6.6 Hz, 2H). 33 ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 146.8, 145.2, 136.5, 127.9, 34 127.4, 126.7, 125.2, 117.3, 113.2, 104.5. HRMS (positive ESI): 35 $[M + H]^+$ Calcd. For $C_{22}H_{14}N_4S_3^+$: 431.0453, Found: 431.0458. 36 Bis(6-phenylimidazo[2,1-b]thiazol-5-yl)sulfane (3x). puri-37 fied by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.55$); Light yellow 38 solid (17.0 mg, 40% yield); mp 228 - 229 °C. 1H NMR (400 39 MHz, CDCl₃) δ 8.08 – 8.05 (m, 4H), 7.56 – 7.53 (m, 4H), 40 7.49 - 7.45 (m, 2H), 6.51 (d, J = 4.5 Hz, 2H), 6.49 (dd, J =41 4.5 Hz, 2H). ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 151.4, 151.3, 42 133.6, 128.6, 128.6, 128.5, 118.4, 112.6, 109.5. HRMS (positive 43 ESI): $[M + H]^+$ Calcd. For $C_{22}H_{14}N_4S_3^+$: 431.0453, Found: 44 431.0456. 45 *Bis(imidazo[1,2-a]pyridin-3-yl)sulfane* (**3***y*). purified by

1

3

4

5

7

55 56

57 58 59

60

46 preparative TLC on silica gel with petroleum ether/EtOAc 47 (1:3) as an eluent ($R_f = 0.24$); Light yellow solid (12.6 mg, 48 48% yield); mp 258 – 259 °C. ¹H NMR (600 MHz, CDCl₃) δ 49 8.58 (d, J = 6.9 Hz, 2H), 7.99 (s, 2H), 7.62 (d, J = 9.0 Hz, 50 2H), 7.27 (t, J = 7.8 Hz, 2H), 6.97 (t, J = 6.6 Hz, 2H). ¹³C{¹H} 51 NMR (151 MHz, CDCl₃) δ 147.6, 140.4, 126.0, 123.9, 118.2, 52 113.4, 109.7. HRMS (positive ESI): [M + H]⁺ Calcd. For 53 C₁₄H₁₀N₄S⁺: 267.0699, Found: 267.0700. 54

Bis(2-(imidazo[1,2-a]pyridin-2-yl)phenyl)sulfane (5a). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent ($R_f = 0.21$); Brown solid (37.3 mg, 89% yield); mp 165 - 166 °C. ¹H NMR (400 MHz, $CDCl_3$) δ 8.18 (dd, J = 7.9, 1.3 Hz, 2H), 8.06 (s, 2H), 8.01 (dd, J = 5.8, 0.9 Hz, 2H, 7.59 (dd, J = 9.2, 0.6 Hz, 2H), 7.36 (td, J = 7.8, 1.4 Hz, 2H, 7.30 - 7.27 (m, 2H), 7.22 - 7.18 (m, 2H), 7.13 – 7.09 (m, 2H), 6.69 (td, J = 6.8, 1.1 Hz, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 144.6, 143.0, 135.3, 133.2, 132.7, 130.8, 128.5, 127.6, 125.9, 124.7, 117.4, 112.3, 112.3. HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{26}H_{18}N_4S^+$: 419.1325, Found: 419.1330.

Bis(2-(6-methylimidazo[1,2-a]pyridin-2-yl)phenyl)sulfane (5b). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:2) as an eluent ($R_f = 0.20$); Light yellow solid (33.2 mg, 74% yield); mp 72 - 73 °C. 1H NMR (400 MHz, CDCl₃) δ 8.8 (dd, J = 7.8, 1.3 Hz, 2H), 7.98 (s, 2 H), 7.79 (s, 2 H), 7.50 (d, J = 9.2 Hz, 2 H), 7.38 - 7.34 (m, 2H), 7.29 - 7.27 (m, 2H), 7.22 - 7,17 (m, 2H), 6.98 (dd, J = 9.2, 1.5 Hz, 2H), 2.25 (s, 3H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 143.6, 142.6, 135.5, 133.0, 132.7, 130.7, 128.3, 127.9, 127.6, 123.6, 121.8, 116.6, 112.2, 18.1. HRMS (positive ESI): [M + H]+ Calcd. For $C_{28}H_{22}N_4S^+$: 447.1638, Found: 447.1643.

Bis(2-(7-methoxyimidazo[1,2-a]pyridin-2-yl)phenyl)sulfane. (5c). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:2) as an eluent ($R_f = 0.21$); Light yellow solid (30.0 mg, 63% yield); mp 85 - 86 °C. ¹H NMR (600 MHz, CDCl₃) δ 8.18 (d, J = 7.9 Hz, 2H), 7.93 (s, 2H), 7.85 (d, J = 7.5 Hz, 1H), 7.35 (t, J = 7.5 Hz, 2H), 7.27 (d, J = 7.5 Hz, 1H), 7.20 - 7.18 (m, 2H), 6.88 (s, 2H), 6.45 (dd, J = 7.4, 2.2 Hz, 2H), 3.84 (s, 6H). ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 158.0, 146.0, 142.4, 135.3, 132.9, 132.7, 130.5, 128.2, 127.6, 126.3, 111.3, 107.5, 94.4, 55.5. HRMS (positive ESI): [M $+ H^{+}$ Calcd. For C₂₈H₂₂N₄O₂S⁺: 479.1536, Found: 479.1541. Bis(2-(7-methylimidazo[1,2-a]pyridin-2-yl)phenyl)sulfane (5*d*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:2) as an eluent ($R_f = 0.20$); Light yellow solid (38.0 mg, 75% yield; 0.42 g, 42% yield for scaleup synthesis); mp 76 – 77 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.17 (dd, J = 7.9, 1.4 Hz, 2H), 7.99 (s, 2H), 7.91 (d, J = 6.9, 2H), 7.37 - 7.33 (m, 4H), 7.29 - 7.26 (m, 2H), 7.21 - 7.17 (m, 2H), 6.55 (dd, J = 6.9, 1.5 Hz, 2H), 2.37 (s, 6H).¹³C{¹H} NMR (101 MHz, CDCl₃) δ 145.0, 142.6, 135.6, 135.5, 133.1, 132.7, 130.7, 128.3, 127.6, 125.1, 115.7, 114.9, 111.8, 21.4. HRMS (positive ESI): $[M + H]^+$ Calcd. For C₂₈H₂₂N₄S⁺: 447.1638, Found: 447.1644.

Bis(2-(7-chloroimidazo[1,2-a]pyridin-2-yl)phenyl)sulfane (5e). purified by preparative TLC on silica gel with petroleum ether/EtOAc (3:1) as an eluent ($R_f = 0.30$); Brown solid (37.2 mg, 77% yield); mp 203 - 204 °C. 'H NMR (600 MHz, CDCl₃) δ 8.12 (d, J = 7.8 Hz, 2H), 8.10 (s, 2H), 8.00 (s, 2H), 7.56 (d, J = 9.5 Hz, 2H), 7.37 (t, J = 7.5 Hz, 2H), 7.29 (d, J = 7.9 Hz, 2H), 7.23 (t, J = 7.5 Hz, 2H), 7.12 (d, J = 9.5)Hz, 2H). ${}^{13}C{}^{1}H$ NMR (151 MHz, CDCl₃) δ 144.0, 142.9, 134.9, 133.2, 132.9, 130.8, 128.8, 127.8, 126.2, 123.6, 120.5, 117.8, 112.7. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₆H₁₆Cl₂N₄S⁺: 487.0545, Found: 487.0550.

8

bis(2-(imidazo[1,2-a]pyridin-2-yl)-5-

1 methoxyphenyl)sulfane.(5f) purified by preparative TLC 2 on silica gel with petroleum ether/EtOAc (1:1) as an eluent 3 $(R_f = 0.21)$; Light yellow solid (40.0 mg, 84% yield); mp 196 4 - 197 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.13 (d, J = 8.9, Hz, 5 $_{2H}$, $_{8.02}$ – $_{8.00}$ (m, $_{4H}$), $_{7.59}$ (d, $_{J}$ = $_{9.1}$ Hz, $_{2H}$), $_{7.14}$ – 6 7.09 (m, 2H), 6.92 (dt, J = 8.6, 2.6 Hz, 2H), 6.85 (d, J = 2.6 7 Hz, 2H), 6.69 (td, J = 7.5, 0.7 Hz, 2H), 3.71 (s, 6H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 159.5, 144.5, 142.9, 133.9, 131.9, 8 128.0, 125.8, 124.6, 117.6, 117.2, 113.7, 112.1, 111.6, 55.3. HRMS 9 (positive ESI): $[M + H]^+$ Calcd. For $C_{28}H_{22}N_4O_2S^+$: 479.1536, 10 Found: 479.1548. 11

12 bis(5-fluoro-2-(imidazo[1,2-a]pyridin-2-

yl)phenyl)sulfane.(**5***q*) purified by preparative TLC on sili-13 ca gel with petroleum ether/EtOAc (1:1) as an eluent (R_f = 14 0.20); Light yellow solid (37.0 mg, 82% yield); mp 139 -15 140 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.15 (dd, J = 8.4, 6.1 16 Hz, 2H), 8.05 (d, J = 6.58 Hz, 2H), 8.00 (s, 2H), 7.60 (d, J = 17 9.3 Hz, 2H), 7.16 (ddd, J = 7.9, 6.7, 1.0 Hz, 2H), 7.00 (td, J = 18 9.1, 2.6 Hz, 2H), 7.00 (dd, J = 9.1, 2.6 Hz, 2H), 6.76 - 6.63 19 (m, 2H). ${}^{13}C{}^{1}H$ NMR (101 MHz, CDCl₃) δ 162.3 (J_{C-F} = 20 247.8 Hz), 144.6, 142.1, 134.5 (J_{C-F} = 7.3 Hz), 132.5 (J_{C-F} = 8.7 21 Hz), 131.6 ($J_{C-F} = 3.1 \text{ Hz}$), 125.9, 124.9, 119.1 ($J_{C-F} = 23.8 \text{ Hz}$), 22 117.4, 115.2 (J_{C-F} = 21.5 Hz), 112.5, 111.9. ¹⁹F NMR (376 MHz, 23 CDCl₃) δ -112.73. HRMS (positive ESI): [M + H]⁺ Calcd. For 24 $C_{26}H_{16}F_2N_4S^+$: 455.1137, Found: 455.1147. 25

bis(5-chloro-2-(imidazo[1,2-a]pyridin-2-

26 *yl)phenyl)sulfane.*(**5***h*) purified by preparative TLC on sili-27 ca gel with petroleum ether/EtOAc (1:1) as an eluent (R_f = 28 0.25); Light yellow solid (42.0 mg, 86% yield); mp 213 - 214 29 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.13 (t, J = 8.4 Hz, 2H), 30 8.05 - 8.04 (m, 4H), 7.60 (t, J = 9.1 Hz, 2H), 7.35 (dd, J = 31 8.4, 2.1 Hz, 2H), 7.25 (d, J = 2.1, 2H), 7.18 – 7.14 (m, 2H), 32 6.74 (td, J = 6.8, 0.9 Hz, 2H). ¹³C{¹H} NMR (101 MHz, 33 $CDCl_3$) δ 144.7, 141.9, 134.2, 134.0, 133.9, 132.0, 128.2, 125.9, 34 125.1, 117.5, 112.5, 112.3. HRMS (positive ESI): [M + H]⁺ 35 Calcd. For C₂₆H₁₆Cl₂N₄S⁺: 487.0545, Found: 487.0551.

36 bis(4-fluoro-2-(imidazo[1,2-a]pyridin-2-

56

57 58 59

60

37 *yl)phenyl)sulfane.*(**5***i*) purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent (R_f = 38 0.31); Light yellow solid (15.0 mg, 33% yield); mp 202 - 203 39 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.11 (s, 2H), 8.05 (d, J = 40 6.7 Hz, 2H), 7.96 (dd, J = 10.2, 2.9 Hz, 2H), 7.61 (d, J = 9.3, 41 2H), 7.27 - 7.22 (m, 2H), 7.20 - 7.15 (m, 2H), 6.93 (td, J = 42 7.8, 2,9 Hz, 2H), 6.75 (td, J = 6.7, 0.8 Hz, 2H). ¹³C{¹H} NMR 43 (101 MHz, CDCl₃) δ 162.3 (*J*_{C-F} = 249.2 Hz), 144.6, 141.9, 44 137.4 ($J_{C-F} = 8.6 \text{ Hz}$), 134.5 ($J_{C-F} = 7.7 \text{ Hz}$), 128.0 ($J_{C-F} = 3.0 \text{ Hz}$) 45 Hz), 126.0, 125.2, 117.5, 117.4(*J*_{C-F} = 24.0 Hz), 115.7 (*J*_{C-F} = 22.0 46 Hz), 112.6, 112.6. ¹⁹F NMR (376 MHz, CDCl₃) δ -114.07. 47 HRMS (positive ESI): $[M + H]^+$ Calcd. For $C_{26}H_{16}F_2N_4S^+$: 48 455.1137, Found: 455.1148. 49

Bis(2-(imidazo[2,1-b]thiazol-6-yl)phenyl)sulfane (5j). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:2) as an eluent ($R_f = 0.25$); Light yellow solid (23.0 mg, 53% yield); mp 230 – 231 °C. 'H NMR (600 MHz, CDCl₃) δ 8.10 (dd, J = 7.8, 1.0 Hz, 2H), 7.97 (s, 2H), 7.37 (d, J = 4.3 Hz, 2H), 7.35 – 7.33 (m, 2H), 7.25 – 7.24 (m, 2H), 7.19 – 7.16 (m, 2H), 6.79 (d, J = 4.5 Hz, 2H). ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 149.2, 144.9, 135.5, 132.6, 132.2, 130.1, 128.0, 127.6, 118.7, 112.5, 112.4. HRMS (positive ESI): [M + H]⁺ Calcd. For C22H₁₄N₄S₃⁺: 431.0453, Found: 431.0454.

Bis(2-(imidazo[1,2-b]quinolin-2-yl)phenyl)sulfane (**5**k). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:2) as an eluent ($R_f = 0.48$); Brown solid (33.2 mg, 64% yield); mp 209 – 210 °C. 'H NMR (600 MHz, CDCl3) δ 8.57 (s, 2H), 8.27 (d, J = 7.0 Hz, 2H), 7.74 (d, J =7.9 Hz, 2H), 7.70 (d, J = 7.5 Hz, 2H), 7.54 (d, J = 8.9 Hz, 2H), 7.49 (t, J = 7.2 Hz, 2H), 7.44 (d, J = 9.4 Hz, 2H), 7.39 – 7.35 (m, 6H), 7.24 (d, J = 7.2 Hz, 2H). '¹³C{'H} NMR (101 MHz, CDCl₃) δ 143.0, 142.1, 135.2, 133.0, 132.8, 132.6, 130.6, 128.9, 128.8, 128.3, 127.8, 126.4, 124.6, 123.3, 116.9, 115.4, 111.2. HRMS (positive ESI): [M + H]⁺ Calcd. For C₃₄H₂₂N₄S⁺: 519.1638, Found: 519.1643.

Bis(*2*-(*pyridin-2-yl*)*phenyl*)*sulfane.* (*5l*). purified by preparative TLC on silica gel with petroleum ether/EtOAc (1:1) as an eluent (R_f = 0.43); Light yellow solid (32.4 mg, 95% yield); mp 77 – 78 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.58 – 8.57 (m, 2H), 7.54 (td, *J* = 7.7, 1.8 Hz, 2H), 7.45 (dd, *J* = 7.5, 1.7 Hz, 2H), 7.40 (d, *J* = 7.8 Hz, 2H), 7.29 – 7.25 (m, 2H), 7.24 – 7.17 (m, 4H), 7.13 – 7.09 (m, 2H). ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 158.0, 149.0, 142.0, 135.7, 135.1, 132.9, 130.4, 129.0, 127.1, 124.1, 121.9. HRMS (positive ESI): [M + H]⁺ Calcd. For C₂₂H₁₆N₂S⁺: 341.1107, Found: 341.111.

1,3-Diphenylthiourea (6). White solid (23.4 mg, 68%). ¹H NMR (600 MHz, DMSO) δ 9.27 (d, *J* = 9.3 Hz, 2H), 7.47 (d, *J* = 7.6 Hz, 4H), 7.27 (t, *J* = 7.6 Hz, 4H), 6.95 (td, *J* = 7.3, 0.8 Hz, 2H).¹³C[¹H] NMR (101 MHz, DMSO) δ 152.5, 139.7, 128.7, 121.8, 118.2.

3-*Iodo-2-phenylimidazo*[*1,2-a*]*pyridine* (7). Yellow solid (10.4 mg, 16%). ¹H NMR (600 MHz, CDCl₃) δ 8.21 (d, *J* = 6.9 Hz, 1H), 8.07 (d, *J* = 8.0 Hz, 2H), 7.61 (d, *J* = 9.0 Hz, 1H), 7.48 (t, *J* = 7.6 Hz, 2H), 7.40 (t, *J* = 7.4 Hz, 1H), 7.26 – 7.23 (m, 1H), 6.90 (t, *J* = 6.9 Hz, 1H). ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 148.2, 148.1, 133.6, 128.6, 128.4, 128.4, 126.5, 125.5, 117.6, 113.2, 59.5.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website.

Experimental procedures and spectra data for **3** and **5** (PDF) Singly-crystal X-ray diffraction data for compounds **3a** and **5a** (CIF)

AUTHOR INFORMATION

Corresponding Author

Email: zhuxinju@zzu.edu.cn Email: mpsong@zzu.edu.cn Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENT

Financial support from the National Natural Science Foundation of China (Grant Nos. 21672192 and 21803059), the China Postdoctoral Science Foundation (Grant Nos. 2016M602254 and 2016M600582), the Program for Science & Technology Innovation Talents in Universities of Henan Province (Grant No. 17HASTIT004), the Aid Project for the Leading Young Teachers in Henan Provincial Institutions (Grant No. 2015GGJS-157), and the Natural Science Foundation of Henan Province (Grant No. 182300410255) is gratefully appreciated.

REFERENCES

1

2

3

4

5

6

7

8

9

10

11

12

13

14

15

16

17

18

19

20

57 58 59

60

(1) (a) Wang, L.; He, W.; Yu, Z. Transition-Metal Mediated Carbon-Sulfur Bond Activation and Transformations. *Chem. Soc. Rev.* **2013**, *42*, 599-621. (b) Pan, F.; Shi, Z.-J. Recent Advances in Transition-Metal-Catalyzed C-S Activation: From Thioester to (Hetero) Aryl Thioether. *ACS Catal.* **2014**, *4*, 280-288. (c) Feng, M.; Tang, B. S.; Liang, H.; Jiang. X. Sulfur Containing Scaffolds in Drugs: Synthesis and Application in Medicinal Chemistry *Curr. Top. Med. Chem.* **2016**, *16*, 1200-1216. (d) Zhang, C.; Zhu, X. Acc. Thieno[3,4-b]thiophene-Based Novel Small-Molecule Optoelectronic Materials. *Chem. Res.* **2017**, *50*, 1342-1350.

(2) (a) Beletskaya, I.-P.; Anikov, V.-P. Transition-MetalCatalyzed C-S, C-Se, and C-Te Bond Formation via CrossCoupling and Atom-Economic Addition Reactions. *Chem. Rev.* 2011, 111, 1596-1636. (b) Liu, H.; Jiang. X.; Transfer of
Sulfur: From Simple to Diverse. *Chem. Asian. J.* 2013, 8,
2546-2564.

(3) (a) Correa, A.; Carril, M.; Bolm, C. Iron-Catalyzed SArylation of Thiols with Aryl Iodides *Angew. Chem., Int. Ed.* 2008, *47*, 2880-3883. (b) Timpa, S.-D.; Pell, C.-J.;
Ozerov, O.-V. A Well-Defined (POCOP)Rh Catalyst for
the Coupling of Aryl Halides with Thiols. *J. Am. Chem.*Soc. 2014, *136*, 14772-14779.

(4) For reviews, please see: (a) Shen, C.; Zhang, P.; Sun, Q.; 33 Bai, S.; Andy Hor, T.-S.; and Liu, X.; Recent Advances in 34 C-S Bond Formation Via C-H Bond Functionalization and 35 Decarboxylation. Chem. Soc. Rev. 2015, 44, 291-314. (b) 36 Qiao, Z.; Jiang, X. Recent Developments in Sulfur-Carbon 37 Bond Formation Reaction Involving Thiosulfates. Org. 38 Biomol. Chem. 2017, 15, 1942-1946. For selected examples, 39 please see: (c) Yang, F.-L.; Tian, S.-K. Iodine-Catalyzed 40 Regioselective Sulfenylation of Indoles with Sulfonyl Hy-41 drazides. Angew. Chem. Int. Ed. 2013, 52, 4929-4932. (d) 42 Xiao, F.; Xie, H.; Liu, S.; Deng, G.-J. Iodine-Catalyzed Re-43 gioselective Sulfenylation of Indoles with Sodium Sul-44 finates. Adv. Synth. Catal. 2014, 356, 364-368. (e) Yan, S.-45 Y.; Liu, Y.-J.; Liu, B.; Liu, Y.-H.; Shi, B.-F. Nickel-46 Ccatalyzed Thiolation of Unactivated Aryl C-H Bonds: 47 Efficient Access to Diverse Aryl Sulfides. Chem. Commun. 48 2015, 51, 4069-4072. (f) Ye, X.; Petersen, J. L.; Shi, X. Nick-49 el-Catalyzed Directed Sulfenylation of sp² and sp³ C-H 50 Bonds. Chem. Commun. 2015, 51, 7863-7866. (g) Xie, W.; 51 Li, B.; Wang, B. Rh(III)-Catalyzed C7-Thiolation and Sele-52 nation of Indolines. J. Org. Chem. 2016, 81, 396-403. (h) 53 Gensch, T.; Klauck, F.-J.-R.; Glorius, F. Cobalt-Catalyzed 54 C-H Thiolation through Dehydrogenative Cross Coupling. 55 Angew. Chem. Int. Ed. 2016, 55, 11287-11291. (i) Gandeepan, 56

P.; Koeller, J.; Ackermann, L. Expedient C-H Chalcogenation of Indolines and Indoles by Positional-Selective Copper Catalysis. *ACS Catal.* **2017**, *7*, 1030-1034. (j) Sun, P.; Yang, D.; Wei, W.; Jiang, M.; Wang, Z.; Zhang, L.; Zhang, H.; Zhang, Z.; Wang, Y.; Wang, H. Visible Light-Induced C-H Sulfenylation using Sulfinic Acids. *Green Chem.* **2017**, *19*, 4785-4791. (k) Dou, Y.; Huang, X.; Wang, H.; Yang, L.; Li, H.; Yuan, B.; Yang, G. Reusable Ccobalt-Phthalocyanine in Water: Efficient Catalytic Aerobic Oxidative Coupling of Thiols to Construct S-N/S-S Bonds. *Green Chem.* **2017**, *19*, 2491-2495. (l) Guin, S.; Deb, A.; Dolui, P.; Chakraborty, S.; Singh, V. K.; Maiti, D. Promoting Highly Diastereoselective γ -C-H Chalcogenation of α -Amino Acids and Aliphatic Carboxylic Acids. *ACS Catal.* **2018**, 8, 2664-2669.

(5) (a) Enguehard-Gueiffier, C.; Gueiffier, A. Recent Progress in the Pharmacology of Imidazo[1,2-*a*]pyridines. *Mini-Rev. Med. Chem.* **2007**, *7*, 888-899. (b) Stasyuk, A.-J.; Banasiewicz, M.; Cyrański, M.-K.; Gryko, D. T. Imidazo[1,2-*a*]pyridines Susceptible to Excited State Intramolecular Proton Transfer: One-Pot Synthesis via an Ortoleva-King Reaction. *J. Org. Chem.* **2012**, *77*, 5552-5558. (c) Dymińska, L. Imidazopyridines as a Source of Biological Activity and Their Pharmacological Potentials-Infrared and Raman Spectroscopic Evidence of Their Content in Pharmaceuticals and Plant Materials. *Bioorg. Med. Chem.* **2015**, *23*, 6087-6099.

(6) (a) Li, K.; Niu, J.-L.; Yang, M.-Z.; Li, Z.; Wu, L.-Y.; Hao, X.-Q.; Song, M.-P. New Type of 2,6-Bis(imidazo[1,2-a]pyridin-2-yl)pyridine-Based Ruthenium Complexes: Active Catalysts for Transfer Hydrogenation of Ketones. *Organometallics* **2015**, *34*, 1170-1176. (b) Cao, X.-N.; Wan, X.-M.; Yang, F.-L.; Li, K.; Hao, X.-Q.; Shao, T.; Zhu, X.; Song, M.-P. NNN Pincer Ru(II)-Complex-Catalyzed α -Alkylation of Ketones with Alcohols. *J. Org. Chem.* **2018**, *8*3, 3657-3668.

(7) (a) Bagdi, A. K.; Santra, S.; Monir, K.; Hajra, A. Synthesis of Imidazo[1,2-*a*]pyridines: a Decade Update. *Chem. Commun.* **2015**, *51*, 1555-1575. (b) Pericherla, K.; Kaswan, P.; Pandey, K.; Kumar, A. Recent Developments in the Synthesis of Imidazo[1,2-*a*]pyridines. *Synthesis* **2015**, *47*, 887-912.

(8) (a) Koubachi, J.; El Kazzouli, S.; Bousmina, M.; Guillaumet, G. Functionalization of Imidazo[1,2-*a*]pyridines by Means of Metal-Catalyzed Cross-Coupling Reactions. *Eur. J. Org. Chem.* **2014**, 2014, 5119-5138. (b) Bagdi, A. K.; Hajra, A. Design, Synthesis, and Functionalization of Imidazoheterocycles. *Chem. Rec.* **2016**, *16*, 1868-1885. (c) Ravi, C.; Adimurthy. Synthesis of imidazo[1,2-*a*]pyridines: C-H functionalization in the direction of C-S bond formation. *S. Chem. Rec.* **2017**, *17*, 1-21.

(9) (a) Cao, H.; Zhan, H.; Lin, Y.; Lin, X.; Du, Z.; Jiang, H. Direct Arylation of Imidazo[1,2-*a*]pyridine at C-3 with Aryl Iodides, Bromides, and Triflates via Copper(I)-Catalyzed C-H Bond Functionalization. *Org. Lett.* **2012**, *14*, 1688-1691. (b) Cao, H.; Lei, S.; Li, N.; Chen, L.; Liu, J.; Cai, H.; Qiu, S.; Tan, J. Cu-Catalyzed Selective C3-Formylation

56

57 58 59

60

of Imidazo[1,2-a]pyridine C-H Bonds with DMSO using Molecular Oxygen. Chem. Commun. 2015, 51, 1823-1825. (c) 2 Lu, S.; Zhu, X.; Guo, Y.-J.; Wang, M.-D.; Zhao, X.-M.; Hao, 3 X.-Q.; Song, M.-P. Reactivity of *p*-Toluenesulfonylmethyl 4 Isocyanide: Iron-Involved C-H Tosylmethylation of Imid-5 azopyridines in Nontoxic Media. J. Org. Chem. 2016, 81, 6 8370-8377. (d) Mondal, S.; Samanta, S.; Singsardar, M.; 7 Hajra, A. Aminomethylation of Imidazoheterocycles with Morpholine. Org. Lett. 2017, 19, 3751-3754. (e) Chang, Q.; 8 Wu, Z.; Yu, L.; Liu, P.; Sun, P. Visible-Light-Mediated C3-9 Azolylation of Imidazo[1,2-*a*]pyridines with 10 2-Bromoazoles. Org. Biomol. Chem. 2017, 15, 5318-5324. (f) 11 Lei, S.; Mai, Y.; Yan, C.; Mao, J.; Cao, H. A. Carbonylation 12 Approach Toward Activation of Csp²-H and Csp³-H Bonds: 13 Cu-Catalyzed Regioselective Cross Coupling of Imid-14 azo[1,2-*a*]pyridines with Methyl Hetarenes. Org. Lett. 2016, 15 18, 3582-3585. (g) Yang, Q.; Li, S.; Wang, J. Cobalt-16 Catalyzed Cross-Dehydrogenative Coupling of Imid-17 azo[1,2-a]pyridines with Isochroman using Molecular Ox-18 ygen as the Oxidant. Org. Chem. Front. 2018, 5, 577-581. 19 (10) (a) Liu, P.; Gao, Y.; Gu, W.; Shen, Z.; Sun, P. Regiose-20 lective Fluorination of Imidazo[1,2-a]pyridines with Se-21 lectfluor in Aqueous Condition. J. Org. Chem. 2015, 80, 22 11559-11565. (b) Yadav, M.; Dara, S.; Saikam, V.; Kumar, M.; 23 Aithagani, S.-K.; Paul, S.; Vishwakarma, R.-A.; Singh, P.-P. 24 Regioselective Oxidative C-H Phosphonation of Imid-25 azo[1,2-a]pyridines and Related Heteroarenes Mediated 26 by Manganese(III) Acetate. Eur. J. Org. Chem. 2015, 2015, 27 6526-6533. (c) Mondal, S.; Samanta, S.; Jana, S.; Hajra, A. 28 (Diacetoxy)iodobenzene-Mediated Oxidative C-H Amina-29 tion of Imidazopyridines at Ambient Temperature. J. Org. 30 Chem. 2017, 82, 4504-4510. (d) Kibriya, G.; Samanta, S.; 31 Jana, S.; Mondal, S.; Hajra, A. Visible Light Organic Pho-32 toredox-Catalyzed C-H Alkoxylation of Imidazopyridine 33 with Alcohol. J. Org. Chem. 2017, 82, 13722-13727. (e)

34 Zhang, J.-R.; Liao, Y.-Y.; Deng, J.-C.; Feng, K.-Y.; Zhang, 35 M.; Ning, Y.-Y.; Lin, Z.-W.; Tang, R.-Y. Oxidative Dual C-H Thiolation of Imidazopyridines with Ethers or Alkanes 36 37 using Elemental Sulphur. Chem. Commun. 2017, 53, 7784-7787. (f) Guo, Y.-J.; Lu, S.; Tian, L.-L.; Huang, E.-L.; Hao, 38 X.-Q.; Zhu, X.; Shao, T.; Song, M.-P. Iodine-Mediated Di-39 functionalization of Imidazopyridines with Sodium Sul-40 finates: Synthesis of Sulfones and Sulfides. J. Org. Chem. 41 2018, 83, 338-349. (g) Yang, D.; Sun, P.; Wei, W.; Liu, F.; 42 Zhang, H.; Wang, H. Copper-Catalyzed Regioselective 43 Cleavage of C-X and C-H Bonds: A Strategy for Sulfur Di-44 oxide Fixation. Chem. Eur. J. 2018, 24, 4423-4427. (h) Lu, 45 S.; Tian, L.-L.; Cui, T.-W.; Zhu, Y.-S.; Zhu, X.; Hao, X.; 46 Song, M.-P. Copper-Mediated C-H Amination of Imidaz-47 opyridines with N-Fluorobenzenesulfonimide. J. Org. 48 Chem. 2018, 83, 13991-14000. 49

(11) (a) Shakoor, S.-M.-A.; Kumari, S.; Khullar, S.; Mandal, 50 S. K.; Kumar, A.; Sakhuja, R. Ruthenium(II)-Catalyzed 51 Regioselective Ortho Amidation of Imidazo Heterocycles 52 with Isocyanates. J. Org. Chem. 2016, 81, 12340-12349. (b) 53 Zhu, X.; Shen, X.-J.; Tian, Z.-Y.; Lu, S.; Tian, L.-L.; Liu, W.-54 B.; Song, B.; Hao, X.-Q. Rhodium-Catalyzed Direct Bis-55

cyanation of Arylimidazo $[1,2-\alpha]$ pyridine via Double C-H Activation. J. Org. Chem. 2017, 82, 6022-6031.

(12) (a) Kaswan, P.; Nandwana, N.-K.; DeBoef, B.; Kumar, A. Vanadyl Acetylacetonate Catalyzed Methylenation of Imidazo[1,2-*a*]pyridines by Using Dimethylacetamide as a Methylene Source: Direct Access to Bis(imidazo[1,2a]pyridin-3-yl)methanes. Adv. Synth. Catal. 2016, 358, 2108-2116. (b) Modi, A.; Ali, W.; Patel, B. K. N,N-Dimethylacetamide (DMA) as a Methylene Synthon for Regioselective Linkage of Imidazo[1,2-a]pyridine. Adv. Synth. Catal. 2016, 358, 2100-2107. (c) Liu, P.; Shen, Z.; Yuan, Y.; Sun, P. Synthesis of Symmetrical Methylene-Bridged Imidazoheterocycles using DMSO as Methylene Source under Metal-Free Conditions Org. Biomol. Chem. 2016, 14, 6523-6530. (d) Patel, O.-P.-S.; Anand, D.; Maurya, R.-K.; Yadav, P.-P. Synthesis of Symmetrical Methylene-Bridged Imidazoheterocycles using DMSO as Methylene Source under Metal-Free Conditions. J. Org. Chem. 2016, 81, 7626-7634.

(13) Shakoor, S. M. A.; Agarwal, D. S.; Khullar, S.; Mandal, S. K.; Sakhuja, R. Solvent-Driven Iodine-Mediated Oxidative Strategies for the Synthesis of Bis(imidazo[1,2a]pyridin-3-yl)sulfanes and Disulfanes. Chem. Asian. J. 2017, 12, 3061-3068.

(14) Shiri, M.; Zolfigol, M.-A.; Kruger, H.-G.; Tanbakouchian, Z. Bis- and Trisindolylmethanes (BIMs and TIMs). Chem. Rev. 2000, 110, 2250-2293.

(15) (a) Palsuledesai, C.-C.; Murru, S.; Sahoo, S.-K.; Patel, B.-K. Acyl-isothiocyanates as Efficient Thiocyanate Transfer Reagents. Org. Lett. 2009, 11, 3382-3385. (b) Sun, N.; Che, L.; Mo, W.; Hu, B.; Shen, Z.; Hu, X. A Mild Copper-Catalyzed Aerobic Oxidative Thiocyanation of Arylboronic Acids with TMSNCS. Org. Biomol. Chem. 2015, 13, 691-696.

(16) (a) Zhao, P.; Yan, X.; Yin, H.; Xi, C. Alkyltriflate-Triggered Annulation of Arylisothiocyanates and Alkynes Leading to Multiply Substituted Quinolines through Domino Electrophilic Activation. Org. Lett. 2014, 16, 1120-1123. (b) Hao, W.; Zeng, J.; Cai, M. The Copper(I)-Catalyzed Tandem Reaction of O-alkynylphenyl Isothiocyanates with Isocyanides: a Rapid Synthesis of 5Hbenzo[d]imidazo-[5,1-b][1,3]thiazines. Chem. Commun. 2014, 50, 11686-11689. (c) Wen, L.-R.; Li, S.-L.; Zhang, J.; Li, M. Convenient Synthesis of Benzo[4,5]thiazolo[2,3-c]-[1,2,4]triazoles with 1 mol% CuCl₂·2H₂O as Catalyst in Water. Green Chem. 2015, 17, 1581-1588.

(17) (a) Tang, X.; Zhu, Z.; Qi, C.; Wu W.; Jiang, H. Copper-Catalyzed Coupling of Oxime Acetates with Isothiocyanates: A Strategy for 2-Aminothiazoles. Org. Lett. 2016, 18, 180-183. (b) He, Y.; Li, J.; Luo, S.; Huang, J.; Zhu, Q. A Metal-Free Synthesis of 2-Aminobenzothiazoles through Aminyl Radical Addition to Aryl Isothiocyanates. Chem. Commun. 2016, 52, 8444-8447. (c) Guo, W.; Zhao, M.; Tan, W.; Zheng, L.; Tao, K.; Liu, L.; Wang, X.; Chen, D.; Fan, X. Visible Light-Promoted Three-Component Tandem Annulation for the Synthesis of 2-Iminothiazolidin-4-ones. J. Org. Chem. 2018, 83, 1402-1413.

(18) (a) Zhao, P.; Liu, Y.; Xi, C. MeOTf-Induced Carboannulation of Isothiocyanates and Aryl Alkynes with C=S Bond Cleavage: Access to Indenones. *Org. Lett.* 2015, *17*, 4388-4391. (b) Chu, J.-J.; Hu, B.-L.; Liao, Z.-Y.; Zhang, X.-G. Copper-Catalyzed Three-Component Tandem Cyclization for One-Pot Synthesis of 1,4-Benzothiazines. *J. Org. Chem.* 2016, *81*, 8647-8652. (c) Modi, A.; Sau, P.; Patel, B. K. Base-Promoted Synthesis of Quinoline-4(1H)-thiones from o-Alkynylanilines and Aroyl Isothiocyanates. *Org. Lett.* 2017, *19*, 6128-6131.

(19) (a) Guin, S.; Rout, S.-K.; Gogoi, A.; Nandi, S.; Ghara, 10 K.-K.; Patel, B.-K. Desulfurization Strategy in the Con-11 struction of Azoles Possessing Additional Nitrogen, Oxy-12 gen or Sulfur using a Copper(I) Catalyst. Adv. Synth. Catal. 13 2012, 354, 2757-2770. (b) Ali, W.; Dahiya, A.; Pandey, R.; 14 Alam, T.; Patel, B.-K. Microwave-Assisted Cascade Strate-15 gy for the Synthesis of Indolo[2,3-b]quinolines from 2-16 (Phenylethynyl)anilines and Aryl Isothiocynates. J. Org. 17 Chem. 2017, 82, 2089-2096. 18

(20) (a) Guo, W.-S.; Dou, Q.; Hou, J.; Wen, L.-R.; Li, M. 19 Synthesis of 6-Phosphorylated Phenanthridines by 20 Mn(II)-Promoted Tandem Reactions of 2-Biaryl Isothio-21 cyanates with Phosphine Oxides. J. Org. Chem. 2017, 82, 22 7015-7022. (b) Wen, L.-R.; Sun, Y.-X.; Zhang, J.-W.; Guo, 23 W.-S.; Li, M. Catalyst- and Solvent-free Bisphosphinyla-24 tion of Isothiocyanates: a Practical Method for the Syn-25 thesis of Bisphosphinoylaminomethanes. Green Chem. 26 2018, 20, 125-129. (c) Zhang, X.; Wang, T.-L.; Huo, C.-D.; 27 Wang, X.-C.; Quan, Z.-J. Base-Controlled Chemoselectivi-28 ty Reaction of Vinylanilines with Isothiocyanates for Syn-29 thesis of Quinolino-2-Thione and 2-Aminoquinoline De-30 rivatives. Chem. Commun. 2018, 54, 3114-3117.

31 (21) (a) Wang, W.; Niu, J.-L.; Liu, W.-B.; Shi, T.-H.; Hao, 32 X.-Q.; Song, M.-P. Rhodium(III)-Catalyzed Annulation of 33 2-Arylimidazo[1,2-*a*]pyridines and Alkynes via Direct 34 Double C-H Activation. Tetrahedron 2015, 71, 8200-8207. 35 (b) Li, K.; Zhu, X.; Lu, S.; Zhou, X.-Y.; Xu, Y.; Hao, X.-Q.; Song, M.-P. Catalyst-Free Friedel-Crafts Alkylation of Im-36 37 idazo[1,2-α]pyridines. Synlett 2016, 27, 387-390. (c) Zhu, M.; Han, X.; Fu, W.; Wang, Z.; Ji, B.; Hao, X.-Q.; Song, M.-38 P.; Xu, C. Regioselective 2,2,2-Trifluoroethylation of Imid-39 azopyridines by Visible Light Photoredox Catalysis. J. Org. 40 Chem. 2016, 81, 7282-7287. 41

(22) Roy, S.; Phukan, P. Biaryl Thioether Synthesis via Cul
Catalyzed Dominothiolation of Aryl Halides in the Presence of DMAP as Ligand. *Tetrahedron Lett.* 2015, 56, 24262429.

(23) (a) Allen, S.-E.; Walvoord, R.-R.; Padilla-Salinas, R.;
Kozlowski, M.-C. Aerobic Copper-Catalyzed Organic Reactions. *Chem. Rev.* 2013, 113, 6234-6458. (b) Rao, W.-H.;
Shi, B.-F. Recent Advances in Copper-Mediated Chelation-Assisted Functionalization of Unactivated C-H Bonds. *Org. Chem. Front.* 2016, 3, 1028-1047.

51 (24) (a) Ramana T.; Saha, P.; Das, M.; Punniyamurthy, T.
52 Copper-Catalyzed Domino Intra- and Intermolecular C-S
53 Cross-Coupling Reactions: Synthesis of 254 (Arylthio)arylcyanamides. Org. Lett. 2010, 12, 84-87. (b)

Ghorbani-Choghamarani, A.; Moradi, Z.; Azadi, G. An Efficient and Recyclable Catalytic System for Carbon– Sulfur Coupling Reaction and Synthesis of 5-Substituted 1H-Tetrazoles. *J. Sulfur. Chem.* **2018**, 39, 237-251.

(25) (a) Bagdi, A.-K.; Rahman, M.; Santra, S.; Majee, A.; Hajra, A. Copper-Catalyzed Synthesis of Imidazo[1,2*a*]pyridines through Tandem Imine Formation-Oxidative Cyclization under Ambient Air: One-Step Synthesis of Zolimidine on a Gram-Scale. *Adv. Synth. Catal.* **2013**, 355, 1741-1747. (b) Pericherla, K.; Khedar, P.; Khungar, B. and Kumar, A. One-Pot Sequential C-N Coupling and Cross Dehydrogenative Couplings: Synthesis of Novel Azole Fused Imidazo[1,2-*a*]pyridines. *Chem. Commun*, **2013**, 49, 2924-2926.

(26) (a) Wang, Z.; Li, B.; Zhang, X.; Fan, X. One-Pot Cascade Reactions Leading to Pyrido[2',1':2,3]Imidazo[4,5c][1,2,3]Triazolo[1,5-*a*]quinolines under Bimetallic Relay Catalysis with Air as the Oxidant. *J. Org. Chem.* **2016**, *81*, 6357–6363. (b) Cao, G.; Chen, Z.; Song, J.; Xu, J.; Miao, M.; Ren, H. Oxidant-Mediated Nitrogenation and Recyclization of Imidazo[1,2-*a*]pyridines with Sodium Azide: Synthesis of 4*H*-Pyrido[1,2-*a*][1,3,5]Triazin-4-Ones. *Adv. Synth. Catal.* **2018**, *360*, 881-886. (c) Wang,T.; Chen, J.; Wang, J.; Xu, S.; Lin, A.; Yao, H.; Jiang, S.; Xu, J. Cobalt-Catalyzed Carbon-Sulfur/Selenium Bond Formation: Synthesis of Benzo[*b*]Thio/Selenophene-Fused Imidazo[1,2*a*]Pyridines. *Org. Biomol. Chem.*, **2018**, *16*, 3721-3725.

(27) Niedermann, K.; Welch, J.-M.; Koller, R.; Cvengroš, J.; Santschi, N. Battaglia, P.; Togni, A. New Hypervalent Iodine Reagents for Electrophilic Trifluoromethylation and their Precursors: Synthesis, Structure, and Reactivity. *Tetrahedron* **2010**, *66*, 5753-5761.

(28) Štrukil, V.; Igrc, M.-D.; Fábián, L.; Eckert-Maksić, M.; Childs, S.-L.; Reid, D.-G.; Duer, M.-J.; Halasz, I.; Mottilloe, C. and Friščić, T. A Model for a Solvent-Free Synthetic Organic Research Laboratory: Click-Mechano Synthesis and Structural Characterization of Thioureas Without Bulk Solvents. *Green Chem.*, **2012**, *14*, 2462-2473.

(29) Mondal, S.; Samanta, S.; Mukta Singsardar, M.; Mishra, S.; Mitra, S.; Hajra, A. Zwitterionic-Type Molten Salt Catalyzed Iodination in Water: Synthesis of Iodoimidazoheterocycles. Synthesis. **2016**, *48*, 4009-4015.

59 60

56

57 58

1

2

3

4

5

6

7

8

9