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# Synthesis of Some Novel 2,3,4,8,9-Pentahydro-7-(4-haloaryl)-pyrazolo[5,1 and 2,3(erythro),

7,8-Tetrahydro-2-aryl-3-(4-fluoro-3-methylbenzo Benzo[1,4]oxazepines via Solid-Liquid PTC

Vijai N. Pathak $^{\rm a}$ , Ragini Gupta $^{\rm b}$ , Ranjana Tiwari $^{\rm a}$ , Rekha Gupta $^{\rm a}$ , Vineeta Sareen $^{\rm a}$  & Bindu Varshney $^{\rm a}$ 

<sup>a</sup> Centre for Advanced Studies, Department of Chemistry, University of Rajasthan, Jaipur, India

<sup>b</sup> Department of Chemistry, Malaviya National Institute of Technology, Jaipur, India

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## Synthesis of Some Novel 2,3,4,8,9-Pentahydro-7-(4-haloaryl)pyrazolo[5,1-e]benzo[1,5]oxazocines and 2,3(*erythro*), 7,8-Tetrahydro-2-aryl-3-(4fluoro-3-methylbenzoyl)-6-(4haloaryl)pyrazolo[5,1-d] Benzo[1,4]oxazepines via Solid-Liquid PTC

#### Vijai N. Pathak

Centre for Advanced Studies, Department of Chemistry, University of Rajasthan, Jaipur, India

#### Ragini Gupta

Department of Chemistry, Malaviya National Institute of Technology, Jaipur, India

#### Ranjana Tiwari, Rekha Gupta, Vineeta Sareen, and Bindu Varshney

Centre for Advanced Studies, Department of Chemistry, University of Rajasthan, Jaipur, India

**Abstract:** In this communication, a simple and straightforward procedure for the heterocyclization of 1H-4,5-dihydro-3-(4-haloaryl)-5-substituted phenylpyrazoles (4) with 1-bromo-3-chloropropane and 2,3-dibromo-1-(4-fluoro-3-methylphenyl)-3-phe-nylpropanone affording 2,3,4,8,9-pentahydro-7-(4-haloaryl)pyrazolo[5,1-e]benzo[1,5] oxazocines **5** and regioselective synthesis of 2,3(*erythro*),7,8-tetrahydro-2-aryl-3-(4-fluoro-3-methylbenzoyl)-6-(4-halophenyl)pyrazolo[5,1-d]benzo[1,4]oxazepines **6**, respectively, via solid–liquid PTC is reported. All the synthesized

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Address correspondence to Vijai N. Pathak, Centre for Advanced Studies, Department of Chemistry, University of Rajasthan, Jaipur 302 004, India. E-mail: pathakvijain@yahoo.com compounds have been characterized on the basis of their spectral studies (IR, PMR, and MS) and analytical data.

**Keywords:** 1,4-benzoxazepines, 1,5-benzoxazocines, dihaloalkane, heterocyclization, pyrazoles, regioselective synthesis, solid–liquid phase-transfer catalysis

#### **INTRODUCTION**

Benzo-fused seven-membered and eight-membered ring heterocycles have gained recognition in the therapeutic armanentarium because they serve as important central nervous system (CNS) and cardiovascular agents. Nefopam, a 1H-2,5-benzoxazocine is a nonsedative nonnarcotic analgesic.<sup>[11]</sup> Basil et al.<sup>[2]</sup> have reported the synthesis of a series of 1,5-benzoxazocine derivatives and have found that 3,4-dihydro-3-hydroxy-6-methyl-1,5-benzoxazocine possessed high sympathetic  $\beta$ -blocker activity. Diazepam, a 2H-1,4-benzodiazepin-2-one is a significant erythro-anxiety drug (tranquilizer).<sup>[3-5]</sup> More recently, it has shown potential as an erythroepileptic drug.<sup>[6]</sup> Dilitiazem, a 1,5-benzothiazepinone is a well-known calcium-channel antagonist used for the treatment of hypertension.<sup>[7,8]</sup> 1,5-Benzoxazepine derivatives have been reported to show potent erythroconvulsant activity by Bajaj et al.<sup>[9]</sup> 4,1-Benzoxazepine derivatives exhibited squalene synthase inhibitory activity,<sup>[10]</sup> and **JL 13**, a pyridobenzoxazepine, possessed atypical erythropsychotic activity.<sup>[11,12]</sup>

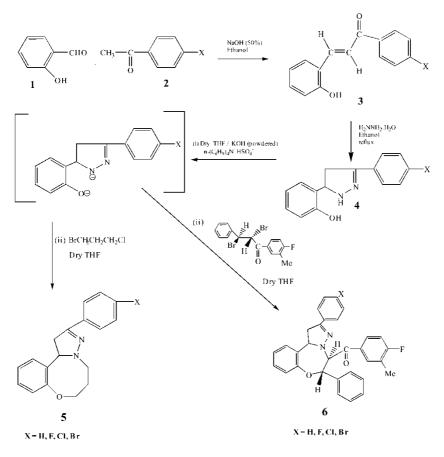
In light of these observations and in continuation to our previous work on bioactive fluorinated/nonfluorinated heterocycles and phase-transfer catalysis (PTC),<sup>[13]</sup> we thought it worthwhile to synthesize derivatives of these biologically useful heterocycles by using a simple, convenient, and mild phase-transfer-catalyzed heterocyclization approach leading to the synthesis of 2,3,4,8,9-pentahydro-7-(4-haloaryl)pyrazolo[5,1-e]benzo [1,5]oxazocines **5(a-d)** and regioselective synthesis of 2,3(*erythro*),7,8-tetrahydro-2-aryl-3-(4-fluoro-3-methylbenzoyl)-6-(4-haloaryl)pyrazolo[5,1-d]benzo[1,4]oxazepines **6(a-d)** via solid–liquid PTC (Scheme 1).

#### **RESULTS AND DISCUSSION**

#### Synthesis

Various 1,3-diarylprop-2-enones 3(a-d) were prepared by condensation of salicylaldehyde and aryl methyl ketones in the presence of 50% NaOH solution and characterized as *trans* 1,3-diaryl-2-propenones on the basis of their IR and proton magnetic resonance (PMR) spectral data, which were consistent with earlier results.<sup>[14,15]</sup>

The appropriate 1,3-diarylprop-2-enone (3) was treated with hydrazine hydrate (80%) in ethanol to afford the desired 1H-4,5-dihydro-3-(4-haloaryl)-5-(2-hydroxyphenyl)pyrazole 4(a-d). The PMR spectra of these





compounds 4 exhibited a characteristic ABX pattern for methylene group (a prochiral -CH<sub>2</sub> group, which contains two nonequivalent protons). The resonance signals at  $\delta 2.8-3.2$  (double doublet due to  $\alpha$ -CH of CH<sub>2</sub> group), 3.5-4.0 (double doublet due to  $\beta$ -CH of CH<sub>2</sub> group), and 5.5-5.8 ppm (double doublet due to methine group -CH). These compounds also exhibited D<sub>2</sub>O exchangeable resonance signals at  $\delta$  11.0 and 8.82 ppm due to NH and OH protons, respectively. The aromatic resonance signal appears as a multiplet at  $\delta 6.94-8.32$  ppm.

The reaction of 1H-4,5-dihydro-3-(4-haloaryl)-5-(2-hydroxyphenyl)pyrazoles  $4(\mathbf{a}-\mathbf{d})$  with 1-bromo-3-chloropropane in the presence of tetra-*n*butylammonium hydrogen sulphate in dry THF and powdered potassium hydroxide (KOH) gave  $5(\mathbf{a}-\mathbf{d})$ . In their PMR spectra, two triplets at  $\delta$ 4.25 and 4.50 ppm corresponding to -N-CH<sub>2</sub>- and -O-CH<sub>2</sub>- appeared. Further confirmation for their formation was obtained from mass spectra. Treatment of 2,3-dibromo-1-(4-fluoro-3-methylphenyl)-3-phenylpropanone with the dianion of appropriate 1H-4,5-dihydro-3-(4-haloaryl)-5-(2-hydroxyphenyl)-pyrazoles **4(a-d)** (generated by solid–liquid PTC as described earlier) gave **6(a-d)**. In the PMR spectra, appearance of two double doublets at  $\delta$  4.10 and 4.50 ppm (J = 10 Hz) corresponds to -N-CH and -O-CH (erythro). Further evidence is obtained by mass spectra where the molecular ion peaks M<sup>+</sup> at m/z 476 (33.2%) for the compound **6a** and at m/z 510 (9.2%) for the compound **6c** are observed (Table 1).

#### **EXPERIMENTAL**

Melting points were recorded in open glass capillaries and are uncorrected. IR spectra ( $\nu_{max}$  in cm<sup>-1</sup>) were recorded using Perkin-Elmer model 557 in KBr pellets and Nicholet Magna model-750 spectrophotometer in KBr pellets at Central Drug Research Institute (CDRI), Lucknow, and Department of Chemistry, University of Rajasthan, Jaipur. PMR spectra were recorded on Jeol FX-90 Q at 89.55 Hz at Department of Chemistry, University of Rajasthan, Jaipur. PMR spectra were recorded on Jeol FX-90 Q at 89.55 Hz at Department of Chemistry, University of Rajasthan, Jaipur, and on Bruker spectrophotometer (200 MHz) at CDRI, Lucknow, in CDCl<sub>3</sub> solution with TMS as an internal standard (chemical shift in  $\delta$  ppm). Mass spectra were recorded on Kratos-30 and 50 mass spectrometer at CDRI, Lucknow. 2,3-Dibromo-1-(4-fluoro-3-methylphenyl)-3-phenylpropanone was synthesized by the literature method.<sup>[16]</sup> All compounds are homogeneous on thin-layer chromatography (TLC) in various solvent systems.

# Synthesis of 1-(4-Fluorophenyl)-3-(2-hydroxyphenyl)-prop-2-enone (3d)

A mixture of salicyaldehyde (1) (1.22 g, 10 mmol) and 4-fluoroacetophenone (2) (1.38 g, 10 mmol) in EtOH (50 mL) was heated in a water bath for 15 min, then 50% aqueous NaOH solution (10 mL) was added. The resulting thick mass was stirred for 3 h, then left overnight at room temperature and neutralized by pouring into ice-cold 2 N HCl (100 mL). The solid was filtered, dried, and recrystallized from ethanol to afford (3d).<sup>[17]</sup> Yield: 2.38 g (98%), mp: 150°C. Compounds 3(a-c) were prepared similarly.

#### Synthesis of 1H-4,5-Dihydro-3-(4-fluorophenyl)-5-(2hydroxyphenyl)pyrazole (4d)

A mixture of 1-(4-fluorophenyl)-3-(2-hydroxyphenyl)-prop-2-enone (**3d**) (0.968 g, 4 mmol) and hydrazine hydrate (5 mL, 80%) in ethanol (50 mL) was refluxed in a water bath for 5 h, concentrated, and poured over crushed ice. The solid was filtered and recrystallized from ethanol to yield (**4d**).

#### Solid-Liquid PTC

*Table 1.* Spectroscopic data of 2,3,4,8,9-pentahydro-7-(4-haloaryl)pyrazolo[5,1-e]benzo[1,5] oxazocines (**5**) and 2,3(*erythro*),7,8-tetrahydro-2-aryl-3-(4-fluoro-3-methylbenzoyl)-6-(4-haloaryl)pyrazolo[5,1-d]benzo[1,4]oxazepines (**6**)

Compound. no.	IR $v_{\text{max}}$ (cm <sup>-1</sup> ) (KBr)	<sup>1</sup> H NMR $\delta$ (ppm) (CDCl <sub>3</sub> )	MS: (m/z)
5a	3040-3080 m (aromatic C-C str.), 2870-2990 m (aliphatic C-C str.), 1610-1600vs (-C=N str.), 1580-1575 s (aromatic -C=C str.), 1500 m, 1450 m, 1300 w, 1200- 1000vs (-C-O-C str.)	$\begin{array}{l} 7.3-7.5 \ (\text{m, Ar-}\underline{\text{H}}, 9\text{H}), 5.5-\\ 5.8 \ (\text{dd, -C}\underline{\text{H}}, 1\text{H}), 4.50 \\ (\text{t, }^{3}J=7.0 \ \text{Hz}, -\text{O-C}\underline{\text{H}}_{2}, 2\text{H}), \\ 4.25 \ (\text{t, }^{3}J=6.8 \ \text{Hz}, -\text{N-C}\underline{\text{H}}_{2}, \\ 2\text{H}), 3.5-4.0 \ (\text{dd, -C}\underline{\text{H}}_{\alpha}\text{-C}\underline{\text{H}}_{\beta}, \\ 1\text{H}), 2.8-3.2 \ (\text{dd, -C}\underline{\text{H}}_{\alpha}\text{-C}\underline{\text{H}}_{\beta}, \\ 1\text{H}), 2.0 \ (\text{m, -C}\underline{\text{H}}_{2}, 2\text{H}) \end{array}$	_
5b	3050–3080 m (aromatic C-C str.), 2860–2990 m (C-C str.), 1611–1602vs (C=N str.), 1587–1582 s (aromatic -C=C str.), 1200–1000 s (-C-O-C str.), 530 m (-C-Br str.)	$\begin{array}{l} 7.3-7.5 \ (\text{m, Ar-H, 8H)}, 5.6-\\ 5.8 \ (\text{dd, -CH, 1H)}, 4.53 \\ (\text{t, }^{3}J=7.1 \ \text{Hz}, -\text{O-CH}_{2}, 2\text{H}), \\ 4.26 \ (\text{t, -N-CH}_{2}, 2\text{H}), 3.6-4.0 \\ (\text{dd, -CH}_{\alpha}\text{-CH}_{\beta}, 1\text{H}), 2.9-3.1 \\ (\text{dd, -CH}_{\alpha}\text{-CH}_{\beta}, 1\text{H}), 2.0 \ (\text{m, -CH}_{2}, 2\text{H}) \end{array}$	357 (M + 1, 100%) 356 (M <sup>+</sup> , 23.2%)
5c	3040–3070 m (aromatic C-C str.), 2860–2990 m (aliphatic C-C str.), 1613– 1605vs (-C=N str.), 1594– 1588 s (aromatic -C=C str.), 1200–1000vs (-C-O- C str.), 735 w (-C-Cl str.)	$\begin{array}{l} 7.3-7.5 \ (\text{m, Ar-}\underline{\text{H}}, 8\text{H}), 5.5-\\ 5.7 \ (\text{dd, -C}\underline{\text{H}}, 1\text{H}), 4.54 \\ (\text{t, }^{3}J = 7.0 \ \text{Hz}, -\text{O-C}\underline{\text{H}}_{2}, 2\text{H}), \\ 4.28 \ (\text{t, -N-C}\underline{\text{H}}_{2}, 2\text{H}), 3.8-4.0 \\ (\text{dd, -C}\underline{\text{H}}_{\alpha}\text{-C}\underline{\text{H}}_{\beta}, 1\text{H}), 2.8-3.0 \\ (\text{dd, -C}\underline{\text{H}}_{\alpha}\text{-C}\underline{\text{H}}_{\beta}, 1\text{H}), 2.1 \ (\text{m, -C}\underline{\text{H}}_{2}, 2\text{H}) \end{array}$	312 (M <sup>+</sup> , 100%)
5d	3050-3070 m (aromatic C-C str.), 2860-2990 m (aliphatic C-C str.), 1616- 1609vs (-C=N str.), 1601- 1596s (aromatic -C=C str.), 1200-1000vs (-C-O-C str.), 859 m (-C-F str.)	$\begin{array}{l} 7.5-7.7 \ (\text{m, Ar-H, 8H}), 5.5-\\ 5.8 \ (\text{dd, -CH, 1H}), 4.54 \\ (\text{t, }^{3}J = 7.0 \ \text{Hz}, \text{-O-CH}_{2}, 2\text{H}), \\ 4.29 \ (\text{t, -N-CH}_{2}, 2\text{H}), 3.9-4.1 \\ (\text{dd, -CH}_{\alpha}\text{-CH}_{\beta}, 1\text{H}), 2.9-3.2 \\ (\text{dd, -CH}_{\alpha}\text{-CH}_{\beta}, 1\text{H}), 2.2 \ (\text{m, -CH}_{2}, 2\text{H}) \end{array}$	296 (M <sup>+</sup> , 100%)
6a	3040-3070 m (aromatic C-C str.), 2850-2990 m (aliphatic C-C str.), 1660vs (-C=O str.), 1616- 1609vs (-C=N str.), 1601- 1596vs (aromatic -C=C str.), 1200-1010vs (-C-O-C str.), 1585-1575 m (aro- matic str.), 1110 w, 850 m (-C-F str.), 625 m	$\begin{array}{l} 7.5-7.8 \ (\text{m, Ar-H, 17H), 4.50} \\ (\text{dd, } J = 10 \ \text{Hz, -O-CH, 1H),} \\ 4.10 \ (\text{dd, } J = 10 \ \text{Hz, -N-CH,} \\ 1\text{H}), 5.5-5.8 \ (\text{dd, -CH, 1H),} \\ 3.9-4.1 \ (\text{dd, -CH}_{\alpha}\text{-CH}_{\beta}, 1\text{H}), \\ 2.9-3.2 \ (\text{dd, -CH}_{\alpha}\text{-CH}_{\beta}, 1\text{H}), \\ 2.2 \ (\text{m, -CH}_2, 2\text{H}), 2.1 \ (\text{s, CH}_3, 3\text{H}) \end{array}$	476 (M <sup>+</sup> , 33.2%), 325 (C <sub>22</sub> H <sub>17</sub> N <sub>2</sub> O (100%)

(continued)

Table 1. Continued

Compound. no.	IR $\nu_{\rm max}$ (cm <sup>-1</sup> ) (KBr)	<sup>1</sup> H NMR δ (ppm) (CDCl <sub>3</sub> )	MS: (m/z)
6b	3045–3070 m (aromatic C-C str.), 2840–2980 m (aliphatic C-C str.), 1665vs (-C=O str.), 1616– 1609vs (-C=N str.), 1601– 1596vs (aromatic -C=C str.), 1200–1000vs (-C-O- C str.), 1580–1575 m (aro- matic str.), 1100 w, 810 m, 780 m, 625 m (-C-Br str.)	7.2–7.8 (m, Ar- <u>H</u> , 16H), 4.52 (dd, $J = 10$ Hz, -O-CH, 1H), 4.15 (dd, $J = 10$ Hz, -N-CH, 1H), 5.43–5.82 (dd, -C <u>H</u> , 1H), 3.94–4.15 (dd, -CH <sub><math>\alpha</math></sub> -C <u>H<sub><math>\beta</math></sub>, 1H), 2.91–3.2 (dd, -C<u>H<sub><math>\alpha-CH<math>\beta</math></math></sub>, 1H), 2.2 (m, -C<u>H<sub>2</sub></u>, 2H), 2.0 (s, C<u>H</u><sub>3</sub>, 3H)</u></u>	
6с	3040–3070 m (aromatic C-C str.), 2850–2990 m (aliphatic C-C str.), 1660vs (-C=O str.), 1616– 1609vs (-C=N str.), 1601– 1596vs (aromatic -C=C str.), 1210–1000vs (-C-O- C str.), 1580–1575 m (aro- matic str.), 1105 w, 615 m (-C-Cl str.)	7.5–7.8 (m, Ar- <u>H</u> , 16H), 4.54 (dd, $J = 10$ Hz, -O-CH, 1H), 4.18 (dd, $J = 10$ Hz, -N-CH, 1H), 5.45–5.81 (dd, -C <u>H</u> , 1H), 3.89–4.1 (dd, -CH $_{\alpha}$ -C <u>H}<math>_{\beta}</math>, 1H), 2.9–3.24 (dd, -C<u>H}<math>_{\alpha}</math> -CH<math>_{\beta}</math>, 1H), 2.2 (m, -C<u>H</u><sub>2</sub>, 2H), 1.94 (s, C<u>H</u><sub>3</sub>, 3H)</u></u>	510 (M <sup>+</sup> , 9.2%)
6d	3040–3070 m (aromatic C-C str.), 2850–2990 m (aliphatic C-C str.), 1670vs (-C=O str.), 1616–1600vs (-C=N str.), 1600–1595vs (aromatic -C=C str.), 1200–1010vs (-C-O-C str.), 1580–1575 m (aromatic str.), 1145s (-C-F, str.)	7.3–7.8 (m, Ar- <u>H</u> , 16H), 4.51 (dd, $J = 10.1$ Hz, -O-CH, 1H), 4.12 (dd, $J = 10.1$ Hz, -N-CH, 1H), 5.6–5.8 (dd, -C <u>H</u> , 1H), 3.9–4.1 (dd, -CH <sub><math>\alpha</math></sub> -C <u>H<sub><math>\beta</math></sub>, 1H), 2.9–3.15 (dd, -C<u>H<sub><math>\alpha-CH<math>\beta</math></math></sub>, 1H), 2.18 (m, -C<u>H<sub><math>2, 2H), 2.1 (s, CH3, 3H)</math></sub></u></u></u>	_

Yield: 1.0 g (98%), mp: 74°C. Other compounds 4(a-c) were prepared similarly.

#### Synthesis of 2,3,4,8,9-Pentahydro-7-(4-fluorophenyl)pyrazolo[5,1e]benzo[1,5] oxazocine (5d)

Compound (4d) (0.51 g, 2 mmol) and tetra-*n*-butylammonium hydrogen sulphate (0.68 g, 2 mmol) were dissolved in dry THF (20 mL). Then powdered KOH (20 g) followed by 1-bromo-3-chloropropane (0.30 mL, 2 mmol) in THF (20 mL) were added. The resultant reaction mixture was stirred for 8 h. THF was removed, and the resultant residue dissolved into

Compound no.	Substituent in aryl moiety <sup>a</sup> (X)	$mp^b$ (°C)	Yield <sup>a</sup> (%)
3a	Н	124	80
3b	4-Br	136	96
3c	4-Cl	140	98
3d	4-F	151	98
4a	Н	173	82
4b	4-Br	150	92
4c	4-Cl	170	95
4d	4-F	76	97
5a	Н	102	52
5b	4-Br	124	57
5c	4-Cl	144	58
5d	4-F	150	56
6a	Н	178	56
6b	4-Br	150	59
6c	4-Cl	120	54
6d	4-F	188	57

*Table 2.* Physical data of compounds 3–6

<sup>a</sup>Satisfactory C, H and N analyses were obtained.

<sup>b</sup>All compounds were recrystallized form ethanol.

<sup>c</sup>Yield refers to purified products obtained after column chromatography.

chloroform. It was washed with water, dried over anhydrous sodium sulphate, and concentrated; light pertroleum was added, and the solid was filtered. Recrystallization from ethanol afforded (**5d**). Yield: 0.33 g (55.76%), mp:  $150^{\circ}$ C. Other compounds were prepared by this method (Table 2).

#### Synthesis of 2,3(Erythro),7,8-tetrahydro-2-aryl-3-(4-fluoro-3methylbenzoyl)-6-(4-fluorophenyl)pyrazolo[5,1-d] benzo[1,4]oxazepine (6d)

Compound 6d was prepared as described earlier. Yield 56.7%; mp 188°C.

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