



An Azeotropic Drying-Free Approach for Copper-mediated Radiofluorination without Addition of Base

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Abstract

Copper-mediated radiofluorination provides a quick and versatile approach for ^{18}F -labeling of arenes and heteroarenes. However, this method is known to be base-sensitive which has been a barrier for preparative scale radiosynthesis. In this report, we provide an approach for copper-mediated radiofluorination without azeotropic drying or adding a base. ^{18}F Fluoride trapped on a PS- HCO_3 Sep-Pak was quantitatively eluted with a solution of 4-dimethylaminopyridinium trifluoromethanesulfonate (DMAP \cdot OTf) in anhydrous *N,N*-dimethylformamide (DMF). The eluted solution was directly used for copper-mediated radiofluorination. Twelve boronic ester substrates were tested, yielding fluorinated products in 27–83% radiochemical yield based on HPLC analysis. This approach was successfully applied to the radiosynthesis of ^{18}F flumazenil, a well-known PET tracer for imaging central benzodiazepine receptors, with a radiochemical yield of 47%. This highly efficient protocol significantly augments the powerful copper-mediated radiofluorination approach.

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1. Introduction

Positron emission tomography (PET) is one of the leading imaging techniques in both clinical and research settings.¹⁻⁴ It provides valuable functional information about a living subject, which helps the physicians and researchers for disease diagnosis, therapy monitoring, and drug evaluation. Among the PET radionuclides, fluorine-18 is undoubtedly the most popular isotope.^{4,5} It has a convenient half-life of 110 min, which allows multi-step syntheses and regional dose transportation. In addition, fluorine-18 has a clean decay profile (97% positron emission) and low β^+ decay energy (0.633 MeV) which is optimal for high resolution PET imaging.⁶

The traditional radiofluorination strategy involves electrophilic or nucleophilic substitution, with the latter being commonly used for decades.^{4,7-9} However, radiofluorination of electron rich arenes or heteroarenes remains challenging.^{10,11} In the past decade, numerous approaches have been developed to prepare fluorine-18 labeled arenes or heteroarenes, including iodonium salt,¹²⁻¹⁴ ylide,^{15,16} and transition metal mediated approaches.¹⁷⁻²⁴ Among these, the copper-mediated radiofluorination approach developed by Gouverneur group has gained a great deal of attention due to the tolerance of a wide range of functional groups on both electron-rich or electron-deficient substrates.²² By treating the pinacolyl arylboronate (arylBPi_n) precursor with Cu(OTf)₂py₄ and [¹⁸F]KF/K₂₂₂, a number of ¹⁸F-labeled arenes and heteroarenes were readily prepared.²² Further optimization of this efficient method has been performed by several research groups. For example, a “low-base” protocol was developed to efficiently produce PET tracers on a preparative scale.²⁵ Other substrates such as boronic acids and arylstannanes have also been successfully evaluated.^{23,26} However, the excess base and phase transfer agent (PTA) can significantly diminish the conversion, since the Cu catalyst possess low stability under basic conditions.^{23,25} It is still challenging to apply this method on scale-up applications and routine productions of PET tracers. Recently, Mossine *et al* reported the use of an organic base,

4-(dimethylamino)pyridine (DMAP)₂ and a customized elution technique for Cu-mediated radiofluorination with up to 58% radiochemical yield (RCY) based on radio-TLC.²⁷ However, the isolated yield is relatively low, possibly due to activity loss during elution and azeotropic drying. In another recent publication a Et₄N·OTf elution approach was developed for successful ¹⁸F-fluorodestannylation through the copper-mediated process.²⁸ Overall, Cu-mediated radiofluorination is a significant improvement to the library of radiofluorination methods, and its full potential is being discovered to benefit radiochemistry communities.

Our group recently developed a “Radiofluorination on the Sep-Pak” method to quickly and efficiently prepare 6-[¹⁸F]fluoronicotinic acid-2,3,5,6-tetrafluorophenyl ester, a prosthetic group for peptide/protein labeling, without the addition of base or azeotropic drying.^{29, 30} Considering the base-sensitivity of the Cu-mediated radiofluorination process, a similar ¹⁸F-elution process could significantly aid this transformation. In this work, [¹⁸F]fluoride trapped on the Sep-Pak was eluted with an anhydrous organic solution of pyridinium salt such as pyridinium trifluoromethanesulfonate (Py·OTf). This approach does not require azeotropic drying, as no water was used for the elution of fluoride. Moreover, the absence of base increases the stability of both precursor and the copper reagent leading to better radiochemical yield. In this paper, we report our findings for an improved Cu-mediated radiofluorination through a no-base-added, azeotropic drying-free ¹⁸F-elution process.

2. Results and Discussion

To begin our study, the elution of [¹⁸F]fluoride from PS-HCO₃ Sep-Pak (**Figure 1**) and subsequent Cu-mediated fluorination was first tested with Py·OTf. Quantitative elution of [¹⁸F]fluoride was obtained (**Table 1**, entry 1) by slowly passing a DMF solution of Py·OTf through the Sep-Pak containing [¹⁸F]fluoride at 0.5 mL/min speed. The eluted [¹⁸F]pyridinium fluoride solution was subjected to the model reaction of 4-[¹⁸F]fluoroacetophenone synthesis via Cu-mediated radiofluorination. However, no product was obtained. Addition of pyridine as

a co-eluent showed no effect on [^{18}F]fluoride elution or radiochemical conversion (entry 2). To further optimize the protocol, 4-(dimethylamino)pyridinium trifluoromethanesulfonate (DMAP \cdot OTf) was tested, as DMAP has been proven more efficient (10-fold higher efficiency) than pyridine in Cu-mediated radiofluorination.²⁷ DMAP was readily converted to the trifluoromethanesulfonic acid salt, DMAP \cdot OTf, by mixing DMAP with trifluoromethanesulfonic acid, and used without purification. The fluoride elution efficiency (95%) was comparable with Py \cdot OTf. More importantly, the desired 4- ^{18}F fluoroacetophenone (entry 3) was prepared with a radiochemical yield of 41%.

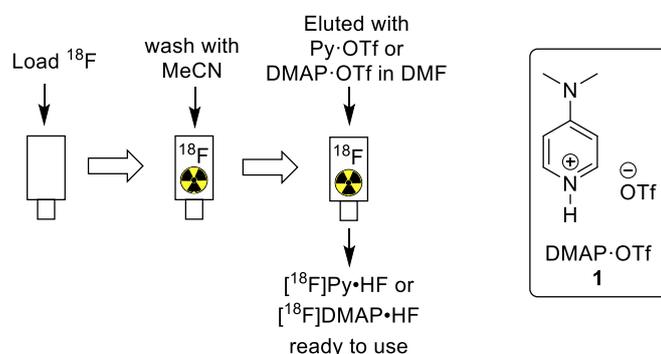
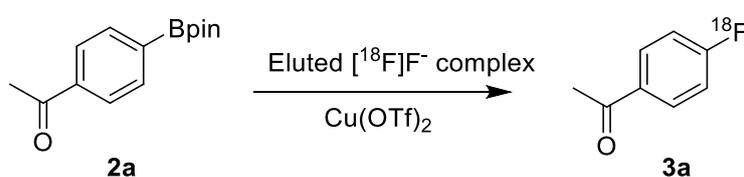


Figure 1. [^{18}F]fluoride elution approach from PS- HCO_3 cartridge.

This result prompted us to further optimize the radiosynthesis. Additional DMAP in the eluting solution was not productive, as it reduced the yield to 9% (entry 4). Increasing the reaction temperature to 120 $^\circ\text{C}$ improved the yield (entry 5). The effect of various amounts of DMAP \cdot OTf was also tested for [^{18}F]fluoride elution (entry 5-7). When 5 mg DMAP \cdot OTf was used, the elution efficiency was slightly decreased to 83% (entry 6). However, larger amounts of DMAP \cdot OTf (15 mg) reduced the radiochemical yield (entry 7). Therefore, we decided to perform the elution of [^{18}F]fluoride with 10 mg of DMAP \cdot OTf as this provided optimal balance between elution efficiency and yield. Various reaction temperatures were also evaluated, and slightly increased yield was observed at higher temperatures (entry 5 and entry 8). Further increasing the temperature to 130 $^\circ\text{C}$ had no effect on radiochemical yield (entry 9). Changing

the solvent to *N,N*-dimethylacetamide (DMA) similarly showed no effect on radiochemical yield (entry 10). Based on the model reaction, we settled on a reaction temperature of 120 °C in DMF or DMA as the optimal conditions for the following studies. Variation of reaction time was not investigated in this optimization. All reactions were heated for 20 min, since it was the standardized time for most related works.^{22, 25, 27, 31}

Table 1. Model reaction of Cu-mediated fluorination.^a



Entry	Eluting agent	Amount (mg)	Co-eluent	Solvent	Elution efficiency (%)	T (°C)	RCY ^b (%)
1	Py·OTf	10	n/a	DMF	>95	120	0
2	Py·OTf	10	Py ^c	DMF	>95	120	0
3	DMAP·OTf	10	n/a	DMF	>95	110	41
4	DMAP·OTf	10	DMAP ^d	DMF	>95	110	9
5	DMAP·OTf	10	n/a	DMF	>95	120	51 ± 2 (n = 3)
6	DMAP·OTf	5	n/a	DMF	83	120	49
7	DMAP·OTf	15	n/a	DMF	>95	120	43
8	DMAP·OTf	10	n/a	DMF	>95	105	40
9	DMAP·OTf	10	n/a	DMF	>95	130	51
10	DMAP·OTf	10	n/a	DMA	>95	120	53

Note: ^a 10 mg ArBpin precursor (40 μmol), 0.37 – 0.76 GBq of [¹⁸F]fluoride were used for each reaction. ^b RCY were determined by analytical HPLC (method A) and radio-TLC. ^c 10 μL of pyridine. ^d 5 mg of DMAP.

The optimized protocol was tested on a substrate scope study of various boronic esters (Figure 2). The selected substrates contain electron deficient (**3a-b**) and electron rich arenes (**3d**, **3g**, **3j-l**), as well as heteroarenes such as quinoline, indole and pyridine moieties (**3c**, **3e-f**, **3h-i**). Using the optimized conditions, medium to high RCY was obtained for most substrates

(up to 83%). The RCYs for few compounds such as **3c** and **3i** are significantly higher than previously reported, possibly due to the improved stability of the precursor under the no-base-added conditions.^{27, 31} The side-by-side comparison of their synthesis under literature “high-base” condition *vs.* this protocol was performed.^{25, 31} For **3c**, the literature method gave 31% RCY, whereas the RCY for the elution protocol was $83 \pm 2\%$. Similar result was obtained for **3i**, which gave <2% RCY for the literature method *vs.* 27% for the elution protocol. The solvent effect was studied for compound **3h**, since the reaction in DMF resulted in a low RCY (19%). Switching the solvent to DMA improved the yield to 40%. Similar yield difference in DMF *vs.* DMA was reported in the literature,^{11, 31} possibly due to the undesired reaction of precursor with dimethylamine (resulting from thermal decomposition of DMF). RCY was determined by analytical HPLC of the crude product and the identity of product was confirmed by co-injection with the standard compounds.

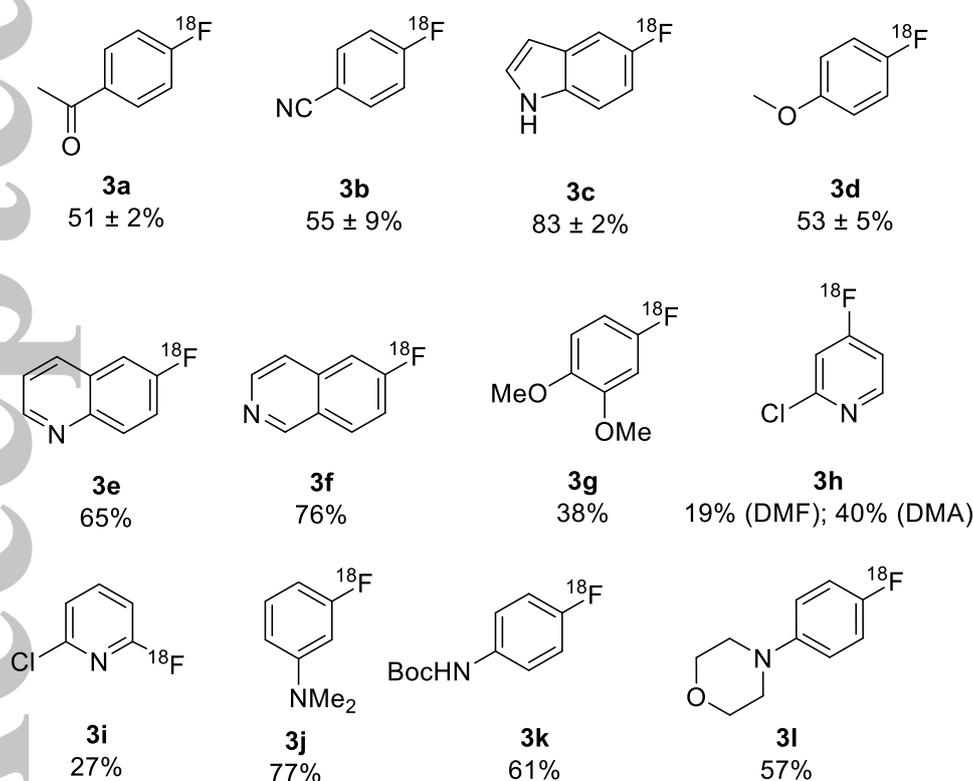
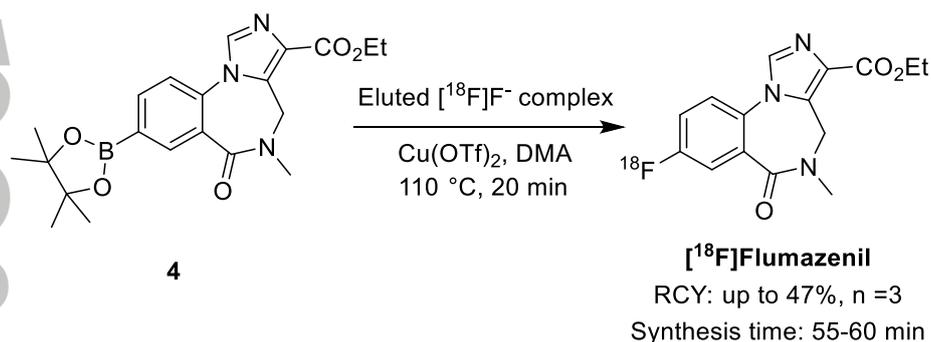


Figure 2. Substrate scope study. Reaction condition: precursor (10 mg), DMAP·OTf (10 mg), Cu(OTf)₂ (3 mg), DMF or DMA (1.3 mL), 120 °C, 20 min.

Fluorine-18 labeled flumazenil is a well-known tracer for quantitative evaluation of central benzodiazepine receptors (**Scheme 1**).^{32, 33} Recently, Preshlock *et al.* reported the radiosynthesis of this tracer via copper-mediated radiofluorination with a RCY of 19%.¹¹ In this study, we tested the labeling efficiency with the DMAP·OTf elution method. The reaction was optimized with low amounts of [¹⁸F]fluoride (0.37 – 0.76 GBq, 10-20 mCi) at varying temperatures, a 51% RCY was obtained at 110 °C with DMAP·OTf in DMA as eluent. When the radiosynthesis was performed on a production scale (6.7 – 7.4 GBq) with HPLC purification, [¹⁸F]flumazenil was successfully obtained with 26 – 47% RCY (isolated, n =3) in 55-60 min. The molar activity is 100 – 126 GBq/μmol at the end of synthesis. The identity of [¹⁸F]flumazenil was confirmed by co-injection with an authentic nonradioactive standard on the analytical HPLC (**Figure 3**).



Scheme 1. Radiosynthesis of [¹⁸F]flumazenil with DMAP·OTf elution method.

We believe our protocol increased the stability of both the precursor and the copper reagent. As a result, less mazenil pinacol boronate (**4**) and Cu(OTf)₂ were needed in the radiosynthesis, which will significantly reduce the cost of the overall procedure. In our improved [¹⁸F]flumazenil synthesis, 2 mg of mazenil pinacol boronate precursor and 3 mg of Cu(OTf)₂ were sufficient to give the purified product in 47% RCY. Low reagent loading and fewer side reactions should generally simplify the purification process, especially for reactions with base-sensitive precursors.

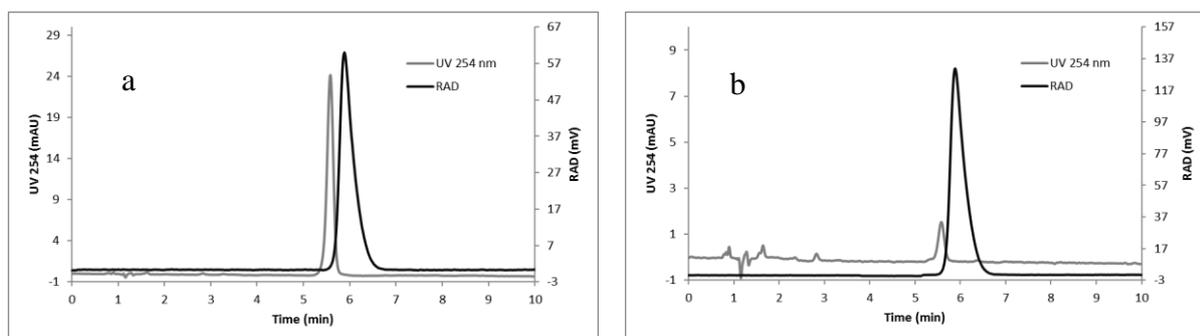


Figure 3. HPLC analysis of a) [^{18}F]flumazenil; b) [^{18}F]flumazenil, co-injected with the nonradioactive standard using Method B. black line, in-line radiodetector; gray line, UV detector at 254 nm.

3. Conclusions

Copper-mediated radiofluorination is an efficient method of incorporating fluorine-18 on both electron-rich and electron-deficient substrates. However, this method requires base to elute fluorine-18 from Sep-Pak and azeotropic drying. The current method requires neither the base to elute fluorine-18 from Sep-Pak nor azeotropic drying. Fluorine-18 eluted from the Sep-Pak with DMAP·OTF can be used directly for radiolabeling. This method will be beneficial for base sensitive precursors and labeled tracers. Moreover, stability of the copper reagent will be higher under these reaction conditions. The current method was successfully validated on 12 aryl- and heteroaryl-boronic ester substrates with a range of electronic properties. Using this approach, the radiosynthesis of [^{18}F]flumazenil was achieved with 47% isolated RCY in 55-60 min synthesis time. The total synthesis time is shorter than the literature reported method (75-80 min) due to no need of azeotropic drying of fluorine-18.³³ This highly efficient method will significantly boost the application scope of the powerful copper-mediated radiofluorination. Further evaluation of this method on other PET tracers is in progress.

4. Methods

4.1. General

Unless otherwise noted, all chemicals and solvents were purchased from Sigma-Aldrich (Milwaukee, WI, USA) or Combi-blocks (San Diego, CA, USA) and used without further purification. Flumazenil and its boronate precursor were purchased from ABX GmbH (Radeberg, Germany). Non-carrier added [^{18}F]fluoride was obtained from the National Institutes of Health cyclotron facility (Bethesda, MD, USA). Chromafix PS-HCO₃ anion-exchange Sep-Pak cartridges were purchased from Synthra (Hamburg, Germany) and the packing material was reduced to half (~20 mg) for better elution efficiency. Ultra-pure water was produced with the Milli-Q[®] Integral water purification system (Billerica, MA, USA). The Fusion 100 syringe pump was purchased from Chemyx (Stafford, TX, USA).

NMR data were recorded on a Bruker 400 MHz spectrometer (Billerica, MA, USA). High resolution mass spectrometry (HRMS) was carried out on an Agilent TOF mass spectrometer (Santa Clara, CA, USA). Radio-TLC was performed on an Eckert-Ziegler AR-2000 radio-TLC Imaging Scanner (Hopkinton, MA, USA). High performance liquid chromatography (HPLC) purification and analytical HPLC were conducted on the Agilent 1260 HPLC system equipped with multi-wavelength UV detector along with a flow count radiodetector (Eckert & Ziegler, B-FC-3500 diode).

HPLC conditions:

Method A: Phenomenex Luna C18 (2) column, 100×4.6 mm, 5 μm. Mobile phase: A: water (0.1% NH₄OH); B: acetonitrile (0.1% NH₄OH). Gradient: 20 - 80% B in 10 min; 1 mL/min.

Method B: Phenomenex Luna C18 (2) column, 100×4.6 mm, 5 μm. Mobile phase: 25% acetonitrile in water (0.1% formic acid); 1.0 mL/min.

Method C: Phenomenex Luna C18 (2) column, 250×10 mm, 5 μm. Mobile phase: 22% acetonitrile in water (25 mM NH₄OAc); 4 mL/min.

4.2. Chemical synthesis

4-Dimethylaminopyridinium trifluoromethanesulfonate (DMAP·OTf)

A solution of trifluoromethanesulfonic acid (735 mg, 4.9 mmol) in dichloromethane (10 mL) was cooled to 0 °C. A solution of 4-dimethylaminopyridine (598 mg, 0.49 mmol) in dichloromethane (5 mL) was added dropwise and the mixture was stirred for 15 min. The precipitated product was collected by vacuum filtration and washed with cold dichloromethane (10 mL). The product was further dried under high vacuum overnight (1.3 g, quantitative yield). No further purification was necessary. ¹H NMR (400 MHz, DMSO) δ 13.15 (s, 1H), 8.21 (d, *J* = 7.6 Hz, 2H), 6.98 (d, *J* = 7.7 Hz, 2H), 3.19 (s, 6H). ¹³C NMR (101 MHz, DMSO) δ 157.43, 139.56, 121.15 (d, *J* = 322.4 Hz), 107.41, 40.09. HRMS (ESI): Calcd for C₇H₁₀N₂ (M+H)⁺ 123.0922, found 123.0919.

4.3. Radiochemistry

General procedure for the Cu-mediated radiofluorination via DMAP·OTf elution

[¹⁸F]fluoride (0.37 – 0.76 GBq, 10 – 20 mCi) in target water was diluted with 2 mL water and passed through a short PS-HCO₃ cartridge (see Section 4.1). The cartridge was washed with anhydrous acetonitrile (6 mL) and dried for 2 min. The [¹⁸F]fluoride on the cartridge was eluted (0.5 mL/min, manually or via syringe pump) with a DMAP·OTf (10 mg) solution in DMF or DMA (1 mL) to a reaction vial containing the boronate precursor (10 mg) and Cu(OTf)₂ (3 mg). The cartridge was further eluted with DMF or DMA (0.3 mL) to the same vial. The reaction mixture was heated at 120 °C for 20 min, an aliquot (~50 μL) was diluted with water (50 μL) and analyzed by analytical HPLC (method A) and radio-TLC for RCY

determination. The identity of the product was confirmed by co-injection with unlabeled standard compound.

General procedure for the Cu-mediated radiofluorination via traditional azeotropic drying process for 3c and 3i

[¹⁸F]fluoride (0.37 – 0.76 GBq, 10 – 20 mCi) in target water was diluted with 2 mL water and passed through a short PS-HCO₃ cartridge (see Section 4.1). [¹⁸F]fluoride was eluted from the cartridge with the eluent (2.8 mg K₂CO₃, 12 mg K₂₂₂ in 1 mL acetonitrile and 280 μL water). The solution was azeotropically dried under vacuum and nitrogen flow at 110 °C. The azeotropic drying process was repeated by adding acetonitrile (1 mL × 3). To the dried activity was added the boronic ester precursor (60 μmol), tetrakis(pyridine)copper(II) triflate (Cu(OTf)₂(py)₄, 15 mg, 22 μmol) in DMF or DMA (0.3 mL). The resulting mixture was heated under 110 °C for 20 min. An aliquot (~50 μL) was diluted with water (50 μL) and analyzed by analytical HPLC (method A). The identity of the product was confirmed by co-injection with unlabeled standard compound.

Radiosynthesis of [¹⁸F]flumazenil

[¹⁸F]fluoride (6.7 GBq, 188 mCi) in target water was diluted with 2 mL water and passed through a short PS-HCO₃ cartridge. The cartridge was washed with anhydrous acetonitrile (6 mL) and dried for 2 min. The [¹⁸F]fluoride on the cartridge was eluted (0.5 mL/min) with a DMAP·OTf (10 mg) solution in DMA (1 mL) to a reaction vial containing mazenil pinacol boronate (2 mg) and Cu (OTf)₂ (3 mg). The cartridge was further eluted with DMA (0.3 mL) to the same vial. The reaction mixture was heated at 110 °C for 20 min, then diluted with aqueous NH₄OAc solution (10 mM, 2 mL), and purified by HPLC using semi-preparative column (method C). The product peak was collected between 26 – 28 min. The radiosynthesis

including [^{18}F]fluoride catch on the Sep-Pak, radiofluorination, and purification is completed in 55-60 min.

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