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Discovery and optimisation of potent, selective, ethanolamine inhibitors of bacterial phenylalanyl tRNA synthetase

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Abstract—High throughput screening of *Staphylococcus aureus* phenylalanyl tRNA synthetase (FRS) identified ethanolamine 1 as a sub-micromolar hit. Optimisation studies led to the enantiospecific lead 64, a single-figure nanomolar inhibitor. The inhibitor series shows selectivity with respect to the mammalian enzyme and the potential for broad spectrum bacterial FRS inhibition. © 2005 Elsevier Ltd. All rights reserved.

The continued spread of bacterial antibiotic resistance and its attendant threat to antimicrobial and general medicine, has been well-documented in medical and scientific publications. The rise of methicillin resistant S. aureus (MRSA) has received particular attention but antibiotic resistance in the major pathogens responsible for respiratory tract infections is also of concern. In the search for new molecular targets for novel antibacterial agents that would not be compromised by established resistance mechanisms, we have focused on bacterial aminoacyl tRNA synthetases. These enzymes couple proteinogenic amino acids to their cognate tRNA in an essential step in protein biosynthesis. The enzyme family is validated clinically by the topical antibiotic mupirocin (marketed as Bactroban®) which acts by inhibition of isoleucyl tRNA synthetase.

We targeted high throughput screening of the aminoacyl tRNA synthetases from *S. aureus*. For example, we have described the identification of a file compound hit against *S. aureus* methionyl tRNA synthetase and optimisation of the series to give highly potent inhibitors with excellent antibacterial activity.^{1–3} Here we describe the discovery and optimisation of a hit against *S. aureus* FRS. Others have also very recently reported the discov-

ery of new inhibitors of bacterial FRS by high throughput screening.^{4,5}

In a pioneering series of studies, Santi et al. characterised FRS from Escherichia coli⁶ and went on to identify phenethylamine derived inhibitors with potency down to 140 nM against the E. coli enzyme.7 Structurally, FRS is a large complex heterodimeric $\alpha_2\beta_2$ enzyme⁸ whose subunits are encoded by a linked transcriptional operon in S. aureus.9 Because of the complexity of the enzyme, the high throughput screening of the Smith-Kline Beecham compound bank was carried out with a crude preparation of FRS isolated from wild-type S. aureus Oxford. Subsequently, the enzyme from S. aureus WCUH29 was cloned, overexpressed in E. coli, purified and characterised.⁹ This recombinant enzyme was used for testing compounds in IC₅₀ assays using the full aminoacylation reaction with an SPA readout.10,11

The ethanolamine 1 was identified as potent hit in the high throughput screen. When tested in the standard aminoacylation assay, 1 was found to inhibit *S. aureus* FRS with an IC_{50} value of 160 nM. Further kinetic characterisation of the enzyme inhibition by 1 showed that its mechanism was best described as competitive with respect to phenylalanine but non-competitive with respect to ATP. Compound 1 thus formed an excellent starting point for a hit-to-lead optimisation programme.

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The most efficient preparation of arrays of analogues of 1 was by reductive alkylation of ethanolamines with aryl aldehydes, as shown in step (iii) of Scheme 1. The limiting factor for this chemistry was the availability of the ethanolamine starting materials 4. An array methodology for the synthesis of these compounds was developed as shown in steps (i) and (ii) of Scheme 1. In a one-pot process, in situ formation of the silylated cyanohydrins 3 was followed by LAH reduction and parallel normal phase chromatography to afford the ethanolamines 4 in moderate yields and good purity.



Scheme 1. Reagents and conditions: (i) TMSCN (1.1 equiv)/ZnI₂ (0.1 equiv)/Et₂O; (ii) LiAlH₄/Et₂O/ Δ ; (iii) R-PhCHO/NaCNBH₃ or resin CNBH₃/AcOH/MeOH.

Initial studies of the SAR of the right hand side phenyl ring showed that substitution was only tolerated at the *ortho* position. The enzyme assay results for *ortho* substituted analogues are reported in Table 1. The carbon-linked and nitrogen-linked substituents of compounds **6–8** resulted in a significant reduction in inhibition. However, several oxygen-linker substituents, such as those of compounds **12** and **13**, afforded a significant enhancement in potency.

Table 1. FRS inhibition of 3-(trifluoromethyl)phenyl ethanolamines



The role of the *meta*-trifluoromethyl group in the left hand side ring was investigated with a series of analogues containing an *ortho* alkoxy substituent in the right hand side ring, as shown in Table 2. It is clear from the positional isomers and di-substitution, that only *meta* substitution is favoured and that further substitution is highly detrimental.

Further analogues to identify the preferred *meta* substituent were then prepared (Table 3). Compact non-polar substituents, such as halogen, methyl and trifluoromethyl were preferred at this position. Larger (21) or polar (25) substituents resulted in a significant reduction in potency.

Amongst the compounds synthesised in the *meta*-substituted array was the bromothiophene analogue **34**. This had an FRS IC₅₀ value of 50 nM, a 3-fold improvement over the corresponding (trifluoromethyl)phenyl analogue. The consistent improvement in properties of the bromothiophene resulted in this becoming the preferred left hand side moiety.

Thus to enhance potency further, the SAR of the 2-alkoxy position was developed in the bromothiophene series. Several groups of analogues were prepared via array chemistry at the benzaldehyde level. Direct alkylations of salicylaldehyde (26) resulted in final compounds such

Table 2. FRS inhibition of (trifluoromethyl)phenyl ethanolamines

X H O R					
No.	Х	R	IC ₅₀ (nM) S. aureus FRS		
14	2-CF ₃	Н	>10,000		
11	3-CF ₃	Н	150		
15	$4-CF_3$	Н	>10,000		
16	2,5-Di-CF ₃	Н	>10,000		
17	3,5-Di-CF ₃	Ph	>10,000		

Table 3. FRS inhibition of 3-(substituted)phenyl ethanolamines

	Ç	OH H	O_R
No.	X	R	IC ₅₀ (nM) S. aureus FRS
11	CF ₃	Н	150
18	Br	Н	110
19	Me	Н	110
20	OCF_3	Н	190
21	OCH ₂ Ph	Η	>10,000
13	CF ₃	Ph	50
22	Cl	Ph	31
23	Ι	Ph	32
24	OMe	Ph	83
25	CH ₂ OH	Ph	570



Scheme 2. Reagents and conditions: (i) $Br(CH_2)_3CO_2Et/K_2CO_3/DMF/\Delta$ (96%); (ii) propiolactone/NaH/DMF/ Δ (35%); (iii) KOH/DMF (94%); (iv) RR'NH/EDC/HOAt/THF/DMF (50–70%).

as **44–48**. More remote diversity was introduced by amide arrays as shown in Scheme 2. The acid intermediates were prepared by bromoester alkylation and base hydrolysis (**30**), or by propiolactone alkylation (**29**), whilst **28** was commercially available. The acids were converted to amides by standard water-soluble carbodi-imide coupling methodology.

In contrast to the relatively tight SAR constraints elsewhere in the molecule, a wide variety of ortho-alkoxy substituents was tolerated as evidenced by many analogues with IC₅₀ values <50 nM in Table 4. A diverse range of functional groups is accommodated, including hydrogen bond donors and acceptors and heterocycles, as well as ionisable groups such as amine and carboxylic acid. Reduced activity was seen only for the phenoxy compound 35 and several of the analogues with a nitrogen atom separated by two carbons from the *ortho* oxygen (39, 40, 51, 52). The diverse functionality possible at this position allows tuning of the overall physicochemical properties of the inhibitors. The tolerance for such a range of polar functional groups suggests that in the enzyme-inhibitor complex, the alkoxy group is likely to be at least partially solvent-exposed on the protein surface.

Representative compounds from the series were tested for selectivity against mammalian FRS, isolated from rat liver. Very little inhibition of the mammalian enzyme was observed. For example, compound 47 had an IC₅₀ of 90,000 nM, resulting in a selectivity ratio of roughly 5000-fold in favour of the bacterial enzyme.

The racemic amine **61** was resolved by chiral HPLC on a Chiralpak AD column, affording the enantiomers **62** (faster eluting) and **63** (slower eluting) with 99.8% ee and 98.0% ee, respectively. The absolute configuration of the enantiomers was assigned by ab initio optical rotation.¹² The enantiomeric amines were taken through in the usual way to the (*R*)- and (*S*)-enantiomers of **38**, compounds **64** and **65**, respectively. When tested against *S. aureus* FRS, **64** had an IC₅₀ value of 3.7 nM whereas for **65** the IC₅₀ was >1000 nM. Thus the inhibition is

highly enantiospecific with a ratio of over 250-fold between the two isomers.



In view of the high affinity of the compounds for S. aureus FRS, we wished to test the spectrum of inhibition against FRS enzymes from organisms that are important pathogens in respiratory tract infections. The FRS enzymes from Streptococcus pneumoniae and Haemophilus influenzae, the key Gram-positive and Gram-negative RTI pathogens, were overexpressed and purified essentially as described for the S. aureus enzyme.⁹ Selected analogues were tested in assays against these enzymes and the results are shown in Table 5.13 The initial hit 1 did not inhibit the H. influenzae enzyme at the highest concentration tested. In contrast, the optimised compounds 48 and 64 showed a relatively balanced spectrum of inhibition across the three species. All three enzymes had a strong preference for the (R)stereochemistry. These results are encouraging for the potential of an RTI spectrum of FRS inhibition.

Despite potent inhibition of *S. aureus* FRS, none of these compounds showed whole cell antibacterial activity against the *S. aureus* organism, when tested at concentrations up to $64 \mu g/mL$. Although it is possible that the target potency needs improving still further, the extremely poor antibacterial activity seems likely to be due to issues of bacterial penetration or efflux. In artificial membrane assays, representative analogues from the series showed high permeability (data not shown). Thus bacterial efflux may be the major factor contributing to the large discrepancy between enzyme potency and antibacterial activity for the ethanolamine inhibitors.

Table 4. FRS inhibition of 4-bromo-2-thienyl ethanolamines



No.	R	IC ₅₀ (nM) S. aureus FRS
34	Me	50
35	Ph	120
36	CH ₂ OMe	35
37	CH ₂ CN	49
38	$(CH_2)_2OH$	8
39	$(CH_2)_2NH_2$	220
40	(CH ₂) ₂ phthalimido	250
41	(CH ₂) ₂ morpholino	18
42	Allyl	18
43	CH ₂ Ph	10
44	CH ₂ (4-MeSO ₂ Ph)	26
45	CH ₂ (2-pyridyl)	43
46	CH ₂ (3-pyridyl)	18
47	CH ₂ (4-pyridyl)	16
48	CH ₂ (2-benzimidazolyl)	26
49	CH ₂ CONH ₂	42
50	CH ₂ CONHEt	32
51	CH ₂ CONHCH ₂ CONHMe	120
52	CH ₂ CONMe(CH ₂) ₂ OH	110
53	(CH ₂) ₂ CO ₂ H	58
54	(CH ₂) ₂ CONHCH ₂ CONHMe	36
55	(CH ₂) ₂ CONMe(CH ₂) ₂ OH	25
56	(CH ₂) ₃ CO ₂ H	43
57	(CH ₂) ₃ CONHCH ₂ CONHMe	30
58	(CH ₂) ₃ CONMe(CH ₂) ₂ OH	26
59	(CH ₂) ₄ NH ₂	22
60	(CH ₂) ₄ phthalimido	69

Table 5. Inhibition of various bacterial FRS enzymes by selected analogues

	Stereochem.	FRS IC ₅₀ (nM)			
		Streptococcus aureus	Streptococcus pneumoniae	Haemophilus influenzae	
1	RS	160	ND	NI @ 300	
43	RS	10	280	1500	
48	RS	26	250	150	
64	R	3.7	190	56	
65	S	>1000	>1000	>1000	

ND-not done; NI-no inhibition.

In summary, a sub-micromolar HTS hit against *S. aur*eus FRS was identified and optimised to give a highly potent sub-10 nM enantiospecific inhibitor. Whilst the series was selective with respect to mammalian FRS and showed significant broad spectrum inhibition, useful antibacterial activity against *S. aureus* was not attained.

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References and notes

- Jarvest, R. L.; Berge, J. M.; Berry, V.; Boyd, H. F.; Brown, M. J.; Elder, J. S.; Forrest, A. K.; Fosberry, A. P.; Gentry, D. R.; Hibbs, M. J.; Jaworski, D. D.; O'Hanlon, P. J.; Pope, A. J.; Rittenhouse, S.; Sheppard, R. J.; Slater-Radosti, C.; Worby, A. J. Med. Chem. 2002, 45, 1959.
- Jarvest, R. L.; Berge, J. M.; Brown, M. J.; Brown, P.; Elder, J. S.; Forrest, A. K.; Houge-Frydrych, C. S. V.; O'Hanlon, P. J.; McNair, D. J.; Rittenhouse, S.; Sheppard, R. J. *Bioorg. Med. Chem. Lett.* 2003, 13, 665.
- Jarvest, R. L.; Berge, J. M.; Brown, P.; Houge-Frydrych, C. S. V.; O'Hanlon, P. J.; McNair, D. J.; Pope, A. J.; Rittenhouse, S. *Bioorg. Med. Chem. Lett.* 2003, 13, 1265.
- (a) Yu, X. Y.; Finn, J.; Hill, J. M.; Wang, Z. G.; Keith, D.; Silverman, J.; Oliver, N. *Bioorg. Med. Chem. Lett.* 2004, 14, 1339; (b) Yu, X. Y.; Finn, J.; Hill, J. M.; Wang, Z. G.; Keith, D.; Silverman, J.; Oliver, N. *Bioorg. Med. Chem. Lett.* 2004, 14, 1343.
- Beyer, D.; Kroll, H.-P.; Endermann, R.; Schiffer, G.; Siegel, S.; Bauser, M.; Pohlmann, J.; Brands, M.; Ziegelbauer, K.; Haebich, D.; Eymann, C.; Brötz-Oesterhelt, H. Antimicrob. Agents Chemother. 2004, 48, 525.
- (a) Santi, D. V.; Danenberg, P. V.; Satterly, P. Biochemistry 1971, 10, 4804; (b) Santi, D. V.; Danenberg, P. V. Biochemistry 1971, 10, 4813; (c) Santi, D. V.; Danenberg, P. V.; Montgomery, K. A. Biochemistry 1971, 10, 4821.
- (a) Anderson, R. T., Jr.; Santi, D. V. J. Med. Chem. 1976, 19, 1270; (b) Santi, D. V.; Cunnion, S. O.; Anderson, R. T., Jr.; Webster, R. W., Jr. J. Med. Chem. 1979, 22, 1260.
- Mosyak, L.; Reshetnikova, L.; Goldgur, Y.; Delarue, M.; Safro, M. G. *Nat. Struct. Biol.* **1995**, *2*, 537.
- Savopoulos, J. W.; Hibbs, M.; Jones, E. J.; Mensah, L.; Richardson, C.; Fosberry, A.; Downes, R.; Fox, S. G.; Brown, J. R.; Jenkins, O. *Protein Expr. Purif.* 2001, 21, 470.
- Macarrón, R.; Mensah, L.; Cid, C.; Carranza, C.; Benson, N.; Pope, A.; Diez, E. Anal. Biochem. 2000, 284, 183.
- 11. Assay conditions used $1 \times K_m$ [Phe], $10 \times K_m$ [ATP] and a large excess of tRNA.
- 12. Specific rotations were calculated for models of 2-amino-1-(4-chlorothien-2-yl)ethanol at the sodium D line using HF/SCF wavefunctions with the 6-31G* basis set (Cl was used in place of Br as fourth row elements are not supported in the current version of the Dalton program).¹⁴ Predicted specific rotations were averaged using Boltzmann statistics. Experimental values were measured in methanol (c = 0.004 mg/mL, 20 °C) at the sodium D line and are compared to the predicted values:

	[α] _D (calcd)	$[\alpha]_{\mathrm{D}}$ (exp)	$\begin{matrix} [\alpha]_{\rm D} \\ ({\rm calcd}-{\rm exp}) \end{matrix}$	
	R	S		R	S
62	+9	-9	+13	-4	-22
63	+9	-9	-14	+22	+5

Both the sign and magnitude of rotation are in good agreement for **62** as (*R*) and **63** as (*S*). The sign of the rotation predicted in this way is correctly predicted in about 95% of cases for small molecules, although the exact magnitude may be more variable.^{15,16} It is considered that the current configurational assignment is highly reliable. Moreover, this assignment is consistent with the sign of rotation of all literature aryl and heteroaryl ethanolamine enantiomers.

- 13. For each bacterial enzyme assay, the concentration of phenylalanine was equal to $K_{\rm m}$ for that enzyme, to allow comparison across the species.
- 14. 'Dalton', a molecular electronic structure program, Release 1.2, **2001**, written by T. Helgaker et al.
- 15. Polavarapu, P. L. Mol. Phys. 1997, 91, 551.
- Stephens, P. J.; Devlin, F. J.; Cheeseman, J. R.; Frisch, M. J. J. Phys. Chem. A 2001, 105, 5356.