

Design and Synthesis of 1,2,4-Oxadiazole Derivatives as Non-steroidal 5 α -Reductase Inhibitors

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The purpose of this study was to synthesize compounds in which the 1,2,4-oxadiazole moiety replaced the amide bond of ONO3805 and to evaluate its 5 α -reductase inhibitory activity as a potential benign prostatic hyperplasia therapeutic target. Four 1,2,4-oxadiazole derivatives, **1**, **2**, **8**, and **20**, were evaluated *in vitro* against 5 α -reductase of rat liver microsome. The prepared **1** and **2** possessed similar binding affinity (K_i) to that of ONO3805. Therefore, the use of 1,2,4-oxadiazole ring as surrogate of the amide bond in ONO3805 has a successful result in this study. It leads not only to enhance chemical stability but also to maintain meaningful inhibitory activity. The butyric acid moiety of these inhibitors is considered to play an important role in mimicking the phosphoric acid portion of coenzyme-NADPH in interacting with the active site of 5 α -reductase.

INTRODUCTION

Steroid 5 α -reductase is a membrane-associated enzyme, which catalyzes the reduction of testosterone to the biologically more active dihydrotestosterone (DHT) in the presence of NADPH.¹ Elevated DHT has been implicated as a causative factor of skin disorders and benign prostatic hyperplasia (BPH).² Therefore, inhibition of 5 α -reductase is an attractive target for therapeutic intervention in disorders associated with elevated levels of DHT such as BPH,^{3,4} acne,⁵ male pattern baldness,⁶ and hirsutism.⁷ A number of 5 α -reductase inhibitors have been reported, including both steroidal⁸ and nonsteroidal inhibitors.⁹ Starting our research program to find a new 5 α -reductase inhibitor, we focused our attention on the development of novel nonsteroidal compounds, especially ONO3805. According to the bioisosterism,¹⁰ some research groups have successfully replaced the ester or amide group of biologically active compounds by an 1,2,4-oxadiazole moiety to improve *in vivo* efficacy.¹¹ The intramolecular acid catalyzed the hydrolysis of the amide bond^{12,13} even though the hydrolysis exhibits a half-life of 7 years under neutral conditions and at room temperature.¹⁴ In addition, we found that the amide bond of ONO3805 was partially cleaved in a DMSO-d₆ solution at 37 °C in four months. Replacement of the amide bond with a 1,2,4-oxadiazole ring for increasing hydrolytic resistance,¹⁵ metabolic stability, and improving pharmacokinetic performance¹⁶ has been reported. Herein, derivatives with 1,2,4-oxadiazole surrogate

of the amide bond in ONO3805 as potential 5 α -reductase inhibitors were designed and synthesized (Fig. 1).

RESULTS AND DISCUSSION

The synthesis of target compound **1** is illustrated in

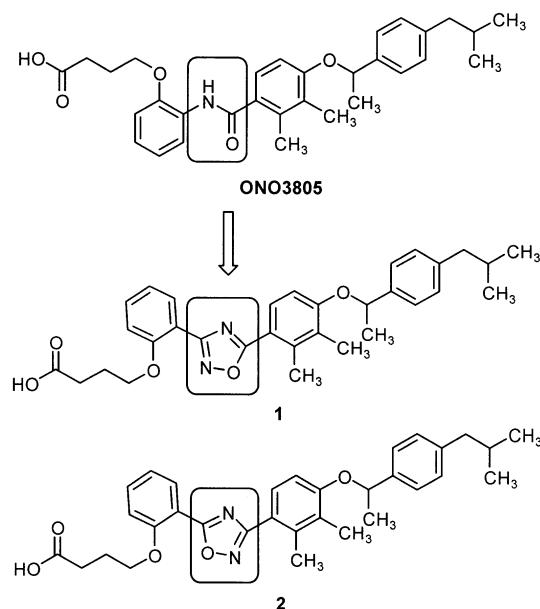
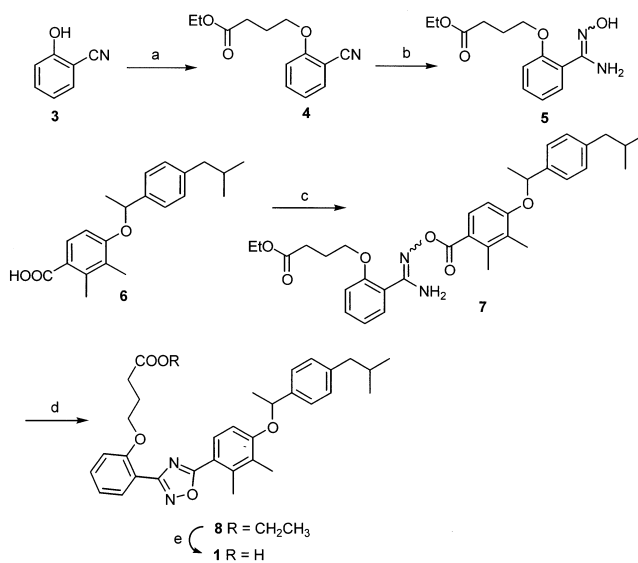


Fig. 1. The amide linkage of ONO3805 was replaced with 1,2,4-oxadiazole bioisostere.

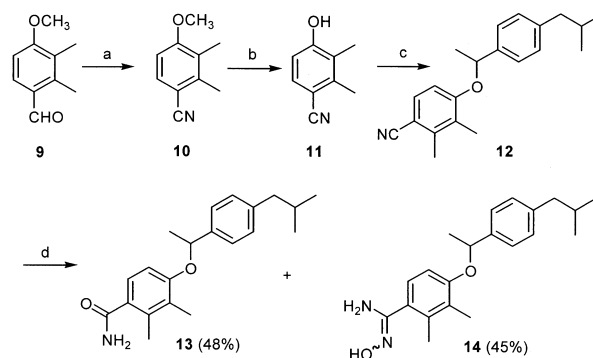
Scheme I. The coupling of substituted amidoxime **5** and 4-[1-(4-isobutylphenyl)ethoxy]-2,3-dimethylbenzoic acid (**6**) using 1,1'-carbonyldiimidazole in THF afforded **7**, which was followed by cyclization in diglyme at 110 °C to form the 1,2,4-oxadiazole derivative **8**.¹⁷ Finally, treatment of **8** with alkaline solution gave 89% yield of **1**. The required **5** was prepared from 2-cyanophenol by alkylation with ethyl γ -bromobutyrate and subsequent treatment with hydroxylamine. The other fragment **6** was prepared by a modified procedure from Harris.¹⁸ The oxadiazole isomer **2** was prepared by initial treatment of substituted amidoxime **14** and diester **16** with NaH in THF to afford ethyl butyrate **17** and a mixture containing **18** and **19**. Finally, hydrolysis of **17** and the mixture of **18** and **19** with aqueous lithium hydroxide afforded 1,2,4-oxadiazole **2** and **20**, respectively (Scheme III). The required amidoxime **14** was prepared as shown in Scheme II. Firstly, the treatment of 2,3-dimethyl-*p*-anisaldehyde with hydroxylamine-*o*-sulfonic acid in a mixed solvent of water and acetonitrile (1:1) to produce benzonitrile **10**,¹⁹ which was demethylation with BBr₃ to give **11**. Then, **11** coupled with 1-(4-isobutylphenyl)ethanol by Mitsunobu reaction to furnish benzonitrile **12**.²⁰ Subsequent treatment of **12** with hydroxylamine in Et₃N yielded amide **13** (48%) and amidoxime **14** (45%). The other fragment **16** was prepared from methyl salicylate by alkylation with ethyl γ -bromobutyrate.

The prepared 1,2,4-oxadiazole derivatives, **1**, **2**, **8**, and **20**, were subjected against 5 α -reductase inhibition assay (Table 1). The 1,2,4-oxadiazole isomers **1** and **2** possessed high

Scheme I

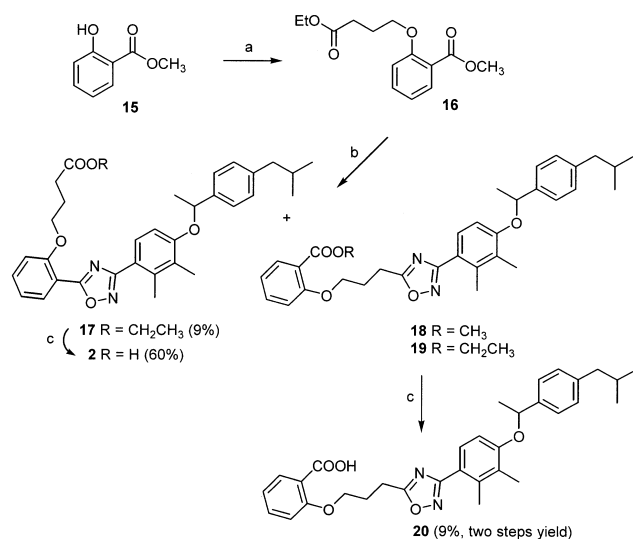


Scheme II



Reagents: (a) HOSA, CH₃CN, H₂O, 60 °C, 96%; (b) BBr₃, CH₂Cl₂, reflux, 98%; (c) 1-(4-isobutylphenyl)ethanol, Ph₃P, DEAD, THF, rt, 49%; (d) NH₂OH-HCl, Et₃N, absolute EtOH, reflux.

Scheme III



Reagents: (a) EtOOC(CH₂)₃Br, K₂CO₃, CH₃CN, reflux, 98%; (b) **14**, NaH, THF, 4 Å MS, 60 °C; (c) 1N LiOH, dioxane, 80 °C.

inhibitory activity to 5 α -reductase with K_i = 15.63 and 22.36 nM, respectively, which is comparable with that of ONO3805 (K_i = 8.46 nM). These results demonstrated that the 1,2,4-oxadiazole moiety is as useful as the amide bioisostere in ONO3805 chemical stability. Holt et al.²¹ proposed that the butanoic acid moiety of ONO3805 was localized in the region of the phosphate groups of NADPH, and the lipophilic part was oriented in the region of the steroidal rings C and D to occupy the hydrophobic pocket of 5 α -reductase. Interestingly, compound **8** lost activity against 5 α -reductase (K_i > 50 μ M). This result might indicate that **8** could not fit into the negative-ionizable region because the carboxylic function was masked by an ethyl group. Surprisingly, compound **20**

Table 1. Binding Affinity Against 5 α -Reductase of Rat Liver Microsome

Compounds	Binding Affinity K _i $\pm$ SD (nM) ^a
1	15.63 $\pm$ 9.69
2	22.36 $\pm$ 6.30
8	NA ^b
20	NA ^b
ONO3805	8.46 $\pm$ 5.92

^a Values are in four experiments.^b Not active at concentration of 50 μ M.

did not show any inhibitory activity. The reason might be that the benzoic acid moiety is unable to function as the butanoic acid moiety in **1** and **2** in fitting into the negative-ionizable region as proposed.

The initial result showed that the 1,2,4-oxadiazole bioisosteres, **1** and **2**, could successfully retain inhibitory activity of 5 α -reductase by comparison with the amide linkage of ONO3805. The oxadiazole **8** lost its activity when butanoic acid moiety was masked by an ethyl group, indicating that the carboxylic acid group was important to mimic the phosphoric acid portion of NADPH. Moreover, the 2-alkoxybenzoic acid moiety also could not fit the active site of 5 α -reductase. Therefore, the negative ionizable region of 5 α -reductase is considered to play an important role in its catalytic activity of reducing testosterone to DHT.

EXPERIMENTAL SECTION

All of the solvents and chemicals were reagent grade. Normal phase silica gels of 9385 (230–400 mesh) for column chromatography, and TLC plates of 573 (Si60 with F₂₅₄, 0.25 mm) were purchased from E. Merck A. G., Darmstadt, Germany. [4-¹⁴C]-Testosterone (2.10 GBq/mmol) was purchased from NEN Life Science Products, Inc., Boston, U.S.A. Melting points measured on a Buchi 510 melting point apparatus were uncorrected. IR spectra were recorded on a Jasco A-100 infrared spectrophotometer. ¹H- and ¹³C-NMR spectra were obtained on Bruker DPX-200 or Bruker AMX-400 spectrophotometers using the solvent peaks as reference standards. EIMS spectra were recorded on Finnigan Mat GCQTM GC/MS and HREIMS on JEOL JMS-HX-100 mass spectrometer. Elemental analyses were carried out on a Perkin-Elmer 240 elemental analyzer and the results were within $\pm 0.4\%$ of the theoretical values.

Ethyl 4-(2-cyanophenoxy)butyrate (**4**)

To a stirred solution of 2-cyanophenol (20 g, 0.17 mol) and K₂CO₃ (25.84 g, 0.19 mol) dissolved in dried acetonitrile (500 mL) was added γ -bromo-n-butyric acid ethyl ester (32.8 g, 0.17 mol) under nitrogen. The mixture was heated at reflux for 11 h, filtered, and concentrated in vacuo. The residue was subjected to chromatography (silica gel, n-hexane/ethyl acetate = 9/1, R_f = 0.19) to afford **4** (36.4 g, 92%) as a colorless oil. ¹H NMR (200 MHz, CDCl₃) δ 7.42–7.51 (m, 2H, ArH), 6.90–6.98 (m, 2H, ArH), 4.03–4.14 (m, 4H, CH₂), 2.52 (t, 2H, J = 7.2 Hz, COCH₂), 2.04–2.17 (m, 2H, CH₂), 1.20 (t, 3H, J = 7.1 Hz, CH₃); ¹³C NMR (50 MHz, CDCl₃) δ 172.9, 160.3, 134.2, 133.6, 120.7, 116.2, 112.1, 101.8, 69.5, 60.4, 30.2, 24.0, 14.0; MS (EI) *m/z* 233 (M⁺); Anal. Calcd. for C₁₃H₁₅NO₃: C, 66.94; H, 6.48; N, 6.00. Found: C, 66.88; H, 6.46; N, 5.98.

Ethyl 4-[2-[amino(hydroxyimino)methyl]phenoxy]butyrate (**5**)

To a stirred solution of hydroxylamine hydrochloride (2.98 g, 42.9 mmol) in absolute ethanol (250 mL) was added K₂CO₃ (5.92 g, 42.86 mmol) at room temperature under nitrogen for 10 min, and then added a solution of **4** (5.0 g, 21.4 mmol) in absolute ethanol (10 mL). The mixture was heated at reflux for 11 h and then evaporated in vacuo. The residue was subjected to chromatography (silica gel, CHCl₃, R_f = 0.15) to give **5** (2.69 g, 47%) as a colorless oil. ¹H NMR (200 MHz, CDCl₃) δ 7.61 (dd, 1H, J = 7.63 Hz, ArH), 7.26–7.31 (m, 1H, ArH), 6.87–6.98 (m, 2H, ArH), 5.37 (br s, 2H, NH₂), 4.01–4.16 (m, 4H, CH₂), 2.48 (t, 2H, J = 7.2 Hz, CH₂), 2.08–2.14 (m, 2H, CH₂), 1.21 (t, 3H, J = 7.1 Hz, CH₃); ¹³C NMR (50 MHz, CDCl₃) δ 173.9, 156.3, 152.2, 130.8, 129.7, 121.1, 120.9, 112.5, 67.6, 60.5, 30.7, 24.4, 14.1; MS (EI) *m/z* 266 (M⁺); Anal. Calcd. for C₁₃H₁₈N₂O₄: C, 58.64; H, 6.81; N, 10.52. Found: C, 58.60; H, 6.58; N, 10.30.

4-[2-[4-[1-(4-Isobutylphenyl)ethoxy]-2,3-dimethyl-N-benzoyloxy]carbamimidoylphenoxy]butanoic acid ethyl ester (**7**)

To a stirred solution of **6** (300 mg, 0.92 mmol) and 1,1'-carbonyldiimidazole (193.9 mg, 1.20 mmol) dissolved in dried THF (30 mL) was added **5** (318.2 mg, 1.19 mmol) at room temperature. The mixture was stirred for 18 h under nitrogen. After the mixture was reacted completely (indicated by phosphomolybdic acid on TLC plate displayed an orange spot), purified by chromatography (silica gel, n-hexane/ethyl acetate = 7/3, R_f = 0.17) to afford **7** which was recrystallized from ether to give a white solid (210 mg, 40%). mp 103–104

$^{\circ}\text{C}$; ^1H NMR (200 MHz, CDCl_3) δ 8.87 (dd, 1H, $J = 7.2$; 1.3 Hz, ArH), 6.89–7.54 (m, 8H, ArH), 6.59 (d, 1H, $J = 8.7$ Hz, ArH), 5.58 (br s, 2H, NH_2), 5.33 (q, 1H, $J = 6.3$ Hz, CH), 4.04–4.15 (m, 4H, OCH_2), 2.51 (s, 3H, CH_3), 2.41–2.47 (m, 4H, $-\text{COCH}_2$; Ar CH_2), 2.28 (s, 3H, CH_3), 2.05–2.18 (m, 2H, CH_2), 1.75–1.89 (m, 1H, CH), 1.63 (d, 3H, $J = 6.5$ Hz, CH_3), 1.20 (t, 3H, $J = 7.1$ Hz, CH_3), 0.87 (d, 6H, $J = 6.6$ Hz, CH_2); ^{13}C NMR (50 MHz, CDCl_3) δ 173.0, 165.8, 158.2, 156.7, 156.3, 141.0, 140.1, 139.4, 131.8, 130.8, 129.3, 128.3, 127.0, 125.1, 122.7, 121.3, 119.6, 112.5, 109.9, 109.6, 76.1, 67.8, 60.6, 45.1, 30.9, 30.1, 24.5, 24.5, 22.4, 17.2, 14.2, 12.2; MS (EI) m/z 574.5 (M^+); HRMS (EI) Calcd. for $\text{C}_{34}\text{H}_{42}\text{N}_2\text{O}_6$ (M^+): 574.3042. Found: 574.3035.

4-[2-[5-[4-[1-(4-Isobutylphenyl)ethoxy]-2,3-dimethylphenyl]-[1,2,4]oxadiazol-3-yl]phenoxy]butanoic acid ethyl ester (8)

A solution of **7** (220 mg, 0.38 mmol) dissolved in dried 2-methoxyethyl ether (diglyme, 10 mL) was heated at 110°C under nitrogen for 10 h. After cooling, the solution was concentrated in vacuo, purified by chromatography (silica gel, n-hexane/ethyl acetate = 9/1) to give **8** ($R_f = 0.27$), and followed by recrystallization from ethyl acetate to furnish a white solid (200 mg, 95%). mp $81.5\text{--}82.0^{\circ}\text{C}$; ^1H NMR (200 MHz, CDCl_3) δ 8.05 (dd, 1H, $J = 7.7$; 1.8 Hz, ArH), 7.76 (d, 1H, $J = 8.8$ Hz, ArH), 7.0–7.39 (m, 7H, ArH), 6.69 (d, 1H, $J = 8.8$ Hz, ArH), 5.37 (q, 1H, $J = 6.4$ Hz, CH), 4.04–4.20 (m, 4H, CH_2), 2.67 (s, 3H, CH_3), 2.64 (t, 2H, $J = 7.0$ Hz, $-\text{COCH}_2$), 2.43 (d, 2H, $J = 7.2$ Hz, Ar CH_2), 2.33 (s, 3H, CH_3), 2.10–2.27 (m, 2H, CH_2), 1.76–1.89 (m, 1H, CH), 1.65 (d, 3H, $J = 6.4$ Hz, CH_3), 1.20 (t, 3H, $J = 7.1$ Hz, CH_3), 0.87 (d, 6H, $J = 6.6$ Hz, CH_2); ^{13}C NMR (50 MHz, CDCl_3) δ 175.7, 173.4, 166.9, 158.5, 157.3, 141.0, 139.9, 138.9, 131.9, 131.3, 129.3, 128.9, 127.1, 125.1, 120.6, 116.7, 116.4, 112.7, 110.6, 76.1, 67.5, 60.3, 45.0, 30.5, 30.1, 24.4, 24.4, 22.3, 17.5, 14.2, 12.3; MS (EI) m/z 556.5 (M^+), 511, 396 (base peak); HRMS (EI) Calcd. for $\text{C}_{34}\text{H}_{40}\text{N}_2\text{O}_5$ (M^+): 556.2937. Found 556.2928.

4-[2-[5-[4-[1-(4-Isobutylphenyl)ethoxy]-2,3-dimethylphenyl]-[1,2,4]oxadiazol-3-yl]phenoxy]butanoic acid (1)

To a stirred suspension of **8** (500 mg, 0.89 mmol) in methanol (150 mL) was added 10% NaOH (15 mL) at room temperature. And the mixture was stirred for 5 days. After cooling in an ice-bath, the mixture was acidified with 6 N HCl to pH 3–4 under vigorous stirring, filtered, and washed with water. The solid was recrystallized from ethyl acetate to afford **1** as a white solid (408 mg, 89%). mp $143\text{--}144^{\circ}\text{C}$; ^1H NMR (200 MHz, CDCl_3) δ 8.03 (dd, 1H, $J = 7.7$ Hz, ArH), 7.73 (d, 1H, $J = 8.7$ Hz, ArH), 7.39–7.47 (m, 1H, ArH), 7.22–

7.26 (m, 2H, ArH), 6.99–7.11 (m, 4H, ArH), 6.68 (d, 1H, $J = 8.9$ Hz, ArH), 5.36 (q, 1H, $J = 6.1$ Hz, CH), 4.18 (t, 2H, $J = 5.8$ Hz, O-CH_2), 2.70 (t, 2H, $J = 7.1$ Hz, $-\text{COCH}_2$), 2.64 (s, 3H, CH_3), 2.42 (d, 2H, $J = 7.1$ Hz, Ar CH_2), 2.32 (s, 3H, CH_3), 2.10–2.22 (m, 2H, CH_2), 1.71–1.91 (m, 1H, CH), 1.64 (d, 3H, $J = 6.7$ Hz, CH_3), 0.86 (d, 6H, $J = 6.6$ Hz, CH_2); ^{13}C NMR (50 MHz, CDCl_3) δ 176.9, 176.0, 166.7, 158.7, 157.1, 141.1, 139.9, 139.0, 132.2, 131.4, 129.4, 129.0, 127.2, 125.1, 120.9, 116.4, 116.2, 112.6, 110.7, 76.4, 66.7, 45.1, 30.6, 30.1, 24.4, 24.4, 22.4, 17.5, 12.3; MS (EI) m/z 528.3 (M^+), 368.1, 161.1 (base peak); HRMS (EI) Calcd. for $\text{C}_{32}\text{H}_{36}\text{N}_2\text{O}_5$ (M^+): 528.2624. Found 528.2636.

2,3-Dimethyl-4-methoxybenzonitrile (10)

To a stirred solution of 2,3-dimethyl-p-anisaldehyde (5 g, 30.5 mmol) dissolved in acetonitrile (200 mL) was added dropwise an aqueous solution (200 mL) of hydroxylamine-sulfonic acid (5.17 mg, 45.7 mmol) at room temperature during 30 min period, and then the mixture was heated at 60°C for 18 h. After cooling, the mixture was neutralized with 10% NaOH and evaporated in vacuo. The residue was extracted with CHCl_3 (5×60 mL), dried over Na_2SO_4 , filtered, concentrated, chromatographed (silica gel, n-hexane/ethyl acetate = 96/4, $R_f = 0.32$), and recrystallized from acetonitrile to furnish **10** as a white solid (4.69 g, 96%). mp 40°C ; IR (KBr): 2950, 2840, 2225, 1590, 1580 cm^{-1} ; ^1H NMR (200 MHz, CD_3OD) δ 7.41 (d, 1H, $J = 8.6$ Hz, ArH), 6.71 (d, 1H, $J = 8.6$ Hz, ArH), 3.84 (s, 3H, OCH_3), 2.41 (s, 3H, CH_3), 2.11 (s, 3H, CH_3); ^{13}C NMR (50 MHz, CDCl_3) δ 160.5, 141.1, 131.4, 126.3, 119.3, 107.9, 104.7, 55.6, 18.1, 11.7; MS (EI) m/z 161 (M^+), 146; Anal. Calcd. for $\text{C}_{10}\text{H}_{11}\text{NO}$: C, 74.51; H, 6.88; N, 8.69. Found: C, 74.63; H, 7.16; N, 8.40.

2,3-Dimethyl-4-hydroxybenzonitrile (11)

To a stirred solution of **10** (4.69 g, 29.1 mmol) dissolved in dry dichloromethane (200 mL) was added BBr_3 (21.9 g, 87.3 mmol) at 0°C under nitrogen. The reaction mixture was warmed up to 50°C for 6 h, and then poured into crushed-ice while stirring vigorously. The resulting mixture was extracted with CHCl_3 (4×150 mL), dried over Na_2SO_4 , filtered, decoloured by active charcoal, and followed by recrystallization from CHCl_3 to give **11** as a white solid (4.21 g, 98%). mp 134°C ; IR (KBr): 3285, 2227 cm^{-1} ; ^1H NMR (400 MHz, CDCl_3) δ 7.32 (d, 1H, $J = 8.4$ Hz, ArH), 6.73 (d, 1H, $J = 8.4$ Hz, ArH), 6.19 (s, 1H, OH), 2.44 (s, 3H, CH_3), 2.17 (s, 3H, CH_3); ^{13}C NMR (50 MHz, CDCl_3) δ 157.8, 142.5, 131.3, 124.4, 119.3, 113.3, 104.4, 18.3, 11.7; MS (EI) m/z 147 (M^+); Anal. Calcd. for $\text{C}_9\text{H}_9\text{NO}$: C, 73.45; H, 6.16; N, 9.52. Found: C, 73.10; H, 6.34; N, 9.31.

2,3-Dimethyl-4-[1-(4-isobutylphenyl)ethoxy]benzonitrile (12)

To a stirred solution of **11** (300 mg, 2.04 mmol), 1-(4-isobutylphenyl)ethanol (472.8 mg, 2.65 mmol) and triphenyl phosphine (802.6 mg, 3.06 mmol) dissolved in THF (30 mL) was added diethylazodicarboxylate (532.9 mg, 3.06 mmol) at room temperature under nitrogen. After 8 h, the colour of solution was changed from colorless to blackish green, concentrated, triturated with ether, filtered, and then the filtrate was purified by chromatography (silica gel, n-hexane, $R_f = 0.1$), followed by recrystallization from n-hexane to give **12** as a white solid (305 mg, 49%). mp 66 °C; IR (KBr): 2960, 2930, 2875, 2225, 1590 cm^{-1} . ^1H NMR (200 MHz, CD_3OD) δ 7.09-7.27 (m, 5H, ArH), 6.60 (d, 1H, $J = 8.7$ Hz, CH), 5.33 (q, 1H, $J = 6.4$ Hz, CH), 2.44 (d, 2H, $J = 7.1$ Hz, CH_2), 2.44 (s, 3H, CH_3), 2.26 (s, 3H, CH_3), 1.77-1.91 (m, 1H, CH), 1.65 (d, 3H, $J = 6.4$ Hz, CH_3), 0.88 (d, 6H, $J = 6.6$ Hz, CH_3); ^{13}C NMR (50 MHz, CDCl_3) δ 158.9, 141.2, 141.1, 139.5, 131.0, 129.3, 126.8, 125.0, 119.2, 110.8, 104.5, 76.3, 45.0, 30.0, 24.3, 22.2, 18.2, 12.0; MS (EI) m/z 307 (M^+); Anal. Calcd. for $\text{C}_{21}\text{H}_{25}\text{NO}$: C, 82.04; H, 8.20; N, 4.56. Found: C, 81.71; H, 8.21; N, 4.51.

4-[1-(4-Isobutylphenyl)ethoxy]-2,3-dimethylbenzamide (13) and 4-[1-(4-Isobutylphenyl)ethoxy]-2,3-dimethylbenzamidoxime (14)

To a stirred solution of hydroxylamine hydrochloride (2.03 g, 29.28 mmol) and triethylamine (2.96 g, 29.28 mmol) dissolved in absolute ethanol (200 mL) was added **12** (0.90 g, 2.93 mmol). The mixture was heated at reflux for 8 days under nitrogen. The mixture was then evaporated in vacuo and the residue was subjected to chromatography (silica gel, n-hexane/ethyl acetate = 7/3). The fast-moving fraction was collected to give **13** ($R_f = 0.33$) which was recrystallized from CHCl_3 to afford a white solid (0.32 g, 48%). mp 125-126 °C; IR (KBr): 3383, 3193, 2920, 1706, 1646, 1596, 1464 cm^{-1} ; ^1H NMR (200 MHz, CD_3OD) δ 7.23 (d, 2H, $J = 8.1$ Hz, ArH), 7.07 (d, 2H, $J = 8.1$ Hz, ArH), 7.04 (d, 1H, $J = 8.5$ Hz, ArH), 6.64 (d, 1H, $J = 8.5$ Hz, ArH), 5.39 (q, 1H, $J = 6.4$ Hz, CH), 4.63 (br s, 2H, NH_2), 2.41 (d, 2H, $J = 7.2$ Hz, CH_2), 2.31 (s, 3H, CH_3), 2.25 (s, 3H, CH_3), 1.74-1.87 (m, 1H, CH), 1.60 (d, 3H, $J = 6.4$ Hz, CH_2), 0.86 (d, 6H, $J = 4.7$ Hz, CH_3); ^{13}C NMR (50 MHz, CD_3OD) δ 175.7, 157.1, 141.1, 140.9, 135.6, 129.3, 126.7, 125.5, 125.2, 110.6, 76.1, 45.1, 39.5, 23.8, 21.7, 16.1, 11.2; LCMS (ESI, acetic acid) m/z 326.2 (MH^+); Anal. Calcd. for $\text{C}_{21}\text{H}_{27}\text{NO}_2$: C, 77.50; H, 8.36; N, 4.30. Found: C, 77.55; H, 8.56; N, 4.24. The low-moving fraction was collected to give **14** ($R_f = 0.18$) which was recrystallized from acetonitrile to give a white solid (0.31 g, 45%). mp 140-141 °C; IR (KBr): 3490, 3381, 3250, 2927, 1640, 1595, 1494

cm^{-1} ; ^1H NMR (200 MHz, CD_3OD) δ 7.23 (d, 2H, $J = 8.0$ Hz, ArH), 7.07 (d, 2H, $J = 8.0$ Hz, ArH), 6.90 (d, 1H, $J = 8.5$ Hz, ArH), 6.55 (d, 1H, $J = 8.5$ Hz, ArH), 5.25 (q, 1H, $J = 6.4$ Hz, OCH), 4.69 (br, 2H, NH_2), 2.42 (d, 2H, $J = 7.1$ Hz, CH_2), 2.28 (s, 3H, CH_3), 1.88 (s, 3H, CH_3), 1.75-1.88 (m, 1H, CH), 1.55 (d, 3H, $J = 6.4$ Hz, CH_3), 0.86 (d, 3H, $J = 6.6$ Hz, CH_3); ^{13}C NMR (50 MHz, CDCl_3) δ 156.5, 153.8, 140.8, 140.4, 136.7, 129.3, 126.8, 126.4, 125.3, 125.2, 110.5, 76.0, 45.1, 30.2, 24.4, 22.4, 16.9, 12.2; MS (ESI, acetic acid) m/z 341.1 (MH^+); Anal. Calcd. for $\text{C}_{21}\text{H}_{28}\text{N}_2\text{O}_2$: C, 74.08; H, 8.29; N, 8.23. Found: C, 74.12; H, 8.10; N, 7.98.

2-(3-Ethoxycarbonylpropoxy)benzoic acid methyl ester (16)

To a stirred suspension of **15** (10 g, 65.72 mmol) and K_2CO_3 (18.2 g, 131.44 mmol) in dried acetonitrile (300 mL) was added γ -bromobutyric acid ethyl ester (12.8 g, 65.72 mmol) at reflux for 24 h under nitrogen. The hot mixture was filtered, washed with acetonitrile, and concentrated. The resulting residue was subjected to chromatography (silica gel, n-hexane/ethyl acetate = 95/5) to obtain **16** (17.1 g, 98%) as a yellowish oil. ^1H NMR (400 MHz, CDCl_3) δ 7.74 (dd, 1H, $J = 1.8, 7.62$ Hz, ArH), 7.37-7.41 (m, 1H, ArH), 6.90-6.94 (m, 2H, ArH), 4.09 (q, 2H, $J = 7.2$ Hz, CH_2), 4.05 (t, 2H, $J = 6.1$ Hz, CH_2), 3.84 (s, 3H, CH_3), 2.54 (t, 2H, $J = 7.3$ Hz, CH_2), 2.07-2.13 (m, 2H, CH_2), 1.21 (t, 3H, $J = 7.3$ Hz, CH_3); ^{13}C NMR (50 MHz, CDCl_3) δ 173.2, 166.7, 158.2, 133.3, 131.6, 120.2, 120.2, 113.1, 67.4, 60.3, 51.8, 30.4, 24.4, 14.1. Anal. Calcd. for $\text{C}_{14}\text{H}_{18}\text{O}_5$: C, 63.15; H, 6.81. Found: C, 63.30; H, 6.78.

4-[2-[3-[4-[1-(Isobutylphenyl)ethoxy]-2,3-dimethylphenyl]-[1,2,4]oxadiazol-5-yl]phenoxy]butyric acid (2) and 2-[3-[3-[4-[1-(Isobutylphenyl)ethoxy]-2,3-dimethylphenyl]-[1,2,4]oxadiazol-5-yl]propoxy]benzoic acid (20)

To a mixture solution of **14** (1 g, 2.94 mmol) dissolved in anhydrous THF (20 mL) containing 4 Å powdered molecular sieves (3 g) was stirred at room temperature for 30 min under nitrogen, and then sodium hydride was added (17.54 mg of 80% dispersion in oil, 3.23 mmol). The mixture was then heated at 60 °C for 1 h. After cooling, to the mixture was added a solution of **16** (2 g, 7.51 mmol) in THF (10 mL) and the resulting mixture was then heated at 60 °C for 3 h. The solution was evaporated in vacuo and the residue was subjected to chromatography (silica gel, n-hexane/ CHCl_3 = 6/4). The fast-moving band was collected to afford **17** ($R_f = 0.28$) as a yellowish oil (145 mg, 9%). IR (neat): 2955, 1734, 1595, 1546, 1464 cm^{-1} ; ^1H NMR (200 MHz, CD_3OD) δ 8.11 (dd, 1H, $J = 1.8; 8.0$ Hz, ArH), 7.62 (d, 1H, $J = 8.6$ Hz, ArH),

7.44-7.53 (m, 1H, ArH), 7.26 (d, 2H, $J = 6.2$ Hz, ArH), 7.01-7.11 (m, 4H, ArH), 6.68 (d, 1H, $J = 8.7$ Hz, ArH), 5.35 (q, 1H, $J = 6.3$ Hz, OCH), 4.04-4.20 (m, 4H, CH₂), 2.64 (t, 2H, $J = 7.2$ Hz, CH₂), 2.55 (s, 3H, CH₃), 2.43 (d, 2H, $J = 7.1$ Hz, CH₂), 2.32 (s, 3H, CH₃), 2.12-2.26 (m, 2H, CH₂), 1.74-1.89 (m, 1H, CH), 1.65 (d, 3H, $J = 6.3$ Hz, CH₃), 1.20 (t, 3H, $J = 7.1$ Hz, CH₃), 0.88 (d, 6H, $J = 6.6$ Hz, CH₃); ¹³C NMR (50 MHz, CDCl₃) δ 173.8, 173.2, 169.4, 157.7, 157.3, 140.9, 140.3, 137.9, 133.8, 131.6, 129.3, 128.5, 126.8, 125.2, 120.7, 119.2, 113.9, 113.0, 110.7, 76.0, 67.6, 60.3, 45.1, 30.5, 30.1, 24.5, 24.3, 22.4, 17.7, 14.2, 12.3. Compound **17** was used for the next reaction without further purification. Subsequently, to the solution of **17** (60 mg, 0.11 mmol) dissolved in dioxane (8 mL) was added 3 mL of 1 N LiOH at 80 °C for 30 min under vigorous stirring. After cooling, the mixture was acidified with 4 N HCl to pH 3-4, evaporated, and subjected to chromatography (silica gel, CHCl₃/CH₃OH = 97/3) to give **2** which was recrystallized from acetonitrile to afford a white solid (35 mg, 60%). mp 103-104 °C; IR (KBr): 2955, 1709, 1595, 1547, 1463, 1339, 1265 cm⁻¹; ¹H NMR (200 MHz, CD₃OD) δ 8.10 (dd, 1H, $J = 1.61$; 6.91 Hz, ArH), 7.61 (d, 1H, $J = 8.66$ Hz, ArH), 7.44-7.52 (m, 1H, ArH), 7.26 (d, 2H, $J = 7.4$ Hz, ArH), 6.99-7.10 (m, 4H, ArH), 6.69 (d, 1H, $J = 8.7$ Hz, ArH), 5.35 (q, 1H, $J = 6.3$ Hz, OCH), 4.17 (t, 2H, $J = 5.9$ Hz, OCH₂), 2.71 (t, 2H, $J = 7.1$ Hz, CH₂), 2.53 (s, 3H, CH₃), 2.42 (d, 2H, $J = 7.2$ Hz, CH₂), 2.31 (s, 3H, CH₃), 2.12-2.27 (m, 2H, CH₂), 1.76-1.89 (m, 1H, CH), 1.63 (d, 3H, $J = 6.3$ Hz, CH₃), 0.87 (d, 6H, $J = 6.6$ Hz, CH₃); ¹³C NMR (50 MHz, CDCl₃) δ 178.1, 173.8, 169.4, 157.6, 157.4, 140.9, 140.3, 137.9, 133.8, 131.6, 129.3, 128.5, 126.8, 125.3, 120.9, 119.2, 113.8, 113.0, 110.8, 76.0, 67.3, 45.1, 30.9, 30.1, 24.3, 24.3, 22.4, 17.7, 12.3; LCMS (ESI, Et₃N) m/z 527.3 (M-1), 441.3 (base peak); HRMS (ESI) Calcd. for C₃₂H₃₅N₂O₅ (M⁺): 528.2624. Found: 528.2618.

The slow-moving band was collected to give a mixture of **18** ($R_f = 0.15$) and **19** ($R_f = 0.17$) as a yellowish oil (0.7 g) and then it was hydrolyzed with 1 N LiOH for 1 h at 60 °C. After it was acidified and evaporated, the solution was eluted by a mixture of CHCl₃/CH₃OH (97/3) to afford **20** ($R_f = 0.14$) as a yellowish oil (132 mg, two steps gave 9% yield). IR (neat): 2954, 1694, 1600, 1460 cm⁻¹; ¹H NMR (200 MHz, CD₃OD) δ 8.13 (dd, 1H, $J = 1.7$; 7.8 Hz, ArH), 7.47-7.55 (m, 2H, ArH), 7.23 (d, 2H, $J = 8.0$ Hz, ArH), 7.0-7.14 (m, 4H, ArH), 6.65 (d, 1H, $J = 6.4$ Hz, ArH), 5.32 (q, 1H, $J = 6.4$ Hz, OCH), 4.37 (t, 2H, $J = 6.4$ Hz, OCH₂), 3.14 (t, 2H, $J = 7.1$ Hz, CH₂), 2.46 (s, 3H, CH₃), 2.42 (d, 2H, $J = 7.1$ Hz, CH₂), 2.36-2.53 (m, 2H, CH₂), 2.29 (s, 3H, CH₃), 1.75-1.88 (m, 1H, CH), 1.63 (d, 3H, $J = 6.4$ Hz, CH₃), 0.86 (d, 6H, $J = 6.6$ Hz, CH₃); ¹³C NMR (50 MHz, CDCl₃) δ 177.2, 169.9, 165.8,

158.0, 157.6, 141.4, 140.7, 138, 135.4, 134.2, 129.7, 128.9, 127.4, 125.6, 122.8, 119.0, 118.6, 113.1, 111.1, 110.0, 76.5, 68.8, 45.5, 30.6, 26.2, 24.8, 23.2, 22.8, 18.2, 12.8; LCMS (ESI, Et₃N) m/z 527.4 (M-1); HRMS (FAB) Calcd. for C₃₂H₃₇O₅N₂ (MH⁺): 529.2701. Found: 529.2702.

Rat 5 α -Reductase Inhibition Assay

The inhibition assay was performed according to that reported by Russell et al.²² with some minor modifications. Briefly, in 200 μ L of 20 mM sodium phosphate buffer (pH 6.5), containing 5 μ M [4-¹⁴C]-testosterone (2.10 GBq/mmol), 200 μ M of NADPH and 0.05% Tween-80, various amounts of the tested compound was added. The reaction was initiated with the addition of 20 μ g of rat liver microsome and incubated at 37 °C for 30 minutes. After incubation, the reaction was terminated with the addition of 1 mL of acetone. The supernatant was removed and dried. The residue was spotted on a TLC plate (silica gel) and developed with dichloromethane/ether 1:1. Radioactivities were recorded with a Bioscan γ -ray detector. The amount of product formed was calculated correspondingly.

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Testosterone; Dihydrotestosterone; 5 α -Reductase; Benign prostatic hyperplasia; Oxadiazole.

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