

A Synthetic Route to The Core Structure of (–)-Retigeranic Acid A

Xiao Wang, Dian Li, Junlin Zhang, Jianxian Gong, Junkai Fu,* and Zhen Yang*



Cite This: *Org. Lett.* 2021, 23, 5092–5097



Read Online

ACCESS |



Metrics & More

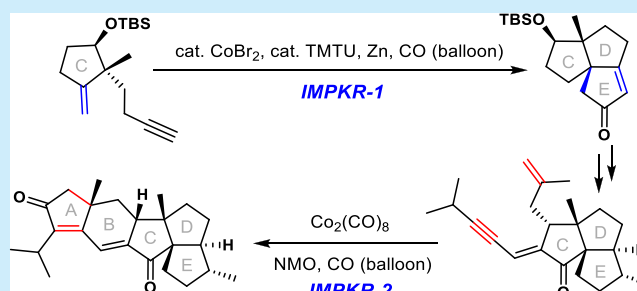


Article Recommendations



Supporting Information

ABSTRACT: Retigeranic acid A is a uniquely structured pentacyclic sesterterpene bearing eight stereogenic centers. We report a concise route to the core structure of (–)-retigeranic acid A. The stereochemistry of its six chiral centers and three quaternary carbon centers was well-controlled. This route features two intramolecular Pauson–Khand reactions (IMPKR): the first forged the D and E rings to deliver the triquinane subunit, and the second constructed the A and B rings and diastereoselectively installed the quaternary C_{6a} center.



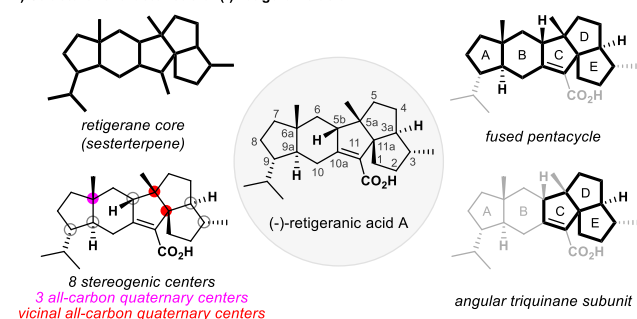
Retigeranic acid A, first isolated by Seshadri in 1965 from the lichens of the *Lobaria retigera* group found in the Himalayas,^{1a,b} is a structurally unique sesterterpene resulting from a remarkably intricate biosynthetic construction.^{1c}

Although the biological activity of retigeranic acid A has not been determined, the significant bioactivities of its parent genus *Lobaria* and the unique structure of this molecule have captured broad attention from synthetic chemists.^{2,3} Retigeranic acid A was the first isolated terpene containing an angular triquinane subunit;⁴ it possesses a unique fused pentacyclic skeleton with eight stereogenic centers, including three all-carbon quaternary centers, especially the vicinal quaternary centers at the bridgehead positions (Figure 1A).

Despite formidable challenges, unremitting efforts have been devoted to accessing this molecule and have assisted the confirmation of its structure. In 1985, Corey and co-workers⁵ developed a linear route to prepare the racemate of retigeranic acid using a Diels–Alder reaction, a [2 + 2] cycloaddition/ring expansion, and a ring contraction as the key steps. Shortly after that, Paquette,⁶ Hudlicky,⁷ and Wender⁸ independently reported convergent asymmetric syntheses of retigeranic acid A using different strategies, e.g., Claisen–Schmidt condensation/1,4-addition, vinylcyclopropane–cyclopentene rearrangement, or alkene–arene *meta* photocycloaddition/Diels–Alder reaction (Figure 1B). Considering the potential biological properties and the attractive structure, developing a supplementary approach to (–)-retigeranic acid A and its analogues is highly desirable for both drug discovery and academic research.

The intramolecular Pauson–Khand^{9a} reaction (IMPKR) is considered a powerful tool to construct cyclopentenone frameworks, which are useful synthetic handles for further manipulations in natural product synthesis.^{9b–h} Remarkable achievements by Schore,^{10a} Ishizaki,^{10b–d} Kerr,^{10e} Jiang,^{10f} Lovely,^{10g} Fox,^{10h} Mukai,¹⁰ⁱ and our group¹¹ have demonstrated that when 1,1-disubstituted alkenes are employed as

A) Structural characteristics of (–)-retigeranic acid A



B) Different strategies to the core skeleton of (–)-retigeranic acid A

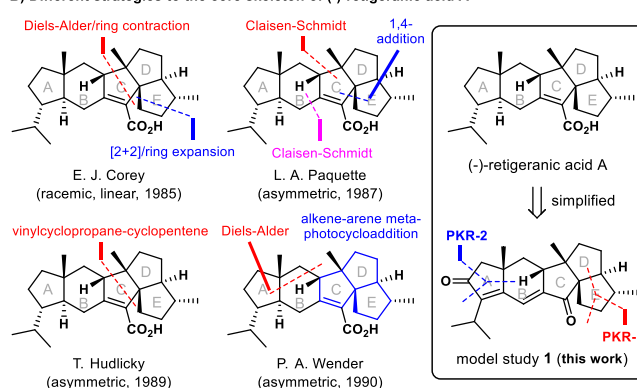
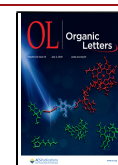


Figure 1. Structural characteristics and different strategies for the synthesis of (–)-retigeranic acid A.

Received: May 13, 2021

Published: June 15, 2021

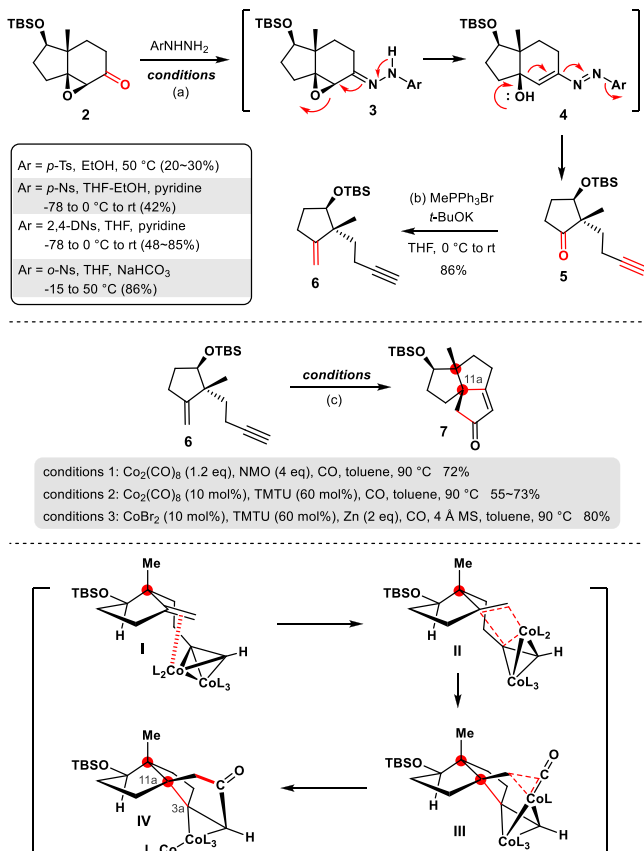


reaction partners, cyclopentenones with a quaternary carbon can be formed.

Herein we report a synthetic study toward (–)-retigeranic acid A, in which two IMPKRs were used as the key steps to forge the complex pentacyclic core structure and the bridged quaternary carbons in a diastereoselective manner.

Our study commenced with known chiral ketone compound **2**¹² (Scheme 1). Cyclic epoxy ketone hydrazone **3**, which was

Scheme 1. Evaluation of the First IMPKR^{4a}



^aReagents and conditions: (a) entry 1: *p*-TsNHNH₂ (1.05 equiv), EtOH, 50 °C, 1 h, 20–30%; entry 2: *p*-NsNHNH₂ (1.1 equiv), pyridine (1.3 equiv), THF/EtOH, –78 to 0 °C to rt, 4 h, 42%; entry 3: 2,4-DNsNHNH₂ (1.05 equiv), pyridine (1.3 equiv), THF, –78 to 0 °C, then rt, 18 h, 48–85%; entry 4: *o*-NsNHNH₂ (1.0 equiv), NaHCO₃ (3 equiv), THF, –15 to 50 °C, 36 h, 86%; (b) MePPh₃Br (1.3 equiv), *t*-BuOK (1.3 equiv), THF, 0 °C to rt, 24 h, 86%; (c) conditions 1: Co₂(CO)₈ (1.2 equiv), NMO (4.0 equiv), CO, toluene, 90 °C, 8 h, 72%; conditions 2: Co₂(CO)₈ (10 mol %), TMTU (60 mol %), CO, toluene, 90 °C, 48 h, 55–73%; conditions 3: CoBr₂ (10 mol %), TMTU (60 mol %), Zn (2 equiv), 4 Å MS, CO, toluene, 90 °C, 18 h, 80%. TBS = *tert*-butyldimethylsilyl, *p*-Ts = *p*-toluenesulfonyl, *p*-Ns = *p*-nitrobenzenesulfonyl, 2,4-DNs = 2,4-dinitrobenzenesulfonyl, *o*-Ns = *o*-nitrobenzenesulfonyl, NMO = 4-methylmorpholine *N*-oxide, TMTU = tetramethylthiourea, MS = molecular sieves.

prepared in situ from the condensation of ketone **2** and the aromatic sulfonylhydrazide, could undergo an Eschenmoser–Tanabe fragmentation¹³ to deliver acyclic alkynone **5**. Preliminary investigation by variation of the sulfonylhydrazide,¹⁴ solvent, and base showed that a yield of 86% could be obtained when the reaction was carried out with *o*-nitrobenzenesulfonylhydrazide¹⁵ and sodium bicarbonate in THF. A Wittig olefination of ynone **5** produced enyne compound **6**.

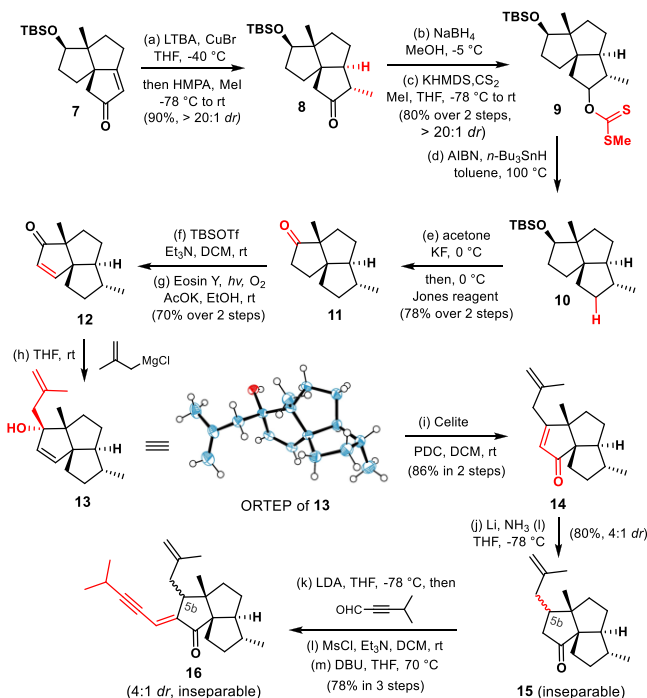
We then tested different conditions for the first IMPKR. Treatment of enyne **6** with stoichiometric Co₂(CO)₈ and NMO under a CO atmosphere resulted in the formation of fused tricyclic compound **7** in 72% yield as a single diastereomer. However, Co₂(CO)₈ is expensive, and this reaction required high-quality NMO. The use of hygroscopic NMO dramatically decreased the efficiency of the reaction.

We therefore explored alternative ways to prepare this key intermediate. Initially, we tested the annulation reaction using a catalytic amount of Co₂(CO)₈ and TMTU under a CO atmosphere,^{11a} but the product yield was affected by the quality of the Co₂(CO)₈, and the yield of **7** was inconsistent, ranging from 55 to 73%. We finally found out that product **7** could be obtained in 80% yield by treatment of air- and moisture-stable CoBr₂ with zinc dust (2.0 equiv) in the presence of TMTU.^{11b} It is noteworthy that the stereochemistry at the C_{11a} position matched that of (–)-retigeranic acid A. A conformational analysis I → IV showed that the enyne–cobalt complex got close to the alkene moiety from below and eventually forged the C_{3a}–C_{11a} bond with excellent diastereoselectivity.

1,4-Reduction of conjugated carbonyl compound **7** with copper hydride using lithium tri-*tert*-butoxyaluminum hydride (LTBA)¹⁶ as the hydrogen source followed by trapping of the resultant enolate anion with iodomethane afforded ketone **8** in 90% yield with excellent diastereoselectivity (Scheme 2). Reduction of the ketone into an alcohol and subsequent removal of the hydroxyl group through a Barton–McCombie radical deoxygenation¹⁷ delivered silyl ether **10**.

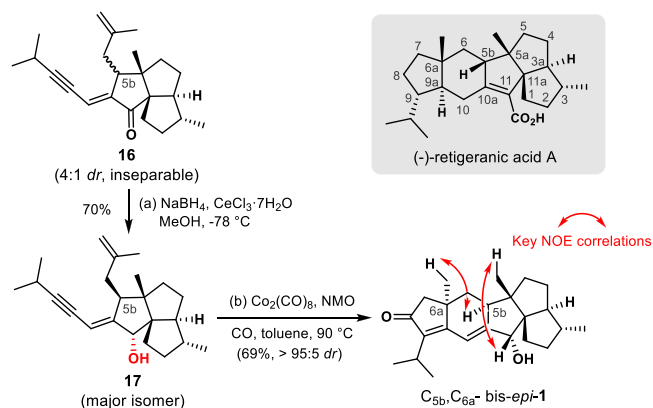
The subsequent desilylation/oxidation sequence¹⁸ transformed **10** into ketone compound **11**. Conversion of the ketone into an enol silyl ether followed by a visible-light-promoted organocatalytic aerobic oxidation developed by our group¹⁹ generated cyclopentenone **12** in 70% yield over two steps. A Grignard reaction of **12** using 2-methylallylmagnesium chloride gave allylic alcohol **13**, and its structure was established by X-ray crystallographic analysis. Pyridinium dichromate (PDC)-mediated oxidative rearrangement²⁰ of allylic alcohol **13** afforded enone **14** in 86% yield over two steps. The next step was a selective reduction of the enone into a ketone in the presence of the terminal alkene moiety, which proved to be difficult because the terminal alkene was easily reduced under many reductive conditions. Eventually, this was realized by treatment of enone **14** with Li/NH₃(l) at –78 °C, and ketone **15** was obtained as an inseparable mixture of diastereoisomers in a ratio of 4:1 and a combined yield of 80%. Subsequent installation of the alkyne moiety at the α-position of the ketone was achieved through a three-step reaction sequence, specifically, aldol addition, activation of the resultant alcohol with methanesulfonyl chloride, and base-promoted β-elimination of the newly generated mesylate, delivering enyne **16** in 78% yield.

Enyne compound **16** was an inseparable mixture of diastereoisomers. After a Luche reduction, allylic alcohol **17** was isolated as the major isomer in 70% yield, but the stereochemistry at C_{5b} could not be identified (Scheme 3). Subjecting **17** to IMPKR conditions successfully produced a fused pentacyclic compound in a highly diastereoselective manner. However, this proved to be C_{5b}C_{6a}-bis-*epi*-1 after a careful analysis of the ¹H–¹H NOESY spectra, meaning that the stereochemistry at C_{5b} of the major diastereoisomer in the mixed enyne **16** was the opposite of that in (–)-retigeranic acid A.

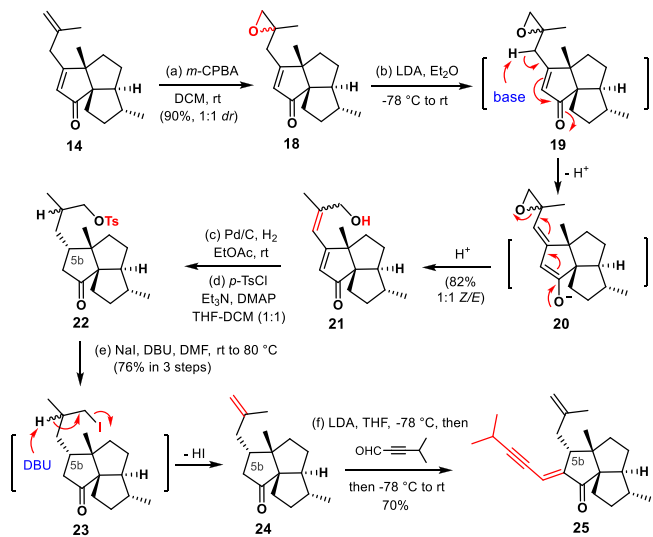
Scheme 2. Synthesis of the Enyne Precursor for the Second IMPKR^a

^aReagents and conditions: (a) CuBr (2.0 equiv), LTBA (1.5 equiv), THF, -40 °C, 2 h, then -78 °C, HMPA (9.0 equiv), MeI (9.0 equiv), -78 °C to rt, 24 h, 92%; (b) NaBH₄ (1.1 equiv), MeOH, -5 °C, 2 h; (c) KHMDS (1.2 equiv), CS₂ (1.2 equiv), THF, -78 °C to rt, 2 h, then MeI (2.4 equiv), 0 °C to rt, 2 h, 89% over two steps; (d) AIBN (0.1 equiv), *n*-Bu₃SnH (2.0 equiv), toluene, 100 °C, 8 h; (e) acetone, KF (8.0 equiv), Jones reagent (5.0 equiv, 2.5 M in 4 M H₂SO₄), 0 °C, 12 h, 78% over two steps; (f) TBSOTf (1.5 equiv), Et₃N (3 equiv), DCM, 0 °C to rt, 8 h; (g) Eosin Y (0.05 equiv), AcOK (1.2 equiv), EtOH, O₂, hv, 8 h, 70% over two steps; (h) 2-methylallylmagnesium chloride (1.5 equiv), THF, 12 h; (i) PDC (3.0 equiv), Celite (1.5 g per mmol of 15), DCM, 12 h, 86% over two steps; (j) Li (20.0 equiv), NH₃ (1 mL per mmol of Li), THF, -78 °C, 5 h, 80%, 4:1 dr; (k) LDA (1.5 equiv), THF, -78 °C, 2 h, then 4-methylpent-2-ynal (3 equiv), -78 °C; (l) MsCl (2 equiv), Et₃N (4 equiv), DCM, 4 h; (m) DBU (4 equiv), THF, 70 °C, 78% over three steps, 4:1 dr. LTBA = lithium tri-*tert*-butoxyaluminum hydride, HMPA = hexamethylphosphoramide, KHMDS = potassium bis-(trimethylsilyl)amide, AIBN = 2,2'-azobis(2-methylpropionitrile), Tf = trifluoromethanesulfonyl, PDC = pyridinium dichromate, LDA = lithium diisopropylamide, MsCl = methanesulfonyl chloride, DBU = 1,8-diazabicyclo[5.4.0]undec-7-ene.

To access the enyne precursor of the IMPKR with the correct stereochemistry at C_{5b}, an alternative route was developed. As shown in Scheme 4, selective epoxidation of the terminal olefin in enone 14 with 3-chloroperoxybenzoic acid (*m*-CPBA) afforded 18 in 90% yield with 1:1 dr. Treatment of epoxide 18 with LDA resulted in the formation of conjugated ketone 21 with a 1:1 *Z/E* ratio.²¹ This reaction was proposed to proceed through deprotonation at the γ -position of the α,β -unsaturated ketone to form conjugated enolate anion intermediate 20, which then underwent ring opening of its epoxide to generate 21. Hydrogenation of the carbon-carbon double bonds followed by activation of the hydroxyl group with *p*-toluenesulfonyl chloride delivered ketone 22. To our delight, the stereochemistry at C_{5b} of 22 was well-controlled and matched that of (-)-retigeranic acid

Scheme 3. Preliminary Evaluation of the Second IMPKR Using the C_{5b}-*epi*-Enyne Precursor^a

^aReagents and conditions: (a) NaBH₄ (1.1 equiv), CeCl₃·7H₂O (1.0 equiv), MeOH, -78 °C, 3 h, 70%; (b) Co₂(CO)₈ (1.2 equiv), NMO (4.0 equiv), CO, toluene, 90 °C, 8 h, 69%.

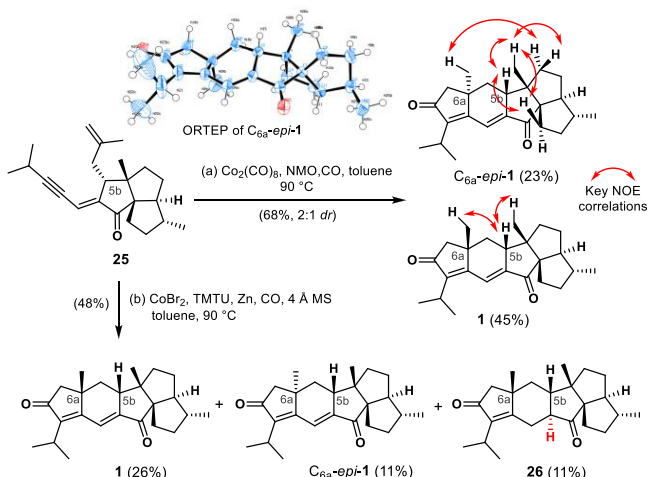
Scheme 4. Synthesis of the Enyne Precursor with Correct Stereochemistry at C_{5b} for the Second IMPKR^a

^aReagents and conditions: (a) *m*-CPBA (2.0 equiv), DCM, rt, overnight, 90%; (b) LDA (2.6 equiv), Et₂O, -78 °C to rt, 2 h, 82%; (c) Pd/C (20 wt %), H₂ (5–8 MPa), EtOAc, 4 h; (d) *p*-TsCl (3.0 equiv), DMAP (3.0 equiv), Et₃N (3.0 equiv), THF/DCM (1:1), overnight; (e) NaI (5.0 equiv), DBU (3.0 equiv), DMF, rt to 80 °C, 6 h, 76% over three steps; (f) LDA (1.5 equiv), THF, -78 °C, 1 h, then 4-methylpent-2-ynal (4.0 equiv), -78 °C to rt, 2 h, 70%. *m*-CPBA = 3-chloroperoxybenzoic acid, DMAP = 4-dimethylaminopyridine.

A. Regeneration of the alkene moiety was then achieved by treatment of 22 with NaI and DBU, which went through replacement of *p*-toluenesulfonate with the iodine anion and base-promoted H–I elimination of intermediate 23.²² The formed alkene 24 underwent an aldol condensation with 4-methylpent-2-ynal in the presence of LDA as the base to afford enyne compound 25 in 70% yield with the correct stereochemistry at C_{5b}.

With enyne 25 in hand, we investigated the second IMPKR (Scheme 5). Two representative sets of conditions were tested.

The IMPKR performed with stoichiometric Co₂(CO)₈ and NMO under a CO atmosphere delivered the desired pentacyclic compound 1 with the correct stereochemistry at

Scheme 5. Evaluation of the Second IMPKR^a

^aReagents and conditions: (a) $\text{Co}_2(\text{CO})_8$ (1.2 equiv), NMO (4.0 equiv), CO, toluene, 90 °C, 18 h, 68%; (b) CoBr_2 (20 mol %), TMTU (120 mol %), Zn (2 equiv), 4 Å MS, CO, toluene, 90 °C, 36 h, 48%.

both C_{5b} and C_{6a} in 45% yield. The C_{6a} epimer of **1** was generated as a minor product (23%). The structure of C_{6a} -*epi*-**1** was further confirmed by X-ray crystallographic analysis, while the stereochemistry of C_{6a} quaternary center of **1** was defined through ^1H – ^1H NOESY spectral analysis.

Treatment of enyne **25** with alternative conditions^{11b} using CoBr_2 (20 mol %), TMTU (1.2 equiv), and zinc dust (2 equiv) under a CO atmosphere, which proved to be the best choice for the first IMPKR, gave inferior yields of both **1** (26%) and C_{6a} -*epi*-**1** (11%). During this reaction, an unexpected reduction product **26** featuring a *trans*-fused [6,5] ring system was obtained in 11% yield, which was probably generated through further reduction of **1** with a small amount of in situ-formed [CoH] species.²³

In summary, the core structure of (–)-retigeranic acid **A** has been synthesized with six well-defined stereogenic centers, all of which matched with those of the target molecule. This synthetic route features two IMPKRs. The application of the first IMPKR forged the D and E rings to afford the triquinane subunit in a highly diastereoselective manner, while the second one constructed the A and B rings and efficiently installed the C_{6a} quaternary center with the correct stereochemistry. Additionally, the Eschenmoser–Tanabe fragmentation, the configuration inversion at C_{5b} , and the diastereoselective 1,4-reduction/methylation sequence are also crucial to the success of this route. Further studies toward the total synthesis of retigeranic acids are currently underway in our laboratory.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.orglett.1c01633>.

Experimental details and characterization data; Tables S1 and S2; Figures S1 and S2; copies of ^1H and ^{13}C NMR spectra; copies and assignments of COSY, HSQC, HMBC, and NOESY spectra; crystallographic data for compound **13** and C_{6a} -*epi*-**1** (PDF)

Accession Codes

CCDC 2074440 and 2074441 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, U.K.; fax: +44 1223 336033.

■ AUTHOR INFORMATION

Corresponding Authors

Junkai Fu – Jilin Province Key Laboratory of Organic Functional Molecular Design & Synthesis, Department of Chemistry, Northeast Normal University, Changchun, Jilin 130024, China; orcid.org/0000-0002-7714-8818; Email: fujk109@nenu.edu.cn

Zhen Yang – State Key Laboratory of Chemical Oncogenomics and Key Laboratory of Chemical Genomics, Peking University Shenzhen Graduate School, Shenzhen, Guangdong 518055, China; Beijing National Laboratory for Molecular Science and Key Laboratory of Bioorganic Chemistry and Molecular Engineering of the Ministry of Education, College of Chemistry and Molecular Engineering and Peking–Tsinghua Center for Life Sciences, Peking University, Beijing 100871, China; Shenzhen Bay Laboratory, Shenzhen, Guangdong 518055, China; orcid.org/0000-0001-8036-934X; Email: zyang@pku.edu.cn

Authors

Xiao Wang – State Key Laboratory of Chemical Oncogenomics and Key Laboratory of Chemical Genomics, Peking University Shenzhen Graduate School, Shenzhen, Guangdong 518055, China

Dian Li – State Key Laboratory of Chemical Oncogenomics and Key Laboratory of Chemical Genomics, Peking University Shenzhen Graduate School, Shenzhen, Guangdong 518055, China

Junlin Zhang – State Key Laboratory of Chemical Oncogenomics and Key Laboratory of Chemical Genomics, Peking University Shenzhen Graduate School, Shenzhen, Guangdong 518055, China

Jianxian Gong – State Key Laboratory of Chemical Oncogenomics and Key Laboratory of Chemical Genomics, Peking University Shenzhen Graduate School, Shenzhen, Guangdong 518055, China

Complete contact information is available at: <https://pubs.acs.org/doi/10.1021/acs.orglett.1c01633>

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

This work was supported by the National Key Research and Development Program of China (Grant 2018YFC0310905), the National Natural Science Foundation of China (Grants 21632002, 21702012, and 21871012), the Guangdong Natural Science Foundation (Grant 2010B03030700002), and the Shenzhen Basic Research Program (Grant 2019SHIBS0004).

■ REFERENCES

- (1) (a) Rao, P. S.; Sarma, K. G.; Seshadri, T. R. Chemical components of the *Lobaria* lichens from the Western Himalayas. *Curr. Sci.* 1965, 34, 9. (b) Rao, P. S.; Sarma, K. G.; Seshadri, T. R. Chemical

components of *Lobaria subsidiosa*, *L. retigera* and *L. subretigera* from the Western Himalayas. *Curr. Sci.* **1966**, 35, 147. (c) Kaneda, M.; Takahashi, R.; Iitaka, Y.; Shibata, S. Retigeranic acid, a novel sesterterpene isolated from the lichens of *lobaria retigera* group. *Tetrahedron Lett.* **1972**, 13, 4609.

(2) (a) Yoshimura, I. The genus *Lobaria* of eastern Asia. *J. Hattori Bot. Lab.* **1971**, 34, 231. (b) Molnár, K.; Farkas, E. Current results on biological activities of lichen secondary metabolites: a review. *Z. Naturforsch., C: J. Biosci.* **2010**, 65, 157. (c) Suzuki, M. T.; Parrot, D.; Berg, G.; Grube, M.; Tomasi, S. Lichens as natural sources of biotechnologically relevant bacteria. *Appl. Microbiol. Biotechnol.* **2016**, 100, 583. (d) Calcott, M. J.; Ackerley, D. F.; Knight, A.; Keyzers, R. A.; Owen, J. G. Secondary metabolism in the lichen symbiosis. *Chem. Soc. Rev.* **2018**, 47, 1730. (e) Cimmino, A.; Nimis, P. L.; Masi, M.; De Gara, L.; van Otterlo, W. A. L.; Kiss, R.; Evidente, A.; Lefranc, F. Have lichenized fungi delivered promising anticancer small molecules? *Phytochem. Rev.* **2019**, 18, 1.

(3) (a) Attah-Poku, S. K.; Chau, F.; Yadav, V. K.; Fallis, A. G. Intramolecular routes to hydrindanes: the Diels-Alder and Michael aldol approach to 6-isopropyl-9-methylbicyclo[4.3.0]nonan-3-one. *J. Org. Chem.* **1985**, 50, 3418. (b) Llera, J. M.; Fraser-Reid, B. An expeditious route to the northern part of retigeranic acid A from (R)-(-)-carvone. *J. Org. Chem.* **1989**, 54, 5544. (c) Hog, D. T.; Mayer, P.; Trauner, D. A unified approach to *trans*-hydrindane sesterterpenoids. *J. Org. Chem.* **2012**, 77, 5838. (d) Zhang, J.; Wang, X.; Li, S.; Li, D.; Liu, S.; Lan, Y.; Gong, J.; Yang, Z. Diastereoselective synthesis of cyclopentanoids: applications to the construction of the ABCD tetracyclic core of retigeranic acid A. *Chem. - Eur. J.* **2015**, 21, 12596. (e) Breitler, S.; Han, Y.; Corey, E. J. Highly diastereoselective cationic cyclization reactions convert a common monocyclic enone to bicyclic precursors for the synthesis of retigeranic acids A and B. *Org. Lett.* **2017**, 19, 6686.

(4) (a) Mehta, G.; Srikrishna, A. Synthesis of polyquinane natural products: an update. *Chem. Rev.* **1997**, 97, 671. (b) Le Bideau, F.; Kousara, M.; Chen, L.; Wei, L.; Dumas, F. Tricyclic sesquiterpenes from marine origin. *Chem. Rev.* **2017**, 117, 6110.

(5) Corey, E. J.; Desai, M. C.; Engler, T. A. Total synthesis of (±)-retigeranic acid. *J. Am. Chem. Soc.* **1985**, 107, 4339.

(6) (a) Paquette, L. A.; Wright, J.; Drtina, G. J.; Roberts, R. A. Enantiospecific total synthesis of natural (-)-retigeranic acid A and two (-)-retigeranic acid B candidates. *J. Org. Chem.* **1987**, 52, 2960. (b) Wright, J.; Drtina, G. J.; Roberts, R. A.; Paquette, L. A. A convergent synthesis of triquinane sesterterpenes. Enantioselective synthesis of (-)-retigeranic acid A. *J. Am. Chem. Soc.* **1988**, 110, 5806.

(7) Hudlicky, T.; Fleming, A.; Radesca, L. The [2 + 3] and [3 + 4] annulation of enones. Enantiocontrolled total synthesis of (-)-retigeranic acid. *J. Am. Chem. Soc.* **1989**, 111, 6691.

(8) Wender, P. A.; Singh, S. K. Synthetic studies on arene-olefin cycloadditions. 11. Total synthesis of (-)-retigeranic acid. *Tetrahedron Lett.* **1990**, 31, 2517.

(9) (a) Khand, I. U.; Knox, G. R.; Pauson, P. L.; Watts, W. E.; Foreman, M. I. Organocobalt complexes. Part II. Reaction of acetylenehexacarbonyl-dicobalt complexes, $(R_1C_2R_2)Co_2(CO)_6$, with norbornene and its derivatives. *J. Chem. Soc., Perkin Trans. 1* **1973**, 977. (b) Blanco-Urgoiti, J.; Añorbe, L.; Pérez-Serrano, L.; Domínguez, G.; Pérez-Castells, J. The Pauson-Khand reaction, a powerful synthetic tool for the synthesis of complex molecules. *Chem. Soc. Rev.* **2004**, 33, 32. (c) Shibata, T. Recent advances in the catalytic Pauson-Khand-type reaction. *Adv. Synth. Catal.* **2006**, 348, 2328. (d) Lee, H.-W.; Kwong, F.-Y. A decade of advancements in Pauson-Khand-type reactions. *Eur. J. Org. Chem.* **2010**, 2010, 789. (e) Simeonov, S. P.; Nunes, J. P. M.; Guerra, K.; Kurteva, V. B.; Afonso, C. A. M. Synthesis of chiral cyclopentenones. *Chem. Rev.* **2016**, 116, 5744. (f) Ricker, J. D.; Geary, L. M. Recent advances in the Pauson-Khand reaction. *Top. Catal.* **2017**, 60, 609. (g) Chen, S.; Jiang, C.; Zheng, N.; Yang, Z.; Shi, L. Evolution of Pauson-Khand reaction: strategic applications in total syntheses of architecturally complex natural products (2016–2020). *Catalysts* **2020**, 10, 1199.

(h) Yang, Z. Navigating the Pauson-Khand reaction in total syntheses of complex natural products. *Acc. Chem. Res.* **2021**, 54, 556.

(10) (a) Knudsen, M. J.; Schore, N. E. Synthesis of the angularly fused triquinane skeleton via intramolecular organometallic cyclization. *J. Org. Chem.* **1984**, 49, 5025. (b) Ishizaki, M.; Iwahara, K.; Kyomura, K.; Hoshino, O. An effective synthesis of angular tricyclic compounds having two contiguous quaternary centers by intramolecular Pauson-Khand reaction of exocyclic enynes. *Synlett* **1999**, 1999, 587. (c) Ishizaki, M.; Iwahara, K.; Niimi, Y.; Satoh, H.; Hoshino, O. Investigation of the synthesis of angular tricyclic compounds by intramolecular Pauson-Khand reaction of exo- and endo-cyclic enynes. *Tetrahedron* **2001**, 57, 2729. (d) Ishizaki, M.; Niimi, Y.; Hoshino, O. Formal total synthesis of (±)-magellanine. *Tetrahedron Lett.* **2003**, 44, 6029. (e) Kerr, W. J.; McLaughlin, M.; Morrison, A. J.; Pauson, P. L. Formal total synthesis of (±)-α- and β-cedrene by preparation of cedrone. Construction of the tricyclic carbon skeleton by the use of a highly efficient intramolecular Khand annulation. *Org. Lett.* **2001**, 3, 2945. (f) Jiang, B.; Xu, M. Highly enantioselective construction of fused pyrrolidine systems that contain a quaternary stereocenter: concise formal synthesis of (+)-conessine. *Angew. Chem., Int. Ed.* **2004**, 43, 2543. (g) Madu, C. E.; Lovely, C. J. A Pauson-Khand Approach to the Hamigerans. *Org. Lett.* **2007**, 9, 4697. (h) Pallerla, M. K.; Fox, J. M. Enantioselective synthesis of (-)-pentalenene. *Org. Lett.* **2007**, 9, 5625. (i) Inagaki, F.; Kinebuchi, M.; Miyakoshi, N.; Mukai, C. Formal synthesis of (+)-nakadomarin A. *Org. Lett.* **2010**, 12, 1800.

(11) (a) Tang, Y.; Deng, L.; Zhang, Y.; Dong, G.; Chen, J.; Yang, Z. Tetramethyl thiourea/ $Co_2(CO)_8$ -catalyzed Pauson-Khand reaction under balloon pressure of CO. *Org. Lett.* **2005**, 7, 593. (b) Wang, Y.; Xu, L.; Yu, R.; Chen, J.; Yang, Z. $CoBr_2$ -TMTU-Zinc-catalysed Pauson-Khand reaction. *Chem. Commun.* **2012**, 48, 8183. (c) Xiao, Q.; Ren, W.-W.; Chen, Z.-X.; Sun, T.-W.; Li, Y.; Ye, Q.-D.; Gong, J.-X.; Meng, F.-K.; You, L.; Liu, Y.-F.; Zhao, M.-Z.; Xu, L.-M.; Shan, Z.-H.; Shi, Y.; Tang, Y.-F.; Chen, J.-H.; Yang, Z. Diastereoselective total synthesis of (±)-schindilactone A. *Angew. Chem., Int. Ed.* **2011**, 50, 7373. (d) Liu, Q.; Yue, G.; Wu, N.; Lin, G.; Li, Y.; Quan, J.; Li, C.-C.; Wang, G.; Yang, Z. Total synthesis of (±)-pentalenolactone A methyl ester. *Angew. Chem., Int. Ed.* **2012**, 51, 12072. (e) Shi, L.-L.; Shen, H.-J.; Fang, L.-C.; Huang, J.; Li, C.-C.; Yang, Z. Development of an expedient intramolecular Pauson-Khand reaction approach to stereoselectively construct the *trans*-decalin with a C1 quaternary chiral center. *Chem. Commun.* **2013**, 49, 8806. (f) Huang, J.; Fang, L.; Long, R.; Shi, L.-L.; Shen, H.-J.; Li, C.-C.; Yang, Z. Asymmetric total synthesis of (+)-fusarisetin A via the intramolecular Pauson-Khand reaction. *Org. Lett.* **2013**, 15, 4018.

(12) Enev, V. S.; Petrov, O. S.; Neh, H.; Nickisch, K. Studies towards a total synthesis of the antiprogesterone onapristone. *Tetrahedron* **1997**, 53, 13709.

(13) For a review, see: (a) Olson, J. A.; Shea, K. M. Critical perspective: named reactions discovered and developed by women. *Acc. Chem. Res.* **2011**, 44, 311. Also see: (b) Drahl, M. A.; Manpadi, M.; Williams, L. J. C-C Fragmentation: origins and recent applications. *Angew. Chem., Int. Ed.* **2013**, 52, 11222. (c) Eschenmoser, A.; Felix, D.; Ohloff, G. Eine neuartige fragmentierung cyclischer alpha beta-ungesättigter carbonylsysteme - synthese von exalton und rac-muscon aus cyclododecanon. *Helv. Chim. Acta* **1967**, 50, 708. (d) Tanabe, M.; Crowe, D. F.; Dehn, R. L. A novel fragmentation reaction of α,β-epoxyketones the synthesis of acetylenic ketones. *Tetrahedron Lett.* **1967**, 8, 3943.

(14) (a) Corey, E. J.; Sachdev, H. S. 2,4-Dinitrobenzenesulfonylhydrazine, a useful reagent for Eschenmoser α,β-cleavage of α,β-epoxy ketones - conformational control of halolactonization. *J. Org. Chem.* **1975**, 40, 579. (b) Thiemann, T.; Noltemeyer, M.; De Meijere, A. Chiral building blocks from (+)-(S)-7,7a-dihydro-7a-methylindane-1,5(6H)-dione (Hajos-Parrish diketone). *Rep. Inst. Adv. Mater. Study Kyushu Univ.* **1997**, 11, 147. (c) Mander, L. N.; McLachlan, M. M. The total synthesis of the galbulimima alkaloid GB 13. *J. Am. Chem. Soc.* **2003**, 125, 2400.

(15) Myers, A. G.; Zheng, B.; Movassaghi, M. Preparation of the reagent *o*-nitrobenzenesulfonylhydrazide. *J. Org. Chem.* **1997**, *62*, 7507.

(16) (a) Masamune, S.; Rossy, P. A.; Bates, G. S. Reductive removal of halo and mesyloxy groups with a copper(I) complex. *J. Am. Chem. Soc.* **1973**, *95*, 6452. For selected reviews, see: (b) Semmelhack, M. F.; Stauffer, R. D.; Yamashita, A. Reductions of conjugated carbonyl compounds with copper hydride - preparative and mechanistic aspects. *J. Org. Chem.* **1977**, *42*, 3180.

(17) (a) Barton, D. H. R.; McCombie, S. W. New method for deoxygenation of secondary alcohols. *J. Chem. Soc., Perkin Trans. 1* **1975**, 1574. (b) Studer, A.; Amrein, S. Tin hydride substitutes in reductive radical chain reactions. *Synthesis* **2002**, 2002, 835. (c) Heravi, M. M.; Bakhtiari, A.; Faghihi, Z. Applications of Barton-McCombie reaction in total syntheses. *Curr. Org. Synth.* **2014**, *11*, 787.

(18) (a) Liu, H.-J.; Han, I.-S. Direct conversion of silyl ethers into carbonyl compounds with Jones reagent in the presence of potassium fluoride. *Synth. Commun.* **1985**, *15*, 759. (b) McKinney, A. R.; Ridley, D. D.; Turner, P. Equine metabolites of norethandrolone: synthesis of a series of 19-nor-17 α -pregnanediols and 19-nor-17 α -pregnanetriols. *Aust. J. Chem.* **2003**, *56*, 829.

(19) Zhang, J.; Wang, L.; Liu, Q.; Yang, Z.; Huang, Y. Synthesis of α , β -unsaturated carbonyl compounds via a visible-light-promoted organocatalytic aerobic oxidation. *Chem. Commun.* **2013**, *49*, 11662.

(20) (a) Babler, J. H.; Coghlan, M. J. Facile method for bis-homologation of ketones to α,β -unsaturated aldehydes - application to synthesis of cyclohexanoid components of Boll-Weevil sex attractant. *Synth. Commun.* **1976**, *6*, 469. (b) Dauben, W. G.; Michno, D. M. Direct oxidation of tertiary allylic alcohols - simple and effective method for alkylative carbonyl transposition. *J. Org. Chem.* **1977**, *42*, 682. (c) Killoran, P. M.; Rossington, S. B.; Wilkinson, J. A.; Hadfield, J. A. Expanding the scope of the Babler-Dauben oxidation: 1,3-oxidative transposition of secondary allylic alcohols. *Tetrahedron Lett.* **2016**, *57*, 3954.

(21) (a) Lemin, A. J.; Djerassi, C. The conversion of diosgenin to cortisone via 11-ketosteroids of the 5β -series. *J. Am. Chem. Soc.* **1954**, *76*, 5672. (b) Ye, H.; Deng, G.; Liu, J.; Qiu, F. G. Expedient construction of the ziegler intermediate useful for the synthesis of forskolin via consecutive rearrangements. *Org. Lett.* **2009**, *11*, 5442.

(22) (a) Bauer, M.; Maier, M. E. Synthesis of the core structure of salicylilhalamide A by intramolecular Suzuki reaction. *Org. Lett.* **2002**, *4*, 2205. (b) Li, W.; LaCour, T. G.; Fuchs, P. L. Dyotropic rearrangement facilitated proximal functionalization and oxidative removal of angular methyl groups: efficient syntheses of 23'-deoxy cephalostatin 1 analogues. *J. Am. Chem. Soc.* **2002**, *124*, 4548.

(23) (a) Lee, H. Y.; An, M. Y. Selective 1,4-reduction of unsaturated carbonyl compounds using $\text{Co}_2(\text{CO})_8\text{-H}_2\text{O}$. *Tetrahedron Lett.* **2003**, *44*, 2775. (b) Lee, H. Y.; An, M. Y.; Sohn, J. H. Studies on reductive Pauson-Khand reaction using cobaltcarbonyls with water. *Bull. Korean Chem. Soc.* **2003**, *24*, 539. (c) Krafft, M. E.; Wright, J. A.; Bonaga, L. Pauson-Khand reactions in water. *Can. J. Chem.* **2005**, *83*, 1006.