



Cite this: *Chem. Commun.*, 2021, 57, 5266

Received 24th February 2021,
Accepted 26th April 2021

DOI: 10.1039/d1cc01016f

rsc.li/chemcomm

Interrupted aza-Wittig reactions using iminophosphoranes to synthesize ^{11}C -carbonyls†

Uzair S. Ismailani,^{ab} Maxime Munch,^{ab} Braeden A. Mair^{bc} and Benjamin H. Rotstein^{ab*}

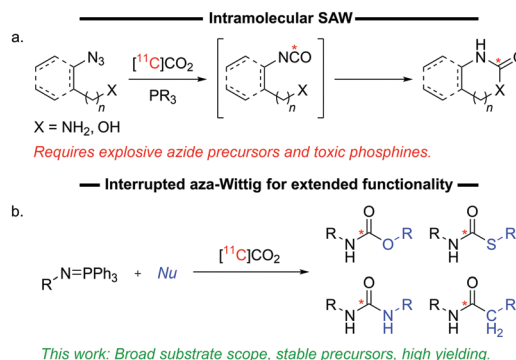
A direct CO_2 -fixation methodology couples structurally diverse iminophosphoranes with various nucleophiles to synthesize ureas, carbamates, thiocarbamates, and amides, and is amenable for ^{11}C radiolabeling. This methodology is practical, as demonstrated by the synthesis of >35 products and isolation of the molecular imaging radiopharmaceuticals [^{11}C]URB694 and [^{11}C]glibenclamide.

Positron emission tomography (PET) is a molecular imaging modality used to evaluate biological processes *in vivo* with short-lived radionuclides. PET radiopharmaceuticals are used to diagnose metastatic and cardiovascular diseases, to detect biomarkers of neurodegeneration, and to probe molecular and functional mechanisms in living systems. Carbon-11 (^{11}C , $t_{1/2} = 20.4$ min) is prized for isotopic labeling of biomolecules, and is routinely incorporated into PET imaging agents for both research and clinical applications.¹ Currently, a lack of diverse methods for directly incorporating [^{11}C]CO₂ into complex molecules has limited its use.² Consequently, [^{11}C]CO₂ is most often converted into reactive secondary precursors such as [^{11}C]CH₃I or [^{11}C]CH₃OTf, which are accompanied by elongated processing times and reductions in radiochemical yield due to sub-quantitative conversions.^{3–6}

Isocyanates are valuable synthetic intermediates that can be readily converted into pharmaceutically-relevant functional groups such as carbamates, ureas, and amides.^{7,8} Current approaches to synthesizing ^{11}C -isocyanates rely on stepwise trapping of [^{11}C]CO₂ with amines, followed by dehydration using POCl₃ or Mitsunobu-type conditions.^{2,9,10} Importantly, the former strategy displays poor tolerance towards anilines and the highly acidic conditions pose challenges for maintaining efficient trapping of [^{11}C]CO₂ in solution. The latter

technique has displayed improved utility for synthesizing asymmetrical ureas¹⁰ and also amides using either Grignard reagents¹¹ or organozinc coupling reactions,¹² though high mass loading of azo reagents and phosphines may complicate radiotracer purification. Each strategy requires careful control of temperature and reagent concentrations during sequential reaction steps in order to prevent formation of complex mixtures of symmetrical byproducts and heterocycles.⁹ Iminophosphoranes have been shown to undergo the aza-Wittig reaction directly with CO₂ to produce isocyanates in high yields.¹³ In the context of ^{11}C , the commercially available precursor *N*-(triphenylphosphoranylidene)aniline (**1a**) was previously reported to prepare acyclic ^{11}C -ureas from [^{11}C]CO₂ in moderate radiochemical yields (RCYs).¹⁴ Del Vecchio *et al.* deployed *o*-azidoanilines and azido alcohols with dimethylphenylphosphine to synthesize cyclic ureas and carbamates in useful yields through a proposed intramolecular Staudinger aza-Wittig sequence (SAW, Scheme 1a) upon heating to 70 °C.^{15,16} An intermolecular variant of this reaction produced a linear carbamate in low RCY at much higher temperature.

We aimed to develop conditions that are high yielding and selective for C–O, C–N, C–C, and C–S bond formation and would be amenable for one-pot [^{11}C]CO₂-fixation to prepare



Scheme 1 Synthetic approaches to iminophosphorane [^{11}C]CO₂-fixation.

^a Department of Biochemistry, Microbiology and Immunology, University of Ottawa, Canada. E-mail: benjamin.rotstein@uottawa.ca

^b University of Ottawa Heart Institute, 40 Ruskin St, Ottawa, ON, Canada

^c Department of Chemistry and Biomolecular Sciences, University of Ottawa, Canada

† Electronic supplementary information (ESI) available. See DOI: 10.1039/d1cc01016f

radiopharmaceuticals and novel tracer candidates. Through the synthesis of functionalized iminophosphorane precursors, this approach would obviate the need for highly acidic POCl₃, Mitsunobu reagents, explosive azide precursors, and toxic phosphines used in current methodologies, enhancing the substrate versatility and practicality of this method for good manufacturing practices (GMP) environments (Scheme 1b). Herein, we describe a versatile and efficient approach to carbonyl ligation using iminophosphorane-CO₂-fixation coupled with intermolecular nucleophilic addition. This method is effective for synthesizing acyclic products with stable isotopes under mild conditions and is suitable for automated synthesis and ¹¹C radiolabeling.

At the outset, we focused on developing a nucleophilic coupling strategy to iminophosphorane-CO₂ fixation conditions using stable isotopes, since no such straightforward procedure for this coupling has been reported.¹³ First, CO₂ was bubbled into a heated toluene solution containing (**1a**) until consumption of the iminophosphorane, followed by the addition of benzyl alcohol (**2**). Low yield of the desired product **4** (6%) was observed using stepwise addition (Table 1, entry 1). The observed major product was the symmetrical *N,N'*-diphenylcarbodiimide (**3**), likely formed by a second aza-Wittig coupling reaction to the isocyanate intermediate. The ratio of **4**:**3** increased to 0.9:1 when the nucleophile was present from the beginning of the reaction (entry 3). Increasing the concentration of **2** led to exclusive formation of **4** in 84% yield (entry 4). This suggests that short-lived free isocyanates are formed in the presence of iminophosphoranes, subject to two competing reaction pathways: aza-Wittig carbodiimide formation and carbamate formation. Thus, at high concentrations of an intercepting nucleophile, it is possible to selectively divert the reaction towards intermolecular ligation. Carbamate **4** could be prepared and isolated in good yields from hydrocarbon, ethereal, and polar solvents using similar conditions (entries 5–8). With our primary focus on establishing easily translatable conditions to ¹¹C radiochemistry, we were delighted to find that two common CO₂ trapping bases, amidine

DBU and phosphazene BEMP (2.6 equiv.), significantly increased the rate of formation of **4**, concomitant with moderate impacts in yield (entries 9–10, Table S1, ESI†). These effects are likely due to the increased availability of soluble CO₂ complexes.^{17,18}

We assessed the compatibility of the iminophosphorane-CO₂-fixation method with a diverse scope of nucleophiles (Scheme 2a). Under our developed conditions, carbamates derived from benzyl, isopropyl, 4-methoxyphenethyl, and *tert*-butyl alcohol were isolated in 83–94% yields (**4**–**7**). In contrast, phenyl carbamates and thiocarbamates (**8**–**10**) required higher nucleophile concentrations to achieve yields ≥ 70%, likely due to their propensity for elimination. Benzyl mercaptan also proved to be a compatible nucleophile, forming the corresponding thiocarbamate **11** in 86% yield. Sterically hindered nucleophiles gave carbamate **7** (83%) and urea **12** (78%). We were gratified to find that amides such as **13** (82%) could be accessed directly by carbon–carbon bond formation using diethyl malonate. Despite this success, some nucleophiles were found to be incompatible with the interrupted aza-Wittig conditions, including Grignard reagents and phenylacetylene. However, several of our synthesized products (**7**–**10**, **12**) stand in as blocked isocyanates, and facilitate indirect nucleophilic substitution (Scheme 2b).¹⁹ *In situ* formed *O*-phenylcarbamate **8** was transformed to amides **14**–**16** in moderate-to-good yields based on **1a**. Both direct and indirect nucleophilic substitutions are robust, and further open the door to selective C–C bond formation using iminophosphoranes.

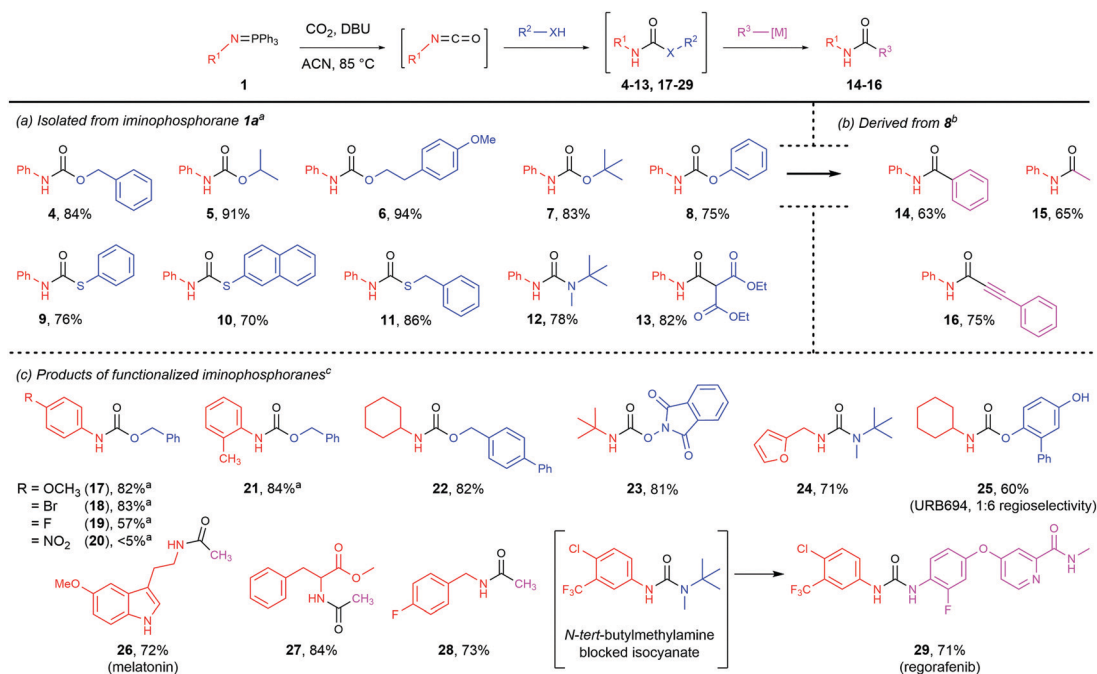
We next investigated the scope of functionalized aryl iminophosphoranes synthesized by the Kirsanov reaction and isolated by our modified general procedure (Scheme 2c, see ESI† for details). First, CBz-protected products (**17**–**21**) were isolated to determine sensitivity to electronic and steric features of iminophosphoranes under the optimized conditions. Electron-rich arenes, aryl bromides, and *ortho*-substitution (**17**–**18**, **21**) were all well-tolerated in comparison with electron-deficient arenes (**19**–**20**). Alkyl iminophosphoranes afforded products such as carbamate **22** (82%) and blocked isocyanates **23**–**24** in good yields. The utility of this method for biopharmaceuticals was assessed by targeting the fatty acid amide hydrolase inhibitor URB694 (**25**), melatonin (**26**), and the oral multi-kinase inhibitor regorafenib (**29**). Hydroquinone carbamate **25** was isolated as a mixture of regioisomers in 60% yield.⁹ Indirect substitution using phenol-blocked isocyanates yielded melatonin **26** (72%), phenylalanine derivative **27** (84%), and electron-deficient amide **28** (73%). Finally, the urea regorafenib **29** was synthesized first by direct nucleophilic coupling, though indirect substitution using an *N*-*tert*-butylmethylamine blocked isocyanate intermediate better facilitated purification of **29** in 71% overall yield.²⁰

Satisfied with the iminophosphorane-CO₂ nucleophilic coupling methodology using stable isotopes, we were determined to apply this method to ¹¹C radiochemistry. Since [¹¹C]CO₂ is the limiting reagent in these processes (typically < 1 μmol), reconsideration of reaction conditions to produce [¹¹C]**4** was required (Table 2). First, we focused on the influence of the concentration of **1a** on product yield using high concentrations

Table 1 Optimization of reaction conditions

Entry	Solvent	[2] (mM)	3 Yield (%)	4 Yield (%)
1 ^{ab}	Toluene	100	77	6
2 ^{ab}	Toluene	150	77	6
3 ^{ac}	Toluene	100	42	36
4 ^{ac}	Toluene	250	0	84
5 ^d	Toluene	250	0	70
6 ^d	1,4-Dioxane	250	0	72
7 ^d	DMF	250	0	75
8 ^d	ACN	250	0	78
9 ^{de}	ACN	250	0	84
10 ^{df}	ACN	250	0	70

^a UPLC/MS yields; 110 °C. ^b Nucleophile added after iminophosphorane consumption. ^c Nucleophile added at *t* = 0. ^d Conditions: **2** (0.625 mmol), solvent (2.5 mL), 85 °C; then **1** (100 mM, 2.5 mL) added over 1 h. Reaction time: 2 h. Isolated yields. ^e DBU (2.6 equiv). ^f BEMP (2.6 equiv).



Scheme 2 For Extended Functionality – Reaction Scope. ^a Conditions: **1** (0.25 mmol), R²-XH (2.5 equiv.), DBU (2.6 equiv.), ACN (2.5 mL), reflux; isolated yields. ^b PhOH (10 equiv.), R³-MgX (20 equiv.) or PhCCH (12 equiv.), THF. ^c R²-XH (10 equiv.), LHMDS (0.99 equiv.), THF. See ESI† for detailed procedures.

of DBU and benzyl alcohol in ACN (entry 1). We noted a low 13% RCY with these initial conditions, mainly due to a large excess of unreacted [¹¹C]CO₂. Increasing the concentration of **1a** (entries 2–3) led to maximum 32% RCY and reducing the concentration of DBU to 100 mM enhanced the selectivity toward [¹¹C]**4** (entry 4–5). Increasing the reaction temperature in DMF to 100 °C resulted in 65% RCY (entries 6–7). Finally, increasing the concentration of nucleophile **2** further improved the selectivity of [¹¹C]**4**, leading to a 91% RCY (entries 7–9). Trapping efficiencies (TE) of [¹¹C]CO₂ during the optimization of [¹¹C]**4** were all greater than 90%.

Structurally diverse iminophosphoranes were also used to radiolabel compounds using this procedure (Scheme 3). Aniline

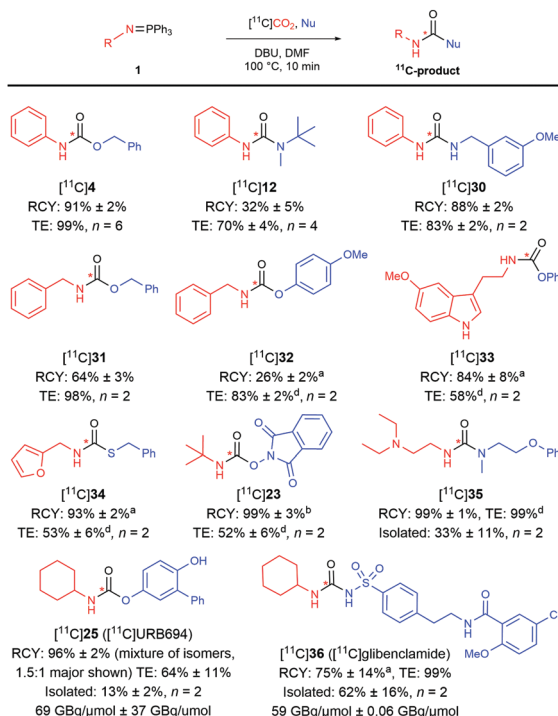
derived products of **1a** include labeled urea [¹¹C]**30** in 88% RCY, and blocked isocyanate [¹¹C]**12** in 32% RCY. Using benzyl iminophosphorane, *O*-benzyl carbamate [¹¹C]**31** was formed in 64% yield. By substituting DABCO for DBU, *O*-aryl carbamate [¹¹C]**32** (26%), 5-methoxytryptamine carbamate [¹¹C]**33** (84%), and thiocarbamate [¹¹C]**34** (93%) could be radiolabeled, with TE ranging from 53–83%. We suspect that mildly basic conditions and higher nucleophile concentrations improve the yields of **32–34** due to their sensitivity towards elimination.²¹ From *tert*-butyl iminophosphorane, *N*-hydroxysuccinimide-derived [¹¹C]**23** (99%) was also radiolabeled.

To further demonstrate the practicality of this technique, [¹¹C]**35**, an experimental antiarrhythmic compound containing the β-glucocerebrosidase activating moiety *N*-methyl-*N*-(2-phenoxyethyl)amine, was isolated using a fully automated method (see ESI†).^{22,23} [¹¹C]**35** was labeled with 99% RCY starting from 15.7 GBq of [¹¹C]CO₂, and obtained in an isolated yield of 33% ± 10.6% (2.7 ± 0.4 GBq) within 22 min of [¹¹C]CO₂ delivery. The fatty acid amide hydrolase inhibitor [¹¹C]URB694 ([¹¹C]**25**, [¹¹C]CURB), used in clinical research to investigate psychiatric illnesses and alcohol use disorder, was prepared from cyclohexyliminophosphorane in 96% ± 2% RCY (2:3 regioselectivity).^{24,25} From 25.9 GBq of [¹¹C]CO₂, 1.9 ± 0.7 GBq of [¹¹C]CURB was obtained as the major isomer in 99% radiochemical purity, with an isolated yield of 13% ± 2%, and molar activity of 69 ± 37 GBq μmol⁻¹ within 17 min from [¹¹C]CO₂ delivery. Lastly, the sulfonylurea glibenclamide, currently used in the treatment of diabetes mellitus type 2 and shown to reduce tissue damage in preclinical models of CNS injuries, was synthesized with 75% ± 14% RCY.²⁶ [¹¹C]Glibenclamide ([¹¹C]**36**) is a substrate for organic anion-transporting

Table 2 Radiochemical optimization

Entry	T (°C)	[1a] (mM)	[2] (mM)	[DBU] (mM)	RCY ^a (%)
1	65	7	200	200	13
2	65	70	200	200	32
3	65	100	200	200	29
4	65	70	200	150	39
5	65	70	200	100	56
6 ^b	65	70	200	100	62
7 ^b	100	70	200	100	65
8 ^b	100	70	600	100	78
9 ^b	100	70	1200	100	91

^a Radiochemical yield calculated by relative integration of radioHPLC chromatograms, see ESI for details; average of *n* ≥ 2; entry 9 *n* = 6; trapping efficiency ≥ 90%. ^b DMF solvent.



Scheme 3 Carbon-11 substrate scope. ^aDBU replaced with DABCO. ^bDBU replaced with LHMDs. ^cK⁺OTf ^d[^{11}C]CO₂ trapped at -60°C . See ESI† for detailed procedures. * indicates position of ^{11}C . TE = trapping efficiency; isolated RCY decay-corrected to end-of-synthesis.

polypeptide (OATP) transporter and can be used to study the integrity of the blood-brain barrier by PET imaging.²⁷ This radiopharmaceutical, which has been synthesized in two-steps using [^{11}C]CH₃OTf,²⁸ was efficiently labeled using an iminophosphorane precursor directly from [^{11}C]CO₂. Following purification, 7.4 ± 1.9 GBq of [^{11}C]glibenclamide was obtained with an isolated yield of $62\% \pm 16\%$ from 25.9 GBq of [^{11}C]CO₂, and a molar activity of 59 ± 0.06 GBq/ μmol^{-1} within 21 minutes from the beginning of synthesis.

In conclusion, we have developed a methodology to synthesize several stable and radiolabeled functional groups using the interrupted aza-Wittig approach. The advantages of this method include direct [^{11}C]CO₂-fixation using stable iminophosphorane precursors prepared from available amines, diverse functional group selectivity, and applicability to PET imaging agents. Radiopharmaceuticals are synthesized under mild reaction conditions with rapid synthesis times and using automated procedures. We anticipate this method contributing to the accessibility of in-demand radiopharmaceuticals such as [^{11}C]CURB and [^{11}C]glibenclamide, among others.

Support for this work was provided by the University of Ottawa Heart Institute, NSERC (RGPIN-2017-06167), CFI (36848), and the Ontario Ministry for Research, Innovation and Science (ER17-13-119). The authors are grateful to Marcelo Muñoz and Gapisha Karunakaran for the isolation of chemical standards and preparation of precursors.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- G. Antoni, *J. Labelled Compd. Radiopharm.*, 2015, **58**(3), 65–72.
- B. H. Rotstein, S. H. Liang, M. S. Placzek, J. M. Hooker, A. D. Gee, F. Dollé, A. A. Wilson and N. Vasdev, *Chem. Soc. Rev.*, 2016, **45**(17), 4708–4726.
- B. Långström, G. Antoni, P. Gullberg, C. Halldin, P. Malmberg, K. Nägren, A. Rimland and H. Svård, *J. Nucl. Med.*, 1987, **28**(6), 1037–1040.
- P. Larsen, J. Ulin, K. Dahlström and M. Jensen, *Appl. Radiat. Isot.*, 1997, **48**(2), 153–157.
- J. M. Link, K. A. Krohn and J. C. Clark, *Nucl. Med. Biol.*, 1997, **24**(1), 93–97.
- D. M. Jewett, *Int. J. Rad. Appl. Instrum. A.*, 1992, **43**(11), 1383–1385.
- E. Delebecq, J.-P. Pascault, B. Boutevin and F. Ganachaud, *Chem. Rev.*, 2013, **113**(1), 80–118.
- A. K. Ghosh, M. Brindisi and A. Sarkar, *ChemMedChem*, 2018, **13**(22), 2351–2373.
- A. A. Wilson, A. Garcia, S. Houle, O. Sadovskii and N. Vasdev, *Chem. – Eur. J.*, 2011, **17**(1), 259–264.
- A. K. Haji Dheere, N. Yusuf and A. Gee, *Chem. Commun.*, 2013, **49**(74), 8193–8195.
- S. Bongarzone, A. Runser, C. Taddei, A. K. Haji Dheere and A. D. Gee, *Chem. Commun.*, 2017, **53**(38), 5334–5337.
- B. A. Mair, M. H. Fouad, U. S. Ismailani, M. Munch and B. H. Rotstein, *Org. Lett.*, 2020, **22**(7), 2746–2750.
- P. I. Molina, M. Alajarin and A. Arques, *Synthesis*, 1982, 596–597.
- E. W. van Tilburg, A. D. Windhorst, M. van der Mey and J. D. M. Herscheid, *J. Labelled Compd. Radiopharm.*, 2006, **49**(4), 321–330.
- A. Del Vecchio, F. Caillé, A. Chevalier, O. Loreau, K. Horkka, C. Halldin, M. Schou, N. Camus, P. Kessler, B. Kuhnast, F. Taran and D. Audisio, *Angew. Chem., Int. Ed.*, 2018, **57**(31), 9744–9748.
- A. Del Vecchio, A. Talbot, F. Caillé, A. Chevalier, A. Sallustiau, O. Loreau, G. Destro, F. Taran and D. Audisio, *Chem. Commun.*, 2020, **56**(78), 11677–11680.
- C. Villiers, J.-P. Dognon, R. Pollet, P. Thuéry and M. Ephritikhine, *Angew. Chem., Int. Ed.*, 2010, **49**(20), 3465–3468.
- M. Hulla and P. J. Dyson, *Angew. Chem., Int. Ed.*, 2020, **59**(3), 1002–1017.
- M. S. Rolph, A. L. J. Markowska, C. N. Warriner and R. K. O'Reilly, *Polym. Chem.*, 2016, **7**(48), 7351–7364.
- M. Hutchby, C. E. Houlden, J. G. Ford, S. N. G. Tyler, M. R. Gagné, G. C. Lloyd-Jones and K. I. Booker-Milburn, *Angew. Chem., Int. Ed.*, 2009, **48**(46), 8721–8724.
- J. S. Derasp and A. M. Beauchemin, *ACS Catal.*, 2019, **9**(9), 8104–8109.
- J. Zheng, S. Jeon, W. Jiang, L. F. Burbulla, D. Ysselstein, K. Oevel, D. Krainc and R. B. Silverman, *J. Med. Chem.*, 2019, **62**(3), 1218–1230.
- J. R. Shanklin, Jr. and C. P. Johnson, III US4500529A, February 19, 1985.
- A. A. Wilson, A. Garcia, J. Parkes, S. Houle, J. Tong and N. Vasdev, *Nucl. Med. Biol.*, 2011, **38**(2), 247–253.
- L. M. Best, B. Williams, B. Le Foll, E. Mansouri, R. P. Bazinet, L. Lin, V. De Luca, D. Lagzdins, P. Rusjan, R. F. Tyndale, A. A. Wilson, C. S. Hendershot, M. Heilig, S. Houle, J. Tong, S. J. Kish and I. Boileau, *Neuropsychopharmacology*, 2020, 1–8.
- D. B. Kurland, C. Tosun, A. Pampori, J. K. Karimy, N. M. Caffes, V. Gerzanich and J. M. Simard, *Pharmaceuticals*, 2013, **6**(10), 1287–1303.
- N. Tournier, W. Saba, S. Cisternino, M.-A. Peyronneau, A. Damont, S. Goutal, A. Dubois, F. Dollé, J.-M. Scherrmann, H. Valette, B. Kuhnast and M. Bottlaender, *AAPS J.*, 2013, **15**(4), 1082–1090.
- F. Caillé, P. Gervais, S. Auvity, C. Coulon, S. Marie, N. Tournier and B. Kuhnast, *Nucl. Med. Biol.*, 2020, **84–85**, 20–27.