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Synthesis of spiro[4.4]thiadiazole derivatives via double 1,3-dipolar cycloaddition of hydrazone chlorides with carbon disulfide†

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An operationally simple and convenient synthesis method toward a series of diverse spiro[4.4]thiadiazole derivatives via double [3 + 2] 1,3-dipolar cycloaddition of nitrilimines generated *in situ* from hydrazone chlorides with carbon disulfide has been achieved under mild reaction conditions.

The spirocyclic compounds having cyclic structures connected through just one carbon atom have attracted much interest from synthetic chemists and medicinal chemists because of their ubiquitous presence in the core of a plethora of natural products and non-natural products, many of which display a broad range of pharmacological and biological activities.¹ Moreover, spiro-compounds are unique because of their rigidity and distinctly three-dimensional structure and have proved to be very interesting for medicinal chemistry or as ligand and catalyst motifs in asymmetric synthesis.² Due to the importance of the spirocyclic architectures, the methods for synthesis of the spirocyclic moiety are too many to enumerate. Some common strategies to afford spirocyclic scaffolds include radical cyclizations,³ Diels–Alder reactions,⁴ cycloaddition,⁵ and ring expansion,⁶ among others. Many of the known methods for synthesizing spiro structures are through constructing a new ring of which the substrates include a carbo- or heterocycle structure.⁷ To the best of our knowledge, a few approaches have offered efficient ways on the formation of two rings through a double 1,3-dipolar cycloaddition in one pot for constructing spirocyclic scaffold.⁸ Therefore, the design and development of innovative and efficient methodologies via double 1,3-dipolar cycloaddition under mild reaction conditions for synthesizing bioactive content spirocyclic scaffolds from readily available precursors is in great demand in both organic and medicinal chemistry.

On the other hand, the 1,3-dipolar cycloaddition reaction (1,3-DCs) has been one of the most prominent reactions to build

five- or six-membered heterocycle in one step in the field of organic synthesis.⁹ In particular, nitrilimines generated *in situ* from the corresponding hydrazone chloride in the presence of a base are highly active intermediates in organic synthesis and have been widely used as useful synthons for preparing bioactive nitrogen heterocyclic derivatives and spirocyclic compounds through the [3 + 2],¹⁰ [3 + 3]¹¹ and [3 + 4]¹² cycloaddition reactions. In addition, the Lu group reported 1,3-dipolar cycloaddition of nitrilimines with carbon dioxide (CO₂), providing elegant access to 1,3,4-oxadiazole-2(3*H*)-ones derivatives.¹³ Meanwhile, carbon disulfide (CS₂) that it is an analogue of CO₂, has been used for the synthesis of various sulfur-containing heterocyclic compounds for agricultural, medicinal, and pharmaceutical applications.¹⁴ Based on the above literatures and in continuation of our interest in synthesis of heterocycles, we disclose a novel protocol for the synthesis of spiro[4.4]thiadiazole derivatives double 1,3-dipolar cycloaddition of nitrilimines generated *in situ* from hydrazone chlorides with carbon disulfide under mild conditions.

In our initial investigation, we chose the hydrazone chloride as the nitrilimine precursor with CS₂ as the model reaction to optimize the reaction conditions. The results of these experiments are summarized in Table 1. No product was observed when the reaction was performed in the absence of base (Table 1, entry 1). Usually, the active nitrilimine is generated *in situ* by dehydrohalogenation of the corresponding hydrazone halide in the presence of an equivalent base. Pleasingly, with the use of TEA as the base, the double 1,3-dipolar cycloaddition reaction proceeded very smoothly in CH₂Cl₂ to give spirothiadiazole product **3a** with 60% yield at room temperature (Table 1, entry 2). The product **3a** was obtained with 56% yield in the presence of DABCO (entry 3). However, the DBU base turned out to be inactive in the double 1,3-dipolar cycloaddition reaction (entry 4). Subsequently, we screened a series of inorganic base to improve the yield, such as Na₂CO₃, K₂CO₃, Cs₂CO₃, NaOH and KOH (entries 5–9). In comparison, the product **3a** was given in

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Table 1 Optimization of reaction conditions^a

| Entry | Base | Solvent | Yield of 3a ^b (%) |
|-----------------|---------------------------------|---------------------------------|------------------------------|
| 1 | None | CH ₂ Cl ₂ | 0 |
| 2 | TEA | CH ₂ Cl ₂ | 60 |
| 3 | DABCO | CH ₂ Cl ₂ | 56 |
| 4 ^b | DBU | CH ₂ Cl ₂ | Trace |
| 5 | Na ₂ CO ₃ | CH ₂ Cl ₂ | 62 |
| 6 | K ₂ CO ₃ | CH ₂ Cl ₂ | 70 |
| 7 | Cs ₂ CO ₃ | CH ₂ Cl ₂ | 92 |
| 8 | NaOH | CH ₂ Cl ₂ | 90 |
| 9 | KOH | CH ₂ Cl ₂ | 88 |
| 10 | Cs ₂ CO ₃ | CHCl ₃ | 80 |
| 11 | Cs ₂ CO ₃ | DCE | 85 |
| 12 | Cs ₂ CO ₃ | EtOAc | 70 |
| 13 | Cs ₂ CO ₃ | Toluene | 61 |
| 14 | Cs ₂ CO ₃ | THF | Trace |
| 15 | Cs ₂ CO ₃ | Et ₂ O | Trace |
| 16 | Cs ₂ CO ₃ | Dioxane | 56 |
| 17 | Cs ₂ CO ₃ | MeCN | 62 |
| 18 ^c | Cs ₂ CO ₃ | CH ₂ Cl ₂ | 80 |
| 19 ^d | Cs ₂ CO ₃ | CH ₂ Cl ₂ | 92 |

^a Unless noted otherwise, reactions were performed with hydrazonoyl chloride **1a** (0.2 mmol), carbon disulfide (0.3 mmol, 1.5 equiv.), base (0.2 mmol, 1 equiv.) in solvent (1.0 mL) at rt for 12 h. ^b Isolated yield by chromatography on silica gel. ^c Reaction was performed with carbon disulfide (0.2 mmol, 1 equiv.) for 24 h. ^d Carbon disulfide (1.0 mmol, 5 equiv.).

moderate yields in the presence of TEA, Na₂CO₃ or K₂CO₃ (entries 2, 5 and 6). To our delight, the yield of the product **3a** was increased to 92% when used Cs₂CO₃ as the base (entry 7). Nevertheless, the yield was slightly reduced with the use of NaOH and KOH as the base (entries 8 and 9). Encouraged by this result, the reaction was examined with different solvents to improve the yield of **3a**. The reaction can be carried out in CHCl₃ and DCE to provide the expected product in 80% yield and 85% yield, respectively (entries 10 and 11). Whereas no the desired product formation take place upon using THF and Et₂O as the solvent (entries 14 and 15). The reactions run with either dioxane or MeCN resulted in moderate conversions (entries 16 and 17). It was found that CH₂Cl₂ was the best solvent for this transformation compared to CHCl₃, DCE, EtOAc, toluene, THF, Et₂O, dioxane and MeCN (entries 10–17). Moreover, when decreased the amounts of CS₂, the reaction time was prolonged and the yield of the product was lowered (entry 18). Whereas, the yield was not further improved when using 5.0 equiv. of CS₂ (entry 19). Therefore, the optimal reaction conditions have been determined which using Cs₂CO₃ as the base in CH₂Cl₂ at room temperature for 12 h (Table 2).

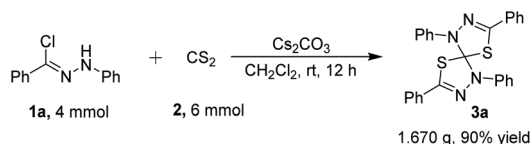
With the optimal reaction conditions in hand, we subsequently investigated the substrate scope and limitation of the nitrilimine precursors. The results are shown in Table 1. Under the optimized conditions, the double 1,3-dipolar cycloaddition reaction could tolerate a variety of hydrazonoyl chlorides bearing different substituents, regardless of the electronic properties

Table 2 Substrate scope of the double 1,3-dipolar cycloaddition^a

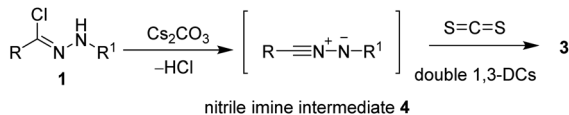
^a Unless noted otherwise, reactions were performed with hydrazonoyl chloride **1** (0.2 mmol), carbon disulfide (0.3 mmol, 1.5 equiv.), base (0.2 mmol, 1 equiv.) in solvent (1.0 mL) at rt for 12 h. Isolated yield by chromatography on silica gel.

(such as electron-donating or electron-withdrawing) or the positions of substituents on the benzoyl chloride moiety, to generate the corresponding products **3** (**3a–i**) in high yields (92–96%). In addition, the structure of product **3h** (CCDC 2041891)¹⁵ was further determined by single-crystal X-ray crystallography analysis. Furthermore, when using the fused aromatic and heteroaromatic hydrazonoyl chlorides as the substrate reacted with CS₂, the reactions were also found to be compatible and gave the products (**3j–l**) in high yields. On the other hand, the hydrazonoyl chlorides containing different substituents (such as methyl and chloro) on phenylhydrazone moiety also worked well in the reaction successfully to obtain the desired cycloadducts **3m** and **3n** in 90% and 93% yield, respectively. Nevertheless, the hydrazonoyl chlorides bearing cyano or nitro group on the benzoyl chloride moiety were not suitable and the expected cycloadduct **3p** and **3q** were not formed. Moreover, the double 1,3-dipolar cycloaddition reaction also didn't work with aliphatic group (**3r**) at the hydrazonoyl chloride.





Scheme 1 Scaled-up version of synthesis of 3a.



Scheme 2 Proposed mechanism of the double [3 + 2] cycloaddition.

To further exhibit the synthetic utility for spirocyclic compounds, under the optimized conditions, a gram scale experiment between 4 mmol of hydrazonil chloride **1a** and 6 mmol of CS₂ proceeded smoothly to afford the desired product **3a** without a significant loss of efficiency (1.670 g, in 90% yield) (Scheme 1). The easy scale-up of this process shows the reaction to be a practical tool for the synthesis of structurally diversified natural product-like molecules possessing privileged scaffold for potential application in biomedical research and other research fields.

As shown in Scheme 2, a plausible mechanism was proposed. Firstly, the nitrilimine intermediate **4** generated *in situ* from the corresponding hydrazonil halide **1** via eliminating of HCl in the presence of a base. Then, the nitrilimine **4** reacts with CS₂ through the double 1,3-dipolar cycloaddition reaction to give the desired product **3**.

In conclusion, we have developed an efficient and simple method to synthesize a broad range of diverse spiro[4.4]thiadiazole derivatives in high yields (up to 96%) through the double 1,3-dipolar cycloaddition of nitrilimines generated *in situ* with CS₂. This reaction proceeds with readily available starting materials, transition-metal free, the experimental simplicity, easy purification, and mild reaction conditions make this procedure highly appropriate for the synthesis of spiro[4.4]thiadiazole derivatives.

Conflicts of interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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