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Site-Specific Incorporation of Multiple Thioamide Substitutions into a Peptide Backbone via Solid Phase Peptide Synthesis

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Supporting Information

ABSTRACT: Among various peptide modification strategies, thioamide substitution by replacing the carbonyl oxygen atom of an amide bond by a sulfur atom, constitutes an invaluable tool for chemical biology including peptide drug discovery and protein structure–function studies. However, thioamide substitution effect has not been well studied because of the lack of synthetic methods to site-specifically incorporating a thioamide bond into a peptide backbone, particularly introducing multi thioamide substitutions to peptide on a solid support. Herein, we reported a highly efficient method to incorporate a thioamide bond to the peptide backbone in a site-specific manner by employing α -thioacyloxyenamides, which are formed from the addition of N-protected monothioamino acids and ynamides, as a kind of novel thioacylating reagents in solid phase peptide synthesis. This method is amenable for 19 of 20 proteinogenic amino acids except for the His. Mono to multiple thioamide substitutions could be



incorporated into a growing peptide with no or low level of epimerization. By using this method, fully thioamide substituted hexapeptide containing up to five continuous thioamide bonds could be synthesized smoothly. This synthetic methodology will spur the application of the thioamide substitution tool for protein engineering and peptide drug discovery.

■ INTRODUCTION

Site-specific modification and functionalization of peptides and proteins can adjust their properties and functionalities and thus evolves into an invaluable tool for peptide drug discovery and protein structure-function studies.¹ Compared to the extensively studied post translation sidechain modifications, little attention has been paid to the peptide backbone modifications.² Replacing the amide bond with other isosteric functional groups is a straightforward strategy for peptide backbone modification. Among various amide bond surrogates, thioamide substitution, in which the amide bond oxygen atom is replaced with a sulfur atom, is the simplest modification that causes minimal perturbation.³ Although numerous studies have revealed that the thioamide is isosteric to an amide bond with the same or similar geometry conformations,^{3b} both groups retain many intrinsic different chemical and physical properties.⁴ For example, the resistance of peptide to enzymatic degradation and bioactive can be dramatically enhanced by replacing the canonical peptide bond with a thioamide linkage.5 Thioamide exhibits higher nucleophilicity and electrophilicity,6 altered hydrogen-bonding propensity,⁷ unique spectroscopic properties⁸ and greater metal binding affinity.⁹ Owing to these features, thioamide has been employed as probes for studying the dynamics of H-bond formation in β -sheets, ¹⁰ α -helices, ^{11, 10c} interaction of peptide bonds,¹² *cis/trans* photoswitches,¹³ fluorescence quenchers,¹⁴ and metal-binding substrates.¹⁵ Although several intriguing features of thioamides have been recruited for studying protein folding, stability, function, and dynamics, the biological behavior of thioamide substitution on peptides and proteins of interest still is unpredictable. Extensive

study of the physical and biological effect of thioamide substitution is impeded mainly by the lack of efficient synthetic strategies to site-specific incorporating thioamide bond into peptide backbone.

A broad range of synthetic methodologies for thioamide have been developed over the past decades.¹⁶ However, only few of them are amenable for incorporating a thioamide bond into a growing peptide because of the difficult availability of chiral starting material and the highly prone to racemization of the α chiral center of the thioamide residue. There are two strategies, by employing thionating reagents and thioacylating reagents, respectively, for synthesis of thioamide substituted peptide. Thionating reagents such as P_4S_{10} ,¹⁷ Lawesson's reagent,¹⁸ and other sulfur reagents¹⁹ have been used to selectively convert the carbonyl oxygen atom of an amide bond to an sp²-hybridized sulfur atom in the presence of ester functionality. However, this method is limited to simple amides and dipeptides because there is no selectivity for substrates containing more than one amide bonds. In this regard, the thioacylating reagents are more flexible because the modular nature enables them to incorporate a thioamide bond in a site-specific manner. Alkyl dithioesters,20 thiobenzimidazolones,²¹ and thioacylnitrobenzotriazoles²² have been employed as active thioacylating reagents for thioamide substituted peptide synthesis. Thioacylnitrobenzotriazoles prepared via three steps transformations from natural α -amino acids (Figure 1, eq. 1) are the only one class of thioacylating reagents that could be used to introduce a thioamide bond in SPPS.^{4, 23} However, this method is limited to the 15 of the 20 common proteinogenic α -amino acids.² In addition, the preparation of thioacylnitrobenzotriazole involves the use of toxic and malo-

Figure 1. Thioacylating Reagents Effective for Incorporating Thioamide Substitution in SPPS.



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dour thionating reagents. Meanwhile, monothioamino acids have been used as thioacyl donor for thioamide incorporation with the assistance of phosphorus coupling reagents.24 However, the co-formation of significant amount of corresponding oxo-amide byproduct rendered it not feasible for SPPS. Recently, we disclosed that ynamide could be used as a racemization/epimerization-free coupling reagent for peptide bond formation.²⁵ Interestingly, thioamide substituted peptide could be obtained in a racemization/epimerization-free manner when monothioamino acid was employed as the acyl donor with α -thioacyloxyenamide formed from the addition of monothioamino acid to ynamide as an efficient thioacylating reagent (Figure 1, eq. 2).²⁶ Owing to the optimal balance between the reactivity and stability of α -thioacyloxyenamides, they can also be used to incorporate a thioamide bond sitespecifically in SPPS. Herein, we provided a systematic study about the use of α -thioacyloxyenamide as thioacylating reagents in SPPS.

RESULTS AND DISCUSSION

In our previous work, we have demonstrated that thioacylating reagents α -thioacyloxyenamides could be obtained as the major product when the addition reaction of monothioamino acids and ynamide (MYTsA) were performed under -40 °C with *m*-xylene as the solvent.²⁶ To systematically evaluate the potency of α -thioacyloxyenamides to incorporate thioamide substitutions site-specifically to a growing peptide backbone on a solid support, all of the α -thioacyloxyenamides (3a-s) originated from 19 of 20 Fmoc-protected proteinogenic α -amino acids were prepared under the optimzed reaction conditions. As shown in Scheme 1, all of the target α thioacyloxyenamides except that of His could be obtained in good yields under the optimized reaction conditions (Scheme 1).²⁶ It should be noted that all of these α -thioacyloxyenamides are stable and can be purified and characterized. No deterioration can be detected after they have been kept in fridge for two months.

With these thioacylating reagents in hand, the systematic and comprehensive optimization of the reaction conditions to incorporate one thioamide unit to the N-terminal of a growing peptide on a solid support was conducted. Initial experiments were carried out with the coupling of α -thioacyloxyenamide derived from Fmoc-protected alanine with 0.03 mmol of tripeptide VGF loaded on trityl chloride resin (2-CTC, which can be cleaved under more milder condition to avoid the side chain reaction and Edman degradation during cleavage of the thioamide substituted peptide from resin^{23d, 27a}) as the model reaction. The effect of solvent, amount of α thioacyloxyenamide, and coupling time on the reaction efficiency have been studied systematically (Table 1). Investigation on solvent disclosed that dimethylformamide (DMF) is the optimal choice (Table 1, entry 6), which is beneficial to suppress epimerization (Table 1, entries 2-4) and the co-formation of oxoamide byproduct efficiently (Table 1, entry 1). Interestingly, decreasing the amount of α thioacyloxyenamide and shortening reaction time from 1 h to 0.5 h have no significant influence on the conversion. On the contrary, prolonged reaction time has deleterious effect on the coupling and resulted more oxoamide byproduct formed. The combination of 3 equiv of α -thioacyloxyenamide, DMF as the solvent, and 1 h coupling time were identified to be the best reaction conditions to incorporate a thioamide substitution to the N-terminal of a growing peptide in SPPS.

Next, the efficiency of incorporating one thioamide substitution to the N-terminal of a tetrapeptide with the tripeptide VGF bearing a bulky Val at the coupling site was evaluated for a-thioacyloxyenamides derived from 19 Fmocprotected proteinogenic α -amino acids. As shown in Table 2, all of them proceeded smoothly to offer the target thioamide substituted tetrapeptide in quantitative conversion. It's notable that amino acids such as Met, Gln, Asn, and Thr, which can not be used for thioacylnitrobenzotriazole method work well in our case (Table 2, entries 3, 7, 9, 13 and 15).² The bulky side-chain protecting groups, such as Boc, 'Bu, and Trt, have no deleterious effect on the conversion (Table 2, entries 4-11). The protected hydrophilic and polar amino acids such as Arg, Asn, Ser, Lys, and Tyr are also compatible with this method (Table 2, entries 20, 21 and Table 3, entries 13-15). Small amount of oxopeptide was formed as the byproduct during the coupling of thiocarbonyl esters of sterically hindered Val and Ile (Table 2, entries 16-17). This might be attributed to the steric hindrance between these bulky residues and bulky Val at the N terminal (data not shown). This phenomenon could be avoided once the bulky Val was replaced by Ala (Table 2, entries 16-17). Low level of epimerization was observed for the thiocarbonyl esters derived from Gln, Asp, Asn, and Ser, which are highly prone to racemization upon activation, under the room temperature conditions (Table 2, entries 7-9, and 11). As shown previously, such slight epimerization could be suppressed by lowering the





^{*a*}These reactions were carried out with **1** (0.15 mmol, 1.5 equiv), **2** (MYTsA, 0.1 mmol), *m*-xylene (1.0 mL), -40 °C, 8 h, Fmoc = fluorenylmethoxy, Boc = tert-butoxycarbonyl, Trt = trityl, 'Bu = tert-butyl, *a*Isolated yield. ^{*b*}CH₂Cl₂, -40 °C, 8 h. ^{*c*}NR = No reaction.

reaction temperature to 0 $^{\circ}C.^{26}$ Scaling up the reaction to 0.4 mmol scale did not alter the coupling efficiency, which showcase the viability of this method (Table 2, entry 1).

The success on the incorporation of monothioamide substitution to the N-terminal of a growing peptide promoted us to challenge the synthesis of multiple thioamide substituted peptides, which involves iterative Fmoc deprotection of endothiopeptide. It is noted that the α -proton of the thioamide (pKa = ca. 13) residue is more acidic than that of the oxoamide and thus it is highly prone to racemization for thioamide residue under the

Table 1. Optimization of the Reaction Conditions^a



^{*a*}Reactions were carried out with 0.03 mmol of **4** at rt. ^{*b*}Cleavage cocktail: TFE:AcOH:CH₂Cl₂ = 1:3:6. ^{*c*}Conversion (**4/5a** ratio) was calculated based on HPLC at 254 nm. ^{*d*}Product diastereomeric purity was determined by HPLC. ^{*e*}8% (oxoamide: thioamide) oxoamide was detected. ^{*f*} DMF/CH₂Cl₂ (v/v) = 1:1. ^{*g*}2% (oxoamide: thioamide) oxoamide was detected.

conventional Fmoc deprotection conditions (piperidine pKa = ca. 10 in DMF).^{4, 27} Petersson and co-works observed that a more sterically hindered and non-nucleophilic base,1,8diazabicyclo[5.4.0]undec-7-ene (DBU), in DMF is beneficial to suppress epimerization during the removal of Fmoc group of thioamide-containing peptides.^{21a, 27a, 28} However, in our hand, iterative exposure of thioamide-containing peptide to such Fmoc deprotection conditions caused racemization during the peptide elongation on a solid support.^{27a, 28a} Fortunately, slight modification of Petersson's deprotection conditions by employing a cocktail of DBU (1%)/n-C₆CH₁₃SH (25%) (v/v in DMF) offered better results. To further illustrate the robustness of our method, the synthesis of a series of dithioamide substituted peptides, including i+1, i+2, and i+3, were studied (Table 3).^{23d, 29} Interestingly, the excellent reaction efficiency could be retained albeit a little longer reaction time was required for the incorporation of the second thioamide substitution. This might be at- tributed to the stronger electron-withdrawing effect of the thio-amide bond than that of oxoamide and thus decreasing the nucleophilicity of the N-terminal amino group of the thioamide-containing peptide. It is found that this electronwithdrawing

Table 2. Synthesis of Mono-Thioamide Substituted Peptides^a

			N_Ts │ (3)		
NH ₂ -V	al-Gly-Phe-	Cleavage coo	→ Fmo	Fmoc-Xaa ^S -Val-Gly-Ph 5	
	entry	Xaa (5)	4/5 ^c	$\mathrm{d}\mathbf{r}^d$	
	1	Ala (5a) ^e	<1:99	>99:1	
	2	Phe (5b)	1:99	99:1	
	3	Met (5c)	1:99	>99:1	
	4	Lys(Boc) (5d)	1:99	>99:1	
	5	Cys(Trt) (5e)	1:99	>99:1	
	6	$\operatorname{Glu}({}^{t}\operatorname{Bu})(\mathbf{5f})$	1:99	>99:1	
	7	Gln(Trt) (5g)	1:99	98:2	
	8	$Asp(^{t}Bu)$ (5h)	<1:99	90:10	
	9	Asn(Trt) (5i)	<1:99	95:5	
	10	Tyr('Bu) (5j)	1:99	99:1	
	11	Ser('Bu) (5k)	<1:99	97:3	
	12	Trp(Boc) (51)	1:99	>99:1	
	13	Gly (5m)	<1:99	>99:1	
	14	Pro (5n)	1:99	>99:1	
	15	Thr(Me) (50)	2:98	>99:1	
	16	Ile (5p) ^{<i>f</i>}	2:98	>99:1	
	17	Val (5q) ^f	2:98	>99:1	
	18	Leu (5r) ^{<i>f</i>}	<1:99	>99:1	
	19	$Arg(Boc)_2(5s)$	2:98	95:5	
	20	Ala (5t) ^{<i>g</i>}	2:98	>99:1	
	21	Glu $(\mathbf{5u})^{h,i}$	<1:99	>99:1	
	22	Ala (5v) ^{j,i}	<1:99	>99:1	

^aReactions were carried with 0.03 mmol of 4 at rt for 1 h. ^bCleavage cocktail: TFE:AcOH:CH₂Cl₂ = 1:3:6. ^cConversion (4/5 ratio) was calculated based on HPLC at 254 nm. ^dProduct diastereomeric purity was determined by HPLC. ^e4 (0.4 mmol) was used. ^fNH₂-A-G-F-resin was used. ^gNH₂-A-R(Pbf)-G-resin was used. ^hNH₂-A-Q(Trt)- W(Boc)-P-K(Boc)-resin was used. ^fThe product peptide was fully deprotected. ^fNH₂-A-Y(^fBu)-N(Trt)-F-resin was used.

induction effect could be interrupted by inserting of the normal amide bond between the two thioamide bonds (Table 3, entries 1, 4, 6, 9, 12 and 13). The reaction time for the introduction of the second thioamide moiety of the i+3 (2 h) was significantly shorter than that of i+1 (4 h), i+2 (3 h). It is expected that dithiopeptide would be highly prone to epimerization and to cause other side reactions. Even though, our protocols gave the required dithiopeptides in excellent conversion (93-99%), and purity (different HPLC methods and HRMS were used to assess the purity of final peptides 7a, 7f, and 7g, for details see SI). No or low level of epimerization was observed for some special amino acid residues. This promising result prompted us to attempt the synthesis of multiple thioamide substituted peptides, the properties of which have not been studied because of their

Table 3. Synthesis of Di- to Multi-Thioamide BondsSubstituted Peptide Sequence^a



0	5 10 15 20 Time (min)	25	30	
entry	sequence	time (h)	6/7 ^c	$\mathrm{d}\mathbf{r}^{d}$
1	A ^s A ^s VGF (7a)	1	1:99	>99:1
2	E('Bu) ^S A ^S VGF (7b)	3	3:97	>99:1
3	S('Bu) ^S A ^S VGF (7c)	1	1:99	92:8
4	M ^s A ^s VGF (7d)	4	4:96	>99:1
5	K(Boc) ^s A ^s VGF (7e)	4	7:93	>99:1
6	A ^s LA ^s VGF (7f)	3	<1:9 9	>99:1
7	F ^s LA ^s VGF (7 g)	1	2:98	99:1
8	K(Boc) ^S LA ^S VGF (7h)	3	6:94	>99:1
9	M ^s LA ^s VGF (7i)	3	4:96	>99:1
10	E('Bu) ^S LA ^S VGF (7j)	5	7:93	>99:1
11	S('Bu) ^S LA ^S VGF (7k)	3	1:99	99:1
12	A ^s LAA ^s VF (71)	2	<1:9 9	>99:1
13	M ^S AMY('Bu) ^S AL (7m)	2	1:99	96:4
14	K(Boc) ^S LK(Boc)A ^S AGF (7n)	2	2:98	>99:1
15	E(^{<i>i</i>} Bu) ^S S(^{<i>i</i>} Bu)VE(^{<i>i</i>} Bu) ^S APFN (Trt)GK(Boc) (70)	2	4:96	96:4
16	$M^{s}A^{s}A^{s}A^{s}A$ (7p)	10	1:99	>99:1
17	A ^s A ^s A ^s A ^s A ^s A ^s A (7q)	5	2:98	>99:1

^aReactions were carried out with 0.03 mmol of **6** at rt for 1-10 h. ^bCleavage cocktail: TFE:AcOH:CH₂Cl₂ = 1:3:6. ^cConversion (**6**/7 ratio) was calculated based on HPLC at 254 nm. ^dProduct diastereomeric purity was determined by HPLC.

difficult availability. Multiple thioamide substituted pep tides are frequently observed in natural products such as thioviridamide family, which contains five continuous thioamide bonds and possesses highly potent pro-apoptotic activities and potent antiproliferative.³⁰ However, chemical synthesis of multiple thioamide substituted peptides on a solid support have never been reported yet. To our delight, α thioacyloxyenamides proved to be effective thioacylating

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reagents to incorporate multiple thioamide substitutions into the peptide backbone in SPPS. As shown in Table 3, tetrathioamide substituted peptide could be easily synthesized by using our method (Table 3, entry 16). Even the fully thioamide substituted homooligo-hexapeptides containing five continuous thioamide bonds, the oxoamide congener of which is highly hydrophobic and has a high tendency to aggregate could also be synthesized on a solid support in an excellent yield by iterative using of α -thioacyloxyenamide derived from Fmoc-protected monothio-Alanine as the thioacylating reagent (Table 3, entry 17).^{28c, 31}

CONCLUSION

In conclusion, we have provided a highly efficient method for site-specific incorporating thioamide substitutions to the backbone of a growing peptide on a solid support. Thiocarbonyl esters of 19 of 20 proteinogenic α -amino acids except His worked well to offer the target thioamide substituted peptides with no or low level of epimerization. Coupling conditions, a modified Fmoc deprotection method and the feasibility of incorporation of multiple thioamide substitutions have been systematically studied. The thioamide substituted hexapeptide containing up to five continuous thioamide bonds could also be synthesized by using this method. This study paves the way for the application of α -thioacyloxyenamides as a kind of thioacylating reagents to incorporate a thioamide substitution to a peptide backbone in a site-specific manner. No doubt, the novel thioacylating strategy reported herein will spur the further application of thioamide substitution tool in peptide and protein chemical biology.

EXPERIMENTAL SECTION

¹H and ¹³C NMR spectra were recorded on a Bruker (400 and 100 MHz for ¹H and ¹³C, respectively) instrument, and are internally referenced to residual solvent signals, CDCl3 referenced at δ 7.26 and 77.06 ppm. Data for ¹H is reported as follows: chemical shift (δ ppm), integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), broad peaks (br), coupling constant (Hz) and assignment. Data for ¹³C NMR are reported in terms of chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constant (Hz) and no special nomenclature is used for equivalent carbons. HRMS (ESI) spectra were obtained by the electrospray ionization time-of-flight (ESI-TOF) mass spectrometry. The diastereomeric ratio (dr) was determined by HPLC with C18 column with H₂O and acetonitrile (ACN) as eluent. Flash column chromatography purification of compound was carried out by gradient elution using ethyl acetate (EA) in light petroleum ether (PE). Unless otherwise noted, materials obtained from commercial suppliers were used without further purification. Analytical RP-HPLC was performed on an UltiMate 3000 HPLC system using commercially available columns (Phenomenex C18, 250 mm x 4.6 mm, 5 µm) at a flow rate of 1.0 mL/min, detection at 220 or 254 nm. Mobile phase: solvent A was 0.045% TFA (v/v) in H₂O, solvent B was 10% H₂O and 0.039% TFA (v/v) in ACN. Gradient: a suitably adjusted gradient of 20% B to 100% B was used.

General Procedure 1 for α -thioacyloxy enamides (3a-s).

To a stirred solution of the appropriate Fmoc protected amino acid (5.0 mmol, 1.0 equiv) and *N*-hydroxysuccinimide (HOSu, 5.5 mmol, 1.1 equiv) in CH_2Cl_2 (20.0 mL), dicyclohexylcarbodiimide (DCC, 5.5 mmol, 1.1 equiv) was added at 25 °C under N₂. Then the mixture was stirred at 25 °C for 2 h. The suspension was filtered to remove the resulting white solid which was washed with 20.0 mL of CH₂Cl₂ repeatedly. The filtrate was concentrated about one-fifth of its original volume and left to stand in the refrigerator for about 2 h. Then the mixture was filtered again and concentrated to give the corresponding activated ester as a colorless foam, which was immediately dissolved in CH₂Cl₂ (20.0 mL), followed by addition of NaHS (15 mmol, 3.0 equiv, 70% purity) and 15-crown-5 (1.5 mmol, 0.3 equiv). Then the reaction mixture was stirred at rt for 3 h, diluted with CH₂Cl₂ and poured into ice water. The pH was adjusted to ~3 by careful addition of 1 M HCl and extracted by CH₂Cl₂ (3 × 20.0 mL). The combined organic layers were washed with brine, dried over Na₂SO₄ and concentrated under reduced pressure. The crude products **2** were used for the next reaction without further purification.^{25a, 26a, 32}

A 5.0 mL round-bottomed flask was charged with MYTsA (0.1 mmol, 1.0 equiv), *m*-xylene (1.0 mL) and **2** (0.15 mmol, 1.5 equiv). The reaction mixture was stirred at -40 °C under air until starting material MYTsA was fully consumed. The reaction mixture was concentrated and purified by silica gel chromatography to afford the α -thioacyloxyenamides **3**. The products **3a**, **3b**, **3m**, **3p**, **3q** and **3r** have been synthesized and characterized in the previous work.²⁵ The data given here are provided for convenience.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-4-(methylthio)

butanethioate (*3c*). Yellow solid (41 mg, 68%); mp 46-50 °C; $R_f = 0.4$ (PE/EA = 5:1); [a]25 D= + 40.8 (c = 0.87, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.77 (d, J = 7.5 Hz, 2H), 7.72 (d, J = 8.2 Hz, 2H), 7.66 – 7.59 (m, 2H), 7.40 (t, J = 7.5 Hz, 2H), 7.35 – 7.28 (m, 4H), 5.67 (d, J = 9.0 Hz, 1H), 4.89 (d, J = 3.0 Hz, 1H), 4.83 – 4.71 (m, 1H), 4.61 (d, J = 3.0 Hz, 1H), 4.48 – 4.42 (m, 1H), 4.43 – 4.33 (m, 1H), 4.25 (t, J = 7.0 Hz, 1H), 3.01 (s, 3H), 2.67 – 2.50 (m, 2H), 2.43 (s, 3H), 2.38 – 2.23 (m, 1H), 2.13 (s, 3H), 2.03 – 1.89 (m, 1H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 218.4, 155.8, 150.0, 144.5, 143.9, 143.8, 141.3, 132.7, 129.6, 128.2, 127.7, 127.1, 125.2, 120.0, 101.7, 67.1, 61.4, 47.2, 38.3, 34.1, 30.0, 21.6, 15.4 ppm; IR (KBr) $\tilde{\nu}$ 3381, 2920, 1723, 1357, 1162, 743 cm⁻¹; HRMS (ESI-TOF) calcd for C₃₀H₃₃N₂O₅S₃ (M + H)⁺: 597.1546, found: 597.1540.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9Hfluoren-9-yl)methoxy)carbonyl)amino)-6-((tert-butoxycarbonyl) amino)hexanethioate (3d). Yellow solid (44 mg, 62%); mp 55-58 °C; $R_f = 0.4$ (PE/EA = 3:1); $[\alpha]25 D = +29.0$ (c = 0.7, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.75 (d, J = 7.5 Hz, 2H), 7.71 (d, J = 8.0 Hz, 2H), 7.62 (t, J = 5.4 Hz, 2H), 7.39 (t, J = 7.5 Hz, 2H), 7.35 - 7.28 (m, 4H), 5.69 (d, J = 9.0 Hz, 1H), 4.88 (d, J = 3.1 Hz, 1H), 4.74 (s, 1H), 4.67 – 4.55 (m, 2H), 4.45 (dd, J = 10.4, 7.0 Hz, 1H), 4.33 (dd, J = 10.6, 7.2 Hz, 1H), 4.23 (t, J = 7.1 Hz, 1H), 3.17 -3.06 (m, 2H), 3.00 (s, 3H), 2.41 (s, 3H), 1.97 - 1.88 (m, 1H), 1.82 - 1.68 (m, 1H), 1.57 - 1.37 (m, 13H); ¹³C{¹H} NMR (100 MHz, CDCl₃) & 219.2, 156.2, 155.8, 150.0, 144.5, 143.9, 143.8, 141.3, 132.6, 129.6, 128.2, 127.7, 127.1, 125.2, 120.0, 101.8, 79.0, 67.0, 61.9, 47.2, 40.1, 38.3, 34.2, 29.6, 28.5, 22.4, 21.6 ppm; IR (KBr) v 3399, 2933, 1717, 1514, 1357, 1156, 737 cm⁻¹; HRMS (ESI-TOF) calcd for C₃₆H₄₃N₃NaO₇S₂ (M + Na)⁺: 716.2435, found: 716.2429.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9Hfluoren-9-yl)methoxy)carbonyl)amino)-3-(trityl thio) propanethioate (3e). Yellow solid (45 mg, 55%); mp 69-73 °C; R_f = 0.4 (PE/EA = 5:1); $[\alpha]$ 25 D= + 40.2 (c = 0.60, CHCl₃); ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3) \delta 7.80 - 7.73 \text{ (m, 2H)}, 7.69 - 7.58 \text{ (m, 4H)}, 7.45$ -7.35 (m, 8H), 7.32 - 7.16 (m, 13H), 5.36 (d, J = 8.9 Hz, 1H), 4.85(d, J = 2.9 Hz, 1H), 4.75 (d, J = 3.0 Hz, 1H), 4.53 - 4.39 (m, 2H),4.38 - 4.28 (m, 1H), 4.23 (t, J = 7.0 Hz, 1H), 2.94 (s, 3H), 2.76 - 4.28 (m, 1H), 4.23 (t, J = 7.0 Hz, 1H), 2.94 (s, 3H), 2.76 - 4.282.60 (m, 2H), 2.37 (s, 3H); ${}^{13}C{}^{1}H$ NMR (100 MHz, CDCl₃) δ 215.8, 155.3, 149.3, 144.4, 144.3, 143.9, 143.8, 141.3, 129.6, 129.6, 128.1, 128.1, 127.7, 127.1, 126.9, 125.2, 120.0, 101.9, 67.3, 67.0, 60.7, 47.2, 37.4, 35.6, 21.6 ppm; IR (KBr) v 3405, 3051, 1728, 1498, 1357, 1163, 743 cm⁻¹; MS (ESI-TOF) calcd for $C_{47}H_{43}N_2O_5S_3 (M + H)^+$: 811.2329, found: 811.2318.

tert-butyl (S)-4-((((9H-fluoren-9-yl)methoxy)carbonyl) amino)-5-((1-((N,4-dimethylphenyl)sulfonamido)vinyl)oxy) -5-thioxopen tanoate (3f). Yellow solid (42 mg, 63%); mp 38-43 °C; $R_f = 0.4$ $(PE/EA = 5:1); [\alpha]_{25} D = +50.7 (c = 0.56, CHCl_3); {}^{1}H NMR (400)$ MHz, CDCl₃) δ 7.75 (d, *J* = 7.5 Hz, 2H), 7.71 (d, *J* = 8.2 Hz, 2H), 7.62 (t, J = 6.7 Hz, 2H), 7.38 (t, J = 7.5 Hz, 2H), 7.34 - 7.26 (m, 4H), 5.79 (d, J = 8.9 Hz, 1H), 4.88 (d, J = 3.0 Hz, 1H), 4.65 (d, J = 3.0 Hz, 1H), 4.64 – 4.57 (m, 1H), 4.45 (dd, *J* = 10.4, 7.0 Hz, 1H), 4.32 (dd, J = 10.5, 7.3 Hz, 1H), 4.22 (t, J = 7.1 Hz, 1H), 2.99 (s, 3H), 2.46 – 2.32 (m, 5H), 2.31 – 2.20 (m, 1H), 2.07 – 1.93 (m, 1H), 1.46 (s, 9H); ${}^{13}C{}^{1}H$ NMR (100 MHz, CDCl₃) δ 218.2, 172.2, 155.8, 149.8, 144.5, 144.0, 143.8, 141.3, 132.9, 129.6, 128.2, 127.7, 127.1, 125.2, 120.0, 101.7, 80.8, 67.1, 61.8, 47.2, 38.0, 31.6, 29.4, 28.1, 21.6 ppm; IR (KBr) \tilde{v} 3375, 2915, 1717, 1363, 1079, 743 cm⁻¹; HRMS (ESI-TOF) calcd for $C_{34}H_{38}N_2NaO_7S_2$ (M + Na)⁺: 673.2013, found: 673.2009.

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14 O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9Hfluoren-9-yl)methoxy)carbonyl)amino)-5-oxo-5-(tritylamino)pen 15 tanethioate (3g). Yellow solid (49 mg, 59%); mp 82-86 °C; $R_f =$ 16 0.3 (PE/EA = 3:1); $[\alpha]$ 25 D= + 58.6 (c = 0.6, CHCl₃); ¹H NMR 17 $(400 \text{ MHz}, \text{CDCl}_3) \delta 7.74 \text{ (d}, J = 7.5 \text{ Hz}, 2\text{H}), 7.65 - 7.55 \text{ (m}, 4\text{H}),$ 18 7.40 - 7.33 (m, 2H), 7.30 - 7.26 (m, 4H), 7.25 - 7.21 (m, 11H), 19 7.20 - 7.15 (m, 4H), 5.91 (d, J = 8.4 Hz, 1H), 4.88 (d, J = 3.1 Hz, 20 1H), 4.67 (q, J = 6.9 Hz, 1H), 4.47 (d, J = 3.1 Hz, 1H), 4.42 (dd, J = 10.4, 7.1 Hz, 1H), 4.29 (dd, J = 10.4, 7.3 Hz, 1H), 4.21 (t, J = 7.221 Hz, 1H), 3.00 (s, 3H), 2.52 - 2.35 (m, 5H), 2.25 (t, J = 7.4 Hz, 2H); 22 $^{13}C{^{1}H}$ NMR (100 MHz, CDCl₃) δ 218.2, 171.2, 155.9, 150.5, 23 144.7, 144.7, 144.0, 143.8, 141.3, 129.7, 128.8, 128.3, 127.9, 127.7, 24 127.2, 126.9, 125.3, 120.0, 101.5, 70.6, 67.2, 61.5, 47.2, 38.8, 33.0, 25 30.1, 21.7 ppm; IR (KBr) v 3387, 3063, 1723, 1498, 1351, 1156, 26 737 cm⁻¹; MS (ESI-TOF) calcd for $C_{49}H_{46}N_3O_6S_2$ (M + H)⁺: 27 836.2823, found: 836.2836.

tert-butyl (S)-3-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-4-((1-((N,4-dimethylphenyl)sulfonamido)vinyl)oxy)-4-thioxo

butanoate (**3h**). Yellow solid (32 mg, 50%); mp 40-45 °C; $R_f = 0.2$ (PE/EA = 10:1); [α]25 D= + 26.8 (c = 1.65, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.77 (d, J = 7.6 Hz, 2H), 7.72 (d, J = 8.3 Hz, 2H), 7.64 (t, J = 8.3 Hz, 2H), 7.40 (t, J = 7.5 Hz, 2H), 7.35 – 7.28 (m, 4H), 5.99 (d, J = 9.3 Hz, 1H), 4.87 (d, J = 2.9 Hz, 1H), 4.85 – 4.77 (m, 1H), 4.68 (d, J = 3.1 Hz, 1H), 4.46 (dd, J = 10.2, 7.0 Hz, 1H), 4.33 (dd, J = 10.3, 7.4 Hz, 1H), 4.26 (t, J = 7.1 Hz, 1H), 3.01 (s, 3H), 2.89 (d, J = 5.6 Hz, 2H), 2.42 (s, 3H), 1.46 (s, 9H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 216.4, 169.2, 155.7, 149.8, 144.4, 143.9, 143.8, 141.3, 129.6, 128.2, 127.7, 127.1, 125.2, 120.0, 101.6, 81.9, 67.3, 58.6, 47.2, 39.6, 37.9, 28.1, 21.6 ppm; IR (KBr) \tilde{v} 3381, 3063, 2974, 1734, 1357, 1162, 737 cm⁻¹; MS (ESI-TOF) calcd for C₃₃H₃₆N₂NaO₇S₂ (M +Na)+: 659.1856, found: 659.1865. O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9H-

fluoren-9-yl)methoxy)carbonyl)amino)-4-oxo-4-(tritylamino) butanethioate (*3i*). Yellow solid (33 mg, 40%); mp 81-83 °C; $R_f = 0.2$ (PE/EA = 3:1); $[\alpha]25 D = + 70.0$ (c = 0.65, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.73 (d, J = 7.5 Hz, 2H), 7.66 (d, J = 7.9 Hz, 2H), 7.61 (t, J = 8.0 Hz, 2H), 7.37 (t, J = 7.6 Hz, 2H), 7.24 (m, 20H), 6.98 (d, J = 8.7 Hz, 1H), 4.81 (m, 2H), 4.46 – 4.31 (m, 2H), 4.18 (m, 2H), 3.25 (dd, J = 15.1, 4.4 Hz, 1H), 3.06 – 2.91 (m, 3H), 2.87 (dd, J = 15.0, 5.0 Hz, 1H), 2.40 (s, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 216.2, 168.8, 156.1, 150.1, 144.6, 144.3, 144.0, 143.9, 141.2, 132.0, 129.6, 128.8, 128.2, 127.9, 127.6, 127.1, 127.0, 125.4, 119.9, 101.1, 71.0, 67.3, 59.1, 47.1, 39.0, 38.5, 21.6 ppm; IR (KBr) \tilde{v} 3381, 2927, 2850, 1663, 1462, 1376, 1161, 739 cm⁻¹; MS (ESI-TOF) calcd for C₄₈H₄₄N₃O₆S₂ (M + H)⁺: 822.2666, found: 822.2660.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-3-(4-(tert-butoxy)phenyl) propanethioate (*3j*). Yellow solid (36 mg, 65%); mp 54-57 °C; $R_f = 0.4$ (PE/EA = 5:1); $[\alpha]25 D = + 40.4$ (c = 0.67, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.83 – 7.66 (m, 5H), 7.59 (t, *J* = 7.4 Hz, 2H),

7.40 (t, J = 7.4 Hz, 2H), 7.36 – 7.28 (m, 5H), 7.09 (d, J = 8.1 Hz, 2H), 6.92 (d, J = 8.3 Hz, 2H), 5.52 (d, J = 8.9 Hz, 1H), 4.87 (q, J = 6.8 Hz, 1H), 4.82 (d, J = 2.6 Hz, 1H), 4.68 (d, J = 2.6 Hz, 1H), 4.43 – 4.32 (m, 2H), 4.21 (t, J = 7.1 Hz, 1H), 3.21 (dd, J = 13.9, 5.6 Hz, 1H), 3.07 – 2.98 (m, 4H), 2.41 (s, 3H), 1.32 (s, 9H); ¹³C {¹H} NMR (100 MHz, CDCl₃) δ 217.7, 155.3, 154.4, 149.7, 144.5, 143.8, 141.3, 133.0, 130.5, 130.0, 129.6, 128.2, 127.7, 127.1, 125.2, 124.1, 120.0, 102.0, 78.4, 67.0, 62.7, 47.2, 39.9, 38.0, 28.9, 21.6 ppm; IR (KBr) $\tilde{\nu}$ 3375, 2986, 1728, 1363, 1168, 743 cm⁻¹; MS (ESI-TOF) calcd for C₃₈H₄₀N₂NaO₆S₂ (M + Na)⁺: 707.2220, found: 707.2233.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9Hfluoren-9-yl)methoxy)carbonyl)amino)-3-(tert-butoxy)propane thioate (3k). Yellow grass solid (40 mg, 63%); $R_f = 0.4$ (PE/EA = 5:1); $[\alpha]25 D = +10.5$ (c = 0.65, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.77 (d, J = 7.5 Hz, 2H), 7.73 (d, J = 8.3 Hz, 2H), 7.66 (dd, J = 11.1, 7.5 Hz, 2H), 7.40 (t, J = 7.5 Hz, 2H), 7.37 - 7.27 (m)4H), 5.83 (d, J = 9.1 Hz, 1H), 4.89 (d, J = 2.8 Hz, 1H), 4.82 (d, J =2.8 Hz, 1H), 4.73 - 4.59 (m, 1H), 4.48 (dd, J = 10.6, 7.0 Hz, 1H), 4.36 (dd, J = 10.5, 7.3 Hz, 1H), 4.28 (t, J = 7.2 Hz, 1H), 3.83 (dd, J = 9.1, 3.6 Hz, 1H), 3.66 (dd, J = 9.1, 4.0 Hz, 1H), 3.01 (s, 3H), 2.41 (s, 3H), 1.16 (s, 9H); ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 216.9, 155.8, 149.4, 144.3, 144.0, 143.9, 141.3, 133.6, 129.6, 128.2, 127.7, 127.1, 125.2, 120.0, 102.2, 73.6, 67.2, 63.5, 62.3, 47.2, 37.3, 27.4, 21.6 ppm; IR (KBr) v 3381, 2969, 1728, 1363, 1162, 737 cm⁻¹; HRMS (ESI-TOF) calcd for $C_{32}H_{36}N_2NaO_6S_2$ (M + Na)⁺: 631.1907, found: 631.1908.

tert-butvl (S)-2-(2-((((9H-fluoren-9-yl)methoxy)carbonyl) amino)-3-((1-((N,4-dimethylphenyl)sulfonamido)vinyl)oxy)-3thioxo propyl)-1H-indole-1-carboxylate (31). Yellow solid (48 mg, 62%); mp 55-59 °C; $R_f = 0.5$ (PE/EA = 5:1); [α]25 D= + 40.9 (c = 0.77, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 8.14 (d, J = 8.2 Hz, 1H), 7.78 – 7.67 (m, 4H), 7.61 (d, J = 7.8 Hz, 1H), 7.58 – 7.46 (m, 3H), 7.42 - 7.34 (m, 2H), 7.33 - 7.22 (m, 6H), 5.61 (d, J = 8.9 Hz, 1H), 4.99 (q, J = 7.0 Hz, 1H), 4.80 (d, J = 3.0 Hz, 1H), 4.63 (d, J =3.0 Hz, 1H, 4.32 (g, J = 10.3, 8.9 Hz, 2H), 4.19 (t, J = 7.3 Hz, 1H),3.41 (dd, J = 14.8, 5.7 Hz, 1H), 3.20 (dd, J = 14.6, 7.2 Hz, 1H), 3.01 (s, 3H), 2.39 (s, 3H), 1.65 (s, 9H); ¹³C{¹H} NMR (100 MHz, CDCl₃) & 217.6, 155.5, 149.9, 149.6, 144.5, 143.8, 141.3, 135.5, 132.8, 130.6, 129.6, 128.2, 127.7, 127.1, 125.2, 124.6, 124.5, 122.7, 119.9, 119.0, 115.3, 114.8, 101.7, 83.7, 67.2, 62.0, 47.1, 38.1, 30.1, 28.2, 21.6 ppm; IR (KBr) v 3387, 2920, 1734, 1357, 1168, 731 cm⁻¹; HRMS (ESI-TOF) calcd for $C_{41}H_{41}N_3NaO_7S_2$ (M + Na)⁺: 774.2278, found: 774.2274.

(9H-fluoren-9-yl)methyl (S)-2-(((1-((N,4-dimethylphenyl) sulfonamido)vinyl)oxy) carbonothioyl)pyrrolidine-1-carboxylate (3*n*). Yellow grass solid (40 mg, 69%); $R_f = 0.5$ (PE/EA = 5:1); $[\alpha]$ 25 D= + 12.9 (c = 0.50, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.83 - 7.50 (m, 6H), 7.44 - 7.05 (m, 6H), 4.90 (d, J = 2.6 Hz, 0.5H),4.82 - 4.74 (m, 1H), 4.72 (d, J = 2.7 Hz, 0.5H), 4.57 (d, J = 2.7 Hz, 0.5H), 4.54 - 4.39 (m, 1.5H), 4.36 - 4.07 (m, 2H), 3.80 - 3.63 (m, 1H), 3.63 – 3.48 (m, 1H), 2.98 (m, 3H), 2.38 (m, 3H), 2.24 (m, 2H), 2.07 (m, 1H), 1.98 - 1.82 (m, 1H); ${}^{13}C{}^{1}H{}$ NMR (100 MHz, CDCl₃) & 219.2, 219.0, 154.7, 154.4, 150.4, 150.0, 144.5, 144.4, 144.3, 143.9, 143.8, 141.3, 141.3, 141.2, 132.7, 132.4, 129.6, 129.6, 128.3, 128.3, 127.7, 127.7, 127.1, 127.1, 125.3, 125.2, 125.2, 125.1, 120.0, 120.0, 101.7, 101.5, 68.2, 67.9, 67.5, 67.4, 47.4, 47.3, 47.2, 46.7, 38.5, 38.2, 33.0, 31.8, 23.8, 22.8, 21.6, 21.6 ppm; IR (KBr) \tilde{v} 3068, 2927, 1705, 1363, 1162, 743 cm⁻¹; HRMS (ESI-TOF) calcd for $C_{30}H_{30}N_2NaO_5S_2$ (M + Na)⁺: 585.1488, found: 585.1484.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (2S,3R)-2-((((9H-fluoren-9-yl)methoxy)carbonyl)amino)-3-methoxybutane thioate (*3o*). Yellow solid (40 mg, 66%); mp 48-50 °C; $R_f = 0.4$ (PE/EA = 8:1); [α]25 *D*= + 47.8 (c = 0.88, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 7.77 (d, *J* = 7.6 Hz, 2H), 7.74 (d, *J* = 8.3 Hz, 2H), 7.69 (d, *J* = 7.4 Hz, 1H), 7.65 (d, *J* = 7.4 Hz, 1H), 7.45 – 7.37 (m,

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2H), 7.36 - 7.29 (m, 4H), 5.76 (d, J = 9.9 Hz, 1H), 4.92 (d, J = 2.9Hz, 1H), 4.67 (d, J = 2.9 Hz, 1H), 4.54 – 4.43 (m, 2H), 4.37 – 4.24 (m, 2H), 4.13 (qd, J = 6.2, 2.1 Hz, 1H), 3.33 (s, 3H), 3.01 (s, 3H),2.43 (s, 3H), 1.25 (d, J = 6.3 Hz, 3H); ¹³C{¹H} NMR (100 MHz, CDCl₃) & 217.2, 156.6, 150.0, 144.4, 144.0, 143.8, 141.3, 132.9, 129.6, 128.3, 127.7, 127.1, 125.3, 120.0, 102.5, 76.9, 67.2, 66.2, 57.2, 47.2, 38.1, 21.6, 16.6 ppm; IR (KBr) v 3428, 3057, 2933, 1723, 1357, 1085, 731 cm⁻¹; HRMS (ESI-TOF) calcd for $C_{30}H_{32}N_2NaO_6S_2 (M + Na)^+: 603.1594$, found: 603.1591.

O-(1-((N,4-dimethylphenyl)sulfonamido)vinyl) (S)-2-((((9Hfluoren-9-yl)methoxy) carbonyl)amino)-5-(1,3-bis(tert-butoxy carbonyl)guanidino)pentanethioate (3s). Yellow solid (45 mg, 10 55%); mp 66-69 °C; $R_f = 0.4$ (PE/EA = 3:1); $[\alpha]25 D = +41.5$ (c = 11 0.48, CHCl₃); ¹H NMR (400 MHz, CDCl₃) δ 11.52 (s, 1H), 8.36 (t, 12 J = 5.6 Hz, 1H), 7.75 (d, J = 7.6 Hz, 2H), 7.71 (d, J = 7.9 Hz, 2H), 13 7.64 (t, J = 9.0 Hz, 2H), 7.39 (t, J = 7.4 Hz, 2H), 7.35 – 7.27 (m, 14 4H), 5.92 (d, J = 9.0 Hz, 1H), 4.87 (d, J = 3.1 Hz, 1H), 4.68 (d, J =3.0 Hz, 1H), 4.64 – 4.53 (m, 1H), 4.50 – 4.33 (m, 2H), 4.23 (t, J = 15 6.9 Hz, 1H), 3.51 (dd, J = 12.8, 6.2 Hz, 1H), 3.39 (dd, J = 12.7, 6.1 16 Hz, 1H), 2.99 (s, 3H), 2.41 (s, 3H), 2.03 - 1.91 (m, 1H), 1.79 - 1.59 17 (m, 3H), 1.50 (s, 9H), 1.48 (s, 9H); ¹³C{¹H} NMR (100 MHz, 18 CDCl₃) § 218.6, 163.6, 156.3, 155.8, 153.3, 149.7, 144.4, 144.0, 19 143.8, 141.3, 133.0, 129.6, 128.2, 127.7, 127.1, 125.2, 119.9, 20 101.8, 83.1, 79.2, 67.0, 62.1, 47.2, 40.3, 37.9, 31.4, 28.3, 28.1, 25.6, 21.6 ppm; IR (KBr) v 3328, 2974, 1728, 1640, 1357, 1126, 743 21 cm^{-1} ; MS (ESI-TOF) calcd for C₄₁H₅₂N₅O₉S₂ (M + H)⁺: 822.3201, 22 found: 822.3195. 23

General Procedure 2 for Incorporation of the Monothioamide on a Solid Support.

25 Normal peptide assembly by Fmoc based SPPS (4). 2-CTC 26 resin (39 mg, 0.03 mmol) with a loading of 0.77 mmol/g, was 27 placed into 5.0 mL fritted syringes, the resin was swelled in CH₂Cl₂ (3.0 mL, 30 min). The syringe was then drained, Fmoc-Phe-OH (35 28 mg, 3.0 equiv), N,N-Diisopropyl-ethylamin (DIPEA, 10.0 µL, 2.0 29 equiv) and 3.0 mL DMF/CH₂Cl₂ (v/v 1:1) were added into the 30 fritted syringe. The loading reaction was left to react 2 h at rt before 31 being drained. Then, the resin was washed with DMF (3.0 mL x 4)32 and CH₂Cl₂ (3.0 mL x 4). The solution of 20% piperidine in DMF 33 (3.0 mL) was added to the fritted syringes to react 2 min before being drained. A further 3.0 mL of 20% piperidine was added to 34 react 18 min. Then, the resin was again carefully washed with DMF 35 (3.0 mL x 4) and CH_2Cl_2 (3.0 mL x 4). The peptide elongation was 36 carried out by adding a solution of Fmoc protected amino acid (3.0 37 equiv), HBTU (35 mg, 3.0 equiv) and DIPEA (30.0 µL, 6.0 equiv) 38 in 3.0 mL DMF/CH₂Cl₂ (1:1). After 1 h, the resin was washed with 39 DMF (3.0 mL x 4) and CH₂Cl₂ (3.0 mL x 4). The resin-bound peptide 4 was obtained by using standard SPPS protocols as 40 described above.23d, 26a, 28a 41

Synthesis of the Thioamide Substituted Tetrapeptide (5). Prior to the thioamide coupling, the 4 (0.03 mmol) was washed with DMF. The solution of 3a (3.0 equiv) in DMF (3.0 mL) was taken into the fritted syringes. The reaction was shaken at rt for 1 h. Then, the solvent was drained off and the resin was thoroughly washed with DMF (3.0 mL x 4) and CH_2Cl_2 (3.0 mL x 4). The peptide 5a was obtained after treat 10 mg resin with 100.0 μ L of cleavage cocktail (2,2,2-Trifluoroethanol (TFE)/AcOH/CH₂Cl₂ - 1:3:6) for 30 min. The CH₂Cl₂ was removed from the cocktail. Then the solution of water/acetonitrile (v/v 1:1, 200.0 µL) was added to the residue. The resin was filtered off and the filtrates was analysed by HPLC and MS. Peptide 5u and 5v were obtained after treat 30 mg resin with 100.0 µL of cleavage cocktail (TFA/TIPS/Thioanisole/ $H_2O = 95:2.5:2.5:2.5$) for 1 h. The cleavage mixture was evaporated to an oil and then cold ether was added. The precipitate was collected and analysed by HPLC and MS.

Synthesis of Fmoc-Ala^S-Val-Gly-Phe-OH (5a). Peptide 5a was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.85% conversion of crude HPLC traces; $t_{\rm R} = 12.513$ min; MS (ESI-TOF) calcd for $C_{34}H_{39}N_4O_6S (M + H)^+$: 631.26, found 630.81.

Synthesis of Fmoc-Phe^S-Val-Gly-Phe-OH (5b). Peptide 5b was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.60% conversion of crude HPLC traces; $t_{\rm R} = 12.091 \text{ min (major)}; \text{ MS (ESI-TOF) calcd for } C_{40}H_{43}N_4O_6S (M$ + H)⁺: 707.29, found 706.97.

Synthesis of Fmoc-Met^S-Val-Gly-Phe-OH (5c). Peptide 5c was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.00% conversion of crude HPLC traces; $t_R = 10.260 \text{ min (major)}; \text{ MS (ESI-TOF) calcd for } C_{36}H_{43}N_4O_6S_2$ $(M + H)^+$: 691.26, found 690.96.

Synthesis of Fmoc-Lys(Boc)^S-Val-Gly-Phe-OH (5d). Peptide 5d was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.03% conversion of crude HPLC traces; $t_{R} = 11.918$ min; MS (ESI-TOF) calcd for $C_{42}H_{54}N_{5}O_{8}S (M + H)^{+}$: 788.37, found 788.18.

Synthesis of Fmoc-Cys(Trt)^S-Val-Gly-Phe-OH (5e). Peptide 5e was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.02% conversion of crude HPLC traces; $t_{R} = 24.000 \text{ min}; \text{ MS (ESI-TOF) calcd for } C_{53}H_{51}N_{4}O_{6}S_{2}(M - H)^{-1}$ 903.33, found 903.14.

Synthesis of Fmoc-Glu('Bu)^S-Val-Gly-Phe-OH (5f). Peptide 5f was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.80% conversion of crude HPLC traces; $t_R = 13.185 \text{ min}; \text{ MS (ESI-TOF) calcd for } C_{40}H_{47}N_4O_8S (M - H)^-$: 743.31, found 743.45.

Synthesis of Fmoc-Gln(Trt)^S-Val-Gly-Phe-OH (5g). Peptide 5g was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.70% conversion of crude HPLC traces; $t_{\rm R} = 14.013$ min (major); MS (ESI-TOF) calcd for C₅₅H₅₆N₅O₇S (M + H)+: 930.39, found 930.23.

Synthesis of Fmoc-Asp('Bu)^S-Val-Gly-Phe-OH (5h). Peptide 5h was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.23% conversion of crude HPLC traces; $t_R = 18.360 \text{ min (major)}; \text{ MS (ESI-TOF) calcd for } C_{39}H_{47}N_4O_8S (M$ + H)⁺: 731.31, found 730.91.

Synthesis of Fmoc-Asn(Trt)^S-Val-Gly-Phe-OH (5i). Peptide 5i was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.19% conversion of crude HPLC traces; $t_{\rm R} = 18.318$ min (major); MS (ESI-TOF) calcd for C₅₄H₅₃N₅O₇S (M + H)⁺: 916.37, found 916.01.

Synthesis of Fmoc-Tyr('Bu)^S-Val-Gly-Phe-OH (5j). Peptide 5j was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 40-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.78% conversion of crude HPLC traces; $t_R = 24.581 \text{ min (major)}; \text{ MS (ESI-TOF) calcd for } C_{44}H_{49}N_4O_7S (M$ - H)-: 777.33, found 777.45.

Synthesis of Fmoc-Ser(^tBu)^S-Val-Gly-Phe-OH (*5k*). Peptide **5k** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.65% conversion of crude HPLC traces; $t_R = 13.370 \text{ min (major)}$; MS (ESI-TOF) calcd for $C_{38}H_{47}N_4O_7S$ (M + H)⁺: 703.32, found 703.12.

Synthesis of Fmoc-Trp(Boc)^S-Val-Gly-Phe-OH (*51*). Peptide **51** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 70-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.17% conversion of crude HPLC traces; $t_R = 14.834$ min; MS (ESI-TOF) calcd for $C_{47}H_{52}N_5O_8S$ (M + H)⁺: 846.35, found 846.33.

Synthesis of Fmoc-Gly^S-Val-Gly-Phe-OH (*5m*). Peptide **5m** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 40-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.58% conversion of crude HPLC traces; $t_R = 15.706$ min; MS (ESI-TOF) calcd for $C_{33}H_{37}N_4O_6S$ (M - H)⁻: 615.23, found 615.67.

Synthesis of Fmoc-Pro^S-Val-Gly-Phe-OH (*5n*). Peptide **5n** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.85% conversion of crude HPLC traces; $t_R = 18.054$ min; MS (ESI-TOF) calcd for $C_{35}H_{39}N_4O_6S$ (M - H)⁻: 655.26, found 655.24.

Synthesis of Fmoc-Thr(Me)^S-Val-Gly-Phe-OH (*5o*). Peptide **5o** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 40-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.72% conversion of crude HPLC traces; $t_R = 18.833$ min; MS (ESI-TOF) calcd for $C_{36}H_{43}N_4O_7S$ (M + H)⁺: 673.27, found 673.20.

Synthesis of Fmoc-Ile^S-Ala-Gly-Phe-OH (*5p*). Peptide **5p** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 97.22% conversion of crude HPLC traces; $t_R = 14.701$ min; MS (ESI-TOF) calcd for $C_{35}H_{39}N_4O_6S$ (M - H)⁻: 643.26, found 643.36.

Synthesis of Fmoc-Val^S-Ala-Gly-Phe-OH (*5q*). Peptide **5q** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 97.94% conversion of crude HPLC traces; $t_R = 13.036$ min; MS (ESI-TOF) calcd for $C_{34}H_{36}N_4O_6S$ (M - H)⁻: 629.24, found 629.20.

Synthesis of Fmoc-Leu^S-Ala-Gly-Phe-OH (*5r*). Peptide **5r** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.88% conversion of crude HPLC traces; $t_R = 15.873$ min; MS (ESI-TOF) calcd for $C_{35}H_{39}N_4O_6S$ (M - H)⁻: 643.26, found 643.38.

Synthesis of Fmoc-Arg(Boc)₂^S-Val-Gly-Phe-OH (5s). Peptide 5s was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.43% conversion of crude HPLC traces; $t_R = 19.433$ min (major); MS (ESI-TOF) calcd for $C_{47}H_{60}N_7O_{10}S$ (M - H)⁻: 914.41, found 914.47.

Synthesis of Fmoc-Ala^S-Val-Arg(pbf)-Gly-OH (*5t*). Peptide **5t** was obtained using the procedure 2. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-80% B with a flow rate of 1.0 mL/min over 30 min

gave the target peptide. 98.01% conversion of crude HPLC traces; $t_R = 17.550$ min. MS (ESI-TOF) calcd for $C_{44}H_{58}N_7O_9S_2^+(M+H)^+$: 892.37, found 892.26.

Synthesis of NH₂-Glu^S-Ala-Gln-Trp-Pro-Lys-OH (**5u**). HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 5-50% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 100% conversion of crude HPLC traces; $t_R = 14.357$ min. MS (ESI-TOF) calcd for $C_{35}H_{52}N_9O_9S^+(M + H)^+$: 774.36, found 774.30.

Synthesis of NH₂-Aal^S-Ala-Tyr-Asn-Phe-OH (**5v**). HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 5-50% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 100% conversion of crude HPLC traces; $t_R = 18.433$ min. MS (ESI-TOF) calcd for C₂₈H₃₇N₆O₇S⁺ (M + H)⁺: 601.24, found 601.15.

General Procedure 3 for the Incorporation of Multithioamide on a Solid Support.

Fmoc Removal. After the first thioamide incorporation, the **6** (0.03 mmol) was washed with DMF. A solution of DBU/n-C₆CH₁₃SH (1%/25%, v/v in DMF, 3.0 mL) was added to the fritted syringes to react 2 min before being drained. A further 3.0 mL of DBU/n-C₆CH₁₃SH was added to react 2 min. Then, the resin was carefully washed with DMF (3.0 mL x 4) and CH₂Cl₂ (3.0 mL x 4).

Subsequent Thioamide Bonds Incorporation. After the first thioamide was incorporated into the peptide sequence, the subsequent Fmoc deprotections and peptide elongation were treated with DBU/*n*-C₆CH₁₃SH and HBTU/DIPEA conditions. The second thioamide incorporation proceeded as described above general procedure 1 to get target 7 on the resin. The peptide 7 was obtained after treat 10 mg resin with 100.0 μ L of cleavage cocktail (TFE/AcOH/CH₂Cl₂ - 1:3:6) for 30 min. The CH₂Cl₂ was removed from the cocktail. Then the solution of water/acetonitrile (v/v 1:1, 200.0 μ L) was added to the residue. The resin was filtered off and the filtrates was analysed by HPLC and MS.

Synthesis of Fmoc-A^SA^SVGF-OH (*7a*). Peptide **7a** was obtained using the procedure 3 after 1 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.42% conversion of crude HPLC traces; $t_R = 14.520$ min; HRMS (ESI-TOF) calcd for $C_{37}H_{42}N_5O_6S_2$ (M -H): 716.2582, found 716.2593.

Synthesis of Fmoc-E('Bu)^SA^SVGF-OH (*7b*). Peptide **7b** was obtained using the procedure 3 after 3 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 96.66% conversion of crude HPLC traces; $t_R = 14.316$ min; MS (ESI-TOF) calcd for $C_{43}H_{54}N_5O_8S_2$ (M + H)⁺: 832.34, found 831.97.

Synthesis of Fmoc-S('Bu)^SA^SVGF-OH (7*c*). Peptide 7*c* was obtained using the procedure 3 after 1 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.25% conversion of crude HPLC traces; $t_R = 13.095$ min (major); MS (ESI-TOF) calcd for $C_{41}H_{52}N_5O_7S_2$ (M + H)⁺: 790.33, found 790.01.

Synthesis of Fmoc-M^SA^SVGF-OH (7*d*). Peptide 7d was obtained using the procedure 3 after 4 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 95.51% conversion of crude HPLC traces; $t_R = 11.121$ min; MS (ESI-TOF) calcd for $C_{39}H_{48}N_5O_6S_3$ (M + H)⁺: 778.28, found 777.92.

Synthesis of Fmoc-K(Boc)^SA^SVGF-OH (*7e*). Peptide **7e** was obtained using the procedure 3 after 4 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 92.92% conversion of crude

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HPLC traces; $t_R = 12.843$ min; MS (ESI-TOF) calcd for $C_{45}H_{59}N_6O_8S_2(M + H)^+$: 875.38, found 875.25.

Synthesis of Fmoc-A^SLA^SVGF-OH (7*f*). Peptide 7*f* was obtained using the procedure 3 after 3 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.63% conversion of crude HPLC traces; $t_R = 11.883$ min; HRMS (ESI-TOF) calcd for $C_{43}H_{53}N_6O_7S_2$ (M - H): 829.3423, found 829.3425.

Synthesis of Fmoc-F^SLA^SVGF-OH (7g). Peptide 7g was obtained using the procedure 3 after 1 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.10% conversion of crude HPLC traces; $t_R = 15.975$ min (major); HRMS (ESI-TOF) calcd for C₄₉H₅₅N₆O₇S₂ (M -H)⁻: 905.3736, found 905.3750.

Synthesis of Fmoc-K(Boc)^SLA^SVGF-OH (7*h*). Peptide 7*h* was obtained using the procedure 3 after 3 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 93.64% conversion of crude HPLC traces; $t_R = 16.006$ min; MS (ESI-TOF) calcd for $C_{51}H_{70}N_7O_9S_2$ (M + H)⁺: 988.47, found 988.39.

Synthesis of Fmoc-M^SLA^SVGF-OH (7*i*). Peptide 7*i* was obtained using the procedure 3 after 3 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 95.62% conversion of crude HPLC traces; $t_R = 13.833$ min; MS (ESI-TOF) calcd for $C_{45}H_{57}N_6O_7S_3$ (M - H): 889.26, found 889.35.

Synthesis of Fmoc-E('Bu)^SLA^SVGF-OH (*7j*). Peptide **7j** was obtained using the procedure 3 after 5 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 93.34% conversion of crude HPLC traces; $t_R = 17.450$ min; MS (ESI-TOF) calcd for $C_{49}H_{65}N_6O_9S_2$ (M + H)⁺: 945.42, found 945.04.

Synthesis of Fmoc-S('Bu)^SLA^SVGF-OH (7k). Peptide 7k was obtained using the procedure 3 after 3 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.54% conversion of crude HPLC traces; $t_R = 17.791$ min (major); MS (ESI-TOF) calcd for $C_{47}H_{63}N_6O_8S_2$ (M + H)⁺: 903.41, found 902.99.

Synthesis of Fmoc-A^SLAA^SVF-OH (71). Peptide 71 was obtained using the procedure 3 after 2 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 60-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 100% conversion of crude HPLC traces; t_R = 13.330 min; MS (ESI-TOF) calcd for C₄₄H₅₅N₆O₇S₂ (M - H)⁻: 843.36, found 843.44.

Synthesis of Fmoc-M^SAMY('Bu)^SAL-OH (7*m*). Peptide 7*m* was obtained using the procedure 3 after 2 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 98.89% conversion of crude HPLC traces; $t_R = 24.743$ min (major); MS (ESI-TOF) calcd for $C_{50}H_{67}N_6O_8S_4$ (M - H): 1007.39, found 1007.66.

Synthesis of Fmoc-K(Boc)^SLK(Boc)A^SAGF-OH (7*n*). Peptide 7**n** was obtained using the procedure 3 after 2 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 97.68% conversion of crude HPLC traces; $t_R = 23.400$ min; MS (ESI-TOF) calcd for $C_{60}H_{84}N_9O_{12}S_2$ (M - H): 1186.57, found 1186.50.

Synthesis of Fmoc-E('Bu)^SS('Bu)VE('Bu)^SAPFN(Trt)GK(Boc)-OH (70). Peptide 70 was obtained using the procedure 3 after 3 h.

HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 80-100% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. $t_R = 24.133$ min (major). MS (ESI-TOF) calcd for $C_{98}H_{127}N_{12}O_{19}S_2^-$ (M - H)⁻: 1839.88, found 1839.75.

Synthesis of Fmoc-M^SA^SA^SA-OH (7*p*). Peptide 7**p** was obtained using the procedure 3 after 10 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 70-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 99.24% conversion of crude HPLC traces; $t_R = 8.151$ min; MS (ESI-TOF) calcd for $C_{32}H_{40}N_5O_4S_5$ (M - H)⁻: 718.17, found 717.94.

Synthesis of Fmoc-A^SA^SA^SA^SA^SA-OH (7*q*). Peptide 7**q** was obtained using the procedure 3 after 5 h. HPLC analysis was performed using C18 analytical column, 254 nm UV detection. Gradient elution with 50-90% B with a flow rate of 1.0 mL/min over 30 min gave the target peptide. 97.75% conversion of crude HPLC traces; $t_R = 16.620$ min; MS (ESI-TOF) calcd for $C_{33}H_{41}N_6O_4S_5$ (M - H)⁻: 745.18, found 745.25.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at http://pubs.acs.org.

Optimization of reaction conditions; ¹H- and ¹³C NMR spectra; HPLC data; ESI-MS spectra; additional experimental procedures (PDF)

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Notes

The authors declare no competing financial interests.

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■ REFERENCES

(1) (a) Boutureira, O.; Bernardes, G. J. Advances in chemical protein modification. *Chem. Rev.* **2015**, *115*, 2174-2195; (b) Avan, I.; Hall, C. D.; Katritzky, A. R. Peptidomimetics via modifications of amino acids and peptide bonds. *Chem. Soc. Rev.* **2014**, *43*, 3575-3594; (c) Ko, E.; Liu, J.; Burgess, K. Minimalist and universal peptidomimetics. *Chem. Soc. Rev.* **2011**, *40*, 4411-4421; (d) Zheng, J. S.; He, Y.; Zuo, C.; Cai, X. Y.; Tang, S.; Wang, Z. A.; Zhang, L. H.; Tian, C. L.; Liu, L. Robust Chemical Synthesis of Membrane Proteins through a General Method of Removable Backbone Modification. *J. Am. Chem. Soc.* **2016**, *138*, 3553-3561.

(2) Mahanta, N.; Szantai-Kis, D. M.; Petersson, E. J.; Mitchell, D. A. Biosynthesis and Chemical Applications of Thioamides. *ACS. Chem. Biol.* **2019**, *14*, 142-163.

(3) (a) Fischer, G. Chemical aspects of peptide bond isomerisation. *Chem. Soc. Rev.* **2000**, *29*, 119-127; (b) Cour, T. F. M.; Hansen, H. A. S.; Clausen, K. I. M.; Lawesson, S. O. The geometry of the thiopeptide unit. *Int. J. Peptide. Protein. Res.* **1983**, *22*, 509-512.

(4) Wang, Y. J.; Szantai-Kis, D. M.; Petersson, E. J. Semi-synthesis of

thioamide containing proteins. Org. Biomol. Chem. 2015, 13, 5074-5081.

(5) Chen, X.; Mietlicki-Baase, E. G.; Barrett, T. M.; McGrath, L. E.; Koch-Laskowski, K.; Ferrie, J. J.; Hayes, M. R.; Petersson, E. J. Thioamide Substitution Selectively Modulates Proteolysis and Receptor Activity of Therapeutic Peptide Hormones. J. Am. Chem. Soc. 2017, 139, 16688-16695.

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3

4

5

6

7

59

60

- (6) Jagodzinski, T. S. Thioamides as useful synthons in the synthesis of heterocycles. *Chem. Rev.* **2003**, *103*, 197-227.
- (7) Lee, H.-J.; Choi, Y.-S.; Lee, K.-B.; Park, J.; Yoon, C.-J. Hydrogen Bonding Abilities of Thioamide. J. Phys. Chem. A. 2002, 106, 7010-7017.
- (8) Judge, R. H.; Moule, D. C.; Goddard, J. D. Thioamide spectroscopy:
 long path length absorption and quantum chemical studies of thioformamide vapour, CHSNH₂/CHSND₂. *Can. J. Chem.* 1987, 65, 2100-2105.
- (9) Bond, M. D.; Holmquist, B.; Vallee, B. L. Thioamide substrate
 probes of metal-substrate interactions in carboxypeptidase a catalysis. *J. Inorg. Biochem.* 1986, 28, 97-105.
- (10) (a) Miwa, J. H.; Patel, A. K.; Vivatrat, N.; Popek, S. M.; Meyer, 16 A. M. Compatibility of the Thioamide Functional Group with β -Sheet 17 Secondary Structure: Incorporation of a Thioamide Linkage into a β -18 Hairpin Peptide. Org. Lett. 2001, 3, 3373-3375; (b) Culik, R. M.; Jo, H.; DeGrado, W. F.; Gai, F. Using Thioamides To Site-Specifically 19 Interrogate the Dynamics of Hydrogen Bond Formation in β -Sheet 20 Folding. J. Am. Chem. Soc. 2012, 134, 8026-8029; (c) Walters, C. R.; 21 Szantai-Kis, D. M.; Zhang, Y.; Reinert, Z. E.; Horne, W. S.; Chenoweth, 22 D. M.; Petersson, E. J. The effects of thioamide backbone substitution 23 on protein stability: a study in α -helical, β -sheet, and polyproline II helical contexts. Chem Sci. 2017, 8, 2868-2877. 24
- 21(11) (a) Miwa, J. H.; Pallivathucal, L.; Gowda, S.; Lee, K. E.
Conformational Stability of Helical Peptides Containing a Thioamide
Linkage. Org. Lett. 2002, 4, 4655-4657; (b) Tran; Zeng, J.; Treutlein,
H.; Burgess, A. W. Effects of Thioamide Substitutions on the
Conformation and Stability of α- and β-Helices. J. Am. Chem. Soc.
2002, 124, 5222-5230; (c) Reiner, A.; Wildemann, D.; Fischer, G.;
Kiefhaber, T. Effect of Thioxopeptide Bonds on α-Helix Structure and
Stability. J. Am. Chem. Soc. 2008, 130, 8079-8084.
- 31(12) Newberry, R. W.; VanVeller, B.; Guzei, I. A.; Raines, R. T. $n \rightarrow \pi^*$ 32Interactions of Amides and Thioamides: Implications for Protein33Stability. J. Am. Chem. Soc. 2013, 135, 7843-7846.
- (13) (a) Wildemann, D.; Schiene-Fischer, C.; Aumüller, T.; Bachmann, 34 A.; Kiefhaber, T.; Lücke, C.; Fischer, G. A Nearly Isosteric 35 Photosensitive Amide-Backbone Substitution Allows Enzyme Activity 36 Switching in Ribonuclease S. J. Am. Chem. Soc. 2007, 129, 4910-4918; (b) Helbing, J.: Bregy, H.: Bredenbeck, J.: Pfister, R.: Hamm, P.: Huber, 37 R.; Wachtveitl, J.; De Vico, L.; Olivucci, M. A fast photoswitch for 38 minimally perturbed peptides: investigation of the trans-cis 39 photoisomerization of N-methylthioacetamide. J. Am. Chem. Soc. 2004, 40 126.8823-8834.
- (14) (a) Goldberg, J. M.; Batjargal, S.; Petersson, E. J. Thioamides as 41 fluorescence quenching probes: minimalist chromophores to monitor 42 protein dynamics. J. Am. Chem. Soc. 2010, 132, 14718-14720; (b) 43 Goldberg, J. M.; Speight, L. C.; Fegley, M. W.; Petersson, E. J. 44 Minimalist probes for studying protein dynamics: thioamide quenching of selectively excitable fluorescent amino acids. J. Am. Chem. Soc. 45 2012, 134, 6088-6091; (c) Goldberg, J. M.; Batjargal, S.; Chen, B. S.; 46 Petersson, E. J. Thioamide quenching of fluorescent probes through 47 photoinduced electron transfer: mechanistic studies and applications. J. 48 Am. Chem. Soc. 2013, 135, 18651-18658; (d) Goldberg, J. M.; Wissner, 49 R. F.; Klein, A. M.; Petersson, E. J. Thioamide quenching of intrinsic protein fluorescence. Chem. Commun. 2012, 48, 1550-1552; (e) 50 Wissner, R. F.; Batjargal, S.; Fadzen, C. M.; Petersson, E. J. Semi-51 synthesis of thioamide /thioamide Forster resonant energy transfer 52 pairs by combining unnatural amino acid mutagenesis and native 53 chemical ligation. J. Am. Chem. Soc. 2013, 135, 6529-6540; (f) 54 Goldberg, J. M.; Chen, X.; Meinhardt, N.; Greenbaum, D. C.; Petersson, E. J. Thioamide-based fluorescent protease sensors. J. Am. Chem. Soc. 55 2014, 136, 2086-2093. 56
- (15) Misra, S. K.; Tewari, U. C. Complexing behaviour of aromatic
 thioamides (ArCSNHCOR). Transition metal complexes of N-

carboethoxy-4-chlorobenzene- and N-carboethoxy-4-bromobenzene thioamide ligands. *Transition Met. Chem.* **2002**, *27*, 120-125.

- (16) (a) Walter, W.; Bode, K. D. Syntheses of Thiocarboxamides. Angew. Chem. Int. Ed. 1966, 5, 447-461; (b) Hurd, R. N.; DeLaMater, G. The Preparation and Chemical Properties of Thionamides. Chem. Rev. 1961, 61, 45-86; (c) Priebbenow, D. L.; Bolm, C. Recent advances in the Willgerodt-Kindler reaction. Chem. Soc. Rev. 2013, 42, 7870-7880; (d) Wei, J.; Li, Y.; Jiang, X. Aqueous Compatible Protocol to Both Alkyl and Aryl Thioamide Synthesis. Org. Lett. 2016, 18, 340-343; (e) Tan, W.; Wang, C.; Jiang, X. Green carbon disulfide surrogate via a combination of potassium sulfide and chloroform for benzothiazine-thione and benzothiazole-thione construction. Org. Chem. Front. 2018, 5, 2390-2394; (f) Tan, W.; Jansch, N.; Ohlmann, T.; Meyer-Almes, F. J.; Jiang, X. Thiocarbonyl Surrogate via Combination of Potassium Sulfide and Chloroform for Dithiocarbamate Construction. Org. Lett. 2019, 21, 7484-7488; (g) Sulfur Chemistry. In Topics in Current Chemistry Collections Jiang, X., Ed.; Springer: Berlin, 2019; Vol. 21.
- (17) Bartlett, P. A.; Spear, K. L.; Jacobsen, N. E. A thioamide substrate of carboxypeptidase A. *Biochemistry*. **1982**, *21*, 1608-1611.
- (18) Clausen, K.; Thorsen, M.; Lawesson, S. O. Studies on amino acids and peptides-I. *Tetrahedron*. **1981**, *37*, 3635-3639.
- (19) Panduranga, V.; Prabhu, G.; Kumar L, R.; Krishnamurthy, M.; Sureshbabu, V. V. Thionation of di and tripeptides employing thiourea as a sulphur transfer reagent. *RSC Adv.* **2016**, *6*, 98141-98146.
- (20) (a) Ried, W.; von der Emden, W. Aminosäure-thionester und Endothiopeptide. *Angew. Chem.* **1960**, *72*, 268-268; (b) Le, H.T.; Mayer, M.; Thoret, S.; Michelot, R. Incorporation of thioamide linkages into a growing peptide under SPPS conditions improved by salt effects. *Int. J. Peptide. Protein. Res.* **1995**, *45*, 138-144.
- (21) (a) Zacharie, B.; Sauvé, G.; Penney, C. Thioacylating agents. Use of thiobenzimidazolone derivatives for the preparation of thiotuftsin analogs. *Tetrahedron.* **1993**, *49*, 10489-10500; (b) Zacharie, B.; Lagraoui, M.; Dimarco, M.; Penney, C. L.; Gagnon, L. Thioamides: Synthesis, Stability, and Immunological Activities of Thioanalogues of Imreg. Preparation of New Thioacylating Agents Using Fluorobenzimidazolone Derivatives. *J. Med. Chem.* **1999**, *42*, 2046-2052.

(22) Shalaby, M. A.; Grote, C. W.; Rapoport, H. Thiopeptide Synthesis. α-Amino Thionoacid Derivatives of Nitrobenzotriazole as Thioacylating Agents. J. Org. Chem. **1996**, *61*, 9045-9048.

(23) (a) Wildemann, D.; Drewello, M.; Fischer, G.; Schutkowski, M. Extremely selective Mg(ClO₄)₂ mediated removal of Bpoc/Ddz moieties suitable for the solid phase peptide synthesis of thioxo peptides. Chem. Commun. 1999, 1809-1810; (b) Frank, R.; Jakob, M.; Thunecke, F.: Fischer, G.: Schutkowski, M. Thioxylation as One-Atom-Substitution Generates a Photoswitchable Element within the Peptide Backbone. Angew. Chem. Int. Ed. 2000, 39, 1120-1122; (c) Caba, J. M.; Rodriguez, I. M.; Manzanares, I.; Giralt, E.; Albericio, F. Solid-Phase Total Synthesis of Trunkamide A1. J. Org. Chem. 2001, 66, 7568-7574; (d) Mukherjee, S.; Verma, H.; Chatterjee, J. Efficient Site-Specific Incorporation of Thioamides into Peptides on a Solid Support. Org. Lett. 2015, 17, 3150-3153; (e) Wang, Y. J.; Szantai-Kis, D. M.; Petersson, E. J. Semi-synthesis of thioamide containing proteins. Org. Biomol. Chem. 2015, 13, 5074-5081; (f) Barrett, T. M.; Fiore, K. E.; Liu, C.; Petersson, E. J. Chemistry of Thioamides: Thioamide-Containing Peptides and Proteins; Springer: Singapore, 2019; (g) Walters, C. R.; Ferrie, J. J.; Petersson, E. J. Chemical Ligation: Thioamide Labeling of Proteins through a Combination of Semisynthetic Methods; John Wiley & Sons, Inc.: Hoboken, 2017.

(24) (a) Høeg-Jensen, T.; Havsteen Jakobsen, M.; Olsen, C. E.; Holm, A. Formation of peptide thioamides by use of Fmoc amino monothioacids and PyBOP. *Tetrahedron Lett.* **1991**, *32*, 7617-7620; (b) Høeg-Jensen, T.; Olsen, C. E.; Holm, A. Thioacylation Achieved by Activation of a Monothiocarboxylic Acid with Phosphorus Reagents. *J. Org. Chem.* **1994**, *59*, 1257-1263; (c) Høeg-Jensen, T. Review: Endothiopeptides Alias Peptide Thioamides. *Phosphorus, Sulfur Silicon Relat. Elem.* **1996**, *108*, 257-278.

(25) (a) Hu, L.; Xu, S.; Zhao, Z.; Yang, Y.; Peng, Z.; Yang, M.; Wang, C.; Zhao, J. Ynamides as Racemization-Free Coupling Reagents for Amide and Peptide Synthesis. *J. Am. Chem. Soc.* **2016**, *138*, 13135-13138; (b) Hu, L.;

Zhao, J. Ynamide: A New Coupling Reagent for Amide and Peptide Synthesis. Synlett 2017, 28, 1663-1670; (c) Tu, Y.; Zeng, X.; Wang, H.; 1 Zhao, J. A Robust One-Step Approach to Ynamides. Org. Lett. 2018, 20, 2 280-283. (d) Zeng, X.; Tu, Y.; Zhang, Z.; You, C.; Wu, J.; Ye, Z.; Zhao, J. 3 Transition-Metal-Free One-Step Synthesis of Ynamides. J. Org. Chem. 2019, 84, 4458-4466. 4 (26) Yang, J.; Wang, C.; Xu, S.; Zhao, J. Ynamide-Mediated 5 Thiopeptide Synthesis. Angew. Chem. Int. Ed. 2019, 58, 1382-1386. 6 (27) (a) Petersson, E. J.; Szantai-Kis, D.; Walters, C.; Barrett, T.; 7 Hoang, E. Thieme Chemistry Journals Awardees - Where Are They 8 Now? Improved Fmoc Deprotection Methods for the Synthesis of Thioamide-Containing Peptides and Proteins. Svnlett. 2017. 28. 1789-9 1794; (b) Internet Bond-energy Databank (iBonD) Home Page: 10 http://ibond.nankai.edu.cn or http://ibond.chem.tsinghua.edu.cn. 11 Internet Bond-energy Databank (iBonD) Home Page 12 http://ibond.nankai.edu.cn or http://ibond.chem.tsinghua.edu.cn; (c) Camacho, L. A.; Lampkin, B. J.; VanVeller, B. A Bottom-Up 13 Approach To Preserve Thioamide Residue Stereochemistry during 14 Fmoc Solid-Phase Peptide Synthesis. Org. Lett. 2019, 21, 7015-7018. 15 (28) (a) Høeg-Jensen, T.; Spatola, A. F.; Holm, A. Amino monothio 16 acids in solid-phase synthesis of peptide thioamides. Int. J. Peptide. 17 Protein. Res. 1996, 47, 190-200; (b) Wade, J. D.; Bedford, J.; Sheppard, R. C.; Tregear, G. W. DBU as an N alpha-deprotecting reagent for the 18 fluorenylmethoxycarbonyl group in continuous flow solid-phase 19 peptide synthesis. Pept. Res. 1991, 4, 194-199; (c) Ralhan, K.; 20 KrishnaKumar, V. G.; Gupta, S. Piperazine and DBU: a safer 21 alternative for rapid and efficient Fmoc deprotection in solid phase peptide synthesis. RSC Adv. 2015, 5, 104417-104425. 22 (29) (a) Mukherjee, S.; Chatterjee, J. Suppressing the epimerization of 23 endothioamide peptides during Fmoc/t-Bu-based solid phase peptide 24 synthesis. J. Pept. Sci. 2016, 22, 664-672; (b) Huang, Y.; Ferrie, J. J.; 25 Chen, X.; Zhang, Y.; Szantai-Kis, D. M.; Chenoweth, D. M.; Petersson, E. J. Electronic interactions of i, i + 1 dithioamides: increased 26 fluorescence quenching and evidence for n-to- π^* interactions. *Chem.* 27 Commun. 2016, 52, 7798-7801; (c) Walters, C. R.; Ferrie, J. J.; 28 Petersson, E. J. Dithioamide substitutions in proteins: effects on 29 thermostability, peptide binding, and fluorescence quenching in 30 calmodulin. Chem. Commun. 2018, 54, 1766-1769. (30) (a) Hayakawa, Y.; Sasaki, K.; Adachi, H.; Furihata, K.; Nagai, K.; 31 Shin-ya, K. Thioviridamide, a novel apoptosis inducer in transformed 32 cells from Streptomyces olivoviridis. J. Antibiot. 2006, 59, 1-5; (b) 33 Hayakawa, Y.; Sasaki, K.; Nagai, K.; Shin-ya, K.; Furihata, K. 34 Structure of thioviridamide, a novel apoptosis inducer from Streptomyces olivoviridis. J. Antibiot. 2006, 59, 6-10; (c) Frattaruolo, 35 L.; Lacret, R.; Cappello, A. R.; Truman, A. W. A Genomics-Based 36 Approach Identifies a Thioviridamide-Like Compound with Selective 37 Anticancer Activity. ACS. Chem. Biol. 2017, 12, 2815-2822; (d) 38 Kjaerulff, L.; Sikandar, A.; Zaburannyi, N.; Adam, S.; Herrmann, J.; 39 Koehnke, J.; Muller, R. Thioholgamides: Thioamide-Containing Cytotoxic RiPP Natural Products. ACS. Chem. Biol. 2017, 12, 2837-40 2841; (e) Kawahara, T.; Izumikawa, M.; Kozone, I.; Hashimoto, J.; 41 Kagaya, N.; Koiwai, H.; Komatsu, M.; Fujie, M.; Sato, N.; Ikeda, H.; 42 Shin-Ya, K. Neothioviridamide, a Polythioamide Compound Produced 43 by Heterologous Expression of a Streptomyces sp. Cryptic RiPP Biosynthetic Gene Cluster. J. Nat. Prod. 2018, 81, 264-269; (f) Tang, 44 J.; Lu, J. X.; Luo, Q. F.; Wang, H. Discovery and biosynthesis of 45 thioviridamide-like compounds. Chin. Chem. Lett. 2018, 29, 1022-46 1028. 47 (31) (a) Paradis-Bas, M.; Tulla-Puche, J.; Albericio, F. The road to the synthesis of "difficult peptides". Chem. Soc. Rev. 2016, 45, 631-654; (b) 48 Larsen, B. D.; Holm, A. Incomplete Fmoc deprotection in solid-phase 49 synthesis of peptides. Int. J. Pept. Protein. Res. 1994, 43, 1-9. 50

(32) Yamashiro, D.; Blake, J. Use of thiol acids in peptide segment coupling in non-aqueous solvents. *Int. J. Pept. Protein. Res.* 1981, *18*, 383-392.

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60