## Article

# Acid-Catalyzed 1,3-Dipolar Cycloaddition of 2H-Azirines with Nitrones: An Unexpected Access to 1,2,4,5-Tetrasubstituted Imidazoles

Anikó Angyal, András Demjén, János Wölfling, László G. Puskás, and Iván Kanizsai\*



straightforward approach toward the regioselective synthesis of 1,2,4,5tetrasubstituted imidazoles, is reported. This trifluoroacetic acid-catalyzed protocol tolerates a broad range of aliphatic and aromatic substrates, offering an efficient access to highly diverse, multisubstituted imidazoles in isolated yields up to 83% under mild conditions.



# INTRODUCTION

2H-Azirines have received significant attention because of their wide applicability as building blocks in organic synthesis.<sup>1</sup> In addition to their ability to act as both nucleophiles and electrophiles,<sup>2</sup> 2*H*-azirines can also operate as dienophiles, dipolarophiles, and dipoles in cycloaddition reactions. Furthermore, they can undergo selective ring cleavage on thermal or photochemical excitation to generate reactive vinyl nitrene, imino carbene, and nitrile vlide species.<sup>4</sup> By exploiting this versatile reactivity, efforts have been made over the past decades on the synthesis of N-heterocycles, such as pyrroles, pyrazoles,<sup>6</sup> pyridines,<sup>7</sup> pyrazines,<sup>8</sup> and indoles,<sup>9</sup> from 2Hazirines.<sup>1</sup> In contrast, the azirine-based assembly of imidazoles has been less extensively explored.<sup>10</sup>

Looking at the 1,3-dipolar cycloaddition (1,3-DC) of 2Hazirines as dipolarophiles, the dipole scope of the reaction is almost exclusively limited to azomethine ylides (Scheme 1a).<sup>11</sup> To the best of our knowledge, there is no precedent for the utilization of nitrones in cycloaddition reactions involving azirines. Inspired by this information and the well-known dipolar cycloaddition reactions of the easily accessible nitrones<sup>12</sup> with alkenes,<sup>13</sup> alkynes,<sup>14</sup> and strained cyclopropenes<sup>15</sup> toward the synthesis of isoxazolidine and isoxazoline scaffolds (Scheme 1b,c), we envisioned a 1,3-DC approach for the construction of an unprecedented aziridine-fused bicyclic framework (Scheme 1d). Unexpectedly, the reaction between 2H-azirines and nitrones led to the discovery of a novel access to 1,2,4,5-tetrasubstituted imidazoles (Scheme 1e), which have drawn considerable attention owing to their wide range of biological and pharmacological properties including, among others, <sup>16</sup> p38 $\alpha$  MAP kinase inhibitory, <sup>17</sup> antiviral, <sup>18</sup> antibacterial, <sup>19</sup> and anticancer<sup>20</sup> activities.

# RESULTS AND DISCUSSION

We initiated our study by investigating the 1,3-DC of Cphenyl-N-methylnitrone 1a and racemic ethyl 3-methyl-2H-

Scheme 1. 2H-Azirines and Nitrones in 1,3-DC



azirine-2-carboxylate  $((\pm)-2a)$  in anhydrous acetonitrile. However, no conversion was observed. In order to increase the reactivity of azirine  $(\pm)$ -2a, several Brønsted and Lewis acid catalysts were tested at room and elevated temperatures (60 °C) (Table 1, entries 2-31; see the Supporting Information for conversions and yields in 8 and 24 h). The

Received: December 6, 2019 Published: February 5, 2020



# Table 1. Catalyst Screening<sup>a</sup>

		Et MeCN 24 h		COOEt
entry	catalyst		yield (%) <sup>b</sup>	yield $(\%)^c$
1			0	0
2	НСООН		36	42
3	AcOH		0	17
4	TFA		62	50
5	MeSO <sub>3</sub> H		52	42
6	$PTSA \times H_2O$		57	44
7	HClO <sub>4</sub> (70%)		56	48
8	$H_2SO_4$		49	41
9	$NaHSO_4$		28	30
10	$B(OH)_3$		35	39
11	$PTA^{f}$		54	49
12	BINOL-phosphoric	acid <sup>g</sup>	52	43
13	silica gel <sup>d,e</sup>		16	41
14	Amberlyst 15 <sup>d</sup>		1	1
15	montmorillonite K	10 <sup>d</sup>	22	22
16	$ZnCl_2$		8	22
17	$Zn(OAc)_2$		7	9
18	$ZnF_2$		0	22
19	InCl <sub>3</sub>		12	21
20	$In(OAc)_3$		2	20
21	$In(OTf)_3$		39	35
22	$FeCl_2 \times 4H_2O$		4	31
23	$\text{FeCl}_3 \times 6\text{H}_2\text{O}$		7	20
24	CuCl		2	13
25	CuCl <sub>2</sub>		3	16
26	$Cu(OAc)_2$		2	2
27	$Cu(OTf)_2$		12	23
28	$Mg(OTf)_2$		56	37
29	Yb(OTf) <sub>3</sub>		57	39
30	$Sc(OTf)_3$		56	42
31	Dy(OTf) <sub>3</sub>	59	39	

<sup>*a*</sup>Reaction conditions: nitrone 1a (0.10 mmol), 2*H*-azirine ( $\pm$ )-2a (0.10 mmol), anhydrous MeCN (0.3 mL), catalyst (10 mol %), 24 h. <sup>*b*</sup>Room temperature. Yields were determined by HPLC-MS analysis. <sup>*c*</sup>60 °C. Yields were determined by HPLC-MS analysis. <sup>*d*</sup>10 w/v % was applied. <sup>*e*</sup>60 Å, 70-230 mesh. <sup>*f*</sup>Phosphotungstic acid. <sup>*g*</sup>1,1'-Binaphthyl-2,2'-diyl hydrogen phosphate.

reaction of 1a and 2a was readily promoted by most of the applied catalysts. Surprisingly, however, the regioselective formation of imidazole-4-carboxylate 3a was observed instead of the expected cycloadduct. The best results were obtained when trifluoroacetic acid (TFA) was utilized at room temperature, affording 3a in 62% high-performance liquid chromatography (HPLC) yield (Table 1, entry 4). Other Brønsted acids such as HClO<sub>4</sub> and *p*-toluenesulfonic acid (PTSA) gave similar yields (Table 1, entries 6 and 7), whereas AcOH and solid acids exhibited poor catalytic activity (Table 1, entries 3, 13–15). Among the tested Lewis acids, only Mg(OTf)<sub>2</sub>, Yb(OTf)<sub>3</sub>, Sc(OTf)<sub>3</sub>, and Dy(OTf)<sub>3</sub> showed considerable efficiency (Table 1, entries 28–31), while zinc, indium, iron, and copper salts proved to be ineffective (Table 1, entries 16–27).

To further optimize the reaction conditions, various solvents were examined at room temperature in the presence of TFA (10 mol %) as a catalyst (Table 2, entries 1-13). However,

#### Table 2. Optimization of the Reaction Conditions<sup>a</sup>

	+	COOEt TFA		COOEt
entry	solvent	2a (equiv)	TFA (mol %)	yield (%) <sup>b</sup>
1	MeCN	1.0	10	62
2	MeOH	1.0	10	39
3	EtOH	1.0	10	46
4	IPA	1.0	10	33
5	TFE <sup>c</sup>	1.0	10	14
6	HFIP <sup>c</sup>	1.0	10	3
7	CHCl <sub>3</sub>	1.0	10	39
8	$CH_2Cl_2$	1.0	10	55
9	Toluene	1.0	10	42
10	THF	1.0	10	54
11	dioxane	1.0	10	47
12	DMF	1.0	10	48
13	DMSO	1.0	10	13
14	MeCN	1.25	10	70
15	MeCN	1.5	10	74
16	MeCN	2.0	10	74
17	MeCN	1.5	1	43
18	MeCN	1.5	2.5	60
19	MeCN	1.5	5	73
20	MeCN	1.5	15	72
21	MeCN	1.5	20	68
22	MeCN	1.5	40	41
23 <sup>d</sup>	MeCN	1.5	5	69
24 <sup>d</sup>	MeCN	1.5	10	$78(72)^{e}$

<sup>*a*</sup>Reaction conditions: nitrone 1a (0.10 mmol), 2*H*-azirine ( $\pm$ )-2a (0.10–0.20 mmol), anhydrous solvent (0.3 mL), TFA (1–40 mol %), rt, 24 h. <sup>*b*</sup>Yields were determined by HPLC–MS analysis. <sup>*c*</sup>Non-dried. <sup>*d*</sup>60 °C, 6 h. <sup>*e*</sup>Isolated yield. Nitrone 1a (1 mmol), 2*H*-azirine ( $\pm$ )-2a (1.5 mmol), anhydrous MeCN (3 mL), TFA (10 mol %), 60 °C, 6 h.

none of the tested media proved superior to acetonitrile (62% HPLC yield, Table 2, entry 1). In general, the formation of 3a was slightly more favored in aprotic solvents, while fluorinated alcohols (entries 5 and 6) and dimethyl sulfoxide (DMSO) were not tolerated. To our delight, by employing 1.5-fold excess of azirine 2a, further improvement in the yield of 3a could be achieved (74% HPLC yield, Table 2, entry 15). On the other hand, when lower or higher catalyst loadings in the range of 1-40 mol % were applied, a trend of decreasing yields was observed (Table 2, entries 17-22). Finally, the reaction performed at 60 °C in the presence of 10 mol % of TFA pleasingly furnished the product 3a in 78% HPLC yield and 72% isolated yield in 6 h (Table 2, entry 24). It is noteworthy that HPLC-MS analysis of the crude reaction mixtures did not indicate the formation of the cycloadduct or other isomeric imidazoles.

To explore the scope of the reaction, a range of *N*-methylnitrones (1a-r) were reacted with azirine 2a under optimized conditions [2a (1.5 equiv), TFA (10 mol %), anhydr. MeCN, 60 °C, 6 h] (Table 3, entries 1–13). The developed method was found to be facile with nitrones bearing both electron-donating (MeO) and electron-withdrawing (F, NO<sub>2</sub>) substituents on the *C*-phenyl ring, affording 3b-d in 68–78% isolated yields (entries 2–4). Moreover, the reaction could be readily extended to heteroaromatic *N*-methylnitrones, as exemplified by the synthesis of pyridyl and furyl imidazoles

Table 3. Reactions of Various Nitrones 1 with 2*H*-Azirine  $(\pm)$ -2a<sup>*a*</sup>

R <sup>2</sup>         	√ 0   +	N COOEt (±)-2a	TFA (10 mol%) MeCN 60 °C, 6 h	$R^2$ $R^1$ N 3a-r	COOEt
entry	1	R <sup>1</sup>	R <sup>2</sup>	3	yield (%)
1	1a	C <sub>6</sub> H <sub>5</sub>	Me	3a	72
2	1b	4-MeOC <sub>6</sub> H <sub>4</sub>	Me	3b	68
3	1c	$4-FC_6H_4$	Me	3c	78
4	1d	$4-NO_2C_6H_4$	Me	3d	69
5	1e	2-pyridyl	Me	3e	82
6	1f	3-pyridyl	Me	3f	81
7	1g	5-Me-2-furyl	Me	3g	41
8	1h	Me	Me	3h	70
9	1i	<i>i</i> -Pr	Me	3i	53
10	1j	t-Bu	Me	3j	49
11	1k	<i>n</i> -heptyl	Me	3k	54
12	11	c-Pr	Me	31	72
13	1m	c-Hex	Me	3m	70
14	1n	<i>i</i> -Pr	<i>i</i> -Pr	3n	55
15	10	$C_6H_5$	Bn	30	63
16	1p	<i>i</i> -Pr	Bn	3p	71
17	1q	$C_6H_5$	c-Hex	3q	37
18	1r	$C_6H_5$	$4-FC_6H_4$	3r	42
<sup>a</sup> Reaction	n condi	itions nitrone	la-r (1 mmol)	). 2.H-azir	ine (+)-2a

(1.5 mmol), anhydrous MeCN (3 mL), TFA (10 mol %), 60 °C, 6 h.

**3e**-g (82, 81, and 41%, entries 5–7). Gratifyingly, C-aliphatic *N*-methylnitrones, including bulky substrate **1**j ( $\mathbb{R}^1 = t$ -Bu), also underwent smooth transformation to furnish multisubstituted imidazoles **3h**-**m** in moderate to good yields (49–72%, entries 8–13). Subsequently, we examined the influence of the N-substituents ( $\mathbb{R}^2$ ) on the efficiency of the reaction (entries 14–18). Interestingly, while *N*-isopropyl- and *N*-benzylnitrones **1n**-**p** were well-tolerated (**3n**-**p**, 55–71%, entries 14–16), the corresponding *N*-cyclohexyl- and *N*-4 (fluorophenyl)imidazoles **3q** and **3r** were obtained in lower isolated yields (37 and 42%, entries 17 and 18).

Next, we focused on exploring the scope of the reaction with respect to azirine, using both a C-aromatic and a C-aliphatic Nmethylnitrone (1a and 1i) for a more reliable comparison (Table 4). As expected, transformations of azirine 2b proceeded smoothly to give the corresponding imidazoles 4a and 5a in yields (Table 4, 67 and 49%) similar to those of ethoxycarbonyl analogue 2a (Table 3, 3a, 72% and 3i, 53%). We were pleased to find that 2,3-diarylazirines 2c-f were fully compatible with the reaction and could be efficiently converted to products 4b-e and 5b-e in good yields, regardless of the electronic nature of the azirine or the nitrone substrate (Table 4, 67-79%). Benzylazirine 2g was also well-tolerated, albeit it provided lower isolated yields (4f, 45% and 5f, 57%). Unfortunately, the application of monosubstituted azirine 2h led to complex reaction mixtures and only trace amounts of the desired products.

Finally, to give a comprehensive evaluation with respect to the generality, we briefly surveyed the reactivity of aromatic azirine **2c** toward some of the nitrones applied previously (Table 5). Pleasingly, both C-aromatic and C-aliphatic *N*methylnitrones could be subjected to the reaction, delivering imidazoles **6a**-**f** in moderate to good yields (Table 5, 50– 79%). It is notable that introducing the sterically demanding

#### Table 4. Scope of 2H-Azirines<sup>4</sup>



"Reaction conditions: nitrone 1a,i (1 mmol), azirine 2b-h (1.5 mmol), anhydrous MeCN (3 mL), TFA (10 mol %), 60 °C, 6 h.

Table 5. Scope of Nitrones with Aromatic 2H-Azirine  $2c^{a}$ 



<sup>a</sup>Reaction conditions: nitrone 1 (1 mmol), azirine 2c (1.5 mmol), anhydrous MeCN (3 mL), TFA (10 mol %), 60 °C, 6 h.

*tert*-butyl group as the  $\mathbb{R}^1$  substituent resulted again in a lower isolated yield (**6d**, 50%). On the other hand, nitrones possessing more electron-donating isopropyl and cyclohexyl substituents on the nitrogen (**1n** and **1q**) were less tolerated

(**6g**, 24% and **6h**, 32%). In contrast, when *N*-benzylnitrone **1p** was employed, product **6i** was isolated in a superior yield of 83%.

Although the detailed reaction mechanism remains to be clarified, a plausible mechanism is proposed in Scheme 2.

Scheme 2. Plausible Mechanism for the Formation of Imidazoles



Initially, 1,3-DC between nitrone 1 and activated 2H-azirine 2 might occur to form cycloadduct **A**, which immediately rearranges to ketoamidine **B** in the presence of the acid catalyst. Following the intramolecular cyclization of intermediate **B**, the final product is achieved by the dehydration of hydroxyimidazoline **C**.

# CONCLUSIONS

In summary, we have developed a novel procedure for the regioselective synthesis of 1,2,4,5-tetrasubstituted imidazoles through an unprecedented 1,3-DC of 2*H*-azirines with nitrones. This TFA-catalyzed cascade reaction is compatible with a variety of 2*H*-azirines and a wide range of C-aliphatic and C-aromatic as well as N-aliphatic and N-arylnitrones, offering a general and straightforward access to highly diverse multisubstituted imidazoles in isolated yields up to 83% under mild conditions.

#### EXPERIMENTAL SECTION

General Information. The NMR spectra were recorded at 298 K on a Bruker Ascend 500 with a 5 mm BBO Prodigy Probe in CDCl<sub>3</sub> $d_1$  or DMSO- $d_6$ . The chemical shifts are reported in  $\delta$  (ppm) relative to the internal standard (tetramethylsilane) or the residual solvent signal. In case of <sup>19</sup>F NMR, hexafluorobenzene was used as the reference compound. Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublet, dt = doublet of triplet, etc.), coupling constants, and integration. Coupling constants (J) are given in hertz (Hz). LC/MS analyses were carried out on an Agilent 1200 Series equipment with an Agilent G1946D MS detector (APCI, operated in a positive mode) with a Kinetex C18 column (100 Å, 5  $\mu$ m, 250 × 4.6 mm, Phenomenex). High-resolution mass spectra (HRMS) were performed on a Thermo Scientific Q Exactive hybrid quadrupole-Orbitrap mass spectrometer using an heated electrospray ionisation (HESI) ion source. Samples (5  $\mu$ L from 1  $\mu$ g/mL solution) were injected to the mass spectrometer using the flow injection method (200 µL/min, water/MeCN 1:1, 0.1% TFA). Melting points (mp) were recorded with a Digital Melting Point Apparatus (Electrothermal, IA 9000 Series). Chromatographic purification of the products was performed on Merck silica gel 60, particle size 0.063-0.200 mm. Thin-layer chromatography (TLC) was performed on fluorescent-indicating plates (aluminum sheets precoated with silica gel 60<sub>F254</sub>, 1.05554, Merck), and visualization was achieved using UV light (254 nm) or by staining with basic potassium permanganate solution. Nitrones 1a-r and racemic 2H-azirines 2a-h were synthesized according to literature procedures.<sup>21</sup> All other reagents

and solvents were commercially available and used without further purification.

General Procedure for the Synthesis of Imidazoles 3a-I, 4a-f, 5a-f, and 6a-i. To a solution of the corresponding nitrone 1 (1 mmol, 1.0 equiv) and 2*H*-azirine 2 (1.5 mmol, 1.5 equiv) in anhydrous MeCN (3 mL, 0.33 M), TFA (7.7  $\mu$ L, 0.1 mmol, 10 mol %) was added at room temperature and stirred at 60 °C for 6 h on a heating block-mounted hot plate magnetic stirrer. After the reaction was complete (monitored by TLC), the solvent was removed under reduced pressure to give the crude product, which was purified by flash column chromatography on silica gel (eluent: hexane/EtOAc or toluene/MeCN) to afford pure 3a-I, 4a-f, 5a-f, and 6a-i.

*Ethyl* 1,5-*Dimethyl*-2-*phenyl*-1*H*-*imidazole*-4-*carboxylate* (**3***a*). White solid, 175 mg, 72% yield, mp 84–85 °C. Silica gel TLC  $R_f$  = 0.20 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ ): δ 7.64 (d, *J* = 7.1 Hz, 2H), 7.57–7.46 (m, 3H), 4.24 (q, *J* = 7.1 Hz, 2H), 3.60 (s, 3H), 2.54 (s, 3H), 1.29 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, DMSO- $d_6$ ): δ 163.8, 146.9, 138.1, 130.6, 129.4, 129.2, 129.0, 128.0, 59.7, 32.5, 14.9, 10.7. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>17</sub>N<sub>2</sub>O<sub>2</sub>, 245.1285; found, 245.1284.

Ethyl 2-(4-Methoxyphenyl)-1,5-dimethyl-1H-imidazole-4-carboxylate (**3b**). White solid, 187 mg, 68% yield, mp 98–99 °C. Silica gel TLC  $R_f$  = 0.11 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.52 (d, *J* = 8.8 Hz, 2H), 6.97 (d, *J* = 8.8 Hz, 2H), 4.39 (q, *J* = 7.1 Hz, 2H), 3.85 (s, 3H), 3.56 (s, 3H), 2.61 (s, 3H), 1.40 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 164.2, 160.2, 147.4, 137.0, 130.7, 128.4, 122.6, 113.9, 60.2, 55.4, 31.9, 14.6, 10.6. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>15</sub>H<sub>19</sub>N<sub>2</sub>O<sub>3</sub>, 275.1391; found, 275.1390.

Ethyl 2-(4-Fluorophenyl)-1,5-dimethyl-1H-imidazole-4-carboxylate (**3c**). White solid, 204 mg, 78% yield, mp 98–100 °C. Silica gel TLC  $R_f = 0.23$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.60–7.54 (m, 2H), 7.15 (t, *J* = 8.7 Hz, 2H), 4.40 (q, *J* = 7.1 Hz, 2H), 3.57 (s, 3H), 2.61 (s, 3H), 1.40 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.2, 163.1 (d, *J* = 223.5 Hz), 146.5, 137.2, 131.2 (d, *J* = 8.3 Hz), 128.6, 126.3, 115.7 (d, *J* = 22.0 Hz), 60.3, 31.9, 14.6, 10.5. <sup>19</sup>F NMR (471 MHz, chloroform-*d*): δ –114.7; HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>16</sub>FN<sub>2</sub>O<sub>2</sub>, 263.1191; found, 263.1191.

*Ethyl* 1,5-*Dimethyl*-2-(4-*nitrophenyl*)-1*H*-*imidazole*-4-*carboxylate* (**3***d*). Yellow solid, 200 mg, 69% yield, mp 145–146 °C. Silica gel TLC  $R_f$  = 0.35 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 8.31 (d, *J* = 8.8 Hz, 2H), 7.83 (d, *J* = 8.8 Hz, 2H), 4.41 (q, *J* = 7.1 Hz, 2H), 3.66 (s, 3H), 2.64 (s, 3H), 1.41 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 163.7, 147.9, 145.0, 138.3, 136.2, 129.9, 129.6, 123.8, 60.6, 32.3, 14.54, 10.6. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>16</sub>N<sub>3</sub>O<sub>4</sub>, 290.1136; found, 290.1138.

*Ethyl* 1,5-*Dimethyl-2-(pyridin-2-yl)-1H-imidazole-4-carboxylate* (**3e**). Light yellow solid, 201 mg, 82% yield, mp 114–115 °C. Silica gel TLC  $R_f$  = 0.13 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 8.61–8.56 (m, 1H), 8.23 (d, *J* = 8.0 Hz, 1H), 7.77 (dt, *J* = 7.8, 1.8 Hz, 1H), 7.26–7.22 (m, 1H), 4.41 (q, *J* = 7.1 Hz, 2H), 4.04 (s, 3H), 2.62 (s, 3H), 1.42 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.0, 150.2, 148.1, 144.7, 138.8, 136.7, 128.6, 124.0, 123.0, 60.4, 33.0, 14.6, 10.4. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>13</sub>H<sub>16</sub>N<sub>3</sub>O<sub>2</sub>, 246.1238; found, 246.1237.

Ethyl 1,5-Dimethyl-2-(pyridin-3-yl)-1H-imidazole-4-carboxylate (**3f**). Orange oil, 198 mg, 81% yield. Silica gel TLC  $R_{\rm f}$  = 0.52 (chloroform/MeOH = 19/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 8.84 (s, 1H), 8.68 (d, *J* = 4.8 Hz, 1H), 7.98 (d, *J* = 7.8 Hz, 1H), 7.41 (dd, *J* = 7.8, 4.8 Hz, 1H), 4.41 (q, *J* = 7.1 Hz, 2H), 3.63 (s, 3H), 2.64 (s, 3H), 1.41 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 163.8, 150.1, 149.6, 144.4, 137.7, 136.9, 129.4, 126.5, 123.5, 60.5, 32.0, 14.6, 10.6. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>13</sub>H<sub>16</sub>N<sub>3</sub>O<sub>2</sub>, 246.1238; found, 246.1235.

Ethyl 1,5-Dimethyl-2-(5-methylfuran-2-yl)-1H-imidazole-4-carboxylate (**3g**). Yellow solid, 102 mg, 41% yield, mp 68–69 °C. Silica gel TLC  $R_f$  = 0.20 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d):  $\delta$  6.79 (d, J = 3.2 Hz, 1H), 6.15–6.04 (m, 1H), 4.38 (q, J = 7.1 Hz, 2H), 3.73 (s, 3H), 2.59 (s, 3H), 2.37 (s, 3H), 1.40 (t, J = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d):  $\delta$  164.0, 153.1, 143.1, 139.4, 136.9, 128.6, 111.7, 107.6, 60.3, 31.8, 14.5, 13.8, 10.2. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>13</sub>H<sub>17</sub>N<sub>2</sub>O<sub>3</sub>, 249.1234; found, 249.1233.

*Ethyl* 1,2,5-*Trimethyl-1H-imidazole-4-carboxylate* (**3***h*). Light yellow oil, 127 mg, 70% yield. Silica gel TLC  $R_f = 0.41$  (chloroform/MeOH = 19/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 4.35 (q, J = 7.2 Hz, 2H), 3.45 (s, 3H), 2.51 (s, 3H), 2.39 (s, 3H), 1.38 (t, J = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 164.0, 144.4, 136.2, 127.1, 60.1, 30.3, 14.6, 13.5, 10.3. HRMS (ESI) m/z:  $[M + H]^+$  calcd for  $C_9H_{15}N_2O_2$ , 183.1129; found, 183.1130.

*Ethyl 2-Isopropyl-1,5-dimethyl-1H-imidazole-4-carboxylate* (**3***i*). Light yellow solid, 111 mg, 53% yield, mp 73–74 °C. Silica gel TLC  $R_f = 0.07$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 4.36 (q, *J* = 7.1 Hz, 2H), 3.48 (s, 3H), 3.02 (hept, *J* = 6.9 Hz, 1H), 2.51 (s, 3H), 1.46–1.27 (m, 9H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.2, 152.6, 136.1, 127.2, 60.2, 30.0, 26.6, 20.9, 14.6, 10.4. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>11</sub>H<sub>19</sub>N<sub>2</sub>O<sub>2</sub>, 211.1442; found, 211.1438.

*Ethyl* 2-(*tert-Butyl*)-1,5-*dimethyl*-1*H*-*imidazole-4*-*carboxylate* (*3j*). Light yellow oil, 110 mg, 49% yield. Silica gel TLC  $R_f = 0.38$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 4.34 (q, *J* = 7.1 Hz, 2H), 3.62 (s, 3H), 2.50 (s, 3H), 1.45 (s, 9H), 1.38 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.3, 154.0, 137.6, 126.3, 60.2, 33.3, 32.5, 29.1, 14.5, 10.5. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>12</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub>, 225.1598; found, 225.1594.

*Ethyl* 2-*Heptyl*-1,5-*dimethyl*-1*H*-*imidazole*-4-*carboxylate* (**3***k*). Light yellow oil, 144 mg, 54% yield. Silica gel TLC  $R_f = 0.21$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 4.36 (q, J = 7.1 Hz, 2H), 3.46 (s, 3H), 2.76–2.64 (m, 2H), 2.51 (s, 3H), 1.76–1.63 (m, 3H), 1.38 (t, J = 7.1 Hz, 3H), 1.36–1.23 (m, 7H), 0.88 (t, J = 6.7 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.2, 148.4, 136.1, 127.3, 60.1, 31.7, 30.3, 29.5, 29.0, 28.1, 27.6, 22.6, 14.6, 14.1, 10.3. HRMS (ESI) m/z:  $[M + H]^+$  calcd for C<sub>15</sub>H<sub>27</sub>N<sub>2</sub>O<sub>2</sub>, 267.2068; found, 267.2069.

*Ethyl* 2-*Cyclopropyl*-1,5-*dimethyl*-1*H*-*imidazole*-4-*carboxylate* (*3I*). Light yellow oil, 150 mg, 72% yield. Silica gel TLC  $R_f = 0.06$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 4.34 (q, *J* = 7.1 Hz, 2H), 3.58 (s, 3H), 2.51 (s, 3H), 1.78–1.70 (m, 1H), 1.37 (t, *J* = 7.1 Hz, 3H), 1.08–1.00 (m, 2H), 0.98–0.90 (m, 2H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.1, 148.9, 136.3, 126.9, 60.1, 30.0, 14.5, 10.3, 7.5, 6.3. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>11</sub>H<sub>17</sub>N<sub>2</sub>O<sub>2</sub>, 209.1285; found, 209.1281.

*Ethyl* 2-Cyclohexyl-1,5-dimethyl-1H-imidazole-4-carboxylate (*3m*). Light yellow solid, 175 mg, 70% yield, mp 85–86 °C. Silica gel TLC  $R_f$  = 0.23 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 4.36 (q, *J* = 7.1 Hz, 2H), 3.47 (s, 3H), 2.63 (tt, *J* = 12.1, 3.4 Hz, 1H), 2.51 (s, 3H), 1.92–1.83 (m, 4H), 1.81–1.68 (m, 3H), 1.38 (t, *J* = 7.1 Hz, 3H), 1.34–1.29 (m, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.2, 151.9, 135.8, 127.4, 60.1, 36.4, 31.0, 30.0, 26.3, 25.6, 14.6, 10.3. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>, 251.1755; found, 251.1756.

*Ethyl* 1,2-*Diisopropyl-5-methyl-1H-imidazole-4-carboxylate* (**3***n*). Light yellow oil, 130 mg, 55% yield. Silica gel TLC  $R_f = 0.36$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 4.53 (hept, *J* = 7.1 Hz, 1H), 4.35 (q, *J* = 7.1 Hz, 2H), 3.06 (hept, *J* = 6.9 Hz, 1H), 2.62 (s, 3H), 1.52 (d, *J* = 7.1 Hz, 6H), 1.38 (t, *J* = 7.1 Hz, 3H), 1.36 (d, *J* = 6.8 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.4, 60.1, 47.2, 27.2, 21.8, 14.6. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>13</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>, 239.1755; found, 239.1754.

Ethyl 1-Benzyl-5-methyl-2-phenyl-1H-imidazole-4-carboxylate (**30**). Light yellow oil, 203 mg, 63% yield. Silica gel TLC  $R_f = 0.22$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.51 (d, J = 6.7 Hz, 2H), 7.41–7.29 (m, 6H), 6.97 (d, J = 7.3 Hz, 2H), 5.19 (s, 2H), 4.42 (q, J = 7.1 Hz, 2H), 2.47 (s, 3H), 1.42 (t, J = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.1, 148.1, 137.1, 136.0, 130.0, 129.4, 129.2, 128.5, 127.9, 125.6, 60.4, 48.1, 14.6, 10.5. HRMS (ESI) m/z:  $[M + H]^+$  calcd for C<sub>20</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub>, 321.1598; found, 321.1596.

Article

*Ethyl* 1-*Benzyl-2-isopropyl-5-methyl-1H-imidazole-4-carboxylate* (*3p*). Light yellow oil, 202 mg, 71% yield. Silica gel TLC  $R_f$  = 0.39 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.34–7.27 (m, 3H), 6.90 (d, *J* = 6.8 Hz, 2H), 5.10 (s, 2H), 4.38 (q, *J* = 7.1 Hz, 2H), 2.92 (hept, *J* = 6.9 Hz, 1H), 2.42 (s, 3H), 1.40 (t, *J* = 7.1 Hz, 3H), 1.29 (d, *J* = 6.9 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.2, 153.3, 135.9, 135.7, 129.1, 127.9, 125.5, 60.3, 46.5, 26.6, 21.6, 14.6, 10.3. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>17</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>, 287.1755; found, 287.1755.

*Ethyl* 1-*Cyclohexyl-5-methyl-2-phenyl-1H-imidazole-4-carboxylate* (*3q*). White solid, 116 mg, 37% yield, mp 103–105 °C. Silica gel TLC  $R_f = 0.50$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.48–7.40 (m, 5H), 4.38 (q, *J* = 7.1 Hz, 2H), 4.08 (tt, *J* = 12.1, 4.0 Hz, 1H), 2.75 (s, 3H), 1.98–1.80 (m, 6H), 1.70–1.63 (m, 1H), 1.39 (t, *J* = 7.1 Hz, 3H), 1.24–1.10 (m, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.3, 147.8, 136.3, 131.4, 129.9, 129.2, 128.3, 60.2, 57.8, 32.0, 26.1, 25.1, 14.6, 12.1. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>25</sub>N<sub>2</sub>O<sub>2</sub>, 313.1911; found, 313.1911.

Ethyl 1-(4-Fluorophenyl)-5-methyl-2-phenyl-1H-imidazole-4carboxylate (**3r**). Beige solid, 135 mg, 42% yield, mp 134–136 °C. Silica gel TLC  $R_f$  = 0.87 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.35 (d, *J* = 7.2 Hz, 2H), 7.27–7.18 (m, 3H), 7.17 (d, *J* = 6.4 Hz, 4H), 4.44 (q, *J* = 7.1 Hz, 2H), 2.41 (s, 3H), 1.43 (t, *J* = 7.1 Hz, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 164.0, 162.6 (d, *J* = 250.8 Hz), 146.9, 138.1, 132.4, 129.8, 129.7, 129.5, 129.2, 128.8, 128.2, 116.96 (d, *J* = 23.0 Hz), 60.5, 14.6, 11.1. <sup>19</sup>F NMR (471 MHz, chloroform-*d*):  $\delta$  –113.9; HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>18</sub>FN<sub>2</sub>O<sub>2</sub>, 325.1347; found, 325.1344.

tert-Butyl 1,5-Dimethyl-2-phenyl-1H-imidazole-4-carboxylate (**4a**). White solid, 183 mg, 67% yield, mp 224–225 °C. Silica gel TLC  $R_f = 0.25$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d):  $\delta$  7.62–7.52 (m, 2H), 7.47–7.38 (m, 3H), 3.57 (s, 3H), 2.58 (s, 3H), 1.60 (s, 9H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d):  $\delta$  163.4, 147.3, 136.1, 130.3, 129.9, 129.2, 129.0, 128.5, 80.6, 32.0, 28.5, 10.7. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>16</sub>H<sub>21</sub>N<sub>2</sub>O<sub>2</sub>, 273.1598; found, 273.1599.

1-Methyl-2,4,5-triphenyl-1H-imidazole (**4b**). White solid, 222 mg, 72% yield, mp 146–148 °C. Silica gel TLC  $R_f$  = 0.78 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.75 (d, *J* = 7.5 Hz, 2H), 7.55 (d, *J* = 7.7 Hz, 2H), 7.52–7.39 (m, 8H), 7.21 (t, *J* = 7.5 Hz, 2H), 7.14 (t, *J* = 7.3 Hz, 1H), 3.51 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 147.9, 137.7, 134.6, 131.2, 131.0, 130.9, 130.5, 129.1, 129.1, 128.8, 128.6, 128.1, 127.0, 126.3, 33.2. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>22</sub>H<sub>19</sub>N<sub>2</sub>, 311.1543; found, 311.1542.

5-(4-Methoxyphenyl)-1-methyl-2,4-diphenyl-1H-imidazole (4c). White solid, 239 mg, 70% yield, mp 157–158 °C. Silica gel TLC  $R_{\rm f}$  = 0.69 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.74 (d, *J* = 7.5 Hz, 2H), 7.57 (d, *J* = 7.6 Hz, 2H), 7.49 (t, *J* = 7.5 Hz, 2H), 7.57 (d, *J* = 8.2 Hz, 2H), 7.42 (t, *J* = 7.4 Hz, 1H), 7.33 (d, *J* = 8.2 Hz, 2H), 7.22 (t, *J* = 7.6 Hz, 2H), 7.14 (t, *J* = 7.3 Hz, 1H), 7.01 (d, *J* = 8.2 Hz, 2H), 3.88 (s, 3H), 3.49 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 159.8, 147.6, 137.5, 134.8, 132.2, 131.1, 130.3, 129.1, 128.7, 128.6, 128.1, 126.8, 126.2, 123.3, 114.5, 55.3, 33.1. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>23</sub>H<sub>21</sub>N<sub>2</sub>O, 341.1649; found, 341.1648.

5-(4-Fluorophenyl)-1-methyl-2,4-diphenyl-1H-imidazole (4d). White solid, 255 mg, 78% yield, mp 168–170 °C. Silica gel TLC  $R_f$  = 0.77 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, DMSO-d<sub>6</sub>): δ 7.79 (d, *J* = 7.5 Hz, 2H), 7.59–7.48 (m, SH), 7.44 (d, *J* = 7.7 Hz, 2H), 7.39 (t, *J* = 8.6 Hz, 2H), 7.25 (t, *J* = 7.6 Hz, 2H), 7.16 (t, *J* = 7.4 Hz, 1H), 3.48 (s, 3H), 3.35 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, DMSO-d<sub>6</sub>): δ 162.7 (d, *J* = 246.1 Hz), 147.4, 137.1, 135.1, 133.5 (d, *J* = 8.3 Hz), 131.1, 129.9, 129.2, 129.0, 128.6, 127.6, 126.7, 116.6 (d, *J* = 21.3 Hz), 33.5. <sup>19</sup>F NMR (471 MHz, chloroform-*d*): δ –115.6; HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>22</sub>H<sub>18</sub>FN<sub>2</sub>, 329.1449; found, 329.1447.

5-(4-Chlorophenyl)-1-methyl-2,4-diphenyl-1H-imidazole (4e). White solid, 272 mg, 79% yield, mp 187–188 °C. Silica gel TLC  $R_f$  = 0.81 (hexane/EtOAc = 2/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.73 (d, *J* = 7.4 Hz, 2H), 7.55–7.47 (m, 4H), 7.47–7.41 (m, 3H), 7.36 (d, *J* = 8.0 Hz, 2H), 7.24 (t, *J* = 7.5 Hz, 2H), 7.17 (t, *J* = 7.3 Hz, 1H), 3.51 (s, 3H).  ${}^{13}C{}^{1}H$  NMR (126 MHz, chloroform-*d*):  $\delta$  148.3, 138.2, 134.7, 134.4, 132.2, 130.8, 129.7, 129.4, 129.1, 128.9, 128.6, 128.2, 127.1, 126.6, 33.2. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>22</sub>H<sub>18</sub>ClN<sub>2</sub>, 345.1154 and 347.1124; found, 345.1155 and 347.1126.

5-Benzyl-1-methyl-2,4-diphenyl-1H-imidazole (4f). White solid, 146 mg, 45% yield, mp 160–161 °C. Silica gel TLC  $R_{\rm f}$  = 0.69 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.71– 7.64 (m, 4H), 7.46 (t, *J* = 7.3 Hz, 2H), 7.41 (t, *J* = 7.4 Hz, 1H), 7.35 (q, *J* = 7.8 Hz, 4H), 7.29–7.22 (m, 2H), 7.21 (d, *J* = 7.5 Hz, 2H), 4.28 (s, 2H), 3.46 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 147.7, 139.4, 138.3, 135.0, 130.9, 129.1, 128.9, 128.7, 128.5, 128.5, 127.9, 127.3, 126.7, 126.7, 32.1, 30.5. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>23</sub>H<sub>21</sub>N<sub>2</sub>, 325.1700; found, 325.1698.

tert-Butyl 2-Isopropyl-1,5-dimethyl-1H-imidazole-4-carboxylate (**5a**). White solid, 116 mg, 49% yield, mp 99–101 °C. Silica gel TLC  $R_{\rm f}$  = 0.26 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 3.45 (s, 3H), 2.98 (hept, *J* = 6.9 Hz, 1H), 2.44 (s, 3H), 1.58 (s, 9H), 1.34 (d, *J* = 6.9 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 163.3, 152.4, 134.3, 128.6, 80.1, 30.0, 28.4, 26.6, 20.9, 10.5. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>13</sub>H<sub>23</sub>N<sub>2</sub>O<sub>2</sub>, 239.1755; found, 239.1753.

2-Isopropyl-1-methyl-4,5-diphenyl-1H-imidazole (**5b**). White solid, 188 mg, 68% yield, mp 92–93 °C. Silica gel TLC  $R_f = 0.70$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.48–7.36 (m, 5H), 7.32 (d, J = 6.1 Hz, 2H), 7.17 (t, J = 7.6 Hz, 2H), 7.09 (t, J = 7.4 Hz, 1H), 3.39 (s, 3H), 3.09 (hept, J = 6.9 Hz, 1H), 1.44 (d, J = 6.8 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 152.8, 136.2, 135.1, 131.5, 130.9, 128.9, 128.6, 128.2, 128.0, 126.9, 125.9, 30.7, 26.7, 21.4. HRMS (ESI) m/z:  $[M + H]^+$  calcd for C<sub>19</sub>H<sub>21</sub>N<sub>2</sub>, 277.1700; found, 277.1701.

2-Isopropyl-5-(4-methoxyphenyl)-1-methyl-4-phenyl-1H-imidazole (5c). White solid, 204 mg, 67% yield, mp 156–158 °C. Silica gel TLC  $R_f = 0.60$  (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.47 (d, J = 7.2 Hz, 2H), 7.24 (d, J = 8.7 Hz, 2H), 7.18 (t, J = 7.6 Hz, 2H), 7.09 (t, J = 7.3 Hz, 1H), 6.96 (d, J = 8.7 Hz, 2H), 3.86 (s, 3H), 3.37 (s, 3H), 3.08 (p, J = 6.9 Hz, 1H), 1.43 (d, J =6.9 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 159.6, 152.5, 136.0, 135.2, 132.2, 128.3, 128.0, 126.8, 125.8, 123.7, 114.4, 55.3, 30.6, 26.7, 21.4. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>23</sub>N<sub>2</sub>O, 307.1805; found, 307.1806.

5-( $\dot{4}$ -Fluorophenyl)-2-isopropyl-1-methyl-4-phenyl-1H-imidazole (**5d**). White solid, 216 mg, 73% yield, mp 138–139 °C. Silica gel TLC  $R_f = 0.29$  (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroformd): δ 7.43 (d, J = 7.0 Hz, 2H), 7.34–7.27 (m, 2H), 7.19 (t, J = 7.6 Hz, 3H), 7.16–7.09 (m, 2H), 3.38 (s, 3H), 3.09 (p, J = 6.9 Hz, 1H), 1.43 (d, J = 6.9 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 162.70 (d, J = 248.1 Hz), 152.9, 136.5, 134.9, 132.8 (d, J = 8.2 Hz), 128.1, 127.5 (d, J = 3.1 Hz), 127.4, 126.9, 126.0, 116.1 (d, J = 21.8 Hz), 30.7, 26.7, 21.4. <sup>19</sup>F NMR (471 MHz, chloroform-d):  $\delta$  –116.3; HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>20</sub>FN<sub>2</sub>, 295.1606; found, 295.1606.

5-(4-Chlorophenyl)-2-isopropyl-1-methyl-4-phenyl-1H-imidazole (**5e**). White solid, 224 mg, 72% yield, mp 132–133 °C. Silica gel TLC  $R_f = 0.36$  (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.47–7.38 (m, 4H), 7.26 (d, *J* = 3.9 Hz, 2H), 7.20 (t, *J* = 7.5 Hz, 2H), 7.12 (t, *J* = 7.3 Hz, 1H), 3.39 (s, 3H), 3.08 (p, *J* = 6.9 Hz, 1H), 1.43 (d, *J* = 6.8 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 153.2, 136.8, 134.8, 134.3, 132.2, 129.9, 129.2, 128.2, 127.2, 127.0, 126.2, 30.8, 26.7, 21.4. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>20</sub>ClN<sub>2</sub>, 311.1310 and 313.1281; found, 311.1311 and 313.1283.

5-Benzyl-2-isopropyl-1-methyl-4-phenyl-1H-imidazole (5f). White solid, 166 mg, 57% yield, mp 104–106 °C. Silica gel TLC  $R_f$  = 0.58 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.60 (d, J = 7.3 Hz, 2H), 7.35–7.28 (m, 4H), 7.25–7.17 (m, 2H), 7.14 (d, J = 7.4 Hz, 2H), 4.16 (s, 2H), 3.32 (s, 3H), 3.03 (p, J = 6.9 Hz, 1H), 1.39 (d, J = 6.9 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 152.7, 138.7, 137.8, 135.5, 128.8, 128.4, 127.8, 127.2, 126.5, 126.3, 124.7, 30.2, 30.1, 26.5, 21.3. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>23</sub>N<sub>2</sub>, 291.1856; found, 291.1857.

2-(4-Methoxyphenyl)-1-methyl-4,5-diphenyl-1H-imidazole (**6a**). White solid, 234 mg, 69% yield, mp 169–171 °C. Silica gel TLC  $R_f$  = 0.65 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.67 (d, *J* = 8.7 Hz, 2H), 7.54 (d, *J* = 7.6 Hz, 2H), 7.50–7.39 (m, SH), 7.21 (t, *J* = 7.5 Hz, 2H), 7.14 (t, *J* = 7.3 Hz, 1H), 7.02 (d, *J* = 8.7 Hz, 2H), 3.87 (s, 3H), 3.48 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 160.0, 147.9, 137.5, 134.7, 130.9, 130.5, 130.1, 129.0, 128.5, 128.1, 126.9, 126.3, 123.4, 114.0, 55.4, 33.1. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>23</sub>H<sub>21</sub>N<sub>2</sub>O, 341.1649; found, 341.1651.

1-Methyl-2-(5-methylfuran-2-yl)-4,5-diphenyl-1H-imidazole (**6b**). White solid, 186 mg, 59% yield, mp 137–139 °C. Silica gel TLC  $R_{\rm f} = 0.44$  (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroformd): δ 7.52–7.48 (m, 2H), 7.47–7.43 (m, 3H), 7.37–7.34 (m, 2H), 7.21–7.16 (m, 2H), 7.15–7.10 (m, 1H), 6.77 (d, *J* = 3.3 Hz, 1H), 6.14–6.11 (m, 1H), 3.60 (s, 3H), 2.40 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 153.1, 143.7, 139.7, 137.9, 134.3, 131.0, 130.7, 130.0, 129.1, 128.7, 128.0, 127.1, 126.4, 111.1, 107.6, 32.7, 13.8. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>21</sub>H<sub>19</sub>N<sub>2</sub>O, 315.1492; found, 315.1493.

1,2-Dimethyl-4,5-diphenyl-1H-imidazole (6c). White solid, 172 mg, 69% yield, mp 109–110 °C. Silica gel TLC  $R_{\rm f}$  = 0.17 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.47–7.40 (m, SH), 7.32 (d, *J* = 7.6 Hz, 2H), 7.18 (t, *J* = 7.5 Hz, 2H), 7.11 (t, *J* = 7.3 Hz, 1H), 3.37 (s, 3H), 2.50 (s, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 144.7, 136.2, 134.7, 131.4, 130.9, 128.9, 128.9, 128.4, 128.1, 126.7, 126.1, 31.1, 13.6. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>17</sub>H<sub>17</sub>N<sub>2</sub>, 249.1387; found, 249.1389.

2-(tert-Butyl)-1-methyl-4,5-diphenyl-1H-imidazole (**6d**). Light beige solid, 146 mg, 50% yield, mp 123–124 °C. Silica gel TLC  $R_f$  = 0.62 (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.48–7.40 (m, 5H), 7.34–7.30 (m, 2H), 7.16 (t, *J* = 7.6 Hz, 2H), 7.08 (t, *J* = 7.3 Hz, 1H), 3.51 (s, 3H), 1.54 (s, 9H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 153.9, 135.1, 135.0, 131.7, 131.2, 130.2, 128.9, 128.4, 128.0, 126.7, 125.8, 33.6, 33.2, 29.5. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>23</sub>N<sub>2</sub>, 291.1856; found, 291.1858.

2-Cyclopropyl-1-methyl-4,5-diphenyl-1H-imidazole (**6e**). White solid, 160 mg, 58% yield, mp 110–111 °C. Silica gel TLC  $R_{\rm f}$  = 0.24 (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.46–7.38 (m, 5H), 7.34–7.30 (m, 2H), 7.16 (t, *J* = 7.8 Hz, 2H), 7.09 (t, *J* = 7.3 Hz, 1H), 3.49 (s, 3H), 1.91–1.80 (m, 1H), 1.20–1.10 (m, 2H), 1.04–0.95 (m, 2H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 149.2, 135.9, 134.9, 131.4, 130.9, 128.9, 128.9, 128.3, 128.0, 126.8, 125.9, 30.8, 7.6, 6.6. HRMS (ESI) *m*/*z*: [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>19</sub>N<sub>2</sub>, 275.1543; found, 275.1545.

2-Cyclohexyl-1-methyl-4,5-diphenyl-1H-imidazole (**6f**). White solid, 249 mg, 79% yield, mp 119–120 °C. Silica gel TLC  $R_{\rm f}$  = 0.58 (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroform-d): δ 7.48–7.36 (m, 5H), 7.35–7.29 (m, 2H), 7.17 (t, J = 7.6 Hz, 2H), 7.09 (t, J = 7.3 Hz, 1H), 3.39 (s, 3H), 2.71 (tt, J = 11.7, 3.5 Hz, 1H), 2.04–1.97 (m, 2H), 1.95–1.87 (m, 2H), 1.87–1.73 (m, 3H), 1.47–1.33 (m, 3H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 152.2, 136.4, 135.2, 131.6, 130.9, 128.9, 128.4, 128.2, 128.0, 126.9, 125.9, 36.6, 31.6, 30.7, 26.5, 25.9. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>22</sub>H<sub>25</sub>N<sub>2</sub>, 317.2013; found, 317.2014.

1,2-Diisopropyl-4,5-diphenyl-1H-imidazole (**6***g*). White solid, 72 mg, 24% yield, mp 171–172 °C. Silica gel TLC  $R_{\rm f}$  = 0.51 (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.44–7.41 (m, 3H), 7.37 (d, *J* = 7.3 Hz, 2H), 7.37–7.30 (m, 2H), 7.13 (t, *J* = 7.5 Hz, 2H), 7.05 (t, *J* = 7.3 Hz, 1H), 4.30 (p, *J* = 7.1 Hz, 1H), 3.18 (p, *J* = 6.8 Hz, 1H), 1.46 (d, *J* = 6.8 Hz, 6H), 1.39 (d, *J* = 7.1 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 152.6, 136.5, 135.2, 132.7, 131.8, 128.8, 128.4, 127.9, 127.4, 126.7, 125.7, 47.1, 27.8, 22.8, 22.6. HRMS (ESI) *m/z*: [M + H]<sup>+</sup> calcd for C<sub>21</sub>H<sub>25</sub>N<sub>2</sub>, 305.2013; found, 305.2014.

*1-Cyclohexyl-2,4,5-triphenyl-1H-imidazole (6h).* White solid, 121 mg, 32% yield, mp 165–166 °C. Silica gel TLC  $R_f$  = 0.81 (hexane/EtOAc = 1/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*):  $\delta$  7.66–7.60 (m, 2H), 7.51–7.39 (m, 10H), 7.14 (t, *J* = 7.5 Hz, 2H), 7.08 (t, *J* = 7.3 Hz, 1H), 3.97 (tt, *J* = 12.3, 3.5 Hz, 1H), 1.85 (d, *J* = 12.4 Hz, 2H), 1.68–1.49 (m, 4H), 1.45 (d, *J* = 13.5 Hz, 1H), 1.04 (qt, *J* = 13.0, 3.6

Hz, 2H), 0.73 (qt, J = 13.2, 3.7 Hz, 1H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-d): δ 147.7, 137.8, 134.7, 132.6, 132.5, 132.2, 130.1, 129.1, 128.9, 128.8, 128.7, 128.4, 127.9, 126.7, 126.0, 58.4, 33.6, 26.2, 25.1. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>27</sub>H<sub>27</sub>N<sub>2</sub>, 379.2169; found, 379.2168.

1-Benzyl-2-isopropyl-4,5-diphenyl-1H-imidazole (**6***i*). White solid, 293 mg, 83% yield, mp 120–121 °C. Silica gel TLC  $R_f$  = 0.58 (hexane/EtOAc = 4/1); <sup>1</sup>H NMR (500 MHz, chloroform-*d*): δ 7.50 (d, *J* = 7.1 Hz, 2H), 7.35–7.24 (m, 5H), 7.24–7.16 (m, 5H), 7.11 (d, *J* = 7.3 Hz, 1H), 6.91 (d, *J* = 7.3 Hz, 2H), 4.99 (s, 2H), 2.90 (hept, *J* = 6.9 Hz, 1H), 1.34 (d, *J* = 6.8 Hz, 6H). <sup>13</sup>C{<sup>1</sup>H} NMR (126 MHz, chloroform-*d*): δ 153.3, 137.6, 131.3, 131.1, 128.8, 128.8, 128.4, 128.3, 128.1, 127.4, 126.9, 126.0, 125.7, 46.7, 26.7, 22.0. HRMS (ESI) m/z: [M + H]<sup>+</sup> calcd for C<sub>25</sub>H<sub>25</sub>N<sub>2</sub>, 353.2013; found, 353.2016.

#### ASSOCIATED CONTENT

#### **9** Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.joc.9b03288.

Detailed optimization of the reaction conditions and copies of  ${}^{1}H$ ,  ${}^{13}C{}^{1}H$ , and 2D NMR spectra (PDF)

#### AUTHOR INFORMATION

#### **Corresponding Author**

Iván Kanizsai – AVIDIN Ltd., Szeged H-6726, Hungary; orcid.org/0000-0003-0109-7097; Email: i.kanizsai@ avidinbiotech.com

### Authors

Anikó Angyal – AVIDIN Ltd., Szeged H-6726, Hungary; Department of Organic Chemistry, University of Szeged, H-6720 Szeged, Hungary

András Demjén – AVIDIN Ltd., Szeged H-6726, Hungary János Wölfling – Department of Organic Chemistry, University of Szeged, H-6720 Szeged, Hungary

László G. Puskás – AVIDIN Ltd., Szeged H-6726, Hungary

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.joc.9b03288

#### Notes

The authors declare no competing financial interest.

#### REFERENCES

(1) For general reviews, see: (a) Palacios, F.; de Retana, A. M. O.; de Marigorta, E. M.; de los Santos, J. M. 2H-Azirines as Synthetic Tools in Organic Chemistry. Eur. J. Org. Chem. 2001, 2401-2414. (b) Palacios, F.; de Retana, A. M. O.; de Marigorta, E. M.; de los Santos, J. M. Preparation, Properties and Synthetic Applications of 2H-Azirines A Review. Org. Prep. Proced. Int. 2002, 34, 219-269 . For selected examples, see:. (c) Angyal, A.; Demjén, A.; Wéber, E.; Kovács, A. K.; Wölfling, J.; Puskás, L. G.; Kanizsai, I. Lewis Acid-Catalyzed Diastereoselective Synthesis of Multisubstituted N-Acylaziridine-2-carboxamides from 2H-Azirines via Joullié-Ugi Three-Component Reaction. J. Org. Chem. 2018, 83, 3570-3581. (d) Nakamura, S. Enantioselective Reaction of 2H-Azirines. Chem.-Asian J. 2019, 14, 1323-1330. (e) Zhao, M.-N.; Zhang, W.; Wang, X.-C.; Zhang, Y.; Yang, D.-S.; Guan, Z.-H. Modular 2,3-diaryl-2H-azirine synthesis from ketoxime acetates via Cs2CO3-mediated cyclization. Org. Biomol. Chem. 2018, 16, 4333-4337.

(2) For general reviews, see: (a) Padwa, A. In Comprehensive Heterocyclic Chemistry III, Volume 1: Three-Membered Heterocycles, Together with All Fused Systems Containing a Three-Membered Heterocyclic Ring; Katritzky, A. R., Scriven, E. F. V., Ramsden, C. A., Taylor, R. J. K., Eds.; Elsevier Science: Oxford, 2008; pp 1–104.
(b) Alves, M. J.; Costa, F. T. In Heterocyclic Targets in Advanced *Organic Synthesis*; Carreiras, M. C., Marco-Contelles, J., Eds.; Research Signpost: Kerala, India, 2011; pp 145–172.

(3) (a) Padwa, A. Cycloaddition and Cyclization Chemistry of 2H-Azirines. Adv. Heterocycl. Chem. 2010, 99, 1–31. (b) Heintzelman, G. R.; Meigh, I. R.; Mahajan, Y. R.; Weinreb, S. M. Diels-Alder Reactions of Imino Dienophiles. Org. React. 2005, 65, 141–599. (c) Shen, M.-H.; Xu, H.-D.; Zhou, H. Evolution of the Aza-Diels–Alder Reaction of 2H-Azirines. Synlett 2016, 27, 2171–2177. (d) Alves, M. J.; Durães, M. M.; Fortes, A. G. Diels–Alder Cycloaddition of 2-Azadienes to Methyl 2-(2,6-Dichlorophenyl)-2H-azirine-3-carboxylate in the Synthesis of Methyl 4-Oxo-1,3-diazabicyclo[4.1.0]heptane-6-carboxylates. Tetrahedron 2004, 60, 6541–6553.

(4) (a) Khlebnikov, A. F.; Novikov, M. S. Recent advances in 2*H*-azirine chemistry. *Tetrahedron* **2013**, *69*, 3363–3401. (b) Müller, F.; Mattay, J. [3+2] Cycloadditions with Azirines under the Conditions of Photoinduced Electron Transfer: A New Method for the Synthesis of Imidazoles and Heterophanes. *Chem. Ber.* **1993**, *126*, 543–549. (c) Wendling, L. A.; Bergman, R. G. Carbon-Carbon Bond Cleavage and Iminocarbene Formation in the Thermal Decomposition of 2*H*-Azirines. *J. Am. Chem. Soc.* **1974**, *96*, 308–309. (d) Cludius-Brandt, S.; Kupracz, L.; Kirschning, A. [3+2]-Cycloadditions of nitrile ylides after photoactivation of vinyl azides under flow conditions. *Beilstein J. Org. Chem.* **2013**, *9*, 1745–1750. (e) Isomura, K.; Ayabe, G.-I.; Hatano, S.; Taniguchi, H. Evidence for Vinyl Nitrene Intermediates in the Thermal Rearrangement of 2*H*-Azirines into Indoles. *J. Chem. Soc., Chem. Commun.* **1980**, 1252–1253.

(5) For selected examples, see: (a) Li, T.; Yan, H.; Li, X.; Wang, C.; Wan, B. Ruthenium-Catalyzed [3+2] Cycloaddition of 2H-Azirines with Alkynes: Access to Polysubstituted Pyrroles. J. Org. Chem. 2016, 81, 12031-12037. , and references cited therein (b) Zhao, M.-N.; Ren, Z.-H.; Yang, D.-S.; Guan, Z.-H. Iron-Catalyzed Radical Cycloaddition of 2H-Azirines and Enamides for the Synthesis of Pyrroles. Org. Lett. 2018, 20, 1287-1290. (c) Wang, Y.; Lei, X.; Tang, Y. Rh(II)-catalyzed cycloadditions of 1-tosyl 1,2,3-triazoles with 2Hazirines: switchable reactivity of Rh-azavinylcarbene as [2C]- or aza-[3C]-synthon. Chem. Commun. 2015, 51, 4507-4510. (d) Xuan, J.; Xia, X.-D.; Zeng, T.-T.; Feng, Z.-J.; Chen, J.-R.; Lu, L.-Q.; Xiao, W.-J. Visible-Light-Induced Formal [3+2] Cycloaddition for Pyrrole Synthesis under Metal-Free Conditions. Angew. Chem., Int. Ed. 2014, 53, 5653-5656. (e) Khlebnikov, A.; Funt, L.; Tomashenko, O.; Novikov, M. An Azirine Strategy for the Synthesis of Alkyl 4-Amino-5-(trifluoromethyl)-1H-pyrrole-2-carboxylates. Synthesis 2018, 50, 4809-4822. , and references cited therein

(6) For selected examples, see: (a) Padwa, A.; Stengel, T. Grubbs and Wilkinson catalyzed reactions of 2-phenyl-3-vinyl substituted 2*H*-azirines. Arkivoc **2005**, 2005, 21–32. (b) Hegedus, L. S.; Kramer, A.; Yijun, C. Reactions of Chromium Carbene Complexes with 1-Azirines. Synthesis of N-Vinylimidates. Organometallics **1985**, *4*, 1747–1750.

(7) For selected examples, see: (a) Prechter, A.; Henrion, G.; dit Bel, P. F.; Gagosz, F. Gold-Catalyzed Synthesis of Functionalized Pyridines by Using 2H-Azirines as Synthetic Equivalents of Alkenyl Nitrenes. Angew. Chem., Int. Ed. 2014, 53, 4959–4963. (b) Sujatha, C.; Bhatt, C. S.; Ravva, M. K.; Suresh, A. K.; Namitharan, K. Copper-Catalyzed Ring-Expansion Cascade of Azirines with Alkynes: Synthesis of Multisubstituted Pyridines at Room Temperature. Org. Lett. 2018, 20, 3241–3244. (c) Loy, N. S. Y.; Singh, A.; Xu, X.; Park, C.-M. Synthesis of Pyridines by Carbenoid-Mediated Ring Opening of 2H-Azirines. Angew. Chem., Int. Ed. 2013, 52, 2212–2216.

(8) For selected examples, see: (a) Loy, N. S. Y.; Kim, S.; Park, C.-M. Synthesis of Unsymmetrical Pyrazines Based on  $\alpha$ -Diazo Oxime Ethers. Org. Lett. **2015**, 17, 395–397. (b) Palacios, F.; de Retana, A. M. O.; Gil, J. I.; de Munain, R. L. Synthesis of Pyrazine-phosphonates and -Phosphine Oxides from 2H-Azirines or Oximes. Org. Lett. **2002**, 4, 2405–2408. (c) Ryu, T.; Baek, Y.; Lee, P. H. Synthesis of Pyrazines from Rhodium-Catalyzed Reaction of 2H-Azirines withN-Sulfonyl 1,2,3-Triazoles. J. Org. Chem. **2015**, 80, 2376–2383.

(9) For selected examples, see: (a) Thangaraj, M.; Bhojgude, S. S.; Jain, S.; Gonnade, R. G.; Biju, A. T. Selective Synthesis of N-

#### The Journal of Organic Chemistry

Unsubstituted and N-Arylindoles by the Reaction of Arynes with Azirines. J. Org. Chem. **2016**, 81, 8604–8611. (b) Wentrup, C.; Freiermuth, B. Pyrolysis of benzotriazoles. Relationships between 1- and 2-vinylbenzotriazoles,  $\alpha$ - and  $\beta$ -azidostyrenes, N-phenylketenimine and indole. Pitfalls in the use of pyrolysis-mass spectrometry in mechanistic studies. J. Anal. Appl. Pyrolysis **2016**, 121, 67–74. (c) Khaidarov, A. R.; Rostovskii, N. V.; Zolotarev, A. A.; Khlebnikov, A. F.; Novikov, M. S. Synthesis of 1-(2-Aminovinyl)indoles and 1,3'-Biindoles by Reaction of 2,2-Diaryl-Substituted 2H-Azirines with  $\alpha$ -Imino Rh(II) Carbenoids. J. Org. Chem. **2019**, 84, 3743–3753. (d) Candito, D. A.; Lautens, M. Exploiting the Chemistry of Strained Rings: Synthesis of Indoles via Domino Reaction of Aryl Iodides with 2H-Azirines. Org. Lett. **2010**, 12, 3312–3315.

(10) (a) Xiang, L.; Niu, Y.; Pang, X.; Yang, X.; Yan, R. I<sub>2</sub>-catalyzed synthesis of substituted imidazoles from vinyl azides and benzylamines. *Chem. Commun.* **2015**, *51*, 6598–6600. (b) Rossa, T. A.; Fantinel, M.; Bortoluzzi, A. J.; Sá, M. M. Multicomponent Synthesis of Structurally Diverse Imidazoles Featuring Azirines, Amines and Aldehydes. *Eur. J. Org. Chem.* **2018**, 4171–4177. (c) Auricchio, S.; Truscello, A. M.; Lauria, M.; Meille, S. V. Ambivalent role of metal chlorides in ring opening reactions of 2*H*-azirines: synthesis of imidazoles, pyrroles and pyrrolinones. *Tetrahedron* **2012**, *68*, 7441– 7449. (d) Shi, S.; Xu, K.; Jiang, C.; Ding, Z. ZnCl<sub>2</sub>-Catalyzed [3+2] Cycloaddition of Benzimidates and 2*H*-Azirines for the Synthesis of Imidazoles. *J. Org. Chem.* **2018**, *83*, 14791–14796. (e) Cardoso, A. L.; Lemos, A.; Pinho e Melo, T. M. V. D. Selective Synthesis of Tetrasubstituted 4-(Tetrazol-5-yl)-1*H*-imidazoles from 2-(Tetrazol-5yl)-2*H*-azirines. *Eur. J. Org. Chem.* **2014**, 5159–5165.

(11) For selected examples, see: (a) Khlebnikov, A. F.; Novikov, M. S.; Petrovskii, P. P.; Stoeckli-Evans, H. An Aza Cyclopropylcarbinyl-Homoallyl Radical Rearrangement-Radical Cyclization Cascade. Synthesis of Dibenzoimidazoazepine and Oxazepine Derivatives. J. Org. Chem. 2011, 76, 5384-5391. (b) Narasimhan, N. S.; Heimgartner, H.; Hansen, H.-J. r.; Schmid, H. Thermische Reaktionen mit 3-Phenyl-2H-azirinen; 1, 3-dipolare Cycloadditionen und En-Reaktionen. Helv. Chim. Acta 1973, 56, 1351-1370. (c) Brown, D.; Brown, G. A.; Andrews, M.; Large, J. M.; Urban, D.; Butts, C. P.; Hales, N. J.; Gallagher, T. The azomethine ylide strategy for  $\beta$ -lactam synthesis. Azapenams and 1-azacephams. J. Chem. Soc., Perkin Trans. 1 2002, 2014-2021. (d) Chandrasekhar, D.; Borra, S.; Nanubolu, J. B.; Maurya, R. A. Visible Light Driven Photocascade Catalysis: Ru(bpy)<sub>3</sub>(PF<sub>6</sub>)<sub>2</sub>/TBHP-Mediated Synthesis of Fused  $\beta$ -Carbolines in Batch and Flow Microreactors. Org. Lett. 2016, 18, 2974-2977. (e) Angyal, A.; Demjén, A.; Harmat, V.; Wölfling, J.; Puskás, L. G.; Kanizsai, I. 1,3-Dipolar Cycloaddition of Isatin-Derived Azomethine Ylides with 2H-Azirines: Stereoselective Synthesis of 1,3-Diazaspiro[bicyclo[3.1.0]hexane]oxindoles. J. Org. Chem. 2019, 84, 4273-4281.

(12) For general reviews, see: (a) Hamer, J.; Macaluso, A. Nitrones. *Chem. Rev.* **1964**, 64, 473–495. (b) Murahashi, S.-I.; Imada, Y. Synthesis and Transformations of Nitrones for Organic Synthesis. *Chem. Rev.* **2019**, 119, 4684–4716. (c) Anderson, L. L. Diverse Applications of Nitrones for the Synthesis of Heterocyclic Compounds. *Asian J. Org. Chem.* **2016**, 5, 9–30.

(13) For selected examples, see: (a) Chen, D.; Wang, Z.; Li, J.; Yang, Z.; Lin, L.; Liu, X.; Feng, X. Catalytic Asymmetric 1,3-Dipolar Cycloaddition of Nitrones to Alkylidene Malonates: Highly Enantioselective Synthesis of Multisubstituted Isoxazolidines. *Chem.*—*Eur. J.* **2011**, *17*, 5226–5229. (b) Xie, L.; Yu, X.; Li, J.; Zhang, Z.; Qin, Z.; Fu, B. Ni(II)-Catalyzed Enantioselective 1,3-Dipolar Cycloaddition of Nitrones with  $\alpha$ ,  $\beta$ -Unsaturated Acyl Carboxylates. *Eur. J. Org. Chem.* **2017**, 657–661. (c) Yang, X.; Cheng, F.; Kou, Y.-D.; Pang, S.; Shen, Y.-C.; Huang, Y.-Y.; Shibata, N. Catalytic Asymmetric 1,3-Dipolar Cycloaddition of  $\beta$ -Fluoroalkylated  $\alpha,\beta$ -Unsaturated 2-Pyridylsulfones with Nitrones for Chiral Fluoroalkylated Isoxazolidines and  $\gamma$ -Amino Alcohols. *Angew. Chem., Int. Ed.* **2017**, *56*, 1510–1514. (d) Hashimoto, T.; Maruoka, K. Recent Advances of Catalytic Asymmetric 1,3-Dipolar Cycloadditions. *Chem. Rev.* **2015**, *115*, 5366–5412. (e) Gothelf, K. V.; Jørgensen, K. A.

Asymmetric 1,3-Dipolar Cycloaddition Reactions. Chem. Rev. 1998, 98, 863-910.

(14) For selected examples, see: (a) Honda, K.; Mikami, K. Asymmetric "Acetylenic" [3+2] Cycloaddition of Nitrones Catalyzed by Cationic Chiral Pd<sup>II</sup> Lewis Acid. Chem.-Asian J. 2018, 13, 2838-2841. (b) Xiao, Z.-F.; Ding, T.-H.; Mao, S.-W.; Ning, X.-S.; Kang, Y.-B. Zinc Iodide-Mediated Direct Synthesis of 2,3-Dihydroisoxazoles from Alkynes and Nitrones. Adv. Synth. Catal. 2016, 358, 1859-1863. (c) Alemán, J.; Fraile, A.; Marzo, L.; Ruano, J. L. G.; Izquierdo, C.; Díaz-Tendero, S. Enantioselective Synthesis of 4-Isoxazolines by 1.3-Dipolar Cycloadditions of Nitrones to Alkynals Catalyzed by Fluorodiphenylmethylpyrrolidines. Adv. Synth. Catal. 2012, 354, 1665-1671. (d) Morse, P. D.; Jamison, T. F. Synthesis and Utilization of Nitroalkyne Equivalents in Batch and Continuous Flow. Angew. Chem., Int. Ed. 2017, 56, 13999-14002. (e) González-Cruz, D.; Tejedor, D.; de Armas, P.; García-Tellado, F. Dual Reactivity Pattern of Allenolates "On Water": The Chemical Basis for Efficient Allenolate-Driven Organocatalytic Systems. Chem.-Eur. J. 2007, 13, 4823-4832.

(15) (a) Akmanova, N. A.; Sagitdinova, K. F.; Balenkova, E. S. Isoxazolidine derivatives with a cyclopropyl residue. *Chem. Heterocycl. Compd.* **1982**, *18*, 910–912. (b) Diev, V. V.; Stetsenko, O. N.; Tung, T. Q.; Kopf, J.; Kostikov, R. R.; Molchanov, A. P. Nitrone Cycloadditions to 1,2-Diphenylcyclopropenes and Subsequent Transformations of the Isoxazolidine Cycloadducts. *J. Org. Chem.* **2008**, *73*, 2396–2399. (c) Adly, F. G.; Marichev, K. O.; Jensen, J. A.; Arman, H.; Doyle, M. P. Enoldiazosulfones for Highly Enantioselective [3+3]-Cycloaddition with Nitrones Catalyzed by Copper(I) with Chiral BOX Ligands. *Org. Lett.* **2019**, *21*, 40–44.

(16) For selected examples, see: (a) Uçucu, Ü.; Karaburun, N. G.; Işikdağ, İ. Synthesis and analgesic activity of some 1-benzyl-2substituted-4,5-diphenyl-1H-imidazole derivatives. Il Farmaco 2001, 56, 285-290. (b) Seo, H. J.; Park, E.-J.; Kim, M. J.; Kang, S. Y.; Lee, S. H.; Kim, H. J.; Lee, K. N.; Jung, M. E.; Lee, M.; Kim, M.-S.; Son, E.-J.; Park, W.-K.; Kim, J.; Lee, J. Design and Synthesis of Novel Arylpiperazine Derivatives Containing the Imidazole Core Targeting 5-HT2A Receptor and 5-HT Transporter. J. Med. Chem. 2011, 54, 6305-6318. (c) Cho, H. J.; Gee, H. Y.; Baek, K.-H.; Ko, S.-K.; Park, J.-M.; Lee, H.; Kim, N.-D.; Lee, M. G.; Shin, I. A Small Molecule That Binds to an ATPase Domain of Hsc70 Promotes Membrane Trafficking of Mutant Cystic Fibrosis Transmembrane Conductance Regulator. J. Am. Chem. Soc. 2011, 133, 20267-20276. (d) Gleave, R. J.; Walter, D. S.; Beswick, P. J.; Fonfria, E.; Michel, A. D.; Roman, S. A.; Tang, S.-P. Synthesis and biological activity of a series of tetrasubstituted-imidazoles as P2X7 antagonists. Bioorg. Med. Chem. Lett. 2010, 20, 4951-4954.

(17) For selected examples, see: (a) Scior, T.; Domeyer, D. M.; Cuanalo-Contreras, K.; Laufer, S. A. Pharmacophore Design of p38 $\alpha$ MAP Kinase Inhibitors with Either 2,4,5-Trisubstituted or 1,2,4,5-Tetrasubstituted Imidazole Scaffold. *Curr. Med. Chem.* 2011, 18, 1526–1539. (b) Laufer, S. A.; Zimmermann, W.; Ruff, K. J. Tetrasubstituted Imidazole Inhibitors of Cytokine Release: Probing Substituents in the N-1 Position. *J. Med. Chem.* 2004, 47, 6311–6325. (c) Muth, F.; Günther, M.; Bauer, S. M.; Döring, E.; Fischer, S.; Maier, J.; Drückes, P.; Köppler, J.; Trappe, J.; Rothbauer, U.; Koch, P.; Laufer, S. A. Tetra-Substituted Pyridinylimidazoles As Dual Inhibitors of p38 $\alpha$  Mitogen-Activated Protein Kinase and c-Jun N-Terminal Kinase 3 for Potential Treatment of Neurodegenerative Diseases. *J. Med. Chem.* 2015, 58, 443–456. , and references cited therein

(18) (a) Ren, J.; Nichols, C.; Bird, L. E.; Fujiwara, T.; Sugimoto, H.; Stuart, D. I.; Stammers, D. K. Binding of the Second Generation Nonnucleoside Inhibitor S-1153 to HIV-1 Reverse Transcriptase Involves Extensive Main Chain Hydrogen Bonding. *J. Biol. Chem.* **2000**, *275*, 14316–14320. (b) Ganguly, S.; Vithlani, V.; Kesharwani, A.; Kuhu, R.; Baskar, L.; Mitramazumder, P.; Sharon, A.; Dev, A. Synthesis, antibacterial and potential anti-HIV activity of some novel imidazole analogs. *Acta Pharm.* **2011**, *61*, 187–201.

# The Journal of Organic Chemistry

(19) (a) Maheta, H. K.; Patel, A. S.; Naliapara, Y. T. Synthesis and microbial study of some novel cyanopyrans and cyanopyridines containing imidazole nucleus. *Int. J. Chem. Sci.* **2012**, *10*, 1815–1829. (b) Fang, Y.; Yuan, R.; Ge, W.-h.; Wang, Y.-j.; Liu, G.-x.; Li, M.-q.; Xu, J.-b.; Wan, Y.; Zhou, S.-l.; Han, X.-g.; Zhang, P.; Liu, J.-j.; Wu, H. Synthesis and biological evaluation of 1,2,4,5-tetrasubstituted imidazoles. *Res. Chem. Intermed.* **2017**, *43*, 4413–4421.

(20) For selected examples, see: (a) Williams, D. R.; Ko, S.-K.; Park, S.; Lee, M.-R.; Shin, I. An Apoptosis-Inducing Small Molecule That Binds to Heat Shock Protein 70. *Angew. Chem., Int. Ed.* **2008**, *47*, 7466–7469. (b) Sharma, G. V. M.; Ramesh, A.; Singh, A.; Srikanth, G.; Jayaram, V.; Duscharla, D.; Jun, J. H.; Ummanni, R.; Malhotra, S. V. Imidazole derivatives show anticancer potential by inducing apoptosis and cellular senescence. *MedChemComm* **2014**, *5*, 1751–1760. (c) Semones, M.; Feng, Y.; Johnson, N.; Adams, J. L.; Winkler, J.; Hansbury, M. Pyridinylimidazole inhibitors of Tie2 kinase. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 4756–4760. (d) Johnson, N. W.; Semones, M.; Adams, J. L.; Hansbury, M.; Winkler, J. Optimization of triarylimidazoles for Tie2: Influence of conformation on potency. *Bioorg. Med. Chem. Lett.* **2007**, *17*, 5514–5517.

(21) (a) Harpp, D. N.; McDonald, J. G. Synthesis of thioamides from aldonitrones utilizing thiocarbonyl transfer reagents. *Tetrahedron Lett.* **1983**, *24*, 4927–4930. (b) Cordier, M.; Archambeau, A. (3 + 3) Cycloaddition of Oxyallyl Cations with Nitrones: Diastereoselective Access to 1,2-Oxazinanes. *Org. Lett.* **2018**, *20*, 2265–2268. (c) An, D.; Guan, X.; Guan, R.; Jin, L.; Zhang, G. Organocatalyzed nucleophilic addition of pyrazoles to 2*H*-azirines: asymmetric synthesis of 3,3disubstituted aziridines and kinetic resolution of racemic 2*H*-azirines. *Chem. Commun.* **2016**, *52*, 11211–11214. (d) Verstappen, M. M. H.; Ariaans, G. J. A.; Zwanenburg, B. Asymmetric Synthesis of 2*H*-Azirine Carboxylic Esters by an Alkaloid-Mediated Neber Reaction. *J. Am. Chem. Soc.* **1996**, *118*, 8491–8492.