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Synthesis and crystal structural characterization of two new 3,5-disubstituted 4-amino-1,2,4triazoles

Abstract: Two new asymmetrical 3-(*p*-R-phenyl)-4-amino-5-(2-pyridyl)-1,2,4-triazoles (**3a**, $R = CH_3$ and **3b**, $R = OCH_3$) were synthesized with a yield of 58% and 69%, respectively. The compounds **3a** and **3b** were characterized with FT-IR, ¹H NMR, ESI-MS spectra, and elemental analysis. Additionally, their molecular and crystal structures were determined by single crystal X-ray analysis.

Keywords: 4-amino-1,2,4-triazole; crystal structure; hydrazine; synthesis.

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Introduction

During the past five decades, substituted 1,2,4-triazoles have aroused wide interest because of their anti-inflammatory, antitumor, fungicide, and herbicide properties [1]. Substituted 1,2,4-triazoles and their derivatives are also useful ligands in coordination chemistry due to their rich and versatile coordination modes [2–4]. In addition, some iron(II) complexes containing substituted 1,2,4-triazoles possess fascinating spin crossover properties which can be applied as molecular-based electronics, displays, and switching materials [5–10].

Recently, a series of symmetrical and asymmetrical 3,5-disubstituted 1,2,4-triazoles have been successfully prepared by our group [11–18] and others [3, 8]. However,

the asymmetrical 3,5-disubstituted 4-amino-1,2,4-triazoles remain largely unexplored. As a continuation of our investigation on substituted 4-amino-1,2,4-triazoles [19, 20], in this article, we present synthesis of two new asymmetrical 3,5-disubstituted 4-amino-1,2,4-triazoles, namely 3-(p-Rphenyl)-4-amino-5-(2-pyridyl)-1,2,4-triazoles **3a** (R = CH₃) and **3b** (R = OCH₃) (Scheme 1). Compounds **3a** and **3b** were characterized by FT-IR, ¹H NMR, ESI-MS spectra, elemental analysis, and single crystal X-ray diffraction. For the purpose of comparison, a known homologous compound, **3c** (R = Cl), is also discussed [21].

Results and discussion

Synthesis of 3,4,5-trisubstituted 1,2,4-triazoles has recently been reviewed [22]. In general, there are three types of methods reported for the preparation of 3,4,5-trisubstituted 1,2,4-triazoles [23]. One of these methods was chosen for the synthesis of our target compounds – 3,5-disubstituted 4-amino-1,2,4-triazoles **3a**–**c**. Compounds **3a**,**b** are new and synthesis of **3c** has been previously published [21].

N-(*p*-R-phenylcarbonyl)-*N*'-(2-pyridylcarbonyl)hydrazines **1a**–**c** were obtained in a yield of 81%, 79%, and 82%, respectively, by stirring pyridine-2-carbonylhydrazine with the corresponding benzoyl chloride in anhydrous pyridine at ambient temperature [24]. Then, 1,4-dichloro-1-(2-pyridyl)-4-(*p*-R-phenyl)-2,3-diaza-1,3-butadienes **2a**–**c** were obtained by stirring **1** with thionyl chloride in dried toluene with the respective yield of 68%, 69%, and 72% [25]. Finally, the cyclization reaction of **2a,b** with anhydrous hydrazine conducted at 120°C for 12 h produced **3a,b** with a yield of 58% and 69%, respectively. Synthesis of the known compound **3c** in a 47% yield has previously been reported [21].

New compounds **3a,b** were characterized by FT-IR, ¹H NMR, ESI-MS, and elemental analysis. The molecular structures of **3a,b** were also confirmed by X-ray crystallography (Table 1). Single crystals of **3a** and **3b** suitable

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Table 2Selected bond lengths (Å) and angles (°) for 3a,b.

3a		3b	
N1-C5	1.338(2)	N1-C5	1.332(4)
N2-N3	1.372(2)	N2-N3	1.365(4)
N4-N5	1.419(2)	N4-N5	1.407(4)
C11-C14	1.508(3)	01-C11	1.361(4)
N1-C1	1.338(2)	N1-C1	1.324(4)
C6-N2-N3	108.2(1)	C6-N2-N3	107.9(2)
C6-C5-N1	118.6(2)	C6-C5-N1	118.8(2)
C7-N4-C6	106.2(1)	C7-N4-C6	105.5(3)
N5-N4-C7	124.8(2)	N5-N4-C7	124.3(2)
C10-C11-C14	121.7(2)	C11-O1-C14	118.4(3)

Scheme 1 Synthesis of 3,5-disubstituted 4-amino-1,2,4-triazoles 3a-c.

for X-ray diffraction study were obtained upon slow concentration of ethanol solution at ambient temperature. The crystal structure of **3c** has been previously published

Table 1 Crystallographic data for 3a-c.

Compound	3a	3b	3c [21]	
Empirical formula	C ₁₄ H ₁₃ N ₅	C ₁₄ H ₁₃ N ₅ O	C ₁₃ H ₁₀ N ₅ Cl	
Formula weight	251.29	267.29	271.71	
Crystal system	Triclinic	Triclinic	Monoclinic	
Space group	PI	PI	C2/c	
a (Å)	6.2709(10)	6.183(10)	28.68(7)	
b (Å)	7.0252(12)	7.245(12)	6.041(16)	
<i>c</i> (Å)	14.572(2)	14.77(2)	14.68(4)	
α (°)	86.277(2)	76.816(17)	90.00	
β (°)	84.922(2)	85.501(18)	103.90(3)	
γ (°)	72.728(2)	73.507(18)	90.00	
V (Å ³)	610.09(17)	617.7(17)	2470(11)	
Ζ	2	2	8	
$D_{c}(\mathbf{g}\cdot\mathbf{cm}^{-3})$	1.368	1.437	1.462	
μ (mm ⁻¹)	0.088	0.097	0.302	
F(000)	264	280	1120	
Crystal size (mm)	$\textbf{0.24} \times \textbf{0.16} \times \textbf{0.10}$	$\textbf{0.14} \times \textbf{0.10} \times \textbf{0.06}$	0.16 imes 0.12 imes 0.10	
heta Range	1.40-25.00	1.42-25.00	1.46-25.00	
Reflections collected	4295	4277	8203	
Independent reflections	2125 $[R_{int} = 0.0174]$	2128 $[R_{int} = 0.0184]$	2175 [$R_{int} = 0.0757$]	
Reflections observed $[I>2\sigma(I)]$	1607	1817	1436	
Data/restraints/parameters	2125/3/180	2128/4/188	2175/3/178	
Goodness-of-fit on F ²	1.067	1.095	1.071	
<i>R/w</i> R [<i>I</i> >2 <i>σ</i> (<i>I</i>)]	0.0449/0.1234	0.0594/0.1852	0.0682/0.1834	
<i>R/w</i> R (all data)	0.0624/0.1324	0.0675/0.1897	0.1027/0.1985	
Max., Min. $\Delta\! ho$ (e·Å·³)	0.227, -0.221	0.397, -0.377	0.801, -0.280	

Note: data of 3c are given for comparison.



Figure 1 Views of 3a and 3b showing the atom-numbering scheme. Hydrogen atoms are omitted for clarity.

Table 3	The interplana	r angles ((°) for 3a,b .
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Compound	Py/Trz	CH ₃ -Ph/Trz	CH ₃ O-Ph/Trz
3a	2.0(1)	30.2(1)	
3b	7.1(1)		30.9(1)

[21]. The crystal structures of **3a,b** with an atom-labeling scheme are shown in Figure 1, and selected bond lengths and bond angles are listed in Table 2. The X-ray crystal-lography analysis indicates that both **3a** and **3b** crystallize in the triclinic space group $P\overline{1}$, which is different from the

Table 4 The hydrogen bonds, C-H··· π and π ··· π interactions (Å, °) in **3a,b**.

Compound	D-H···A	d(D-H)/Å	d(H···A)/Å	d(DA)/Å	∠D-H…A/°
3a	C1 ^v -H1A ^v ····N5	0.93	2.70	3.420(2)	135
	C4-H4A····N2 ⁱ	0.93	2.82	3.394(2)	121
	N5-H5A···N1	0.89	2.15	2.908(2)	142
	N5-H5B····N2"	0.88	2.41	3.106(2)	137
	С14-H14С… <i>л</i> (С8-С13) ^{ііі}	0.96	2.98	3.656(2)	128
	$\pi \cdots \pi$ interaction		cent…cent (Å)		Dihedral angle (°)
	π (Py) $\cdots \pi$ (Trz) ^{iv}		3.687(2)		1.9
3b	C1 ^{viii} -H1A ^{viii} ····N5	0.93	2.76	3.378(6)	124
	C2 ^{vi} -H2A ^{vi} ····O1	0.93	2.62	3.533(6)	168
	N5-H5B…N1	0.85	2.27	2.902(5)	131
	C14-H14A…π (C8-C13) ^{νii}	0.96	3.29	3.927(5)	126

Symmetry codes: (i) -x, 1-y, 2-z; (ii) 1+x, y, z; (iii) 2-x, -y, 1-z; (iv) 1-x, -y, 2-z; (v) 2-x, -y, 2-z; (vi) x, y-1, 1+z; (vii) 2-x, -1-y, 2-z; (viii) 2-x, -y, 1-z.



Figure 2 View of the crystal packing for **3a** showing hydrogen bonds, C-H $\cdots \pi$ and $\pi \cdots \pi$ interactions.



Figure 3 View of the crystal packing for **3b** showing hydrogen bonds and C-H \cdots π interactions.

Brought to you by | New York University Bobst Library Technical Services Authenticated Download Date | 5/26/15 4:00 PM homologous compound 3c(C2/c). The central 1,2,4-triazole in **3a** is oriented at interplanar angles of 2.0(1)° and 30.2(1)° with respect to the pyridyl ring and the *p*-methylphenyl ring, respectively. Similarly, the central 1,2,4-triazole ring in **3b** is oriented at interplanar angles of $7.1(1)^{\circ}$ and $30.9(1)^{\circ}$ with respect to the pyridyl ring and the *p*-methoxyphenyl ring, respectively (Table 3). In addition, the N atom of the pyridyl group is oriented towards the amino group on the N4 position in both **3a** and **3b**. This feature is similar to that of the related substituted 4-amino-1,2,4-triazoles compounds [19, 20]. The bond lengths and bond angles of 3a and **3b** are similar and in normal ranges [23, 26, 27]. The hydrogen bonds, a C-H··· π interaction and a non-bonding $\pi \cdots \pi$ interaction in these compounds are given in Table 4. These interactions stabilize the crystal packing of **3a** and 3b. The crystal packing diagrams of 3a and 3b are shown in Figures 2 and 3, respectively.

Experimental

Melting points were determined on an X-4 digital microscope melting-point apparatus (Beijing) and are uncorrected. FT-IR spectra were recorded on a Nicolet 380 FT-IR instrument using KBr disks. ¹H NMR spectra were measured on a Bruker AM 500 MHz spectrometer at ambient temperature in DMSO- d_6 for **1** and **2** but in CDCl₃ for **3**. Electrospray ionization mass spectra (ESI-MS) were recorded on an LCQ ADVANTAGE MAX mass spectrometer. The spray voltage, the capillary voltage, and the capillary temperature were 4 kV, 40 V, and 260°C, respectively. Elemental analyses were carried out with a Thermo Finnigan Flash 1112A elemental analyzer. All chemicals were of analytical grade and used as received. Toluene was freshly distilled from Na/benzophenone, whereas pyridine was distilled over NaOH. Pyridine-2-carbonylhydrazine was prepared according to a previously reported method [23].

Preparation of *N*-(*p*-R-phenylcarbonyl)-*N*'-(2pyridylcarbonyl)hydrazines 1a-c

A suspension of the corresponding *p*-substituted benzoyl chloride (48 mmol) in anhydrous pyridine (10 mL) was added dropwise to a stirred solution of pyridine-2-carbonylhydrazine (5.49 g, 40 mmol) in dichloromethane (80 mL) at 0°C. The mixture was stirred for 8 h at ambient temperature and then poured into 80 mL of ice water. The resultant precipitate was filtered, dried *in vacuo*, and crystallized from anhydrous ethanol to give **1a**–**c** as a white solid.

N-(*p*-Methylphenylcarbonyl)-*N*'-(2-pyridylcarbonyl)hydrazine (1a) Yield 81%; mp 182–184°C; IR: 3189, 3010, 2922, 1674, 1633, 1607, 1500, 1280, 1114, 836, 743 cm⁻¹; ¹H NMR: δ 2.44 (3H, s, CH₃), 7.13 (2H, d, J = 8 Hz, ArH), 7.54 (1H, dd, $J_1 = 5$ Hz, $J_2 = 2$ Hz, PyH), 8.01–8.05 (3H, m, PyH, ArH), 8.20 (1H, d, J = 8 Hz, PyH), 8.63 (1H, d, J = 5 Hz, PyH), 10.48 (1H, s, NH), 10.51 (1H, s, NH). Anal. Calcd for C₁₄H₁₃N₃O₂: C, 65.87; H, 5.13; N, 16.46. Found: C, 65.73; H, 5.31; N, 16.67. *N*-(*p*-Methoxyphenylcarbonyl)-*N*'-(2-pyridylcarbonyl)hydrazine (1b) Yield 79%; mp 154–156°C (lit. mp 153–155°C [22]); IR: 3285, 3021, 2964, 1703, 1642, 1610, 1566, 1508, 1488, 1303, 1260, 1026, 998 cm⁻¹; ¹H NMR: δ 3.88 (3H, s, CH₃), 7.19 (2H, d, J = 8 Hz, ArH), 7.65 (1H, dd, $J_1 = 5$ Hz, $J_2 = 2$ Hz, PyH), 8.05–8.09 (3H, m, PyH, ArH), 8.25 (1H, d, J = 8 Hz, PyH), 8.80 (1H, d, J = 5 Hz, PyH), 10.51 (1H, s, NH), 10.53 (1H, s, NH). Anal. Calcd for C₁₄H₁₃N₃O₃: C, 61.99; H, 4.83; N, 15.49. Found: C, 61.82; H, 4.71; N, 15.77.

N-(*p*-Chlorophenylcarbonyl)-*N*'-(2-pyridylcarbonyl)hydrazine (1c) Yield 82%; mp 179–181°C; IR: 3177, 3007, 1688, 1642, 1597, 1570, 1513, 1488, 1334, 1092, 1016, 844 cm⁻¹; ¹H NMR: δ 7.62 (2H, d, *J* = 8 Hz, ArH), 7.68 (1H, dd, *J*₁ = 5 Hz, *J*₂ = 2 Hz, PyH), 7.94 (2H, d, *J* = 8 Hz, ArH), 8.05 (2H, m, PyH), 8.72 (1H, d, *J* = 5 Hz, PyH), 10.63 (1H, s, NH), 10.67 (1H, s, NH). Anal. Calcd for C₁₃H₁₀N₃O₂Cl: C, 56.64; H, 3.66; N, 15.24. Found: C, 56.42; H, 3.54; N, 15.39.

Preparation of 1,4-dichloro-1-(2-pyridyl)-4-(p-R-phenyl)-2,3-diaza-1,3-butadienes 2a-c

A solution of thionyl chloride (5 mL) in anhydrous toluene (10 mL) was added dropwise to a stirred solution of **1** (16 mmol) in anhydrous toluene (50 mL). The mixture was heated under reflux for 8 h at 120°C and then the solvent was removed under reduced pressure. The residue was crystallized from ethanol to give **2** as colorless crystals.

1,4-Dichloro-1-(2-pyridyl)-4-(*p***-methylphenyl)-2,3-diaza-1,3-butadiene (2a)** Yield 68%; mp 156–158°C; IR: 3054, 2993, 1613, 1588, 1556, 1494, 1458, 1443, 1247, 1183, 1079, 824, 743, 728 cm⁻¹; ¹H NMR: δ 2.45 (3H, s, CH₃), 7.32–7.34 (2H, d, *J* = 8 Hz, ArH), 7.47 (1H, dd, *J*₁ = 5 Hz, *J*₂ = 2 Hz, PyH), 7.85 (1H, m, PyH), 8.09 (2H, d, *J* = 7 Hz, ArH), 8.32 (1H, d, *J* = 8 Hz, PyH), 8.72 (1H, d, *J* = 5 Hz, PyH). Anal. Calcd for C₁₄H₁₁N₃Cl₂: C, 57.55; H, 3.79; N, 14.38. Found: C, 57.31; H, 3.43; N, 14.49.

1,4-Dichloro-1-(2-pyridyl)-4-(*p***-methoxyphenyl)-2,3-diaza-1,3-butadiene (2b)** Yield 69%; mp 152–154°C; IR: 3052, 2978, 1615, 1585, 1498, 1458, 1443, 1268, 1176, 1085, 1019, 832, 790, 741 cm⁻¹; 'H NMR: δ 3.85 (3H, s, CH₃), 7.02 (2H, d, *J* = 8 Hz, ArH), 7.45 (1H, dd, *J*₁ = 5 Hz, *J*₂ = 2 Hz, PyH), 7.83 (1H, m, PyH), 8.14 (2H, d, *J* = 8 Hz, ArH), 8.31 (1H, d, *J* = 8 Hz, PyH), 8.61 (1H, d, *J* = 5 Hz, PyH). Anal. Calcd for C₁₄H₁₁N₃Cl₂O: C, 54.57; H, 3.60; N, 13.64. Found: C, 54.36; H, 3.45; N, 13.41.

1,4-Dichloro-1-(2-pyridyl)-4-(*p***-chlorophenyl)-2,3-diaza-1,3-butadiene (2c)** Yield 72%; mp 146–148°C; IR: 1605, 1586, 1553, 1479, 1463, 1442, 1275, 1151, 1090, 835, 794, 731 cm⁻¹; ¹H NMR: δ 7.45 (1H, dd, $J_1 = 5$ Hz, $J_2 = 2$ Hz, PyH), 7.50 (2H, d, J = 8 Hz, ArH), 7.92 (1H, m, PyH), 8.16 (2H, d, J = 8 Hz, ArH), 8.32 (1H, d, J = 8 Hz, PyH), 8.82 (1H, d, J = 5 Hz, PyH). Anal. Calcd for C₁₃H₈N₃Cl₃: C, 49.95; H, 2.58; N, 13.44. Found: C, 49.66; H, 2.32; N, 13.68.

Preparation of 3-(*p*-R-phenyl)-4-amino-5-(2pyridyl)-1,2,4-triazoles 3a-c

A mixture of **2** (2 mmol) and anhydrous hydrazine (2 mL) was sealed in a 25 mL Teflon-lined stainless steel reactor and heated for 12 h at 120°C and then cooled to ambient temperature. The precipitate was filtered, washed with deionized water, and crystallized from anhydrous ethanol to give **3** as colorless crystals.

3-(*p*-**Methylphenyl**)-**4-**amino-**5-**(**2**-**pyridyl**)-**1**,**2**,**4**-**triazole** (**3a**) Yield 58%; mp 205–207°C; IR: 3345, 3272, 3193, 1640, 1591, 1472, 1422, 1284, 1151, 1034, 993, 961, 818, 786, 728, 696, 583 cm⁻¹; ¹H NMR: δ 2.42 (3H, s, CH₃), 6.45 (2H, br, NH₂), 7.30 (2H, d, *J* = 9 Hz, ArH), 7.36 (1H, dd, *J*₁ = 5 Hz, *J*₂ = 2 Hz, PyH), 7.85 (1H, m, PyH), 8.08 (2H, d, *J* = 9 Hz, ArH), 8.36 (1H, d, *J* = 7 Hz, PyH), 8.62 (1H, d, *J* = 5 Hz, PyH); ESI-MS: *m*/*z* 252.25 (M+H)⁺. Anal. Calcd for C₁₄H₁₃N₅: C, 66.92; H, 5.21; N, 27.87. Found: C, 66.76; H, 5.05; N, 27.99.

3-(*p*-Methoxyphenyl)-4-amino-5-(2-pyridyl)-1,2,4-triazole (**3b**) Yield 69%; mp 194–195°C; IR: 3342, 3278, 3157, 2993, 2833, 1611, 1582, 1488, 1441, 1247, 1188, 1035, 972, 840, 737, 696, 594 cm⁻¹; ¹H NMR: δ 3.86 (3H, s, CH₃), 6.44 (2H, br, NH₂), 6.99 (2H, d, *J* = 9 Hz, ArH), 7.35 (1H, dd, *J*₁ = 5 Hz, *J*₂ = 2 Hz, PyH), 7.84 (1H, m, PyH), 8.14 (2H, d, *J* = 9 Hz, ArH), 8.32 (1H, d, *J* = 7 Hz, PyH), 8.61 (1H, d, *J* = 5 Hz, PyH); ESI-MS: *m/z* 268.3 (M+H)⁺. Anal. Calcd for C₁₄H₁₃N₅O: C, 62.91; H, 4.90; N, 26.20. Found: C, 62.74; H, 4.78; N, 26.31.

3-(p-Chlorophenyl)-4-amino-5-(2-pyridyl)-1,2,4-triazole (**3c**) See reference [21].

Single-crystal X-ray diffraction analysis of 3a and 3b The single crystals of **3a** and **3b** were selected for lattice parameter determination and collection of intensity data at 296 K on a Bruker SMART CCD diffractometer with a detector distance of 5 cm and frame exposure time of 10 s using a graphite-monochromated Mo- K_{α} ($\lambda = 0.71073$ Å) radiation. The structures were solved by direct methods and refined on F^2 by full-matrix least squares procedures using SHELXTL software [28]. All non-hydrogen atoms were anisotropically refined. The H atoms were placed on calculated positions (C-H 0.96 Å) and assigned isotropic thermal parameters riding on their parent atoms; the H atoms of $-NH_2$ were located from difference Fourier maps and refined isotropically. Details on crystal data of **3a** and **3b** are summarized in Table 1.

CCDC 983076 (**3a**) and 983077 (**3b**) contain the supplementary crystallographic data for this article. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www. ccdc.cam.ac.uk/data_request/cif.

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References

- Potts, K. T. The chemistry of 1,2,4-triazoles. Chem. Rev. 1961, 61, 87–127.
- [2] Haasnoot, J. G. Mononuclear, oligonuclear and polynuclear metal coordination compounds with 1,2,4-triazole derivatives as ligands. *Coord. Chem. Rev.* 2000, 200–202, 131–185.
- [3] Klingele, M. H.; Brooker, S. The coordination chemistry of 4-substituted 3,5-di(2-pyridyl) -4H-1,2,4-triazoles and related ligands. *Coord. Chem. Rev.* 2003, 241, 119–132.

- Beckmann, U.; Brooker, S. Cobalt(II) complexes of pyridazine or triazole containing ligands: spin-state control. *Coord. Chem. Rev.* 2003, 245, 17–29.
- [5] Kahn, O.; Martinez, C. J. Spin-transition polymers: from molecular materials toward memory devices. *Science* **1998**, *279*, 44–48.
- [6] Zhu, D.; Xu, Y.; Yu, Z.; Guo, Z.; Sang, H.; Liu, T.; You, X. A novel bis(*trans*-thiocyanate)iron(II) spin-transition molecular material with bidentate triaryltriazole ligands and its bis(*cis*thiocyanate)iron(II) high-spin isomer. *Chem. Mater.* 2002, 14, 838–843.
- [7] van Koningsbruggen, P. J. Special classes of iron(II) azole spin crossover compounds. *Top. Curr. Chem.* 2004, 233, 123–149.
- [8] Kitchen, J. A.; Brooker, S. Spin crossover in iron(II) complexes of 3,5-di(2-pyridyl)-1,2,4-triazoles and 3,5-di(2-pyridyl)-1,2,4triazolates. *Coord. Chem. Rev.* 2008, 252, 2072–2092.
- [9] Malcolm, A. H. Structure: function relationships in molecular spin-crossover complexes. *Chem. Soc. Rev.* 2011, 40, 4119– 4142.
- [10] Shen, G. P.; Qi, L.; Wang, L.; Xu, Y.; Jiang, J. J.; Zhu, D.; Liu, X. Q.; You, X. Spin-crossover in a *trans*-[FeL₂(NCS)₂] family (L = triaryltriazole): remote substituent effects on spin transition modes and temperature. *J. Chem. Soc. Dalton. Trans.* 2013, 42, 10144–10152.
- [11] Zhu, D. R.; Song, Y.; Xu, Y.; Zhang, Y.; Raj, S. S. S.; Fun, H. K.; You, X. Z. Syntheses, crystal structures and properties of the novel Co(II) and Ni(II) complexes with 4-(*p*-methylphenyl)-3,5bis(pyridin-2-yl)-1,2,4-triazole. *Polyhedron* 2000, *19*, 2019– 2025.
- [12] Zhu, D. R.; Xu, Y.; Zhang, Y.; Wang, T. W.; You, X. Z. 4-Phenyl-3,5-bis(2-pyridyl)-4H-1,2,4-triazole. *Acta Crystallogr.* 2000, *C56*, 895–896.
- [13] Zhu, D.; Xu, Y.; Mei, Y.; Shi, Y.; Tu, C.; You, X. Crystal and molecular structure of a novel complex [Mn(MOBPT)(H₂O)₂(NCS)₂] (MOBPT = 4-(*p*-methoxyphenyl)-3,5-bis(pyridin-2-yl)-1,2,4-triazole). *J. Mol. Struct.* 2001, *559*, 119–125.
- [14] Zhou, J.; Yang, J.; Qi, L.; Shen, X.; Zhu, D.; Xu, Y.; Song, Y. Synthesis, crystal structure and magnetic property of the novel dinuclear nickel(II) complex with 4-(*p*-methoxyphenyl)-3,5bis(pyridine-2-yl)-1,2,4-triazole. *Transition Met. Chem.* 2007, 32, 711–715.
- [15] Yang, J.; Bao, W. W.; Ren, X. M.; Xu, Y.; Shen, X.; Zhu, D. R. Synthesis, crystal structure, and magnetism of a dinuclear nickel(II) complex [Ni₂(MOBPT)₂(N₃)₄]·2H₂O. *J. Coord. Chem.* 2009, *62*, 1809–1816.
- [16] Chen, L.; Cheng, H. M.; Jiang, J. J.; Shen, X.; Zhu, D. R. Synthesis, crystal structure and thermal stability of [CuL₂(ClO₄)₂] (L = 3-(*p*-bromophenyl)-4-(*p*-methylphenyl)-5-(2-pyridyl)-1,2,4-triazole). *Chinese J. Inorg. Chem.* 2012, *28*, 381–385.
- [17] Lu, W.; Zhu, D. R.; Xu, Y.; Cheng, H. M.; Zhao, J.; Shen, X. Syntheses and crystal structures of two novel Cu(II) and Co(II) complexes with 3-methyl-4-(p-bromophenyl) -5-(2-pyridyl)-1,2,4-triazole. *Struct. Chem.* 2010, *21*, 237–244.
- [18] Zhao, J.; Cheng, H. M.; Shen, G. P.; Xu, Y.; Zhu, D. R. Syntheses, crystal structures, and spectral properties of Mn(II) and Co(II) complexes with 3-(p-chlorophenyl)-4-(p-methylphenyl)-5-(2pyridyl)-1,2,4-triazole. J. Coord. Chem. 2011, 64, 942–951.
- [19] Shen, G. P.; Jiang, J. J.; Sun, F.; Shen, X.; Zhu, D. R.; Liu, X. Q. Syntheses, crystal structures, and spectral characterization of

two novel quinolyl substituted triazoles. J. Heterocycl. Chem. 2013, 50, 1152–1156.

- [20] Gong, J.; He, X.; Chen, L.; Shen, X.; Zhu, D. Syntheses, crystal structures, and spectral characterization of an asymmetrically substituted 1,2,4-triazole and its iron(II) complex. J. Coord. Chem. 2013, 66, 2875–2884.
- [21] Wang, L.; Jiang, J. J.; Chen, L.; Shen, X.; Zhu, D. R. Coexistence of two distinct axial ligands and their thermo-induced substitution in *trans*-[FeL₂(NCS)₂][FeL₂(CH₃OH)₂](NCS)₂ (L = 4-amino-3-(*p*-chlorophenyl)-5-(2-pyridy1)-1,2,4-triazole). *Inorg. Chem. Commun.* 2013, *28*, 104–108.
- [22] Moulin, A.; Bibian, M.; Blayo, A. L.; Habnouni, S. E.; Martinez, J.; Fehrentz, J. A. Synthesis of 3,4,5-trisubstituted-1,2,4-triazoles. *Chem. Rev.* 2010, *110*, 1809–1827.
- [23] Zhao, J.; Shen, G. P.; Zhang, Y.; Shen X.; Zhu, D. R. Syntheses, crystal structures, and spectral characterization of three new asymmetrical substituted triaryltriazoles. *J. Heterocycl. Chem.* 2012, 49, 1114–1119.

- Hua, G.; Li, Y.; Fuller, A. L.; Slawin, A. M. Z.; Woollins, J. D. Facile synthesis and structure of novel 2,5-disubstituted 1,3,4-selenadiazoles. *Eur. J. Org. Chem.* 2009, 10, 1612–1618.
- [25] Gautun, O. R.; Carisen, P. H. Synthesis of bis(αalkylaminobenzylidene)hydrazines and their transformation into 4-alky-4H-1,2,4-triazoles. *Acta Chem. Scand.* **1991**, 45, 609–615.
- [26] Jiang, J. J.; Shen, G. P.; He, X.; Shen, X.; Zhu, D. R. One-pot synthesis of a 3,4,5-triaryltriazole from the substituted 1,3,4-oxadiazole: crystal structures characterization. *J. Heterocycl. Chem.*, 2014, http://dx.doi.org/10.1002/jhet.1894.
- [27] Cheng, H. M.; Zhu, D. R.; Lu, W.; Xu, R. H.; Shen, X. Synthesis and crystal structure characterization of 3,5-bis(2quinolyl)-1,2,4-triazole. J. Heterocycl. Chem. 2010, 47, 210–213.
- [28] Sheldrick, G. M. A short history of SHELX. *Acta Crystallogr*. 2008, *A64*, 112–122.