Mild Conditions for Copper-Catalyzed N-Arylation of Imidazoles

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Abstract: An efficient copper(I) bromide catalyzed N-arylation of azoles with a variety of aromatic bromides and iodides under mild conditions is reported. This reaction displayed great functional group compatibility and excellent reactive selectivity.

Key words: copper, aryl halides, imidazoles, azoles, N-arylation

N-Arylimidazoles represent an attractive class of chemicals that are widely used in the preparation of agrochemicals, pharmaceuticals, biological active compounds, and N-heterocyclic carbene chemistry.^{1,2} Classical synthetic methods for these compounds included nucleophilic aromatic substitution of active aryl halides with imidazole and Ullmann-type coupling reactions etc.³ However, these methods suffer from several drawbacks such as harsh reaction conditions (>200 °C) and/or the use of stoichiometric amounts of copper.⁴ Much progress has been achieved over the last few decades in copper-catalyzed C-N bond formation.³ Employing different mono- and bidentate chelators as effective ancillary ligands, N-arylimidazoles were successfully produced under lower temperatures (100-150 °C) in 24-40 hours.⁵ Most recently, a ligandfree N-arylation of imidazoles from aryl iodides was also reported at 35-40 °C in 40 hours.⁶ Nevertheless, a new strategy for the assembly of N-arylimidazoles under mild conditions, especially at lower temperatures and with shorter reaction times, remains highly desirable.

Recently, we identified pyridin-2-yl β -ketones as new efficient supporting ligands for copper-catalyzed coupling reactions of aliphatic amines, ammonia, and phenols with aryl halides under mild conditions.⁷ Herein we report the efficient N-arylation of azoles with a variety of aromatic bromides and iodides under the catalysis of a combination of copper(I) bromide/pyridin-2-yl β -ketone.

Initially, we chose the coupling of iodobenzene (1a) with 1*H*-imidazole (2a) as a model to optimize the reaction conditions. As shown in Table 1, the yield of 3a was poor, less than 10%, when the reaction was carried out in the ab-

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Table 1 Optimization of the Reaction Conditions for the N-Arylation of 1H-Imidazole (2a) with Iodobenzene (1a)^a



^a Reaction conditions: PhI (**1a**, 1.0 mmol), 1*H*-imidazole (**2a**, 1.2 mmol), [Cu] (5 mmol%), L**1** (10 mmol%) (unless otherwise noted), base (2.0 mmol), solvent (1.0 mL), 60 °C, N_2 , 5 h.

CuBr

CuBr

85

8

DMSO

DMSO

^b Without ligand.

14

15

^c L1 (5 mmol%) was used. d L2 (10 mmol%) was used.

^d L2 (10 mmol%) was used.

NaOH

Et₃N

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sence of ligand or copper at 60 °C (Table 1, entries 1 and 2). However, when 5 mmol% copper(I) bromide and 5 mmol% 1-(5,6,7,8-tetrahydroquinolin-8-yl)ethanone (L1) were employed, **3a** was produced in moderate yield (Table 1, entry 3). The isolated yield of **3a** improved significantly (~95%) when 10 mmol% of supporting ligand L1 was utilized. The results also revealed that the simplest and commercially available pyridin-2-yl β -ketone ana-

logue, 1-pyridin-2-ylpropan-2-one (**L2**), also successfully promoted the coupling reaction with a slightly lower yield (Table 1, entry 5). Other conditions, such as copper sources, solvents, and bases, were also screened. The results indicated that optimal combination of 5 mmol% copper(I) bromide, 10 mmol% **L1**, and 2.0 equivalents of cesium carbonate in dimethyl sulfoxide (1.0 mL).

ArX + N H 1 2a	$\begin{array}{c} \text{CuBr/L1} \\ \hline \text{Cs}_2\text{CO}_3, \text{DMSO} \\ \hline 60 \text{ °C}, 5 \text{ h} \end{array} \qquad \text{Ar} \xrightarrow{N} N$			
Entry	Ar-X	Product		Yield (%)
1		3b		83
2	OMe I	3c	OMe N	78
3	Br	3d	Br-NNN	91
4		3e		89
5	Br	3a		86
6	Br	3f		82
7	MeO Br	3g	Meo	80
8	MeO-Br	3h	MeO	82
9	F O Br	3 i	F C C N	86
10	CI-Br	3j		95
11	H ₂ N F ₂ C	3k		86 ^b
12	o Br	31		92
13	NCBr	3m		88

 Table 2
 Copper(I) Bromide/L1-Catalyzed N-Arylation of 1H-Imidazole (2a) with Aryl and Heteroaryl Halides 1^a

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ArX + $\begin{pmatrix} N \\ N \\ H \\ H \end{pmatrix}$	$\begin{array}{c} CuBr/L1 \\ \hline Cs_2CO_3, DMSO \\ \hline 60 \ ^\circC, 5 \ h \end{array} \qquad \qquad$			
Entry	Ar-X	Product		Yield (%)
14	HO-Br	3n		72 ^b
15	H ₂ N-Br	30		81 ^b
16	HO	3р	HO	80 ^b
17	Br	3q		92
18	CI CI	3a		0
19	N N Br	3r		78
20	CIBr	3s		76
21	Ser Br	3t		77
22	MeO-	3u		80
23	Br	3v		73
24	Br — K Br	3w		75°

 Table 2
 Copper(I) Bromide/L1-Catalyzed N-Arylation of 1H-Imidazole (2a) with Aryl and Heteroaryl Halides 1^a (continued)

 $\overline{}$

^a Reaction conditions: Ar-X **1** (1.0 mmol), 1*H*-imidazole (**2a**, 1.2 mmol), CuBr (5 mmol%), L**1** (10 mmol%), Cs₂CO₃ (2.0 mmol), DMSO (1.0 mL), 60 °C, N₂, 5 h.

^b 120 °C, 12 h.

^c Performed with 2.0 mmol of imidazole.

The scope of the copper(I) bromide/L1-catalyzed N-arylation of 1*H*-imidazole (2a) was explored by using a variety of aryl iodides and aryl bromides 1 under the optimized conditions. As summarized in Table 2, the results showed that all the tested aryl iodides coupled with 1*H*-imidazole (2a) to afford the desired products 3b–e with satisfactory yields (entries 1–4). For aryl bromides, the catalyst system efficiently promoted the coupling reactions in most cases. The results also indicated that electron-withdrawing groups in the aryl bromide substrates benefited the coupling reaction to give the products 3j,l,m in excellent yields (Table 2, entries 10, 12, and 13). The coupling reactions were also successfully performed using electron-rich substrates, such as 3-bromotoluene, 3bromoanisole, and 4-bromoanisole, giving **3f-h** in good yields (Table 2, entries 6–8). Furthermore, the copper(I) bromide/**L1** system was well tolerated by multiple functional groups such as ketones, nitriles, free amines, and free hydroxy groups (Table 2, entries 11–16). Although this protocol did not work well for electron-rich aryl chlorides (i.e., chlorobenzene, Table 2, entry 18) even if the reactions were carried out at high temperatures (120 °C) with longer reaction time, the catalyst system significantly improved the reaction efficiency of electron-deficient aryl chlorides with 1*H*-imidazole (**2a**). For example, coupling of 1-chloro-4-nitrobenzene with 1*H*-imidazole (**2a**) at 100 °C in the absence of **L1** gave 1-(4-nitrophenyl)-1*H*-imidazole in only 41% yield, but addition of 10% **L1** at 100 °C gave the product in a dramatically improved 81% yield (results not shown). Utilizing 4-chlorobenzonitrile as the substrate with 1*H*-imidazole (**2a**) gave **3m** in 75% isolated yield (results not shown).



^a Reaction conditions: Ar-X **1** (1.0 mmol), Het-NH **2** (1.2 mmol), CuBr (5 mmol%), L**1** (10 mmol%), Cs₂CO₃ (2.0 mmol), DMSO (1.0 mL), 80 °C, N₂, 12 h. ^b 60 °C.

^d The regioselectivity (19:1) was determined by GC-MS analysis and the major pure product was isolated by recrystallization.

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The N-arylation of imidazole with various halopyridines and halopyrimidines was also performed with great success. Almost all the halopyridines and halopyrimidines coupled with 1*H*-imidazole (**2a**) to give **3r**–**w** in excellent or good yields under the optimized conditions (Table 2, entries 19–24). In addition, the double-coupling product **3w** was obtained when 2,5-dibromopyridine was reacted with 1*H*-imidazole (**2a**) (Table 2, entry 24).

To further expand the scope of this methodology, the catalytic system was applied to other nitrogen-containing heterocycles 2. We were pleased to find that most of azoles 2 coupled with aryl halides 1 to afford the corresponding products 4 in good yields (Table 3), but that slightly higher reaction temperatures (80 °C) and longer reaction times (12 h) were required. It should be noted that sterically hindered 2-methyl-1H-imidazole and 2-propyl-1H-imidazole gave the desired coupling products 4d,e,h in good yields (Table 3, entries 4, 5, and 8). Aryl bromides bearing a free amino group could successfully undergo selective N-arylation of 4-methyl-1H-imidazole and 4methyl-1H-pyrazole to assemble the corresponding products 4f,g in 81% and 88% yields, respectively (Table 3, entries 6 and 7). In addition, excellent N1 regioselectivity (19:1) for the coupling of 4-methyl-1*H*-imidazole was observed (Table 3, entry 6).

In summary, we have successfully developed an efficient copper(I) bromide/L1 catalyst system for the N-arylation of azoles with a variety of aromatic bromides and iodides. Employing this catalytic system, *N*-arylazoles can be synthesized under much milder conditions (60–80 °C, 5–12 h). Also, this new protocol displays great functional groups compatibility and excellent reactive selectivity. Our effort may provide an attractive addition to the copper-catalyzed synthesis of *N*-arylazoles.

All reactions were carried out in 5-mL sealed tubes, under a pure and dry argon atmosphere. Solvents, base and all other materials were used commercially without further purification. **L1**, **L2** were prepared as previously described.^{7a} All new compounds are indicated as such: **3i**, **3j**, **3k**, **3r**, **3s**, **3v**, **3w**, **4e**, **4g**, **4h**; for all known compounds, adequate characterization was obtained by ¹H NMR spectroscopy and mass spectrometry. NMR spectra were recorded at 20 °C on a 300 or 400 MHz spectrometer working respectively at 300 or 400 MHz for ¹H, at 75 or 100 MHz for ¹³C.

Arylazoles 3 and 4; General Procedure

A sealed tube equipped with a Teflon valve was charged with a magnetic stir bar, CuBr (7 mg, 0.05 mmol, 5 mol%), Cs_2CO_3 (650 mg, 2 mmol) and any remaining solids (azoles 2 and/or aryl halides 1). The tube was evacuated and backfilled with N_2 (3 ×). Under a counter flow of N_2 , azole 2 (1.2 mmol, if liquid), aryl halide 1 (1 mmol, if liquid), DMSO (1.0 mL), and L1 (18 mg, 0.10 mmol, 10 mol%) were added by syringe. The tube was evacuated and again backfilled with N_2 (3 ×) and sealed. The mixture was heated to the indicated temperature for the required time period. After cooling to r.t., the mixture was diluted with EtOAc (25 mL), passed through a fritted glass filter to remove the inorganic salts and the filtrate was washed by H₂O, brine, then removed under vacuum. The residue was purified by column chromatography (silica gel) and the product was dried under high vacuum (at least 1 h).

^{° 120 °}C.

1-Phenyl-1*H*-imidazole (3a)^{5e}

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.28$ (s, 1 H), 7.76 (t, 1 H), 7.67 (q, J = 5.8 Hz, 2 H), 7.53 (t, 2 H), 7.37 (t, 1 H), 7.14 (s, 1 H). MS: m/z = 144.

1-(4-Tolyl)-1H-imidazole (3b)^{5t}

¹H NMR (300 MHz, CDCl₃): δ = 8.19 (s, 1 H), 7.69 (s, 1 H), 7.52 (d, *J* = 6.8 Hz, 2 H), 7.31 (d, *J* = 7.2 Hz, 2 H), 7.09 (s, 1 H), 2.34 (s, 3 H).

MS: m/z = 158.

1-(2-Methoxyphenyl)-1H-imidazole (3c)⁶

¹H NMR (300 MHz, CDCl₃): δ = 7.89 (s, 1 H), 7.40 (t, *J* = 7.5 Hz, 3 H), 7.25 (d, *J* = 8.5 Hz, 1 H), 7.07 (t, *J* = 7.5 Hz, 2 H), 3.82 (s, 3 H).

MS: m/z = 174.

1-(4-Bromophenyl)-1*H*-imidazole (3d)^{5e}

¹H NMR (300 MHz, CDCl₃): δ = 8.28 (s, 1 H), 7.84 (s, 1 H), 7.72 (d, *J* = 9.0 Hz, 2 H), 7.63 (d, *J* = 9.0 Hz, 2 H), 7.11 (s, 1 H). MS: *m*/*z* = 223.

1-Biphenyl-4-yl-1*H*-imidazole (3e)⁵ⁱ

¹H NMR (300 MHz, CDCl₃): δ = 8.33 (s, 1 H), 7.81 (m, 3 H), 7.74 (m, 4 H), 7.49 (m, 2 H), 7.39 (t, *J* = 7.3 Hz, 1 H), 7.13 (s, 1 H). MS: *m*/*z* = 220.

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1-(3-Tolyl)-1*H*-imidazole (3f)^{5r}

¹H NMR (300 MHz, CDCl₃): δ = 8.23 (s, 1 H), 7.73 (s, 1 H), 7.49 (s, 1 H), 7.39 (t, 2 H), 7.18 (t, *J* = 7.0 Hz, 1 H), 7.10 (s, 1 H), 2.38 (s, 3 H).

MS: m/z = 158.

1-(3-Methoxyphenyl)-1*H*-imidazole (3g)^{5r}

¹H NMR (300 MHz, CDCl₃): δ = 8.29 (s, 1 H), 7.77 (s, 1 H), 7.42 (t, *J* = 8.5 Hz, 1 H), 7.22 (t, 2 H), 7.11 (s, 1 H), 6.94 (q, 1 H), 3.84 (s, 3 H).

MS: m/z = 174.

1-(4-Methoxyphenyl)-1*H*-imidazole (3h)⁶

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.12$ (s, 1 H), 7.64 (s, 1 H), 7.55 (d, J = 8.8 Hz, 2 H), 7.06 (d, J = 8.8 Hz, 3 H), 3.80 (s, 3 H).

MS: m/z = 174.

1-[3-(4-Fluorophenoxy)phenyl]-1H-imidazole (3i)

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.30$ (s, 1 H), 7.78 (s, 1 H), 7.45 (m, 3 H), 7.28 (m, 2 H), 7.15 (m, 2 H), 7.10 (s, 1 H), 6.88 (t, 1 H). ¹³C NMR (75 MHz, CDCl₃): $\delta = 161.0$, 159.2, 157.8, 151.7, 138.6, 135.5, 131.0, 130.4, 123.4, 121.3, 121.2, 120.3, 118.1, 116.8, 116.5, 116.4, 115.6, 114.9, 111.0.

MS: m/z = 254.

HRMS (EI): $m/z [M + H]^+$ calcd for $C_{15}H_{12}FN_2O$: 255.0928; found: 255.0936.

1-[4-Chloro-3-(trifluoromethyl)phenyl]-1*H*-imidazole (3j)

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.42$ (s, 1 H), 8.12 (d, 1 H), 8.02 (m, 1 H), 7.90 (m, 2 H), 7.15 (s, 1 H).

¹³C NMR (75 MHz, CDCl₃): δ = 136.0, 135.5, 133.1, 131.4, 131.2, 130.4, 127.5, 125.4, 123.9, 120.75, 120.67, 120.61, 120.54, 120.29, 118.1.

MS: m/z = 246.

HRMS (EI): m/z [M]⁺ calcd for $C_{10}H_7ClF_3N_2$: 247.0244; found: 247.0240.

3-(1H-Imidazol-1-yl)-5-(trifluoromethyl)aniline (3k)

¹H NMR (300 MHz, DMSO-*d*₆): δ = 8.20 (br, 1 H), 7.72 (br, 1 H), 7.10 (s, 1 H), 7.01 (s, 1 H), 6.97 (s, 1 H), 6.84 (s, 1 H), 5.89 (s, 2 H).

MS: m/z = 227.

HRMS (EI): m/z [M + H]⁺ calcd for C₁₀H₇ClF₃N₂: 228.0743; found: 228.0748.

1-[4-(1*H*-Imidazol-1-yl)phenyl]ethanone (3l)^{5t}

¹H NMR (300 MHz, DMSO- d_6): δ = 8.48 (s, 1 H), 8.15 (d, J = 8.8 Hz, 2 H), 7.94 (s, 1 H), 7.90 (d, J = 8.8 Hz, 2 H), 7.22 (s, 1 H), 2.67 (s, 3 H).

MS: m/z = 186.

4-(1H-Imidazol-1-yl)benzonitrile (3m)5t

¹H NMR (400 MHz, DMSO-*d*₆): δ = 8.43 (s, 1 H), 7.97 (d, *J* = 8.8 Hz, 2 H), 7.88 (d, *J* = 8.8 Hz, 3 H), 7.15 (s, 1 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 140.57, 136.25, 134.67, 130.99, 120.95, 118.81, 118.21, 109.53.

MS: m/z = 169.

4-(1H-Imidazol-1-yl)phenol (3n)5t

¹H NMR (400 MHz, DMSO- d_6): δ = 9.81 (br, 1 H), 8.15 (s, 1 H), 7.60 (s, 1 H), 7.39 (d, *J* = 8.8 Hz, 2 H), 7.11 (s, 1 H), 6.85 (d, *J* = 8.8 Hz, 2 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 157.03, 143.02, 129.24, 129.06, 122.86, 119.21, 116.57.

MS: m/z = 160.

4-(1H-Imidazol-1-yl)aniline (3o)^{5t}

¹H NMR (400 MHz, DMSO- d_6): δ = 7.95 (br, 1 H), 7.47 (s, 1 H), 7.23–7.19 (m, 2 H), 7.01 (s, 1 H), 6.67–6.63 (m, 2 H), 5.24 (br, 2 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 148.42, 135.81, 129.47, 126.72, 122.45, 118.90, 114.73.

MS: m/z = 159.

[4-(1*H*-Imidazol-1-yl)phenyl]methanol (3p)⁶

¹H NMR (400 MHz, DMSO- d_6): δ = 8.21 (s, 1 H), 7.70 (s, 1 H), 7.57 (d, *J* = 8.8 Hz, 2 H), 7.42 (d, *J* = 8.4 Hz, 2 H), 7.09 (s, 1 H), 5.32 (t, *J* = 6.0 Hz, 1 H), 4.52 (d, *J* = 6.0 Hz, 2 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 141.74, 135.95, 135.90, 130.17, 128.24, 120.55, 118.52, 115.48, 62.71.

MS: m/z = 174.

1-Naphthalen-1-yl-1*H*-imidazole (3q)^{5e}

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.15$ (d, 2 H), 8.00 (s, 1 H), 7.66 (m, 1 H), 7.60 (m, 2 H), 7.56 (m, 2 H), 7.50 (m, 1 H), 7.20 (s, 1 H).

MS: m/z = 194.

4-{[5-(1*H*-Imidazol-1-yl)pyridin-2-yl]methyl}morpholine (3r)

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.85$ (d, 1 H), 8.31 (s, 1 H), 8.07 (dd, J = 7.2, 3.6 Hz, 1 H), 7.81 (s, 1 H), 7.60 (m, 1 H), 7.15 (s, 1 H), 3.64 (s, 2 H), 3.60 (m, 4 H), 2.45 (m, 4 H).

¹³C NMR (75 MHz, CDCl₃): δ = 157.8, 142.2, 135.5, 132.6, 130.9, 129.4, 123.8, 118.1, 66.8, 64.2, 53.7.

MS: m/z = 244.

HRMS (EI): m/z [M + H]⁺ calcd for C₁₃H₁₇N₄O: 245.1397; found: 245.1392.

5-Chloro-2-(1*H*-imidazol-1-yl)pyrimidine (3s)

¹H NMR (300 MHz, DMSO- d_6): δ = 9.00 (s, 2 H), 8.56 (s, 1 H), 7.93 (s, 1 H), 7.16 (s, 1 H).

¹³C NMR (75 MHz, CDCl₃): δ = 190.5, 157.1, 136.3, 131.0, 116.6, 106.3.

MS: m/z = 180.

HRMS (EI): m/z [M + H]⁺ calcd for C₇H₆ClN₄: 181.0276; found: 181.0279.

3-(1*H*-Imidazol-1-yl)pyridine (3t)^{5u}

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.96$ (d, J = 3.2 Hz, 1 H), 8.57 (q, 1 H), 8.35 (s, 1 H), 8.11 (m, 1 H), 7.84 (s, 1 H), 7.57 (m, 1 H), 7.16 (s, 1 H).

MS: m/z = 145.

5-(1*H*-Imidazol-1-yl)-2-methoxypyridine (3u)^{5u}

¹H NMR (300 MHz, DMSO- d_6): δ = 8.48 (d, J = 4.0 Hz, 1 H), 8.17 (s, 1 H), 8.01 (dd, J = 8.0, 4.0 Hz, 1 H), 7.68 (s, 1 H), 7.11 (s, 1 H), 6.98 (d, J = 11.6 Hz, 1 H), 3.90 (s, 3 H).

MS: m/z = 175.

3-(1*H*-Imidazol-1-yl)-2-methylpyridine (3v)

¹H NMR (300 MHz, DMSO- d_6): δ = 8.56 (m, 1 H), 7.90 (s, 1 H), 7.75 (m, 1 H), 7.67 (s, 1 H), 7.43 (m, 1 H), 7.13 (s, 1 H), 2.36 (s, 3 H).

¹³C NMR (75 MHz, CDCl₃): δ = 154.6, 149.4, 137.3, 133.9, 132.8, 130.0, 121.7, 120.3, 40.9.

MS: m/z = 159.

HRMS (EI): m/z [M + H]⁺ calcd for C₉H₁₀N₃: 160.0869; found: 160.0871.

2,5-Bis(1*H*-imidazol-1-yl)pyridine (3w)

¹H NMR (300 MHz, DMSO- d_6): $\delta = 8.87$ (d, J = 3.6 Hz, 1 H), 8.58 (s, 1 H), 8.38 (s, 1 H), 8.34 (d, J = 3.6 Hz, 1 H), 8.00 (m, 2 H), 7.88 (s, 1 H), 7.16 (d, J = 8.0 Hz, 2 H).

¹³C NMR (75 MHz, CDCl₃): δ = 148.1, 142.1, 135.5, 135.0, 132.4, 132.1, 131.4, 131.2, 118.1, 116.1, 112.8.

MS: m/z = 211.

HRMS (EI): m/z [M + H]⁺ calcd for C₁₁H₁₀N₅: 212.0931; found: 212.0933.

1-Phenyl-1*H*-1,2,4-triazole (4a)^{5r}

H NMR (300 MHz, DMSO- d_6): δ = 9.30 (s, 1 H), 8.24 (s, 1 H), 7.88 (m, 2 H), 7.58 (m, 2 H), 7.43 (t, J = 7.2 Hz, 1 H).

MS: m/z = 145.

1-Biphenyl-4-yl-1*H*-1,2,4-triazole (4b)^{8b}

¹H NMR (300 MHz, DMSO- d_6): $\delta = 9.36$ (s, 1 H), 8.27 (s, 1 H), 7.95–7.98 (d, J = 8.9 Hz, 2 H), 7.85–7.89 (d, J = 8.8 Hz, 2 H), 7.73–7.76 (t, J = 7.2 Hz, 2 H), 7.47–7.53 (t, J = 7.8 Hz, 2 H), 7.38–7.43 (t, J = 7.3 Hz, 1 H).

MS: m/z = 221.

4-(1*H*-Benzoimidazol-1-yl)benzonitrile (4c)^{5s}

¹H NMR (400 MHz, DMSO- d_6): δ = 8.65 (s, 1 H), 8.06 (d, *J* = 8.4 Hz, 2 H), 7.90 (d, *J* = 8.8 Hz, 2 H), 7.79–7.77 (m, 1 H), 7.71–7.68 (m, 1 H), 7.36–7.31 (m, 2 H).

¹³C NMR (100 MHz, DMSO-*d*₆): δ = 144.43, 143.57, 140.20, 134.75, 132.77, 124.45, 124.30, 123.54, 120.62, 118.77, 111.35, 110.30.

MS: m/z = 219.

2-Methyl-1-phenyl-1*H*-imidazole (4d)^{5r}

¹H NMR (300 MHz, DMSO- d_6): δ = 7.54 (m, 2 H), 7.45 (m, 3 H), 7.27 (s, 1 H), 6.92 (s, 1 H), 2.28 (s, 3 H).

MS: m/z = 158.

1-Phenyl-2-propyl-1*H*-imidazole (4e)

¹H NMR (300 MHz, DMSO- d_6): δ = 7.51–7.56 (m, 3 H), 7.41–7.48 (m, 2 H), 7.23 (s, 1 H), 6.94 (s, 1 H), 2.50–2.58 (m, 2 H), 1.53–1.60 (m, 2 H), 0.78–0.81 (m, 3 H).

¹³C NMR (75 MHz, CDCl₃): δ = 148.5, 137.9, 129.3, 128.2, 127.5, 125.8, 120.6, 28.9, 21.4, 13.7.

MS: m/z = 186.

HRMS (EI): m/z [M + H]⁺ calcd for C₁₂H₁₅N₂: 187.1230; found: 187.1233.

3-(4-Methyl-1*H***-imidazol-1-yl)-5-(trifluoromethyl)aniline (4f)^{8a}** ¹H NMR (400 MHz, DMSO- d_6): δ = 8.06 (d, *J* = 0.8 Hz, 1 H), 7.35 (s, 1 H), 6.97 (s, 1 H), 6.94 (s, 1 H), 6.82 (s, 1 H), 5.87 (s, 2 H), 2.15 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 151.29, 138.89, 135.17, 135.81, 131.50, 125.81, 123.10, 114.62, 108.32, 103.69, 103.65, 13.97.

MS: m/z = 241.

3-(4-Methyl-1*H*-pyrazol-1-yl)-5-(trifluoromethyl)aniline (4g)

¹H NMR (400 MHz, DMSO- d_6): $\delta = 8.22$ (s, 1 H), 7.54 (s, 1 H), 7.24 (s, 1 H), 7.14 (s, 1 H), 6.74 (s, 1 H), 5.82 (s, 2 H), 2.08 (s, 3 H).

¹³C NMR (100 MHz, DMSO- d_6): δ = 151.05, 142.17, 141.58, 131.73, 131.42, 131.11, 130.80, 126.63, 125.98, 123.27, 118.30, 107.50, 107.46, 106.38, 101.55, 101.50, 9.13.

MS: m/z = 241.

HRMS (EI): m/z [M + H]⁺ calcd for C₁₆H₁₄FN₂O: 242.0900; found: 242.0908.

1-[3-(4-Fluorophenoxy)phenyl]-2-methyl-1*H*-imidazole (4h)

¹H NMR (300 MHz, DMSO- d_6): δ = 7.48–7.54 (t, J = 7.8 Hz, 1 H), 7.26–7.30 (m, 3 H), 7.20–7.25 (m, 3 H), 7.14–7.19 (m, 1 H), 6.99–7.06 (m, 1 H), 6.89 (s, 1 H), 2.28 (s, 3 H).

¹³C NMR (75 MHz, DMSO- d_6): δ = 161.0, 158.8, 157.8, 151.73, 151.7, 144.5, 139.3, 130.5, 127.8, 121.30, 121.25, 121.19, 120.5, 119.7, 117.1, 116.8, 116.5, 114.8, 13.8.

MS: m/z = 268.

HRMS (EI): $m/z \,[M + H]^+$ calcd for $C_{16}H_{14}FN_2O$: 269.1085; found: 269.1094.

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