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Asymmetric Coupling of Carbon-Centered Radical Adjacent to Nitrogen: Copper-Catalyzed Cyanation and Etherification of Enamides

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Dedication to the celebration of 70th anniversary of SIOC

Abstract: The first copper-catalyzed asymmetric cyanation and etherification of enamides have been established, where a carboncentered radical adjacent to a nitrogen atom (CRAN) is enantioselectively trapped by a chiral copper (II) species. Moreover, the asymmetric cyanation of vinyl esters was disclosed as well. These reactions featuring very mild reaction conditions and high functional group tolerance give a series of chiral α -cyano amides, α -cyano esters and α -hemiaminals in good yields with excellent enantioselectivities, in which chiral α -cyano amides can be easily converted into enantio-enriched 1,2-diamines and amino acids.

Optically pure amines and amides represent not only one of most privileged moieties frequently found in natural products and drugs, but also act as useful synthons in organic synthesis.^[1] Therefore, much effort has been devoted to their synthesis and many well-established methods have been documented,^[2] such as asymmetric hydrogenation,^[3] hydroboration,^[4] Michael addition,^[5] and so on. Owing to highly reactive radical species, difunctionalization of alkenes via radical pathways recently serves as a powerful tool in organic synthesis.^[6] Among them, reactions of enamides present an attractive method for the concise and efficient synthesis of amides, where a carbon radical adjacent to a nitrogen atom (CRAN) is considered as a key intermediate.^[7] To survey the possible asymmetric reaction, the enantioseletive capture of CRAN becomes an inevitable problem.^[8] Recently, Phipps developed the first CPA-catalyzed asymmetric decarboxylative Minisci-type reaction, where a chiral phosphoric acid (CPA) catalyst could promote highly enenatioselective radical addition of CRAN to pyridines via a hydrogen-bonding process.^[9] Later on, Studer and coworkers untilized this strategy to disclose the elegant asymmetric radical difunctionation of enamides (Scheme 1a).^{[10}

As our ongoing research interest in asymmetric radical transformations (ATRs),^[11] our group has developed a series of coppercatalyzed asymmetric diffunctionalizations of styrenes, such as cyanation, arylation and alkynylation reactions, where a benzylic radical intermedaite generated by radical addition across styrenes could be efficiently and enantioselectively trapped by the corresponding chiral (Box)Cu(II) species (Box = bisoxazoline), leading to the formation of a chiral C-C bond (Scheme 1b, *i*).^[11a, 12] However, when

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Scheme 1. Asymmetric radical difunctionalization of enamides.

aliphatic alkenes were employed as substrate, these reactions involving a non-benzylic radical intermediate furnished the corresponding products with low enantioselectivities. Although benzylic,^[12a-b,13] allylic^[14] and propargylic radicals^[15] could be efficiently and enantioselectively trapped by chiral (Box)Cu(CN)₂ species to deliver enantioenriched organonitriles, the asymmetric cyanation of alkylsubstituted carbon-centered radicals yielding the dialkyl-substituted organonitriles with low levels of enantioselective induction still remains a big challenging issue, presumably due to the less stabilities of the transiently generated alkyl-substituted carbon-centered radical. It is well known that a carbon-centered radical can be stablized by heteroatoms adjacent to the radical such as nitrogen,^[16] thus promoting us to survey the enantioselective enantioselective cyanation of CRAN generated by radical addition to enamides; in addition, a coordination of the carbonyl group in the enamide to a copper center woud be highly beneficial to the enantioselective induction (Scheme 1b, ii). Moreover, Stahl and coworkers recently reported a copper-catalyzed etherification of benzylic C-H bonds.^[17] However, employing chiral Box ligand failed to achieve an enantioselective reaction, yielding racemic oxygenation products. The significantly different properties between CRAN and benzylic radicals also intrigued us to investigate the asymmetric coupling of CRAN to alcohols (Scheme 1b, iii). However, it is noteworthy that CRAN can be easily over oxidized to an iminium cation,^[18] making the enantioselective coupling reaction more challenging. Herein, we communicate the first copper-catalyzed asymmetric cyanation of enamides, involving highly enantioselective capture of CRAN by chiral L*Cu(CN)2 species. Moreover, the enantioselective coupling of enamides to alcohols is also realized by the same strategy (Scheme 1b). In addition, our method features excellent functional group compatibility and provides an easy and

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efficient access to a wide array of enantiomerically enriched CF₃containing amides.

To test the above hypothesis, we initially focused on the asymmetric cyanation of enamides as the mode reaction. First, the reaction of N-vinyl benzamide 1a with Togni-I reagent was surveyed, using 5 mol% Cu(CH₃CN)₄PF₆ and 6 mol% privileged Box ligand in our previous reports.^[12a] As shown in Table 1, the Box ligand L1 with the indane moiety yielded the desired product 2a in 53% yield with poor enantioselectivity (19% ee, entry 1). To our delight, the enantioselectivity of 2a could be significantly improved by introducing substituents into the Box ligands at the gem-carbon position (L2-L3), but a yield of 2a was dramatically decreased (entries 2-3); for instance, the reaction with ligand L3 provided 2a in 18% yield with 76% ee. The Box ligands L4-L6 with benzyl groups on the oxazoline rings were also investigated, and L4 exhibited a slightly lower enantioselectivity; using L6 led to similar results as using L3 (entries 4-6). In the presence of L3, copper catalyst screening revealed that CuOAc exhibited the best reactivity to provide 2a in 64% yield, albelt with 69% ee (entry 7, for details, see SI). Moreover, solvents also played a key role in the reaction, and methyl tert-butyl ether (MTBE) proved to be the best to afford 2a in 98% yield with 87% ee (entry 9). Notably, 1,2,4,5tetrafluorobenzene gave 2a with a better enantioselectivity but in a moderate yield (entry 8). Decreasing the reaction temperature resulted in better enantioselectivities without loss of the reaction efficiency. The reaction performed at -20 °C and using L3 as ligand in MTBE provided 2a in 98% yield with 96% ee as the best results (entries 10-12). Again, L6 and L3 exhibited similar reactivities under the optimized reaction conditions (entries 12-13).

 Table 1. Condition Screening^[a]

| | o II ^ | cat. Cu(I)/ L* TMSCN (2.0 equi [,] | v) C | CN L CE | |
|-----------|---------------------------------|---|---------------------------------|------------------------------|----|
| Ph´ | H 1a | Togni I CF ₃ * (2.0 eq Solvent, rt, <i>N</i> ₂ | uiv) Ph | N 2a | |
| Entry | Ligand | Cu catalyst | Solvent | 2a yield (ee) ^[b] | |
| 1 | L1 | Cu(CH ₃ CN) ₄ PF ₆ | CH ₂ Cl ₂ | 53% (19%) | b. |
| 2 | L2 | Cu(CH ₃ CN) ₄ PF ₆ | CH_2CI_2 | 20% (35%) | |
| 3 | L3 | Cu(CH ₃ CN) ₄ PF ₆ | CH ₂ Cl ₂ | 18% (76%) | |
| 4 | L4 | Cu(CH ₃ CN) ₄ PF ₆ | CH ₂ Cl ₂ | 35% (62%) | |
| 5 | L5 | Cu(CH ₃ CN) ₄ PF ₆ | CH ₂ Cl ₂ | 13% (70%) | 7 |
| 6 | L6 | Cu(CH ₃ CN) ₄ PF ₆ | CH ₂ Cl ₂ | 18% (74%) | |
| 7 | L3 | CuOAc | CH ₂ Cl ₂ | 64% (69%) | |
| 8 | L3 | CuOAc | $C_6H_2F_4$ | 58% (94%) | |
| 9 | L3 | CuOAc | MTBE | 98% (87%) | |
| 10 | L3 | CuOAc | MTBE | 98% (92%) ^[c] | 1 |
| 11 | L3 | CuOAc | MTBE | 98% (95%) ^[d] | |
| 12 | L3 | CuOAc | MTBE | 98% (96%) ^[e] | |
| 13 | L6 | CuOAc | МТВЕ | 98% (92%) ^[e] | |
| | RR | | p1 p2 | | |
| ,0. | $\mathcal{X}_{\mathcal{V}}^{O}$ | | | | |
| \square | N N-/ | | | \rangle | |
| | N. | | | L4 R = Et | |
| | | LZ R = Me D | | L5 R = Bn | |
| | | | | LO R= 2-Np-CH ₂ | |

[a] Reaction conditions: **1a** (0.1 mmol), Cu salt (5 mol%), ligand (6 mol%), Togni-I (0.2 mmol) and TMSCN (0.2 mmol) in solvent (1 mL) under a nitrogen atmosphere at room temperature; [b] Yields were determined by crude ¹H NMR with CH_2Br_2 as an internal standard, and enantiomeric excess (ee) value was determined by HPLC on a chiral stationary phase; [c] At 0 °C. [d] At -10 °C. [e] At -20 °C.

With the optimized reaction conditions in hand, the substrate scope with respect to enamides was then explored, and these results were summarized in Table 2A. *N*-Vinyl benzamides with various substituents on the aromatic ring were suitable for the reaction to provide the corresponding products **2a-2h** in excellent yields (85%-98%) with excellent enantioselectivities (92%-98% ee). In addition, reactions of *N*-vinyl-amides with 1- or 2-naphthalenyl groups also proceeded very well to give products **2i** in 90% yield (93% ee) and **2j**

in 93% yield (92% ee), respectively. Moreover, substrates derived from alkyl carboxylic acids were also suitable, and the reactions provided 2k and 21 in 92-97% yields with 93-94% ee. Interestingly, reactions of Nvinyl Boc-amides also worked nicely to furnish 2m in 72% yield with 98% ee, which is considered as a valuable synthon in organic synthesis. Notably, N-vinyl benzamides, in which the N-H bond was protected by phenyl or Boc groups, also proved amenable to the reaction, yielding products 2n-2o in good yields with excellent enantioselectivities (86-93% ee). N-vinyl phthalimde and N-vinly lactams could also be employed as substrate to afford the corresponding products 2p-2r in excellent yields with 87-90% ee. More importantly, the compatiblity of various heteroarenes (furan, thiophene and pyridine) was examined under the current reaction conditions, and the corresponding products 2s-2u were furnished in 70-94% yields with 93-96% ee. Unfortunately, β -substitued N-vinyl amides as substrate failed to deliver tetrasubstituted product (2v). Notably, the reaction of 1a could be performed on a 10 mmol scale without loss of reaction efficiency and enantioselectivity. The absolute structure of (S)-2a was unambiguously determined by the X-ray crystallography.[19

A carbon-centered radical adjacent to oxygen has been extensively studied, and its properties were similar to CRAN. We then turned our attention to testing the asymmetric cyanation of the carbon-centered radicals adjacent to an oxygen atom. To our delight, the asymmetric trifluoromethylcyanation reaction of 3a indeed worked, in the presence of L6 (Table 2B, for the optimization of the reaction conditions, see SI), giving the product 4a with excellent enantioselectivity (90% ee), and a series of enantioenriched α -cyano esters were successfully synthesized. Both vinyl aryl esters and vinyl alkyl esters yielded the corresponding products 4a-4e in good to excellent yields (70-99%) with excellent enantioselectivities (87-91% ee). Notably, the reaction displays high functional group tolerance, such as methoxyl, chloro and trifluoromethyl moieties were compatible with the current reaction conditions. More importantly, substrates bearing heterocycles such as indole and thiophene, were suitable for the reaction to delivere the desired product 4f-4g in good to excellent yields (70-90%) with excellent enantioselectivities (89-90% ee). When chiral vinyl amino acid ester 3h was employed as substrate, to our delight, the desired product 4h was obtained in good yields with excellent diastereoselectivities (d.r. = 10:1). Lastly, other various vinyl esters as substrate were tested, vinyl phosphinate 3i gave product 4i in 86% vield with 92% ee, while the reaction of vinyl-OTs 3j just afforded 4j in 65% yield with moderate enantioselectivity (60% ee).

Above results demonstrated that the generated CRAN could be enantioselectively trapped by chiral copper(II) cyanide. Inspired by these results, then we turned our attention to testing the asymmetric coupling of CRAN with alcohols. Under the cyanation reaction conditions, (MeO)₄Si instead of TMSCN was initially employed as a methoxide source, the reaction of enamide **1a** with (MeO)₄Si afforded the desired trifluoromethoxylation product **6a** in 37% yield with 53% ee, in the presence of 10 mol% Cu(CH₃CN)₄PF₆/12 mol% **L3**, and 51% yield and 67% ee were obtained when 10 mol% Cu(CH₃CN)₄PF₆/12 mol% **L6** were used. After systematic screening, 10 mol% Cu(CH₃)₄PF₆/12 mol% **L7** as catalyst provided a better performance to give **6a** in 59% yield with 89% ee. Gratifyingly, simple MeOH could be directly applied to the reaction, providing **6a** in 84% yield with 94% ee as the best results (for details, see SI).

After having established the optimal reaction conditions, the substrate scope of the oxygenation of enamides was also explored under the optimized reaction conditions, and the results were summarized in Table 3. The reaction showed excellent functional group tolerance, *N*-vinyl benzamides with various substituents on the aromatic ring such as electron-donating groups (methyl, methoxyl) and electron-deficient groups (halogen, phenyl, cyano, trifluoromethyl) were suitable for the reaction to provide the corresponding products **6a-6j** in moderate to excellent yield (55-91%) with excellent enantio-selectivities (84%-96% ee). Notably, the *ortho*-substituted substrate exhibited slightly lower enantioselectivity (**6i**, 86% ee). In addition, *N*-vinyl-amides bearing 1- or 2-naphthalenyl groups could also give products **6k** in 65% yield (98% ee) and **6l** in 74% yield (84% ee), respectively. Enamide substrates derived from alkyl carboxylic acids

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Table 2. Scope of enamides and vinyl esters.^[a,b]



[a] All reactions were run on a 0.2 mmol scale in MTBE (2 mL) under a nitrogen atmosphere. [b] Isolated yields, and enantiomeric excess (ee) values were determined by HPLC on a chiral stationary phase. [c] On a 10 mmol scale. [d] Using L6.

Table 3. Scope of enamides and alcohols.^[a,b]



[a] All reactions were run on 0.2 mmol scale in DMAc (1 mL) under a nitrogen atmosphere. [b] Isolated yields, and enantiomeric excess (ee) values determined by HPLC on a chiral stationary phase. [c] Reaction at -20 $^{\circ}$ C.

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were also suitable for reaction to give 6m in 55% yield with 86% ee. Moreover, enamides bearing heterocycles, such as pyrridine and furan, were also compatible with the reaction conditions to deliver products 6n and 60 in good yields (80-87%) with excellent enantioselectivities (95% ee).

Lastly, a series of alcohols as coupling partners were investigated. Reactions of simple alcohols, such as ethanol, propanol, pentanol and benzyl alcohol, proceeded smoothyl to yield corresponding products 6p-6s with excellent results. Notably, the oxygenation reaction of CRAN also featured high functional group tolerance, a series of alcohols containing various functional groups such as cyclopropanyl (6t), diol (6u), sulfonyl (6v), alkynyl (6w), aniline (6x) and methylenedioxybenzene (6y) were subjected to the reaction conditions, giving the desired products in 60-86% vields with 80-97% ee. Moreover, reactions of alcohols bearing heteroarenes including furan and thiophene also proceeded very well to afford 6z-7a in good yields with excellent enantioselectivities, except that the reaction of alcohol with benzoxazoline gave product 7b with moderate enantioselectivity (60% ee). In addition, reactions of enamides with chiral alcohols also furnished the products 7c-7e in good yields with high levels of diastereoselective induction (94-97% de). Alcohols with a long chain (7f) also proved amenable to the asymmetric etherification.

In order to showcase the synthetic utility of the method, further transformations of the trifluoromethylcyanation products were investigated (Scheme 2). Optically pure 1,2-diamine **8** could be readily prepared through the hydrogenation of **2a** using Raney nickel catalyst in 90% yield with 99% ee. Moreover, **2q** could be transformed into amide **9** in 86% yield with 86% ee, the core structure of which is similar to that of antiepileptic drug Levetiracetam;^[20] Interestingly, further recrystallization of product **9** in isopropanol yielded a small amount of precipitate only, and we were delighted to find that the enantioselectivity of **9** in solution was accumulated to 95% ee in 75% yield.



Scheme 2. Synthetic applications.

Some control experiments revealed that the reactions involves a radical pathway (for details, see SI). In addition, compared to vinyl enolate 3a, the reaction of vinyl benzyl ethers yielded the desired product with low enantioselectivity (10, 16% ee, Scheme 3, i), indicating that the carbonyl groups in vinyl substrates were essential for the asymmetric coupling. There are two possible roles for the carbonyls: (1) decreasing the electron density of a carboncentered radical and thus avoiding its overoxidation to a carbocation; and (2) acting as a coordinating site for chiral copper(II) species to facilitate the asymmetric coupling of the chiral copper(II) species with carbon-centered radicals. For vinyl phosphite (3i) and sulfonate (3j) bearing a stronger electronwithdrawing group than the carbonyl group in 3a, the reaction of 3i led to slightly higher enantioselectivity than 3a, while the reaction of 3j gave product 4j with much lower ee value (Scheme 3, i). We reasoned that these outcomes might be attributed to the order of the chelating ability of groups: phosphinyl > acyl > sulfonyl. Moreover, compared to the CRAN generated from substrate 1p (90% ee), when two more carbon atoms were added to the carbon-centered radical connecting to a nitrogen atom, the cyanation product 11 was produced with lower enantioselectivity (70% ee), and increasing the carbon atoms resulted in even worse enantioselectivity (<5% ee, 12) (Scheme 3b, *ii*). Based on these studies, we concluded that the carbonyl group coordinating to chiral Cu(II) species played an essential role in the asymmetric radical coupling step.



Scheme 3. Preliminary mechanistic studies.

In conclusion, we have developed the first copper-catalyzed asymmetric cyanation and oxygenation of enamides proceeding through a highly enantioselective capture of CRAN by a chiral copper complex; and the asymmetric cyanation of vinyl esters was illustrated as well. These methods allow for the straightforward and efficient synthesis of various α -cyano amides, α -cyano ethers and hemiaminals in good yields with excellent enantioselectivities under very mild conditions. Preliminary mechanistic studies reveal that the carbonyl group in the enamides plays a crucial role in achieving high levels of enantioselectivities in the asymmetric cyanation and etherification of the carbon-centered radicals adjacent to a heteroatom.

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Keywords: asymmetric radical reaction • copper catalysis • cyanation • etherification • enamides

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cyanation of vinyl esters was illustrated as well. The reactions featuring very mild reaction conditions and high functional group tolerance give a series of chiral a-cyano amides, a-cyano esters and hemiaminals in good yields with excellent enantioselectivities.