Gold-Catalyzed Reactions between Alkenyldiazo Carbonyl Species and Acetals

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Supporting Information

ABSTRACT: In the presence of catalyst IPrAuSbF₆ catalyst (IPr = 1,3-bis(diisopropylphenyl)imidazol-2-ylidene), alkenyldiazo carbonyl species react with organic acetals to give *E*configured alkyl 3,5-dimethoxy-5-pent-2-enoates stereoselectively. This reaction sequence comprises an initial Prins-type reaction, followed by gold carbene formation.

lkenyldiazo carbonyl species are useful reagents to access complicated acyclic and cyclic molecules.¹ The major use of these reagents relies on their initial diazo decomposition with a metal catalyst to generate reactive alkenylmetal carbene intermediates, further enabling stereo- or regioselective [3 + n]cycloadditions (n = 2-4) with suitable dipolarophiles.²⁻⁴ This reaction route has found widespread applications to access complicated carbo- and heterocyclic compounds.²⁻⁴ Alternatively, in the presence of suitable Lewis acid catalysts, these alkenyl diazo reagent work as nucleophiles to react with aldehydes and enones to give addition products to retain their diazo functionalities.⁵ We recently reported⁶ formal [4 + 2]cycloadditions of these diazo species (1) with N-arylimines (Povarov reaction)⁷ to give isolable diazo-containing cycloadducts (3) using HOTf (3 mol %) or IPrAuCl/AgNTf₂ (2.5 mol %, IPr = 1, 3-bis(diisopropylphenyl)imidazol-2-ylidene), as depicted in eq 1. This reaction sequence is applicable also to





diphenoxymethylbenzenes (4) to effect an unprecedented oxa-Povarov reaction, further affording oxacyclic products (5) stereoselectively (eq 2).⁶ In this work, we report a distinct and new route between alkenyldiazo carbonyl species 1 and commonly used acetal substrates 6; this process generates 1,3-dialkoxy-4-diazo intermediate I, further giving *E*-configured ethyl 3,5-dimethoxy-5-pent-2-enoate 7 stereoselectively (eq 3).



Table 1 shows the reactions between vinyldiazo ester 1a and dimethoxymethylbenzene 6a using various acid catalysts. We

Table 1. Catalyst Screenings over Various Acid Catalysts

	QMe			OMe I	
	D ₂ Et OMe catalyst Ph + Ph OMe DCE, 25 °C MeO 6a 7a		I <mark>}</mark> Ph ₂O M 8h	е0 7а'	
			con (yi	npounds eld, %) ^b	
entry	catalyst (mol %)	time (h)	1a	7a	
1	IPrAuCl/AgNTf ₂ (2.5)	0.1		80	
2	IPrAuCl/AgSbF ₆ (2.5)	0.1		84 (81) ^c	
3	$IPrAuCl/AgSbF_{6}$ (2.5)	1.0		77^d	
4	$LAuCl/AgSbF_{6}$ (2.5)	0.1		67	
5	$PPh_{3}AuCl/AgSbF_{6}$ (5)	0.3		57	
6	$AgSbF_6$ (2.5)	0.3		46	
7	$Rh_{2}(OAc)_{4}$ (2.5)	1.2			
8	$Cu(OTf)_2(5)$	1.5			
9	$Zn(OTf)_2(5)$	2.0			
10	HOTf (3)	2.5			
11	$Sc(OTf)_3$ (5)	6.0	46		
12	$In(OTf)_3$ (5)	6.0	51		
13	HOTf (2) /PhCH $(NH_2)Me(3)$	6.0	76		

^{*a*}**1a** (1.0 equiv, 0.50 M), **6a** (1.0 equiv), IPr = 1,3-bis-(diisopropylphenyl)imidazol-2-ylidene, L = P(*t*-Bu)2(*o*-biphenyl). ^{*b*}Product yields are given after purification from a silica column. ^{*c*}The value in parentheses corresponds to a prior removal of AgCl. ^{*d*}this reaction was operated at 1.0 g scale with 0.1 M.

first tested the reactions with IPrAuCl/AgX (2.5 mol %, X = NTf₂ and SbF₆) that were found to be chemoselective toward the Povarov reaction^{6,7} of vinyldiazo species **1a** (see eq 1). Herein, we obtained ethyl 3,5-dimethoxy-5-pent-2-enoate 7**a** in 80–84% yields (entries 1 and 2). With IPrAuCl/AgSbF₆, the yield of compound 7**a** was kept at 81% with a prior filtration of

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Note

Table 2. Reaction Scope with Various Acetal Substrates



^aProduct yields are given after purification from a silica column.

AgCl through a Celite bed. If the reaction was operated with diazo reagent 1a (0.1 M) on a gram scale, the desired 7a was obtained with 77% yield (entry 3). Other gold catalysts LAuCl/AgSbF₆ (L = P(*t*-Bu)₂(*o*-biphenyl)) and PPh₃AuCl/AgSbF₆ appear to be less productive, giving desired unsaturated ester 7a in 57–67% yields (entries 4 and 5). The use of AgSbF₆ alone gave ester 7a in 46% yield (entry 6). Other catalysts Rh₂(OAc)₄, Cu(OTf)₂, Zn(OTf)₂, and HOTf, each at 2.5–5 mol % loading (entries 7–10), gave a complicated mixture of products whereas Sc(OTf)₃, In(OTf)₃, and HOTf/PhCH-(NH₂)Me led to recovery of vinyldiazo species 1a in large proportions (46–76%, entries 11–13). The *E*-configuration of ester 7a is inferred from the X-ray diffraction⁸ of its acid derivative 7a', which was obtained by LiOH-catalyzed hydrolysis in a THF/H₂O (2:1) solution.

Table 2 shows the scope of this gold-catalyzed reaction with various acetals 6b-p. The reaction is applicable to acetals 6b-6c bearing various alkoxy groups (R = Et; allyl), giving desired unsaturated esters 7b and 7c in 63% and 57% yields, respectively. The scope of the reaction is expanded with its applicability to phenyldimethoxymethanes 6d-g bearing

various phenyl substituents including electron-rich 4-methyl, 4-*tert*-butyl, 4-methoxy, and 2-methoxy, affording desired unsaturated esters 7d-g in good yields (73–89%, entries 3–6). The same reactions worked well with electron-deficient phenyl substrates **6h** and **6i** bearing 4-chloro and 4-bromo substituents, giving the expected products **7h** and **7i** in 74% and 78% yields, respectively (entries 7 and 8). We also tested the reactions on heteroaryl substrates **6j–n** including 3-furanyl, 3-thienyl, 2-benzofuranyl, 3-indolyl, and 2-benzothienyl; their corresponding products **7j–n** had 67–77% yields (entries 9–13). We tested the reactions on alkenyl acetals **60** and **6p** to provide unsaturated esters **70** and **7p** in 86–87% yields (entries 14 and 15).

To expand the reaction scope, we also prepared various diazovinyl carbonyl reagents 1b-g; the results are provided in Table 3. As shown in entries 1–3, the reactions between various vinyldiazo esters 1b-d and acetal **6a** proceeded smoothly to afford desired unsaturated esters 8b-d (X = $O^{t}Bu$), Oallyl, OBn) in 72–82% yields, respectively. The same reactions were applicable to vinyldiazo ketones 1e-g (X = "Pr,

Table 3. Reaction Scope with Various Vinyldiazo Substrates



2-thienyl, phenyl), giving unsaturated ketones 8e-g in satisfactory yields (72–86%).

Shown in Scheme 1 are additional experiments undertaken to understand the reaction mechanism. We prepared deuterated





sample d_1 -1a with its alkenyl C(2)-proton fully deuterated; its gold-catalyzed reaction with phenyldimethoxymethane d_0-6a gave the desired d_1 -7a bearing a fully deuterated =CDCO₂Et, indicative of a 1,2-hydrogen shift. We performed the same reaction using acetals d_6 -6a and d_9 -6f in equimolar proportions, their reactions with vinyldiazo ester 1a (2 equiv) gave unsaturated esters d₃-7a and d₃-7f in 76% and 78% yields, respectively. The two methoxy groups of each product contained 50% deuterium content at the benzyl and alkenyl positions according to ¹H NMR analysis whereas their mass spectra revealed a 1:2:1 ratio, corresponding to the ratios of $d_0:d_3:d_6$ samples. Notably, the $d_0:d_3:d_6$ ratio remains the same even though diazo reagent 1a was absent. This information confirms an intermolecular gold-catalyzed exchange of methoxy between two acetals to give d_3 -6a and d_3 -6f before their reactions with diazo species 1a.

Scheme 2 shows a plausible mechanism involving the intermediacy of gold carbene species **D**. As $IPrAuSbF_6$ enables a rapid methoxy exchange between two acetals, as shown in Scheme 1, this observation indicates a facile formation of the oxonium intermediate **A**. We envisage that gold activates a Prins-type reaction^{9,10} between vinyldiazo ester **1a** and acetal **6a** to generate 1,3-dimethoxy-4-diazo ester **C**; the success of this reaction relies on the stabilization of an adjacent



carbocation by daizo functionality as in intermediate **B**. The postulated carbene intermediate **D** is supported by a 1,2-hydrogen shift^{11,12} in the deuterium labeling experiment (Scheme 1). The excellent *E*-selectivity in this 1,2-shift is indicative also of gold carbene species^{11,13} whereas the Z-selectivity is known for rhodium carbenes.^{12,13}

In summary, gold-catalyzed reactions between vinyldiazo carbonyl species and acetals are described; the resulting *E*-configured alkyl 3,5-dimethoxy-5-pent-2- enoates are obtained in good yields. According to our experimental data, this reaction sequence comprises an initial Prins-type reaction, followed by gold-carbene generation. The success of this Prins-type reaction indicates a stabilization effect of the diazo functionality on the adjacent carbocation; this information allows new designs between alkenyldiazo species and other electrophiles.

EXPERIMENTAL SECTION

General Comments. Unless otherwise noted, all reactions to prepare the substrates were performed in oven-dried glassware under nitrogen atmosphere with freshly distilled solvents. The catalytic reactions were performed under nitrogen atmosphere. DCE, DCM, and CH₃CN were distilled from CaH₂ under nitrogen. THF were distilled from Na metal under nitrogen. All other commercial reagents were used without further purification, unless otherwise indicated. ¹H and ¹³C NMR spectra were recorded on spectrometers 400, 500, and 600 MHz using chloroform- d_1 (CDCl₃), and dimethyl sulfoxide- d_6 (DMSO) as internal standards. Alkenyldiazo compounds (1a–g) were prepared according to literature references.^{4a,6,14}

General Procedure for the Gold-Catalyzed Reaction between Acetal (6a) and Vinyldiazo Ester (1a). A dichloroethane (DCE, 1.0 mL) solution of IPrAuCl (7.8 mg, 0.012 mmol) and AgSbF₆ (4.3 mg, 0.012 mmol) was stirred at 25 °C for 5 min, and to this solution was added a DCE (1.5 mL) solution of alkenyl diazoacetates (1a) (70 mg, 0.50 mmol) and acetal (6a) (76 mg, 0.50 mmol) over 5 min. The resulting solution was stirred at 25 °C. After 10 min, the solution was filtered over a short silica bed; the solvent was evaporated under reduced pressure. The residue was purified on a flash silica gel column (hexane/ethyl acetate = 9/1) to give compound 7a as a colorless liquid (110 mg, 84% yield).

(*E*)-Ethyl 3,5-dimethoxy-5-phenylpent-2-enoate (7a): colorless liquid (110 mg, 84%); ¹H NMR (600 MHz, CDCl₃) δ 7.34–7.29 (m, 4H), 7.23 (t, *J* = 7.0 Hz, 1H), 5.00 (s, 1H), 4.51 (dd, *J* = 5.3, 8.7 Hz, 1H), 4.11 (q, *J* = 7.0 Hz, 2H), 3.55 (s, 3H), 3.19 (s, 3H), 3.16 (dd, *J* = 5.2, 13.6 Hz, 1H), 3.07 (dd, *J* = 5.3, 13.6 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 173.0, 167.3, 141.4, 128.2, 127.5, 126.6, 91.9, 81.9, 59.3, 56.9, 55.5, 40.8, 14.4; HRMS calcd for C₁₅H₂₀O₄ 264.1362, found 264.1362.

Deuterated (*E*)-ethyl 3,5-dimethoxy-5-phenylpent-2-enoate (d₁-7a): colorless liquid (91 mg,70%); ¹H NMR (600 MHz, CDCl₃) δ 7.34–7.29 (m, 4H), 7.23 (d, *J* = 7.1 Hz, 1H), 4.51 (dd, *J* = 5.4, 8.7 Hz, 1H), 4.11 (q, *J* = 7.1 Hz, 2H), 3.59 (s, 3H), 3.20–3.15 (m, 4H), 3.08 (dd, *J* = 5.5, 13.4 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150

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MHz, CDCl₃) δ 172.9, 167.3, 141.4, 128.2, 127.5, 126.7, 91.7 (t, *J* = 24 Hz), 81.9, 59.3, 56.9, 55.5, 40.8, 14.4; HRMS calcd for C₁₅H₁₉DO₄ 265.1424, found 265.1428.

Deuterated (*E***)-ethyl 3,5-dimethoxy-5-phenylpent-2-enoate (d₃-7a):** colorless liquid (101 mg, 76%) ; ¹H NMR (600 MHz, CDCl₃) δ 7.35–7.29 (m, 4H), 7.23 (t, *J* = 7.1 Hz, 1H), 5.00 (d, *J* = 3.4 Hz, 1H), 4.51 (dd, *J* = 5.2, 8.7 Hz, 1H), 4.11 (q, *J* = 7.2 Hz, 2H), 3.56 (s, 1.5H), 3.19 (s, 1.47 H), 3.15 (dd, *J* = 8.7, 13.5 Hz, 1H), 3.08 (dd, *J* = 5.3, 13.5 Hz, 1H), 1.25 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 173.0, 167.4, 141.4 (d, *J* = 4.8 Hz), 128.2, 127.5, 126.6, 91.9 (d, *J* = 6.3 Hz), 81.9 (*J* = 15.1 Hz), 59.3, 56.9, 55.5, 40.8, 14.4; HRMS calcd for C₁₅H₂₀O₄ 264.1362, found 264.1373, 267.1552, 270.1739.

(*E*)-Ethyl 3,5-dimethoxy-5-phenylpent-2-enoate (7a'): white solid (56 mg, 90%); ¹H NMR (500 MHz, CDCl₃) δ 7.33 (d, *J* = 7.5 Hz, 2H), 7.31–7.24 (m, 3H) 4.96 (s, 1H), 4.45 (t, *J* = 8.5 Hz, 1H), 3.49 (s, 3H), 3.20 (dd, *J* = 8.0, 14.0 Hz, 1H), 3.06 (s, 3H), 2.95 (dd, *J* = 6.5, 14.0 Hz, 1H); ¹³C NMR (125 MHz, CDCl₃) δ 171.4, 168.0, 141.2, 128.2, 127.5, 126.4, 92.7, 81.0, 56.1, 55.4, 39.1; HRMS calcd for C₁₃H₁₆O₄ 236.1049, found 236.1055.

(*E*)-Ethyl 3,5-diethoxy-5-phenylpent-2-enoate (7b): light yellow liquid (91 mg, 63%); ¹H NMR (600 MHz, CDCl₃) δ 7.32 (d, *J* = 7.8 Hz, 2H), 7.28 (t, *J* = 8.2 Hz, 2H), 7.22 (d, *J* = 7.2 Hz, 1H), 4.93 (s, 1H), 4.60 (dd, *J* = 6.4, 7.9 Hz, 1H), 4.10 (q, *J* = 7.1 Hz, 2H), 3.71 (q, *J* = 7.0 Hz, 1H), 3.62 (q, *J* = 7.0 Hz, 1H), 3.39–3.32 (m, 2H), 3.26 (dd, *J* = 8.0, 13.2 Hz, 1H), 3.00 (dd, *J* = 6.3, 13.3 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H), 1.16 (t, *J* = 7.0 Hz, 3H), 1.12 (t, *J* = 7.0 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.2, 167.5, 142.1, 128.0, 127.3, 126.7, 92.1, 79.5, 64.2, 63.7, 59.3, 40.9, 15.3, 14.4, 14.0; HRMS calcd for C₁₇H₂₄O₄ 292.1675, found 292.1668.

(*E*)-Ethyl 3,5-bis(allyloxy)-5-phenylpent-2-enoate (7c): light yellow liquid (90 mg, 57%); ¹H NMR (600 MHz, CDCl₃) δ 7.34 (d, *J* = 7.0 Hz, 2H), 7.30 (t, *J* = 7.3 Hz, 2H), 7.23 (d, *J* = 7.2 Hz, 1H), 5.84–5.78 (m, 2H), 5.27–5.09 (m, 4H), 4.98 (s, 1H), 4.68 (dd, *J* = 5.9, 8.4 Hz, 1H), 4.22–4.18 (m, 2H), 4.11 (q, *J* = 7.1 Hz, 2H), 3.90 (dd, *J* = 4.9, 13.0 Hz, 1H), 3.74 (dd, *J* = 5.8, 13.0 Hz, 1H), 3.34 (dd, *J* = 8.4, 13.5 Hz, 1H), 3.05 (dd, *J* = 5.9, 13.5 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 171.5, 167.3, 141.5, 134.9, 131.8, 128.2, 127.5, 126.8, 118.1, 116.3, 93.0, 79.3, 69.5, 68.9, 59.4, 40.7, 14.4; HRMS calcd for C₁₉H₂₄O₄ 316.1675, found 316.1668.

(*E*)-Ethyl 3,5-dimethoxy-5-p-tolylpent-2-enoate (7d): color-less liquid (123 mg, 89%); ¹H NMR (600 MHz, CDCl₃) δ 7.23 (d, *J* = 8.0 Hz, 2H), 7.12 (d, *J* = 7.8 Hz, 2H), 5.01 (s, 1H), 4.49 (dd, *J* = 5.2, 8.7 Hz, 1H), 4.12 (q, *J* = 7.1 Hz, 2H), 3.57 (s, 3H), 3.18 (s, 3H), 3.15 (dd, *J* = 5.2, 13.5 Hz, 1H), 3.07 (dd, *J* = 5.2, 13.5 Hz, 1H), 2.31 (s, 3H), 1.25 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 173.1, 167.4, 138.3, 137.1, 128.9, 126.5, 91.8, 81.7, 59.3, 56.8, 55.5, 40.8, 21.1, 14.4; HRMS calcd for C₁₆H₂₂O₄ 278.1518, found 278.1518.

(*E*)-Ethyl 5-(4-*tert*-butylphenyl)-3,5dimethoxypent-2-enoate (7e): light yellow liquid (137 mg, 86%) ; ¹H NMR (600 MHz, CDCl₃) δ 7.32 (d, *J* = 8.4 Hz, 2H), 7.26 (d, *J* = 8.3 Hz, 2H), 5.01 (s, 1H), 4.48 (dd, *J* = 5.3, 8.7 Hz, 1H), 4.12 (q, *J* = 7.1 Hz, 2H), 3.57 (s, 3H), 3.19 (s, 3H), 3.14 (dd, *J* = 5.1, 13.5 Hz, 1H), 3.06 (dd, *J* = 5.1, 13.5 Hz, 1H), 1.28 (s, 9H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 173.3, 167.4, 150.4, 138.4, 126.3, 125.4, 91.8, 81.7, 59.4, 56.9, 55.5, 40.9, 33.8, 31.3, 14.4; HRMS calcd for C₁₉H₂₈O₄ 320.1988, found 320.1993

(*E*)-Ethyl 3,5-dimethoxy-5-(4-methoxyphenyl)pent-2-enoate (7f): light yellow liquid (130 mg, 89%); ¹H NMR (600 MHz, CDCl₃) δ 7.25 (d, *J* = 8.3 Hz, 2H), 6.83 (d, *J* = 8.4 Hz, 2H), 4.98 (s, 1H), 4.46 (dd, *J* = 5.5, 8.5 Hz, 1H), 4.11 (q, *J* = 7.0 Hz, 2H), 3.77 (s, 3H), 3.54 (s, 3H), 3.16–3.13 (m, 4H), 3.06 (dd, *J* = 5.5, 13.5 Hz, 1H), 1.23 (t, *J* = 7.0 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 173.0, 167.3, 159.0, 133.4, 127.8, 113.5, 91.9, 81.4, 59.3, 56.6, 55.4, 55.1, 40.7, 14.3; HRMS calcd for C₁₆H₂₂O₅ 294.1467, found 294.1436.

Deuterated (*É*)-ethyl 3,5-dimethoxy-5-(4-methoxyphenyl)pent-2-enoate (d₃-7f): light yellow liquid (113 mg, 78%); ¹H NMR (600 MHz, CDCl₃) δ 7.26–7.24 (m, 2H), 6.84 (d, *J* = 8.6 Hz, 2H), 4.99 (d, *J* = 3.4 Hz, 1H), 4.46 (dd, *J* = 5.5, 8.5 Hz, 1H), 4.11 (q, *J* = 7.1 Hz, 2H), 3.77 (s, 3H), 3.55 (s, 1.5H), 3.16–3.13 (m, 2.5H), 3.06 (dd, *J* = 5.4, 13.5 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 173.1, 167.4, 159.0, 133.4 (d, J = 5.1 Hz), 127.0, 113.5, 91.8 (d, J = 6 Hz), 81.4 (d, J = 15.3 Hz), 59.4, 56.6, 55.5, 55.2, 40.7, 14.4; HRMS calcd for C₁₆H₂₂O₅ 294.1467, found 294.1473, 297.1641, 300.1809.

(*E*)-Ethyl 3,5-dimethoxy-5-(2-methoxyphenyl)pent-2-enoate (7g): light yellow liquid; (107 mg, 73%); ¹H NMR (600 MHz, CDCl₃) δ 7.39 (d, *J* = 7.5 Hz, 1H), 7.19 (t, *J* = 7.8 Hz, 1H), 6.93 (t, *J* = 7.4 Hz, 1H), 6.81 (d, *J* = 8.1 Hz, 1H), 4.99 (s, 1H), 4.96 (dd, *J* = 5.1, 8.7 Hz, 1H), 4.07 (q, *J* = 7.1 Hz, 2H), 3.78 (s, 3H), 3.57 (s, 3H), 3.49 (dd, *J* = 5.1, 14.0 Hz, 1H), 3.19 (s, 3H), 2.90 (dd, *J* = 5.1, 14.0 Hz, 1H), 1.21 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.7, 167.3, 157.0, 129.5, 128.2, 127.1, 120.6, 110.1, 92.1, 74.9, 59.2, 56.8, 55.3, 55.2, 38.4, 14.3; HRMS calcd for C₁₆H₂₂O₅ 294.1467, found 294.1436.

(*E*)-Ethyl 5-(4-chlorophenyl)-3,5-dimethoxypent-2-enoate (7h): yellow liquid (110 mg, 74%); ¹H NMR (600 MHz, CDCl₃) δ 7.27–7.26 (m, 4H), 5.00 (s, 1H), 4.48 (dd, J = 5.9, 8.1 Hz, 1H), 4.10 (q, J = 7.1 Hz, 2H), 3.55 (s, 3H), 3.18 (s, 3H), 3.10–3.08 (m, 2H), 1.24 (t, J = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.6, 167.3, 139.9, 133.2, 128.4, 128.0, 92.0, 81.3, 59.4, 56.9, 55.5, 40.7, 14.2; HRMS calcd for C₁₅H₁₉ClO₄ 298.0972, found 298.0973.

(*E*)-Ethyl 5-(4-bromophenyl)-3,5-dimethoxypent-2-enoate (7i): yellow liquid (133 mg, 78%); ¹H NMR (600 MHz, CDCl₃) δ 7.42 (d, *J* = 8.8 Hz, 2H), 7.22 (d, *J* = 8.4 Hz, 2H), 5.00 (s, 1H), 4.47 (dd, *J* = 5.2, 8.7 Hz, 1H), 4.10 (q, *J* = 7.1 Hz, 2H), 3.56 (s, 3H), 3.18 (s, 3H), 3.08 (dd, *J* = 1.7, 7.9 Hz, 2H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.6, 167.3, 140.4, 131.4, 128.4, 121.4, 92.1, 81.4, 59.5, 56.9, 55.5, 40.7, 14.4; HRMS calcd for C₁₅H₁₉BrO₄ 342.0467, found 342.0461.

(*E*)-Ethyl 5-(furan-3-yl)-3,5-dimethoxypent-2-enoate (7j): light yellow liquid (91 mg, 72%); ¹H NMR (600 MHz, CDCl₃) δ 7.35 (d, *J* = 3.6 Hz, 1H), 7.34 (d, *J* = 3.5 Hz, 1H), 6.41 (s, 1H), 5.01 (s, 1H), 4.53 (dd, *J* = 6.0, 8.2 Hz, 1H), 4.12 (q, *J* = 7.2 Hz, 2H), 3.57 (s, 3H), 3.25 (dd, *J* = 5.2, 13.5 Hz, 1H), 3.22 (s, 3H), 3.11 (dd, *J* = 6.0, 13.5 Hz, 1H), 1.24 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.7, 167.4, 143.1, 140.1, 125.3, 108.9, 92.1, 73.7, 59.4, 56.3, 55.5, 38.9, 14.4; HRMS calcd for C₁₃H₁₈O₅ 254.1154, found 254.1145.

(*E*)-Ethyl 3,5-dimethoxy-5-(thiophen-3-yl)pent-2-enoate (7k): light yellow liquid (103 mg, 77%); ¹H NMR (600 MHz, CDCl₃) δ 7.24 (dd, *J* = 3.0, 5.1 Hz, 1H), 7.16 (d, *J* = 2.9 Hz, 1H), 7.08 (d, *J* = 5.9 Hz, 1H), 5.00 (s, 1H), 4.63 (dd, *J* = 5.7, 8.4 Hz, 1H), 4.11 (q, *J* = 7.2 Hz, 2H), 3.55 (s, 3H), 3.24–3.20 (m, 4H), 3.11 (dd, *J* = 5.8, 13.6 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.8, 167.3, 142.7, 126.1, 125.6, 121.8, 91.9, 77.6, 59.4, 56.7, 55.5, 39.8, 14.4; HRMS calcd for C₁₃H₁₈O₄S 270.0926, found 270.0922.

(*É*)-Ethyl 5-(benzofuran-3-yl)-3,5-dimethoxypent-2-enoate (7l): light yellow liquid (101 mg, 67%); ¹H NMR (500 MHz, CDCl₃) δ 7.51 (d, *J* = 7.5 Hz, 1H), 7.45 (d, *J* = 8.0 Hz, 1H), 7.23 (t, *J* = 7.5 Hz, 1H), 7.18 (t, *J* = 7.5 Hz, 1H), 6.67 (s, 1H), 5.05 (s, 1H), 4.73 (dd, *J* = 6.0, 8.5 Hz, 1H), 4.12 (q, *J* = 7.5 Hz, 2H), 3.55–3.51 (m,4H), 3.31 (s, 3H), 3.22 (dd, *J* = 6.0, 14.0 Hz, 1H), 1.24 (t, *J* = 7.5 Hz, 3H); ¹³C NMR (125 MHz, CDCl₃) δ 171.8, 167.3, 156.3, 155.0, 127.9, 124.1, 122.6, 120.9, 111.4,104.6, 92.4, 74.8, 59.5, 56.9, 55.6, 36.9, 14.3; HRMS calcd for C₁₇H₂₀O₅ 304.1311, found 304.1302.

(*E*)-Ethyl 3,5-dimethoxy-5-(1-tosyl-1*H*-indol-3-yl)pent-2enoate (7m): yellow semisolid (231 mg, 71%); ¹H NMR (600 MHz, CDCl₃) δ 7.92 (d, *J* = 8.3 Hz, 1H), 7.75 (d, *J* = 7.8 Hz, 1H), 7.71 (d, *J* = 8.3 Hz, 2H), 7.50 (s, 1H), 7.27 (t, *J* = 7.8 Hz, 1H), 7.22–7.17 (m, 3H), 4.99 (s, 1H), 4.44 (dd, *J* = 5.2, 8.6 Hz, 1H), 4.09 (q, *J* = 7.1 Hz, 2H), 3.47 (s, 3H), 3.42 (dd, *J* = 8.3, 13.5 Hz, 1H), 3.30 (s, 3H), 3.17 (dd, *J* = 6.0, 13.5 Hz, 1H), 2.20 (s, 3H), 1.22 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.3, 167.3, 144.8, 135.4, 135.2, 129.8, 129.2, 126.7, 124.7, 123.9, 123.2, 122.6, 120.9, 113.5, 92.2, 74.9, 59.5, 56.6, 55.4, 38.4, 21.5, 14.3; HRMS calcd for C₂₄H₂₇NO₆S 457.1559, found 457.1571.

(*E*)-Ethyl 5-(benzo[*b*]thiophene-3-yl)-3,5-dimethoxypent-2enoate (7n): light yellow liquid (118 mg, 74%); ¹H NMR (600 MHz, CDCl₃) δ 7.79 (d, *J* = 7.8 Hz, 1H), 7.69 (d, *J* = 8.2 Hz, 1H), 7.30–7.26 (m, 2H), 7.21 (s, 1H), 5.05 (s, 1H), 4.88 (dd, *J* = 5.6, 8.4 Hz, 1H), 4.12 (q, *J* = 7.1 Hz, 2H), 3.57 (s, 3H), 3.35 (dd, *J* = 5.2, 13.7

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Hz, 1H), 3.30 (s, 3H), 3.24 (dd, J = 5.6, 13.6 Hz, 1H), 1.24 (t, J = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.1, 167.3, 146.3, 139.7, 139.4, 124.0 (CH × 2), 123.3, 122.5, 121.7, 92.3, 77.9, 59.5, 56.9, 55.5, 40.7, 14.3; HRMS calcd for C₁₇H₂₀O₄S 320.1082, found 320.1091.

(2*E*,6*E*)-Ethyl 3,5-dimethoxy-7-phenylhepta-2,6-dienoate (70): light yellow liquid (124.7 mg, 86%); ¹H NMR (600 MHz, CDCl₃) δ 7.36 (d, *J* = 7.3 Hz, 2H), 7.28 (t, *J* = 7.4 Hz, 2H), 7.21 (d, *J* = 7.3 Hz, 1H), 6.51 (d, *J* = 15.9 Hz, 1H), 6.09 (dd, *J* = 7.9, 15.9 Hz, 1H), 5.03 (s, 1H), 4.13-4.07 (m, 3H), 3.58 (s, 3H), 3.30 (s, 3H), 3.16 (dd, *J* = 7.4, 13.2 Hz, 1H),), 3.01 (dd, *J* = 6.6, 13.2 Hz, 1H), 1.24 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.5, 167.3, 136.5, 132.1, 129.4, 128.4, 127.6, 126.4, 92.2, 80.5, 59.3, 56.4, 55.4, 38.3, 14.3; HRMS calcd for C₁₇H₂₂O₄ 290.1518, found 290.1515.

(2*E*,6*E*)-Ethyl 3,5-Dimethoxy-7-(thiophene-2-yl)hepta-2,6-dienoate (7p): light yellow liquid (128 mg, 87%); ¹H NMR (600 MHz, CDCl₃) δ 7.12 (dd, *J* = 4.1,2.5 Hz, 1H), 6.92 (t, *J* = 1.5 Hz, 2H), 6.64 (d, *J* = 15.7 Hz, 1H), 5.92 (dd, *J* = 7.8,15.7 Hz, 1H), 5.03 (s, 1H), 4.11(q, *J* = 7.2 Hz, 2H), 4.03(q, *J* = 6.7 Hz, 1H), 3.59 (s, 3H), 3.29 (s, 3H), 3.14 (dd, *J* = 7.5, 13.2 Hz, 1H),), 2.96 (dd, *J* = 6.4, 13.2 Hz, 1H), 1.24 (t, *J* = 7.2 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 172.5, 167.4, 141.8, 129.1, 127.3, 125.7, 125.1, 124.3, 92.2, 80.2, 59.4, 56.5, 55.5, 38.3, 14.3; HRMS calcd for C₁₅H₂₀O₄S 296.1082, found 296.1081.

(*E*)-*tert*-Butyl 3,5-dimethoxy-5-phenylpent-2-enoate (8b): light yellow liquid (99 mg, 82%); ¹H NMR (600 MHz, CDCl₃) δ 7.31–7.28 (m, 4H), 7.24 (t, *J* = 8.4 Hz, 1H), 4.91 (s, 1H), 4.48 (dd, *J* = 5.9, 8.3 Hz, 1H), 3.52 (s, 3H), 3.19 (s, 3H), 3.15 (dd, *J* = 8.3, 13.6 Hz, 1H), 3.08 (dd, *J* = 5.8, 13.6 Hz, 1H), 1.48 (s, 9H); ¹³C NMR (150 MHz, CDCl₃) δ 171.6, 166.7, 141.3, 128.2, 127.5, 126.7, 93.8, 81.9, 79.2, 56.8, 55.3, 40.3, 28.3; HRMS calcd for C₁₇H₂₄O₄ 292.1675, found 292.1668.

(*E*)-Allyl 3,5-dimethoxy-5-phenylpent-2-enoate (8c): light yellow liquid (96 mg, 76%); ¹H NMR (600 MHz, CDCl₃) δ 7.32–7.28 (m, 4H), 7.23 (d, *J* = 7.4 Hz, 1H), 5.95–5.89 (m, 1H), 5.29 (dd, *J* = 1.5, 17.2 Hz, 1H), 5.20 (dd, *J* = 1.3, 10.4 Hz, 1H), 5.04 (s, 1H), 4.57–4.56 (m, 2H), 4.50 (dd, *J* = 5.2, 8.7 Hz, 1H), 3.56 (s, 3H), 3.20–3.14 (m, 4H), 3.08 (dd, *J* = 5.2, 13.6 Hz, 1H); ¹³C NMR (150 MHz, CDCl₃) δ 173.5, 166.9, 141.3, 132.8, 128.2, 127.6, 126.6, 117.7, 91.6, 81.9, 64.3, 56.9, 55.6, 40.9; HRMS calcd for C₁₆H₂₀O₄ 276.1362, found 276.1364.

(*E*)-Benzyl 3,5-dimethoxy-5-phenylpent-2-enoate (8d): light yellow liquid (82 mg, 74%); ¹H NMR (600 MHz, CDCl₃) δ 7.34–7.28 (m, 10H), 5.19–5.08 (m, 2H), 5.07 (s, 1H), 4.51 (dd, *J* = 5.2, 8.7 Hz, 1H), 3.56 (s, 3H), 3.20 (s, 3H), 3.18–3.06 (m, 2H); ¹³C NMR (150 MHz, CDCl₃) δ 173.6, 167.2, 141.3, 136.6, 128.6 128.5, 128.4, 128.3, 128.2, 128.1, 91.7, 81.9, 65.4, 56.9, 55.6, 40.9; HRMS calcd for C₂₀H₂₂O₄ 326.1518, found 326.1512.

(*E*)-6,8-Dimethoxy-8-phenyloct-5-en-4-one (8e): light yellow liquid (95 mg, 72%); ¹H NMR (600 MHz, CDCl₃) δ 7.34 (d, *J* = 7.3 Hz, 2H), 7.31 (t, *J* = 7.8 Hz, 2H), 7.22 (d, *J* = 7.3 Hz, 1H), 5.43 (s, 1H), 4.51 (dd, *J* = 5.1, 8.8 Hz, 1H), 3.58 (s, 3H), 3.18 (s, 3H), 3.13 (dd, *J* = 5.1, 13.3 Hz, 1H), 3.03 (dd, *J* = 8.8, 13.3 Hz, 1H), 2.37–2.34 (m, 2H), 1.62–1.57 (m, 2H), 0.90 (t, *J* = 7.3 Hz, 3H); ¹³C NMR (150 MHz, CDCl₃) δ 199.2, 172.7, 141.5, 128.2, 127.5, 126.6, 99.5, 81.9, 56.9, 55.4, 46.7, 41.6, 18.3, 13.8; HRMS calcd for C₁₆H₂₂O₃ 262.1569, found 262.1576.

(*E*)-3,5-Dimethoxy-5-phenyl-1-(thiophene-2-yl)pent-2-en-1one (8f): light yellow liquid (100 mg, 85%); ¹H NMR (500 MHz, CDCl₃) δ 7.54 (d, *J* = 5.0 Hz, 1H), 7.40 (d, *J* = 7.5 Hz, 1H), 7.30 (t, *J* = 7.5 Hz, 2H), 7.21–7.07 (m, 4H), 6.06 (s, 1H), 4.64 (dd, *J* = 5.0, 8.5 Hz, 1H), 3.70 (s, 3H), 3.31–3.13 (m, 5H); ¹³C NMR (125 MHz, CDCl₃) δ 181.9, 174.9, 145.0, 141.4, 134.1, 132.2, 128.2, 128.1, 126.6, 124.2, 96.4, 81.8, 56.9, 55.7, 42.3; HRMS calcd for C₁₇H₁₈O₃S 302.0977, found 302.0974.

(*E*)-3,5-Dimethoxy-1,5-diphenylpent-2-en-1-one (8g): light yellow liquid (103 mg, 86%); ¹H NMR (400 MHz, CDCl₃) δ 7.95 (d, *J* = 8.4 Hz, 1H), 7.86 (d, *J* = 7.8 Hz, 1H), 7.54–7.40 (m, 6H), 7.34–7.30 (m, 2H), 6.18 (s, 1H), 4.67 (dd, *J* = 7,2, 13.2 Hz, 1H), 3.71 (s, 3H), 3.28–3.12 (m, 5H); ¹³C NMR (100 MHz, CDCl₃) δ 190.2, 174.9, 141.5, 128.5, 128.3, 128.2, 127.9, 127.6, 127.5, 127.0, 96.8, 81.9, 56.9, 55.7, 42.2; HRMS calcd for C₁₉H₂₀O₃ 296.1412, found 296.1410.

ASSOCIATED CONTENT

S Supporting Information

¹H and ¹³C NMR spectra of compounds 7a-p, d_1 -7a, d_3 -7a, d_3 -7f, 8b-g, and crystallographic data for compound 7a'. This material is available free of charge via the Internet at http:// pubs.acs.org.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) Selected reviews: (a) Doyle, M. P.; McKervy, M. A.; Ye, T. Modern Catalytic Methods for Organic Synthesis with Diazo Compounds; John Wiley & Sons: New York, 1998. (b) Padwa, A.; Weingarten, M. D. Chem. Rev. 1996, 96, 223. (c) Davies, H. M. L.; Denton, J. R. Chem. Soc. Rev. 2009, 38, 3061. (d) Doyle, M. P.; Duffy, R.; Ratnikov, M.; Zhou, L. Chem. Rev. 2010, 110, 704. (e) Davies, H. M. L.; Morton, D. Chem. Soc. Rev. 2011, 40, 1857. (f) Zhang, Z.; Wang, J. Tetrahedron 2008, 64, 6577.

(2) For selected examples for carbocyclic cycloadducts, see:
(a) Deng, L.; Giessert, A. J.; Gerlitz, O. O.; Dai, X.; Diver, S. T.; Davies, H. M. L. J. Am. Chem. Soc. 2005, 127, 1342. (b) Davies, H. M. L. Adv. Cycloaddition 1999, 5, 119. (c) Davies, H. M. L.; Xing, B.; Kong, N.; Stafford, D. G. J. Am. Chem. Soc. 2001, 123, 7461. (d) Davies, H. M. L.; Clark, T. J.; Smith, H. D. J. Org. Chem. 1991, 56, 3819. (e) Liu, Y.; Bakshi, K.; Zavalij, P.; Doyle, M. P. Org. Lett. 2010, 12, 4304. (f) Olson, J. P.; Davies, H. M. L. Org. Lett. 2008, 10, 573. (3) For oxacyclic cycloadducts, see: (a) Xu, X.; Hu, W.-H.; Zavalij, P. Y.; Doyle, M. P. Angew. Chem., Int. Ed. 2011, 50, 11152. (b) Doyle, M. P.; Hu, W.; Timmons, D. J. Org. Lett. 2001, 3, 3741.

(4) For azacyclic cycloadducts, see selected reviews: (a) Doyle, M. P.;
Yan, M.; Hu, W.; Gronenberg, L. J. Am. Chem. Soc. 2003, 125, 4692.
(b) Barluenga, J.; Lonzi, G.; Riesgo, L.; López, L. A.; Tomas, M. J. Am. Chem. Soc. 2010, 132, 13200. (c) Yang, M.; Jacobsen, N.; Hu, W.;
Gronenberg, L. S.; Doyle, M. P.; Colyer, J. T.; Bykowski, D. Angew. Chem., Int. Ed. 2004, 43, 6713. (d) Wang, X.; Xu, X.; Zavalij, P.; Doyle, M. P. J. Am. Chem. Soc. 2011, 133, 16402. (e) Lian, Y.; Davies, H. M. L. J. Am. Chem. Soc. 2010, 132, 440. (f) Xu, X.; Ratnikov, M. O.; Zavalij, P. Y.; Doyle, M. P. Org. Lett. 2011, 13, 6122.

(5) (a) Doyle, M. P.; Kundu, K.; Russel, A. E. Org. Lett. 2005, 7, 5171–5174. (b) Liu, Y.; Zhang, Y.; Jee, N.; Doyle, M. P. Org. Lett. 2008, 10, 1605–1608.

(6) Jadhav, A. M.; Pagar, V. V.; Liu, R.-S. Angew. Chem., Int. Ed. 2012, 51, 11809.

(7) Reviews for the Povarov reactions, see: (a) Povarov, L. S. *Russ. Chem. Rev.* **1967**, *36*, 656. (b) Kouznetsov, V. V. *Tetrahedron* **2009**, *65*, 2721. (c) Bello, D.; Ramón, R.; Lavilla, R. *Curr. Org. Chem* **2010**, *14*, 332. (d) McCarrick, M. A.; Wu, Y. D.; Houk, K. N. J. Org. Chem. **1993**, *58*, 3330.

(8) The crystallographic data of unsaturated acid 7a' was provided in Supporting Information.

(9) For reviews on the Prins cyclization, see: (a) Adams, D. R.; Bhatnagar, S. P. Synthesis 1977, 661. (b) Snider, B. B. In The Prins Reactions and Carbonyl-ene Reactions; Trost, B. M., Fleming, I., Heathcock, C. H., Eds.; Pergamon Press: New York, 1991; Vol. 2, pp 527. (c) Overman, L. E.; Pennington, L. D. J. Org. Chem. 2003, 68, 7143.

(10) For selected examples, see: (a) Jasti, R.; Rychnovsky, S. D. J. Am. Chem. Soc. 2006, 128, 13640. (b) Lee, C.-H. A.; Loh, T.-P.

The Journal of Organic Chemistry

Tetrahedron Lett. 2006, 47, 1641–1644. (c) Overman, L. E.; Veltthuisen, E. J. J. Org. Chem. 2006, 71, 1581–1587. (d) Bolla, M. L.; Patterson, B.; Rychnovsky, S. D. J. Am. Chem. Soc. 2005, 127, 16044. (e) Barry, C. S. J.; Crosby, S. R.; Harding, J. R.; Hughes, R. A.; King, C. D.; Parker, G. D.; Willis, C. L. Org. Lett. 2003, 5, 2429. (f) Miranda, P. O.; Diaz, D. D.; Padron, J. I.; Bermejo, J.; Martin, V. S. Org. Lett. 2003, 5, 1979.

(11) For the preferable *E*-alkene selectivity of gold carbenes, see ref 5 and selected examples: (a) Lu, B.; Li, C.; Zhang, L. *J. Am. Chem. Soc.* 2010, 132, 14070. (b) Li, G.; Zhang, G.; Zhang, L. *J. Am. Chem. Soc.* 2008, 130, 3740. (c) Wang, S.; Zhang, L. *Org. Lett.* 2006, 8, 4585. (d) Davies, P. W.; Cremonesi, A.; Martin, N. *Chem. Commun.* 2011, 47, 379.

(12) For the Z-alkene selectivity of rhodium carbenes, see selected examples: (a) Taber, D. F.; Joshi, P. V. J. Org. Chem. 2004, 69, 4276.
(b) Davies, H. M. L.; Hedley, S. J. Chem. Soc. Rev. 2007, 36, 1109.
(c) Ota, K.; Chatani, N. Chem. Commun. 2008, 2906. (d) Lian, Y.; Davies, M. L. H. Org. Lett. 2009, 12, 924. (e) Lian, Y.; Davies, H. M. L. Org. Lett. 2012, 14, 1934.

(13) Prior to 1,2-hydride shift, a preferable conformation is shown in state **D**, in which the front C–H bond is aligned with the rear empty p-orbital with its small Rs group lying on the same side with metal (M). After this 1,2-shift, the E- and Z-selectivity of products depends on metal, as depicted by E'. Dimeric Rh has a restricted rotation barrier to show the Z-selectivity with a small rotation (60°). In contrast, gold prefers thermodynamically favored *E*-configured products because of its facile rotation with 120°.



(14) (a) Davis, H. M. L.; Hougland, P. W.; Cantrell, W. R., Jr. Synth. Commun. 1992, 22, 971. (b) Padwa, A.; Kulkarni, Y. S.; Zhang, Z. J. J. Org. Chem. 1990, 55, 4144. (c) Davies, H. M. L.; Ahmed, G.; Churchill, M. R. J. Am. Chem. Soc. 1996, 118, 10774.