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Synthesis of new class of spirocarbocycle derivatives by multicomponent domino reaction and their evaluation for antimicrobial, anticancer activity and Molecular docking studies

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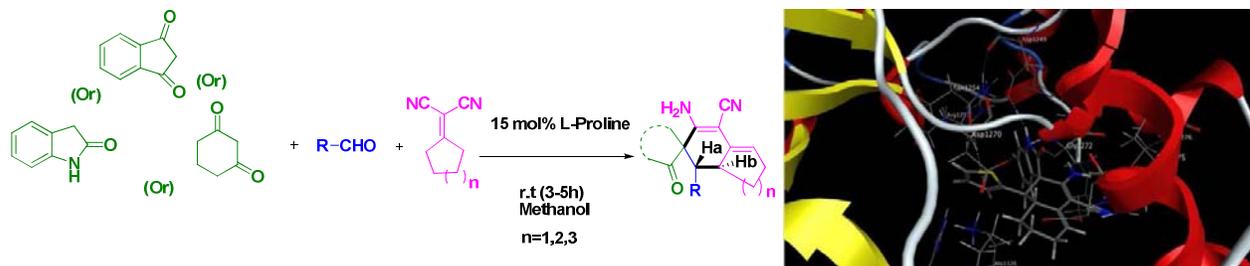


**Highlights**

1. One pot synthesis of spirocarbocycles was developed using L-proline.
2. Ordinary reaction conditions, wide substrate scope, excellent yield (72-90%).
3. Most of the compounds exhibited moderate to good activity against bacteria.
4. Docking study of ligands with receptor was further supported by *in vitro* results.
5. Compound 6i showed very good affinity towards the ALK receptor.

## Graphical Abstract

Synthesis of spirocarbocycles was achieved by a three component reaction of cyclic nucleophiles, vinyl malononitriles and aldehydes with variable substitution patterns.



1     **Synthesis of new class of spirocarbocycle derivatives by multicomponent**  
2     **domino reaction and their evaluation for antimicrobial, anticancer activity**  
3                     **and Molecular docking studies**

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12  
13    **Abstract:**

14    A series of 25 new spirocarbocycles were synthesized by a three component reaction that  
15    involves few cyclic nucleophiles, vinyl malononitriles and aldehydes with variable  
16    substitution patterns. All the synthesized compounds were evaluated for their antimicrobial  
17    activity and the compounds showed significant activity. Synthesized compounds **4c**, **4i** and **6i**  
18    showed good anticancer activity against A549 cancer cell line. Molecular docking studies  
19    indicated that compound **4i** had the greatest affinity for DNA gyrase receptor than others and  
20    compound **6i** had the greatest affinity for anaplastic lymphoma kinase (ALK) receptor. These  
21    compounds can be better therapeutic agents for microbial and cancer cell lines.

22    *Keywords:* Spirocarbocycle, multicomponent reaction, vinylogous Michael addition,  
23    antimicrobial activity, anticancer activity, Molecular docking.

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## 1 **1. Introduction**

2 Spiro compounds exhibit a broad spectrum of important bioactivities such as antitumor [1],  
3 antidiabetic [2], antibacterial [3], antitubercular [4], antiinflammatory [5] activities etc [6]  
4 due to their interesting structures. Spirooxindoles are privileged medicinal scaffolds that are  
5 found in natural and unnatural biologically active compounds [Figure 1]. The key paradigm  
6 of modern drug discovery and the goal of synthetic organic chemistry are being achieved by  
7 the rapid assembly of structurally different compounds. One of the most potential strategies  
8 for the synthesis and library production of these spiro compounds is achieved through  
9 multicomponent reactions (MCRs) [7]. They have been used widely for carbon-carbon bond  
10 formation by synthetic chemists. They also offer an eloquent tool for the one pot synthesis of  
11 distinct and complex molecule as well as small and drug like heterocycles.

12 <<Figure 1>>

13 Electron-deficient dicyanoalkenes have been reported to behave as good hydride acceptors in  
14 conjugate reduction reactions [8] and also act as versatile direct vinylogous donors in  
15 asymmetric Michael addition reactions with excellent chemo- and stereo selectivity [9]. The  
16 strong electron-withdrawing groups activate the  $\gamma$ -position and further it undergoes Michael  
17 addition. Vinylogous Michael addition was reported as a key step in the preparation of many  
18 spiro compounds [10].

19 The methodologies that are accessible for the synthesis of spirocyclic compounds are  
20 alkylations [11], rearrangement reactions [12], cycloadditions [13], transition metal catalyzed  
21 reactions [14] and cleavage of bridged systems [15]. The synthetic and pharmacological  
22 importance of spiro compounds are anchored in their spirocyclic motifs. The captivating  
23 framework of spirooxindoles increased the thirst of chemists in preparing these compounds

1 by numerous methods. Our group has always been interested on the synthesis of novel  
2 spirooxindole compounds and screening for their biocidal activity [16].

## 3 **2. Result and Discussion**

### 4 **2.1 Chemistry**

5 The biological importance of the spiro compounds directed us to synthesize spiro-  
6 carbocycles. In continuation of our studies in the area of multicomponent domino reaction  
7 using vinylogous Michael addition methodology [17] we herein report the one pot  
8 multicomponent reaction of vinyl malononitriles with different chalcones and their  
9 antimicrobial, anticancer activity and molecular docking results of ligands to the DNA gyrase  
10 and ALK receptor. To the best of our knowledge, there have been no reports for the synthesis  
11 of the spiro moieties derived from oxindole, indanedione and 1,3-cyclohexanedione  
12 chalcones. In the present study the vinyl malononitriles undergo vinylogous Michael addition  
13 with the *in situ* generated chalcones followed by intramolecular nucleophilic addition and  
14 isomerization respectively to afford spirocarbocycles in moderate to high yields (**Scheme 1**).

15 <<**Scheme 1**>>

16 We initiated our study by performing the reaction of cyclohexylidene malononitrile  
17 **3a** with *in situ* generated indanedione chalcone without any base catalyst and using methanol  
18 as solvent which afforded compound **4a** in 40% yield (Table 1, entry 1). Then the  
19 investigatory experiment was performed to improve the yield of the product by varying  
20 reaction conditions *viz.* solvent, temperature and base catalyst. All the reaction products are  
21 purified by recrystallization using ethanol as a solvent. The observations (**Table 1**) led us to  
22 the conclusion that the base has an obvious effect on the reaction. Among the selected bases  
23 (Table 1, entries 2-6), commercially available organocatalyst L-proline was proven to be the  
24 most suitable (Table 1, entries 6-9) as it gives single diastereomer. The reaction of 1, 3-

1 indandione **1** with aldehyde **2a** and alkylidene malononitrile **3a** in the presence of L-proline  
2 was completely diastereoselective in affording only the trans diastereomer whereas the other  
3 bases yield mixture of diastereomers.

4 Several solvents were investigated (Table 1, entries 6, 7, 10-12) and methanol was the best  
5 one (Table 1, entries 7-9), although ethanol also produced comparable results (Table 1, entry  
6 6). The stoichiometry ratio 15 mol% of the catalyst was chosen to be the best for performing  
7 the reaction since further increase had no impact on the yield of the reaction (Table 1 entries  
8 7-9). Thus the best result was obtained when 1, 3-indandione **1** and aldehyde **2a** were stirred  
9 for ten minutes in the presence of 15 mol% of L-proline as a catalyst for the *in situ*  
10 generation of chalcone followed by the addition of cyclohexylidene malononitrile **3a** to  
11 provide a precipitate which was filtered off and recrystallized from ethanol to yield the  
12 desired product **4a** in 89% yield.

13 << **Table 1** >>

14 To explain the mechanism of this tandem reaction, a postulated reaction course is depicted in  
15 (**Scheme 2**). In the first step 1, 3-indandione **1** undergoes condensation reaction with  
16 aldehyde by the removal of water molecule and affords chalcone **1a**. In the second step the  
17 proton of vinyl malononitrile **3a** is removed by the mild base to furnish vinylogous carbanion  
18 which attacks the activated double bond of chalcone **1a** forming the intermediate **1b**. Among  
19 the two paths the first one shows the steric repulsion and fails to afford a product. The  
20 nucleophilic addition to the nitrile carbon from the least hindered side through path B results  
21 in the formation of intermediate **1c** which on isomerisation gives only one spirocarbocyclic  
22 diastereomer.

23 << **Scheme 2** >>

1 With these results in hand, we then investigated the substrate scopes and limitations of the  
2 synthesis of spirocarbocycles. First we started the reaction with 1,3-indandione by varying  
3 aldehydes and alkylidene malononitrile to afford spirocarbocycles (**Scheme 3**) the results are  
4 listed in **Table 2**. Aromatic aldehydes having electron donating groups like 4-methyl and 4-  
5 methoxy benzaldehydes (Table 2, entries 1-5) reacted at faster rates compared with those that  
6 substituted with electron withdrawing groups like 2-fluoro and 4-chloro benzaldehydes  
7 (Table 2, entries 6-10). Various aromatic aldehydes such as naphthaldehyde were also  
8 examined (Table 2, entry 11). Besides, our methodology has been used successfully in order  
9 of acid and base sensitive materials such as heteroaromatic aldehydes and corresponding  
10 spirocarbocycles were obtained in excellent yields without the formation of any by products  
11 (Table 2, entries 12, 13). As it is clear from the obtained results, the presented methodology  
12 can be used in oxygen and sulphur containing heteroaromatic aldehydes. Thus the reaction  
13 was found to be general and has a broad scope due to its applicability to a variety of  
14 substrates and the products **4a-n** were isolated in excellent yields (79-90%) under milder  
15 reaction conditions. The results are summarized in **Table 2**.

16 <<**Scheme 3**>>

17 <<**Table 2**>>

18 To extend the scope of this protocol for this multicomponent reaction, studies were continued  
19 with other chalcones like oxindole chalcones using the same protocol which resulted in the  
20 formation of corresponding spiroxindoles (**Scheme 4**). The reaction provided spiroxindoles  
21 in good yields (74-86%) and the results are shown in **Table 3**. The promising results  
22 prompted us to further explore the scope of this protocol. We investigated the reaction of  
23 cyclohexylidene malononitrile with 1, 3-cyclohexanedione chalcone. The spirocarbocycles  
24 (**Scheme 5**) were obtained in moderate yields (74-76%) and the results are mentioned in

1 **Table 4.**

2 <<Scheme 4>>

3 <<Table 3>>

4 The structure of all prepared compounds were elucidated with the aid of IR,  $^1\text{H}$  and  $^{13}\text{C}$   
5 NMR, Mass spectroscopy and elemental analysis data were discussed for a representative  
6 compound **6a**. The IR spectrum of **6a** exhibited a sharp peak at  $3422\text{ cm}^{-1}$  which corresponds  
7 to the -NH stretching of -NH<sub>2</sub> group and peaks at  $2200$  and  $1702\text{ cm}^{-1}$  corresponded to the  
8 nitrile and amide carbonyls respectively. The  $^1\text{H}$  NMR spectra of **6a** showed two singlets at  $\delta$   
9  $5.38$  and  $10.35$  for the -NH<sub>2</sub> and -NH protons (D<sub>2</sub>O exchangeable) respectively, clearly  
10 indicating the incorporation of both moieties in the product. The doublet appearing for the H<sub>a</sub>  
11 proton of the product **6a** showed a coupling constant value of  $12.4\text{ Hz}$  indicative of the trans  
12 stereochemistry between the H<sub>a</sub> and H<sub>b</sub> protons which is also clearly evidenced from the  
13 single crystal X-ray analysis of compound **6a** (**Figure 2**) [18] which further supports the  $^1\text{H}$   
14 NMR spectroscopy. In  $^{13}\text{C}$  NMR spectra, the spiro carbon atom displayed a signal at  $\delta 82.0$   
15 and the amide carbonyl carbon atom resonated at  $\delta 176.6$ . The mass spectra also exhibited a  
16 distinguishing peak at  $m/z 382\text{ [M+H]}^+$ . This shows that the reaction of oxindole **5** with  
17 aldehyde **2a** and alkylidene malononitrile **3a** is completely diastereoselective in affording  
18 only the trans diastereomer. The structures of all spirocarbocycles were consistent with the  
19 above mentioned data.

20 <<Figure 2>>

21 <<Table 4>>

22

## 1 2.2. Pharmacology

### 2 2.2.1 Antimicrobial activity

3 In the present work, the antimicrobial activities of 25 synthesized compounds were screened  
4 against nine bacteria and two fungi using *in vitro* disc diffusion method. The results revealed  
5 that most of the synthesized compounds exhibited antimicrobial activities against *E.*  
6 *aerogens*, *S. epidermidis*, *S. aureus* (MRSA), *S. typhimurium*, *K. pneumonia* and *M. luteus*.  
7 The results are summarized in **Table 5** and **Figure 3**. Compounds **4b**, **4c**, **4i** and **4k** have  
8 shown excellent activities more than the standard drug against both gram-positive and gram-  
9 negative bacteria at 1mg/disc. Moreover the compounds **4a**, **4i**, **4l**, **4m**, **6a**, **6b**, **6i**, **6j** and **8b**  
10 showed good antibacterial activity over the others. All tested compounds showed moderate  
11 antifungal activity against *C. albican* and *M. pachydermatis*.

12 <<Table 5>>

13 <<Figure 3>>

14 The Minimum Inhibitory Concentration (MIC) values of active compounds against bacteria  
15 are given in **Table 6** and **Figure 4**. Significant MIC values were observed against gram  
16 positive and gram negative bacteria. The results revealed that the spirocarbocycles **4c**, **4i**, **4k**  
17 and **6i** have shown good antibacterial activity against tested organisms. Among all tested  
18 compounds 4-methyl substituted aromatic ring containing compound **4c** has shown  
19 significant MIC values against *K. pneumonia*, *P. vulgaris*, *S. flexneri*, 4-chloro substituted  
20 aromatic ring containing compound **4i** is potent against *S. aureus* (MRSA), *P. vulgaris*, *S.*  
21 *flexneri* and *M. luteus*. The naphthyl group containing compound **4k** is active against *S.*  
22 *aureus* (MRSA), *S. flexneri* and *M. luteus*. The 2-thiophenyl containing spirooxindole  
23 compound **6i** showed significant MIC values against *S. aureus* (MRSA), *P. vulgaris* and *S.*  
24 *flexneri*.

25 <<Table 6>>

1 <<Figure 4>>

2 **2.2.2. Anticancer activity**

3 Anti cancer activity studies have been performed for the synthesized compounds **4i**, **4c** and  
4 **6i**. They showed potent anticancer activity *in vitro* against A549 lung adenocarcinoma cancer  
5 cell line. Compound **4c** showed 59.6% activity at the dose of 50 µg/mL with IC<sub>50</sub> value of 50  
6 µg/mL. Compound **4i** showed 78.8% activity at the dose of 50 µg/mL with IC<sub>50</sub> value of 30  
7 µg/mL. Compound **6i** showed 83.9% activity at the dose of 50 µg/mL with IC<sub>50</sub> value of 20  
8 µg/mL. All concentrations used in the experiment decreased the cell viability significantly  
9 (P<0.05) in a concentration-dependent manner (Table 7).

10

11 <<Table 7>>

12 **2.2.3 Molecular docking studies**

13 Docking studies were performed to gain insight into the protein inhibitor interactions inside  
14 the enzyme binding sites. Over the past decade DNA gyrase receptor remains one of the most  
15 investigated and validated targets for the development of anti bacterials [19]. Most of the  
16 synthesized spirocarbocyclic compounds have shown significant activity against *S. aureus*  
17 (MRSA) hence it is thought worthwhile to do docking studies to support the *in vitro* activity.  
18 All the synthesized new spirocarbocycles were subjected to docking using MOE 2011  
19 software version 7.1. All the prepared compounds were chosen for the docking study of  
20 ligands with the DNA gyrase receptor. To find the potential of these molecules against the  
21 human lung cancer cells the compounds were also docked to the Anaplastic Lymphoma  
22 Kinase (ALK) receptor [20].

23 To verify the reproducibility of docking calculations the bound ligand was extracted from the  
24 complexes and submitted for one ligand run calculation. The final docked conformations fall  
25 within 0.5 to 1 Å root-mean-square deviation [21]. Hence it was concluded that this method

1 could be used for the docking of other compounds (**Figure 5a& b**). We have also performed  
2 the docking of the standard ligand Streptomycin with the DNA gyrase receptor for method  
3 validation (**Figure 6**).

4 <<**Figure 5**>>

5 <<**Figure 6**>>

6 The docked ligand conformations were analyzed in terms of free energy of binding (FEB),  
7 hydrogen bonding and hydrophobic interactions. One hundred (100) docking runs were  
8 performed and the best docked representation of the ligand was selected based on the  
9 conformation with lowest value of FEB.

10 Among all compounds docked to the DNA gyrase receptor, compound **4i** was the most  
11 active. It had a high binding energy of -11.64 kcal/mol. This compound exactly fits as that of  
12 ligand and shows strong interaction with ARG 1122 aminoacid. (**Figure 8**)

13 <<**Figure 7**>>

14 The intermediate active compound **4m** binds with the DNA gyrase receptor and the  
15 corresponding binding energy is -9.51 kcal/mol. It shows interaction with the aminoacid  
16 GLN 1056 (**Figure 8**). The least active compound **6h** has binding energy value -8.14  
17 kcal/mol, and it shows interaction with ALA 1068 aminoacid (**Figure 9**).

18 <<**Figure 8**>>

19 <<**Figure 9**>>

20 Molecular docking studies of synthesized molecules to the ALK receptor show that, the most  
21 active compound **6i**, has a very high binding energy value, -18.43 kcal/mol and it interacts  
22 with the three aminoacids namely, ASP 1249, ASN 1254 and GLY 1272 (**Figure10**).  
23 Compound **4h**, a moderately active compound, has a binding energy value of -12.58 kcal/mol  
24 and it interacts with GLU 1210 as shown in **Figure 11**. The least active compound **6h** has a  
25 binding energy of -11.23 kcal/mol. It interacts with ARG 1209 aminoacid. (**Figure 12**)

1 <<Figure 10>>

2 <<Figure 11>>

3 <<Figure 12>>

4 The preparation of various other spirooxindoles and the detailed study of biological activity  
5 are underway.

### 6 **3. Conclusion**

7 We have synthesized a new series of 25 spirocarbocycle derivatives through vinylogous  
8 Michael addition. A new, simple, efficient and environmentally benign method involving the  
9 usage of L-proline for the synthesis of spirocarbocycle was developed. By this method, a  
10 diverse spirocarbocycle library has been rapidly constructed with excellent yields without  
11 involving tedious extraction and isolation procedures. All the compounds were evaluated for  
12 their activities against 4 gram positive bacteria, 5 gram negative bacteria and 2 fungi. Most of  
13 the compounds were found to exhibit significant antimicrobial activity. Among them **4c**, **4i**  
14 and **6i** have shown excellent activities and hence are promising candidates as antibacterial  
15 agents. Compounds **4c**, **4i** and **6i** also showed good anticancer activity against A549 lung  
16 cancer cell line. The docking scores show that the spirocarbocyclic molecules have good  
17 potential against the human lung cancer cells.

### 18 **4. Experimental**

#### 19 *4.1. Chemistry*

20 Analytical TLC was performed on precoated aluminium sheets of silica gel 60F254 of 0.2  
21 mm thickness (Merck, Germany). Melting points were determined on Gallenkamp melting  
22 point apparatus and are uncorrected. Infrared (IR) spectra were recorded on a Perkin Elmer  
23 FT-IR spectrometer as KBr pellets. <sup>1</sup>H NMR (400 MHz) and <sup>13</sup>C NMR (100 MHz) spectra  
24 were recorded in DMSO-d<sub>6</sub> solutions with TMS as an internal standard on a Bruker Avance  
25 DPX-400 MHz instrument. Proton chemical shifts (δ) are relative to tetramethylsilane (TMS,

1  $\delta = 0.00$ ) as internal standard and expressed in parts per million. The number of protons (n)  
2 for a given resonance was indicated as nH. Coupling constants (J) are given in hertz. Spin  
3 multiplicities are given as s (singlet), d (doublet), t (triplet) and m (multiplet). Mass spectra  
4 were recorded under HRMS (ESI) using Thermo Scientific Exactive Orbitrap mass  
5 spectrometer and Thermo Finnigan LCQ Advantage MAX 6000 ESI mass spectrometer and  
6 Perkin-Elmer GC-MS. Elemental analysis data were recorded using Thermo Finnigan  
7 FLASH EA 1112 CHN analyzer.

#### 8 4.1.1. Experimental procedure for the synthesis of (4a-m)

9 Indanedione **1** (1 mmol), aldehydes **2** (1 mmol) **2a-h** were stirred in MeOH in the presence of  
10 L-proline (20 mol%) at room temperature (r.t.) for ten minutes followed by the addition of  
11 alkylidene malononitrile **3a-c** (1 mmol) at r.t. for 3 h. The solid precipitated out, and then  
12 was filtered off and purified by recrystallization from ethanol to afford product **4a-m** as  
13 yellow crystalline solid.

##### 14 4.1.1.1 3'-amino-1,3-dioxo-1'-p-tolyl-1,3,6',7',8',8a'-hexahydro-1'H-spiro[indene-2,2'- 15 naphthalene]-4'-carbonitrile (**4a**)

16 Yellow solid; mp: 224-226°C (Decomposes); IR (cm<sup>-1</sup>): 755, 1246, 1589, 1661, 1704, 1742,  
17 2205, 2921 3246, 3345, 3408. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  .0.71 (q, 1H,  $J = 12.4$ Hz),  
18 1.32 – 1.42 (m, 2H), 1.64 – 1.66 (m, 1H), 1.99 (s, 3H), 2.08 – 2.20 (m, 2H), 2.98 (d, 1H,  $J =$   
19 12.8Hz), 3.04 – 3.07 (m, 1H), 5.59 – 5.61 (m, 1H), 6.12 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable),  
20 6.61 (d, 1H,  $J = 7.6$  Hz), 6.72 – 6.79 (m, 3H), 7.67 (d, 1H,  $J = 7.6$  Hz), 7.75 - 7.77 (m, 3H).  
21 <sup>13</sup>C NMR (100MHz, DMSO-d<sub>6</sub>):  $\delta$  20.0, 22.1, 25.4, 27.9, 33.5, 52.2, 63.6, 82.9, 117.4, 118.1,  
22 123.1, 123.7, 126.6, 129.0, 129.2, 129.2, 131.4, 131.8, 133.1, 136.4, 136.6, 142.7, 143.3,  
23 151.9, 199.7, 200.2. HRMS (ESI): Mass calculated for C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>Na [M+Na]<sup>+</sup> 417.1573,  
24 found, [M+Na]<sup>+</sup>, 417.1573. Anal. Calcd. For: (C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>) C, 79.16; H, 5.62; N, 7.10  
25 Found: C, 79.05; H, 5.73; N, 7.01.

1 4.1.1.2 6-amino-1',3'-dioxo-4-p-tolyl-1',2,3,3a,3',4-hexahydro-2,5'-spirobi[indene]-7-  
2 carbonitrile (**4b**)

3 Yellow solid; mp 215-216 °C (Decomposes); IR (cm<sup>-1</sup>): 790, 1244, 1573, 1664, 1703, 1741,  
4 2206, 2925, 3246, 3347 3406. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.98 (q, 1H, *J* = 10.4), 1.67  
5 – 1.74 (m, 1H), 1.98 (s, 3H), 2.22 – 2.34 (m, 2H), 3.03 (d, 1H, *J* = 12.4 Hz), 3.50 – 3.55 (m,  
6 1H), 5.33 – 5.35 (m, 1H), 6.48 (brs, 2H, NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.72 – 6.77 (m, 4H), 7.69  
7 – 7.75 (m, 4H); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 20.8, 30.4, 31.3, 42.4, 52.8, 65.2, 78.4,  
8 116.0, 117.9, 123.2, 123.7, 129.1, 133.1, 136.4, 136.6, 136.9, 138.5, 142.3, 143.5, 154.6,  
9 199.4, 199.9. MS *m/z* = 381 [M+H]<sup>+</sup>; Anal. Calcd for C<sub>25</sub>H<sub>20</sub>N<sub>2</sub>O<sub>2</sub>: C, 78.93; H, 5.30; N, 7.36  
10 Found: C, 79.94; H, 5.21; N, 7.24.

11 4.1.1.3. 3-amino-1',3'-dioxo-1-p-tolyl-1',3',6,7,8,9,9a-octahydrospiro[benzo[7]annulene-  
12 2,2'-indene]-4-carbonitrile (**4c**)

13 Yellow solid; mp 210-212 °C (Decomposes); IR (cm<sup>-1</sup>): 754, 1252, 1590, 1637, 1704, 1741,  
14 2200, 2925, 3253, 3366, 3535. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 1.14-1.24 (m, 3H), 1.34-  
15 1.36 (m, 1H).1.63-1.64 (m, 2H), 1.99 (s, 3H), 2.18-2.31 (m, 2H), 3.13(d, 1H, *J* = 11.6 Hz),  
16 3.49-3.50 (m, 1H prton merged with solvent peak), 5.76 (t, 1H, *J* = 6.2 Hz), 6.08(brs, 2H,  
17 NH<sub>2</sub>, D<sub>2</sub>O exchangeable)), 6.70 (d, 2H, *J* = 8.4 Hz), 6.79 (d, 2H, *J* = 8.0Hz), 7.63-7.66 (m,  
18 1H), 7.74-7.81 (m, 3H); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 20.7, 25.9, 27.9, 29.8, 31.8, 38.0,  
19 52.2,63.0, 83.3, 118.6, 121.2, 123.1, 124.0, 129.2, 134.6, 136.5, 136.7, 136.9, 137.3, 142.2,  
20 143.1, 152.3, 199.0, 199.4. HRMS (ESI): Mass calculated for C<sub>27</sub>H<sub>25</sub>N<sub>2</sub>O<sub>2</sub> [M+H]<sup>+</sup> 409.1911  
21 found, [M+H]<sup>+</sup> 409.1916; Anal. Calcd for C<sub>27</sub>H<sub>24</sub>N<sub>2</sub>O<sub>2</sub>: C, 79.39; H, 5.92; N, 6.86;. Found: C,  
22 79.28; H, 6.02; N, 7.94.

23 4.1.1.4. 3'-amino-1'-(4-methoxyphenyl)-1,3-dioxo-1,3,6',7',8',8a'-hexahydro-1'H-  
24 spiro[indene-2,2'-naphthalene]-4'-carbonitrile (**4d**)

1 Yellow solid; mp 194-197 °C (Decomposes) ; IR (cm<sup>-1</sup>): 758, 1246, 1588, 1656, 1702, 1739,  
2 2202, 2927, 3250, 3343, 3429. <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.69 (q, 1H, *J* = 12.4 Hz),  
3 1.13-1.17 (m, 2H), 1.61-1.67 (m, 1H), 2.14-2.22 (m, 2H), 2.95 (d, 1H, *J* = 12.4 Hz), 3.37-3.43  
4 (m, 1H), 3.48 (s, 3H), 5.60-5.62 (m, 1H), 5.94 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.49-6.52  
5 (m, 2H), 6.64 (d, 1H, *J* = 8 Hz), 6.74 (d, 1H, *J* = 8.8 Hz), 7.67 (d, 1H, *J* = 7.2 Hz), 7.73-7.78 (m,  
6 3H); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 22.1, 25.3, 27.7, 30.4, 34.5, 55.3, 63.7, 85.1, 113.6,  
7 114.7, 123.3, 127.5, 131.4, 134.2, 136.8, 140.4, 146.8, 158.3, 172.2, 197.8, 199.0; MS *m/z* =  
8 411 [M+H]<sup>+</sup>; Anal. Calcd for C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>3</sub>: C, 76.08; H, 5.40; N, 6.82 Found: C, 76.19; H,  
9 5.30; N, 6.71.

10 4.1.1.5. *6-amino-4-(4-methoxyphenyl)-1',3'-dioxo-1',2,3,3a,3',4-hexahydro-2,5'-*  
11 *spirobi[indene]-7-carbonitrile (4e)*

12 Yellow solid; mp 206-209 °C (Decomposes) ; IR (cm<sup>-1</sup>): 1254, 1512, 1573, 1661, 1702, 1739,  
13 2204, 2932, 3244, 3345, 3413; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.23-0.28 (m, 1H), 0.98-  
14 1.05 (m, 1H), 1.57-1.63 (m, 2H), 2.34 (d, 1H, *J* = 7.6 Hz), 2.77 (s, 3H), 2.82-2.90 (m, 1H),  
15 4.65-4.67 (m, 1H), 5.62 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 5.80 (d, 2H, *J* = 8 Hz), 6.04-  
16 6.08 (m, 2H), 7.00-7.06 (m, 4H); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 30.4, 31.3, 42.5, 46.4,  
17 55.3, 65.3, 78.7, 116.8, 117.9, 123.3, 123.6, 127.8, 136.7, 136.9, 138.0, 142.2, 143.3, 154.4,  
18 158.6, 199.6, 200.3; MS *m/z* = 397 [M+H]<sup>+</sup>; Anal. Calcd for C<sub>25</sub>H<sub>20</sub>N<sub>2</sub>O<sub>3</sub>, C, 75.74; H, 5.08;  
19 N, 7.07 Found: C, 75.83; H, 5.19; N, 7.17.

20 4.1.1.6. *3'-amino-1'-(2-fluorophenyl)-1,3-dioxo-1,3,6',7',8',8a'-hexahydro-1'H-spiro[indene-*  
21 *2,2'-naphthalene]-4'-carbonitrile (4f)*

22 Yellow solid; mp 222-224 °C (Decomposes) ; IR (cm<sup>-1</sup>): 759, 1247, 1588, 1662, 1704, 1742,  
23 2205, 2919, 3249, 3349, 3410; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.66 (q, 1H, *J* = 12.2 Hz),  
24 1.26-1.34 (m, 2H) 1.62-1.65 (m, 1H), 2.05-2.17 (m, 2H), 3.44 (d, 1H, *J* = 12.5 Hz), 4.48-

1 4.51 (m, 1H), 5.62-5.64 (m, 1H), 6.06 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.79-6.84 (m,  
2 2H), 6.87-6.90 (m, 1H), 6.93-6.96 (m, 1H), 7.70 (d, 1H, *J* = 5 Hz), 7.73-7.76 (m, 3H); <sup>13</sup>C  
3 NMR (100 MHz, DMSO-d<sub>6</sub>): δ 21.7, 25.3, 27.9, 33.4, 42.9, 63.1, 83.6, 118.1, 118.8, 122.9,  
4 123.6, 124.5, 129.2, 130.2, 131.4, 137.3, 142.8, 151.9, 159.2, 198.8, 200.1; MS *m/z* = 399  
5 [M+H]<sup>+</sup>; Anal. Calcd for C<sub>25</sub>H<sub>19</sub>FN<sub>2</sub>O<sub>2</sub>: C, 75.36; H, 4.81; N, 7.03 Found: C, 75.47; H, 4.72;  
6 N, 7.12.

7 *4.1.1.7. 6-amino-4-(2-fluorophenyl)-1',3'-dioxo-1',2,3,3a,3',4-hexahydro-2,5'-spirobi[indene]-*  
8 *7-carbonitrile (4g)*

9 Yellow solid; mp 222-225 °C (Decomposes); IR (cm<sup>-1</sup>): 761, 1245, 1573, 1646, 1708, 1742,  
10 2199, 2829, 3258, 3359, 3445; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.25 (q, 1H, *J* = 11.6 Hz),  
11 1.03-1.04 (m, 1H), 1.57-1.66 (m, 2H), 2.42 (d, 1H, *J* = 12.4 Hz), 2.91-2.97 (m, 1H), 4.71-  
12 4.72 (m, 1H), 5.78 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.09 (t, 1H, *J* = 9.2 Hz), 6.16-6.20  
13 (m, 1H), 6.27-6.34 (m, 2H), 7.00-7.01 (m, 1H), 7.08 (d, 3H, *J* = 2.9 Hz); <sup>13</sup>C NMR (100  
14 MHz, DMSO-d<sub>6</sub>): δ 35.0, 36.0, 47.2, 48.9, 69.5, 83.4, 120.1, 121.8, 122.6, 128.1, 128.3,  
15 129.6, 133.9, 135.1, 141.6, 141.7, 142.2, 147.0, 147.6, 159.2, 163.2, 203.5, 204.0; MS *m/z* =  
16 385 [M+H]<sup>+</sup>; Anal. Calcd for C<sub>24</sub>H<sub>17</sub>FN<sub>2</sub>O<sub>2</sub>: C, 74.99; H, 4.46; N, 7.29. Found: C, 75.17; H,  
17 4.35; N, 7.38.

18 *4.1.1.8. 3'-amino-1'-(4-chlorophenyl)-1,3-dioxo-1,3,6',7',8',8a'-hexahydro-1'H-spiro[indene-*  
19 *2,2'-naphthalene]-4'-carbonitrile (4h)*

20 Yellow solid; mp 246-248 °C (Decomposes); IR (cm<sup>-1</sup>): 833, 1093, 1244, 1590, 1636, 1697,  
21 1738, 2201, 2923, 3251, 3360, 3452; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.73 (q, 1H, *J* = 12.4  
22 Hz), 1.24-1.32 (m, 1H), 1.40-1.43 (m, 1H), 1.64-1.67 (m, 1H), 1.97-2.07 (m, 1H), 2.15-2.20  
23 (m, 1H), 3.06 (d, 1H, *J* = 12.4 Hz) 3.09-3.10 (m, 1H), 5.60-5.62 (m, 1H), 6.19 (brs, 2H, -  
24 NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.76 (d, 1H, *J* = 7.2 Hz), 6.86(d, 1H, *J* = 7.6 Hz), 7.04 (d, 2H, *J* =

1 8.4 Hz), 7.69 (d, 1H,  $J = 7.6$  Hz), 7.78 (q, 3H,  $J = 7.0$  Hz);  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ ):  
2  $\delta$  22.0, 25.3, 27.7, 33.3, 51.7, 63.5, 82.8, 117.7, 118.0, 123.2, 123.8, 128.5, 128.6, 128.7,  
3 131.4, 132.3, 135.2, 136.7, 136.8, 142.5, 143.2, 151.6, 199.5, 200.0; MS  $m/z = 415$   $[\text{M}+\text{H}]^+$ ;  
4 Anal. Calcd for  $\text{C}_{25}\text{H}_{19}\text{ClN}_2\text{O}_2$ : C, 72.37; H, 4.62; N, 6.75 Found: C, 72.28; H, 4.73; N, 6.65.

5

6 4.1.1.9. *3-amino-1-(4-chlorophenyl)-1',3'-dioxo-1,1',3',6,7,8,9,9a-*  
7 *octahydrospiro[benzo[7]annulene-2,2'-indene]-4-carbonitrile (4i)*

8 Yellow solid; mp 180-181 °C (Decomposes); IR ( $\text{cm}^{-1}$ ): 833, 1091, 1490, 1590, 1635, 1700,  
9 1741, 2199, 2929, 3366, 3455;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  0.50 (m, 3H), 1.34-1.38  
10 (m, 1H), 1.66-1.69 (m, 2H), 2.02-2.05 (m, 1H), 2.36-2.40 (m, 1H), 2.51 (d, 1H,  $J = 11.7$  Hz)  
11 2.98-3.00 (m, 1H), 5.07 (t, 1H,  $J = 5.78$  Hz), 5.27 (brs, 2H,  $-\text{NH}_2$ ,  $\text{D}_2\text{O}$  exchangeable), 6.09-  
12 6.15 (m, 3H), 6.47(d, 1H,  $J = 8.2$  Hz), 6.96 (d, 1H,  $J = 7.4$  Hz), 7.07-7.12 (m, 3H);  $^{13}\text{C}$  NMR  
13 (100 MHz, DMSO- $d_6$ ):  $\delta$  25.6, 27.8, 29.6, 30.1, 36.3, 52.4, 62.9, 83.4, 118.5, 122.1, 123.3,  
14 123.9, 126.4, 128.4, 129.3, 132.5, 136.5, 137.1, 141.9, 143.0, 151.9, 199.0, 199.1; MS  $m/z =$   
15 429  $[\text{M}+\text{H}]^+$ ; Anal. Calcd for  $\text{C}_{26}\text{H}_{21}\text{ClN}_2\text{O}_2$ : C, 72.81; H, 4.94; N, 6.53. Found: C, 72.75; H,  
16 5.15; N, 6.42.

17 4.1.1.10. *3'-amino-1'-(4-bromophenyl)-1,3-dioxo-1,3,6',7',8',8a'-hexahydro-1'H-*  
18 *spiro[indene-2,2'-naphthalene]-4'-carbonitrile (4j)*

19 Yellow solid; mp 254-256 °C (Decomposes); IR ( $\text{cm}^{-1}$ ): 805, 1243, 1589, 1636, 1696, 1738,  
20 2200, 2948, 3250, 3361, 3452;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  0.72 (q, 1H,  $J = 12$  Hz),  
21 1.30 (d, 1H,  $J = 11.2$  Hz), 1.39-1.42 (m, 1H), 1.65(d, 1H,  $J = 9.2$  Hz), 2.06-2.07 (m, 1H),  
22 2.15-2.20 (m, 1H), 3.03 (d, 1H,  $J = 11.2$  Hz) 3.34-3.41 (m, 1H), 5.60-5.62 (m, 1H), 6.18 (brs,  
23 2H,  $-\text{NH}_2$ ,  $\text{D}_2\text{O}$  exchangeable), 6.70 (d, 1H,  $J = 7.6$  Hz), 7.17 (d, 1H,  $J = 8.4$  Hz), 7.69 (d,

1 2H,  $J = 8.4$  Hz), 7.69(d, 1H,  $J = 7.6$  Hz), 7.74-7.81(m, 3H);  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ ):  
2  $\delta$  22.0, 25.3, 27.7, 33.3, 51.8, 63.4, 82.8, 117.7, 118.0, 120.9, 123.3, 123.8, 129.0,  
3 131.4, 131.5, 133.5, 135.6, 136.7, 136.8, 142.5, 143.2, 151.6, 199.5, 200.0; MS  $m/z = 459$   
4  $[\text{M}+\text{H}]^+$ ; Anal. Calcd for  $\text{C}_{25}\text{H}_{19}\text{BrN}_2\text{O}_2$ : C, 65.37; H, 4.17; N, 6.10. Found: C, 65.26; H,  
5 4.27; N, 6.20.

6 4.1.1.11. *3'-amino-1'-(naphthalen-1-yl)-1,3-dioxo-1,3,6',7',8',8a'-hexahydro-1'H-*  
7 *spiro[indene-2,2'-naphthalene]-4'-carbonitrile (4k)*

8 Yellow solid; mp 213-215 °C (Decomposes); IR ( $\text{cm}^{-1}$ ): 774, 1256, 1488, 1585, 1633, 1721,  
9 2195, 2926, 3420, 3466;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  0.65 (q, 1H,  $J = 12.1$  Hz), 1.23-  
10 1.26 (m, 1H), 1.73-1.79 (m, 2H), 2.13-2.22 (m, 1H), 2.37 (d, 1H,  $J = 9.4$  Hz), 2.65 (d, 1H,  $J$   
11 = 10.8 Hz) 3.14-3.20 (m, 1H), 5.65-5.67 (m, 1H), 6.10 (brs, 2H,  $-\text{NH}_2$ ,  $\text{D}_2\text{O}$  exchangeable),  
12 7.09-7.11 (m, 2H), 7.23-7.28(m, 2H), 7.45-7.51(m, 2H), 7.63-7.70 (m, 4H), 7.79-7.82(m,  
13 1H);  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ ):  $\delta$ , 27.3, 35.1, 36.9, 44.9, 52.5, 63.8, 83.2, 118.0,  
14 121.8, 123.0, 123.1, 123.7, 125.1, 126.0, 126.7, 128.8, 129.3, 132.1, 133.7, 136.1, 142.7,  
15 152.3, 167.6, 199.5, 200.5; MS  $m/z = 431$   $[\text{M}+\text{H}]^+$ ; Anal. Calcd for  $\text{C}_{29}\text{H}_{22}\text{N}_2\text{O}_2$ : C, 80.91; H,  
16 5.15; N, 6.51. Found: C, 80.82; H, 5.26; N, 6.60.

17 4.1.1.12. *3'-amino-1'-(furan-3-yl)-1,3-dioxo-1,3,6',7',8',8a'-hexahydro-1'H-spiro[indene-2,2'-*  
18 *naphthalene]-4'-carbonitrile (4l)*

19 Yellow solid; mp 247-248 °C (Decomposes); IR ( $\text{cm}^{-1}$ ): 1243, 1589, 1660, 1704, 2206, 2923,  
20 3253, 3347, 3422;  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  0.79 (q, 1H,  $J = 12$  Hz), 1.41-1.43 (m,  
21 2H), 1.69-1.72 (m, 1H), 1.94-2.19 (m, 2H), 2.87-2.92 (m, 1H), 2.98 (d, 1H,  $J = 12.4$  Hz),  
22 5.57-5.59 (m, 1H), 5.87(s, 1H) 6.14 (brs, 2H,  $-\text{NH}_2$ ,  $\text{D}_2\text{O}$  exchangeable), 7.13 (s, 1H), 7.20(s,  
23 1H), 7.79-7.80(m, 1H), 7.84-7.85(m, 1H);  $^{13}\text{C}$  NMR (100 MHz, DMSO- $d_6$ ):  $\delta$  26.7, 30.1,  
24 32.6, 38.2, 47.6, 67.5, 87.6, 122.2, 122.8, 125.1, 128.1, 128.6, 136.2, 141.5, 146.5, 147.4,

1 148.1, 148.9, 156.3, 204.7, 204.9; MS  $m/z = 371$   $[MH]^+$ ; Anal. Calcd for  $C_{23}H_{18}N_2O_3$ : C,  
2 74.58; H, 4.90; N, 7.56. Found: C, 74.48; H, 4.79; N, 7.65

3 *4.1.1.13. 3'-amino-1,3-dioxo-1'-(thiophen-2-yl)-1,3,6',7',8',8a'-hexahydro-1'H-spiro[indene-*  
4 *2,2'-naphthalene]-4'-carbonitrile (4m)*

5 Yellow solid; mp 248-250 °C (Decomposes); IR ( $cm^{-1}$ ): 1244, 1587, 1662, 1704, 1741, 2205,  
6 2923, 3250, 3347, 3411;  $^1H$  NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  0.83 (q, 1H,  $J = 12.1$  Hz), 1.39-  
7 1.48 (m, 2H), 1.68-1.69 (m, 1H), 2.08-2.21 (m, 2H), 2.93-2.99 (m, 1H), 3.35-3.40(m, 1H),  
8 5.59-5.61 (m, 1H),6.15 (brs, 2H,  $-NH_2$ ,  $D_2O$  exchangeable), 6.55-6.61(m, 2H), 7.08-7.09(m,  
9 1H), 7.75-7.82(m, 4H);  $^{13}C$  NMR (100 MHz, DMSO- $d_6$ ):  $\delta$  22.0, 25.4, 27.7, 35.5, 47.6, 63.3,  
10 82.8, 117.8, 118.0, 123.4, 124.0, 131.2, 136.6, 136.8, 142.8, 143.3, 151.3, 199.7, 200.0; MS  
11  $m/z = 387$   $[M+H]^+$ ; Anal. Calcd for  $C_{23}H_{18}N_2O_2S$ : C, 71.48; H, 4.69; N, 7.25. Found: C,  
12 71.36; H, 4.80; N, 7.16

13 *4.1.2. Experimental procedure for the synthesis of (6a-j)*

14 Oxindole **5** (1 mmol), aldehydes **2** (1 mmol) **2a-j** were stirred in MeOH in the presence of L-  
15 proline (20 mol%) at room temperature (r.t.) for ten minutes followed by the addition of  
16 alkylidene malononitrile **3a-c** (1 mmol) at r.t. for 5 h. The solid precipitated out was filtered  
17 off and purified by recrystallization from ethanol to afford product **6a-j** as white crystalline  
18 solid.

19 *4.1.2.1. 3'-amino-2-oxo-1'-p-tolyl-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-*  
20 *naphthalene]-4'-carbonitrile (6a)*

21 White solid; mp 248-249 °C (Decomposes); IR ( $cm^{-1}$ ):752, 1191, 1388, 1580, 1627, 1702,  
22 2200, 2926. 3057, 3299, 3329, 3422;  $^1H$  NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  0.72 (q, 1H,  $J = 12.1$   
23 Hz), 1.26-1.29 (m, 1H), 1.37-1.41 (m, 1H), 1.64-1.67 (m, 1H), 2.03-2.07(m, 1H), 2.12 (s,  
24 3H), 2.16-2.21 (m, 1H), 3.02 (d, 1H,  $J = 12.4$  Hz), 3.37-3.41 (m, 1H), 5.38 (brs, 2H,  $-NH_2$ ,

1 D<sub>2</sub>O exchangeable), 5.58-5.60 (m, 1H), 6.37 (d, 1H,  $J = 8.8$  Hz), 6.49 (d, 1H,  $J = 8$ Hz), 6.66  
2 (d, 1H,  $J = 7.6$  Hz), 6.93-7.00 (m, 2H), 7.06 (t, 1H,  $J = 7.6$  Hz), 7.13 (d, 1H,  $J = 8$ Hz), 7.28  
3 (d, 1H,  $J = 7.6$  Hz), 10.35 (brs, 1H, -NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR (100 MHz, DMSO-  
4 d<sub>6</sub>):  $\delta$  20.9, 22.3, 25.5 28.3, 32.7, 52.6, 57.9, 82.0, 110.0, 116.8, 118.5, 122.2, 124.7, 126.4,  
5 128.1, 128.9, 129.2, 129.7, 132.0, 132.6, 134.5, 135.9, 142.9, 154.5, 176.6; HRMS (ESI):  
6 Mass calculated for C<sub>26</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>Na [M+Na]<sup>+</sup>: 404.1733, found: 404.1734; Anal. Calcd for  
7 C<sub>25</sub>H<sub>23</sub>N<sub>3</sub>O: C, 78.71; H, 6.08; N, 11.02. Found: C, 78.60; H, 6.17; N, 11.12.

8 4.1.2.2. 3-amino-2'-oxo-1-p-tolyl-1,6,7,8,9,9a-hexahydrospiro[benzo[7]annulene-2,3'-  
9 indoline]-4-carbonitrile (**6b**)

10 White solid; mp 238-240 °C (Decomposes); IR (cm<sup>-1</sup>):755, 1475, 1577, 1622, 1693, 1716,  
11 2200, 2927, 3263, 3350, 3460; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  1.01 (q, 1H,  $J = 11.3$  Hz),  
12 1.40-1.49 (m, 1H), 1.60-1.67 (m, 3H), 1.71-1.74 (m, 1H), 2.10(s, 3H), 2.24-2.28 (m, 1H),  
13 2.67-2.69 (m, 1H), 3.20 (d, 1H,  $J = 12.2$  Hz), 3.27-3.28 (m, 1H), 5.26 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O  
14 exchangeable), 5.70-5.74 (m, 1H), 6.92-6.97 (m, 1H), 6.99-7.03 (m, 2H), 7.05-7.10 (m, 2H),  
15 7.14-7.21 (m, 1H), 7.33-7.40 (m, 2H), 10.42 (brs, 1H, -NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR  
16 (100 MHz, DMSO-d<sub>6</sub>):  $\delta$  20.9, 26.0, 28.1, 30.3, 32.3, 37.7, 52.6, 57.6, 82.5, 107.5, 110.0,  
17 119.1, 120.5, 122.2, 125.4, 127.2, 127.6, 128.3, 128.6, 129.0, 129.4, 129.5, 135.9, 136.3,  
18 138.0, 138.3, 142.7, 154.5, 176.0; MS  $m/z = 396$  [M+H]<sup>+</sup>; Anal. Calcd for C<sub>26</sub>H<sub>25</sub>N<sub>3</sub>O: C,  
19 78.96; H, 6.37; N, 10.62. Found: C, 78.87; H, 6.45; N, 10.53

20 4.1.2.3. 3'-amino-1'-(4-methoxyphenyl)-2-oxo-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-  
21 naphthalene]-4'-carbonitrile (**6c**)

22 White solid; mp 241-243 °C (Decomposes); IR (cm<sup>-1</sup>): 833, 1185, 1266, 1512, 1573,1618,  
23 1701, 2195, 2224, 2952, 3276, 3324, 3425; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>):  $\delta$  0.72 (q, 1H,  $J$   
24 = 12.1 Hz), 1.27-1.30 (m, 1H), 1.37-1.50 (m, 1H), 1.65-1.74 (m, 1H), 2.06-2.08(m, 1H),

1 2.16-2.21 (m, 1H), 3.01 (d, 1H,  $J = 12.8$  Hz), 3.31-3.34 (m, 1H), 3.60 (s, 3H), 5.43 (brs, 2H, -  
2  $NH_2$ ,  $D_2O$  exchangeable), 5.57-5.59 (m, 1H), 6.39 (s, 2H), 6.50 (d, 1H,  $J = 7.6$  Hz), 6.76 (d,  
3 1H,  $J = 8.5$  Hz), 6.93-6.97 (m, 1H), 7.05-7.09 (m, 1H), 7.14 (d, 1H,  $J = 8.6$  Hz), 7.28 (d, 1H,  
4  $J = 7.3$  Hz), 10.37 (brs, 1H, - $NH$ ,  $D_2O$  exchangeable);  $^{13}C$  NMR (100 MHz,  $DMSO-d_6$ ):  $\delta$   
5 22.3, 25.5, 28.3, 32.8, 52.2, 55.2, 58.0, 81.8, 110.0, 112.6, 113.9, 114.3, 116.6, 118.5, 122.2,  
6 124.7, 127.6, 129.2, 129.3, 129.7, 129.8, 132.6, 133.1, 142.9, 154.5, 158.1, 176.7; MS  $m/z =$   
7 398  $[M+H]^+$ ; Anal. Calcd for  $C_{25}H_{23}N_3O_2$ : C, 75.54; H, 5.83; N, 10.57. Found: C, 75.45; H,  
8 5.72; N, 10.66.

9 4.1.2.4. *3'-amino-1'-(2-fluorophenyl)-2-oxo-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-*  
10 *naphthalene]-4'-carbonitrile (6d)*

11 White solid; mp 265-267 °C (Decomposes); IR ( $cm^{-1}$ ): 753, 1203, 1227, 1489, 1577, 1625,  
12 1701, 2198, 2831, 2939, 3335, 3427;  $^1H$  NMR (400 MHz,  $DMSO-d_6$ ):  $\delta$  0.74 (q, 1H,  $J = 12.0$   
13 Hz), 1.20-1.23 (m, 1H), 1.36-1.44 (m, 1H), 1.65-1.67 (m, 1H), 2.04-2.21 (m, 2H), 3.39-3.43  
14 (m, 1H), 3.50 (d, 1H,  $J = 12.5$  Hz), 5.52 (brs, 2H, - $NH_2$ ,  $D_2O$  exchangeable), 5.59-5.61 (m,  
15 1H), 6.50 (d, 1H,  $J = 7.7$  Hz), 6.77 (t, 1H,  $J = 9.0$  Hz), 6.92 (t, 1H,  $J = 7.6$  Hz), 7.04-7.10  
16 (m, 3H), 7.21 (d, 1H,  $J = 7.4$  Hz), 7.31 (t, 1H,  $J = 7.4$  Hz), 10.56 (brs, 1H, - $NH$ ,  $D_2O$   
17 exchangeable);  $^{13}C$  NMR (100 MHz,  $DMSO-d_6$ ):  $\delta$  22.1, 25.4, 28.0, 32.5, 43.5, 57.4, 81.7,  
18 109.8, 115.0, 115.2, 117.0, 118.4, 122.2, 124.6, 124.8, 128.5, 128.7, 129.1, 129.2, 129.4,  
19 132.2, 142.7, 154.4, 176.6; HRMS (ESI): Mass calculated for  $C_{24}H_{20}FN_3NaO$   $[M+Na]^+$   
20 408.1488, found,  $[M+Na]^+$  408.1489. Anal. Calcd for  $C_{24}H_{20}FN_3O$ : C, 74.79; H, 5.23; N,  
21 10.90. Found: C, 74.68; H, 5.14; N, 10.79.

22 4.1.2.5. *3'-amino-1'-(4-chlorophenyl)-2-oxo-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-*  
23 *naphthalene]-4'-carbonitrile (6e)*

24 White solid; mp 258-259 °C (Decomposes); IR ( $cm^{-1}$ ): 747, 1487, 1579, 1629, 1725, 2195,

1 2923, 3334, 3453; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.76 (q, 1H, *J* = 12.1 Hz), 1.24-1.27  
2 (m, 1H), 1.38-1.42 (m, 1H), 1.66-1.68 (m, 1H), 2.04-2.08(m, 1H),2.17-2.21 (m, 1H), 3.14 (d,  
3 1H, *J* = 12.4 Hz), 3.37-3.41 (m,1H) 5.52 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 5.59-5.60 (m.  
4 1H), 6.50 (d, 2H, *J* = 7.8 Hz)), 6.91-6.97 (m, 2H), 7.06-7.08 (m, 1H), 7.22-7.29 (m, 2H),  
5 7.30-7.32 (m, 1H),10.43 (brs, 1H, -NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR (100 MHz, DMSO-  
6 d<sub>6</sub>): δ 22.2, 25.5, 28.2, 32.4, 52.2, 57.7, 81.7, 110.0, 116.8, 118.5, 122.3, 124.8, 127.4, 128.3,  
7 128.5, 129.4, 131.7, 132.3, 133.8, 136.6, 142.8, 154.3, 176.4; MS *m/z* = 402 [M+H]<sup>+</sup>; Anal.  
8 Calcd for C<sub>24</sub>H<sub>20</sub>ClN<sub>3</sub>O: C, 71.73; H, 5.02; N, 10.46. Found: C, 71.63; H, 5.11; N, 10.35.

9 4.1.2.6. 3'-amino-1'-(4-bromophenyl)-2-oxo-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-  
10 naphthalene]-4'-carbonitrile (**6f**)

11 White solid; mp 264-266 °C (Decomposes); IR (cm<sup>-1</sup>): 748, 1485, 1577, 1631, 1725, 2195,  
12 2922, 3331, 3446; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.75 (q, 1H, *J* = 12.1 Hz), 1.24-1.26  
13 (m, 1H), 1.38-1.41 (m, 1H), 1.65-1.68 (m, 1H), 2.06-2.08(m, 1H),2.16-2.20 (m, 1H), 3.11 (d,  
14 1H, *J* = 12.4 Hz), 3.34-3.37 (m,1H) 5.46 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 5.58-5.60 (m.  
15 1H), 6.45 (d, 1H, *J* = 8.2 Hz)), 6.52 (d, 1H, *J* = 7.6 Hz) 6.96 (t, 1H, *J* = 7.5 Hz)), 7.03-7.10  
16 (m, 2H), 7.17 (d, 1H, *J* = 8.4 Hz), 7.30 (d, 1H, *J* = 7.4 Hz), 7.39 (d, 1H, *J* = 8.4 Hz), 10.42  
17 (brs, 1H, -NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 22.2, 25.5, 28.2,  
18 32.4, 52.3, 57.7, 81.9, 110.1, 117.0, 118.4, 120.3, 122.4, 124.8, 128.9, 129.3, 129.4, 130.4,  
19 131..2, 132.2, 134.1, 137.0, 142.8, 154.2, 176.5; MS *m/z* = 446 [M+H]<sup>+</sup>; Anal. Calcd for  
20 C<sub>24</sub>H<sub>20</sub>BrN<sub>3</sub>O: C, 64.58; H, 4.52; N, 9.41. Found: C, 64.69; H, 4.43; N, 9.52.

21 4.1.2.7 3'-amino-1'-(naphthalen-1-yl)-2-oxo-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-  
22 naphthalene]-4'-carbonitrile (**6g**)

23 White solid; mp 245-247 °C (Decomposes); IR (cm<sup>-1</sup>): 783, 1391, 1580, 1631, 1703, 2220,  
24 2948, 3333, 3426; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.68 (q, 1H, *J* = 12.1 Hz), 1.00-1.06

1 (m, 1H), 1.32-1.34 (m, 2H), 1.97-2.18 (m, 2H), 3.41-3.46 (m, 1H), 4.24 (d, 1H,  $J = 12.0$  Hz),  
2 5.37 (brs, 2H,  $-NH_2$ ,  $D_2O$  exchangeable), 5.61-5.64 (m, 1H), 6.34 (d, 1H,  $J = 7.7$  Hz), 6.60-  
3 6.64 (m, 1H) 6.78 (t, 1H,  $J = 7.6$  Hz), 7.27-7.37 (m, 2H), 7.42-7.48 (m, 2H), 7.54-7.56 (m,  
4 1H), 7.59-7.67 (m, 2H), 8.16 (d, 1H,  $J = 8.0$  Hz), 10.56 (brs, 1H,  $-NH$ ,  $D_2O$  exchangeable);  
5  $^{13}C$  NMR (100 MHz,  $DMSO-d_6$ ):  $\delta$  22.2, 25.5, 27.6, 34.5, 44.6, 58.2, 82.1, 109.7, 116.9,  
6 118.6, 121.9, 124.2, 124.3, 125.3, 125.4, 125.5, 125.8, 126.0, 126.6, 127.6, 128.6, 128.7,  
7 129.1, 130.9, 132.7, 132.9, 133.3, 133.8, 135.0, 142.4, 155.1, 177.4; MS  $m/z = 418$   $[M+H]^+$ ;  
8 Anal. Calcd for  $C_{28}H_{23}N_3O$ : C, 80.55; H, 5.55; N, 10.06. Found: C, 80.65; H, 5.44; N, 10.17.

9 4.1.2.8. *3'-amino-1'-(furan-3-yl)-2-oxo-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-*  
10 *naphthalene]-4'-carbonitrile (6h)*

11 White solid; mp 236-238 °C. (Decomposes); IR ( $cm^{-1}$ ): 754, 1476, 1583, 1633, 1702, 2199,  
12 2930, 3279, 3422;  $^1H$  NMR (400 MHz,  $DMSO-d_6$ ):  $\delta$  0.83 (q, 1H,  $J = 12.4$  Hz), 1.40-1.48  
13 (m, 2H), 1.70-1.73 (m, 1H), 2.03-2.21 (m, 2H), 2.97 (d, 1H,  $J = 12.0$  Hz), 3.17-3.24 (m, 1H),  
14 5.44 (brs, 2H,  $-NH_2$ ,  $D_2O$  exchangeable), 5.55-5.57 (m, 1H), 6.07 (s, 1H), 6.62 (d, 1H,  $J =$   
15  $7.7$  Hz) 6.81 (s, 1H), 6.97 (t, 1H,  $J = 7.5$  Hz), 7.12-7.19 (m, 1H), 7.36 (s, 1H), 10.44 (brs, 1H,  
16  $-NH$ ,  $D_2O$  exchangeable);  $^{13}C$  NMR (100 MHz,  $DMSO-d_6$ ):  $\delta$  22.2, 25.5, 28.3, 32.6, 44.1,  
17 57.2, 81.8, 110.1, 116.8, 118.5, 121.2, 122.4, 124.2, 129.3, 130.2, 132.1, 141.2, 143.1, 143.3,  
18 154.0, 177.1; MS  $m/z = 358$   $[M+H]^+$ ; Anal. Calcd for  $C_{22}H_{19}N_3O_2$ : C, 73.93; H, 5.36; N,  
19 11.76. Found: C, 73.85; H, 5.45; N, 11.67.

20 4.1.2.9. *3'-amino-2-oxo-1'-(thiophen-2-yl)-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-*  
21 *naphthalene]-4'-carbonitrile (6i)*

22 White solid; mp 267-269 °C (Decomposes); IR ( $cm^{-1}$ ): 696, 1203, 1476, 1578, 1629, 1701,  
23 2200, 2935, 3278, 3423;  $^1H$  NMR (400 MHz,  $DMSO-d_6$ ):  $\delta$  0.85 (q, 1H,  $J = 12.2$  Hz), 1.39-  
24 1.42 (m, 2H), 1.69-1.71 (m, 1H), 2.07-2.21 (m, 2H), 3.30 (d, 1H,  $J = 10.8$  Hz), 3.46-3.52 (m,

1 1H), 5.44 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 5.57-5.59 (m, 1H), 6.57 (d, 2H, *J* = 7.6 Hz),  
2 6.72 (s, 1H) 6.96 (t, 1H, *J* = 7.4 Hz), 7.10 (t, 1H, *J* = 7.6 Hz), 7.15-7.16 (m, 1H), 7.27 (d, 1H,  
3 *J* = 7.4 Hz), 10.48 (brs, 1H, -NH, D<sub>2</sub>O exchangeable); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ  
4 22.2, 25.5, 28.1, 34.6, 48.9, 57.9, 81.8, 110.1, 117.0, 118.4, 122.5, 124.6, 125.5, 126.7, 129.4,  
5 129.7, 132.0, 140.3, 143.2, 153.9, 176.6; MS *m/z* = 374 [M+H]<sup>+</sup>; Anal. Calcd for  
6 C<sub>22</sub>H<sub>19</sub>N<sub>3</sub>OS: C, 70.75; H, 5.13; N, 11.25. Found: C, 70.64; H, 5.02; N, 11.34.

7 4.1.2.10. 3'-amino-2-oxo-1'-(pyridin-2-yl)-6',7',8',8a'-tetrahydro-1'H-spiro[indoline-3,2'-  
8 naphthalene]-4'-carbonitrile (**6j**)

9 White solid; mp 270-271 °C (Decomposes); IR (cm<sup>-1</sup>): 751, 1199, 1475, 1574, 1627, 1699,  
10 2199, 2830, 2937, 3325, 3424; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.83 (q, 1H, *J* = 12.0 Hz),  
11 1.00-1.03 (m, 1H), 1.35-1.38 (m, 1H), 1.63-1.66 (m, 1H), 2.05-2.20 (m, 2H), 3.28 (d, 1H, *J* =  
12 12.1 Hz), 3.53-3.61 (m, 1H), 5.39 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 5.56-5.58 (m, 1H),  
13 6.46 (d, 1H, *J* = 7.7 Hz), 6.63-6.74 (m, 1H) 6.86-6.89 (m, 1H), 6.99-7.03 (m, 2H), 7.24 (d,  
14 1H, *J* = 7.3 Hz), 7.30-7.37 (m, 1H), 8.26-8.33 (m, 1H), 10.26 (brs, 1H, -NH, D<sub>2</sub>O  
15 exchangeable); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 22.2, 25.4, 27.8, 32.5, 54.8, 56.8, 81.7,  
16 109.7, 116.4, 118.5, 121.9, 122.2, 124.9, 128.7, 129.2, 133.0, 135.7, 143.1, 148.8, 155.1,  
17 157.9, 175.9; HRMS (ESI): Mass calculated for C<sub>23</sub>H<sub>21</sub>N<sub>4</sub>O [M+H]<sup>+</sup> 369.1710, found,  
18 [M+H]<sup>+</sup> 369.1717; Anal. Calcd for C<sub>23</sub>H<sub>20</sub>N<sub>4</sub>O: C, 74.98; H, 5.47; N, 15.21. Found: C, 74.88;  
19 H, 5.36; N, 15.10.

20 4.1.3. Experimental procedure for the synthesis of (**8a** & **8b**)

21 1,3-Cyclohexanedione **7** (1 mmol), aldehydes **2** (1 mmol) **2a-b** were stirred in MeOH in the  
22 presence of L-proline (20 mol%) at room temperature (r.t.) for ten minutes followed by the  
23 addition of alkylidene malononitrile **3a** (1 mmol) at r.t. for 3 h. The solid precipitated out,

1 and then filtered off and purified by recrystallization from ethanol to afford product **8a** and  
2 **8b** as white crystalline solid.

3 4.1.3.1. *3'-amino-2,6-dioxo-1'-p-tolyl-6',7',8',8a'-tetrahydro-1'H-spiro[cyclohexane-1,2'-*  
4 *naphthalene]-4'-carbonitrile (8a)*

5 White solid; mp 245-246 °C (Decomposes); IR (cm<sup>-1</sup>):822 1394, 1588, 1657, 1688, 1717,  
6 2202, 2923, 3250, 3345, 3422; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.11 (q, 1H, *J* = 11.7 Hz),  
7 0.57-0.65 (m, 1H), 1.29-1.32 (m, 2H), 1.42-1.48 (m, 1H), 1.59-1.64 (m, 1H), 1.95-2.13 (m,  
8 4H), 2.24 (s, 3H), 2.47-2.52 (m, 1H), 2.73-2.82 (m, 1H), 2.89 (s,2H), 5.52-5.54 (m, 1H), 5.91  
9 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.71 (d, 1H, *J* = 7.6 Hz), 6.83 (d, 1H, *J* = 8.0 Hz)) 7.08  
10 (d, 1H, *J* = 7.6 Hz), 7.16 (d, 1H, *J* = 7.6 Hz); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 14.6, 21.1,  
11 22.1, 25.4, 27.5, 33.3, 54.9, 71.2, 83.1, 117.1, 118.2, 127.5, 129.9, 130.1, 131.3, 132.0, 133.9,  
12 137.8, 154.3, 209.8, 210.9; HRMS (ESI): Mass calculated for C<sub>23</sub>H<sub>24</sub>N<sub>2</sub>NaO<sub>2</sub> [M+Na]<sup>+</sup>  
13 383.1730, found, [M+Na]<sup>+</sup> 383.1731 Anal. Calcd for C<sub>23</sub>H<sub>24</sub>N<sub>2</sub>O<sub>2</sub>: C, 76.64; H, 6.71; N,  
14 7.77. Found: C, 76.75; H, 6.60; N, 7.68.

15 4.1.3.2. *3'-amino-1'-(4-methoxyphenyl)-2,6-dioxo-6',7',8',8a'-tetrahydro-1'H-*  
16 *spiro[cyclohexane-1,2'-naphthalene]-4'-carbonitrile (8b)*

17 White solid; mp 241-243 °C (Decomposes); IR (cm<sup>-1</sup>):830 1262, 1512, 1591, 1659, 1684,  
18 1715, 2204, 2951, 3246, 3346, 3416; <sup>1</sup>H NMR (400 MHz, DMSO-d<sub>6</sub>): δ 0.24 (q, 1H, *J* = 13.3  
19 Hz), 0.62 (q, 1H, *J* = 10.8 Hz), 1.32-1.38 (m, 2H), 1.49-1.55 (m, 1H), 1.63-1.66 (m, 1H),  
20 1.99-2.16 (m, 4H), 2.53-2.58 (m, 1H), 2.76-2.84 (m, 1H), 2.88-2.91 (m, 2H), 3.72 (s, 3H),  
21 5.52-5.54 (m, 1H), 5.93 (brs, 2H, -NH<sub>2</sub>, D<sub>2</sub>O exchangeable), 6.77 (d, 1H, *J* = 8.2 Hz), 6.87 (t,  
22 2H, *J* = 7.8 Hz)) 6.94 (d, 1H, *J* = 8.6 Hz); <sup>13</sup>C NMR (100 MHz, DMSO-d<sub>6</sub>): δ 14.7, 22.1,  
23 25.4, 27.5, 33.4, 54.4, 55.5, 83.1, 114.2, 115.3, 117.1, 118.2, 128.6, 128.7, 132.0, 132.4,  
24 154.3, 159.3, 209.9, 211.0; HRMS (ESI): Mass calculated for C<sub>23</sub>H<sub>25</sub>N<sub>2</sub>O<sub>3</sub> [M+H]<sup>+</sup> 377.1860,

1 found,  $[M+H]^+$  377.1860; Anal. Calcd for  $C_{23}H_{24}N_2O_3$ : C, 73.38; H, 6.43; N, 7.44. Found:  
2 C, 73.47; H, 6.34; N, 7.54.

### 3 4.2. Biological assays

#### 4 4.2.1. Materials and methods for antimicrobial activity

5 Streptomycin (Sigma) was used as positive control against bacteria. Ketoconazole (Himedia,  
6 Mumbai) was used as positive control against fungi.

#### 7 4.2.2. Tested microbes

8 The following bacteria and fungi were used for the experiment. Bacteria; *Shigella flexneri*  
9 MTCC 1457, *Micrococcus luteus* MTCC 106, *Enterobacter aerogenes* MTCC 111,  
10 *Staphylococcus aureus* MTCC 96, *Klebsiella pneumoniae* MTCC 109, *Staphylococcus*  
11 *epidermidis* MTCC 3615, *Proteus vulgaris* MTCC 1771, *Salmonella typhimurium* MTCC  
12 1251 and *Staphylococcus aureus* (MRSA- methicillin resistant). The reference cultures were  
13 obtained from Institute of Microbial Technology (IMTECH), Chandigarh, India-160 036;  
14 fungi: *Malassezia pachydermatis* and *Candida albicans* MTCC 227. All the cultures were  
15 obtained from the Department of Microbiology, Christian Medical College, Vellore, Tamil  
16 Nadu, India.

#### 17 4.2.3. Preparation of inoculums

18 Bacterial inoculums were prepared by growing cells in Mueller Hinton broth (MHB)  
19 (Himedia) for 24 h at 37°C. The filamentous fungi were grown on sabouraud dextrose agar  
20 (SDA) slants at 28°C for 10 days and the spores were collected using sterile doubled distilled  
21 water and homogenized. Yeast was grown on sabouraud dextrose broth (SDB) at 28°C for  
22 48-72 h.

#### 23 4.2.4. Disc diffusion assay

24 Antimicrobial activities were carried out using disc diffusion method [22]. Petri plates were  
25 prepared with 20 ml of sterile Mueller Hinton Agar (MHA) (Hi-media, Mumbai). The test

1 cultures were swabbed on the top of the solidified media and allowed to dry for 10 min and a  
2 specific amount of synthesised compound at 1mg/disc was added to each disc separately. The  
3 loaded discs were placed on the surface of the medium and left for 30 min at room  
4 temperature for compound diffusion. Negative control was prepared using respective  
5 solvents. Streptomycin was used as positive control against bacteria. Ketoconazole was used  
6 as positive control for fungi. The plates were incubated for 24 h at 37°C for bacteria and for  
7 48 h at 28°C for fungi. Zones of inhibition were recorded in millimetres and the experiment  
8 was repeated twice.

#### 9 *4.2.5. Minimum inhibitory concentration (MIC)*

10 Minimum inhibitory concentration studies of 17 compounds were performed according to  
11 the standard reference methods for antibacterial activity [23]. The required concentrations  
12 (1000 µg/mL, 500 µg/mL, 250 µg/mL, 125 µg/mL, 62.5 µg/mL, 31.25 µg/mL, 15.62 µg/mL  
13 and 7.81µg/mL) of the compound were dissolved in DMSO (2%), and diluted to give serial  
14 two-fold dilutions that were added to each medium in 96 well plates. An inoculum of 100 µl  
15 from each well was inoculated. The antifungal agent Ketoconazole for fungi and  
16 Streptomycin for bacteria were included in the assay as positive controls. For fungi, the  
17 plates were incubated for 48 to 72 hours at 28°C and for bacteria the plates were incubated  
18 for 24 h at 37°C. The MIC for fungi was defined as the lowest extract concentration, showing  
19 no visible fungal growth after incubation time. 5 µl of tested broth was placed on the sterile  
20 MHA plates for bacteria and incubated at respective temperature. The MIC for bacteria was  
21 determined as the lowest concentration of the compound inhibiting the visual growth of the  
22 test cultures on the agar plate.

#### 23 *4.2.6. Cytotoxic properties*

24 A549 lung adenocarcinoma cancer cell line was obtained from National Institute of  
25 Cell Sciences, Pune. A549 cell line was maintained in complete tissue culture medium

1 Dulbecco's Modified Eagle's Medium with 10 % Fetal Bovine Serum and 2mM L-Glutamine,  
2 along with antibiotics (about 100 International Unit/mL of penicillin, 100 µg/mL of  
3 streptomycin) with the pH adjusted to 7.2. The cytotoxicity was determined according to the  
4 method of Balachandran et al. [24] with some changes. Cells (5000 cells/well) were seeded  
5 in 96 well plates containing medium with different concentrations such as 50, 40, 30, 20, 10  
6 and 5 µg/mL. The cells were cultivated at 37 °C with 5% CO<sub>2</sub> and 95% air in 100% relative  
7 humidity. After various durations of cultivation, the solution in the medium was removed. An  
8 aliquot of 100 µL of medium containing 1 mg/mL of 3-(4, 5-dimethylthiazol-2-yl)-2, 5-  
9 diphenyl-tetrazolium bromide was loaded in the plate. The cells were cultured for 4 h and  
10 then the solution in the medium was removed. An aliquot of 100 µL of DMSO was added to  
11 the plate, which was shaken until the crystals were dissolved. The cytotoxicity against cancer  
12 cells was determined by measuring the absorbance of the converted dye at 540 nm in an  
13 Enzyme linked immune sorbant assay reader. Cytotoxicity of each sample was expressed as  
14 the half maximal inhibitory concentration (IC<sub>50</sub>) value. The IC<sub>50</sub> value is the concentration  
15 of test sample that causes 50% inhibition of cell growth, averaged from three replicate  
16 experiments.

#### 17 *4.3. Molecular docking studies*

18 Protein and ligand preparation were carried out using MOE (Molecular Operating  
19 Environment) 2011 software tool version 7.1. A PDB entry 2XCS was selected for  
20 antibacterial docking study and was processed with MOE software. The structural issues  
21 such as capping, completing residues with missing atoms and selecting appropriate alternate  
22 locations were corrected automatically using structure preparation application available with  
23 the MOE software. An additional step was carried out to delete extraneous cofactors or  
24 unbound water and the co-crystallized ligand. The active site of protein was identified by  
25 using MOE's site finder application. Ligand 2D structures were drawn using ChemBioDraw

1 Ultra 11.0 (ChemOffice 2008). Ligand 2D structures were converted to mol2 format using  
2 the file convertor Open Babel GUI. Energy minimization was performed to adjust hydrogen  
3 and lone pairs and to calculate partial charges. The Protein and ligand structures were energy  
4 minimized using of MMFF94x force field implemented in MOE. The energy was minimized  
5 to the minimum gradient of 0.05. The final refined poses were ranked by the MM/GBVI  
6 binding free energy estimation. The best docked representation of the ligand was chosen  
7 based on the conformation with lowest binding free energy out of 100 docking runs. The  
8 final docked conformations were within the range of 0.5 to 1 Å root-mean-square deviation.  
9 For the docking of synthesized molecules to the ALK structure with the PDB id 2XP2, the  
10 same procedure followed as for the DNA gyrase receptor.

## 12 Acknowledgements

13 N.S.P thanks the Council of Scientific and Industrial Research (CSIR), New Delhi, India for  
14 the research fellowship and SAIF, IITM, Chennai for single crystal XRD analysis.

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10 Cambridge Crystallographic Data centre as supplemental publication no. CCDC- 972722.  
11 Copies of the data can be obtained, free of charge on application to CCDC, 12 Union Road,  
12 Cambridge CB2 1EZ, UK (fax: +44 01223 336033 or email: [deposit@ccdc.cam.ac.uk](mailto:deposit@ccdc.cam.ac.uk)).
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### 7 **Table, Figures and Schemes captions**

8 **Table 1.** Optimization of reaction conditions for the preparation of spirocarbocycles.

9 **Table 2.** Synthesis of spirocarbocycle derivatives from 1, 3-indandione, substituted  
10 aldehydes and alkylidene malononitrile.

11 **Table 3.** Synthesis of spirooxindole derivatives from oxindole or 1,3-cyclohexanedione,  
12 substituted aldehydes and alkylidene malononitrile.

13 **Table 4.** Synthesis of spirocarbocycle derivatives from substituted aldehydes and alkylidene  
14 malononitrile.

15 **Table 4 .** Crystal data and structure refinement parameters for compound **6a**.

16 **Table 5.** *In vitro* antimicrobial activity of synthesized compounds.

17 **Table 6.** MIC ( $\mu\text{g/ml}$ ) of compounds tested against bacteria.

18 **Table 7.** Anticancer activity of synthesised compounds against A549 cancer cell line and  
19 calculated binding energy with ALK receptor of synthesized spirocarbocycles.

20 **Figure 1.** Some biologically important spirooxindole compounds.

21 **Figure 2.** ORTEP diagram of synthesized compound **6a**.

- 1 **Figure 3.** Comparison of antimicrobial activity of synthesized compounds and standard  
2 drugs.
- 3 **Figure 4.** Comparison of MIC ( $\mu\text{g/ml}$ ) values of synthesized compounds and standard drugs.
- 4 **Figure 5.** Docking of co-crystallized ligand and standard drug Streptomycin with the DNA  
5 gyrase receptor for method validation
- 6 **Figure 6.** Docking of co-crystallized ligand (crizotinib) with the ALK receptor for method  
7 validation
- 8 **Figure 7.** 2D and 3D binding mode of most active compound **4i** (FEB = -11.64 kcal/mol)  
9 with DNA gyrase receptor.
- 10 **Figure 8.** 2D and 3D binding mode of moderate active compound **4m** (FEB = -9.51  
11 kcal/mol) with DNA gyrase receptor.
- 12 **Figure 9.** 2D and 3D binding mode of least active compound **6h** (FEB = -8.14 kcal/mol)  
13 with DNA gyrase receptor.
- 14 **Figure 10.** 2D and 3D binding mode of most active compound **6i** (FEB = -18.43 kcal/mol)  
15 with ALK receptor.
- 16 **Figure 11.** 2D and 3D binding mode of intermediate active compound **4h** (FEB = -12.58  
17 kcal/mol) with ALK receptor.
- 18 **Figure 12.** 2D and 3D binding mode of intermediate active compound **6h** (FEB = -11.23  
19 kcal/mol) with ALK receptor.
- 20 **Scheme 1.** Synthesis of spirocarbocycle.
- 21 **Scheme 2.** Plausible mechanism for the formation of spirocarbocycles.

1 **Scheme 3.** Synthesis of spirocarbocycle derivatives from 1, 3-indandione, substituted  
2 aldehydes and alkylidene malononitrile.

3 **Scheme 4.** Synthesis of spirooxindole derivatives from oxindole, substituted aldehydes and  
4 alkylidene malononitrile

5 **Scheme 5.** Synthesis of spirocarbocycle derivatives from 1,3-cyclohexanedione, substituted  
6 aldehydes and alkylidene malononitrile.

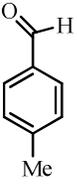
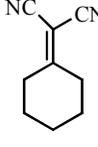
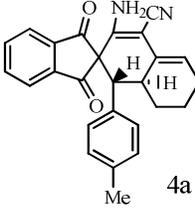
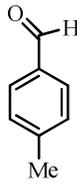
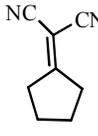
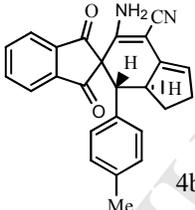
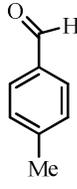
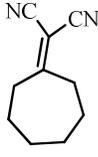
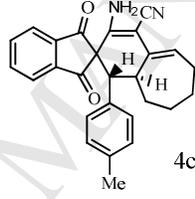
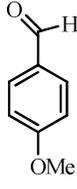
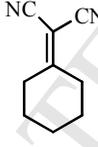
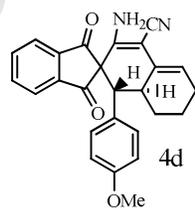
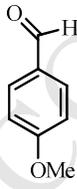
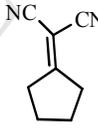
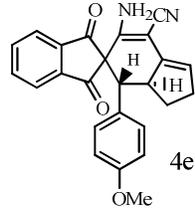
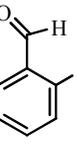
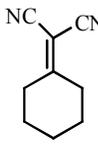
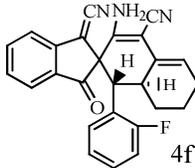
7 **Table 1**

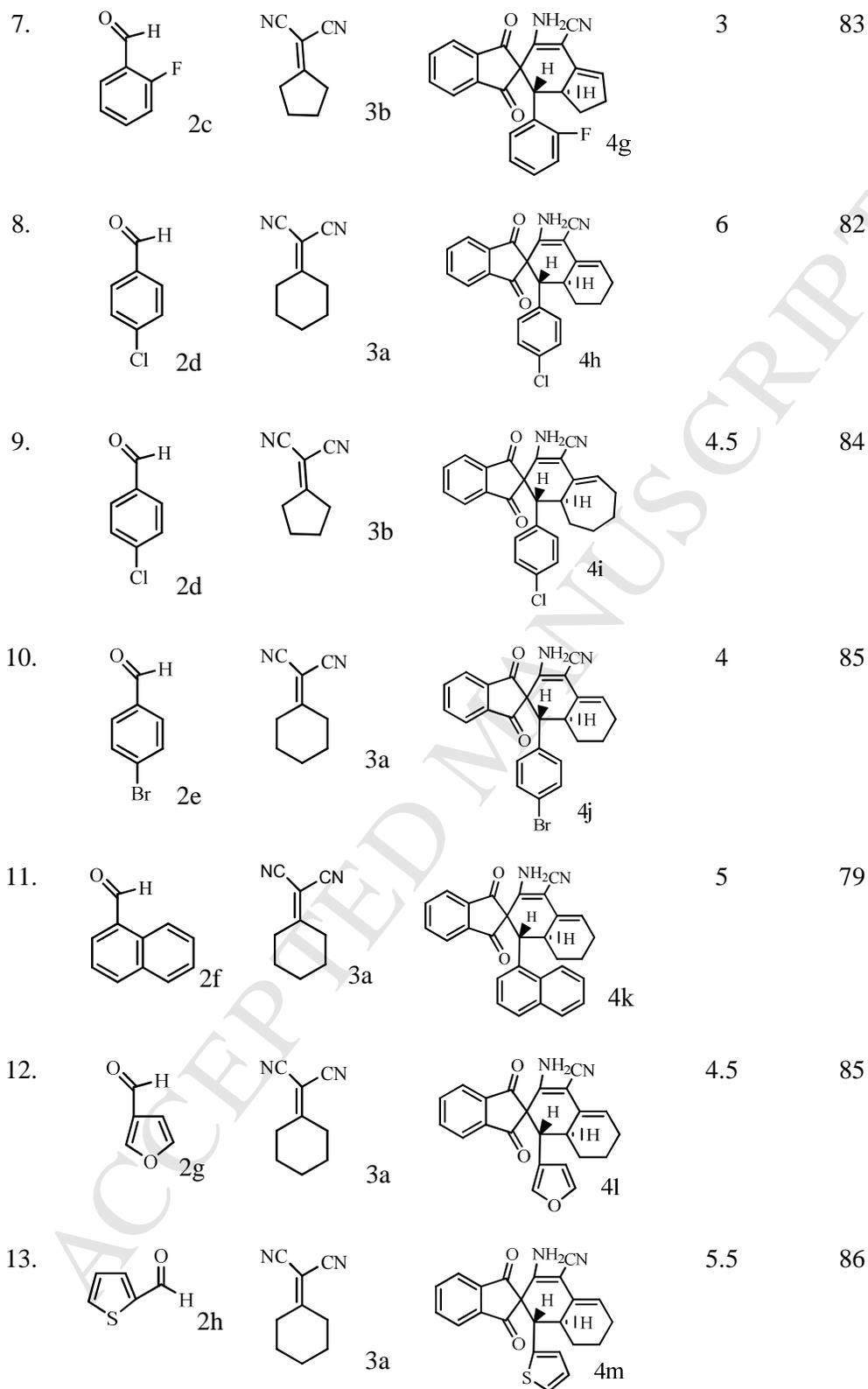
Entry	Solvent	Catalyst (10 mol%)	Conditions	Time (h)	Yield <sup>a</sup> (%)
1.	MeOH	-	r.t.	48	40
2.	EtOH	K <sub>2</sub> CO <sub>3</sub>	r.t.	6	62
3.	MeOH	Et <sub>3</sub> N	r.t.	4	80
4.	EtOH	DABCO	Reflux	6	54
5.	EtOH	NaOEt	r.t.	5	67
6.	EtOH	L-proline	r.t.	3	85
7.	MeOH	L-proline	r.t.	3	86
<b>8.</b>	<b>MeOH</b>	<b>L-proline</b>	<b>r.t.</b>	<b>3</b>	<b>89<sup>b</sup></b>
9.	MeOH	L-proline	r.t.	3	87 <sup>c</sup>
10.	CH <sub>3</sub> CN	L-proline	r.t.	5	69
11.	H <sub>2</sub> O	L-proline	r.t.	6	72
12.	DMF	L-proline	r.t.	8	58

8 [a] Isolated yield of 4a after recrystallization [b]The reaction was performed using 15 mol%  
9 of the catalyst [c]The reaction was performed using 20 mol% of the catalyst.

1

Table 2

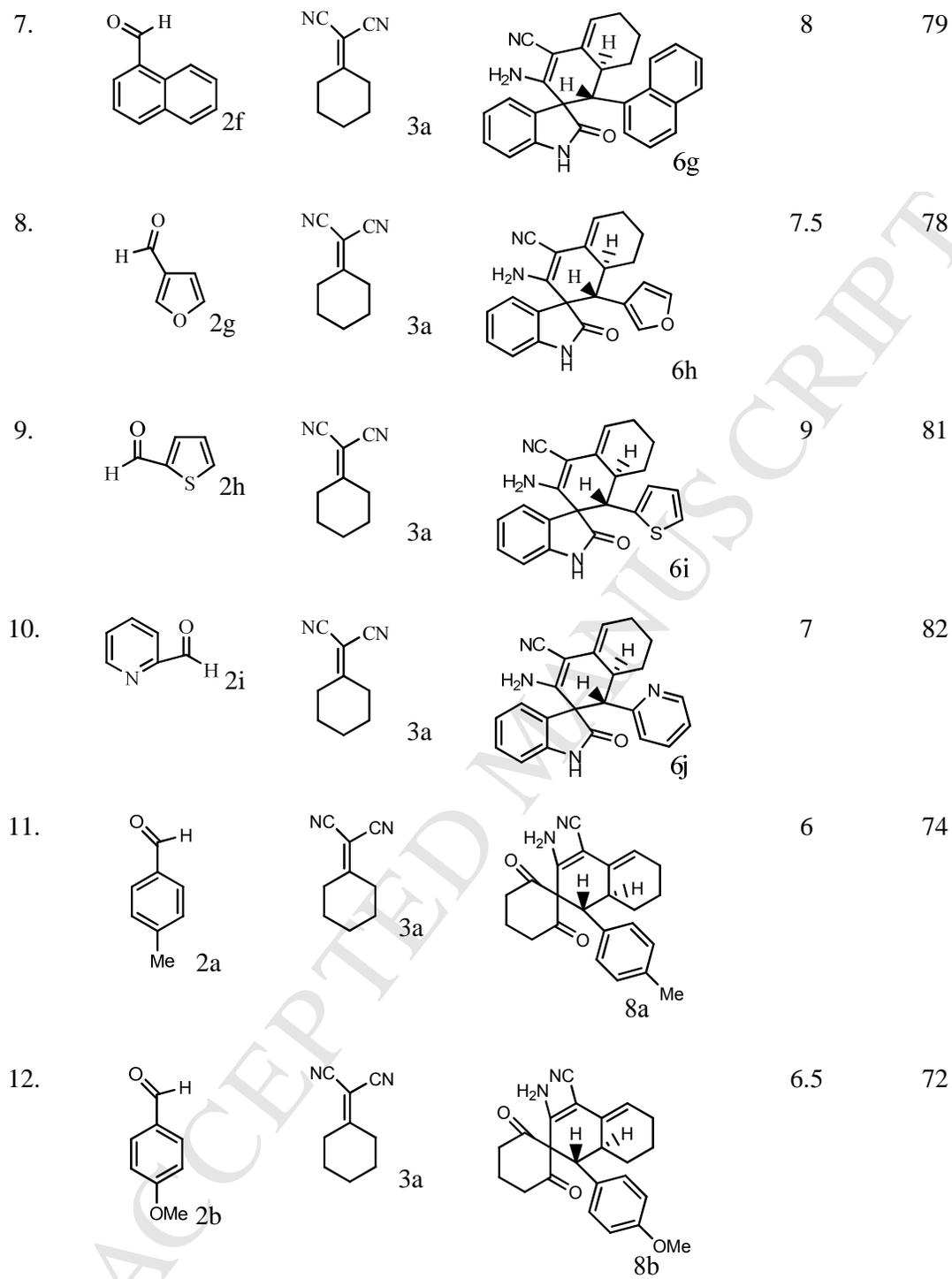
Entry	Aldehyde	Vinylogous malononitrile	Product	Time	Yield
1.	 Me 2a	 3a	 4a	3	89
2.	 Me 2a	 3b	 4b	2.5	90
3.	 Me 2a	 3c	 4c	4	86
4.	 OMe 2b	 3a	 4d	3.5	87
5.	 OMe 2b	 3b	 4e	2.5	88
6.	 2c	 3a	 4f	4	81



1

Table 3

Entry	Aldehyde/ketone	Vinylogous malononitrile	Product	Time	Yield
1.	2a	3a		5	86
2.	2a	3c		6	83
3.	2b	3a		7	84
4.	2c	3a		8	78
5.	2d	3a		10	81
6.	2e	3a		9	80



1

2

3

4

1

**Table 4**

2 Crystal data and structure refinement parameters for compound 6a

Empirical formula	C <sub>25</sub> H <sub>23</sub> N <sub>3</sub> O
Formula weight	381.46
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P21/n
Unit cell dimensions	$a = 9.3858(3)$ Å; $b = 21.8349(6)$ Å; $c = 9.6525(2)$ Å $\alpha = \gamma = 90^\circ$ $\beta = 93.3610(10)^\circ$
Volume	1974.76(9) Å <sup>3</sup>
Z, Calculated density	4, 1.283 Mgm <sup>-3</sup>
Absorption coefficient	0.080 mm <sup>-1</sup>
F(000)	808
Crystal size	0.35 x 0.35 x 0.30 mm
$\theta$ range for data collection	2.31 to 25.00°
Limiting indices	-11 ≤ h ≤ 11, -25 ≤ k ≤ 25, -11 ≤ l ≤ 11
Reflections collected / unique	18019 / 3486 [R(int) = 0.0335]
Completeness to $\theta = 25.00$	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9865 and 0.9632
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3486 / 0 / 275
Goodness-of-fit on F <sup>2</sup>	1.030
Final R indices [I > 2σ(I)]	R1 = 0.0400, wR2 = 0.0965

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R indices (all data)	$R_1 = 0.0587, wR_2 = 0.1072$
Extinction coefficient	0.0079(11)
Largest diff. peak and hole	0.240 and -0.158 e.Å <sup>-3</sup>
CCDC	972722

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ACCEPTED MANUSCRIPT

Table 5

Compounds	Zone of inhibition in mm										
	Gram positive bacteria				Gram negative bacteria				Fungi		
	<i>S. epidermidis</i>	<i>S. aureus</i>	<i>S. aureus (MRSA)</i>	<i>M. luteus</i>	<i>E. aerogens</i>	<i>S. typhimurium</i>	<i>K. pneumonia</i>	<i>P. vulgaris</i>	<i>S. flexneri</i>	<i>C. albicans</i>	<i>M. pachydermatis</i>
4a	10	10	11	9	13	14	15	10	8	10	11
4b	14	17	12	12	15	10	15	16	13	10	13
4c	16	13	17	10	18	19	20	22	23	10	10
4d	12	11	NI	11	10	NI	10	NI	NI	NI	10
4e	NI	NI	10	8	NI	NI	11	NI	NI	NI	NI

4f	14	14	13	15	12	10	16	12	14	10	NI
4g	17	14	16	12	15	12	15	14	10	NI	NI
4h	8	9	13	NI	13	10	NI	NI	8	NI	NI
4i	17	13	21	21	18	18	17	23	22	13	12
4j	NI	14	13	NI	NI	12	10	NI	NI	10	NI
4k	15	12	21	22	17	18	17	19	21	10	9
4l	12	14	13	18	15	10	12	15	14	13	12
4m	12	10	18	10	15	16	14	22	24	10	11
6a	12	11	9	8	13	9	10	12	9	8	10
6b	14	13	15	13	12	16	12	12	11	9	8
6c	9	10	11	8	10	8	9	NI	8	10	13
6d	13	17	12	10	14	11	10	8	9	10	NI

6e	13	NI	NI	9	10	12	16	17	10	9	11
6f	10	12	12	8	15	11	10	9	8	10	10
6g	15	NI	NI	10	18	14	16	NI	12	NI	9
6h	NI	NI	10	8	NI						
6i	19	13	21	15	17	15	16	23	25	10	12
6j	9	NI	NI	NI	8	11	NI	10	NI	12	11
8a	NI	NI	NI	NI	NI	NI	10	9	NI	NI	NI
8b	13	14	15	21	14	11	16	17	19	10	9
Streptomycin	26	14	30	26	22	18	20	30	30	NA	NA
Ketoconazole	NA	28	26								

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NA-not applicable NI – no inhibition.

Table 6

Compounds	Minimum inhibitory concentration ( $\mu\text{g/ml}$ )								
	Gram positive bacteria				Gram negative bacteria				
	<i>S. epidermidis</i>	<i>S. aureus</i>	<i>S. aureus</i> (MRSA)	<i>M.</i> <i>luteus</i>	<i>E. aerogens</i>	<i>S.</i> <i>typhimurium</i>	<i>K.</i> <i>pneumonia</i>	<i>P.</i> <i>vulgaris</i>	<i>S.</i> <i>flexneri</i>
4a	500	250	250	1000	250	250	250	500	1000
4b	250	125	250	250	250	500	250	125	250
4c	250	250	125	250	125	125	62.5	62.5	62.5
4f	250	250	250	250	250	500	125	250	250
4g	125	250	125	250	250	250	250	250	250
4i	125	250	62.5	62.5	125	125	125	62.5	62.5
4k	125	250	62.5	62.5	125	125	125	125	62.5

4l	250	250	250	125	250	500	250	250	250
4m	250	250	125	250	250	125	250	62.5	62.5
6a	250	250	1000	1000	250	1000	500	250	1000
6b	250	250	250	250	250	125	250	250	250
6d	250	125	250	250	250	250	500	1000	1000
6e	250	NI	NI	1000	250	250	125	125	250
6f	500	250	250	1000	250	250	500	1000	1000
6g	125	NI	NI	250	125	250	125	NI	500
6i	125	250	62.5	250	125	250	125	62.5	62.5
8b	250	250	250	250	250	250	125	125	125
Streptomycin	6.25	6.25	6.25	6.25	25	30	6.25	6.25	6.25

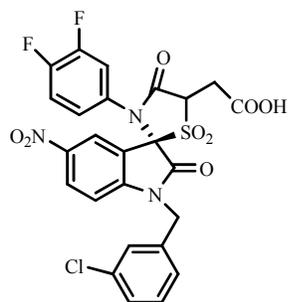
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NI – no inhibition

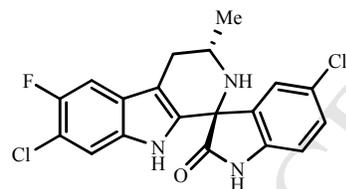
Table 7

Concentration ( $\mu\text{g/mL}$ )	Cell inhibition					
	4c		4i		6i	
	%	Mean $\pm$ S.D	%	Mean $\pm$ S.D	%	Mean $\pm$ S.D
5	3.7	0.341 $\pm$ 0.00406	12.1	0.311 $\pm$ 0.00435	25.7	0.263 $\pm$ 0.00491
10	10.5	0.317 $\pm$ 0.00296	28.1	0.255 $\pm$ 0.00648	45.8	0.192 $\pm$ 0.00322
20	18.6	0.288 $\pm$ 0.00586	47.5	0.186 $\pm$ 0.00435	62.4	0.133 $\pm$ 0.00296
30	30.8	0.245 $\pm$ 0.00346	56.2	0.155 $\pm$ 0.00529	67.8	0.114 $\pm$ 0.00291
40	46.6	0.189 $\pm$ 0.00462	62.4	0.133 $\pm$ 0.00520	75.9	0.085 $\pm$ 0.00364
50	59.6	0.143 $\pm$ 0.00230	78.8	0.075 $\pm$ 0.00491	83.9	0.057 $\pm$ 0.00462
Free energy of binding (kcal/mol)	4c		4i		6i	
	-14.55		-16.13		-18.43	

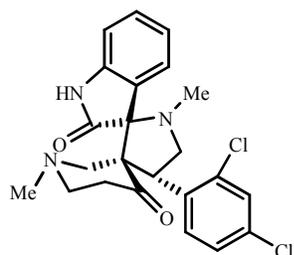
Figure 1



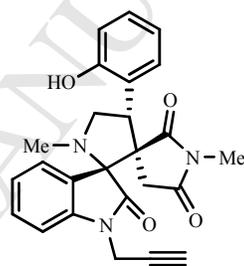
antituberculous agent



antimalarial agent



antimycobacterial agent



antimicrobial agent

Figure 2

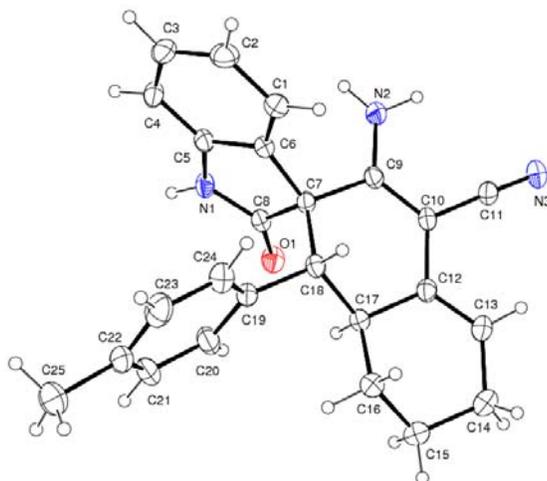
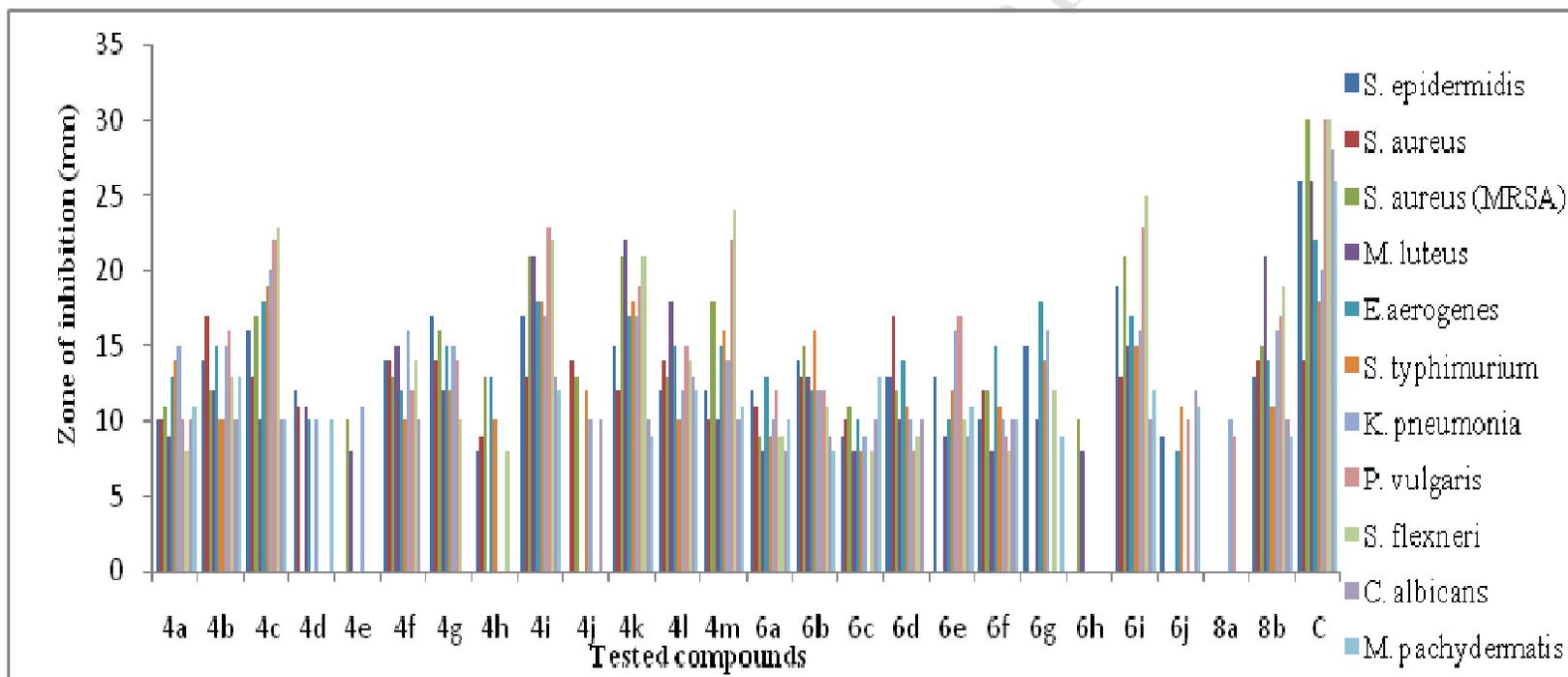


Figure 3



C = Streptomycin for Bacteria, Ketoconazole for fungi

Figure 4

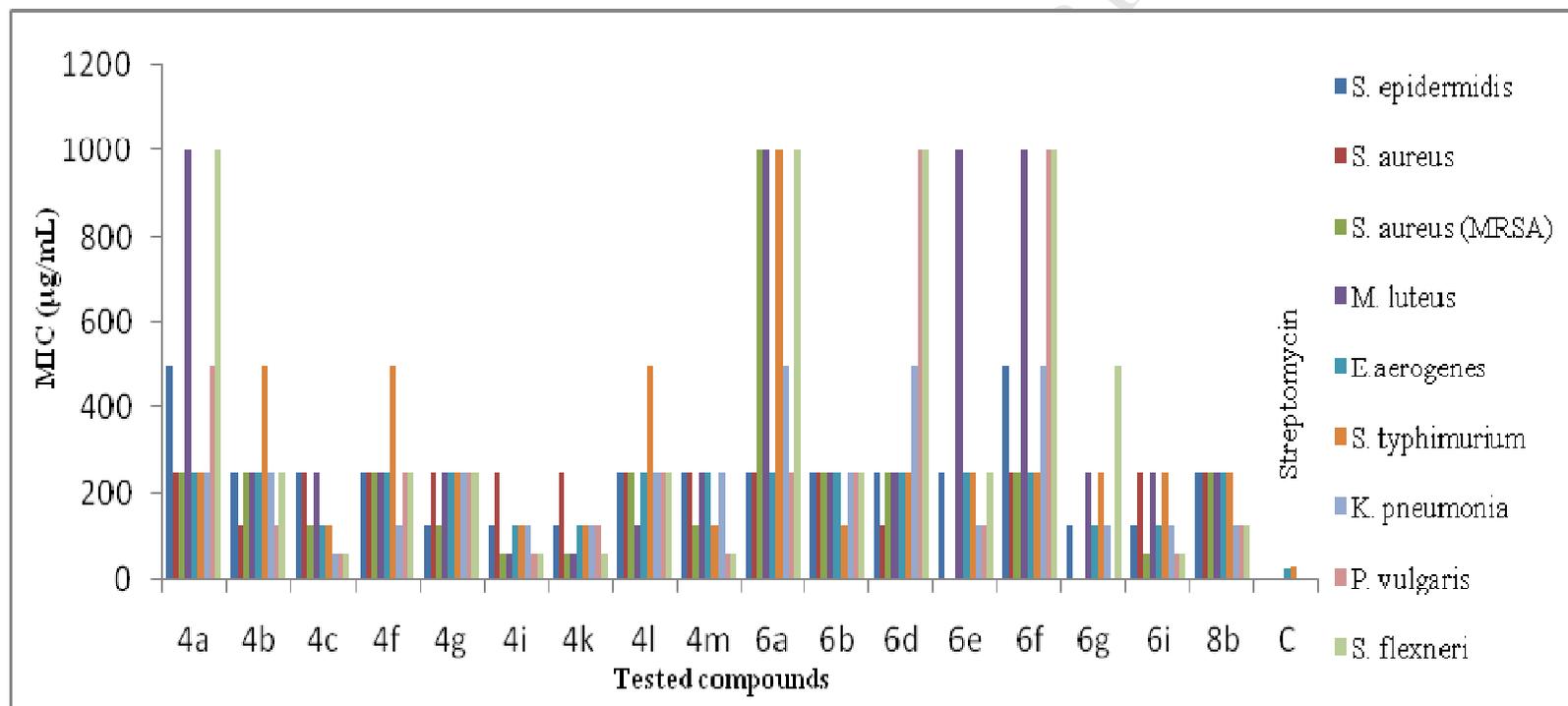


Figure 5a &amp; 5b

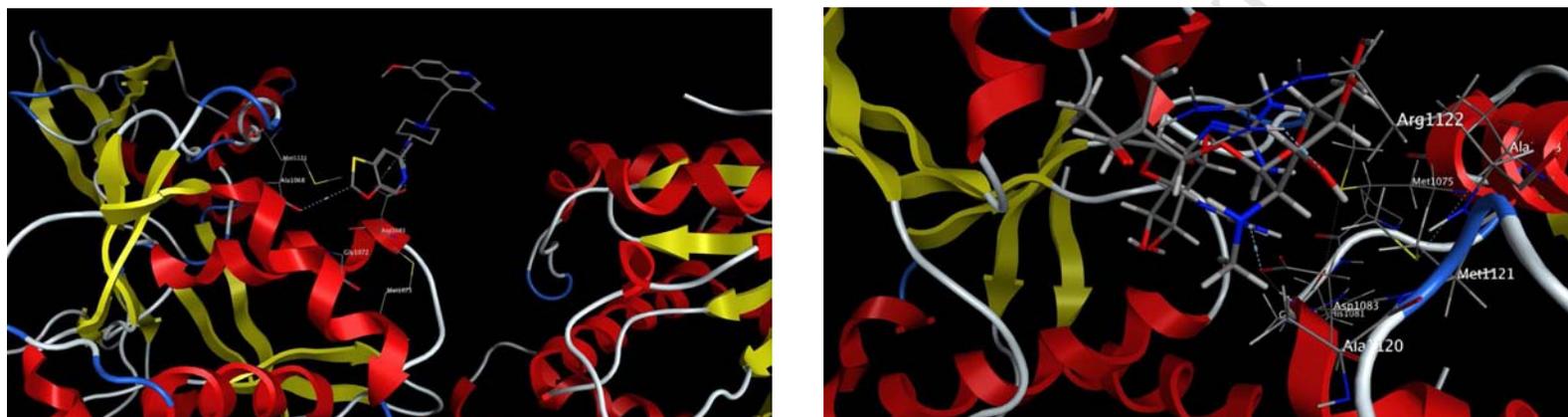
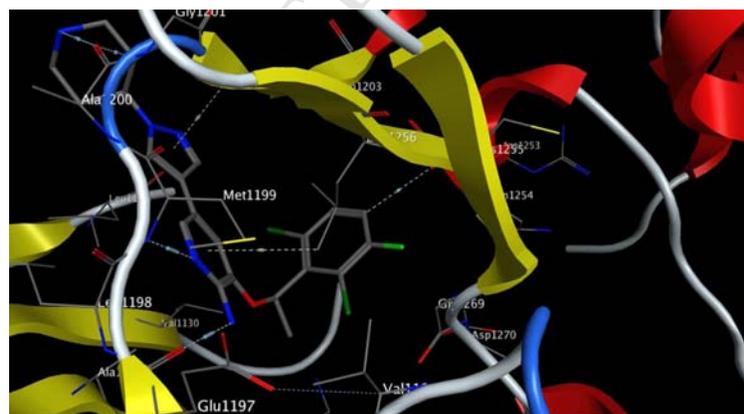
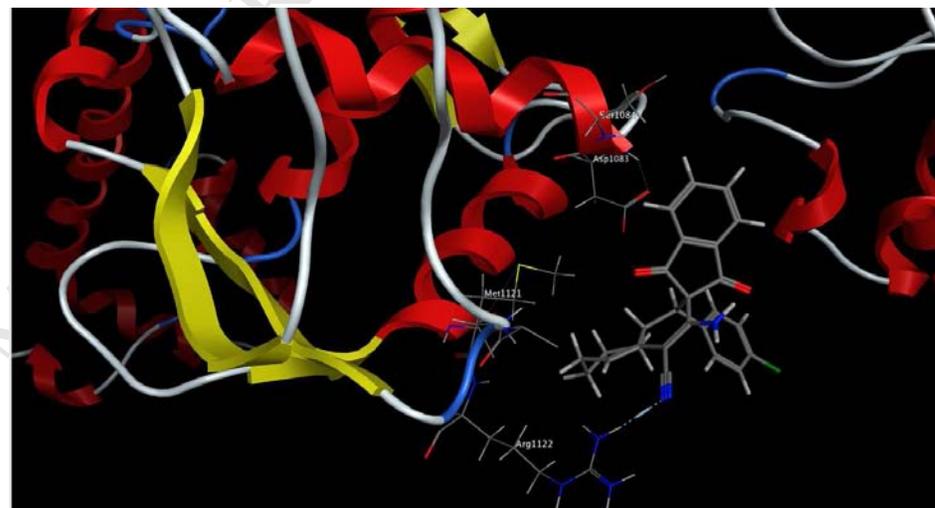
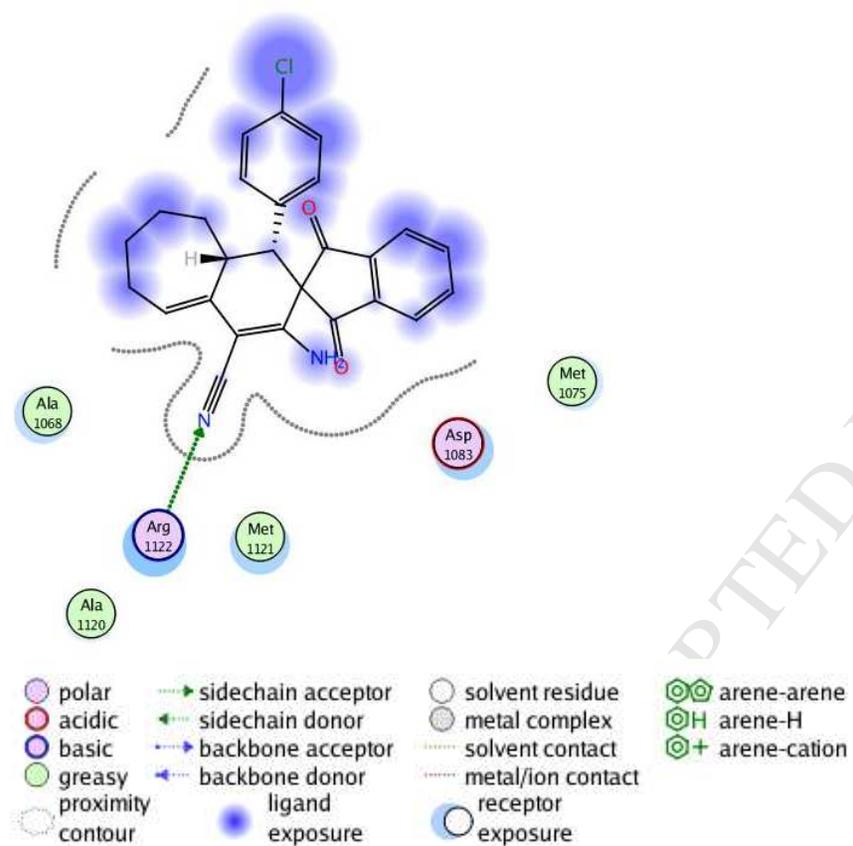


Figure 6



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Figure 7



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Figure 8

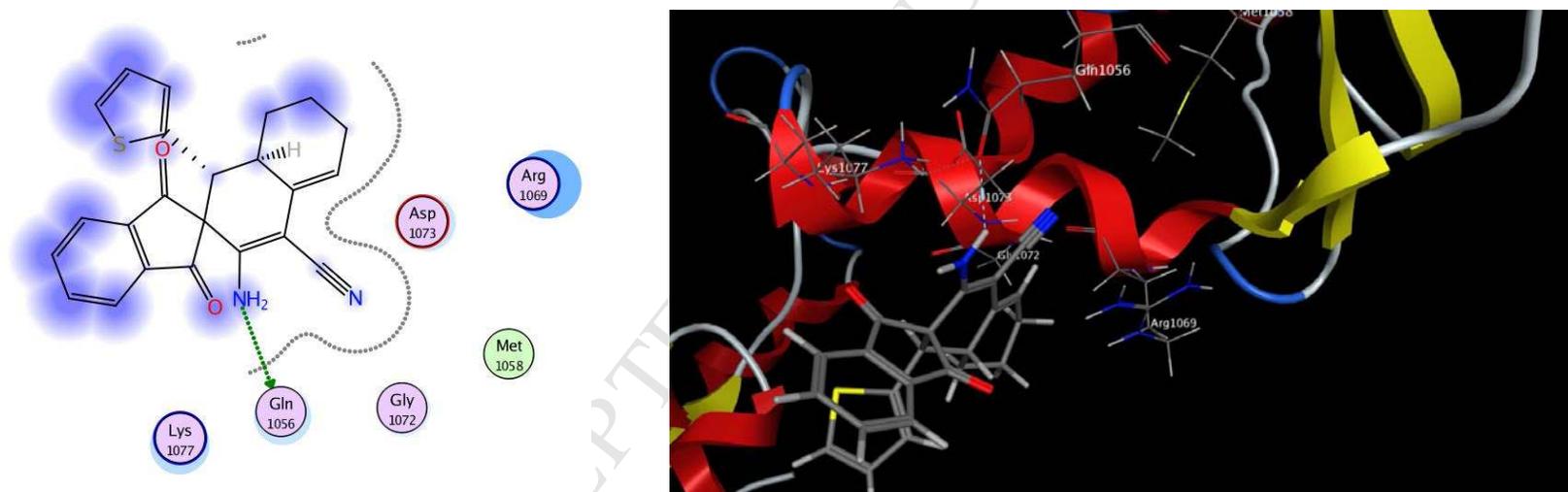


Figure 9



Figure 10

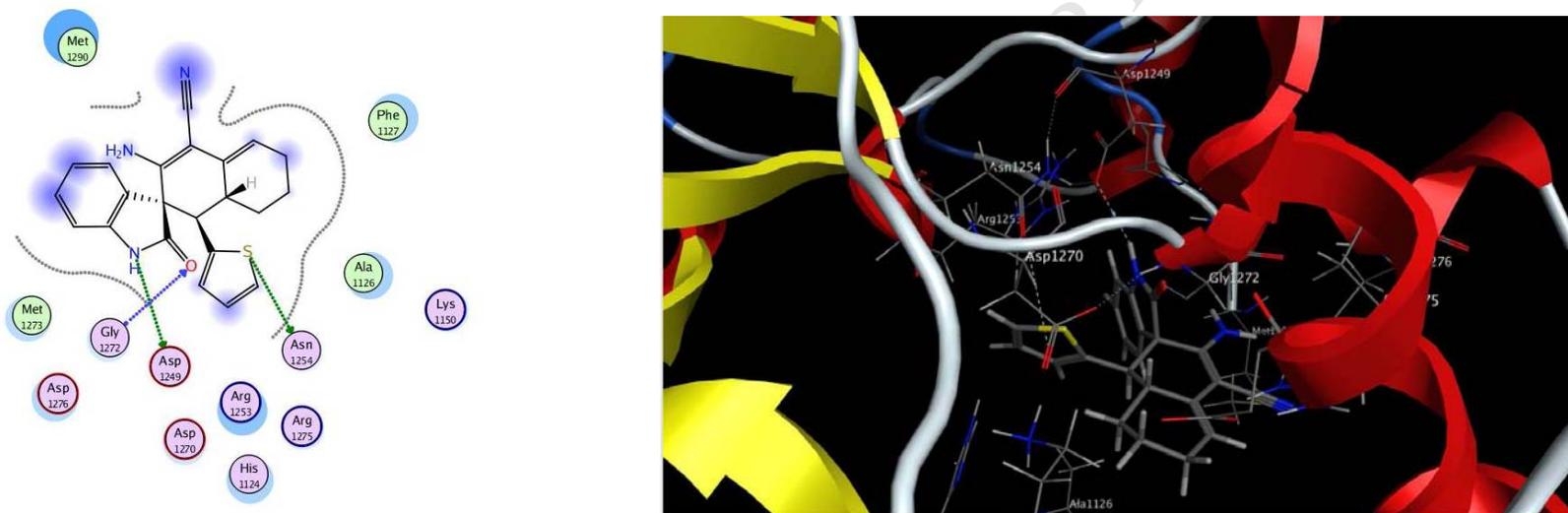
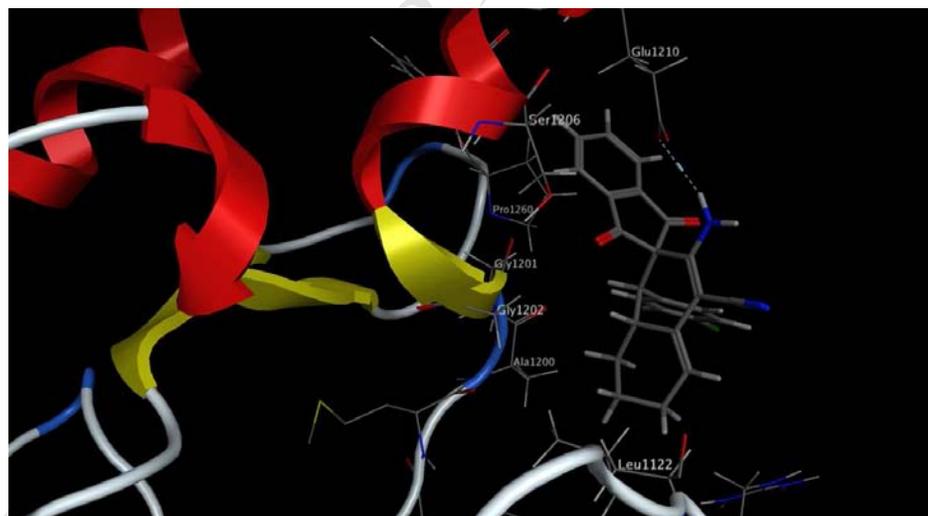
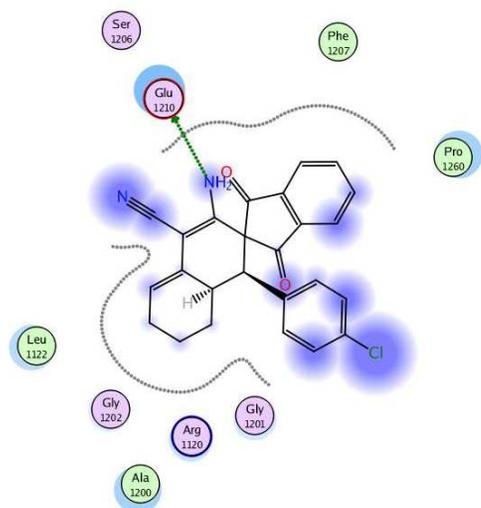
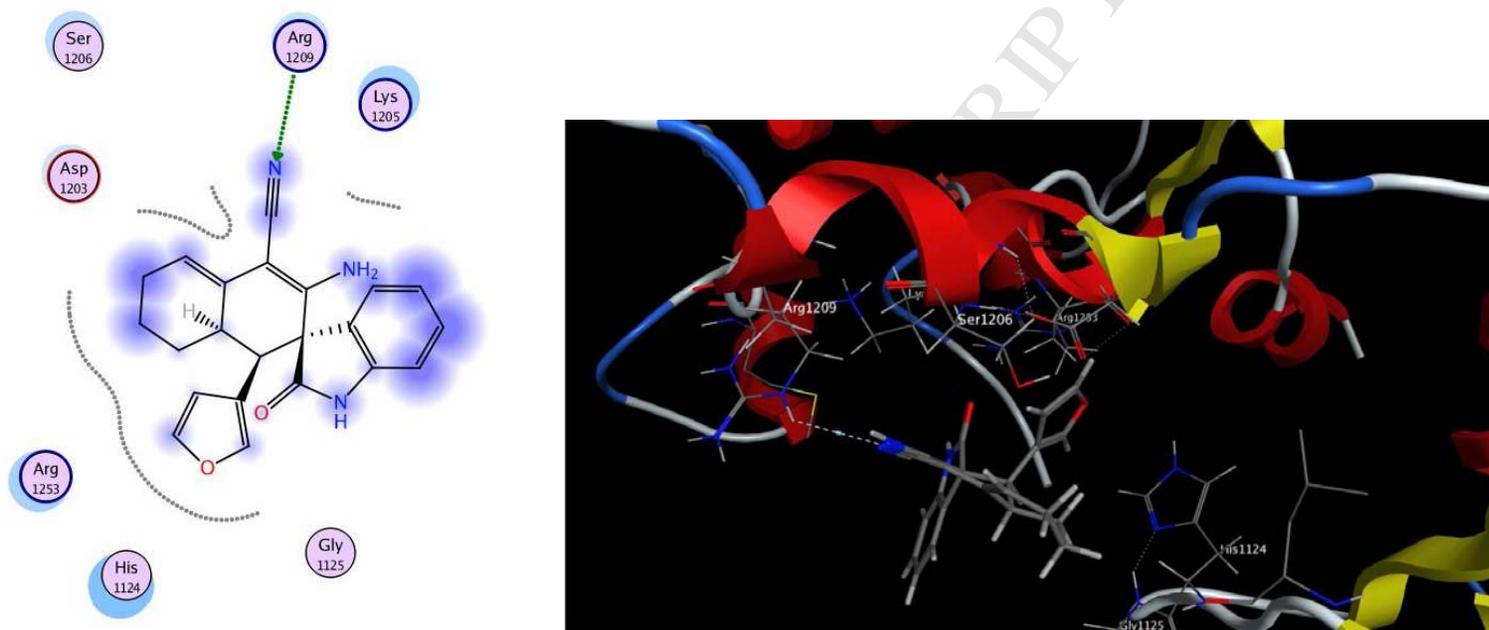


Figure 11



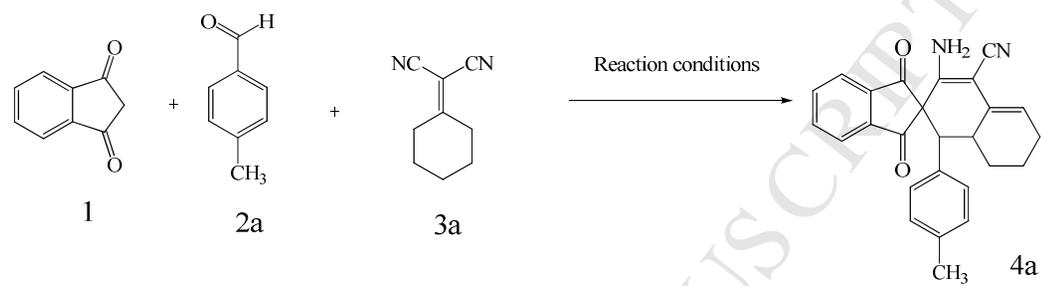
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Figure 12

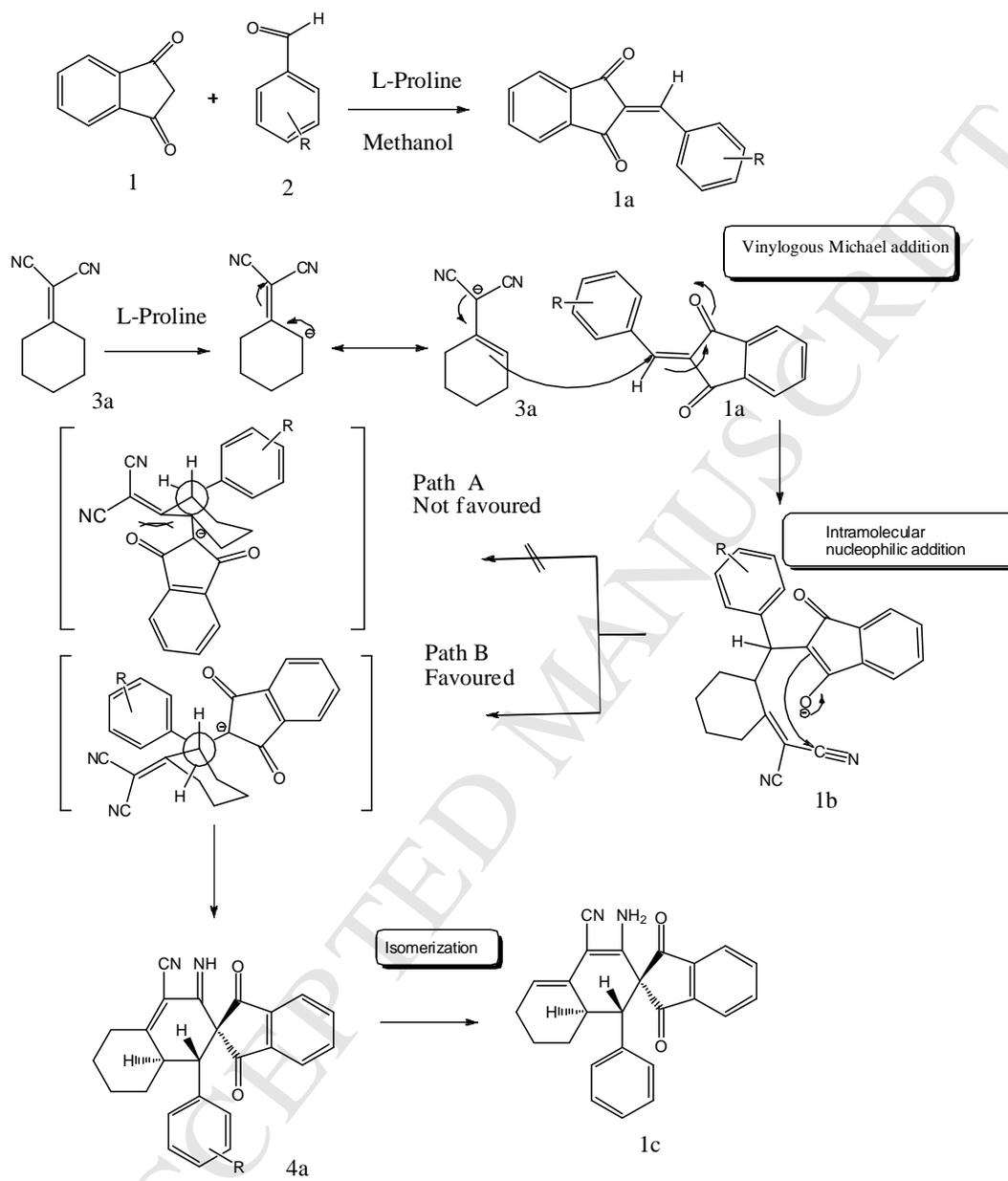


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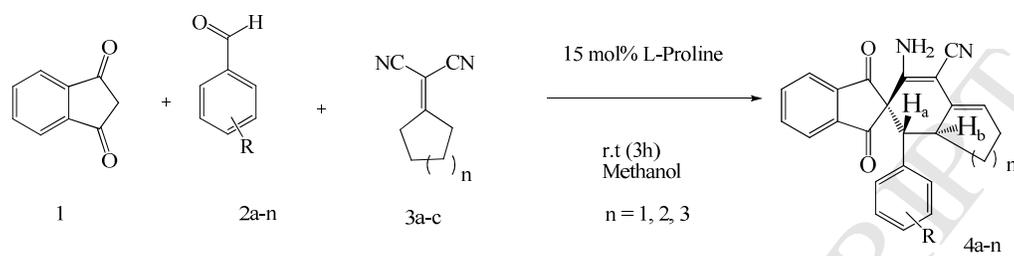
## Scheme 1



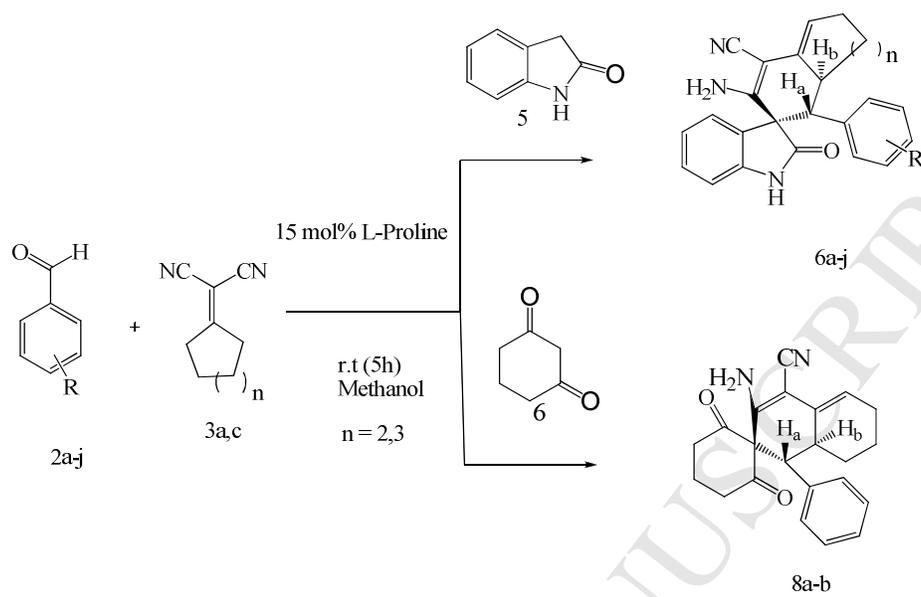
## Scheme 2



## Scheme 3



## Scheme 4



**Figures and Schemes captions**

**Figure 1.** Some biologically important spirooxindole compounds.

**Figure 2.** ORTEP diagram of synthesized compound **6a**.

**Figure 3.** Comparison of antimicrobial activity of synthesized compounds and standard drugs.

**Figure 4.** Comparison of MIC ( $\mu\text{g/ml}$ ) values of synthesized compounds and standard drugs.

**Figure 5.** Docking of co-crystallized ligand and standard drug Streptomycin with the DNA gyrase receptor for method validation

**Figure 6.** Docking of co-crystallized ligand (crizotinib) with the ALK receptor for method validation

**Figure 7.** 2D and 3D binding mode of most active compound **4i** (FEB = -11.64 kcal/mol) with DNA gyrase receptor.

**Figure 8.** 2D and 3D binding mode of moderate active compound **4m** (FEB = -9.51 kcal/mol) with DNA gyrase receptor.

**Figure 9.** 2D and 3D binding mode of least active compound **6h** (FEB = -8.14 kcal/mol) with DNA gyrase receptor.

**Figure 10.** 2D and 3D binding mode of most active compound **6i** (FEB = -18.43 kcal/mol) with ALK receptor.

**Figure 11.** 2D and 3D binding mode of intermediate active compound **4h** (FEB = -12.58 kcal/mol) with ALK receptor.

**Figure 12.** 2D and 3D binding mode of intermediate active compound **6h** (FEB = -11.23 kcal/mol) with ALK receptor.

**Scheme 1.** Synthesis of spirocarbocycle.

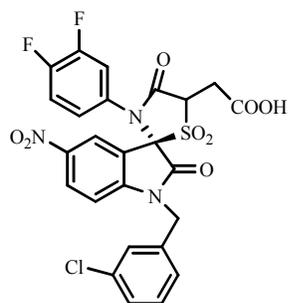
**Scheme 2.** Plausible mechanism for the formation of spirocarbocycles.

**Scheme 3.** Synthesis of spirocarbocycle derivatives from 1, 3-indandione, substituted aldehydes and alkylidene malononitrile.

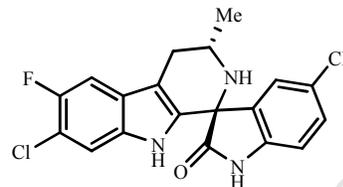
**Scheme 4.** Synthesis of spirooxindole derivatives from oxindole, substituted aldehydes and alkylidene malononitrile

**Scheme 5.** Synthesis of spirocarbocycle derivatives from 1,3-cyclohexanedione, substituted aldehydes and alkylidene malononitrile.

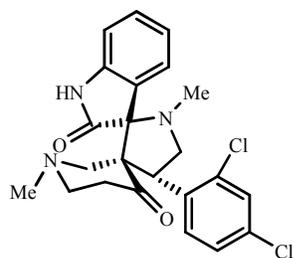
Figure 1



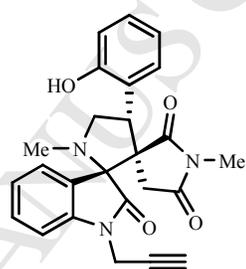
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antimalarial agent



antimycobacterial agent



antimicrobial agent

Figure 2

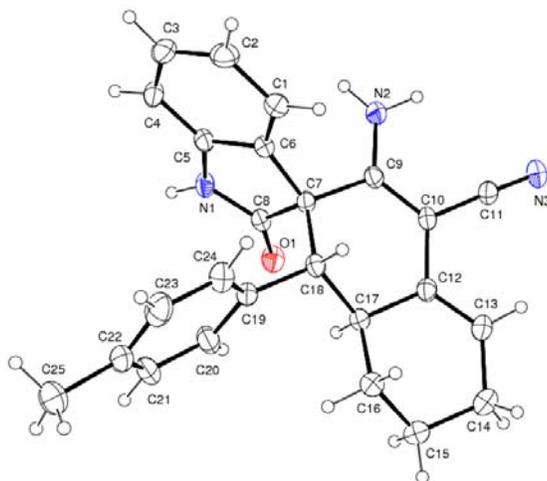
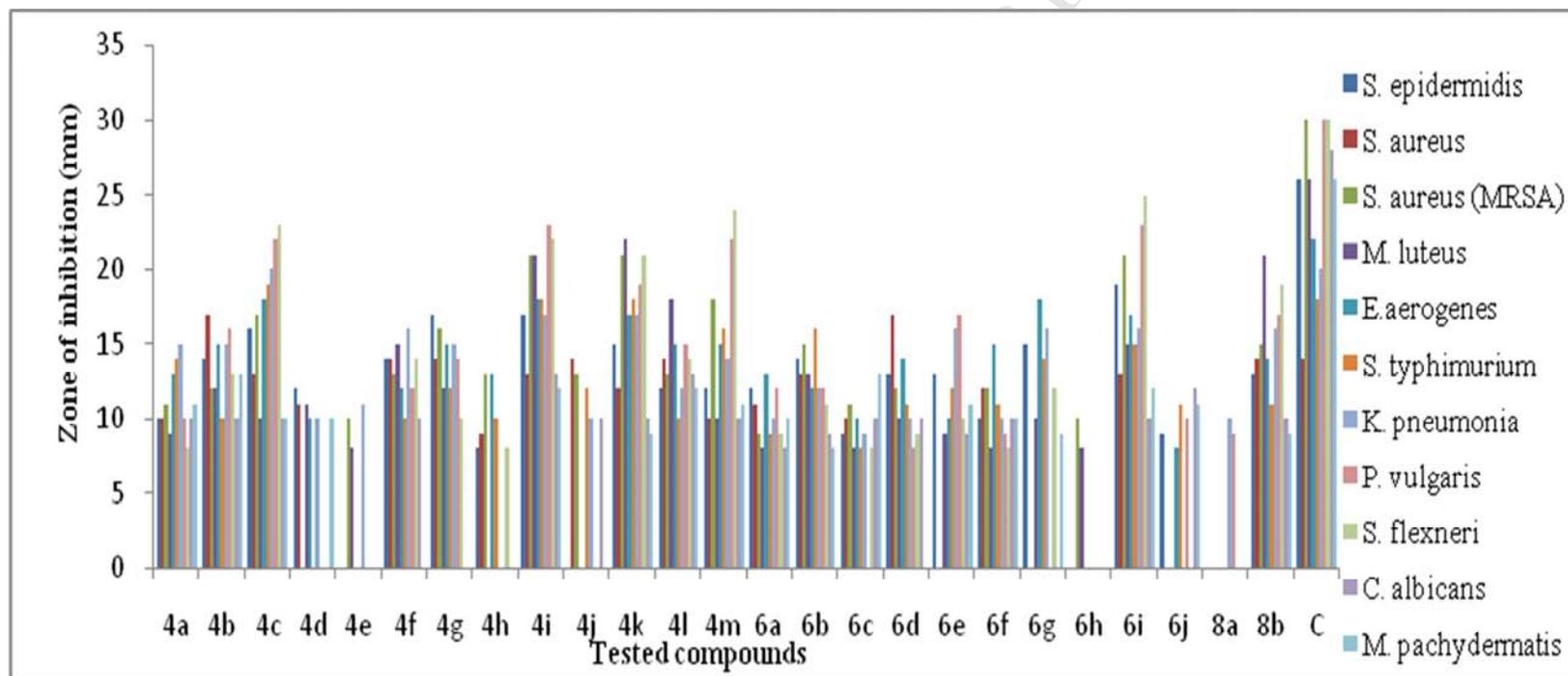
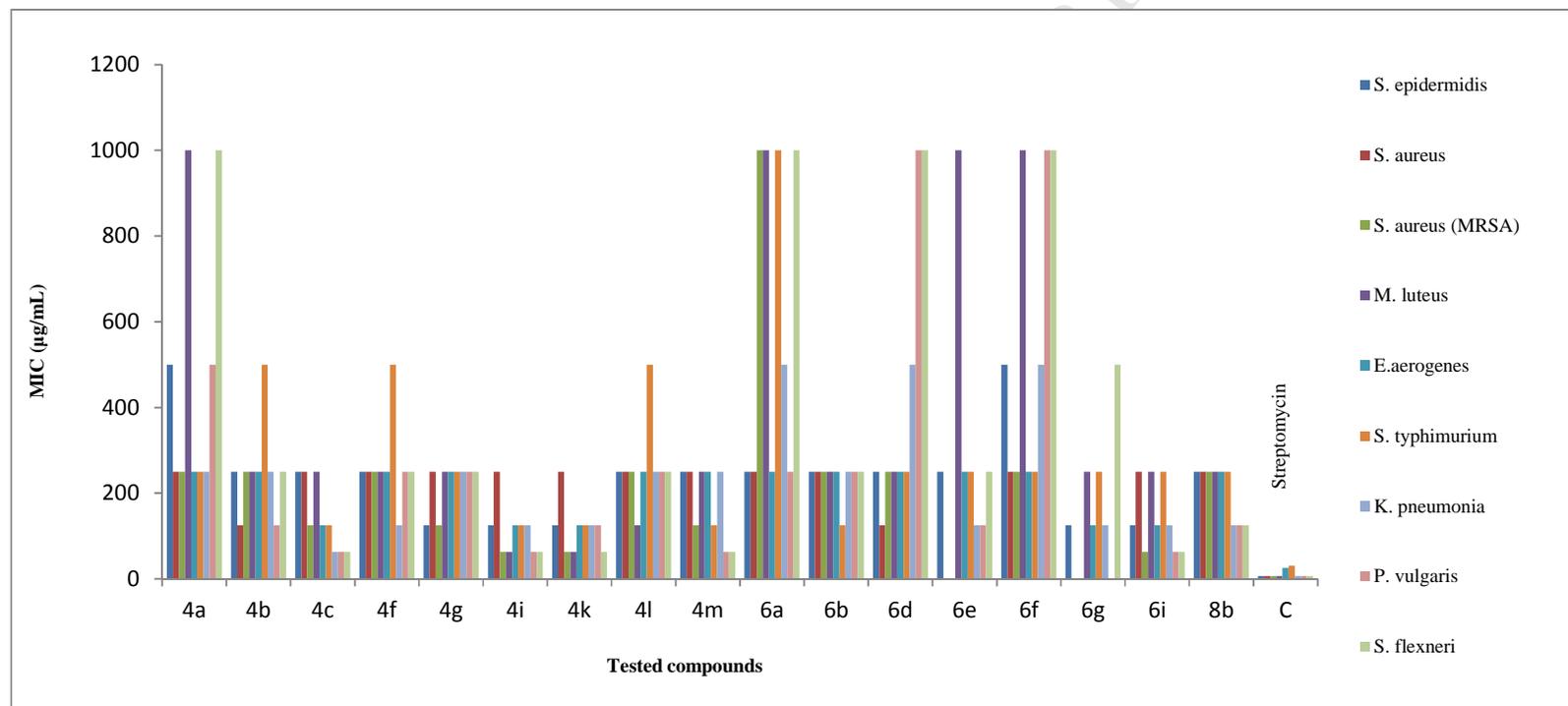


Figure 3



C = Streptomycin for Bacteria, Ketoconazole for fungi

Figure 4



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Figure 5a &amp; 5b

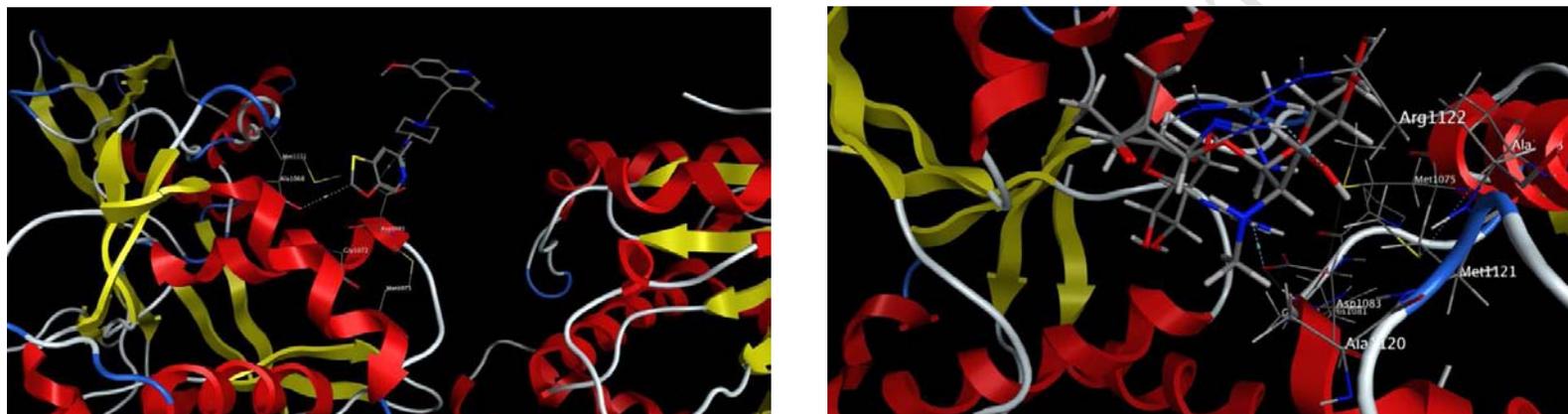
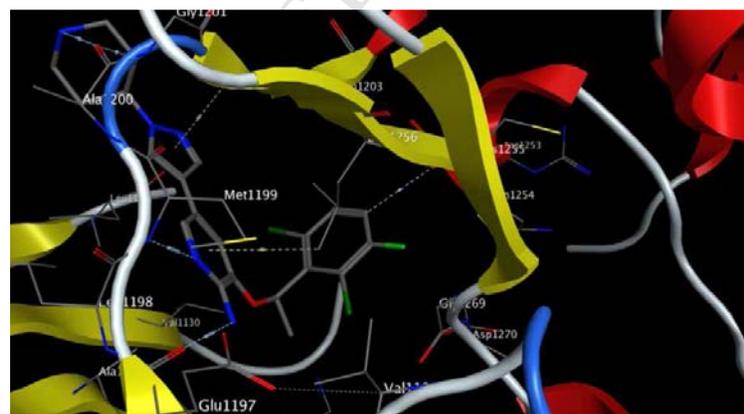


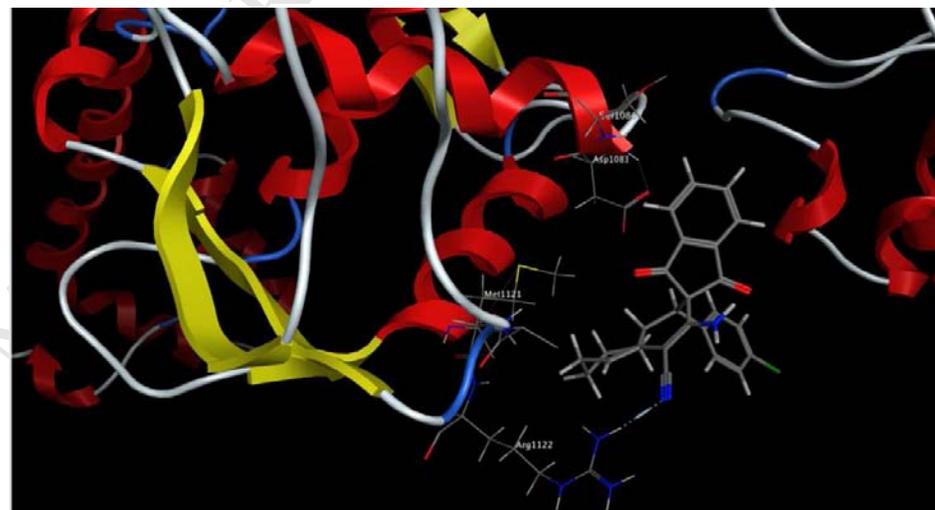
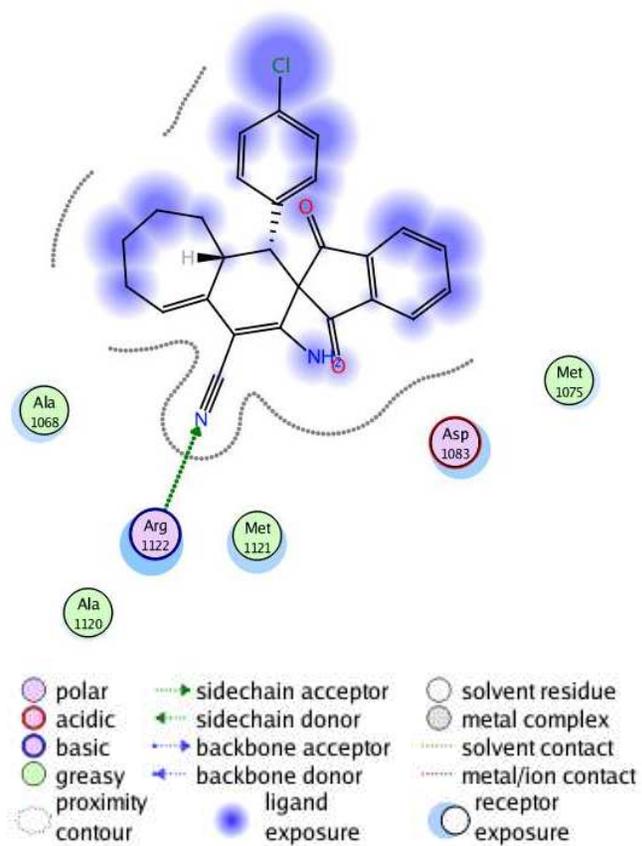
Figure 6



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Figure 7



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Figure 8

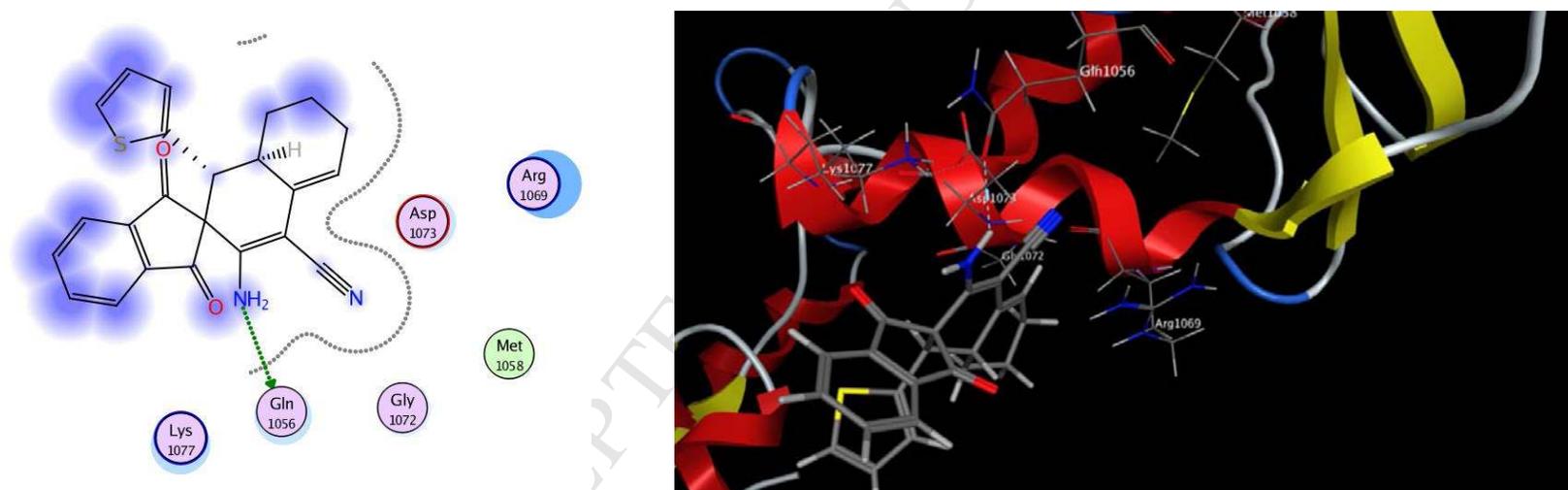


Figure 9

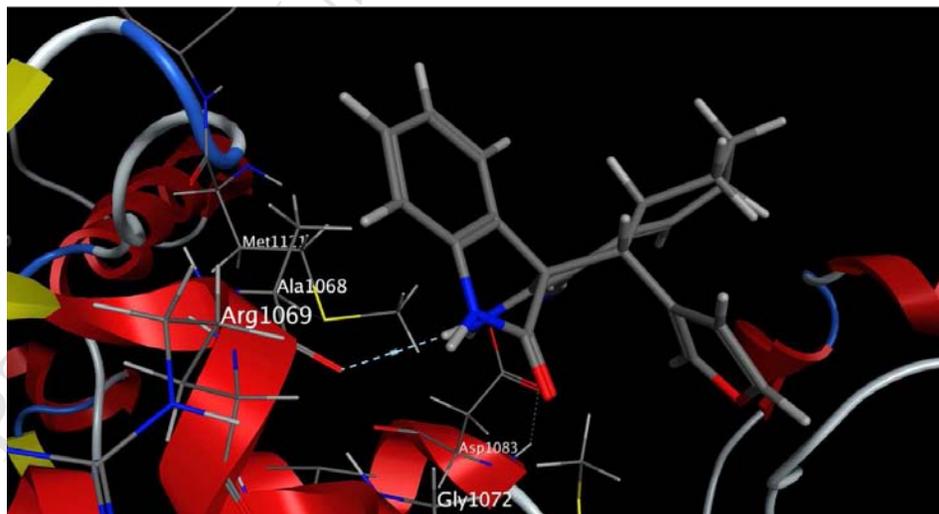
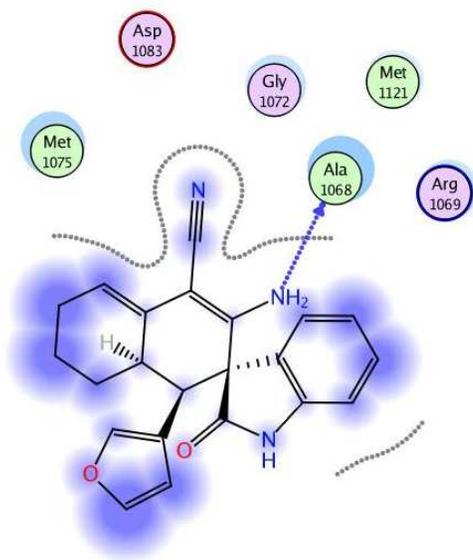
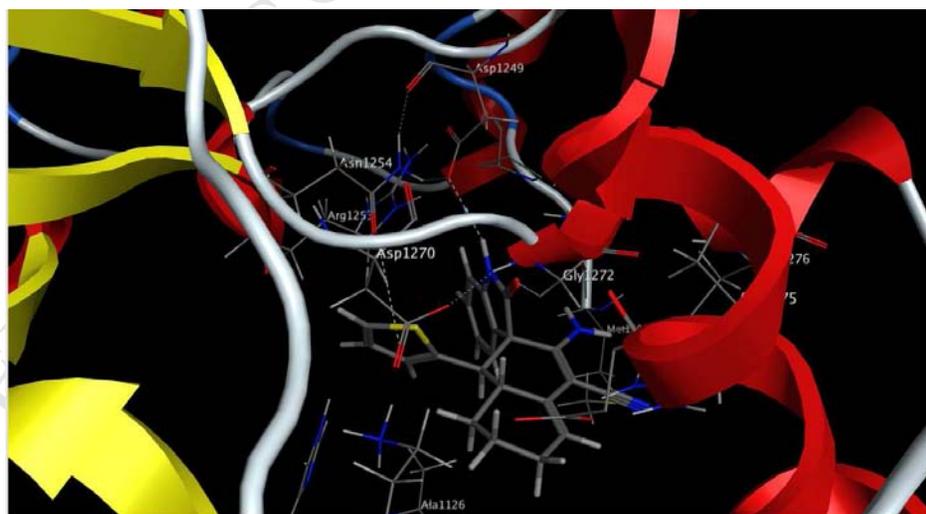
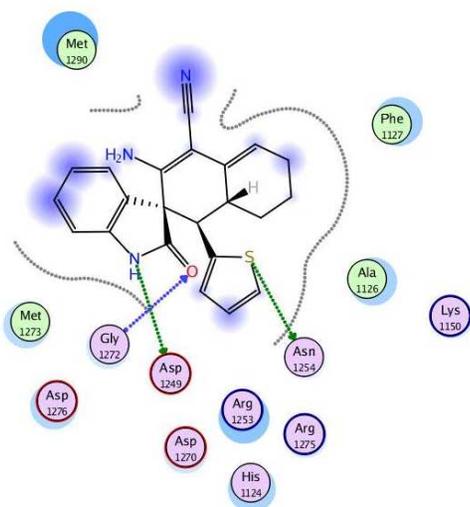
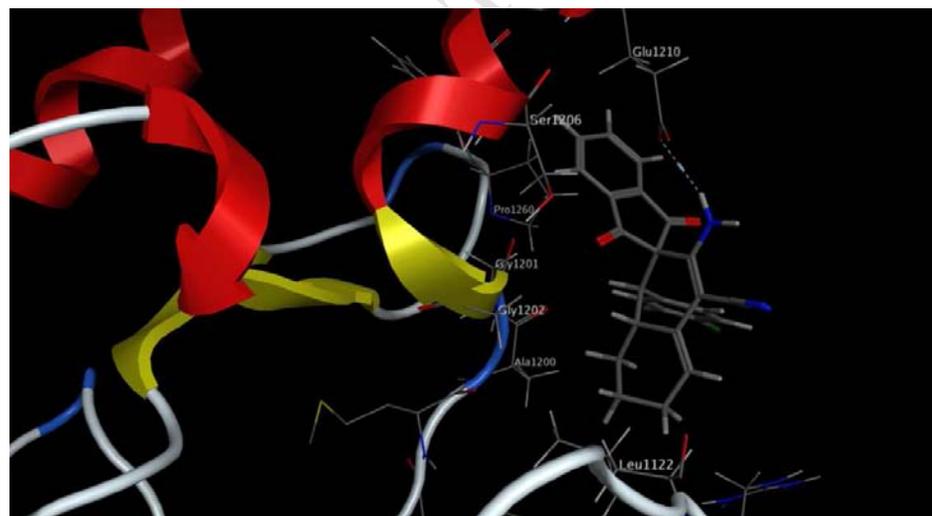
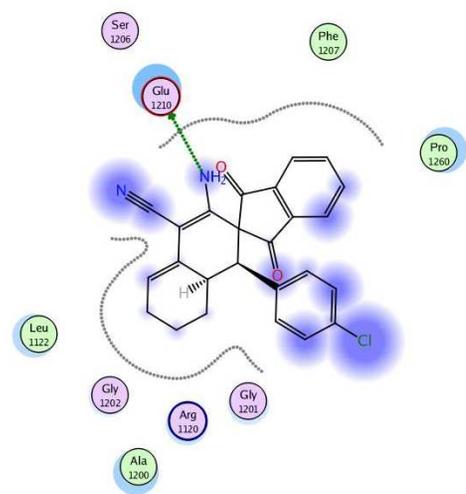


Figure 10



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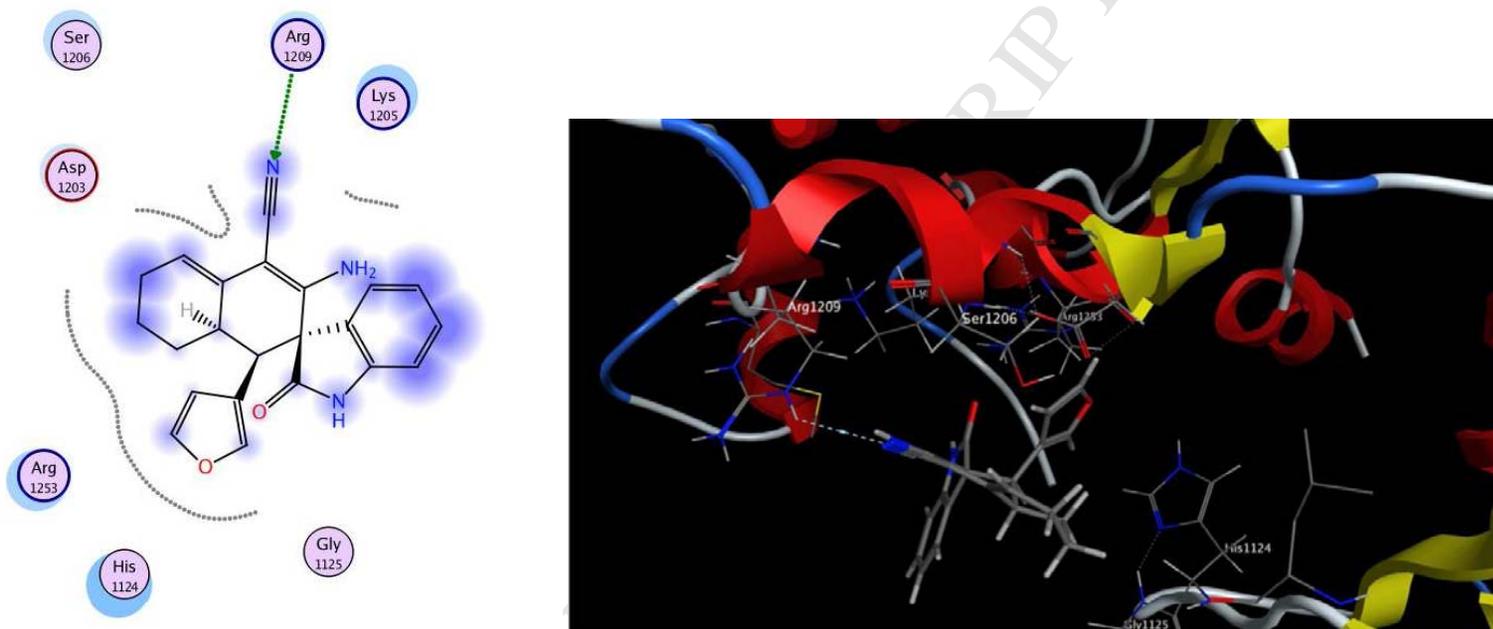
Figure 11



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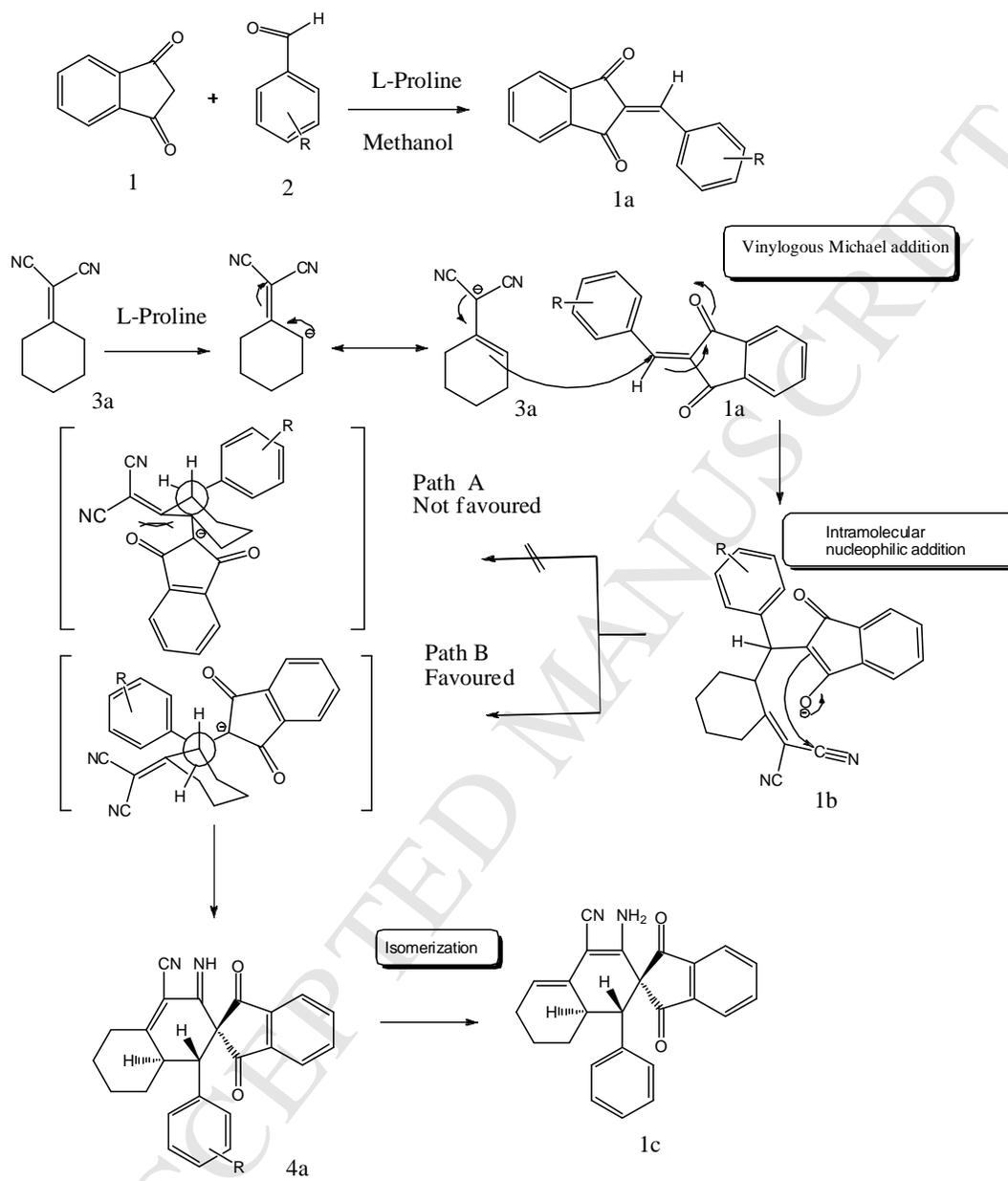
Figure 12



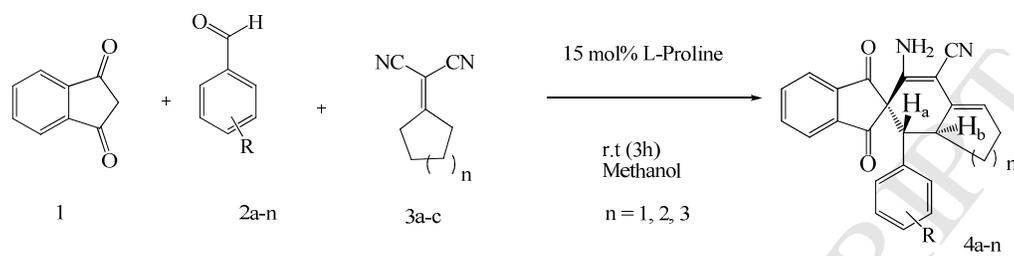
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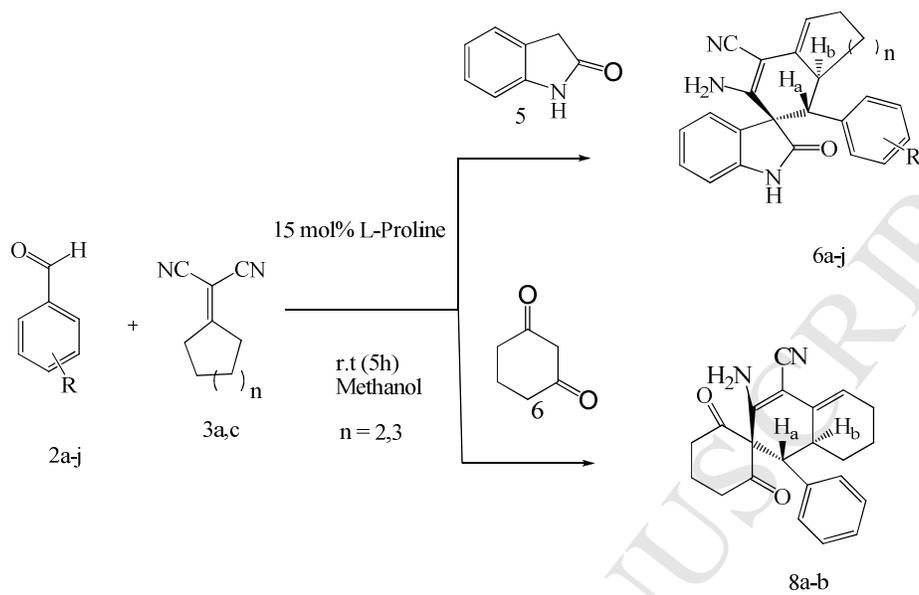
## Scheme 2



## Scheme 3



## Scheme 4



1 **Tables**

2 **Table 1.** Optimization of reaction conditions for the preparation of spirocarbocycles.

3 **Table 2.** Synthesis of spirocarbocycle derivatives from 1, 3-indandione, substituted  
4 aldehydes and alkylidene malononitrile.

5 **Table 3.** Synthesis of spirooxindole derivatives from oxindole or 1,3-cyclohexanedione,  
6 substituted aldehydes and alkylidene malononitrile.

7 **Table 4.** Synthesis of spirocarbocycle derivatives from substituted aldehydes and alkylidene  
8 malononitrile.

9 **Table 4 .** Crystal data and structure refinement parameters for compound **6a**.

10 **Table 5.** *In vitro* antimicrobial activity of synthesized compounds.

11 **Table 6.** MIC ( $\mu\text{g/ml}$ ) of compounds tested against bacteria.

12 **Table 7.** Anticancer activity of synthesised compounds against A549 cancer cell line and  
13 calculated binding energy with ALK receptor of synthesized spirocarbocycles.

14

1

**Table 1**

Entry	Solvent	Catalyst (10 mol%)	Conditions	Time (h)	Yield <sup>a</sup> (%)
1.	MeOH	-	r.t.	48	40
2.	EtOH	K <sub>2</sub> CO <sub>3</sub>	r.t.	6	62
3.	MeOH	Et <sub>3</sub> N	r.t.	4	80
4.	EtOH	DABCO	Reflux	6	54
5.	EtOH	NaOEt	r.t.	5	67
6.	EtOH	L-proline	r.t.	3	85
7.	MeOH	L-proline	r.t.	3	86
<b>8.</b>	<b>MeOH</b>	<b>L-proline</b>	<b>r.t.</b>	<b>3</b>	<b>89<sup>b</sup></b>
9.	MeOH	L-proline	r.t.	3	87 <sup>c</sup>
10.	CH <sub>3</sub> CN	L-proline	r.t.	5	69
11.	H <sub>2</sub> O	L-proline	r.t.	6	72
12.	DMF	L-proline	r.t.	8	58

2 [a] Isolated yield of 4a after recrystallization [b]The reaction was performed using 15 mol%  
3 of the catalyst [c]The reaction was performed using 20 mol% of the catalyst.

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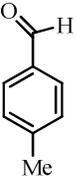
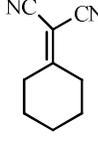
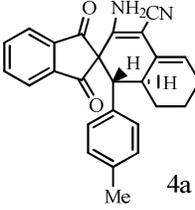
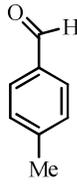
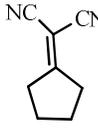
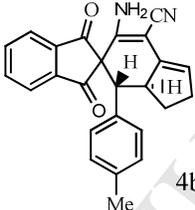
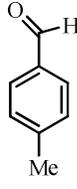
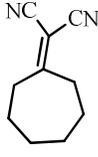
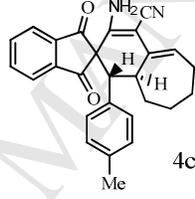
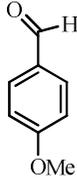
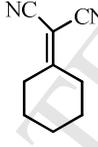
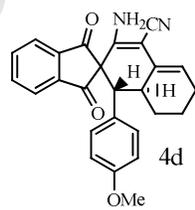
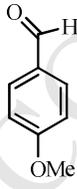
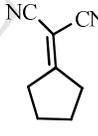
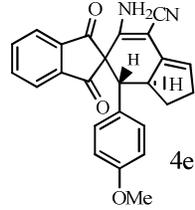
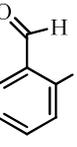
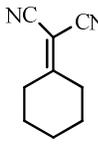
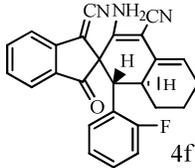
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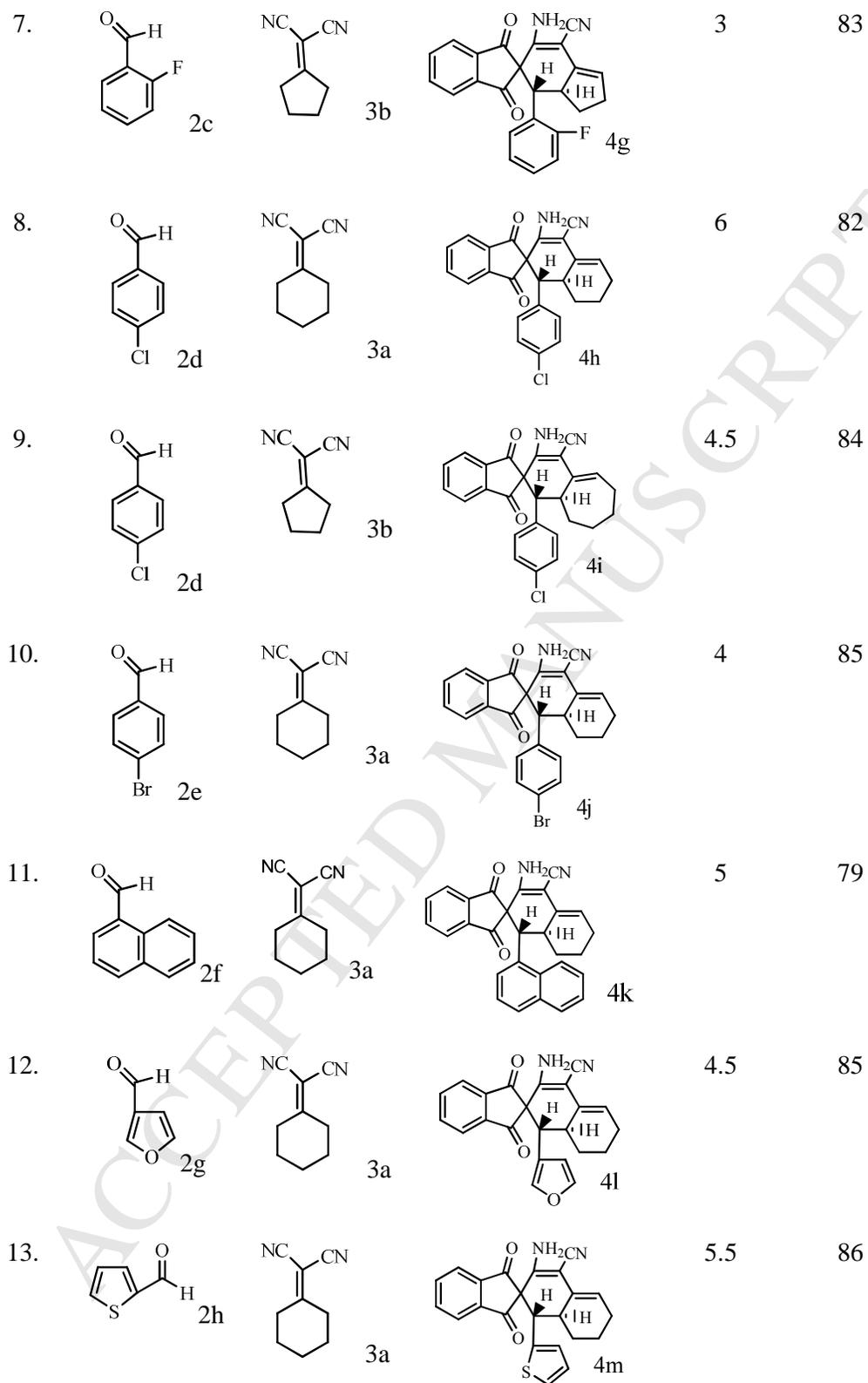
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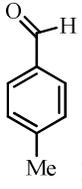
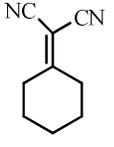
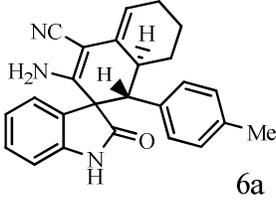
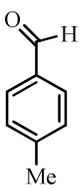
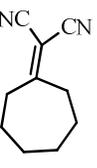
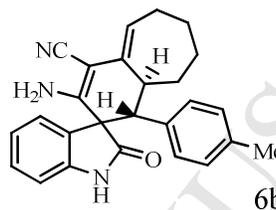
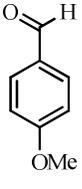
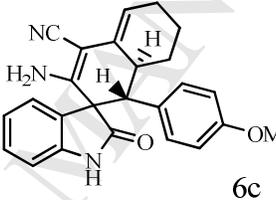
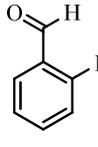
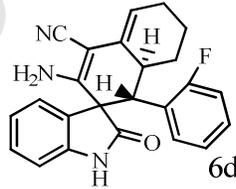
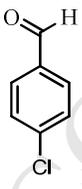
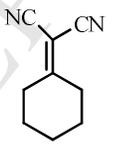
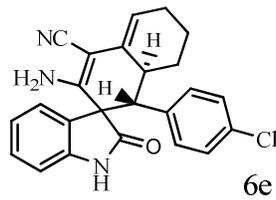
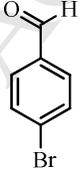
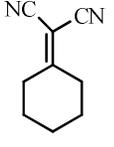
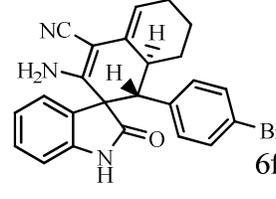
Table 2

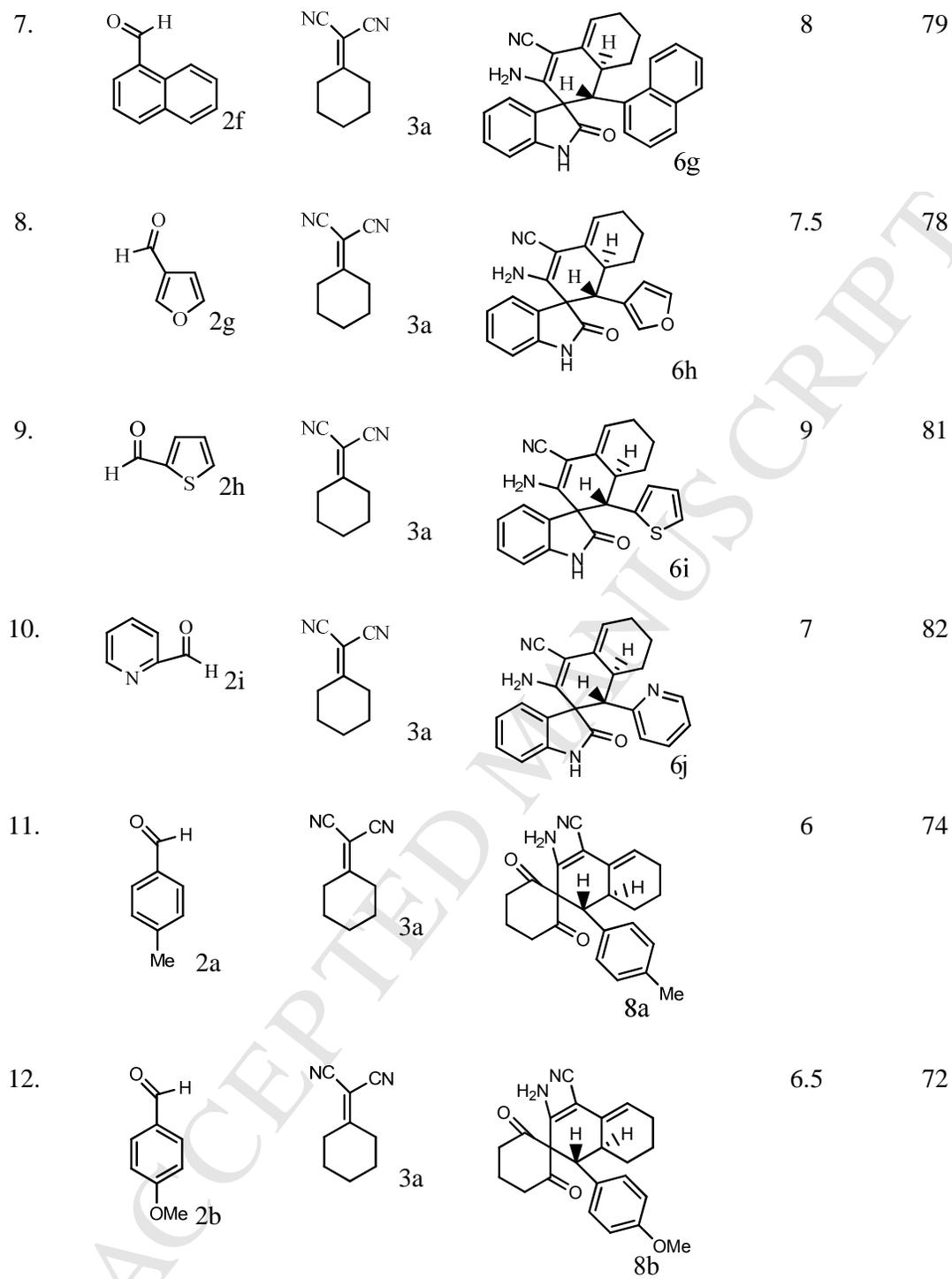
Entry	Aldehyde	Vinylogous malononitrile	Product	Time	Yield
1.	 Me 2a	 3a	 4a	3	89
2.	 Me 2a	 3b	 4b	2.5	90
3.	 Me 2a	 3c	 4c	4	86
4.	 OMe 2b	 3a	 4d	3.5	87
5.	 OMe 2b	 3b	 4e	2.5	88
6.	 2c	 3a	 4f	4	81



1

Table 3

Entry	Aldehyde/ketone	Vinylogous malononitrile	Product	Time	Yield
1.	 2a	 3a	 6a	5	86
2.	 2a	 3c	 6b	6	83
3.	 2b	 3a	 6c	7	84
4.	 2c	 3a	 6d	8	78
5.	 2d	 3a	 6e	10	81
6.	 2e	 3a	 6f	9	80



1

2

3

4

1

**Table 4**

2 Crystal data and structure refinement parameters for compound 6a

Empirical formula	$C_{25}H_{23}N_3O$
Formula weight	381.46
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P21/n
Unit cell dimensions	$a = 9.3858(3)$ Å; $b = 21.8349(6)$ Å; $c = 9.6525(2)$ Å $\alpha = \gamma = 90^\circ$ $\beta = 93.3610(10)^\circ$
Volume	1974.76(9) Å <sup>3</sup>
Z, Calculated density	4, 1.283 Mgm <sup>-3</sup>
Absorption coefficient	0.080 mm <sup>-1</sup>
F(000)	808
Crystal size	0.35 x 0.35 x 0.30 mm
$\theta$ range for data collection	2.31 to 25.00°
Limiting indices	-11 ≤ h ≤ 11, -25 ≤ k ≤ 25, -11 ≤ l ≤ 11
Reflections collected / unique	18019 / 3486 [R(int) = 0.0335]
Completeness to $\theta = 25.00$	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9865 and 0.9632
Refinement method	Full-matrix least-squares on F <sup>2</sup>
Data / restraints / parameters	3486 / 0 / 275
Goodness-of-fit on F <sup>2</sup>	1.030
Final R indices [I > 2σ(I)]	R1 = 0.0400, wR2 = 0.0965

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R indices (all data)	$R_1 = 0.0587, wR_2 = 0.1072$
Extinction coefficient	0.0079(11)
Largest diff. peak and hole	0.240 and -0.158 e.Å <sup>-3</sup>
CCDC	972722

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Table 5

Compounds	Zone of inhibition in mm										
	Gram positive bacteria				Gram negative bacteria					Fungi	
	<i>S. epidermidis</i>	<i>S. aureus</i>	<i>S. aureus (MRSA)</i>	<i>M. luteus</i>	<i>E. aerogens</i>	<i>S. typhimurium</i>	<i>K. pneumonia</i>	<i>P. vulgaris</i>	<i>S. flexneri</i>	<i>C. albicans</i>	<i>M. pachydermatis</i>
4a	10	10	11	9	13	14	15	10	8	10	11
4b	14	17	12	12	15	10	15	16	13	10	13
4c	16	13	17	10	18	19	20	22	23	10	10
4d	12	11	NI	11	10	NI	10	NI	NI	NI	10
4e	NI	NI	10	8	NI	NI	11	NI	NI	NI	NI

4f	14	14	13	15	12	10	16	12	14	10	NI
4g	17	14	16	12	15	12	15	14	10	NI	NI
4h	8	9	13	NI	13	10	NI	NI	8	NI	NI
4i	17	13	21	21	18	18	17	23	22	13	12
4j	NI	14	13	NI	NI	12	10	NI	NI	10	NI
4k	15	12	21	22	17	18	17	19	21	10	9
4l	12	14	13	18	15	10	12	15	14	13	12
4m	12	10	18	10	15	16	14	22	24	10	11
6a	12	11	9	8	13	9	10	12	9	8	10
6b	14	13	15	13	12	16	12	12	11	9	8
6c	9	10	11	8	10	8	9	NI	8	10	13
6d	13	17	12	10	14	11	10	8	9	10	NI

6e	13	NI	NI	9	10	12	16	17	10	9	11
6f	10	12	12	8	15	11	10	9	8	10	10
6g	15	NI	NI	10	18	14	16	NI	12	NI	9
6h	NI	NI	10	8	NI						
6i	19	13	21	15	17	15	16	23	25	10	12
6j	9	NI	NI	NI	8	11	NI	10	NI	12	11
8a	NI	NI	NI	NI	NI	NI	10	9	NI	NI	NI
8b	13	14	15	21	14	11	16	17	19	10	9
Streptomycin	26	14	30	26	22	18	20	30	30	NA	NA
Ketoconazole	NA	28	26								

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NA-not applicable NI – no inhibition.

Table 6

Compounds	Minimum inhibitory concentration ( $\mu\text{g/ml}$ )								
	Gram positive bacteria				Gram negative bacteria				
	<i>S. epidermidis</i>	<i>S. aureus</i>	<i>S. aureus</i> (MRSA)	<i>M.</i> <i>luteus</i>	<i>E. aerogens</i>	<i>S.</i> <i>typhimurium</i>	<i>K.</i> <i>pneumonia</i>	<i>P.</i> <i>vulgaris</i>	<i>S.</i> <i>flexneri</i>
4a	500	250	250	1000	250	250	250	500	1000
4b	250	125	250	250	250	500	250	125	250
4c	250	250	125	250	125	125	62.5	62.5	62.5
4f	250	250	250	250	250	500	125	250	250
4g	125	250	125	250	250	250	250	250	250
4i	125	250	62.5	62.5	125	125	125	62.5	62.5
4k	125	250	62.5	62.5	125	125	125	125	62.5

4l	250	250	250	125	250	500	250	250	250
4m	250	250	125	250	250	125	250	62.5	62.5
6a	250	250	1000	1000	250	1000	500	250	1000
6b	250	250	250	250	250	125	250	250	250
6d	250	125	250	250	250	250	500	1000	1000
6e	250	NI	NI	1000	250	250	125	125	250
6f	500	250	250	1000	250	250	500	1000	1000
6g	125	NI	NI	250	125	250	125	NI	500
6i	125	250	62.5	250	125	250	125	62.5	62.5
8b	250	250	250	250	250	250	125	125	125
Streptomycin	6.25	6.25	6.25	6.25	25	30	6.25	6.25	6.25

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 NI – no inhibition

**Table 7**

Concentration ( $\mu\text{g/mL}$ )	Cell inhibition					
	4c		4i		6i	
	%	Mean $\pm$ S.D	%	Mean $\pm$ S.D	%	Mean $\pm$ S.D
5	3.7	0.341 $\pm$ 0.00406	12.1	0.311 $\pm$ 0.00435	25.7	0.263 $\pm$ 0.00491
10	10.5	0.317 $\pm$ 0.00296	28.1	0.255 $\pm$ 0.00648	45.8	0.192 $\pm$ 0.00322
20	18.6	0.288 $\pm$ 0.00586	47.5	0.186 $\pm$ 0.00435	62.4	0.133 $\pm$ 0.00296
30	30.8	0.245 $\pm$ 0.00346	56.2	0.155 $\pm$ 0.00529	67.8	0.114 $\pm$ 0.00291
40	46.6	0.189 $\pm$ 0.00462	62.4	0.133 $\pm$ 0.00520	75.9	0.085 $\pm$ 0.00364
50	59.6	0.143 $\pm$ 0.00230	78.8	0.075 $\pm$ 0.00491	83.9	0.057 $\pm$ 0.00462
Free energy of binding	4c		4i		6i	
(kcal/mol)	-14.55		-16.13		-18.43	

**Synthesis of new class of spirocarbocycle derivatives by multicomponent domino reaction and their evaluation for antimicrobial activity and Molecular docking studies**

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<sup>a</sup>*Organic Chemistry Division, CSIR-Central Leather Research Institute, Adyar, Chennai 600 020, India*

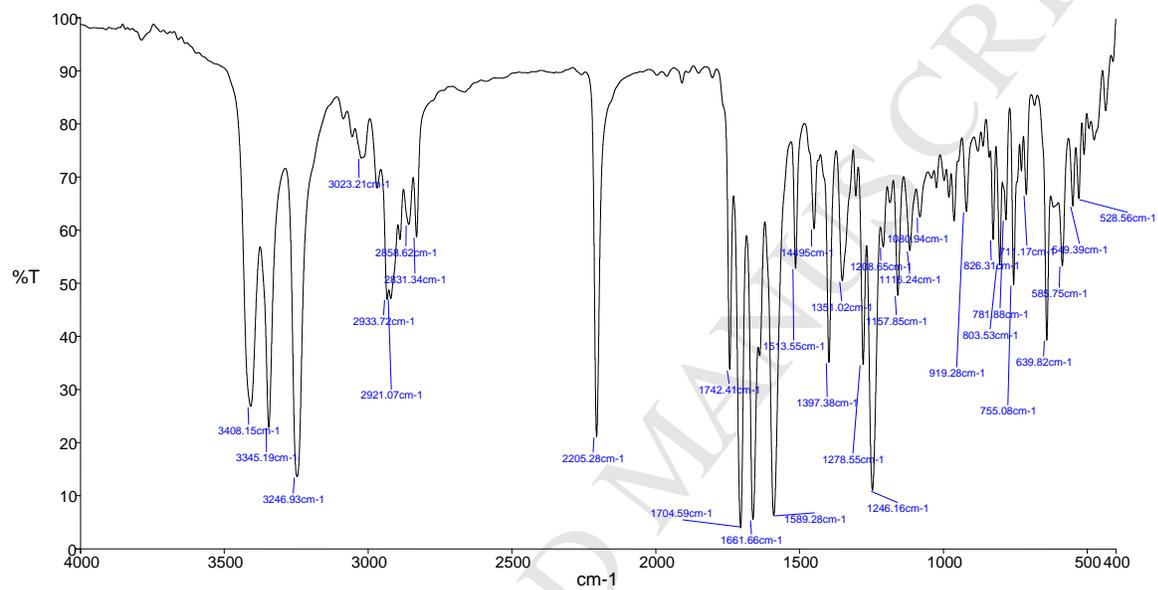
<sup>b</sup>*Division of Microbiology, Entomology Research Institute, Loyola College, Chennai 600 034, India*

<sup>c</sup>*Department of Biotechnology, Indian Institute of Technology, Chennai 600 036, India*

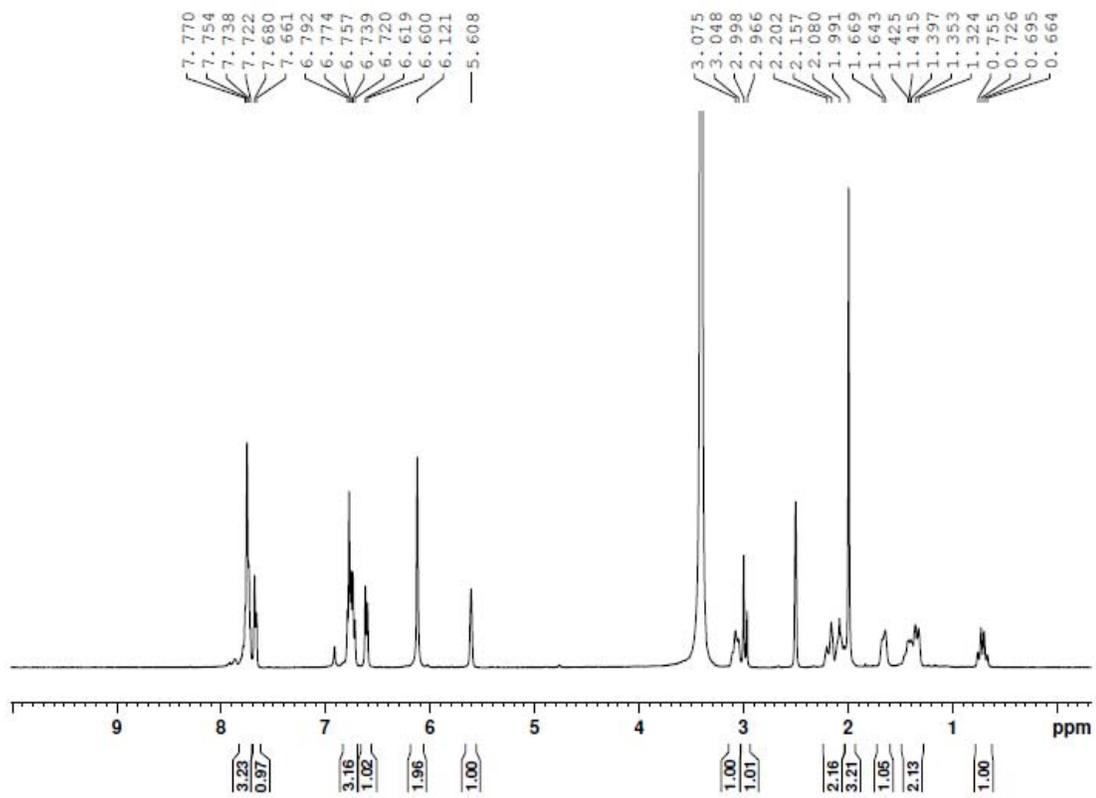
*Fax: +91 44 24911589. E-mail: ptperumal@gmail.com*

**Supplementary material**

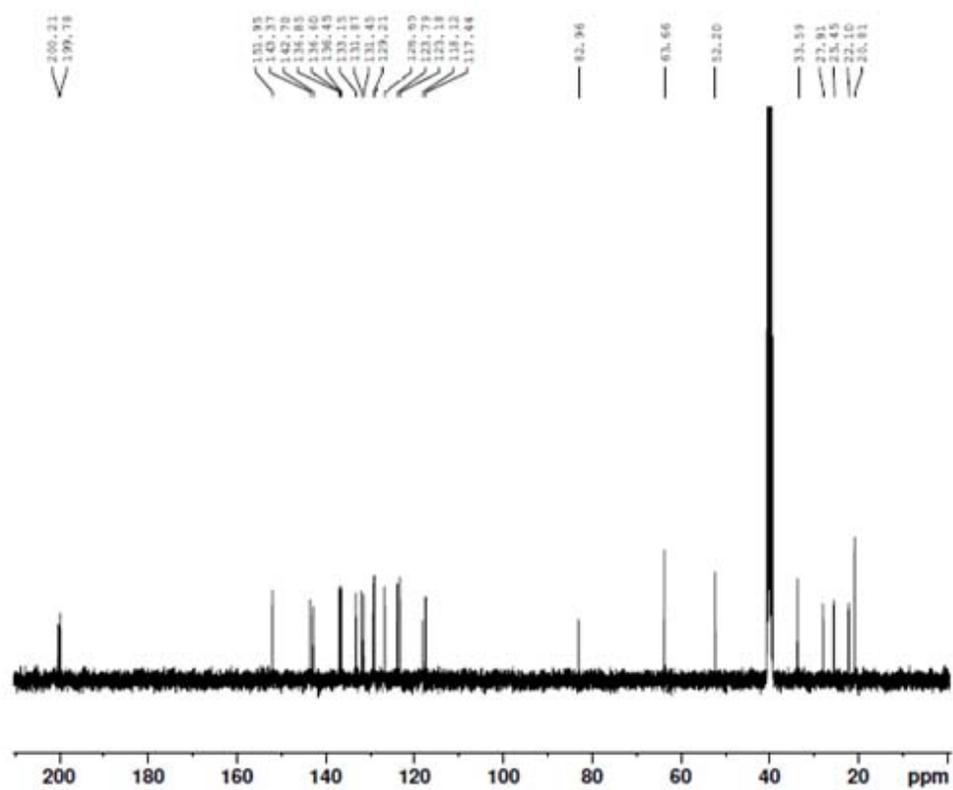
## IR Spectrum of compound 4a



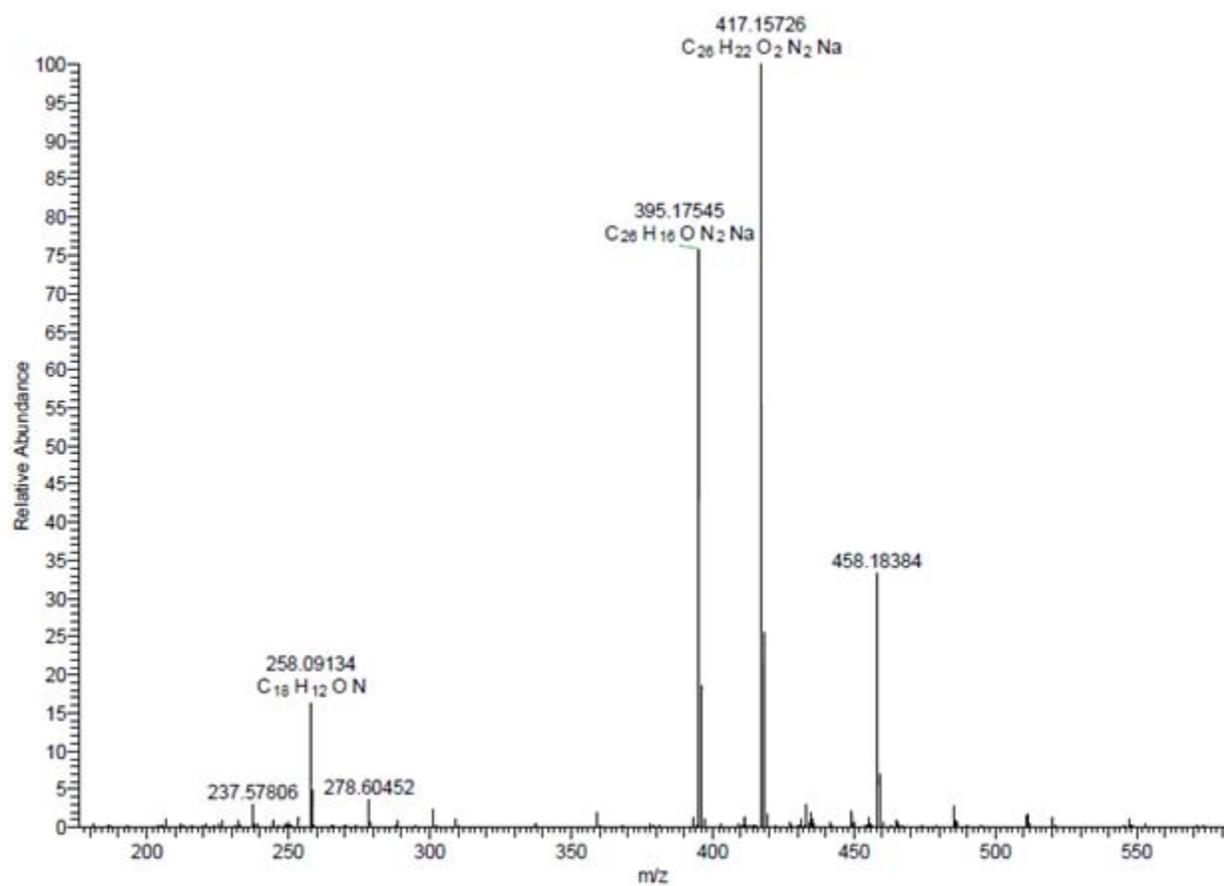
$^1\text{H}$  NMR spectrum of 4a



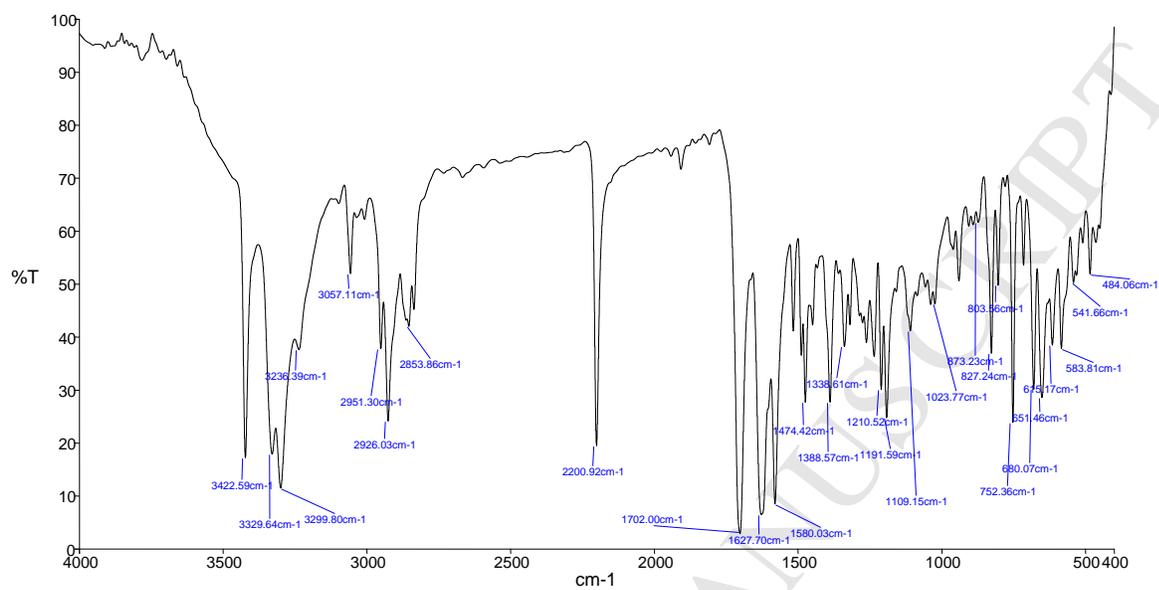
$^{13}\text{C}$  NMR spectrum of 4a



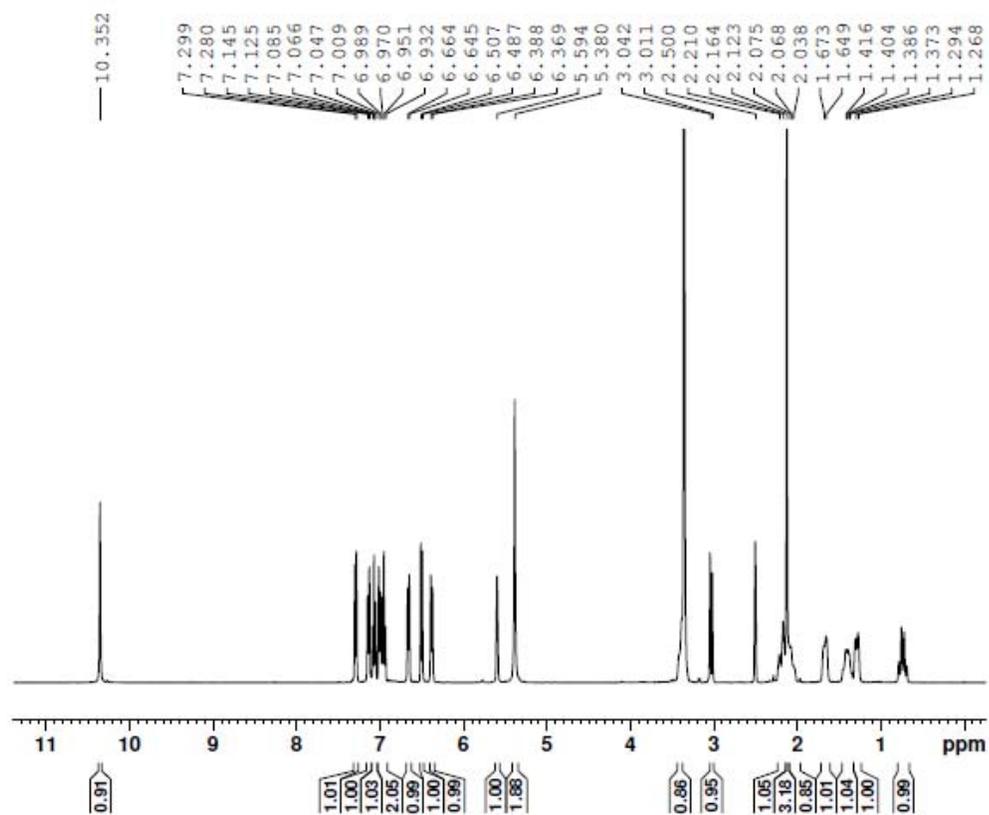
## Mass spectrum of 4a



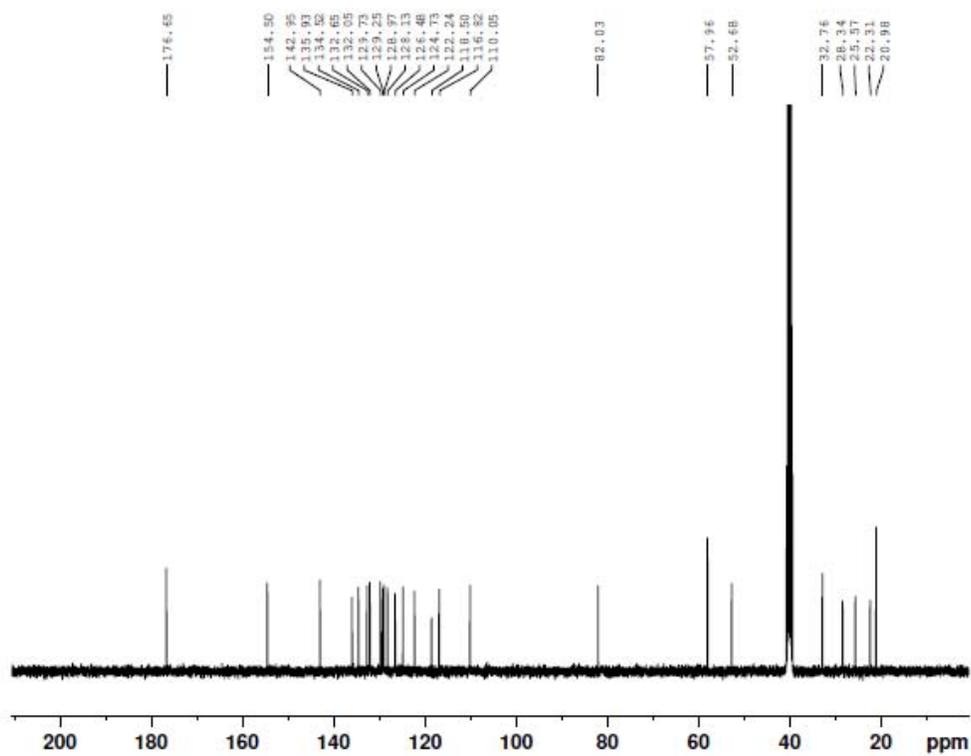
## IR Spectrum of compound 6a



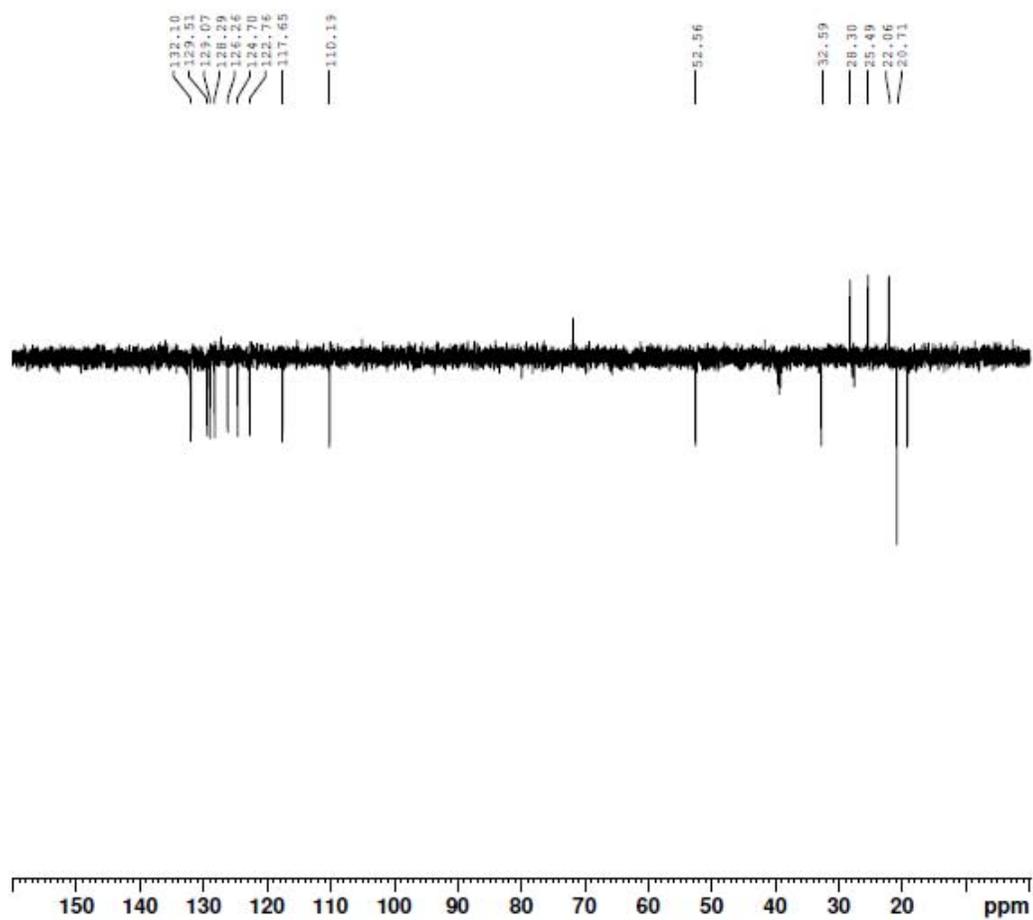
$^1\text{H}$  NMR spectrum of 6a



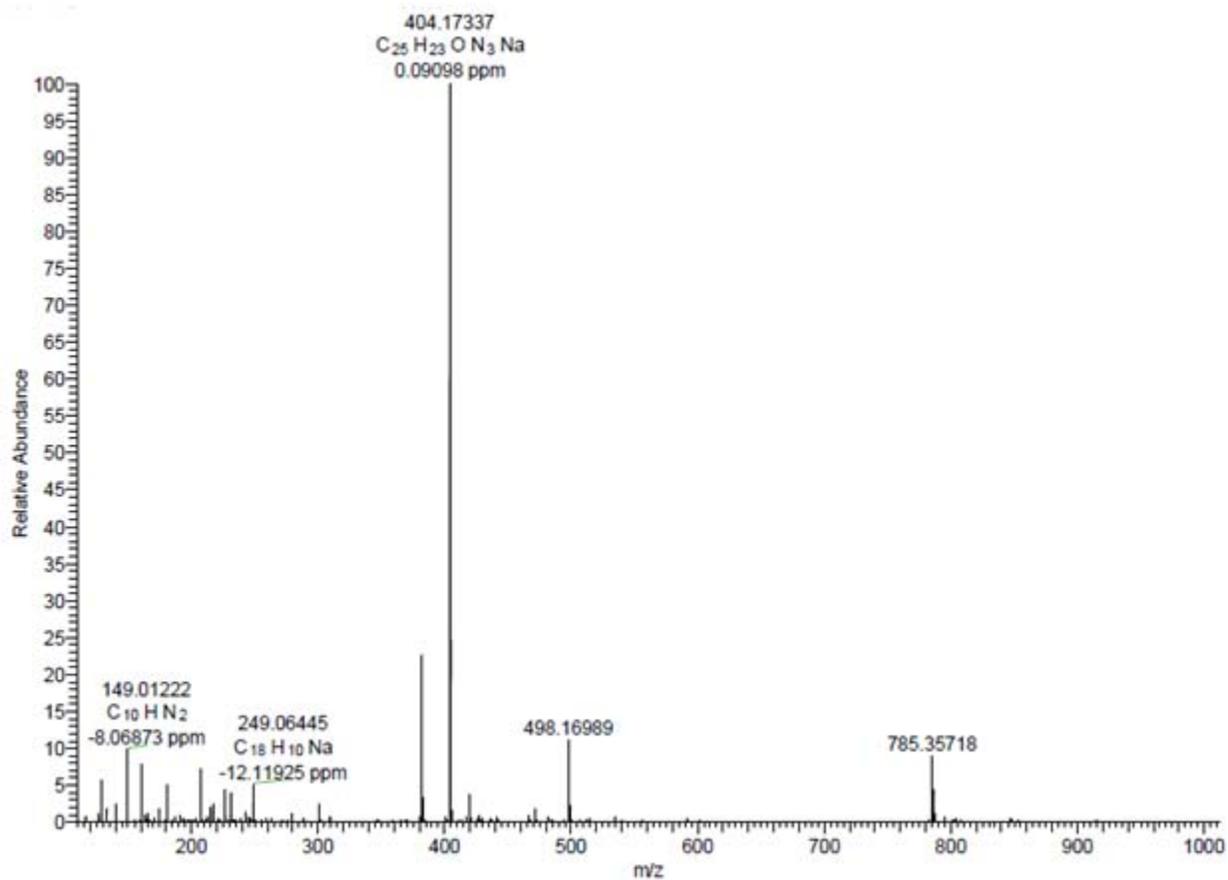
$^{13}\text{C}$  NMR spectrum of 6a

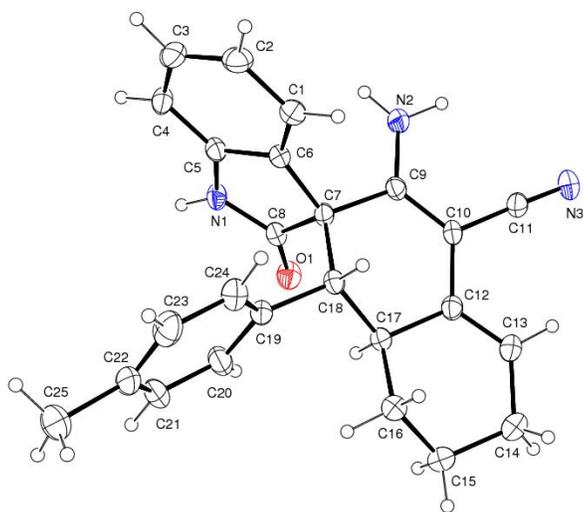


DEPT of compound 6a

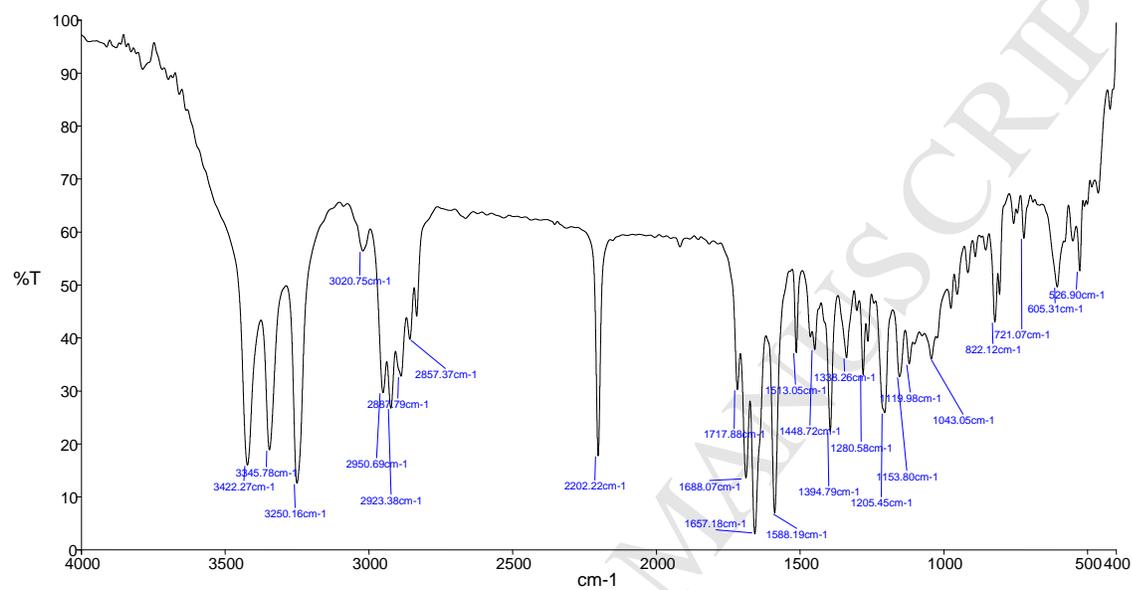


## Mass spectrum of 6a

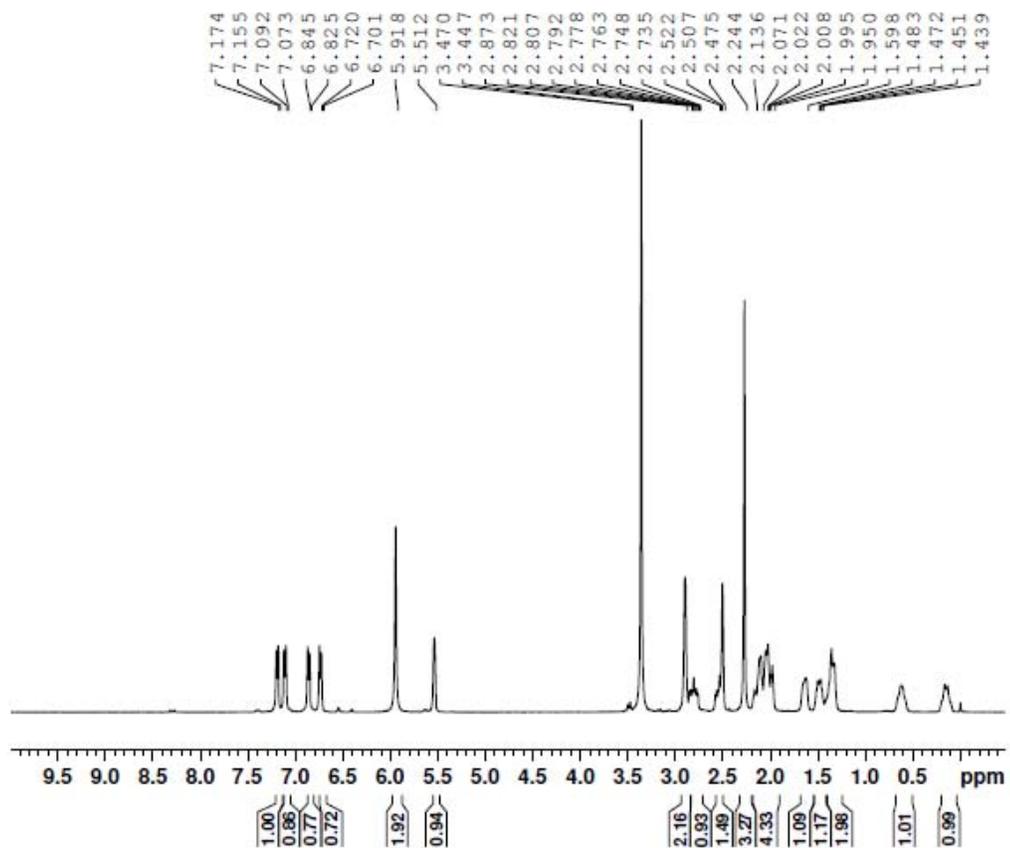


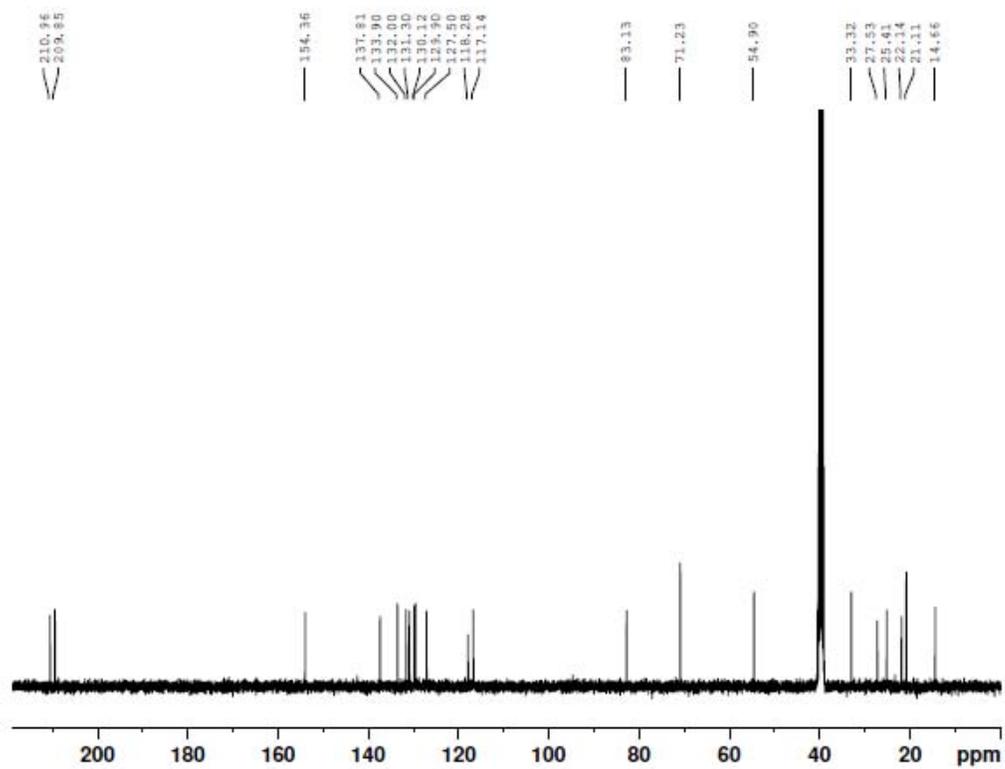
ORTEP diagram of compound **6a**

## IR Spectrum of compound 8a

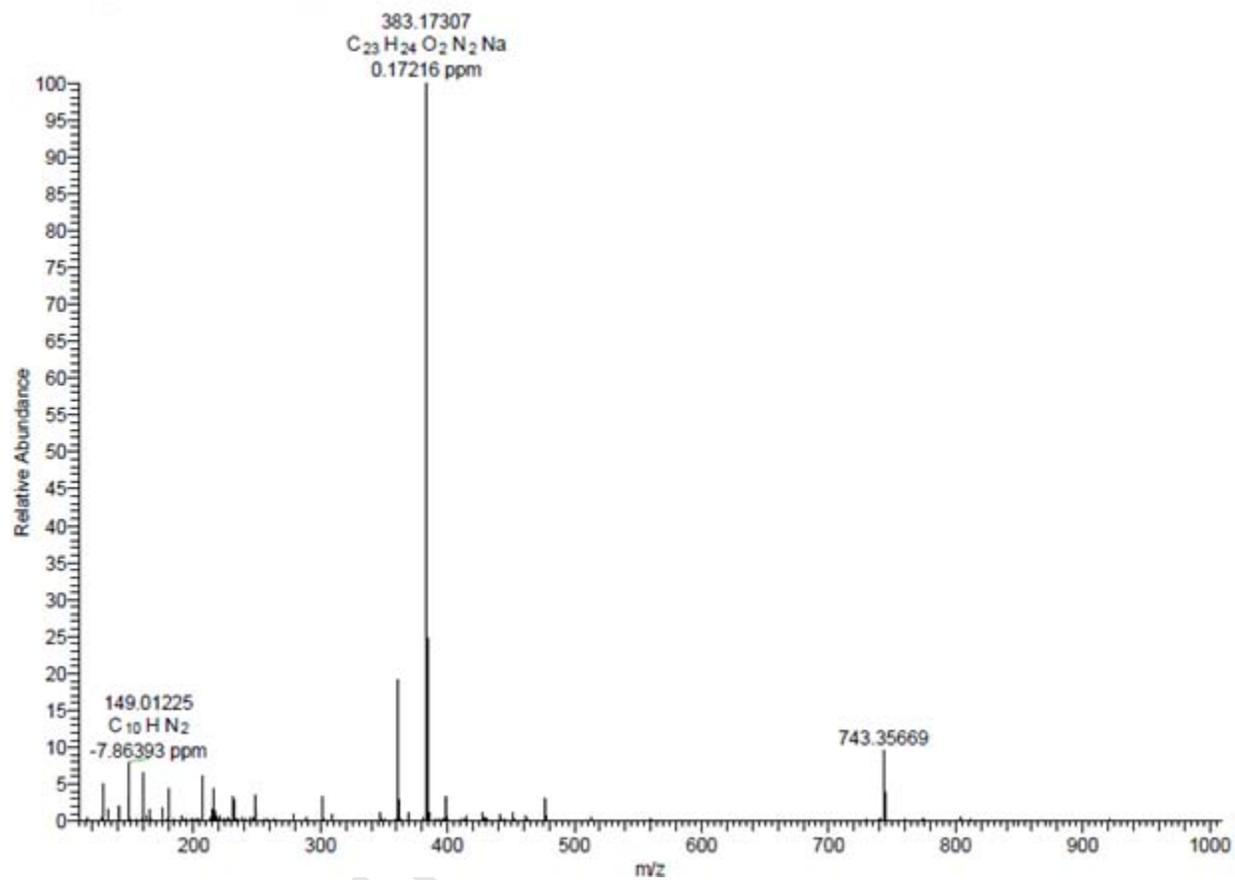


$^1\text{H}$  NMR spectrum of 8a

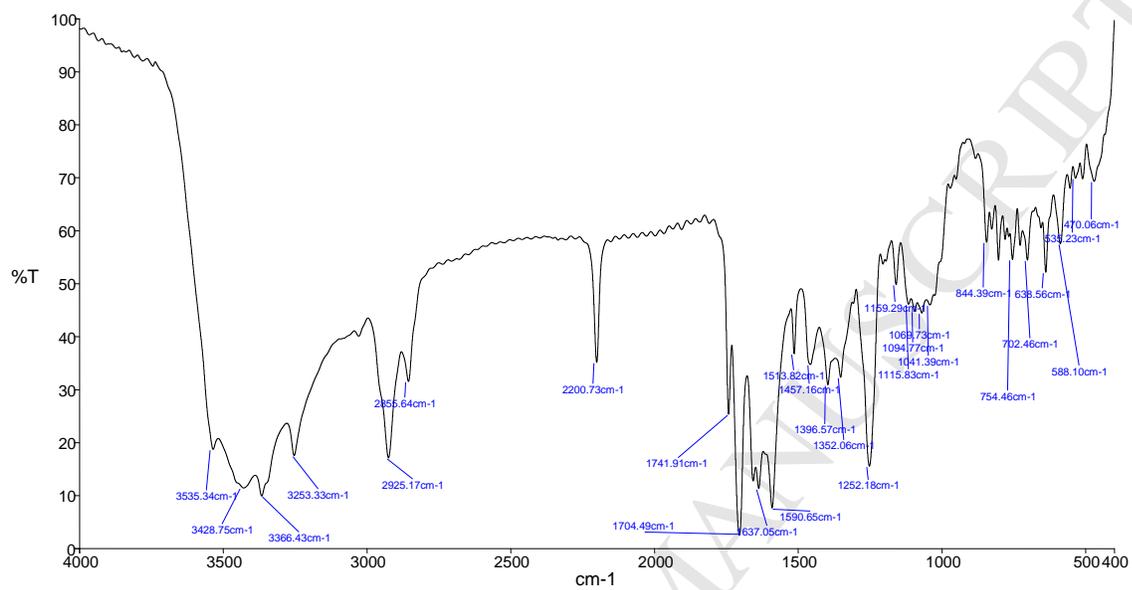


$^{13}\text{C}$  NMR spectrum of 8a

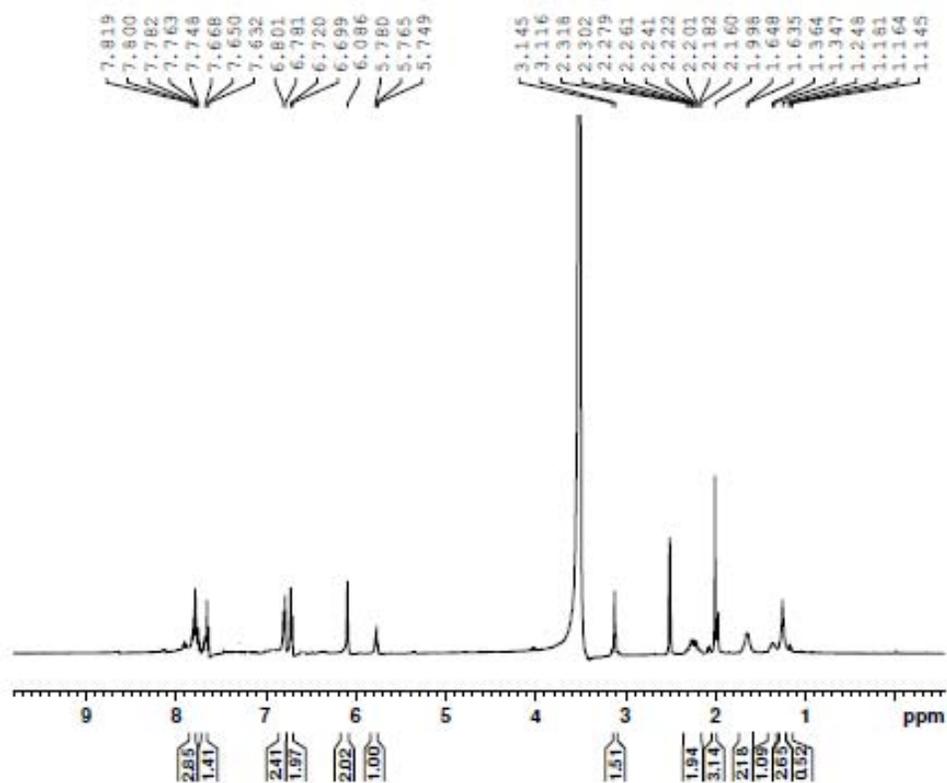
## Mass spectrum of 8a



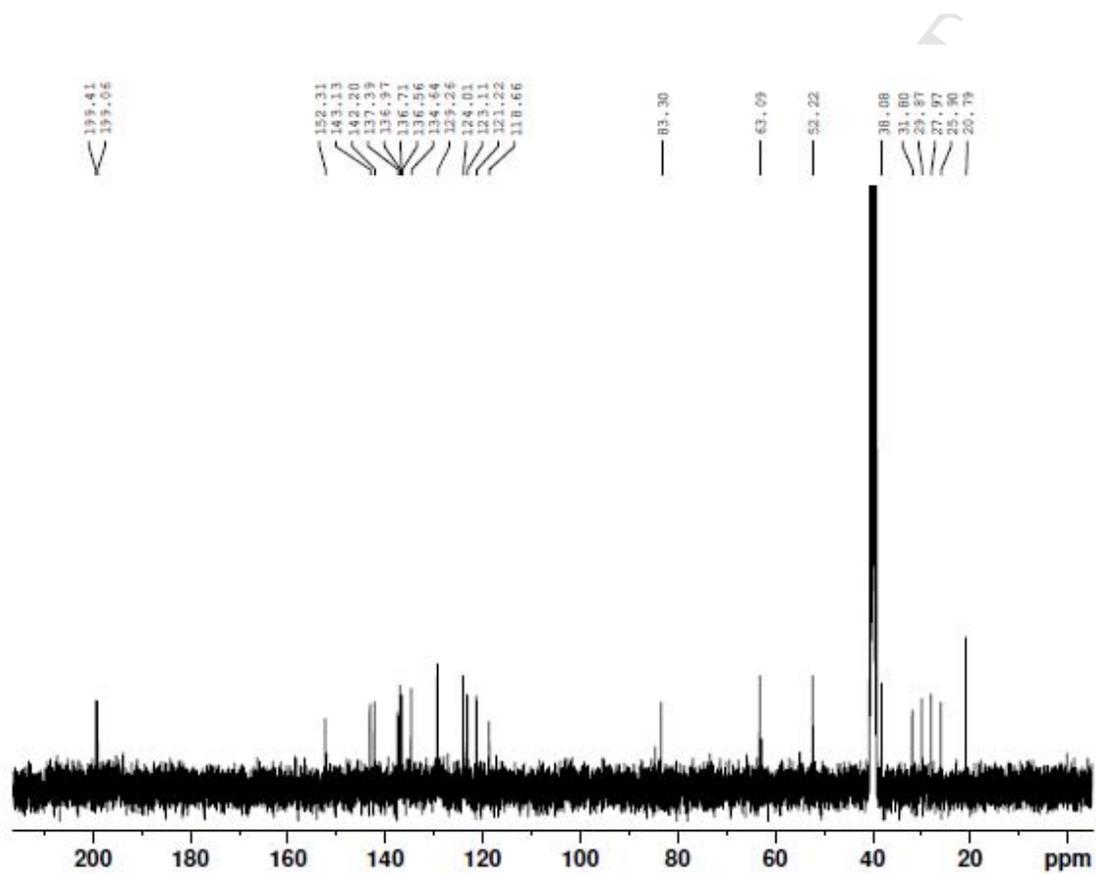
## IR Spectrum of compound 4c



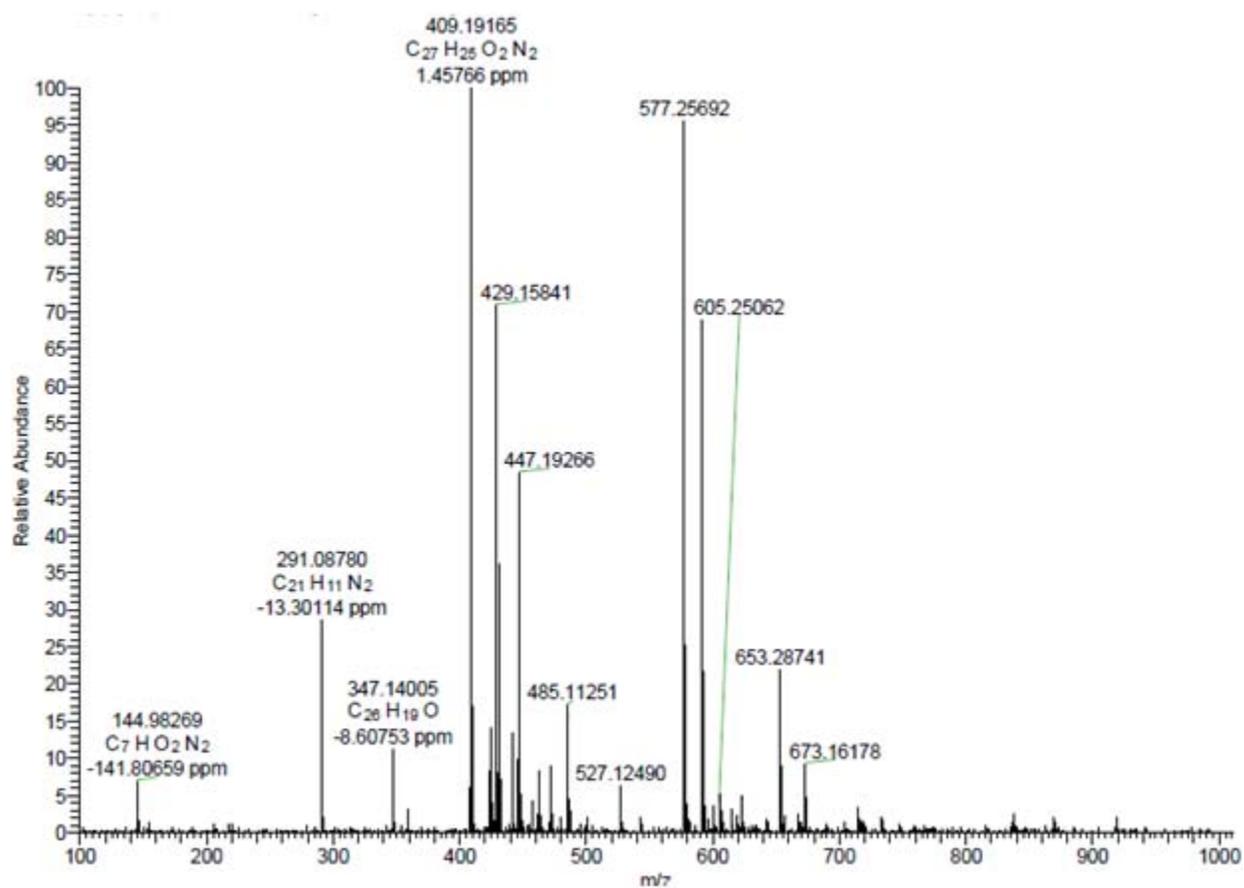
$^1\text{H}$  NMR spectrum of 4c



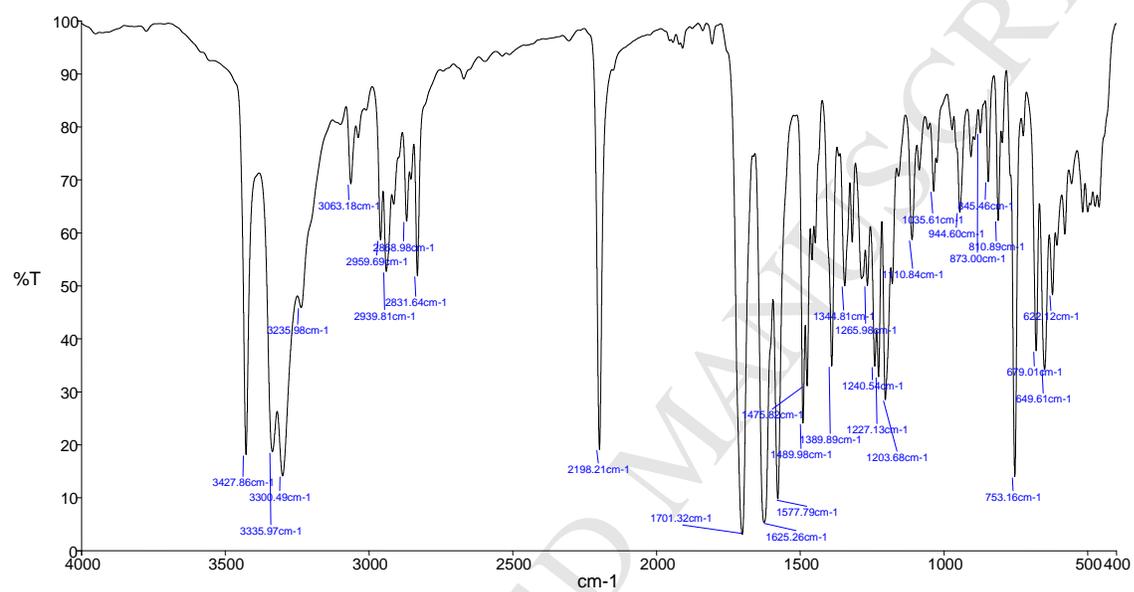
$^{13}\text{C}$  NMR spectrum of 4c



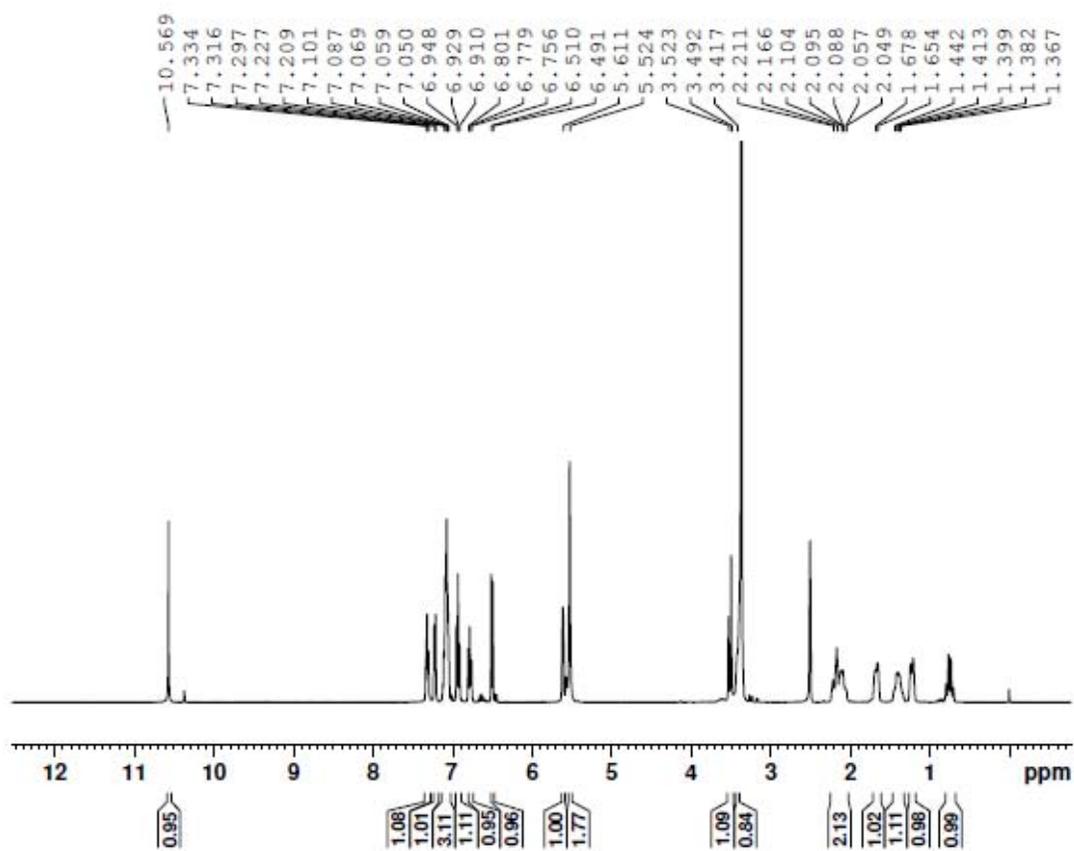
## Mass spectrum of 4c

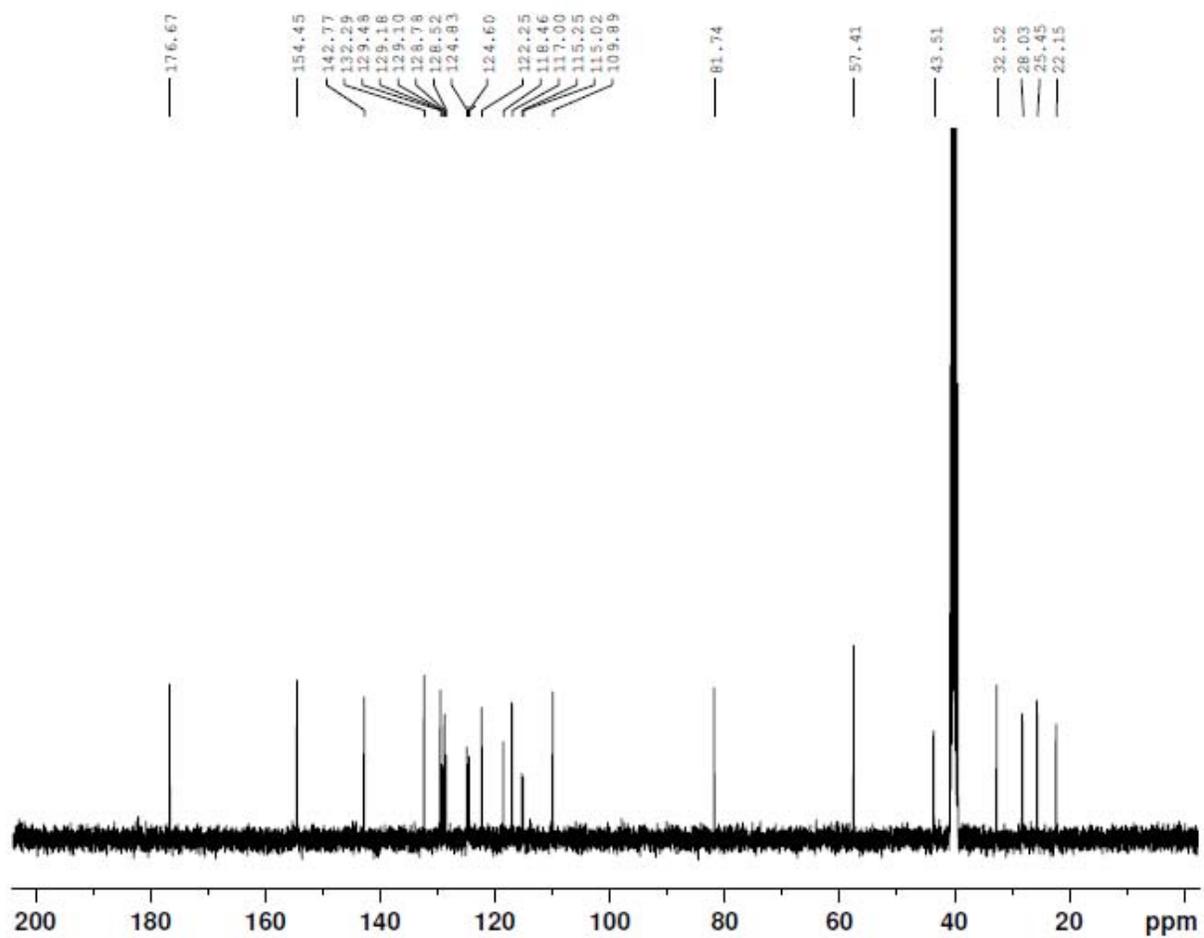


## IR Spectrum of compound 6d

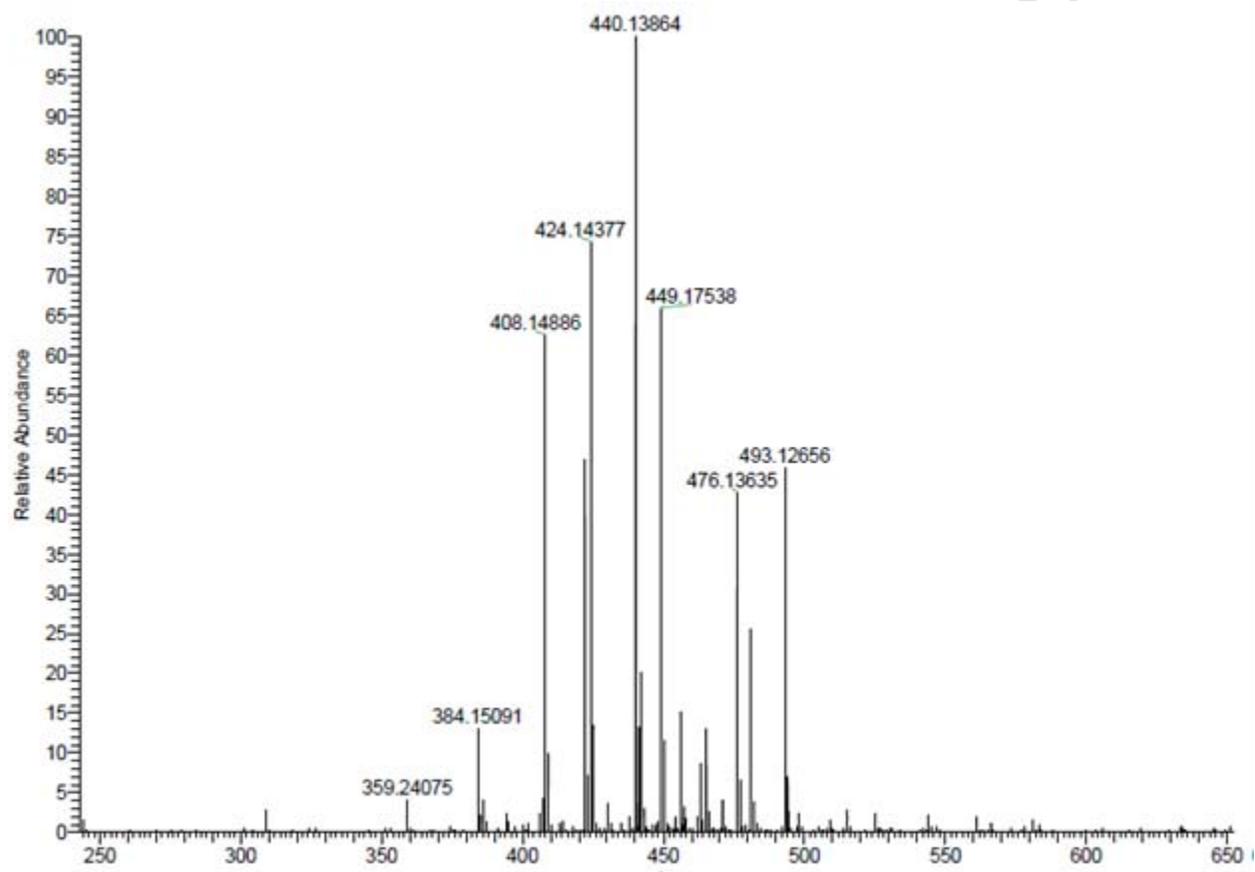


$^1\text{H}$  NMR spectrum of 6d

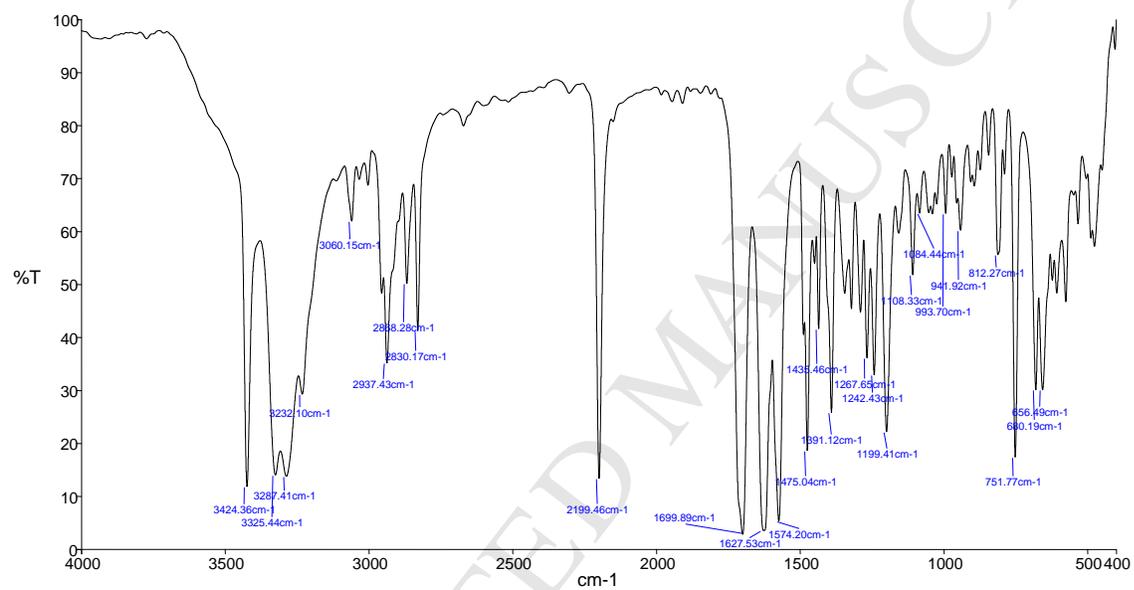


$^{13}\text{C}$  NMR spectrum of 6d

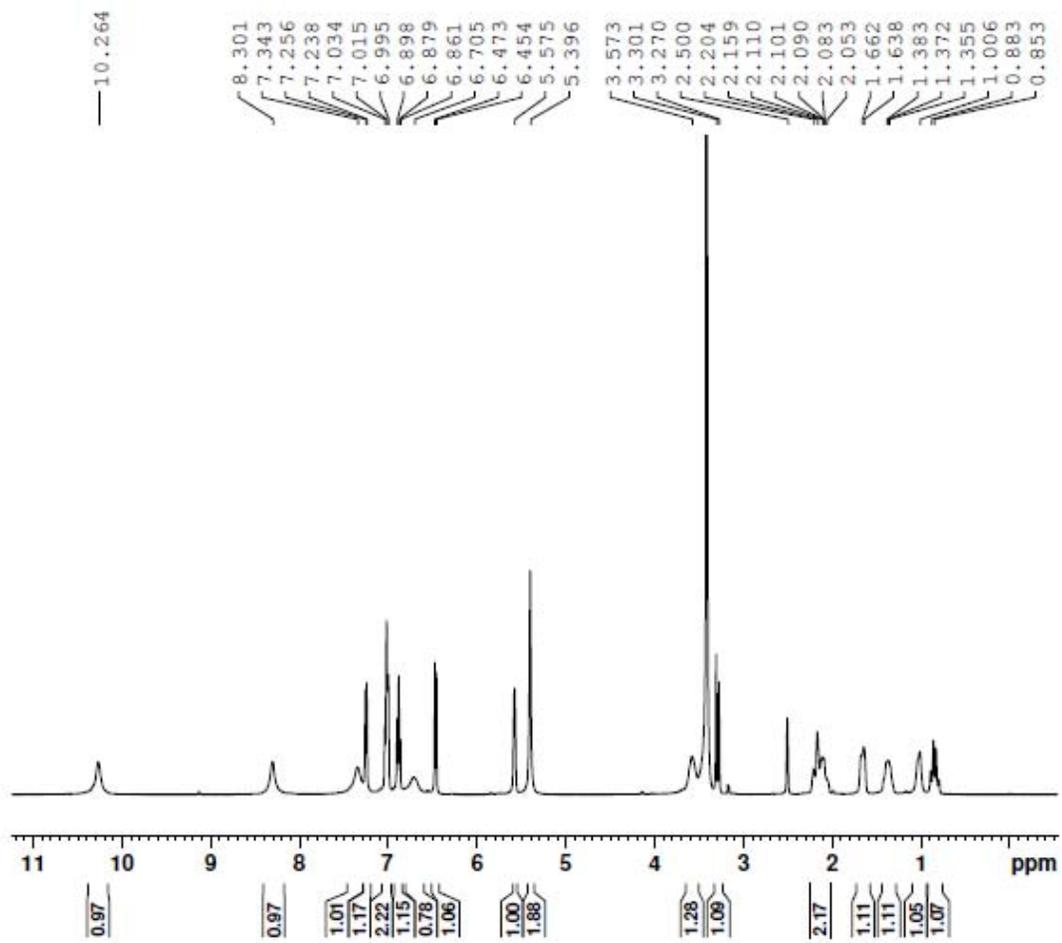
## Mass spectrum of 6d



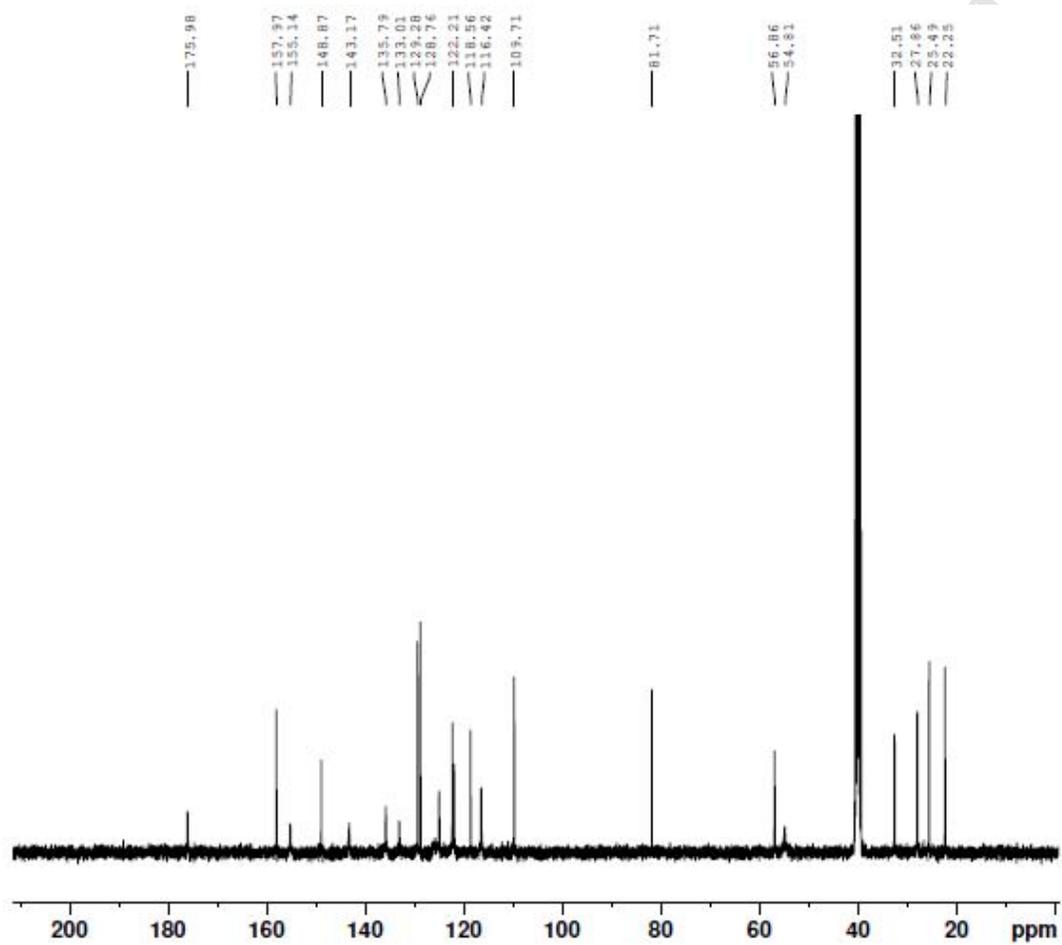
## IR Spectrum of compound 6j



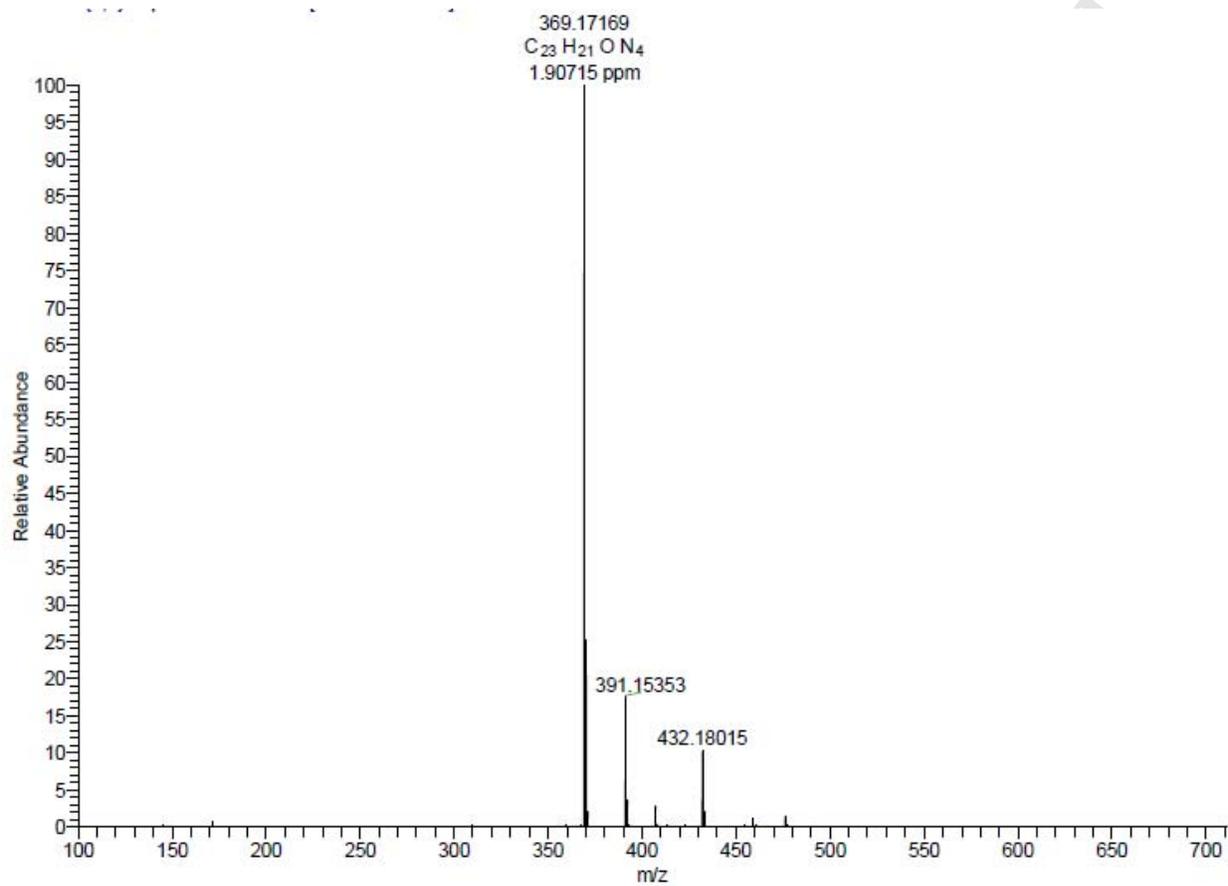
$^1\text{H}$  NMR spectrum of 6j



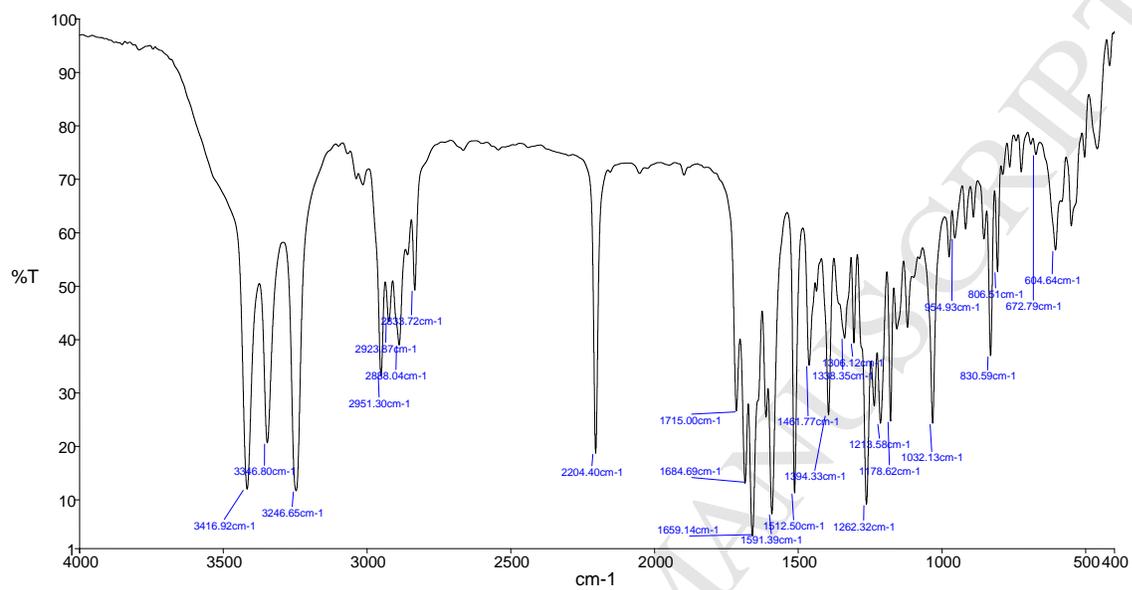
$^{13}\text{C}$  NMR spectrum of 6j

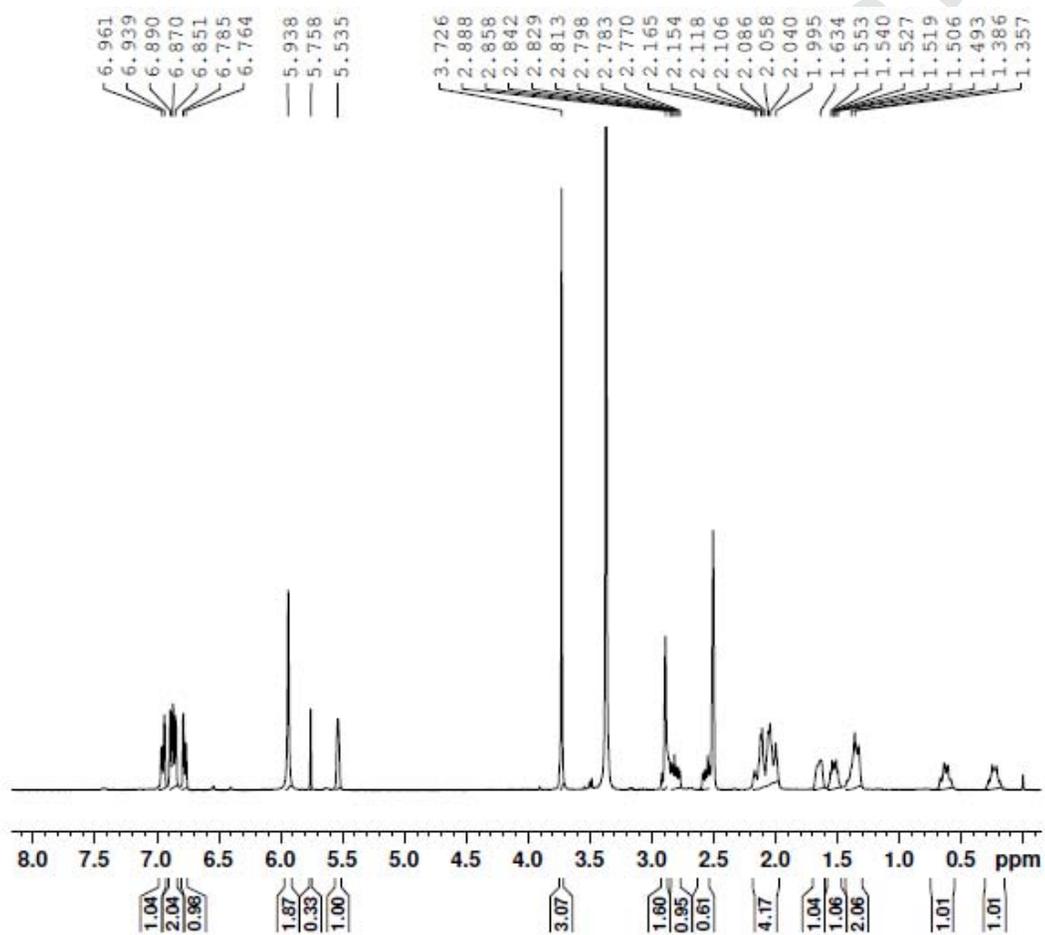


## Mass spectrum of 6j

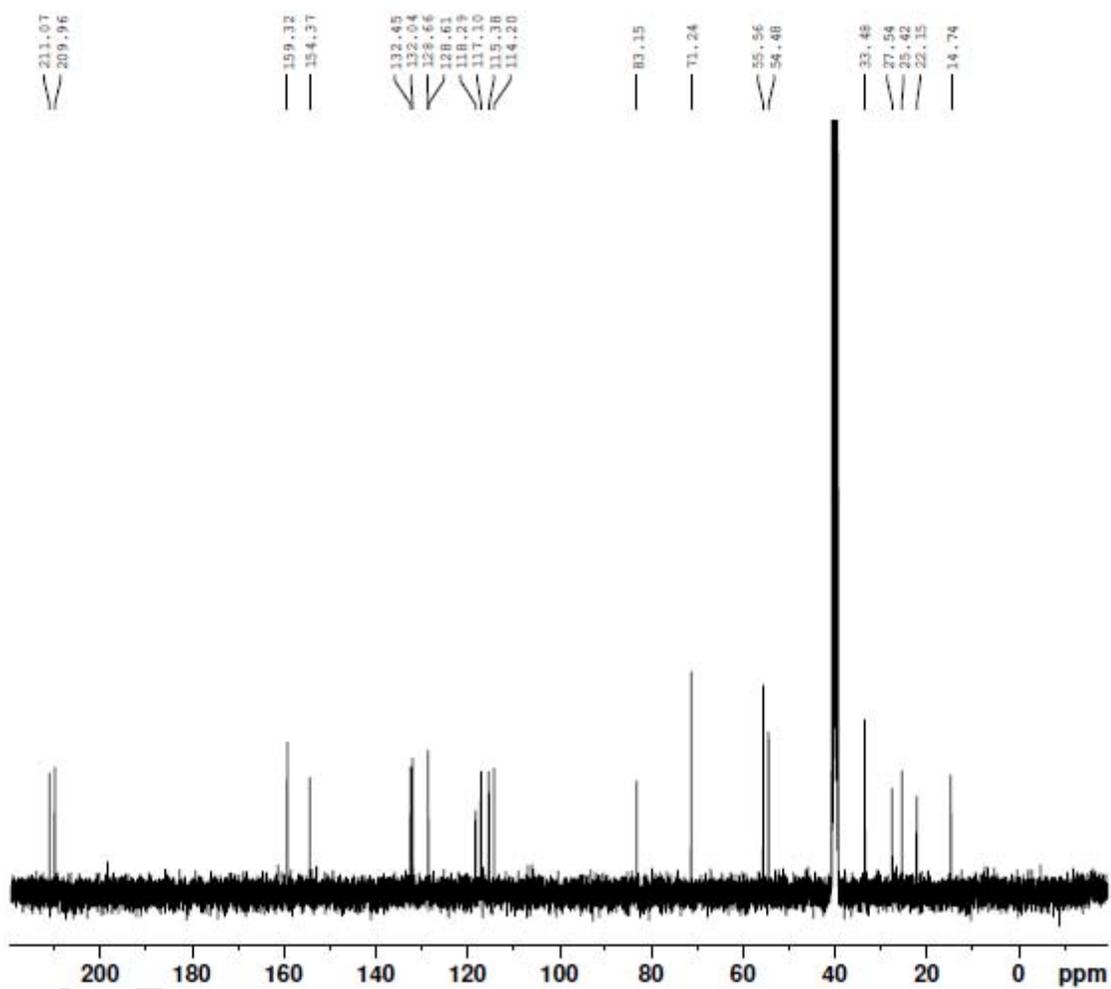


## IR Spectrum of compound 8b

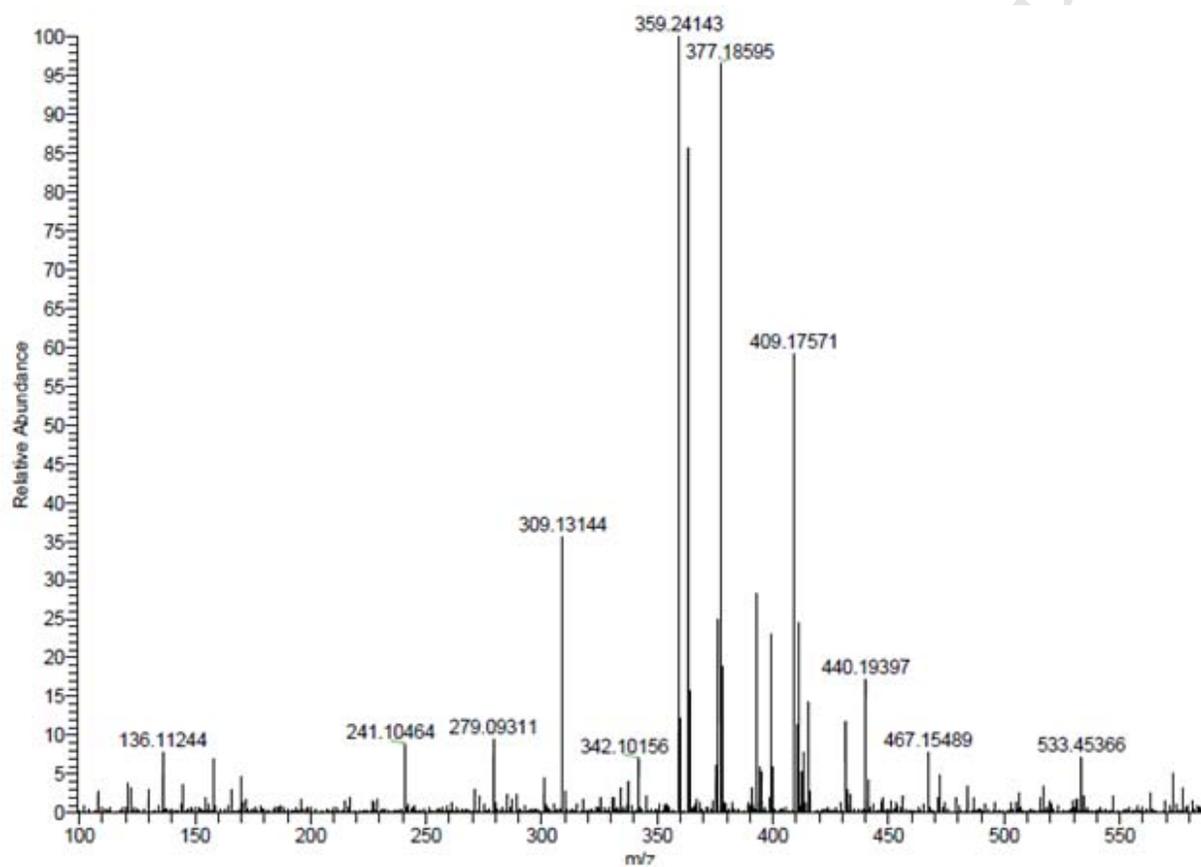


$^1\text{H}$  NMR spectrum of 8b

$^{13}\text{C}$  NMR spectrum of 8b



## Mass spectrum of 8b



## Binding energy and the interaction of ligands with the DNA gyrase receptor

<b>Ligand</b>	<b>Interaction</b>	<b>Free energy of Binding (kcal/mol)</b>
4a	ARG 1069 H-acceptor GLY 1072 Pi-H	-10.34
4b	LYS 1077 (B) H-acceptor ARG 1069 (B) pi-cation	-10.63
4c	ARG 1069 (B) H-acceptor	-9.90
4d	ARG 1122 (B) H-acceptor MET 1075 (B) pi-H	-11.52
4e	ARG 1069 (B) pi-H	-8.33
4f	GLN 1056 (B) H-donor	-9.49
4g	ARG 1069 (B) H-acceptor	-8.45
4h	MET 1121 (B) H-acceptor	-8.33
4i	ARG 1122	-11.64
4j	GLY 1076 (B) pi-H	-10.09
4k	MET 1075 (B) pi-H ARG 1122 (B) pi-cation	-10.36
4l	MET 1075 (B) pi-H	-9.86
4m	GLN 1056 (B) H-donor	-9.51
6a	ARG 1069 (B) H-acceptor	-11.52
6b	ARG 1069 (B) H-acceptor	-11.20
6c	ARG 1122 (B) pi-H	-8.66
6d	GLY 1072 (B) H-donor	-11.01
6e	ASP 1073 (B) H-donor LYS 1077 (B) H-acceptor RG 1069 (B) pi-cation	-8.66
6f	ARG 1069 (B) pi-cation	-9.61
6g	ARG 1122 (B) pi-H ARG 1122 (B) pi-H	-9.17
6h	ALA 1068 (B) H-donor	-8.14
6i	ASP 1073 (B) H-donor LYS 1077 (B) H-acceptor	-10.66
6j	ARG 1069 (B) H-acceptor ARG 1069 (B) H-acceptor	-10.82
8a	ARG 1069 (B) H-acceptor	-9.61
8b	ARG 1069 (B) H-acceptor	-10.52

## Binding energy and the interaction of ligands with the ALK receptor

Ligand	Interaction	Free energy of Binding (kcal/mol)
4a	ASP 1203 H-donor	-15.20
4b	ASP 1203 Pi-H	-14.05
4c	ARG 1253 H-acceptor HIS 1124 H-acceptor	-14.55
4d	GLU 1210 H-donor	-13.26
4e	ARG 1253 Pi-cation	-14.16
4f	ASP 1203 H-donor	-12.85
4g	MET 1328 H-donor HIS 1124 Pi-H	-12.29
4h	GLU 1210 H-donor	-12.58
4i	HIS 1124 H-acceptor	-16.13
4j	HIS 1124 Pi-H	-11.84
4k	HIS 1124 H-acceptor	-14.37
4l	GLU 1210 H-donor	-12.60
4m	MET 1328 H-donor HIS 1124 Pi-H	-12.29
6a	GLU 1210 H-donor	-11.57
6b	GLN 1177 H-acceptor	-11.79
6c	ARG 1253 H-acceptor	-12.69
6d	ASP 1249 H-donor GLY 1272 H-acceptor	-12.45
6e	LYS 1267 H-acceptor GLN 1146 Pi-H	-11.55
6f	GLU 1197 H-donor	-11.27
6g	ARG 1253 Pi-cation	-12.72
6h	ARG 1209 H-acceptor	-11.23
6i	ASP 1249 H-donor ASN 1254 H-donor GLY 1272 H-acceptor	-18.43
6j	HIS 1124 Pi-H LYS 1205 Pi-H	-13.13
8a	LEU 1198 H-acceptor	-11.98
8b	HIS 1124 H-acceptor HIS 1124 Pi-H	-13.97