Total synthesis of naamine C and pyronaamidine, antitumor marine imidazole alkaloids

PERKIN

Seikou Nakamura, Ikuo Kawasaki, Miki Kunimura, Miyuki Matsui, Yoko Noma, Masayuki Yamashita and Shunsaku Ohta*

Kyoto Pharmaceutical University, Misasagi Yamashinaku, Kyoto 607-8414, Japan. E-mail: sohta@mb.kyoto-phu.ac.jp

Received (in Cambridge, UK) 23rd January 2002, Accepted 6th March 2002 First published as an Advance Article on the web 19th March 2002

The first total synthesis of naamine C and pyronaamidine, highly substituted and cytotoxic imidazole marine alkaloids of a certain kind of sponge, was achieved through an eight-step reaction starting from 1-methyl-2-phenylthio-1*H*-imidazole.

Introduction

Many imidazole alkaloids containing highly substituted imidazole ring(s) have been isolated from a bright yellow sponge, *Leucetta chagosensis*, and several of their structures are shown in Fig 1.¹ These alkaloids generally have interesting biological

MeO
$$X^3$$
 Me X^2 MeO X^3 Me X^2 MeO X^3 Naamine B X^3 Naamine B X^3 Naamine B X^3 Naamine B X^3

$$\begin{array}{c|c} & & & & & & \\ & & & & & \\ MeO & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\ & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & &$$

Pyronaamidine (4): $X^1 = OMe$, $X^2 = OMe$, $X^3 = OH$ Naamidine A (5): $X^1 = OH$, $X^2 = X^3 = H$ Naamidine B (6): $X^1 = OMe$, $X^2 = OH$, $X^3 = H$ Naamidine G (7): $X^1 = OMe$, $X^2 = X^3 = H$

Clathridine (8)

Kealiiquinone (9): X = OH 2-Deoxy-2-amino-Kealiiquinone (10): X = NH₂

Fig. 1

properties such as antitumor and antifungal activities. For example, it was reported that naamidine A, B and G 5-7^{1a-e} showed antifungal activity against Cryptococcus neoformans, and, in particular, pyronaamidine 41e was cytotoxic against KB cells, minimum inhibitory concentration (MIC) = 5 μ g mL⁻¹. A structural characteristic of these alkaloids is that one or two alkoxybenzyl group(s) are located at the 4 and/or 5-position of the 1-methyl-1*H*-imidazole ring. So far as the alkaloids **4–8** are concerned, the 2-position of the ring is substituted with the (1-methyl-2,5-dioxo-3*H*-imidazolin-4-yl)amino moiety. Pyronaamidine 4 has been considered to be a possible biometabolic intermediate in the biochemical production of the tricyclic alkaloids, kealiiquinone 91e and 2-deoxy-2-aminokealiiquinone 10.1 We have investigated the total synthesis of these imidazole natural products and already reported the first total synthesis of several marine imidazole alkaloids, 1,2 3,3 5,2 84 and 9.5 The most important key step in the total synthesis of 4 may be the construction of the (1-methyl-2,5-dioxo-3H-imidazolin-4-yl)amino side chain. In this paper, we report the first total synthesis of 4 through naamine C 2.

Results and discussion

First, the preparation of the 1,2,4,5-tetrasubstituted imidazole derivative 17 was attempted (Scheme 1). 1-Methyl-2-phenylthio-1H-imidazole 11 was converted to the 5-substituted imidazole 14 according to our previously reported method.⁵ When the 4position of 14 was lithiated by treatment with tert-butyllithium, followed by quenching with p-anisaldehyde, a diastereomeric mixture of the alcohol 15 was obtained in 43% yield. The TBDMS group was removed by treatment with TBAF. The alcohol 16 was reduced with zinc powder in conc. HCl-acetic acid at 80 °C to give not the desired 1,2,4,5-tetrasubstituted imidazole 17 but a tricyclic naphthoimidazole 18 as a major product (yield 75%) along with many minor uncharacterized products. The structure of 18 was supported by a 1H-NMR spectroscopic study based on the data obtained in the previous investigation of the total synthesis of kealiiquinone 9.5 It could be considered that an intramolecular Friedel-Crafts type cyclization of 16 occurred under such acidic conditions to give 18. Reduction of 16 with nickel boride,6 which was used in the previous report,² resulted unfortunately in formation of a complex mixture of many compounds such as the corresponding alcohols and deprotected phenols.

To overcome these problems, the Et₃SiH reduction method ⁷ was applied to the present system (Scheme 2). The benzyl alcohol **20** having a TBDMS group instead of the MOM group of

J. Chem. Soc., Perkin Trans. 1, 2002, 1061-1066

Scheme 1 Reagents: (a) lithium 2,2,6,6,-tetramethylpiperidinide (LTMP), THF, 89%; (b) TBDMSCl, imidazole, DMF, quant; (c) NBS, THF, 64%; (d) (i) tert-BuLi THF; (ii) p-anisaldehyde, 43%; (e) TBAF, THF, quant: (f) Zn, conc. HCl, AcOH, 18: 75%; (g) NaBH₄, NiCl₂·6H₂O, THF, MeOH, produced complex mixture.

12 was prepared starting from 11 and the aldehyde 19b8 (R = TBDMS) similarly as above. The compound 20 was smoothly reduced to the 5-benzylimidazole 21 in 98% yield by treatment with triethylsilane in the presence of trifluoroacetic acid according to Kobayashi's procedure.7 The 5-benzylimidazole 21 was brominated by NBS to give the 4-bromoimidazole 22 in 81% yield. When the bromide 22 was subjected to lithiation with tert-butyllithium at -78 °C for 15 min followed by quenching with p-anisaldehyde, the required 4-alkylated product 23 was obtained in only 15% yield along with a mixture of many minor uncharacterized products. This result might be attributable to intermediate formation of an equilibrium mixture containing the kinetic product 27 and the thermodynamic product 28 before addition of p-anisaldehyde (Scheme 3). Thus, tertbutyllithium was added into a mixed solution of 23 and p-anisaldehyde in THF at -78 °C in order to avoid the equilibrium, and the yield of 23 increased as expected, and reached

Reductive removal of the hydroxy group of 23 with Et₃SiH in the presence of TFA unfortunately resulted in formation of the tricyclic imidazole derivative 26 in 79% yield instead of the desired reductant 24 because intramolecular Friedel–Crafts type cyclization of 23 occurred under such acidic conditions. On the other hand, reduction of 23 with nickel boride⁶ gave successfully the 2-unsubstituted 4,5-dibenzylimidazole 25 in 62% yield. The structure of 25 was supported by ¹H-NMR spectra and other analytical data (Scheme 2).

The imidazole **25** was brominated by NBS to give the 2-bromoimidazole **29**, which was subjected to lithiation with *tert*-butyllithium followed by treatment with trisyl azide ⁹† to afford the 2-azido **30** in 46% overall yield from **25**. The TBDMS

Scheme 2 Reagents: (a) LTMP, THF, 78%; (b) Et₃SiH, TFA, DCM, 98%; (c) NBS, THF, 81%; (d) (i) tert-BuLi (2 equiv.), THF; (ii) p-anisaldehyde (5 equiv.), 15%; (e) tert-BuLi (6 equiv.), p-anisaldehyde (5 equiv.), THF, 89%; (f) Et₃SiH, TFA, DCM, **26**: 79%; (g) NaBH₄, NiCl₂·6H₂O, THF, MeOH, **25** 62%.

(24: R1 = SPh)

 $\textbf{Scheme 3} \quad \text{A possible equilibrium after the lithiation of } \textbf{22}.$

group of **30** was removed by treatment with TBAF, and the subsequent hydrogenation over 10% Pd–C gave naamine C **2** in 84% yield from **30** in 2 steps. Naamine C **2** was isolated as a yellow powder, the physical and spectral data of which almost agreed with those of the natural product reported ^{1/5,10} (Scheme 4).

The final step was construction of the 2-(1-methyl-2,5-dioxo-3*H*-imidazolin-4-yl)amino moiety. We have already reported a method for the regio-selective condensation of arylamine with 1-methylparabanic acid ‡ **32** for constructing the side chain, and its application to the total synthesis of clathridine **8**.⁴ This time, naamine C **2** was treated with **32** in the presence of TMSCl and *N*,*N*-diisopropylethylamine according to the previous procedure ¹¹ to give successfully pyronaamidine **4** in 28% yield as yellow needles, the melting point and spectral data of which were all consistent with those of the natural product reported by Scheuer ¹*e*, ¹² (Scheme 4).

While pyrronaamidine 4 was isolated at a relatively early stage among many imidazole alkaloids of sponges, its total

MeO 11 СНО Мe TBDMSO ÓН OTBDMS 20 b OMe MeO d or e MeO MeO Ме TBDMSO MeO 21: R = H Мe 22: R = Br **TBDMS**O 23 ОМе g OMe MeO SPh MeO Мe **TBDMS**O MeO 26 MeO Мe TBDMSO **25**: $R^1 = H$

[†] The IUPAC name for trisyl azide is azidotriphenylsilane.

[‡] The IUPAC name for parabanic acid is imidazolidinetrione.

Scheme 4 Reagents: (a) NBS, THF, 59%; (b) (i) tert-BuLi, THF, (ii) trisyl azide, 78%; (c) TBAF, THF, 87%; (d) H₂, 10% Pd–C, EtOH, 97%; (e) N,N-diisopropylethylamine, TMSCl, CHCl₃, 28%.

synthesis had not been reported, and we were fortunately able to achieve the first total synthesis of 4 through 2.

Experimental

All melting points were measured with a Yanaco MP micromelting points apparatus without correction. IR was taken with a Shimadzu IR-435 spectrometer. 1 H-NMR spectra were measured on a Varian INOBA 400NB (1 H: 400 MHz, 13 C: 100.6 MHz) with tetramethylsilane as an internal standard and chemical shifts δ are reported in ppm. Abbreviations of 1 H-NMR signal patterns are as follows: s (singlet); d (doublet); t (triplet); m (multiplet). Mass spectra (MS) and high-resonance MS (HRMS) were obtained on a JEOL JMS BU-20 spectrometer under EI ionizing conditions. Silica gel (Merck Art. 7734) was used for column chromatography.

5-[1-(*tert*-Butyldimethylsiloxy)-1-(3,4-dimethoxy-2-methoxy-methoxyphenyl)methyl]-4-[1-hydroxy-1-(4-methoxyphenyl)methyl]-1-methyl-2-phenylthio-1*H*-imidazole (15)

A solution of tert-BuLi in n-pentane (1.51 M; 0.44 mL, 0.66 mmol) was added dropwise to a solution of 14⁵ (200 mg, 0.33 mmol) in THF (2 mL) under an N_2 atmosphere at -78 °C. Stirring was continued for 1 h, then a solution of p-anisaldehyde (0.20 mL, 1.64 mmol) in THF (0.5 mL) was added dropwise at -78 °C. Stirring was continued for 3 h at -78 °C, then water (2 mL) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt-n-hexane 1:5) on silica gel to give 15 (95 mg, 43%), a diastereomeric mixture (ca. 1:1) as a pale yellow viscous material; v_{max} (CHCl₃) 2917, 1473, 1242, 1083, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) -0.28 (s, 3H, SiCH₃), -0.11 (s, 3H, SiCH₃), -0.07 $(s, 3H, SiCH_3), 0.04 (s, 3H, SiCH_3), 0.84 [s, 9H, SiC(CH_3)_3], 0.85$ [s, 9H, SiC(CH_3)₃], 3.29 (s, 3H, NC H_3), 3.40 (s, 3H, NC H_3), 3.56 (s, 3H, OCH₃), 3.62 (s, 3H, OCH₃), 3.75 (s, 3H, OCH₃), 3.76 $(s, 3H, OCH_3), 3.78 (s, 6H, 2 \times OCH_3), 3.85 (s, 3H, OCH_3), 3.86$ (s, 3H, OC H_3), 4.61 (d, 1H, J = 4.9 Hz, OC H_a H_bO), 4.92 (d, 1H, $J = 5.0 \text{ Hz}, \text{ OC}H_aH_bO), 4.97 \text{ (d, 1H, } J = 5.1 \text{ Hz, OC}H_aH_bO),$ 5.09 (d, 1H, J = 4.9 Hz, OCH_a H_b O), 5.78 (d, 1H, J = 7.5 Hz, ArCH(OH)Ar), 6.00 (d, 1H, J = 7.5 Hz, ArCH(OH)Ar), 6.28 (s, 1H, ArCH(OTBDMS)Ar), 6.37 (s, 1H, ArCH(OTBDMS)-Ar), 6.67 (d, 2H, J = 8.8 Hz, Ar–H), 6.796 (d, 2H, J = 8.8 Hz, Ar-H), 6.802 (d, 2H, J = 8.8 Hz, Ar-H), 7.03-7.38 (m, 16H, Ar-H) [HRMS m/z Calc. for $C_{35}H_{46}N_2O_7SSi$: M, 666.2795. Found: M⁺, 666.2799].

5-[1-Hydroxy-1-(3,4-dimethoxy-2-methoxymethoxyphenyl)-methyl]-4-[1-hydroxy-1-(4-methoxyphenyl)methyl]-1-methyl-2-phenylthio-1H-imidazole (16)

A solution of TBAF in THF (1 M; 0.14 mL, 0.14 mmol) was

added dropwise to a solution of 15 (78 mg, 0.12 mmol) in THF at room temperature. The mixture was stirred for 10 min at room temperature. Water (0.5 mL) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt) on silica gel to give 16 (57 mg, 88%), a diastereomeric mixture (ca. 1:1), as a pale yellow viscous material; v_{max} (CHCl₃) 3384, 2920, 1596, 1450, 1240, 1090 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 3.37 (s, 3H, NCH₃), 3.42 (s, 3H, NCH₃), 3.50 (s, 3H, OCH_3), 3.52 (s, 3H, OCH_3), 3.75 (s, 3H, OCH_3), 3.78 (s, 3H, OCH_3), 3.82 (s, 6H, 2 × OCH_3), 3.83 (s, 3H, OCH_3), 3.84 (s, 3H, OC H_3), 5.137 (d, 1H, J = 5.7 Hz, OC H_a H_bO), 5.141 (d, 1H, J = 5.9 Hz, OC H_aH_bO), 5.16 (d, 1H, J = 5.9 Hz, OCH_aH_bO), 5.18 (d, 1H, J = 5.0 Hz, OCH_aH_bO), 5.94 (s, 1H, ArCH(OH)Ar), 6.00 (s, 1H, ArCH(OH)Ar), 6.19 (s, 2H, 2 × ArCH(OH)Ar), 6.36 (d, 1H, J = 8.4 Hz, Ar-H), 6.47 (d, 1H, J = 8.8 Hz, Ar-H, 6.55 (d, 1H, J = 8.8 Hz, Ar-H, 6.58 (d, 1H, J = 8.8 Hz, Ar-H), 6.58 (d, 1H, J = 8.8 Hz, Ar-H, 6.58 (d, 1H, J = 8.8 Hz, Ar-H), 6.58 (d, 1H, J = 8.8 Hz, Ar-H),J = 8.6 Hz, Ar-H, 6.76 (d, 2H, J = 8.8 Hz, Ar-H, 6.82 (d, 2H, J = 8.8 Hz, Ar-H)2H, J = 8.8 Hz, Ar-H), 7.09-7.41 (m, 14H, Ar-H) [HRMS] m/z Calc. for C₂₉H₃₂N₂O₇S: M, 552.1930. Found: M⁺ 552.1933].

6,7-Dimethoxy-8-hydroxy-4-(4-methoxyphenyl)-1-methyl-1*H*-naphtho[2,3-*d*]imidazole (18)

Zn powder (114 mg) was added to a mixture of acetic acid (0.5 mL), conc. HCl (0.05 mL) and 16 (42 mg, 0.08 mmol), and then the whole was stirred at 80 °C for 1 h. The reaction mixture was filtered through a cotton plug, and the filtrate was evaporated under reduced pressure. After addition of water (0.5 mL), K₂CO₃ powder was added to basify, and the whole was extracted with AcOEt (2 mL). The organic phase was dried over anhydrous sodium sulfate and evaporated to give an oily residue, which was purified by column chromatography (CHCl₃: MeOH = 10:1) on silica gel to give 18 (27 mg, 75%) as a pale yellow viscous material; v_{max} (CHCl₃) 3484, 2979, 1655, 1477, 1240, 1096 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 3.75 (s, 3H, NCH₃), 3.82 (s, 3H, OCH_3), 3.90 (s, 3H, OCH_3), 3.99 (s, 3H, OCH_3), 6.33 (s, 1H, ArOH), 6.97 (s, 1H, Ar–H), 7.08 (d, 2H, J = 8.8 Hz, Ar–H), 7.21-7.36 (m, 5H, Ar–H), 7.59 (d, 2H, J = 8.8 Hz, Ar–H), 7.94(s, 1H, Ar–H); δ (CDCl₃) 31.0, 55.3, 55.5, 61.4, 97.0, 99.1, 113.8, 118.6, 125.9, 127.29, 127.32, 128.8, 129.3, 129.5, 132.1, 132.4, 132.6, 135.2, 141.6, 143.2, 150.3, 150.8, 158.9 [HRMS m/z Calc. for $C_{27}H_{24}N_2O_4S$: M, 472.1457. Found: M^+ , 472.1458].

5-[1-Hydroxy-1-(2-*tert*-butyldimethylsiloxy-3,4-dimethoxy-phenyl)methyl]-1-methyl-2-phenylthio-1*H*-imidazole (20)

A solution of *n*-BuLi in *n*-hexane (1.6 M; 12.7 mL, 20.3 mmol) was added dropwise to a solution of 2,2,6,6-tetramethylpiperidine (3.8 mL, 22.3 mmol) in THF (50 mL) under an N_2 atmosphere at -78 °C, and the mixture was stirred for 15 min. A solution of 11 (3.86 g, 20.3 mmol) in THF (6 mL) was added

dropwise to the mixture, and the mixture was stirred for 1 h at -78 °C. A solution of 2-tert-butyldimethylsiloxy-3,4-dimethoxybenzaldehyde 198 (6.02 g, 20.3 mmol) in THF (6 mL) was added dropwise to the mixture and the whole was stirred at the same temperature for 2 h. Water (20 mL) was added, and the mixture was extracted with AcOEt (50 mL). The organic layer was dried over anhydrous sodium sulfate and evaporated to give a crystalline mass, which was purified by column chromatography (AcOEt-n-hexane = 1 : 1) on silica gel followed by recrystallization from n-hexane-AcOEt to afford 20 (7.70 g, 78%), mp 122.3–122.5 °C (colorless needles); v_{max} (CHCl₃) 3150, 2928, 1597, 1455, 1277, 1098, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 0.09 (s, 3H, $SiCH_3$), 0.23 (s, 3H, $SiCH_3$), 0.88 [s, 9H, $SiC(CH_3)_3$], 3.63 (s, 3H, NCH₃), 3.76 (s, 3H, OCH₃), 3.88 (s, 3H, OCH₃), 6.09 (s, 1H, ArCH(OH)Im), 6.60 (d, 1H, J = 8.6 Hz, Ar-H), 6.69 (s, 1H, Im-H), 7.01 (d, 1H, J = 8.8 Hz, Ar-H), 7.12-7.26 (m, 5H, Ar-H); δ (CDCl₃) -4.5, -4.3, 18.6, 25.9, 31.8, 55.9, 60.4, 63.1, 105.2, 122.2, 124.9, 126.5, 128.0, 129.2, 129.5, 134.8, 136.7, 139.0, 139.4, 146.5, 153.4 [Calcd for C₂₅H₃₄N₂O₄SSi: C, 61.69; H, 7.04; N, 5.76. Found; C, 61.45; H, 7.07; N, 5.99%. HRMS m/z Calc. for $C_{25}H_{34}N_2O_4SSi: M$, 486.2008. Found: M^+ , 486.2015].

5-[1-(2-tert-Butyldimethylsiloxy-3,4-dimethoxyphenyl)methyl]-1-methyl-2-phenylthio-1*H*-imidazole (21)

To a solution of 20 (2.0 g, 4.1 mmol) in 12 ml of CH₂Cl₂ were added a solution of triethylsilane (3.3 ml, 20.5 mmol) and a solution of TFA (1.9 ml, 24.6 mmol). The solution was stirred for 12 h at rt under N2 and quenched by the addition of saturated aqueous NaHCO₃ solution (15 ml). The mixture was extracted with CHCl₃ (15 mL). The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue, which was purified by column chromatography (AcOEt-nhexane = 1 : 2) on silica gel to give 21 (1.90 g, 98%) as a pale yellow viscous material; v_{max} (CHCl₃) 2916, 1457, 1098, 834 cm^{-1} ; δ_{H} (CDCl₃) 0.21 [s, 6H, Si(CH₃)₂], 0.97 [s, 9H, SiC(CH₃)₃], 3.40 (s, 3H, NC H_3), 3.76 (s, 3H, OC H_3), 3.84 (s, 3H, OC H_3), 3.86 (s, 2H, ArC H_2 Im), 6.47 (d, 1H, J = 8.6 Hz, Ar–H), 6.55 (d, 1H, J = 8.6 Hz, Ar–H), 6.90 (s, 1H, Im–H), 7.10–7.26 (m, 5H, Ar-H); δ (CDCl₃) -4.2, 18.7, 25.4, 26.0, 31.1, 55.8, 60.3, 105.0, 121.4, 123.5, 126.3, 127.6, 129.1, 129.2, 134.3, 135.4, 137.2, 139.8, 147.1, 152.4 [HRMS m/z Calc. for $C_{25}H_{34}N_2O_3SSi$: M, 470.2059. Found: M⁺, 470.2050].

4-Bromo-5-[1-(2-tert-butyldimethylsiloxy-3,4-dimethoxyphenyl)methyl]-1-methyl-2-phenylthio-1*H*-imidazole (22)

NBS (182 mg, 1.02 mmol) was added to a solution of 21 (482 mg, 1.02 mmol) in THF (4 ml) under an N₂ atmosphere at 0 °C, and the whole was stirred for 1 h at 0 °C. Then water (1 ml) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt-n-hexane = 1 : 2) on silica gel to afford 22 (454 mg, 81%), mp 86.6-88.4 °C (colorless crystals, recrystallized from *n*-hexane–AcOEt); v_{max} (CHCl₃) 2915, 1599, 1457, 1253, 1098, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 0.25 [s, 6H, $Si(CH_3)_2$], 1.03 [s, 9H, $SiC(CH_3)_3$], 3.36 (s, 3H, NCH_3), 3.75 (s, 3H, OCH_3), 3.82 (s, 3H, OCH_3), 3.92 (s, 2H, ArC H_2 Im), 6.38 (d, 1H, J = 8.6 Hz, Ar–H), 6.44 (d, 1H, J =8.6 Hz, Ar-H), 7.16-7.29 (m, 5H, Ar-H); δ (CDCl₃) -4.1, 18.8, 24.3, 26.1, 32.0, 55.8, 60.3, 105.2, 116.1, 120.8, 122.4, 126.8, 128.0, 129.3, 132.1, 134.4, 137.1, 139.8, 146.9, 152.4 [Calc. For C₂₅H₃₃BrN₂O₃SSi: C, 54.63; H, 6.05; N, 5.10. Found: C, 54.41; H, 5.99; N, 5.19%. MS m/z (% base): 551 (3), 550 (7), 549 (2), 548 (6), 496 (2), 495 (12), 494 (28), 493 (100), 492 (26), 491 (92), 478 (18), 476 (15), 397 (15), 209 (27), 199 (24). HRMS m/z Calc. for C₂₅H₃₃BrN₂O₃SSi: M, 548.1164. Found: M⁺, 548.1177].

5-[1-(2-tert-Butyldimethylsiloxy-3,4-dimethoxyphenyl)methyl]-4-[1-hydroxy-1-(4-methoxyphenyl)methyl]-1-methyl-2-phenylthio-1*H*-imidazole (23)

[Method A]. A solution of tert-BuLi in n-pentane (1.56 M; 0.17 mL, 0.26 mmol) was added dropwise to a solution of 22 (72 mg, 0.13 mmol) in THF (1.0 mL) under an N₂ atmosphere at -78 °C. Stirring was continued for 1 h, then a solution of p-anisaldehyde (0.08 mL, 0.66 mmol) in THF (0.5 mL) was added dropwise at -78 °C. Stirring was continued for 3 h at -78 °C, then water (2 mL) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt-n-hexane = 1 : 2) on silica gel to give 23 (12 mg, 15%) as a pale yellow viscous material.

[Method B]. A solution of tert-BuLi in n-pentane (1.56 M; 0.23 mL, 0.36 mmol) was added dropwise to a mixed solution of 22 (100 mg, 0.18 mmol) and p-anisaldehyde (0.11 mL, 0.91 mmol) in THF (1.5 mL) under an N₂ atmosphere at -78 °C. Stirring was continued at -78 °C and a solution of *tert*-BuLi in n-pentane [1.56 M; 0.46 mL (0.23 mL × 2), 0.72 mmol] was added to the reaction mixture every 15 min until TLC of the reaction mixture indicated disappearance of the starting compound 22. After stirring was continued for 1 h at -78 °C, water (1 mL) was added to the mixture. The mixture was extracted with AcOEt, and the organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt-n-hexane = 1 : 2) on silica gel to give 23 (98 mg, 89%) as a pale yellow viscous material; v_{max} (CHCl₃) 3400, 2916, 1602, 1457, 1246, 1098, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 0.197 (s, 3H, SiC H_3), 0.203 (s, 3H, SiC H_3), 0.98 [s, 9H, SiC(C H_3)₃], 3.29 (s, 3H, NCH₃), 3.61 (br s, 1H, OH), 3.73 (s, 3H, OCH₃), 3.75 (s, 2H, $ArCH_3Im$), 3.76 (s, 3H, OCH₃), 3.79 (s, 3H, OCH₃), 5.73 (d, 1H, J = 3.7 Hz, ArCH(OH)Im), 6.10 (d, 1H, J = 8.6 Hz, Ar-H), 6.32 (d, 1H, J = 8.6 Hz, Ar–H), 6.79 (d, 2H, J = 8.8 Hz, Ar–H), 7.10–7.35 (m, 7H, Ar–H); δ (CDCl₃) –4.2, 18.7, 24.0, 26.1, 31.1, 55.2, 55.8, 60.3, 69.7, 104.9, 113.6, 121.4, 122.3, 126.4, 127.4, 127.9, 127.9, 129.2, 135.1, 135.6, 136.0, 139.7, 142.8, 146.8, 152.2, 158.9 [HRMS m/z Calc. for $C_{33}H_{42}N_2O_5SSi: M$, 606.2583. Found: M⁺, 606.2591].

8-(tert-Butyldimethylsilyloxy)-6,7-dimethoxy-4-(4-methoxyphenyl)-1-methyl-2-phenylthio-4,9-dihydro-1H-naphtho[2,3-d]imidazole (26)

To a solution of 23 (54 mg, 0.09 mmol) in 1.5 ml of CH₂Cl₂ were added a solution of triethylsilane (0.02 mL, 0.13 mmol) and a solution of TFA (0.02 mL, 0.27 mmol). The solution was stirred for 12 h at rt under N₂ and quenched by the addition of saturated aqueous NaHCO₃ solution (2 ml). The mixture was extracted with CHCl₃ (2 mL). The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue, which was purified by column chromatography (AcOEt-nhexane = 1:1) on silica gel to give 26 (41 mg, 79%) as colorless crystals; v_{max} (CHCl₃) 2917, 1603, 1490, 1246, 1126, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 0.27 (s, 3H, SiCH₃), 0.28 (s, 3H, SiCH₃), 1.07 [s, 9H, $SiC(CH_3)_3$], 3.55 (s, 3H, NC H_3), 3.746 (s, 3H, OC H_3), 3.752 (s, 3H, OC H_3), 3.754 (s, 3H, OC H_3), 3.83 (d, 1H, J = 3.7 Hz, $ArCH_aH_bIm$), 3.87 (d, 1H, J = 3.5 Hz, $ArCH_aH_bIm$), 5.22 (t, 1H, J = 3.4 Hz, ArCH(Ar)Im), 6.40 (s, 1H, Ar–H), 6.78 (d, 2H, J = 8.6 Hz, Ar-H, 7.02-7.22 (m, 7H, Ar-H) [HRMS m/z Calc.for C₃₃H₄₀N₂O₄SSi: M, 588.2478. Found: M⁺, 588.2470].

5-[1-(2-tert-Butyldimethylsiloxy-3,4-dimethoxyphenyl)methyl]-4-[1-(4-methoxyphenyl)methyl]-1-methyl-1*H*-imidazole (25)

Sodium borohydride (1.08 g, 28.51 mmol) was added to a solution of 23 (412 mg, 0.68 mmol) and nickel(II) chloride hexahydrate (2.26 g, 9.51 mmol) in MeOH–THF = 1:1 (20 mL) under an N₂ atmosphere at 0 °C, and the whole was refluxed for 2 h. The solvent was evaporated off, then water (20 mL) was added to the residue. The mixture was extracted with CHCl₃. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (CHCl₃-MeOH = 20:1) on silica gel to give 25 (202 mg, 62%) as a pale yellow viscous material; v_{max} (CHCl₃) 2918, 1601, 1457, 1241, 1099, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 0.25 [s, 6H, Si(C H_3)₂], 1.03 [s, 9H, $SiC(CH_3)_3$], 3.33 (s, 3H, NC H_3), 3.75 (s, 3H, OC H_3), 3.76 (s, 3H, OCH_3), 3.81 (s, 3H, OCH_3), 3.82 (s, 2H, $ArCH_2Im$), 3.84 (s, 2H, ArC H_2 Im), 6.21 (d, 1H, J = 8.6 Hz, Ar–H), 6.36 (d, 1H, 2H, J = 8.6 Hz, Ar–H), 7.37 (s, 1H, Im–H); δ (CDCl₃) –4.1, 18.8, 23.0, 26.1, 31.4, 33.0, 55.2, 55.8, 60.3, 104.8, 113.7, 122.5, 122.7, 125.1, 129.5, 133.0, 136.6, 139.2, 146.8, 152.1, 157.7 [HRMS m/z Calc. for C₂₇H₃₈N₂O₄Si: M, 482.2601. Found: M⁺, 482.2592].

2-Bromo-5-[1-(2-tert-butyldimethylsiloxy-3,4-dimethoxyphenyl)-methyl]-4-[1-(4-methoxyphenyl)methyl]-1-methyl-1H-imidazole (29)

NBS (9 mg, 0.05 mmol) was added to a solution of 25 (25 mg, 0.05 mmol) in THF (0.5 ml) under an N_2 atmosphere at 0 °C, and then the whole was stirred for 1 h at 0 °C. Then water (0.5 ml) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt-n-hexane = 1 : 2) on silica gel to give 29 (17 mg, 59%) as a pale yellow viscous material; v_{max} (CHCl₃) 2917, 1602, 1460, 1244, 1098, 834 cm⁻¹ $\delta_{\rm H}$ (CDCl₃) 0.24 [s, 6H, Si(CH₃)₂], 1.02 [s, 9H, SiC(CH₃)₃], 3.27 (s, 3H, NC H_3), 3.76 (s, 6H, 2 × OC H_3), 3.810 (s, 2H, ArC H_2 Im), 3.813 (s, 3H, OC H_3), 3.83 (s, 2H, ArC H_2 Im), 6.24 (d, 1H, J =8.6 Hz, Ar-H), 6.38 (d, 1H, J = 8.6 Hz, Ar-H), 6.77 (d, 2H, J =8.8 Hz, Ar-H), 7.15 (d, 2H, J = 8.8 Hz, Ar-H); δ (CDCl₃) -4.1, 18.8, 23.9, 26.1, 32.0, 33.0, 55.2, 55.8, 60.3, 104.9, 113.7, 118.3, 121.9, 122.5, 128.3, 129.5, 132.5, 139.7, 139.8, 146.8, 152.2, 157.8 [MS m/z (% base): 562 (2), 560 (2), 507 (2), 506 (8), 505 (28), 504 (8), 503 (26), 383 (10), 381 (10), 279 (6), 209 (6), 121 (100). HRMS m/z Calc. for $C_{27}H_{37}BrN_2O_4Si$: M, 560.1705. Found: M⁺, 560.1714].

2-Azido-5-[1-(2-*tert*-butyldimethylsiloxy-3,4-dimethoxyphenyl)-methyl]-4-[1-(4-methoxyphenyl)methyl]-1-methyl-1*H*-imidazole (30)

A solution of tert-BuLi in n-pentane (1.56 M; 0.43 mL, 0.66 mmol) was added dropwise to a solution of 29 (124 mg, 0.22 mmol) under an N_2 atmosphere at -78 °C. The mixture was stirred for 15 min at -78 °C, then trisyl azide (205 mg, 0.66 mmol) was added, and then the whole was stirred for 1 h at -78 °C. Water (1 mL) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (AcOEt-n-hexane = 1:3) on silica gel to give 30 (90 mg, 78%) as a pale yellow viscous material; v_{max} (CHCl₃) 2916, 2120, 1601, 1499, 1245, 1098, 834 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 0.23 [s, 6H, $Si(CH_3)_2$], 1.01 [s, 9H, $SiC(CH_3)_3$], 3.07 (s, 3H, NCH_3), 3.746 (s, 3H, OCH₃), 3.753 (s, 2H, ArCH₂Im), 3.76 (s, 3H, OCH₃), 3.79 (s, 2H, ArC H_2 Im), 3.81 (s, 3H, OC H_3), 6.27 (d, 1H, J = 8.6 Hz, Ar-H), 6.37 (d, 1H, J = 8.6 Hz, Ar-H), 6.78 (d, 2H, J = 8.8 Hz, Ar-H), 7.16 (d, 2H, J = 8.8 Hz, Ar-H); δ (CDCl₃) -4.1, 18.8, 23.3, 26.1, 29.2, 32.7, 55.2, 55.8, 60.3, 104.9, 113.7, 122.4, 122.5, 124.5, 129.4, 132.7, 136.5, 138.9, 139.7, 146.8, 152.1, 157.8 [HRMS m/z Calc. for $C_{27}H_{37}N_5O_4Si: M$, 523.2614. Found: M^+ , 523.2612].

2-Azido-5-[1-(2-hydroxy-3,4-dimethoxyphenyl)methyl]-4-[1-(4-methoxyphenyl)methyl]-1-methyl-1*H*-imidazole (31)

A solution of TBAF in THF (1 M; 1.00 mL, 1.00 mmol) was added dropwise to a solution of 30 (476 mg, 0.91 mmol) in THF (5 mL) at room temperature. The mixture was stirred for 5 min at room temperature. Water (2 mL) was added, and the mixture was extracted with AcOEt. The organic layer was dried over anhydrous sodium sulfate and evaporated to give an oily residue. The crude product was purified by column chromatography (CHCl₃-MeOH = 50 : 1) on silica gel to give 31, yellow crystals (322 mg, 87%), mp 140 °C dec. (recrystallized from CHCl₃-n-hexane); v_{max} (CHCl₃) 3475, 2924, 2121, 1607, 1500, 1458, 1240, 1169, 1093 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 3.13 (s, 3H, NC H_3), 3.76 (s, 3H, OCH_3), 3.78 (s, 2H, $ArCH_2Im$), 3.816 (s, 3H, OCH₃), 3.823 (s, 2H, ArCH₂Im), 3.89 (s, 3H, OCH₃), 6.31 (d, 1H, J = 8.8 Hz, Ar-H), 6.38 (d, 1H, J = 8.6 Hz, Ar-H), 6.78 (d, 2H, J = 8.8 Hz, Ar-H), 7.16 (d, 2H, J = 8.8 Hz, Ar-H); δ (CDCl₃) 22.4, 29.3, 32.6, 55.2, 55.8, 60.9, 103.5, 113.7, 117.1, 123.3, 124.4, 129.4, 132.8, 135.2, 136.2, 138.8, 146.9, 150.9, 157.8 [HRMS m/z Calc. for C₂₁H₂₃N₅O₄: M, 409.1750. Found: M+, 409.1749].

Naamine C 2

A mixture of 31 (52 mg, 0.13 mmol) and 10% Pd/C (10 mg) in EtOH (3 mL) was stirred for 24 h under an H₂ atmosphere at room temperature. The catalyst was removed by filtration with CHCl₃ and the filtrate was evaporated to give an oily residue. The crude product was purified by column chromatography $(CHCl_2 : MeOH = 5 : 1)$ on silica gel to give 2 as a vellow powder ^{1/510} (47 mg, 97%); v_{max} (CHCl₃) 3251, 3107, 2932, 1660, 1608, 1504, 1458, 1236, 1093 cm⁻¹; $\delta_{\rm H}$ (CDCl₃) 3.25 (s, 3H, NCH₃), 3.736 (br s, 2H, ArCH₂Im), 3.742 (s, 3H, OCH₃), 3.75 (br s, 2H, ArC H_2 Im), 3.83 (s, 3H, OC H_3), 3.89 (s, 3H, OC H_3), 6.35 (d, 1H, J = 8.6 Hz, Ar–H), 6.47 (d, 1H, J = 8.6 Hz, Ar–H), 6.77 (d, 2H, J = 8.8 Hz, Ar–H), 7.10 (d, 2H, J = 8.8 Hz, Ar–H); δ (CDCl₃) 22.1, 29.7, 29.9, 55.2, 55.8, 61.0, 103.7, 114.0, 115.9, 120.9, 123.3, 124.7, 129.5, 130.1, 135.5, 146.7, 147.2, 151.3, 158.2 [HRMS m/z Calc. for C₂₁H₂₅N₃O₄: M, 383.1845. Found: M+, 383.1843].

Pyronaamidine 4

A solution of 1-methylparabanic acid 32 (30 mg, 0.24 mmol) and N,N-diisopropylethylamine (0.11 mL, 0.61 mmol) and trimethylsilyl chloride (0.06 mL, 0.49 mmol) in CHCl₃ (0.5 mL) was stirred for 5 min under an N2 atmosphere at 0 °C, and then the stirring was continued for 2 h at room temperature. A solution of naamine C 2 (90 mg, 0.24 mmol) in CHCl₃ (0.5 mL) was added to the mixture, the whole was refluxed for 48 h. The solution was evaporated to give an oily residue, which was purified by column chromatography (CHCl₃-MeOH = 5:1) on silica gel to afford 4, yellow crystals (32 mg, 28%), mp 182.3–184.5 °C (recrystallized from CHCl₃–*n*-hexane; lit. ^{1e,12} mp 185–187 °C); ν_{max} (CHCl₃) 3476, 2977, 1783, 1732, 1659, 1609, 1563, 1504, 1457, 1388, 1297, 1240, 1172, 1144, 1094, 1031 cm⁻¹; $\delta_{\rm H}$ (CD₂Cl₂) 3.09 (s, 3H, NCH₃), 3.54 (s, 3H, $-OCH_3$), 3.75 (s, 3H, $-OCH_3$), 3.80 (s, 3H, $-OCH_3$), 3.87 (s, 3H, $-OCH_3$), 3.88 (s, 2H, ArC H_2 Im), 3.90 (s, 2H, ArC H_2 Im), 6.35 (d, 1H, J = 8.6 Hz, Ar-H), 6.44 (d, 1H, J = 8.6 Hz, Ar-H), 6.79(d, 2H, J = 8.8 Hz, Ar–H), 7.10 (d, 2H, J = 8.8 Hz, Ar–H); δ (CD₂Cl₂) 23.2, 24.7, 30.0, 32.2, 55.5, 56.1, 61.2, 104.0, 114.1, 116.4, 123.7, 126.9, 129.7, 131.9, 135.2, 135.8, 146.5, 146.8, 147.5, 151.6, 156.4, 158.5, 162.7 [Calc. For $C_{25}H_{27}N_5O_6 \cdot {}^{1/2}H_2O$: C, 59.75; H, 5.62; N, 13.94. Found: C, 59.56; H, 5.53; N, 13.81%. HRMS m/z Calc. for $C_{25}H_{27}N_5O_6$: M, 493.1961. Found: M+, 493.1974].

Acknowledgements

This research was financially supported, in part, by a Grant-In-Aid for the promotion of the advancement of education and

research in graduate schools in Subsidies for ordinary expenses of private schools from the Promotion and Mutual Aid Corporation for Private Schools, and the Frontier Research Program and a Grant-In-Aid for Encouragement of Young Scientists (to I. K.) from the Ministry of Education, Culture, Sports, Science and Technology (MEXT) of Japan.

References

- (a) S. Carmely and Y. Kashman, Tetrahedron Lett., 1987, 28, 3003;
 (b) S. Carmely, M. Ilan and Y. Kashman, Tetrahedron, 1989, 45, 2193;
 (c) B. R. Copp, C. R. Fairchild, L. Cornell, A. M. Casazza, S. Robinson and C. M. Ireland, J. Med. Chem., 1998, 41, 3909;
 (d) D. C. Dunbar, J. M. Rimoldi, A. M. Clark, M. Kelly and M. T. Hamann, Tetrahedron, 2000, 56, 8795;
 (e) R. K. Akee, T. R. Carroll, W. Y. Yoshida, P. J. Scheuer, T. J. Stout and J. Clardy, J. Org. Chem., 1990, 55, 1944;
 (f) X. Fu, J. R. Barnes, T. Do and F. J. Schmitz, J. Nat. Prod., 1997, 60, 497;
 (g) A. Plubrukarn, D. W. Smith, R. E. Cramer and B. S. Davidson, J. Nat. Prod., 1997, 60, 712;
 (h) X. Fu, F. J. Schmitz, R. S. Tanner and M. Kelly-Borges, J. Nat. Prod., 1998, 61, 384.
- 2 S. Ohta, N. Tsuno, S. Nakamura, N. Taguchi, M. Yamashita, I. Kawasaki and M. Fujieda, *Heterocycles*, 2000, **53**, 1939.
- 3 I. Kawasaki, S. Nakamura, S. Yanagitani, A. Kakuno, M. Yamashita and S. Ohta, *J. Chem. Soc., Perkin Trans. 1*, 2001, 3095.
- 4 S. Ohta, N. Tsuno, K. Maeda, S. Nakamura, N. Taguchi, M. Yamashita and I. Kawasaki, *Tetrahedron Lett.*, 2000, **41**, 4623.
- 5 (a) I. Kawasaki, N. Taguchi, T. Yamamoto, M. Yamashita and S. Ohta, *Tetrahedron Lett.*, 1995, **36**, 8251; (b) I. Kawasaki, N. Taguchi,

- M. Yamashita and S. Ohta, *Chem. Pharm. Bull.*, 1997, **45**, 1393; (c) S. Nakamura, N. Tsuno, M. Yamashita, I. Kawasaki, S. Ohta and Y. Ohishi, *J. Chem. Soc., Perkin Trans.* 1, 2001, 429.
- 6 T. G. Back, D. L. Baron and K. Yang, J. Org. Chem., 1993, 58, 2407
- 7 N. Kawai, Y. Fujibayashi, S. Kuwabara, K. Takao, Y. Ijuin and S. Kobayashi, *Tetrahedron*, 2000, **56**, 6467.
- 8 S. B. Singh and G. R. Pettit, J. Org. Chem., 1989, 54, 4105.
- R. E. Harmon, G. Wellman and S. K. Gupta, J. Org. Chem., 1973, 38, 11.
- 10 The reported physical data for the natural naamine C (ref. 1/): yellow powder; $\delta_{\rm H}$ (500 MHz; CDCl₃) 3.26 (s, 3H), 3.72 (s, 3H), 3.73 (br s, 4H), 3.81 (s, 3H), 3.87 (s, 3H), 6.34 (d, 1H, J = 8.7 Hz), 6.46 (d, 1H, J = 8.7 Hz), 6.76 (d, 2H, J = 8.7 Hz), 7.06 (d, 2H, J = 8.7 Hz); δ (125 MHz; CDCl₃) 22.6, 29.3, 29.6, 55.3, 55.8, 61.0, 103.8, 114.2, 115.1, 121.1, 122.7, 123.3, 129.0, 129.5, 135.5, 146.4, 147.1, 151.5, 158.5. IR was not reported in ref. 1f.
- 11 At this time *N,N*-diisopropylethylamine was used instead of Et₃N and imidazole, which were used in ref. 4. Although use of the latter bases resulted in low yield of 4 (10% yield), use of the former base somewhat improved the yield of 4 (28% yield).
- 12 The reported physical data for the natural pyronaamidine (ref. 1*e*): mp 185–187 °C (yellow feathery crystals); $\nu_{\rm max}$ (CHCl₃) 3403 (br) 1790, 1732, 1664, 1613, 1567, 1510, 1444, 1392, 1302, 1246, 1178, 1148, 1096, 1034, 968, 752, 606 cm⁻¹; $\delta_{\rm H}$ (500 MHz; CD₂Cl₂) 3.09 (s, 3H), 3.54 (s, 3H), 3.75 (s, 3H), 3.80 (s, 3H), 3.87 (s, 3H), 3.88 (s, 2H), 3.90 (s, 2H), 6.34 (d, 1H), 6.44 (d, 1H), 6.78 (d, 2H), 7.15 (d, 2H), 8.14 (br s, 1H), 8.14 (br s, 1H); δ (125 MHz; CD₂Cl₂) 23.2, 24.7, 29.9, 32.1, 55.5, 56.0, 61.1, 104.0, 114.1, 116.4, 123.7, 126.9, 129.7, 131.9, 135.1, 135.8, 146.5, 146.7, 147.5, 151.7, 156.4, 158.5, 162.7