

## Design, synthesis, and cytotoxic activities of new 2,4,5-triarylimidazoles

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**Abstract** A new group of 2,4,5-triarylimidazoles containing *N,N*-dimethylaminoethoxy or piperidinyl ethoxy group at the *para* position of the C-5 phenyl ring were synthesized and their cytotoxic activities were evaluated on three different breast cancer cell lines using MTT assay. The compounds contain various substituents at the *para* position of C-2 phenyl ring. Among the synthesized compounds, 4-(5-(4-(2-piperidin-1-yl)ethoxy)phenyl)-4-phenyl-1*H*-imidazol-2-yl)phenol (**11e**) and 1-2-(4-(2,4-diphenyl-1*H*-imidazol-5-yl)phenoxy)ethyl piperidine (**11h**) with IC<sub>50</sub>s of less than 0.1 μM on all three cell lines were the most potent cytotoxic compounds.

**Keywords** Synthesis · 2,4,5-Triarylimidazoles · Cytotoxic activity · MTT · Docking study

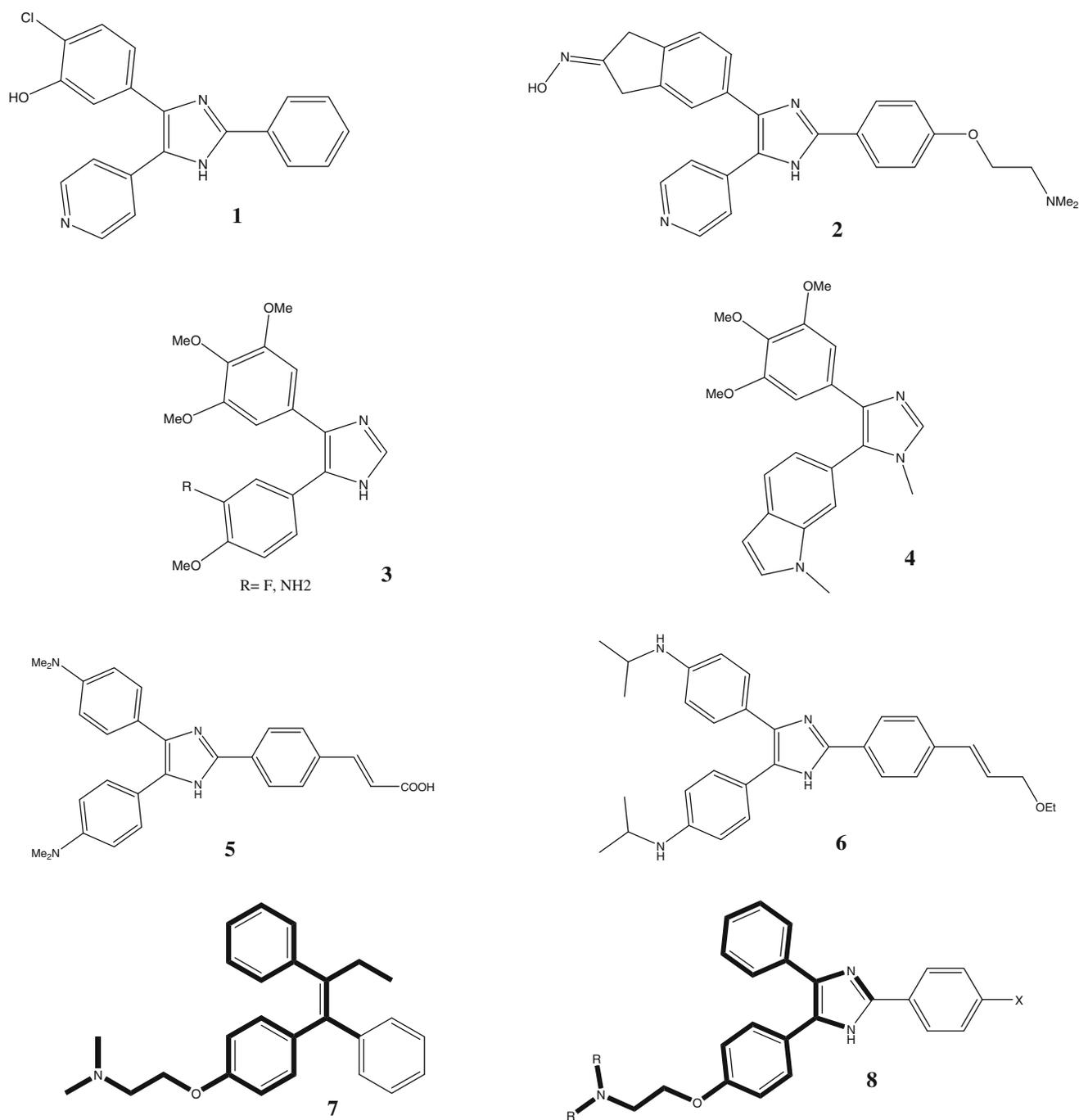
### Introduction

Imidazole scaffold exists in many significant biomolecules including biotin, histamine, and amino acids such as histidine. Many compounds possessing significant biochemical and pharmacological roles contain imidazole nucleus. Members of this class of heterocycles are known

to possess antibiotic (Brogden *et al.*, 1978), anti-fungal (Di Santo *et al.*, 2005), anti-inflammatory (Mano *et al.*, 2003), COX-2 inhibitors (Zarghi *et al.*, 2012), and cytotoxic activities (Narasimhan *et al.*, 2011). Imidazole compounds have been an interesting source for researchers for more than a century and many of them are being used as chemotherapy compounds today. A wide variety of imidazole containing compounds having cytotoxic activities on different cancer cell lines with different mechanisms of action have been developed over years, such as inhibitors of B-Raf kinase **1,2** (Takle *et al.*, 2006), combretastatin analogs **3,4** (Wang *et al.*, 2002), and modulators of P-glycoprotein (P-gp)-mediated multidrug resistance (MDR) **5,6** (Newman *et al.*, 2000; Robert and Jarry, 2003) shown in Fig. 1. Tamoxifen **7** was developed more than 30 years ago for the treatment of advanced breast cancer (Osborne, 1998). Tamoxifen, like other SERMs, binds to estrogen receptor (ER) and, in breast cancer cells, antagonizes the effect of estrogen on a variety of growth-regulatory genes (Dhingra, 1999). The predominant effect of tamoxifen and many other SERMs is cytostatic, by inducing a G1 cell cycle block, thereby slowing cell proliferation. Tamoxifen may also induce apoptosis, although this process has not always been easy to demonstrate in tumors (Ellis *et al.*, 1997). Based upon the anti-cancer properties of imidazole compounds in Fig. 1 and anti-breast cancer effects of tamoxifen, new 2,4,5-triaryl-1*H*-imidazoles were designed, synthesized, and their cytotoxic activities were evaluated on three breast cancer cell lines. For designing the new imidazoles, the vicinal diaryl moiety and the basic side chain (*N,N*-dialkylamino ethoxyphenyl) was derived from the structure of tamoxifen, while the total 2,4,5-triarylimidazole resembles the other cytotoxic imidazole compounds shown in Fig. 1.

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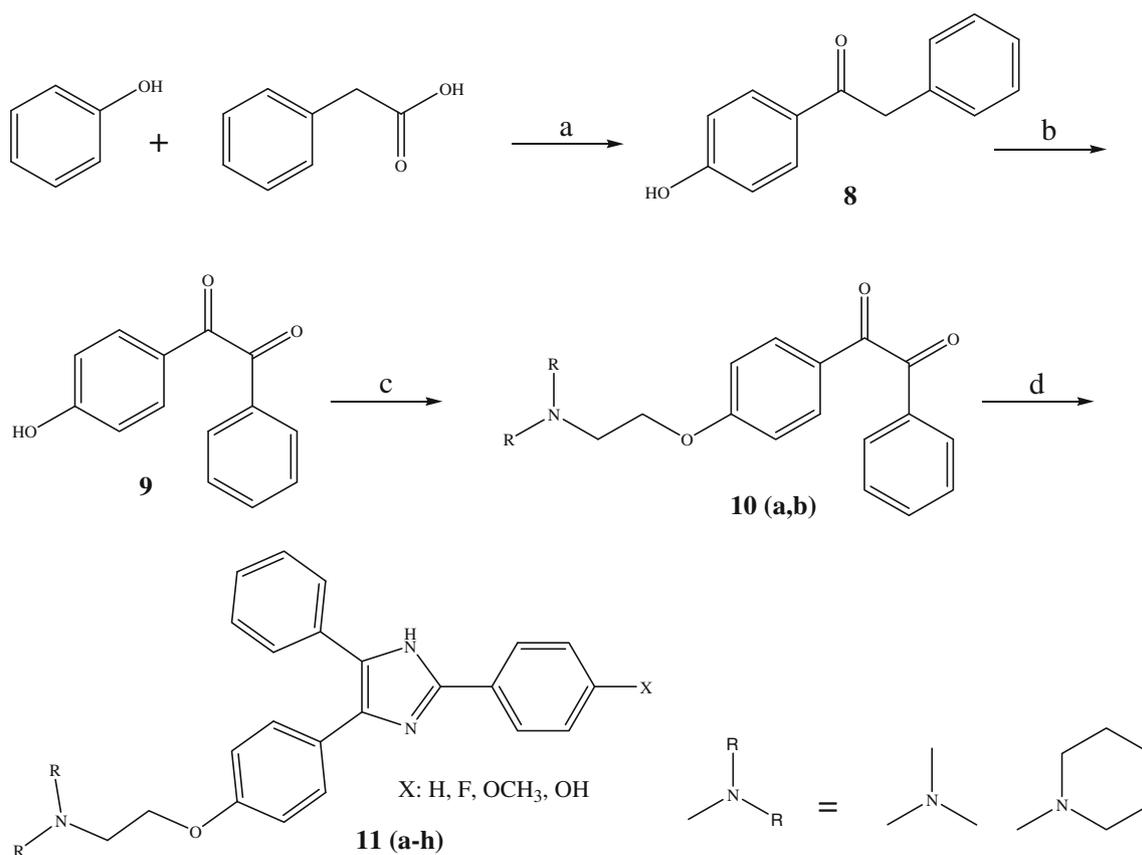
**Fig. 1** Structures of imidazole compounds, tamoxifen (7) and our designed molecules (8)

## Results and discussions

### Chemistry

As shown in Scheme 1 the target 2,4,5-triaryl-1*H*-imidazole derivatives **11a–h** were synthesized via a one-pot microwave reaction between 1,2-diketone **10** and the appropriate aromatic aldehyde in the presence of ammonium acetate under microwave irradiation (Wolkenberg *et al.*, 2004; Zarghi

*et al.*, 2012). Primarily 1-(4-hydroxyphenyl)-2-phenylethanone **8** was synthesized using Friedel–Crafts reaction between phenol and phenylacetic acid in the presence of Lewis acid  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (Wahala and Hase, 1991). Compound **8** was then oxidized to diketone **9** using selenium dioxide (Caturla *et al.*, 2004). Diketone **9** undergone a nucleophilic substitution reaction with appropriate reagents (*N*-(2-chloroethyl)-*N,N*-dimethylammonium chloride or *N*-(2-chloroethyl)-piperidinium chloride) to add the basic side chain and



**Scheme 1** Reagents and conditions: **a**  $\text{BF}_3 \cdot \text{Et}_2\text{O}$ , 80 °C; **b**  $\text{SeO}_2$ , dioxane/ $\text{H}_2\text{O}$ , reflux; **c** *N*-(2-chloroethyl)-*N,N*-dimethylammonium chloride or *N*-(2-chloroethyl)-piperidinium chloride,  $\text{K}_2\text{CO}_3$ , acetonitrile, reflux; **d** Ar-CHO,  $\text{NH}_4\text{OAc}$ , AcOH, MW, 180 W, 10 min

was converted to diketone **10 (a, b)** (Um et al., 2003). 1,2-Diketone **10** was used to synthesize the final 2,4,5-triaryl-1*H*-imidazoles **11a–h** via the multi-component microwave reaction. The purity of all products was determined by thin layer chromatography using several solvent systems of different polarity. All compounds were pure and stable. The compounds were characterized by  $^1\text{H}$ NMR, MS, IR, and CHN analysis. The physical data of final synthesized derivatives were summarized in Table 1.

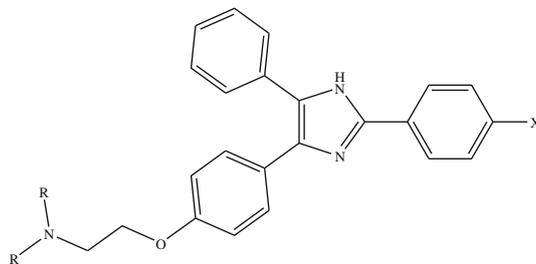
### Cytotoxicity

The cytotoxicity of compounds **11a–h** was determined on breast cancer cell lines using MTT assay (Mosmann, 1983). The MTT assay of the compounds was performed on three different breast cancer cell lines: MCF-7 and T47D (containing medium to high levels of estrogen receptors) and MDA-MB-231 (without estrogen receptors) (Siciliano et al., 1979; Dickstein et al., 1993; Keydar et al., 1979) to indicate that the anti-proliferative activities of designed compounds are mediated through hormone-dependent or hormone-independent mechanisms. The results of MTT assay of these compounds are shown in Table 2. According

to these results, all compounds **11a–h** were cytotoxic against three different breast cancer cell lines. In addition, our results indicated that the synthesized compounds **11a–h** were more cytotoxic against T47D cells which contain high amounts of estrogen receptors in comparison with those of MCF7 and MDA-MB-231. This may be due to the ability of synthesized compounds to blockade estrogen receptors in breast cancer cell lines. However, our results showed that some compounds (**11d**, **11e**, and **11h**) had similar cytotoxic activities on MDA-MB-231 cell line as well which indicated that other anti-cancer mechanisms may be involved in addition to blockade estrogen receptors in breast cancer cell lines. Moreover, our results indicated that among the synthesized compounds, **11e** and **11h** containing piperidino basic side chain with  $\text{IC}_{50}\text{s}$  of less than 0.1  $\mu\text{M}$  on all three cell lines had the highest cytotoxic effects and were the most potent compounds in this group.

### Docking study

Regarding the resemblance between the structures of the designed 2,4,5-triaryl-1*H*-imidazoles **11a–h** and the structure of tamoxifen and obtained MTT assay results, it could be

**Table 1** Physicochemical properties of compounds 2,4,5-triarylimidazoles **11a–h**

Compound	X		MP °C	Yield%	Molecular formula	Molecular weight
<b>11a</b>	OH		147– 149	36	C <sub>25</sub> H <sub>25</sub> N <sub>3</sub> O <sub>2</sub>	399.5
<b>11b</b>	F		172– 173	30	C <sub>25</sub> H <sub>24</sub> N <sub>3</sub> OF	401.5
<b>11c</b>	OMe		159– 160	46	C <sub>26</sub> H <sub>27</sub> N <sub>3</sub> O <sub>2</sub>	413.5
<b>11d</b>	H		150– 151	35	C <sub>25</sub> H <sub>25</sub> N <sub>3</sub> O	383.5
<b>11e</b>	OH		225– 227	20	C <sub>28</sub> H <sub>29</sub> N <sub>3</sub> O <sub>2</sub>	439.5
<b>11f</b>	F		188– 190	26	C <sub>28</sub> H <sub>28</sub> N <sub>3</sub> OF	441.5
<b>11g</b>	OMe		118– 120	30	C <sub>29</sub> H <sub>31</sub> N <sub>3</sub> O <sub>2</sub>	453.6
<b>11h</b>	H		117– 118	34	C <sub>28</sub> H <sub>29</sub> N <sub>3</sub> O	423.5

assumed that one of the mechanisms of cytotoxic activities of compounds **11a–h** on breast cancer cell lines might be through acting on estrogen receptors. Therefore, the orientation of 4-(5-(4-(2-(dimethylamino)ethoxy)phenyl)-4-phenyl-1*H*-imidazol-2-yl)phenol **11a** in the estrogen receptor  $\alpha$  (ER $\alpha$ ) active site was examined by a docking experiment

(Fig. 2). This molecular modeling study shows that compound **11a** was well bound in the active site of ER $\alpha$  so that the N atom of the tertiary amino group of the basic side chain (dimethylaminophenoxy), which can be protonized in physiologic pH, is in the vicinity of the oxygen of carboxylate group of Asp<sup>351</sup> (distance = 4.7 Å) and is capable of

**Table 2** In vitro cytotoxicity data of compounds **11a–h** using MTT assay

Compound	MDA-MB-231 IC <sub>50</sub> (μM)	T47D IC <sub>50</sub> (μM)	MCF-7 IC <sub>50</sub> (μM)
<b>11a</b>	82 ± 6	63 ± 3	70 ± 5
<b>11b</b>	76 ± 4	61 ± 3	82 ± 2
<b>11c</b>	53 ± 2	28 ± 2	42 ± 4
<b>11d</b>	58 ± 4	50 ± 3	53 ± 4
<b>11e</b>	<0.1	<0.1	<0.1
<b>11f</b>	>100	61 ± 4	78 ± 4
<b>11g</b>	71 ± 2	54 ± 2	65 ± 4
<b>11h</b>	<0.1	<0.1	<0.1
Tamoxifen	55.5 ± 2.7	12.1 ± 0.9	25.1 ± 1.5

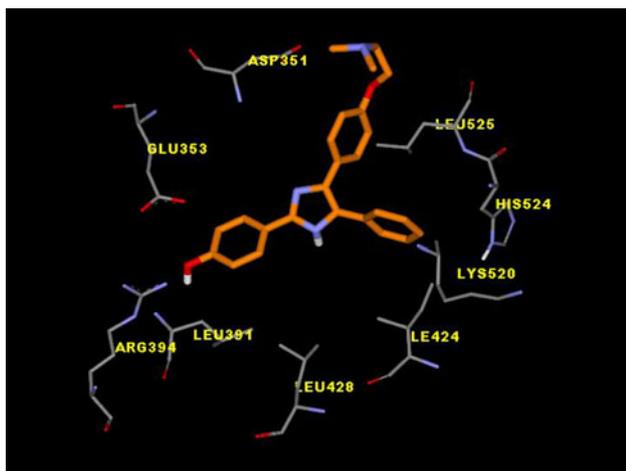
Experimental values represent the average for two experiments performed in triplicate along with the standard deviation (SD) between the assay values

binding to this amino acid. The OH group at the *para* position of C-2 phenyl ring is in the vicinity of ASP<sup>394</sup> and Glu<sup>353</sup> so that the H atom of OH group is in a 4.3 Å distance from the O atom of carboxylate group in Glu<sup>353</sup> and the O atom of the OH group is in a 3.0 Å distance from H atom of the NH group in Arg<sup>394</sup> (Barrett *et al.*, 2008). These data together with biological results may explain that one of the mechanisms of cytotoxic activities of compounds **11a–h** on breast cancer cell lines might be through acting on estrogen receptors.

## Experimental

### Materials

All reagents were purchased from the Aldrich (USA) or Merck (Germany) Chemical Company and were used without further purifications.



**Fig. 2** Docking **11a** (in orange) in the active site of human ER $\alpha$ . Hydrogen atoms have been removed to improve clarity (Color figure online)

### General

Melting points (mp) were determined using a Thomas Hoover melting point apparatus (Philadelphia, USA). Synthesis 3000 microwave oven (Anton Paar, Austria) was used to synthesize the final compounds. Infrared spectra were acquired on a Perkin Elmer 1420 ratio recording spectrometer. A Bruker FT-500 MHz instrument (Bruker Biosciences, Germany) was used to acquire <sup>1</sup>H NMR spectra and chloroform-D was used as solvent. Coupling constant (*J*) values are estimated in hertz (Hz) and spin multiples are given as s (singlet), d (double), t (triplet), q (quartet), m (multiplet), and br (broad). The mass spectral measurements were performed on an 6410 Agilent LCMS triple quadrupole mass spectrometer (LCMS) with an electrospray ionization (ESI) interface. Elemental analyses were carried out with a Perkin Elmer Model 240-C apparatus (Perkin Elmer, Norwalk, CT, USA). The results of the elemental analyses (C, H, N) were within ± 0.4 % of the calculated amounts.

### Preparation of 1-(4-hydroxyphenyl)-2-phenylethanone **8**

Equimolar amounts of phenol and phenyl acetic acid were added to 40 ml of BF<sub>3</sub>·Et<sub>2</sub>O and stirred under Argon atmosphere in 0 °C for 15 min. The reaction temperature was then allowed to gradually reach the room temperature and subsequently the reaction mixture was refluxed in 75 °C for 20 h. After cooling, the reaction mixture was poured onto crushed ice, 20 ml of ethyl acetate was added and the organic phase was separated and washed primarily with water and afterward with saturated NaHCO<sub>3</sub> solution, dried with sodium sulfate and evaporated under vacuum. The oily residue was dissolved in DCM and precipitates formed by adding hexane drop wise. The precipitate was filtered and recrystallized in DCM to obtain white powder. Yield: 21 %; Mp: 148–150 °C; IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3,340 (OH), 1,670 (C=O); MS *m/z* (%): 212.1 (M<sup>+</sup>, 10), 181.0 (15), 165.0 (10), 152.0 (10), 120.9 (10), 93.1 (60).

### Preparation of 1-(4-hydroxyphenyl)-2-phenylethan-1,2-dione **9**

4 g (0.0188 mmol) of compound **8** was dissolved in 15 ml dioxane and added to a solution of 12 g SeO<sub>2</sub> in 50 ml dioxane and 10 ml water. The reaction mixture was refluxed for 24 h. After cooling, selenium was filtered off and the filtrate was poured onto crushed ice and extracted with ethyl acetate. The organic phase was washed with water and dried with sodium sulfate and evaporated under vacuum. The oily residue was dissolved in DCM and the precipitates were formed by adding hexane drop wise. The precipitate was filtered and recrystallized in DCM to obtain orange powder. Yield: 64 %; Mp: 127–129 °C; IR (KBr):  $\nu$  (cm<sup>-1</sup>) 3,385

(OH), 1,670, 1,645 (C=O); MS *m/z* (%): 226.1 (M<sup>+</sup>, 10), 121.0 (100), 105.1 (75), 93.1 (70), 77.0 (85).

*Preparation of 1-(4-(2-(piperidin-1-yl or dimethylamino)ethoxy)phenyl)-2-phenylethan-1,2-dione (10a, b)*

1 g (4.42) mmol of diketone **9** was dissolved in 40 ml acetonitrile and 5.8 g K<sub>2</sub>CO<sub>3</sub> was added while stirring. Four equivalents of *N*-(2-chloroethyl)piperidinium chloride or *N*-(2-chloroethyl)-*N,N*-dimethylammonium chloride was added subsequently and the reaction mixture was refluxed for 24 h. The inorganic material was filtered off and the filtrate was evaporated under vacuum, dissolved in DCM and washed with brine. The organic phase was dried with sodium sulfate and evaporated. The oily residue was purified using column chromatography with a mobile phase of CHCl<sub>3</sub>/*n*-hexane (9:1 v/v), to obtain a pure yellow viscous oil.

*1-(4-(2-(Piperidin-1-yl)ethoxy)phenyl)-2-phenylethan-1,2-dione 10a*

Yield: 30 %; IR (CCl<sub>4</sub>): *v* (cm<sup>-1</sup>) 1,715, 1,685 (C=O); LC-MS (ESI) *m/z*: 338.8 (M + 1) (100); <sup>1</sup>HNMR (CDCl<sub>3</sub>, 500 MHz): δ 1.48–1.52 (m, 2H, piperidine CH<sub>2</sub>), 1.65–1.69 (m, 4H, piperidine CH<sub>2</sub>), 2.57–2.60 (m, 4H, piperidine CH<sub>2</sub>), 2.86–2.88 (t, 2H, CH<sub>2</sub>N, *J* = 5.9 Hz), 4.23–4.25 (t, 2H, OCH<sub>2</sub>, *J* = 5.9 Hz), 6.99 (d, 2H, 1-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 9.0 Hz), 7.53–7.56 (m, 2H, 2-phenyl H<sub>3</sub> and H<sub>5</sub>), 7.67–7.70 (t, 1H, 2-phenyl H<sub>4</sub>, *J* = 7.4 Hz), 7.96 (d, 2H, 1-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 9.0 Hz), 8.00 (d, 2H, 2-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 7.2 Hz).

*1-(4-(2-(Dimethylamino)ethoxy)phenyl)-2-phenylethan-1,2-dione 10b*

Yield: 32 %; IR (CCl<sub>4</sub>): *v* (cm<sup>-1</sup>) 1,720, 1,680 (C=O); LC-MS (ESI) *m/z*: 298.9 (M + 1) (100); <sup>1</sup>HNMR (CDCl<sub>3</sub>, 500 MHz): δ 2.39 (s, 6H, N(CH<sub>3</sub>)<sub>2</sub>), 2.80–2.82 (t, 2H, CH<sub>2</sub>N, *J* = 5.6 Hz), 4.18–4.20 (t, 2H, OCH<sub>2</sub>, *J* = 5.6 Hz), 7.03 (d, 2H, 1-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 8.6 Hz), 7.53–7.56 (m, 2H, 2-phenyl H<sub>3</sub> and H<sub>5</sub>), 7.67–7.70 (t, 1H, 2-phenyl H<sub>4</sub>, *J* = 7.3 Hz), 7.97 (d, 2H, 1-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 8.6 Hz), 8.01 (d, 2H, 2-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 8.2 Hz).

General procedure for preparation of 2,4,5-triaryl-1*H*-imidazoles **11a–h**

Equivalent amounts of diketone **10** (**a** or **b**) and appropriate aromatic aldehyde along with 2 g ammonium acetate in 6 ml glacial acetic acid were placed in microwave reactor for 10 min, while the power was set at 180 W. After cooling, the solution was neutralized with aqueous ammonia in which the

product precipitated immediately. The precipitate was filtered and washed with water and purified using column chromatography (Yields: 20–46 %).

*4-(5-(4-(2-(Dimethylamino)ethoxy)phenyl)-4-phenyl-1*H*-imidazol-2-yl)phenol 11a*

IR (KBr): *v* (cm<sup>-1</sup>) 3,500–2,500 (OH), 3,058 (NH); LC-MS (ESI) *m/z*: 400.8 (M + 1) (100); <sup>1</sup>HNMR (CDCl<sub>3</sub>, 500 MHz): δ 2.27 (s, 6H, N(CH<sub>3</sub>)<sub>2</sub>), 2.66–2.68 (t, 2H, CH<sub>2</sub>N, *J* = 5.7 Hz), 3.99–4.01 (t, 2H, OCH<sub>2</sub>, *J* = 5.7 Hz), 6.77–6.79 (d, 4H, 2-phenyl H<sub>3</sub> and H<sub>5</sub> and 5-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 8.7 Hz), 7.12–7.15 (m, 1H, 4-phenyl H<sub>4</sub>), 7.19–7.22 (m, 2H, 4-phenyl H<sub>3</sub> and H<sub>5</sub>), 7.37 (d, 2H, 4-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 8.6 Hz), 7.47 (d, 2H, 5-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 7.5 Hz), 7.79 (d, 2H, 2-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 8.6 Hz), 9.32 (s, 1H, NH); Anal. Calcd. for: C<sub>25</sub>H<sub>25</sub>N<sub>3</sub>O<sub>2</sub>: C, 75.16; H, 6.31; N, 10.52. Found: C, 75.32; H, 6.65; N, 10.39.

*2-(4-(2-(4-Fluorophenyl)-4-phenyl-1*H*-imidazol-5-yl)phenoxy)-*N,N*-dimethylethanamine 11b*

IR (KBr): *v* (cm<sup>-1</sup>) 3,050 (NH); LC-MS (ESI) *m/z*: 402.8 (M + 1) (100); <sup>1</sup>HNMR (CDCl<sub>3</sub>, 500 MHz): δ 2.39 (s, 6H, N(CH<sub>3</sub>)<sub>2</sub>), 2.79–2.81 (t, 2H, CH<sub>2</sub>N, *J* = 5.4 Hz), 4.12–4.14 (t, 2H, OCH<sub>2</sub>, *J* = 5.4 Hz), 6.90 (d, 2H, 5-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 8.0 Hz), 7.12–7.15 (t, 2H, 2-phenyl H<sub>3</sub> and H<sub>5</sub>), 7.29 (m, 1H, 4-phenyl H<sub>4</sub>), 7.33–7.35 (m, 2H, 5-phenyl H<sub>2</sub> and H<sub>6</sub>), 7.45 (m, 2H, 4-phenyl H<sub>3</sub> and H<sub>5</sub>), 7.58 (m, 2H, 4-phenyl H<sub>2</sub> and H<sub>6</sub>), 7.90–7.92 (t, 2H, 2-phenyl H<sub>2</sub> and H<sub>6</sub>), 9.44 (s, 1H, NH). Anal. Calcd. for: C<sub>25</sub>H<sub>24</sub>N<sub>3</sub>O: C, 74.79; H, 6.03; N, 10.47. Found: C, 74.99; H, 6.25; N, 10.31.

*2-(4-(2-(4-Methoxyphenyl)-4-phenyl-1*H*-imidazol-5-yl)phenoxy)-*N,N*-dimethylethanamine 11c*

IR (KBr): *v* (cm<sup>-1</sup>) 3,070 (NH); LC-MS (ESI) *m/z*: 414.8 (M + 1) (100); <sup>1</sup>HNMR (CDCl<sub>3</sub>, 500 MHz): δ 2.37 (s, 6H, N(CH<sub>3</sub>)<sub>2</sub>), 2.77–2.80 (t, 2H, CH<sub>2</sub>N, *J* = 5.7 Hz), 3.87 (s, 3H, OCH<sub>3</sub>), 4.10–4.12 (t, 2H, OCH<sub>2</sub>, *J* = 5.7 Hz), 6.88 (d, 2H, 5-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 8.7 Hz), 6.95 (d, 2H, 2-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 8.7 Hz), 7.25–7.28 (m, 1H, 4-phenyl H<sub>4</sub>), 7.32–7.34 (m, 3H, 5-phenyl H<sub>2</sub> and H<sub>6</sub>, and NH), 7.44 (d, 2H, 4-phenyl H<sub>3</sub> and H<sub>5</sub>, *J* = 7.6 Hz), 7.56 (d, 2H, 4-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 7.6 Hz), 7.86 (d, 2H, 2-phenyl H<sub>2</sub> and H<sub>6</sub>, *J* = 8.7 Hz); Anal. Calcd. for: C<sub>26</sub>H<sub>27</sub>N<sub>3</sub>O<sub>2</sub>: C, 75.52; H, 6.58; N, 10.16. Found: C, 75.66; H, 6.35; N, 10.22.

*2-(4-(2,4-Diphenyl-1*H*-imidazol-5-yl)phenoxy)-*N,N*-dimethylethanamine 11d*

IR (KBr): *v* (cm<sup>-1</sup>) 3,065 (NH); LC-MS (ESI) *m/z*: 384.9 (M + 1) (100); <sup>1</sup>HNMR (CDCl<sub>3</sub>, 500 MHz): δ 2.48 (s, 6H,

$N(CH_3)_2$ , 2.89–2.91 (t, 2H,  $CH_2N$ ,  $J = 5.2$  Hz), 4.18–4.20 (t, 2H,  $OCH_2$ ,  $J = 5.2$  Hz), 6.91 (d, 2H, 5-phenyl  $H_3$  and  $H_5$ ,  $J = 8.5$  Hz), 7.35–7.60 (m, 10H, 5-phenyl  $H_2$  and  $H_6$ , 4-phenyl and 2-phenyl  $H_3$ – $H_5$ ), 7.97 (d, 2H, 2-phenyl  $H_2$  and  $H_6$ ,  $J = 7.68$  Hz), 9.38 (s, 1H, NH); Anal. Calcd. for.  $C_{25}H_{25}N_3O$ : C, 78.30; H, 6.57; N, 10.96. Found: C, 78.61; H, 6.85; N, 11.12.

*4-(5-(4-(2-Piperidin-1-yl)ethoxy)phenyl)-4-phenyl-1H-imidazol-2-yl)phenol IIe*

IR (KBr):  $\nu$  ( $cm^{-1}$ ) 3,500–2,500 (OH), 3,056 (NH); LC–MS (ESI)  $m/z$ : 440.2 ( $M + 1$ ) (100);  $^1H$ NMR ( $CDCl_3$ , 500 MHz):  $\delta$  1.43–1.45 (m, 2H, piperidine  $CH_2$ ), 1.60–1.64 (m, 4H, piperidine  $CH_2$ ), 2.50–2.54 (m, 4H, piperidine  $CH_2$ ), 2.78–2.81 (t, 2H,  $CH_2N$ ,  $J = 5.9$  Hz), 4.10–4.12 (t, 2H,  $OCH_2$ ,  $J = 5.9$  Hz), 6.80 (d, 2H, 2-phenyl  $H_3$  and  $H_5$ ,  $J = 8.8$  Hz), 6.83 (d, 2H, 5-phenyl  $H_3$  and  $H_5$ ,  $J = 8.7$  Hz), 7.19–7.21 (m, 1H, 4-phenyl  $H_4$ ), 7.26–7.30 (m, 3H, 4-phenyl  $H_2$  and  $H_6$ , NH), 7.42 (d, 2H, 4-phenyl  $H_3$  and  $H_5$ ,  $J = 8.5$  Hz), 7.53 (d, 2H, 4-phenyl  $H_2$  and  $H_6$ ,  $J = 7.4$  Hz), 7.81 (d, 2H, 5-phenyl  $H_2$  and  $H_6$ ,  $J = 8.7$  Hz); Anal. Calcd. for.  $C_{28}H_{29}N_3O_2$ : C, 76.51; H, 6.65; N, 9.56. Found: C, 76.36; H, 6.45; N, 9.75.

*1-(2-(4-(2-(4-Fluorophenyl)-4-phenyl-1H-imidazol-5-yl)phenoxy)ethyl)piperidine IIIf*

IR (KBr):  $\nu$  ( $cm^{-1}$ ) 3,050 (NH); LC–MS (ESI)  $m/z$ : 442.8 ( $M + 1$ ) (100);  $^1H$ NMR ( $CDCl_3$ , 500 MHz):  $\delta$  1.49–1.53 (m, 2H, piperidine  $CH_2$ ), 1.65–1.69 (m, 4H, piperidine  $CH_2$ ), 2.57–2.61 (m, 4H, piperidine  $CH_2$ ), 2.85–2.87 (t, 2H,  $CH_2N$ ,  $J = 5.5$  Hz), 4.15–4.18 (t, 2H,  $OCH_2$ ,  $J = 5.5$  Hz), 6.90–6.94 (m, 2H, 5-phenyl  $H_3$  and  $H_5$ ), 7.15–7.18 (t, 2H, 2-phenyl  $H_3$  and  $H_5$ ,  $J = 7.0$  Hz), 7.34–7.69 (m, 7H, 5-phenyl  $H_2$  and  $H_6$  and 4-phenyl), 7.91–7.94 (dd, 2H, 2-phenyl  $H_2$  and  $H_6$ ), 9.46 (s, 1H, NH); Anal. Calcd. for.  $C_{28}H_{28}N_3OF$ : C, 76.16; H, 6.39; N, 9.52. Found: C, 76.23; H, 6.54; N, 9.69.

*1-(2-(4-(2-(4-Methoxyphenyl)-4-phenyl-1H-imidazol-5-yl)phenoxy)ethyl)piperidine IIg*

IR (KBr):  $\nu$  ( $cm^{-1}$ ) 3,068 (NH); LC–MS (ESI)  $m/z$ : 454.7 ( $M + 1$ ) (100);  $^1H$ NMR ( $CDCl_3$ , 500 MHz):  $\delta$  1.48–1.51 (m, 2H, piperidine  $CH_2$ ), 1.65–1.68 (m, 4H, piperidine  $CH_2$ ), 2.55–2.58 (m, 4H, piperidine  $CH_2$ ), 2.83–2.85 (t, 2H,  $CH_2N$ ,  $J = 5.5$  Hz), 3.90 (s, 3H,  $OCH_3$ ), 4.16–4.18 (t, 2H,  $OCH_2$ ,  $J = 5.5$  Hz), 6.79–6.80 (m, 2H, 5-phenyl  $H_3$  and  $H_5$ ), 7.00 (d, 2H, 2-phenyl  $H_3$  and  $H_5$ ,  $J = 7.9$  Hz), 7.35–7.69 (m, 6H, 5-phenyl  $H_2$  and  $H_6$ , 4-phenyl), 7.87 (d, 2H, 2-phenyl  $H_2$  and  $H_6$ ,  $J = 8.0$  Hz), 9.31 (s, 1H, NH); Anal. Calcd. for.  $C_{29}H_{31}N_3O_2$ : C, 76.79; H, 6.89; N, 9.26. Found: C, 76.99; H, 7.02; N, 8.98.

*1-2-(4-(2,4-Diphenyl-1H-imidazol-5-yl)phenoxy)ethyl)piperidine IIh*

IR (KBr):  $\nu$  ( $cm^{-1}$ ) 3,078 (NH); LC–MS (ESI)  $m/z$ : 424.7 ( $M + 1$ ) (100);  $^1H$ NMR ( $CDCl_3$ , 500 MHz):  $\delta$  1.55–1.58 (m, 2H, piperidine  $CH_2$ ), 1.81–1.84 (m, 4H, piperidine  $CH_2$ ), 2.78–2.83 (m, 4H, piperidine  $CH_2$ ), 3.02–3.04 (m, 2H,  $CH_2N$ ), 4.30–4.32 (m, 2H,  $OCH_2$ ), 6.88 (d, 2H, 5-phenyl  $H_3$  and  $H_5$ ,  $J = 7.5$  Hz), 7.37–7.59 (m, 10H, 5-phenyl  $H_2$  and  $H_6$ , 4-phenyl and 2-phenyl  $H_3$ – $H_5$ ), 8.00 (d, 2H, 2-phenyl  $H_2$  and  $H_6$ ,  $J = 7.6$  Hz), 8.88 (s, 1H, NH); Anal. Calcd. for.  $C_{28}H_{29}N_3O$ : C, 79.40; H, 6.90; N, 9.92. Found: C, 79.68; H, 6.77; N, 10.11.

Cytotoxicity studies

The anti-proliferative activities of the compounds were determined on three breast cancer cell lines using MTT assay.  $1 \times 10^4$  cells/well were seeded in 198  $\mu$ l RPMI (phenol red free) medium, supplemented with 10 % FBS in each well of 96-well micro culture plates and incubated for 24 h at 37 °C in a  $CO_2$  incubator. 2  $\mu$ l of the compounds, diluted to the desired concentrations in DMSO, were added to the wells with respective vehicle control. After 72 h of incubation, media were removed and to each well 20  $\mu$ l MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide) (5 mg/ml) was added and the plates were further incubated for 4.5 h. Then, the supernatant from each well was carefully removed, formazan crystals were dissolved in 200  $\mu$ l of DMSO and absorbance at 540 nm wavelength were recorded.

Molecular modeling (docking) studies

Docking studies were performed using Autodock Vina software (Trott and Olson, 2010). The coordinates of the X-ray crystal structure of the selective estrogen receptor modulator lasofoxifene bound to the human estrogen receptor  $\alpha$  was obtained from the RCSB Protein Data Bank (2OUZ) and hydrogens were added, Kollman charge was calculated and non-polar hydrogens were deleted. The ligand molecule was constructed using the Hyperchem 8.0.3 and was energy minimized using AMBER MM force field, Polak–Ribier algorithm for 1,000 iterations reaching a convergence of 0.01 kcal/mol Å. A grid box of 24–24–24 Å with the central X–Y–Z coordinates of X: 30.6130 Y: –1.2140 Z: 27.6360 was built for calculation of the energy map. In the end of docking process, conformations having optimal docking energy were visualized using Viewer Lite 5.0 software, residues with atoms  $>7.5$  Å from the docking box were removed for efficiency and the distances between atoms of ligand and amino acids in the active site were calculated. For docking validation, lasofoxifene was docked in the active

site of ER $\alpha$  with exactly similar conditions and the docked conformation having lowest docking energy was aligned with lasofoxifene in crystallography with ER $\alpha$  (2OUZ), using Pymol software and RMSD acquired was 1.74 Å showing that the docking method was valid (Ali *et al.*, 2007).

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