organic compounds

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4-(4-Hydroxybenzylideneamino)-4H-1,2,4-triazole hemihydrate

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In the title compound, 4-(4H-1,2,4-triazol-4-yliminomethyl)phenol hemihydrate, C₉H₈N₄O·0.5H₂O or (I)·0.5H₂O, molecules of (I) are arranged as layers running along the b axis through intermolecular $O-H\cdots N$ and $C-H\cdots O$ hydrogen bonds. These layers are stabilized by hydrogen-bonded water molecules to form three-dimensional networks.

Comment

The aroyl Schiff bases of 4-amino-1,2,4-triazole have received considerable attention over the past few decades (Kitaev et al., 1971; Mazza et al., 1976; Kargin et al., 1988). It is of interest that some of them are anti-inflammatory agents (Gupta & Bhargava, 1978) and new coccidiostatic drugs (Colautti et al., 1971). As a continuation of our study on the structure of Schiff-base-containing substituted 1,2,4-triazole (Shanmuga Sundara Raj et al., 1999), we report here the title structure, (I).



An ORTEPII (Johnson, 1976) diagram of (I) with numbering scheme is shown in Fig. 1. Compared to 9-(4H-1,2,4-triazol-4-ylimino)-4,5-diazafluorene (Shanmuga Sundara Raj et al., 1999), molecules of (I) are essentially planar [the maximum displacement from the least-squares mean plane through the whole molecule is 0.073 (2) Å for C9]. The bond lengths and angles observed in the structure are in the normal ranges. In contrast with 2-(2-hydroxybenzylidene)-1-(2-picoloyl)hydrazine hemihydrate (Wang et al., 1998), molecules of (I) are arranged as layers running along the b axis through strong intermolecular O1-H11···N4 $\left(-\frac{1}{2}+x, -\frac{1}{2}-y, \frac{1}{2}+z\right)$ hydrogen bonds and weak C9-H9···O1 $(\frac{1}{2} - x, -\frac{1}{2} + y)$, $-\frac{1}{2}-z$) ones. These layers are stabilized by water molecules to form three-dimensional networks through strong O1W-H1W···N3 hydrogen bonds and weak C7–H7···O1W(x, -y, $\frac{1}{2} + z$) ones. The geometry of these interactions are listed in





The structure of (I) showing 50% probability displacement ellipsoids with the numbering scheme.

Table 2. In the packing of molecules of the title compound, the water molecules (O1W) lie on crystallographic twofold axes.

Experimental

The title compound was prepared by condensation of equivalent amounts of p-hydroxybenzaldehyde and 4-amino-1,2,4-triazole in ethanol for 5 h (Kitaev et al., 1971). Diffraction-quality crystals were obtained by recrystallization from ethanol.

Crystal data

$C_9H_8N_4O\cdot0.5H_2O$	$D_x = 1.413 \text{ Mg m}^{-3}$
$M_r = 197.20$	Mo $K\alpha$ radiation
Monoclinic, C_2/c	Cell parameters from 36
u = 14.134(2)Å	reflections
p = 12.491 (2) Å	$\theta = 5.22 - 9.01^{\circ}$
r = 12.063 (2) Å	$\mu = 0.102 \text{ mm}^{-1}$
$B = 119.483 \ (10)^{\circ}$	T = 293 (2) K
$V = 1854.0(5) \text{ Å}^3$	Block, colourless
Z = 8	$0.30 \times 0.26 \times 0.20 \text{ mm}$

Data collection

 $R_{\rm int} = 0.021$ Siemens P4 diffractometer $\theta_{\rm max} = 25^{\circ}$ $2\theta/\omega$ scans $h = -1 \rightarrow 16$ Absorption correction: empirical (North et al., 1968) $k = -1 \rightarrow 14$ $T_{\min} = 0.954, \ T_{\max} = 0.968$ $l = -14 \rightarrow 12$ 2043 measured reflections 3 standard reflections 1642 independent reflections 1100 reflections with $I > 2\sigma(I)$

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0851P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.057$ + 1.2289P $wR(F^2) = 0.155$ where $P = (F_o^2 + 2F_c^2)/3$ S = 1.166 $(\Delta/\sigma)_{\rm max} = -0.006$ $\Delta \rho_{\rm max} = 0.28 \ {\rm e} \ {\rm \AA}^{-3}$ 1637 reflections $\Delta \rho_{\rm min} = -0.21 \text{ e} \text{ Å}^{-3}$ 133 parameters Extinction correction: SHELXTL H atoms treated by a mixture of independent and constrained (Siemens, 1995) refinement Extinction coefficient: 0.0022 (6)

every 97 reflections

intensity decay: 5.84%

Table 1

Selected geometric parameters (Å, °).

N1-C7	1.267 (2)	N3-N4	1.383 (2)
N1-N2	1.409 (2)	N4-C9	1.302 (2)
N2-C9	1.357 (2)	C1-O1	1.350 (2)
N2-C8	1.362 (2)	C4-C7	1.456 (3)
N3-C8	1.304 (3)		
C7-N1-N2	115.4 (2)	C8-N2-N1	132.3 (2)
C9-N2-C8	105.7 (2)	O1-C1-C6	123.0 (2)
C9-N2-N1	122.0 (2)	N1-C7-C4	122.5 (2)

Table 2Hydrogen-bonding geometry (Å, °).

$D-\mathrm{H}\cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdots A$
$O1-H11\cdots N4^i$	1.01	1.74	2.717 (2)	161
C9−H9···O1 ⁱⁱ	1.18	2.15	3.293 (3)	164
$O1W - H1W \cdot \cdot \cdot N3$	1.02	1.89	2.898 (2)	169
$C7 - H7 \cdots O1W^{iii}$	1.10	2.47	3.214 (2)	123

Symmetry codes: (i) $x - \frac{1}{2}, -\frac{1}{2} - y, \frac{1}{2} + z$; (ii) $\frac{1}{2} - x, y - \frac{1}{2}, -\frac{1}{2} - z$; (iii) $x, -y, \frac{1}{2} + z$.

Data collection: *XSCANS* (Siemens, 1994); cell refinement: *XSCANS*; data reduction: *SHELXTL* (Siemens, 1995); program(s) used to solve structure: *SHELXTL*; program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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References

- Colautti, A., Ferlauto, R. J., Maurich, V., De Nardo, M., Nisi, C., Rubessa, F. & Runti, C. (1971). *Chim. Ther.* **6**, 367–379.
- Gupta, A. K. & Bhargava, K. P. (1978). Pharmazie, 33, 430-431.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Kargin, Yu. M., Kitaeva, M. Yu., Latypova, V. Z., Vafina, A. A., Zaripova, R. M. & Il'yasov, A. V. (1988). *Izv. Akad. Nauk SSSR Ser. Khim.* 3, 607–611.
- Kitaev, Yu. P., Savin, V. I., Zverev, V. V. & Popova, G. V. (1971). *Khim. Geterotsikl. Soedin.* 7, 559–564.
- Mazza, M., Montanari, L. & Pavanetto, F. (1976). Farmaco Ed. Sci. 31, 334-344.
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351– 359.
- Shanmuga Sundara Raj, S., Fun, H.-K., Zhu, D.-R., Jian, F.-F., Zhang, K.-L. & You, X.-Z. (1999). Acta Cryst. C55, 1526–1528.
- Siemens (1994). XSCANS. Version 2.1. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Siemens (1995). SHELXTL. Version 5.0. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Wang, Z.-X., Jian, F.-F., Duan, C.-Y., Bai, Z.-P. & You, X.-Z. (1998). Acta Cryst. C54, 1927–1929.