

www.angewandte.de

Angewander GDCh Eine Zeitschrift der Gesellschaft Deutscher Chemiker

Akzeptierter Artikel

Titel: Ligand-Controlled Copper-Catalyzed Regiodivergent Carbonylative Synthesis of α-Amino Ketones and α-Boryl Amides from Imines and Alkyl lodides

Autoren: Xiao-Feng Wu and Fu-Peng Wu

Dieser Beitrag wurde nach Begutachtung und Überarbeitung sofort als "akzeptierter Artikel" (Accepted Article; AA) publiziert und kann unter Angabe der unten stehenden Digitalobjekt-Identifizierungsnummer (DOI) zitiert werden. Die deutsche Übersetzung wird gemeinsam mit der endgültigen englischen Fassung erscheinen. Die endgültige englische Fassung (Version of Record) wird ehestmöglich nach dem Redigieren und einem Korrekturgang als Early-View-Beitrag erscheinen und kann sich naturgemäß von der AA-Fassung unterscheiden. Leser sollten daher die endgültige Fassung, sobald sie veröffentlicht ist, verwenden. Für die AA-Fassung trägt der Autor die alleinige Verantwortung.

Zitierweise: Angew. Chem. Int. Ed. 10.1002/anie.202012251

Link zur VoR: https://doi.org/10.1002/anie.202012251

WILEY-VCH

WILEY-VCH

Ligand-Controlled Copper-Catalyzed Regiodivergent Carbonylative Synthesis of *α*-Amino Ketones and *α*-Boryl Amides from Imines and Alkyl Iodides

Fu-Peng Wu and Xiao-Feng Wu*

[*] F.-P. Wu, Prof. Dr. X.-F. Wu, Leibniz-Institut für Katalyse e. V. an der Universität Rostock, Albert-Einstein-Straße 29a, 18059 Rostock, Germany; Prof. Dr. X.-F. Wu, Dalian National Laboratory for Clean Energy, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, 116023, Dalian, Liaoning, China, E-mail: xiao-feng.wu@catalysis.de

Supporting information for this article is given via a link at the end of the document.

Abstract: Regioselective transformation is among the bng-standing challenges in organic synthesis. In this communication, a coppercatalyzed selectivity controlled regiodivergent borocarbonylation of imines with alkyl iodides has been developed. Various *α*-amino ketones and *α*-boryl amides were produced in moderate to good yields from the same substrates. The choosing of the appropriate ligand is the key for the regioselectivity control: *α*-amino ketones were produced selectively in good yields with (*p*-CF₃C₆H₄)₃P as the ligand; while the corresponding *α*-boryl amides were obtained in high regioselectivity with ^{Me}IMes as the ligand.

Over the past few decades, carbonylation reactions have experienced significant progresses with CO as an attractive C1 source for the preparation of various carbonyl-containing compounds and extending the carbon chain at the same time.^[1] Among the numerous transformations, transition-metal catalyzed borocarbonylation represents a novel strategy for converting CO/boron(ester) into complexed carbonyl-boryl-compounds. Some borocarbonylation reactions of alkynes and alkene have been achieved by using Cu^[2] or Cu/Pd^[3] catalyst systems. The key step in these reactions is the addition of LCu-Bpin to unsaturated bonds to generate an active alkenyl- or alky-CuL intermediate, which can subsequently react with CO and electrophiles to deliver the final products (Scheme 1a). Thus, as an inexpensive and abundant metal, copper is crucial and unique in borocarbonylation reactions.^[4]

On the other hand, achieving regioselective transformation of unsaturated chemical bonds is among the long-standing challenges in organic synthesis.^[5] Based on the continues efforts from organic chemists, great successes have been made during the past decades. In particular, transition-metal catalysts such as copper,^[6] palladium,^[7] and rhodium^[8] have been extensively studied in this topic. Furthermore, ligands design provides more opportunities for regioselective transformations, which can affect the metallic complex's structural and electronic properties by the electronic and steric properties of the ligands. Although the regioselective carbonylative reaction is even more challenge as CO coordinates to the metal and decrease its electron density, a variety of ligand-controlled regiodivergent carbonylation have been developed recently. For example, Liu et al. reported a regioselective alkoxycarbonylation of allenes in 2015,^[9] and Alper and co-workers develop an elegant palladium-catalyzed regioselective aminocarbonylation of styrenes.[10] Additionally, several other original regioselective transformations on carbonylation have been established in recent years as well.[11] Despite these achievements, regiodivergent carbonylative conversion of C=N double has rarely been reported.[12] This is due to the polarized imine (C=N) groups place a part of positive charge at the carbon atom, making the carbon positive (electrophilic)-attracted by negatively charged nucleophiles

(Scheme 1b).^[13] Therefore, inverting the inherent polarity of the imines is the key to reach regiodivergent transformation.^[14] Additionally, borocarbonylation reaction of C=N double bonds offers an ideal option for the synthesis of α -amino ketones^[15] or α -boryl amides^[16] if the regioselectivity of the C=N double bond could be altered, which represent important building blocks in organic synthesis as well as with potent biomedical applications. Herein, we developed a ligand-controlled copper-catalyzed highly regioselective boro carbonylation of imines with B₂pin₂ and alkyl iodides to give diverse α -amino ketones and α -boryl amides (Scheme 1c).







c) Ligand-controlled Cu-catalyzed borocarbonylation of imine (*this study*)



Scheme 1. Strategies for regiodivergent borocarbonylation of imines.

We commenced our studies with N,1-diphenylmethanimine (1a), 1-iodobutane (2a) and B₂pin₂. Ancillary ligands were thought to be the crucial factor for the reaction, thus we screened ligands under our conditions (For details of the optimization process, see supporting information). As shown in Figure 1, both bipyridine (L1) as nitrogen ligand and BuPAd₂ (L2) as bulky and electron-rich ligand failed to produce the desired product a-amino ketones 3a or a-boryl amides 4a. To our surprise, by using the classic ligand triphenylphosphine (L3), we been able to obtain 53% yield of 3a with high selectivity (>20:1), and the major byproduct was N-(1-phenylpentyl)aniline. Then this type of ligand with different electronic properties were examined, phosphine ligands with electron-withdrawing group (L4-L6) improved the reaction efficiency. On the other side, electron-donating ligands (L7, L8) decreased the reaction efficiency and steric hindrance (L8) reduced the reaction selectivity meanwhile. Bisphosphine ligands such as DCyPE (L9), DCyPP (L10), DCyPB (L11) and Cy-Xantphos (L12) delivered the wished 3a in low yields and with poor selectivity. In the absence of ligand, 3a could be formed in 61% yield and without 4a formation. Impressively, only α -boryl amides 4a (10%) yield, 3a:4a <1:20 selectivity) was obtained by switching to NHC

Angewandte Chemie

WILEY-VCH

Vanuscri

ligands IMes (L14). The starting material 1a was completely converted and the main byproduct was N-(phenyl(4,4,5,5tetramethyl-1.3.2-dioxaborolan-2-yl)methyl)aniline due to the complete consumption of 2a. Finally, the desire product 4a can be generated preferentially (62% yield, <20:1) by careful finetuning the loading of 2a and other reaction parameters. Among the other analyzed NHC ligands (L15-L18), ^{Me}IMes ligand (L15) with enhanced donating-electron property improved the yield of

4a to 71% with high selectivity (<20:1). These results imply that the regioselective addition of LCuBpin to C=N double bonds is mainly dictated by the electronic properties of the ligands applied. Specifically, phosphine ligands with electronwithdrawing group mainly produce α -amino ketones **3a**, while ligands with strong coordination property tend to give α -boryl amides 4a.



Figure 1. Cu-catalyzed borocarbonylation of imine: Influence of ligands. [a] Reaction conditions: 1a (0.2 mmol), 2a (2.4 equiv.), CuCl (10 mol%), ligand (L1 and L9-L13: 10 mol%; L2-L8: 20 mol%), B2pin2 (3.0 equiv.), NaO^tBu (3.0 equiv.), CO (10 bar), THF/toluene (4:1, 0.2 M), 80 °C, 16 h; in the case of carbene ligand, NHC-CuCl complex was used. [b] 2a (3 equiv.), NHC-CuCl complex (10 mol%), B2pin2 (1.5 equiv.), NaO^tBu (1.5 equiv.), toluene (0.2 M).

With the two sets of optimized reaction conditions in hand, we examined the scope of this regiodivergent copper-catalyzed carbonylation of imines with alkyl iodides for the synthesis of α amino ketones. As shown in the Table 1, alkyl iodides with different chain length (3a-3e) showed good reactivity, delivered the corresponding products in moderate yields. Substrates containing ether or thiophene produced the desired products in moderate yields as well. However, the reaction failed in the case of secondary iodoalkane (3i). Subsequently, we evaluated a series of imines. Imines derived from benzaldehyde and anilines bearing electron-withdrawing groups such as F, CI, and Br

showed excellent reactivity, afforded the corresponding α -amino ketones in moderate to excellent yields (3j, 3n, 3p and 3q). Amide, morpholine, methoxyl-containing imines were also compatible with the reaction conditions to produce the target products in slightly decreased yields (3k, 3l, 3m and 3o). The results reveal that electronic properties of imines are consistent with electronic effects from the ligand. Moreover, ortho-ether or thioether substituted imines were also tolerated well (3r and 3s). Finally, imines derived from anilines and electron-poor or electron-rich benzaldehyde were tested; the corresponding products were successfully prepared in moderated yields (3t-3cc).





[a] Reaction conditions: 1 (0.2 mmol), 2 (2.4 equiv.), CuCl (10 mol%), L5 (20 mol%), B₂pin₂ (3.0 equiv.), NaO^tBu (3.0 equiv.), CO (10 bar), THF/toluene (4:1, 0.2 M), stirred at 80 °C for 16 h, isolated yield.

Subsequently, a scope on *a*-boryl amides production was performed (Table 2). Similarly, alkyl iodides bearing difference chain length, ether or trifluoromethyl can be utilized without any problem (**4a-4h**). Aldimines spanning a range of electronic properties were also tested for this transformation. Polar functional group such as morpholine (**4i**), amide (**4j**) and ether (**4k**, **4n** and **4o**) could also be employed. Halides (**4r-4t**) and Heterocyclic imine (**4u**) were also suitable reactants here. The absolute configuration of compound **4r**, isolated as a colorless crystalline solid, was clearly confirmed by X-ray crystallography. The coordination between the boron atom and oxygen of amide is observed and consistent with ¹¹B NMR.^[17] Notably, the obtained products **4** are ready for further synthetic transformations.^[17]

Concerning the reaction pathway, based on the experimental results and related literatures,^{1-4,12-14,18} a possible reaction pathway is proposed (Figure 2). Initially, LCu-Bpin I, generated from CuCl, B_2pin_2 and NaO'Bu, inserts into the C=N

bond of imine to give α-boryl amido-copper complex II. In the case of presenting electron-deficient ligand, the N-Cu bond is weaker and the copper specie II trend to occur intramolecular 1,2-rearrangement to afford α-amino alkylcopper complex III (Figure 2, left catalytic pathway).^[18] Afterwards, alkylcopper III reacts with alkyl iodide to generate copper complex IVa. Then, CO coordinates with complex Iva to produce alkylacyl-copper Va after an insertion step. After reductive elimination and work up with MeOH, the desired product α -amino ketone 3 was formed and meanwhile regenerate LCuX for the next catalytic cycle. In the catalytic cycle on the right, under the assistance of ^{Me}IMes ligand, the cross-coupling reaction between electron-rich intermediate copper complex II and alkyl iodide quickly gives IVb. After insertion and reductive elimination steps, the corresponding a-boryl amide was produced. Here it is worth to mention that radical intermediates were involved in the reactions between copper complexes (II and III) and alkyl iodide. We also can not exclude the possibility that intermediate III was formed directly from LCuBpin I and imine.





Table 2: Reaction scopes: Testing of alkylio dides and imines for α-boryl amides production.^[a]

[a] Reaction conditions: 1 (0.2 mmol), 2 (3.0 equiv.), ^{Me}IMes•CuCl (10 mol%), B_2pin_2 (1.5 equiv.), NaO'Bu (1.5 equiv.), CO (10 bar), toluene (0.2 M), stirred at 80 °C for 16 h, isolated yield. [b] Only one of the two molecules of the asymmetric unit is shown. Hydrogen atoms are omitted for clarity. Displacement ellipsoids correspond to 30% probability.



Angewandte Chemie

In summary, a novel ligand-controlled copper-catalyzed boro carbonylation for the selective synthesis of α -amino ketones and α -boryl amides from imines and alkyl iodides has been developed. In this catalyst system, the choosing of the appropriate ligand is the key for the regioselectivity control: α -amino ketones were produced selectively in good yields with (p-CF₃C₆H₄)₃P as the ligand; while the corresponding α -boryl amides were obtained in high regioselectivity with ^{Me}IMes as the ligand.

Acknowledgements

We thank the China Scholarship Council for a Ph.D. Scholarship. We appreciate the analytical department of the Leibniz-Institute for Catalysis at the University of Rostock for their assistance. We also thank Dr. Anke Spannenberg (LIKAT) for the X-ray crystal structure analysis of compounds **4r**.

Keywords: regioselective • carbonylation • α -amino ketones • α -boryl amides • io doalkane

- a) J.-B. Peng, F.-P. Wu, X.-F. Wu, *Chem. Rev.* 2019, *11*9, 2090-2127;
 b) Z. Yin, J.-X. Xu, X.-F. Wu, *ACS Catal.* 2020, *10*, 6510-6531.
- [2] a) L. J. Cheng, N. P. Mankad, Angew. Chem. Int. Ed. 2018, 57, 10328-10332; Angew. Chem. 2018, 130, 10485-10489; b) F. -P. Wu, Y. Yuan, C. Schunemann, P. C. J. Kamer, X.-F. Wu, Angew. Chem. Int. Ed 2020. 59, 10451-10455; Angew. Chem. 2020, 132, 10537-10541; c) F.-P. Wu, X. Luo, U. Radius, T. B. Marder, X.-F. Wu, J. Am. Chem. Soc. 2020. 142, 14074-14079.
- [3] Y. Yuan, F.-P. Wu, J.-X. Xu, X.-F. Wu, Angew. Chem. Int. Ed. 2020, 59, 17055-17061.
- [4] a) D. Hemming, R. Fritzemeier, S. A. Westcott, W. L. Santos, P. G. Steel, *Chem. Soc. Rev.* 2018, *47*, 7477-7494; b) J.-B. Peng, H.-Q. Geng, X.-F. Wu, *Chem* 2019, 5, 526-552; .c) J. Cheng, N. P. Markad, *Chem. Soc. Rev.* 2020, doi:10.1039/d0cs00316f
- [5] a) M. Beller, J. Seayad, A. Tillack, H. Jiao, *Angew. Chem. Int. Ed.* 2004, 43, 3368-3398; b) C. Najera, I. P. Beletskaya, M. Yus, *Chem. Soc. Rev.* 2019, 48, 4515-4618.
- [6] a) Y. Ye, S. T. Kim, J. Jeong, M. H. Baik, S. L. Buchwald, J. Am. Chem. Soc. 2019, 141, 3901-3909: b) T. Jia, Q. He, R. E. Ruscoe, Al. P. Pulis, D. J. Procter, Angew. Chem. Int. Ed 2018, 57, 11305-11309; Angew. Chem. 2018, 130, 11475-11479; c) W. Su, T.-J. Gong, Q. Zhang, Q. Zhang, B. Xiao, Y. Fu, ACS Catal. 2016, 6, 6417-6421; d). J. Feng, Y. Xu, M. Oestreich, Chem. Sci 2019, 10, 9679-9683; e). Sakae, K. Hirano, M. Miura, J. Am. Chem. Soc. 2015, 137, 6460-6463; f) R. Y. Liu, Y. Yang, S. L. Buchwald, Angew. Chem. Int. Ed 2016, 55, 14077-14080; Angew. Chem. 2016, 128, 14283-14286.
- a) V. Debrauwer, A. Turlik, L. Rummler, A. Prescimone, N. Blanchard, K. N. Houk, V. Bizet, *J. Am. Chem. Soc.* 2020, *142*, 11153-11164; b) Z. Huang, Y. Cheng, X. Chen, H. F. Wang, C. X. Du, Y. Li, Chem. Commun. 2018, *54*, 3967-3970.
- [8] a) A. J. Bochat, V. M. Shoba, J. M. Takacs, Angew. Chem. Int. Ed. 2019, 58, 9434-9438; Angew. Chem. 2019, 131, 9534-9538. b) P. A. Evans, J. R. Sawyer, P. A. Inglesby, Angew. Chem. Int. Ed. 2010, 49, 5746-5749; Angew. Chem. 2010, 122, 5882-5885; c) M. D. Wodrich, B. Ye, J. F. Gonthier, C. Corminboeuf, N. Cramer, Chem. Eur. J. 2014, 20, 15409-15418.
- [9] J. Liu, Q. Liu, R. Franke, R. Jackstell, M. Beller, J. Am. Chem. Soc. 2015, 137, 8556-8563.
- [10] T. Xu, F. Sha, H. Alper, J. Am. Chem. Soc. 2016, 138, 6629-6635.
- [11] a) D. Ding, G. Zhu, X. Jiang, Angew. Chem. Int. Ed 2018, 57, 9028-9032; Angew. Chem. 2018, 130, 9166-9170; b) F. Sha, H. Alper, ACS

Catal. **2017**, 7, 2220-2229; c) W. Liu, W. Ren, J. Li, Y. Shi, W. Chang, Y. Shi, *Org. Lett.* **2017**, *19*, 1748-1751; d) T. Xu, H. Alper, *J. Am. Chem. Soc.* **2014**, *136*, 16970-16973; e) Ren, W. Chang, Y. Wang, J. Li, Y. Shi, *Org. Lett.* **2015**, *17*, 3544-3547; f) J. Liu, Z. Han, X. Wang, Z. Wang, K. Ding, J. *Am. Chem. Soc.* **2015**, *137*, 15346-15349; g) Ren, W. Chang, J. Dai, Y. Shi, J. Li, Y. Shi, *J. Am. Chem. Soc.* **2016**, *138*, 14864-14867.

- [12] a) Y. Hoshimoto, K. Ashida, Y. Sasaoka, R. Kumar, K. Kamikawa, X. Verdaguer, A. Riera, M. Ohashi, S. Ogoshi, *Angew. Chem. Int. Ed* **2017**, *56*, 8206-8210; *Angew. Chem.* **2017**, *129*, 8318-8322; b) S. Oda, B. Sam, M. J. Krische, *Angew. Chem. Int. Ed* **2015**, *54*, 8525-8528; *Angew. Chem* **2015**, *127*, 8645-8648; c) X. Zhou, Y. Ding, H. Huang, *Chem Asian J.* **2020**, *15*, 1678-1682; d) K. Ashida, Y. Hoshimoto, N. Tohnai, D. E. Scott, M. Ohashi, H. Imaizumi, Y. Tsuchiya, S. Ogoshi, *J. Am. Chem Soc.* **2020**, *142*, 1594-1602; e) Y. Hoshimoto, T. Ohata, Y. Sasaoka, M. Ohashi, S. Ogoshi, *J. Am. Chem. Soc.* **2014**, *136*, 15877-15880.
- [13] a) F. J. T. Talbot, Q. Dherbæssy, S. Manna, C. Shi, S. Zhang, G. P. Howell, G. J. P. Perry, D. J. Procter, Angew. Chem Int. Ed 2020 doi:10.1002/anie.202007251; b) K. Morisaki, H. Morimoto, T. Ohshima, ACS Catal 2020, 10, 6924-6951; c) J. Tjutrins, B. A. Amdtsen, Chem. Sci. 2017, 8, 1002-1007; d) Z. Li, L. Zhang, M. Nishiura, G. Luo, Y. Luo, Z. Hou, J. Am. Chem. Soc. 2020, 142, 1966-1974; e) S. Manna, Q. Dherbæssy, G. J. P. Perry, D. J. Procter, Angew. Chem Int. Ed 2020, 59, 4879-4882; Angew. Chem. 2020, 132, 4909-4912; f) T. Itoh, Y. Kanzaki, Y. Shimizu, M. Kanai, Angew. Chem. Int. Ed 2018, 57, 8265-8269; Angew. Chem 2018, 130, 8397-8401; g) Q. Xia, H. R. Chang, J. Li, J. Y. Wang, Y. Q. Peng, G. H. Song, J. Org. Chem 2020, 85, 2716-2724.
- [14] a) A. Patra, S. Mukherjee, T. K. Das, S. Jain, R. G. Gonnade, A. T. Biju, Angew. Chem. Int. Ed 2017, 56, 2730-2734; Angew. Chem. 2017, 129, 2774-2778; b) Y. Wu, L. Hu, Z. Li, Nature, 2015, 523, 445–450.
- [15] F. I. Carroll, B. E. Blough, P. Abraham, A. C. Mills, J. A. Holleman, S. A. Wockenhauer, A. M. Decker, A. Landavazo, K. T. McElroy, H. A. Navarro, M. B. Gatch, M. J. Forster, J. *Med. Chem* **2009**, *52*, 6768-6781.
- [16] E. Caselli, C. Romagnoli, R. Vahabi, M. A. Taracila, R. A. Bonomo, F. Prati, J. Med. Chem. 2015, 58, 5445-5458.
- [17] a) L Chen, X. Zou, H. Zhao, S. Xu, Org. Lett. 2017, 19, 3676-3679. b)
 N. Hu, G. Zhao, Y. Zhang, X. Liu, G. Li, W. Tang, J. Am Chem Soc. 2015, 137, 6746-6749.
- [18] Z. Li, L. Zhang, M. Nishiura, Z. Hou, ACSCatal. 2019, 9, 4388-4393.

WILEY-VCH



A novel ligand-controlled copper-catalyzed borocarbonylation for the selective synthesis of α -amino ketones and α -boryl amides from imines and alkyl iodides has been developed. In this catalyst system, the choosing of the appropriate ligand is the key for the regioselectivity control: α -amino ketones were produced selectively in good yields with $(p-CF_3C_6H_4)_3P$ as the ligand; while the corresponding α -boryl amides were obtained in high regioselectivity with ^{Me}IMes as the ligand.

This article is protected by copyright. All rights reserved.