

4-Alkynylphenyl Imidazolylpropyl Ethers as Selective Histamine H₃-Receptor Antagonists with High Oral Central Nervous System Activity

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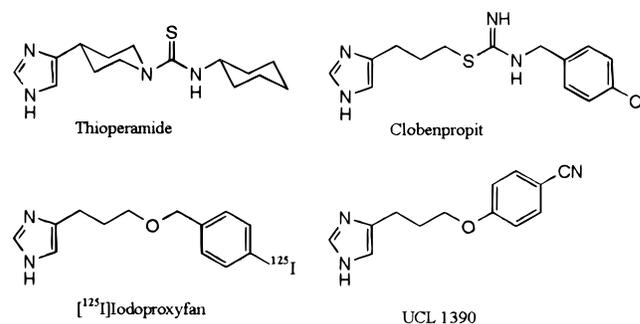
In search for potent and therapeutically useful H₃-receptor antagonists, we prepared novel 4-alkynylphenyl ether derivatives of 3-(1*H*-imidazol-4-yl)propanol in a convenient synthetic route. All compounds were tested for in vitro and in vivo H₃-receptor antagonist activity as well as for H₃-receptor selectivity versus H₁- and H₂-receptors. The presented 4-alkynylphenyl ethers are highly potent and selective H₃ antagonists showing oral activity and improved brain penetration. Particularly 4-ethynylphenyl 3-(1*H*-imidazol-4-yl)propyl ether (**14a**) displays striking in vitro and in vivo activity with a $-\log K_i$ value of 8.6 and an ED₅₀ value of 0.12 mg/kg. At present **14a** is the most potent H₃-receptor antagonist in vivo and may therefore be a potential drug for the therapy of H₃-receptor-dependent diseases of the central nervous system (CNS).

Introduction

The histamine H₃-receptor was demonstrated to be located presynaptically on histaminergic neurons in the central nervous system (CNS) where it regulates histamine synthesis and release as an inhibitory autoreceptor.^{1,2} Inhibition of this negative feedback mechanism by H₃-receptor antagonists thus increases concentration of the neurotransmitter histamine released.³ Moreover, H₃-receptors function as heteroreceptors on nonhistaminergic neurons in the brain and the periphery inhibiting the release of neuropeptides⁴ and several other neurotransmitters.⁵ In radiolabeling studies the highest density of H₃-receptors was found in distinct areas of the CNS,⁶ and it is thus suggested that the potential therapeutic role of H₃-receptor antagonists may be the treatment of various neurological and psychiatric diseases, e.g., epilepsy, narcolepsy, schizophrenia, or dementia.⁷

The first selective H₃-receptor antagonist, thioperamide,⁸ which has become a useful tool for the pharmacological investigation of H₃-receptors, does not appear to be suitable for clinical studies because of hepatotoxicity that is probably related to its thiourea group (Chart 1). Similar drawbacks are likely to occur with the isothiourea derivative clobenpropit⁹ which is more active than thioperamide in vitro but less effective in vivo presumably as a consequence of limited bioavailability (cf. Table 1). To circumvent the toxicological and bioavailability problems at H₃-receptors, a number of histamine H₃-receptor antagonists that are devoid of sulfur-containing functionalities were developed, e.g., amides, carbamates, esters, ethers, guanidines, heteroaryl compounds,⁷ and less polar alkynes derived from the marine natural product verongamine.¹⁰

Chart 1. Histamine H₃-Receptor Antagonists



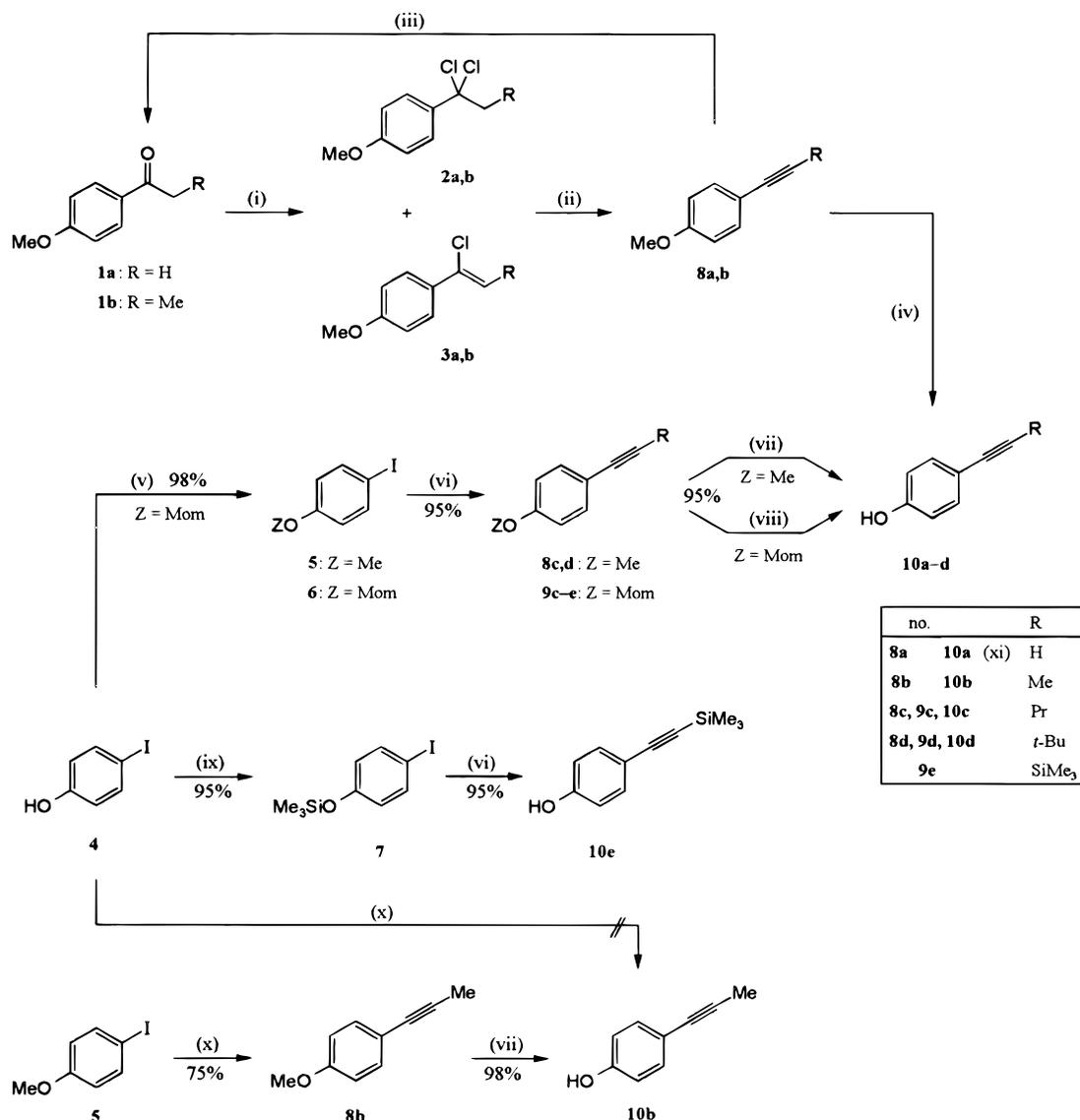
The aim of the present study was to design potent histamine H₃-receptor antagonists that exhibit improved oral bioavailability and brain penetration and lack potentially toxic sulfur-containing functionalities. The current research focused on imidazolylpropyl ethers because their potency was clearly pointed out in previous studies. In particular, the para-iodinated benzylic ether iodoproxyfan was selected for radiolabeling from a series of recently described aliphatic imidazolylalkyl ethers due to its in vitro potency and selectivity (Chart 1).^{6,11} In an attempt to explore structure–activity relationships of H₃-receptor ligands, Ganellin et al. reported aromatic imidazolylalkyl ethers to be potent H₃-receptor antagonists that provide high in vitro activity at H₃-receptors.¹² Thus, para-substituted phenyl ether derivatives such as 4-cyanophenyl 3-(1*H*-imidazol-4-yl)propyl ether (UCL 1390; Chart 1) served as lead compounds in the search for novel antagonists. To obtain selective, orally active, and centrally penetrating H₃-receptor antagonists, we introduced alkynyl moieties in the para-position of phenyl ether derivatives of 3-(1*H*-imidazol-4-yl)propanol. The synthetical approach and pharmacological evaluation of those 4-alkynylphenyl ethers are presented, all of which were screened for their in vitro and in vivo H₃-receptor

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Scheme 1. Synthetic Pathways To Obtain 4-Alkynylphenols **10a–e**^a

^a (i) PCl₅, benzene, 24 h at ambient temperature; (ii) *t*-BuOK, 18-crown-6, C₆H₁₂, 0 °C → reflux, 3 h; (iii) Me₃SiBr or Me₃SiI, CH₂Cl₂, 48 h at ambient temperature → 50 °C, 30 min, MeOH/H₂O; (iv) EtSNa, DMF, reflux, 1 h; (v) MomCl, Et₃N, Et₂O, 48 h at ambient temperature; (vi) HC≡CR, Pd^{II}Cl₂(PPh₃)₂, CuI, Et₃N, 1 h at ambient temperature; (vii) EtSNa, DMF, reflux, 1 h; (viii) HCl, MeOH, THF, 24 h at ambient temperature; (ix) Me₃SiCl, Et₃N, Et₂O, reflux, 18 h; (x) Me₃SiC≡CMe, (η³-C₃H₅Pd^{II}Cl)₂, TBAF, THF, ambient temperature → 50 °C, 15 min; (xi) identified by TLC but not isolated due to instability.

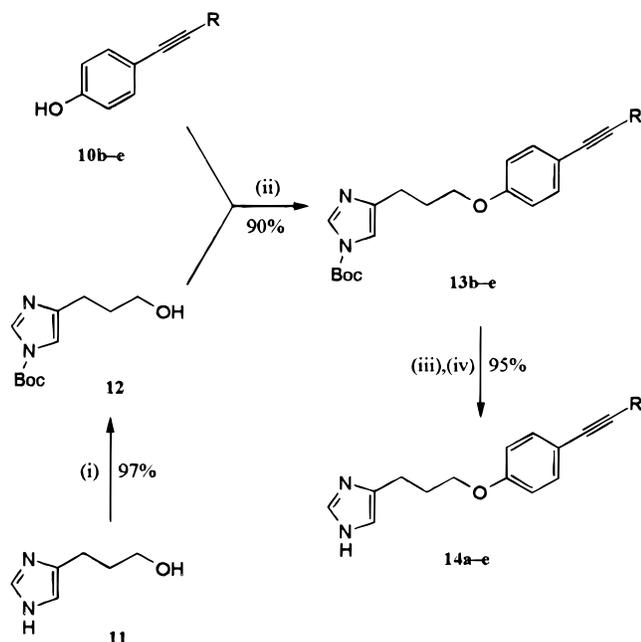
antagonist activity as well as for their H₃-receptor selectivity versus H₁- and H₂-receptors.

Chemistry

The classical route to prepare aromatic alkynes **8a,b** proceeds from aromatic ketones **1a,b** which are converted to *gem*-dichlorides **2a,b** by treatment with PCl₅ (Scheme 1). β-Elimination of HCl may follow, and vinylic chlorides **3a,b** are frequent side products. The C≡C triple bond formation is accomplished by successive dehydrohalogenation in the presence of potassium *tert*-butanolate. Although this method was optimized by means of phase-transfer catalysis as recently reported,¹³ the alkynes **8a,b** were obtained with overall yields of only about 20%. Cleavage of the phenol protecting methyl ether was finally achieved by treatment with sodium ethanethiolate in refluxing dimethylformamide¹⁴ because the procedure with trimethylsilyl bromide or iodide¹⁵ led to hydration of the triple bond

giving the corresponding ketones **1a,b**. In contrast to 4-propynylphenol (**10b**) the ethynyl analogue **10a** was found to be too labile to be isolated from solution due to polymerization, although its formation was clearly proved by means of TLC.

An efficient and mild alternative route is provided by arylation of alkynes, known as the Heck reaction,¹⁶ which was achieved by treatment with an "arylpalladium" reagent (Scheme 1). The latter was generated in situ by treatment of an aryl iodide (**5**, **6**) with a palladium(II)–triphenylphosphine complex in the presence of triethylamine and cuprous(I) iodide as cocatalyst.¹⁷ The reaction was facilitated by cuprous(I) iodide so that it was completed after 30 min at room temperature. Subsequent cleavage of the protecting groups, either methyl or methoxymethyl (Mom),¹⁸ yielded the stable internal alkynes **10c,d**. As the trimethylsilyl protecting group of **9e** was not stable under the hydrolytic conditions used,¹⁹ the labile terminal 4-alkynylphe-

Scheme 2. Synthesis of 4-Alkynylphenyl Ethers **14**^a

^a (i) Boc_2O , MeCN, Et_3N , dioxane, H_2O , DMAP, 2 h at ambient temperature; (ii) PPh_3 , DEAD, THF, 72 h at ambient temperature; (iii) NH_3 , MeOH, 6 h at ambient temperature; (iv) H_2NNH_2 , MeOH, 30 min at ambient temperature.

nol **10a** was formed instead of **10e** but could not be isolated as mentioned above.

Synthesis of the desired trimethylsilyl-protected alkyne **10e** was accomplished by Heck reaction using trimethylsilyl-protected 4-iodophenol (**7**) as substrate (Scheme 1). During the reaction the trimethylsilyl protecting group of the phenol was cleaved, hence giving **10e** almost quantitatively in a tandem reaction.

Although gaseous propyne could have been used effectively under these conditions, the Heck reaction did not seem to be attractive for the synthesis of **10b** due to its problematic utilization. However, the fluoride-mediated cross-coupling reaction of trimethylsilylpropyne with 4-iodoanisole yielding **8b** is an efficient alternative to dehydrohalogenation (Scheme 1).²⁰ Although this cross-coupling reaction was reported to be highly chemoselective in the presence of an unprotected hydroxy group, we found protection of the phenol to be essential because coupling of 4-iodophenol (**4**) failed to give **10b**. Compound **8b** was deprotected by sodium ethanethiolate in the final stage to give **10b** with an overall yield of 74%.

To obtain the desired 4-alkynylphenyl ethers **14**, the phenol precursors **10** were coupled to *tert*-butoxycarbonyl (Boc)-protected 3-(1*H*-imidazol-4-yl)propanol (**12**)²¹ according to the Mitsunobu protocol (Scheme 2).²² The Boc protection of the imidazole was readily accomplished at ambient temperature by means of di-*tert*-butyl dicarbonate in the presence of 4-(*N,N*-dimethylamino)pyridine. The Boc residue was primarily chosen as protecting group because the commonly used trityl group could not be cleaved without affecting the $\text{C}\equiv\text{C}$ triple bond, neither hydrolytically under various acidic conditions nor hydrogenolytically in the presence of palladium on charcoal. However, cleavage of the Boc group by methanolic ammonia or hydrazine at ambient temperature proved to be ideal in this reaction sequence.

Even desilylation of the alkyne **13e**, which was expected to occur quantitatively under these conditions, came about just partially so that not only the unprotected target compound 4-ethynylphenyl 3-(1*H*-imidazol-4-yl)propyl ether (**14a**) was obtained (60% yield) but also its trimethylsilyl-protected analogue **14e** (30% yield), separation of which was performed by means of column chromatography.

Pharmacological Results and Discussion

In Vitro Testing. The presented compounds were tested for their antagonist potency at H_3 -receptors in an assay with K^+ -evoked [^3H]histamine release from synaptosomes of rat cerebral cortex.²³

The novel 4-alkynylphenyl ethers **14** show moderate to high H_3 -receptor antagonist potency, which was found to be dependent on the residue of the aromatic alkyne (Table 1). The terminal alkyne **14a** is the most potent compound of this series, whereas substitution of the aromatic alkyne decreased antagonist activity. It is assumed that the affinity to H_3 -receptors is reduced most likely due to steric effects as the methyl group in **14b** led only to a slight decrease in potency as compared to the bulkier propyl (**14c**), *tert*-butyl (**14d**), and trimethylsilyl (**14e**) groups, respectively. The ethynyl derivative **14a**, however, which has a $-\log K_i$ value of 8.6 appears to be even more potent than the standard H_3 -receptor antagonist thioperamide ($-\log K_i = 8.4$).⁸ The pronounced in vitro potency of **14a** was further substantiated in a functional H_3 -receptor assay on the guinea pig ileum,⁶ in which it acts as a competitive antagonist displaying a pA_2 value of 8.1 ± 0.15 ($\bar{x} \pm \text{SEM}$, $n = 16$).

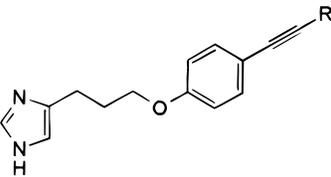
In Vivo Results on Mice. The in vivo antagonist activity at H_3 -receptors was screened in an assay measuring the effect on brain histamine turnover after oral application to mice.²³

Corresponding to the effects observed in vitro, the 4-ethynylphenyl ether **14a** showed improved potency as compared to the internal alkyne derivatives **14b–e**, which are nevertheless highly effective H_3 -receptor antagonists exhibiting about the same potency as thioperamide (Table 1). However, the striking CNS activity following oral administration of **14a** is clearly pointed out by these results as it is approximately 6.5 times higher on a molar basis than thioperamide. It is even more potent than any other H_3 -receptor antagonist published so far and might thus become a potential therapeutic agent in neurology or psychiatry.

Screening at Other Histamine Receptor Subtypes. In addition, compounds **14a–e** were tested for their H_1 -receptor activity on the guinea pig ileum as well as for their H_2 -receptor activity on the guinea pig atrium.²⁴ As a result, all compounds were observed to be moderately or highly selective for the H_3 -receptor versus H_1 - and H_2 -receptors, respectively (Table 1). Particularly the most potent compound (**14a**) exhibits pronounced selectivity for the H_3 -receptor at a ratio of approximately 6000/1 vs H_1 - and 1250/1 vs H_2 -receptors.

Conclusions

The presented 4-alkynylphenyl ether derivatives of 3-(1*H*-imidazol-4-yl)propanol were prepared in a mild and convenient synthetic route providing almost quan-

Table 1. Chemical Data and Pharmacological Results of Screening for Histamine H₃-Receptor Antagonist Activity and Selectivity


compd	R	yield (%)	mp ^c (°C)	formula	M _r	in vivo		in vitro		
						ED ₅₀ ^a (mg kg ⁻¹) $\bar{x} \pm \text{SEM}$	K _i ^b (nM) $\bar{x} \pm \text{SEM}$	-log K		
								H ₃ ^b	H ₂ ^d	H ₁ ^e
14a	H	60 ^f	150	C ₁₄ H ₁₄ N ₂ O·C ₄ H ₄ O ₄ ·0.25H ₂ O	346.9	0.12 ± 0.07	2.3 ± 0.8	8.6	<4.8	<5.5
14b	Me	87	137	C ₁₅ H ₁₆ N ₂ O·C ₄ H ₄ O ₄ ·0.5H ₂ O	365.4	1.3 ± 0.4	7.5 ± 1.2	8.1	<4.7	<5.0
14c	Pr	85	126	C ₁₇ H ₂₀ N ₂ O·C ₄ H ₄ O ₄ ·0.25H ₂ O	388.9	2.2 ± 0.9	140 ± 58	6.9	<4.7	<5.9
14d	<i>t</i> -Bu	95	132	C ₁₈ H ₂₂ N ₂ O·C ₄ H ₄ O ₄ ·0.5H ₂ O	407.5	2.2 ± 0.3	123 ± 17	6.9	<4.6	<5.6
14e	SiMe ₃	30 ^f	126	C ₁₇ H ₂₂ N ₂ OSi·C ₄ H ₄ O ₄ ·0.75H ₂ O	428.1	1.1 ± 0.3	36 ± 15	7.4	<4.6	<5.0
thioperamide						1.0 ± 0.5 ^g	4 ± 1 ^h			
clobenpropit						26 ± 7 ⁱ	0.6 ± 0.1 ^j			

^a In vivo screening on central H₃-receptor activity after po application to mice.²³ ED₅₀ values expressed in mg of base kg⁻¹. ^b Functional H₃-receptor test in vitro on synaptosomes of rat cerebral cortex.²³ ^c Crystallization solvent: Et₂O/EtOH. ^d Functional H₂-receptor test on guinea pig atrium.²⁴ ^e Functional H₁-receptor test on guinea pig ileum.²⁴ ^f Compounds **14a,e** obtained from a single experiment. ^g Reference 12. ^h Reference 8. ⁱ Reference 11. ^j Reference 6.

titative yields in each step, except for compounds **14a,e** which were obtained together from a single experiment. In vitro studies proved their H₃-receptor antagonist potency and selectivity for H₃-receptors versus H₁- and H₂-receptors. Furthermore, the new compounds display high oral CNS activity with particularly the terminal alkyne being 6.5 times more active on a molar basis than thioperamide and more effective than any other H₃-receptor antagonist published to date. Although increasing substitution of the alkyne reduced both in vitro and in vivo activity, the internal alkynes proved to be almost equipotent with thioperamide. These data clearly indicate the efficacy of the structural modifications applied. Furthermore, the (trimethylsilyl)ethynyl derivative **14e** is, to our knowledge, the first compound containing a silyl moiety which shows H₃-receptor antagonist activity.

Experimental Section

Chemistry. General Procedures. Melting points were determined on an Electrothermal IA 9000 digital melting point apparatus and are uncorrected. ¹H NMR spectra were recorded on a Bruker AC 300 (300 MHz) spectrometer. Chemical shifts are expressed in ppm downfield from internal Me₄Si as reference. ¹H NMR data are reported in the order: multiplicity (br, broad; s, singlet; d, doublet; t, triplet; m, multiplet; *, exchangeable by D₂O), number of protons, and approximate coupling constants in hertz. A Finnigan MATCH/A (EI; 170 °C, 70 eV) or a Finnigan MAT CH5DF (FAB⁺; Xe, Me₂SO/glycerol) mass spectrometer was used. IR spectra were obtained on a Perkin-Elmer spectrophotometer 297. Elemental analyses (C, H, N) were measured on a Perkin-Elmer 240 C instrument and were within ±0.4% of the theoretical values unless otherwise indicated. Analytical TLC was performed on silica gel F₂₅₄ plates (Merck). Preparative column chromatography was carried out using silica gel, 63–200 μm (Merck).

1-Ethynyl-4-methoxybenzene (8a).²⁵ A solution of 4-methoxyphenylethanone (**1a**; 50 mmol, 7.51 g) in 10 mL of benzene was added to a suspension of PCl₅ (50 mmol, 10.43 g) in 100 mL of benzene. The mixture was stirred for 24 h at ambient temperature, hydrolyzed on ice, and extracted with light petroleum (bp 50–70 °C). The organic layer was washed (H₂O, NaHCO₃, H₂O), dried (MgSO₄), and evaporated to dryness. The resulting oil (**2a**; 44 mmol, 9.0 g) was dissolved in 40 mL of cyclohexane and cooled in an ice bath. Then *t*-BuOK (88 mmol, 9.87 g) and 18-crown-6 (0.09 mmol, 23 mg) were added at 0–5

°C followed by an additional 3 h under reflux. After the mixture was hydrolyzed on ice, the organic layer was separated, washed (H₂O), dried (MgSO₄), and evaporated. The resulting product was subjected to column chromatography [eluent: CH₂Cl₂ (25%), light petroleum (75%)] to afford the title compound as an oil which was crystallized from light petroleum (yield, 1.5 g, 25%): mp 29 °C; ¹H NMR (CDCl₃) δ 7.42 (d, *J* = 8.7 Hz, 2H, Ph-2-H, Ph-6-H), 6.83 (d, *J* = 8.7 Hz, 2H, Ph-3-H, Ph-5-H), 3.81 (s, 1H, Me), 2.99 (s, 1H, C≡CH); MS *m/z* 132 (M⁺, 100), 117 (27), 89 (26); IR (KBr) $\tilde{\nu}$ 3287 (s, C≡C-H), 2959 (m, C-H), 2838 (m, C-H), 2105 (m, C≡C), 1605 (s, C=C), 1570 (w, C=C), 1507 (s, C=C). Anal. (C₉H₈O·0.2H₂O) C, H, N.

4-Methoxy-1-propynylbenzene (8b).²⁶ **Method A:** Prepared according to **8a** (yield, 20%). **Method B:** To a solution of 4-iodoanisole (**5**; 10 mmol, 2.34 g) in 40 mL of THF were subsequently added (trimethylsilyl)propyne (10 mmol, 1.18 g), TBAF (10 mmol, 2.61 g), and (η^3 -C₃H₅PdCl)₂ (0.25 mmol, 0.09 g) followed by 15 min of stirring at 50 °C under argon atmosphere. The cooled mixture was filtered and evaporated under reduced pressure. The resulting product was purified as described for **8a** (yield, 1.1 g, 75%): mp 22 °C; ¹H NMR (CDCl₃) δ 7.32 (d, *J* = 8.8 Hz, 2H, Ph-2-H, Ph-6-H), 6.81 (d, *J* = 8.8 Hz, 2H, Ph-3-H, Ph-5-H), 3.79 (s, 3H, O-Me), 2.03 (s, 3H, C≡C-Me); MS *m/z* 146 (M⁺, 43), 131 (18), 111 (100), 90 (22); IR (KBr) $\tilde{\nu}$ 2955 (m, C-H), 2850 (m, C-H), 2043 (w, C≡C), 1607 (s, C=C), 1567 (m, C=C), 1510 (vs, C=C). Anal. (C₁₀H₁₀O·0.25H₂O) C, H, N.

4-Methoxy-1-pentynylbenzene (8c). To a solution of 4-iodoanisole (**5**; 20 mmol, 4.68 g) in 50 mL of Et₃N were subsequently added (PPh₃)₂PdCl₂ (0.2 mmol, 0.14 g), CuI (0.2 mmol, 0.04 g), and pentyne (40 mmol, 2.72 g) followed by stirring for 1 h at ambient temperature under Ar atmosphere. The mixture was then filtered, concentrated under reduced pressure, and subjected to column chromatography (eluent: light petroleum). The product was obtained as a light-yellow oil (yield, 3.4 g, 98%): ¹H NMR (CDCl₃) δ 7.32 (d, *J* = 8.8 Hz, 2H, Ph-2-H, Ph-6-H), 6.81 (d, *J* = 8.8 Hz, 2H, Ph-3-H, Ph-5-H), 3.79 (s, 3H, O-Me), 2.36 (t, *J* = 7.0 Hz, 2H, C≡C-CH₂), 1.63 (m, 2H, CH₂-Me), 1.04 (t, *J* = 7.4 Hz, 3H, CH₂-Me); MS *m/z* 174 (M⁺, 85), 145 (100), 102 (13). Anal. (C₁₂H₁₄O) C, H, N.

1-(3,3-Dimethylbutynyl)-4-methoxybenzene (8d): prepared according to **8c**; crystallized from light petroleum (yield, 93%); mp 39 °C; ¹H NMR (CDCl₃) δ 7.32 (d, *J* = 8.8 Hz, 2H, Ph-2-H, Ph-6-H), 6.79 (d, *J* = 8.8 Hz, 2H, Ph-3-H, Ph-5-H), 3.78 (s, 3H, O-Me), 1.30 (s, 9H, *t*-Bu); MS *m/z* 188 (M⁺, 81), 175 (100), 160 (21), 133 (11), 115 (15). Anal. (C₁₃H₁₆O) C, H, N.

4-(Methoxymethoxy)-1-pentynylbenzene (9c): prepared according to **8c** (yield, 98%); oil; $^1\text{H NMR}$ (CDCl_3) δ 7.32 (d, J = 8.7 Hz, 2H, Ph-2-H, Ph-6-H), 6.94 (d, J = 8.7 Hz, 2H, Ph-3-H, Ph-5-H), 5.16 (s, 2H, O-CH₂), 3.47 (s, 3H, O-Me), 2.36 (t, J = 7.0 Hz, 2H, C \equiv C-CH₂), 1.61 (m, 2H, CH₂-Me), 1.04 (t, J = 7.4 Hz, 3H, CH₂-Me); MS m/z 204 (M^+ , 43), 145 (11), 45 (100). Anal. ($\text{C}_{13}\text{H}_{16}\text{O}_2$) C, H, N.

1-(3,3-Dimethylbutynyl)-4-(methoxymethoxy)benzene (9d): prepared according to **8c** (yield, 98%); oil; $^1\text{H NMR}$ (CDCl_3) δ 7.31 (d, J = 8.8 Hz, 2H, Ph-2-H, Ph-6-H), 6.93 (d, J = 8.8 Hz, 2H, Ph-3-H, Ph-5-H), 5.16 (s, 2H, CH₂), 3.46 (s, 3H, O-Me), 1.30 (s, 9H, *t*-Bu); MS m/z 218 (M^+ , 49), 203 (10), 173 (27), 45 (100). Anal. ($\text{C}_{14}\text{H}_{18}\text{O}_2 \cdot 0.25\text{H}_2\text{O}$) C, H, N.

4-(Methoxymethoxy)-1-[(trimethylsilyl)ethynyl]benzene (9e): prepared according to **8c** (yield, 79%); oil; $^1\text{H NMR}$ (CDCl_3) δ 7.39 (d, J = 8.6 Hz, 2H, Ph-2-H, Ph-6-H), 6.95 (d, J = 8.6 Hz, 2H, Ph-3-H, Ph-5-H), 5.17 (s, 2H, CH₂), 3.47 (s, 3H, O-Me), 0.24 (s, 9H, Si-Me₃); MS m/z 234 (M^+ , 41), 219 (15), 189 (17), 130 (25), 45 (100). Anal. ($\text{C}_{13}\text{H}_{18}\text{O}_2\text{Si}$) C, H, N.

4-Propynylphenol (10b). A solution of **8b** (20 mmol, 3.0 g) in 100 mL of DMF was heated together with Et₃Na (50 mmol, 4.21 g) under reflux for 1 h. The resulting suspension was then concentrated under reduced pressure, extracted with CH₂Cl₂, filtered, and evaporated again. Column chromatography (eluent: CH₂Cl₂) afforded the title compound as a light-yellow oil (yield, 2.67 g, 98%); $^1\text{H NMR}$ (CDCl_3) δ 7.27 (d, J = 8.6 Hz, 2H, Ph-3-H, Ph-5-H), 6.81 (d, J = 8.6 Hz, 2H, Ph-2-H, Ph-6-H), 5.22 (br*, 1H, OH), 2.02 (s, 3H, Me); MS m/z 132 (M^+ , 100), 103 (18). Anal. ($\text{C}_9\text{H}_8\text{O} \cdot 0.8\text{H}_2\text{O}$) C, H, N.

4-Pentynylphenol (10c). **Method A:** Prepared according to **10b** (yield, 98%). **Method B:** To a solution of **9c** (10 mmol, 2.04 g) in 20 mL of THF was added 5 mL of 10% methanolic HCl followed by stirring at ambient temperature for 24 h. Removal of the solvent under reduced pressure and purification by column chromatography (eluent: CH₂Cl₂) afforded the product as a light-yellow oil (yield, 1.52 g, 95%); $^1\text{H NMR}$ (CDCl_3) δ 7.27 (d, J = 8.7 Hz, 2H, Ph-3-H, Ph-5-H), 6.74 (d, J = 8.7 Hz, 2H, Ph-2-H, Ph-6-H), 4.99 (s*, 1H, OH), 2.36 (t, J = 7.0 Hz, 2H, C \equiv C-CH₂), 1.61 (m, 2H, CH₂-Me), 1.04 (t, J = 7.3 Hz, 3H, CH₂-Me); MS m/z 160 (M^+ , 79), 131 (100). Anal. ($\text{C}_{11}\text{H}_{12}\text{O} \cdot 0.25\text{H}_2\text{O}$) C, H, N.

4-(3,3-Dimethylbutynyl)phenol (10d). **Method A:** Prepared according to **10b** (yield, 95%). **Method B:** Prepared according to **10c** (yield, 90%); crystallized from light petroleum; mp 90 °C; $^1\text{H NMR}$ (CDCl_3) δ 7.26 (d, J = 8.6 Hz, 2H, Ph-3-H, Ph-5-H), 6.73 (d, J = 8.6 Hz, 2H, Ph-2-H, Ph-6-H), 4.86 (br*, 1H, OH), 1.30 (s, 9H, *t*-Bu); MS m/z 174 (M^+ , 57), 159 (100), 144 (13). Anal. ($\text{C}_{12}\text{H}_{14}\text{O}$) C, H, N.

4-[(Trimethylsilyl)ethynyl]phenol (10e): prepared from (4-iodophenoxy)trimethylsilane (**7**) according to **8c**; crystallized from light petroleum (yield, 85%); mp 68 °C; $^1\text{H NMR}$ (CDCl_3) δ 7.33 (d, J = 8.6 Hz, 2H, Ph-3-H, Ph-5-H), 6.72 (d, J = 8.6 Hz, 2H, Ph-2-H, Ph-6-H), 5.43 (br*, 1H, OH), 0.23 (s, 9H, Si-Me₃); MS m/z 190 (M^+ , 31), 175 (100), 116 (13). Anal. ($\text{C}_{11}\text{H}_{14}\text{OSi}$) C, H, N.

4-(3-Hydroxypropyl)-1H-imidazole-1-carboxylic Acid 1,1-Dimethylethyl Ester (12).²⁷ 3-(1H-imidazol-4-yl)propanol (**11**; 10 mmol, 3.7 g)¹¹ was dissolved in 20 mL of MeCN and 5 mL of dioxane/H₂O (1:1). After addition of 5 mL of Et₃N, DMAP (1 mmol, 0.12 g), and Boc₂O (12 mmol, 2.62 g), the solution was stirred at ambient temperature for 2 h followed by evaporation to dryness. The residue was subjected to column chromatography [eluent: CH₂Cl₂ (90%), MeOH (10%)] to give the title compound as an oil which crystallized at 4 °C (yield, 2.21 g, 97%); mp 76–78 °C; $^1\text{H NMR}$ ($\text{Me}_2\text{SO}-d_6$) δ 8.08 (s, 1H, Im-2-H), 7.21 (s, 1H, Im-5-H), 4.47 (t, J = 5.1 Hz, 1H, OH), 3.43 (m, 2H, O-CH₂), 2.50 (t, J = 7.7 Hz, 2H, Im-CH₂), 1.71 (m, 2H, Im-CH₂-CH₂), 1.56 (s, 9H, *t*-Bu); MS m/z (FAB⁺) 227 ($\text{M} + \text{H}^+$, 17), 171 (81), 127 (40), 109 (15), 81 (11). Anal. ($\text{C}_{11}\text{H}_{18}\text{N}_2\text{O}_3 \cdot 0.5\text{H}_2\text{O}$) C, H, N.

4-Ethynylphenyl 3-(1H-imidazol-4-yl)propyl ether (14a): for preparation see **14e**; crystallized as hydrogen maleate from EtOH/Et₂O; $^1\text{H NMR}$ ($\text{Me}_2\text{SO}-d_6$) δ 8.89 (s, 1H, Im-2-H), 7.42 (m, 3H, Im-5-H, Ph-2-H, Ph-6-H), 6.94 (d, J = 8.4 Hz, 2H, Ph-

3-H, Ph-5-H), 6.07 (s, 2H, Mal), 4.06 (m, 3H, O-CH₂, CH), 2.82 (t, J = 7.4 Hz, 2H, Im-CH₂), 2.09 (m, 2H, Im-CH₂-CH₂); MS m/z 226 (M^+ , 12), 209 (1), 153 (1), 109 (100), 82 (36). Anal. ($\text{C}_{14}\text{H}_{14}\text{N}_2\text{O} \cdot \text{C}_4\text{H}_4\text{O}_4 \cdot 0.25\text{H}_2\text{O}$) C, H, N.

3-(1H-imidazol-4-yl)propyl 4-Propynylphenyl Ether (14b). Triphenylphosphine (6 mmol, 1.57 g) was dissolved in 20 mL of THF under N₂ atmosphere together with **12** (5 mmol, 1.13 g) and **10c** (5 mmol, 0.80 g) and cooled in an ice bath. Then diethyl azodicarboxylate (6 mmol, 0.95 mL) was added dropwise followed by additional stirring for 72 h at ambient temperature. Evaporation under reduced pressure and purification by column chromatography (eluent: EtOAc) afforded the Boc-protected product **13b** which was deprotected by stirring in 25 mL of methanolic NH₃ (10%) at ambient temperature for 6 h. The residue was concentrated under reduced pressure and subjected to column chromatography [eluent: CH₂Cl₂ (90%), MeOH (10%)] to give the title compound as an oil, which was crystallized as hydrogen maleate from EtOH/Et₂O; $^1\text{H NMR}$ ($\text{Me}_2\text{SO}-d_6$) δ 8.88 (s, 1H, Im-2-H), 7.42 (s, 1H, Im-5-H), 7.30 (d, J = 8.6 Hz, 2H, Ph-2-H, Ph-6-H), 6.87 (d, J = 8.6 Hz, 2H, Ph-3-H, Ph-5-H), 6.04 (s, 2H, Mal), 4.01 (t, J = 6.1 Hz, 2H, O-CH₂), 2.78 (t, J = 7.5 Hz, 2H, Im-CH₂), 2.06 (m, 2H, Im-CH₂-CH₂), 2.00 (s, 3H, Me); MS m/z 240 (M^+ , 16), 158 (100), 109 (100), 82 (23). Anal. ($\text{C}_{15}\text{H}_{16}\text{N}_2\text{O} \cdot \text{C}_4\text{H}_4\text{O}_4 \cdot 0.5\text{H}_2\text{O}$) C, H, N.

3-(1H-imidazol-4-yl)propyl 4-pentynylphenyl ether (14c): prepared according to **14b**; crystallized as hydrogen maleate from EtOH/Et₂O; $^1\text{H NMR}$ ($\text{Me}_2\text{SO}-d_6$) δ 8.89 (s, 1H, Im-2-H), 7.43 (s, 1H, Im-5-H), 7.30 (d, J = 8.6 Hz, 2H, Ph-2-H, Ph-6-H), 6.87 (d, J = 8.6 Hz, 2H, Ph-3-H, Ph-5-H), 6.05 (s, 2H, Mal), 4.01 (t, J = 6.1 Hz, 2H, O-CH₂), 2.79 (t, J = 7.4 Hz, 2H, Im-CH₂), 2.36 (t, J = 7.0 Hz, 2H, C \equiv C-CH₂), 2.06 (m, 2H, Im-CH₂-CH₂), 1.53 (m, 2H, CH₂-Me), 0.98 (t, J = 7.3 Hz, 3H, Me); MS m/z 268 (M^+ , 13), 131 (6), 109 (100), 82 (14). Anal. ($\text{C}_{17}\text{H}_{20}\text{N}_2\text{O} \cdot \text{C}_4\text{H}_4\text{O}_4 \cdot 0.25\text{H}_2\text{O}$) C, H, N.

4-(3,3-Dimethylbutynyl)phenyl 3-(1H-imidazol-4-yl)propyl ether (14d): prepared according to **14b**; crystallized as hydrogen maleate from EtOH/Et₂O; $^1\text{H NMR}$ ($\text{Me}_2\text{SO}-d_6$) δ 8.90 (s, 1H, Im-2-H), 7.43 (s, 1H, Im-5-H), 7.27 (d, J = 8.7 Hz, 2H, Ph-2-H, Ph-6-H), 6.87 (d, J = 8.7 Hz, 2H, Ph-3-H, Ph-5-H), 6.06 (s, 2H, Mal), 4.01 (t, J = 6.1 Hz, 2H, O-CH₂), 2.80 (t, J = 7.5 Hz, 2H, Im-CH₂), 2.07 (m, 2H, Im-CH₂-CH₂), 1.27 (s, 9H, *t*-Bu); MS m/z 282 (M^+ , 14), 267 (4), 159 (7), 109 (100), 82 (11). Anal. ($\text{C}_{18}\text{H}_{22}\text{N}_2\text{O} \cdot \text{C}_4\text{H}_4\text{O}_4 \cdot 0.5\text{H}_2\text{O}$) C, H, N.

3-(1H-imidazol-4-yl)propyl 4-[(Trimethylsilyl)ethynyl]phenyl Ether (14e). This was prepared according to **14b**. The Boc-protected product **13e** was deprotected by stirring in 25 mL of methanolic hydrazine solution (10%) at ambient temperature for 30 min. The residue was concentrated under reduced pressure and subjected to column chromatography [eluent: CH₂Cl₂ (90%), MeOH (10%)] to separate the title compound from **14a** giving it as an oil which was crystallized as hydrogen maleate from EtOH/Et₂O; $^1\text{H NMR}$ ($\text{Me}_2\text{SO}-d_6$) δ 8.86 (s, 1H, Im-2-H), 7.42 (m, 3H, Im-5-H, Ph-2-H, Ph-6-H), 6.93 (d, J = 8.5 Hz, Ph-3-H, Ph-5-H), 6.06 (s, 2H, Mal), 4.06 (t, 2H, O-CH₂), 2.81 (t, J = 7.3 Hz, 2H, Im-CH₂), 2.09 (m, 2H, Im-CH₂-CH₂), 0.23 (s, 9H, Si-Me₃); MS m/z 298 (M^+ , 6), 283 (3), 175 (5), 147 (4), 109 (100), 82 (23); IR (KBr) $\tilde{\nu}$ 3432 (br), 3141 (m), 2957 (m, C-H), 2877 (m, C-H), 2155 (w, C \equiv C), 1574 (vs, C=C), 1510 (vs, C=C). Anal. ($\text{C}_{17}\text{H}_{22}\text{N}_2\text{OSi} \cdot \text{C}_4\text{H}_4\text{O}_4 \cdot 0.75\text{H}_2\text{O}$) C, N; H: calcd, 6.48; found, 6.04.

Pharmacology. General Methods: Histamine H₃-Receptor Assay on Synaptosomes from Rat Cerebral Cortex. The new compounds were tested for their H₃-receptor antagonist activity at least in triplicate in an assay with K⁺-evoked depolarization-induced release of [³H]histamine from synaptosomes according to Garbarg et al.²³

Histamine H₃-Receptor Antagonist Activity in Vivo in Mouse. In vivo testing of all compounds was performed at least in triplicate after oral administration to Swiss mice as described by Garbarg et al.²³

In Vitro Screening at Other Histamine Receptors. All compounds were screened in duplicate for histamine H₂-receptor activity at isolated spontaneously beating guinea pig right

atrium as well as for H₁-receptor activity at isolated guinea pig ileum by standard methods described by Hirschfeld et al.²⁴

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