Pyrazoles as Potential Histamine H3-Receptor Antagonists

Katarzyna Kieć-Kononowicz

Department of Chemical Technology of Drugs, Collegium Medicum, Jagiellonian University, ul. Medyczna 9, 30-688 Kraków, Poland

Xavier Ligneau and Jean-Charles Schwartz

Unité de Neurobiologie et Pharmacologie, Centre Paul Broca de l'INSERM, 2 ter, rue d'Alésia, 75014 Paris, France

Walter Schunack*

Institut für Pharmazie, Freie Universität Berlin, Königin-Luise-Strasse 2+4, 14195 Berlin, Germany

Received December 23, 1994

In search of structure-activity relationships among histamine H3-receptor antagonists, the imidazole ring of known H3-receptor antagonists was replaced by different heteroaromatic ring systems. Thus, pyrazoles with ether (4,5) and carbamate (6,7) moieties as functional groups were synthesized. Reaction of the hydrochloride of 4-(3-hydroxypropyl)pyrazole (1) with phenyl or benzyl isocyanates mainly gave the carbamates 6 and 7, whereas a similar reaction with 1 as the free base furnished the N-carbamoylpyrazoles 8 and 9. The bifunctional pyrazoles 10 and 11 were formed as by-products. The compounds obtained did not show significant H3-receptor antagonist activity in vitro (rat brain cortex) or in vivo (mouse brain). These results demonstrate the importance of the imidazole moiety for H3-receptor antagonists. The new compounds were also screened for H1-receptor antagonist activity on the isolated guinea-pig ileum and for H2-receptor antagonist activity on the isolated spontaneously beating guinea-pig right atrium. The substances showed only weak antagonistic activity at both histamine receptors H1 and H2.

The existence of a third histamine receptor was reported for the first time in 1983¹⁾. It proved to be an autoreceptor with presynaptical location. Interaction of histamine with H₃-receptors leads to inhibition of histamine synthesis in and histamine release from histaminergic neurons.

Thioperamide was the first highly potent and selective H_3 -receptor antagonist to be designed²⁾, since then H_3 -receptor antagonists with very different functional groups have been developed ³⁻⁵⁾. So far most of the H_3 -receptor antagonists have been imidazole derivatives. The very few non-imidazoles with H_3 -receptor antagonist activity do not reach the potency of compounds like thioperamide. As a general construction pattern for H_3 -receptor antagonists the existence of a nitrogen-containing heterocycle connected to a polar group *via* an alkyl chain seems to be essential for a potent interaction with H_3 -receptors. A lipophilic residue linked by a spacer with the polar group seems to enable the molecule to reaching additional binding areas, thereby increasing the H_3 -receptor antagonist activity of the resulting molecule ³⁾.

In our previous work azine and diazine analogues of known H₃-receptor antagonistically active imidazoles were synthesized $^{6)}$. The aim of this work was the synthesis of pyrazole analogues of those azines.

Pyrazole als potentielle Histamin-H3-Rezeptorantagonisten

Im Rahmen von Untersuchungen von Histamin-H3-Rezeptorantagonisten wurde der Imidazolring bekannter H3-Rezeptorantagonisten gegen verschiedene heteroaromatische Ringsysteme ausgetauscht. Es wurden Pyrazole mit Ether-(4,5) und Urethan-Strukturelementen (6,7) als funktionelle Gruppen synthetisiert. Die Reaktion des Hydrochlorids von 4-(3-Hydroxypropyl)pyrazol (1) mit Phenyl- oder Benzylisocyanaten ergab hauptsächlich die Urethane 6 und 7, während eine vergleichbare Umsetzung mit 1 als freier Base die N-Carbamoylpyrazole 8 und 9 lieferte. Als Nebenprodukte wurden die bifunktionellen Pyrazole 10 und 11 erhalten. Die dargestellten Substanzen zeigten in vitro (Zerebralkortex der Ratte) und in vivo (Mäusehirn) keine bedeutsame H3-antagonistische Aktivität. Diese Ergebnisse weisen auf die Bedeutung des Imidazolrings für H3-Rezeptorantagonisten hin. Die neuen Verbindungen wurden auch auf H1-antagonistische Aktivität am Meerschweinchendünndarm und auf H2-Antagonismus am isolierten, spontan schlagenden rechten Vorhof des Meerschweinchens getestet. Die Substanzen zeigten an H1 und H2-Rezeptoren nur schwache antagonistische Aktivität.

Results and Discussion-Synthesis

The starting material in the synthesis of all new compounds was 4-(3-hydroxypropyl)pyrazole (1) obtained according to ref. ⁷⁾. For the synthesis of the ethers 4 and 5 the N-protected pyrazole 3 was used (Scheme 1). The reaction of 1 with a stoichiometric amount of triphenylmethyl chloride in CH2Cl2 in the presence of triethylamine gave the N-tritylpyrazole 3 and small amounts of the N,O-ditritylated derivative 2. As a result of the reaction of 3 with haloalkanes (3-phenylpropyl bromide or 3-cyclohexylpropyl chloride) performed in toluene in the presence of NaH and 15-crown-5 and subsequent acidic hydrolysis, ethers 4 and 5 were obtained. Ether 4 was separated as the hydrochloride, 5 as the free base. The reaction of 1xHCl with stoichiometric amounts of benzyl or phenyl isocyanate in acetonitrile produced mainly the carbamates 6 and 7 besides minor amounts of the N-carbamoylpyrazoles 8 and 9 and the bifunctional pyrazoles 10 and 11. The same reaction performed with 1 as the free base, gave mainly 8 and 9 and smaller amounts of 6 and 7 as well as 10 and 11. The compounds 8-11 were separated by column chromatography. The structures of the compounds obtainedwere confirmed on the basis of elemental and spectral analyses (¹H-NMR, MS).



Scheme 1

Pharmacology

Selected pyrazoles were tested for their *in vitro* (rat brain cortex) and *in vivo* (mouse brain) H₃-receptor antagonist (or agonist) activity ¹). Compounds 4–11 also were screened for H₂-receptor activity on the isolated spontaneously beating guinea-pig right atrium as well as for H₁-receptor activity on the isolated guinea-pig ileum. Results are listed in Table 1.

Table 1: Antagonistic activity of 4-11 on histamine H₃-, H₂-, and H₁-receptors.

Compound	H3 (brain cortex) Ki (rat) ED50, EC50 (mouse)	H ₂ (atrium) —logK _B (guinea pig)	H1 (ileum) —logKB (guinea pig)
4	$K_i > 5 \times 10^{-7} M$ ED ₅₀ > 10 mg/kg p.o.	a)	4.29
5	$K_i > 5 \times 10^{-7} M$	4.28	4.37
6	$K_i > 5x10^{-7} M$	a)	4.38
7	$K_i > 5x10^{-7} M$ $F_{CS0} > 5x10^{-7} M$	a)	4.21
8	$K_i > 5x 10^{-7} M$ ED50 > 10 mg/kg p.o.	3.62	4.23
9		3.63	4.29
10		4.12	4.15
11		3.85	5.45

^{a)} No effect was observed up to 1×10^{-4} M.

The obtained pyrazoles possess hardly any H₃-receptor antagonist properties. Replacement of the imidazole ring with a pyrazole moiety leads to compounds with negligible *in vitro* activity at histamine H₃-receptors. The results clearly demonstrate the major importance of the imidazole ring for potent antagonism at H₃-receptors. Compounds **4–8** were also tested in *in vivo* experiments with mice after peroral administration. None of the tested compounds showed any central H₃-receptor antagonist activity. Compound 7 at a dose of 10 mg/kg *p.o.* to mice reduced the N^{t} -methylhistamine level by about 30%. This effect is usually observed with H₃-receptor agonists. Thus, 7 was also screened for H₃-receptor agonist activity, but when tested *in vitro* 7 did not show significant effects (indicated as EC₅₀ value in Table 1). The reduction of the N^{t} -methylhistamine level might be the result of another mechanism, which needs further investigation.

The compounds also showed only weak antagonistic activity at the two other histamine receptors, H_1 and H_2 . The antagonism was in most cases of the competitive/non-competitive type. Compound 5 is a partial H₂-agonist, the maximum response was only 48% that of histamine.

This work was supported by the European Community Research Programme 'Biomedical and Health Research'. The financial contribution of the Commission for the Concerted Action 'Histamine H₃ Agonists and Antagonists as Drugs' (EEC BMH1 CT92–1087) is highly acknowledged. The study was also supported by a grant from the Verband der Chemischen Industrie, Fonds der Chemischen Industrie. The technical assistance of Mrs. *H. Luka* and *I. Walther*, who performed the pharmacological experiments on the isolated atrium and ileum of the guinea-pig, is gratefully acknowledged.

Experimental Part

Chemistry

M.p.s: Büchi 512 (uncorrected). – Elementary analyses: Perkin-Elmer 240 B and 240 C. – IR: Perkin-Elmer 1420 Ratio Recording IR Spectraphotometer (KBr). – ¹H-NMR: Bruker AC 300, δ [ppm] ref. to TMS. *– signals exchangeable with D₂O. – MS: Finnigan MAT CH7A (170 °C, El/70 eV) or MAT 711 (200 °C, El/80 eV); m/z (%); direct inlet. FAB (*FAB, xenon, DMSO/glycerol) Finnigan MAT CH5 DF. – TLC: Al sheets 0.2 mm layer silica gel (60 F₂₅₄; Merck); solvent systems: I: CHCl₃ : AcOEt (1:1); II: toluene : acetone (20:1.5). – Rotational chromatography: Chromatotron 7924T (Harrison Research); glass rotors, 4 mm layers of silica gel 60 PF₂₅₄ containing gypsum (Merck); developing systems: CHCl₃ : AcOEt (1:1). – Column chromatography (cc): silica gel (Baker) 0.05–0.2 mm. Starting materials: Reaction of 2,3-dihydropyran and ethyl orthoformate in the presence of BF3 yielded 2-ethoxy-3-tetrahydropyran carbaldehyde diethyl acetal which with hydrazine dihydrochloride gave 4-(3-hydroxypropyl)pyrazole (1), see ref.⁷⁾.

Reaction of 1 with chlorotriphenylmethane

A solution of 2.80 g (10 mmol) chlorotriphenylmethane in CH₂Cl₂ (25 ml) was added dropwise to a stirred mixture of 1.26 g (10 mmol) of 1 and 1.01 g (10 mmol) triethylamine in CH₂Cl₂ (25 ml) with cooling. The reaction mixture was stirred at room temp. for 12 h and then washed with 2×50 ml of 2% HCl and 3×50 ml of water; the org. layer was dried (MgSO₄) and evaporated *in vacuo*. 100 ml of EtOH was added, the mixture was refluxed for 10 min, and after cooling the precipitate of 2 was separated; m.p. 84–86 °C (EtOH); Rr(10.90; yield 19%, – C44H₃₈N₂O (610.8) Calcd. C 86.5 H 6.27 N 4.59 Found C 86.4 H 6.26 N 4.42.– ¹H-NMR (CDCl₃): 1.82 (quint, J = 7.3 Hz, 2H, CH₂CH₂CH₂), 2.54 (t, J = 7.6 Hz, 2H, PyrCH₂), 3.05 (t, J = 6.3 Hz, 2H, CH₂O), 7.09–7.43 (m, 32H, 30 Ph-H, Pyr-3-H and Pyr-5-H).– IR: 3048, 3020 (aromat. CH), 2931 (aliphat. CH), 1595 (C=N), 1488, 1445, 1066, 748, 701.– MS: ^{*}FAB: 611 ([M+H]^{*}, 0.6), 243 (100, CPh₃), 228(2), 215(2), 202(2), 178(2), 165(17), 154(4), 136(3), 115(2), 107(1), 77(5).

The ethanolic filtrate was evaporated *in vacuo* to dryness, **3** was obtained by addition of diethyl ether. – m.p. 129–131 °C (Et₂O); R_F(I) 0.58; yield 71%; C₂5H₂₄N₂O (368.4) Calcd. C 81.5 H 6.57 N 7.60 Found C 81.7 H 6.64 N 7.43.– ¹H-NMR (CDCl₃): 1.28* (br. s, 1H, OH), 1.79 (quint, J = 7.5 Hz, 2H, CH₂CH₂CH₂), 2.52 (t, J = 7.6 Hz, 2H, PyrCH₂), 3.64 (t, J = 6.3 Hz, 2H, CH₂O), 7.12–7.31 (m, 16H, 15 Ph-H and Pyr-3-H), 7.49 (s, 1H, Pyr-5-H).– IR: 3424 (OH), 3052, 3021 (aromat. CH), 2936 (aliphat. CH) 1595 (C=N), 1489, 1443, 1129, 757, 745, 700.– EI-MS: 368 (M⁺⁺, 5), 243 (100, CPh₃), 228 (5), 215 (4), 165 (50), 126 (2), 108 (3), 91 (2), 81 (7), 77 (2).

3-Phenylpropyl-3-(4-pyrazolyl)propyl ether (4)

0.24 g (6 mmol) of NaH (60% suspension in mineral oil) and 0.1 ml (0.5 mmol) of 1,4,7,10,13-pentaoxacylcopentadecane (15-crown-5) were added to a solution of 1.84 g (5 mmol) 3 in dry toluene (10 ml). The reaction mixture was stirred at room temp. for 8 h. 2.0 g (10 mmol) of 1-bromo-3phenylpropane was then added and the mixture was stirred on an oil bath at 70-80 °C for 20 h (reaction followed by TLC). After cooling, the toluene was evaporated, 30 ml 2N HCl and 20 ml acetone were added and the mixture was warmed on an oil bath at 70 °C for 2 h. After completion of hydrolysis the solution was cooled, and acetone was evaporated in vacuo. The separated solid (triphenylmethanol) was filtered off, the water layer washed with 3×50 ml diethyl ether, made alkaline with Na2CO3 and extracted thoroughly with CH₂Cl₂. The solution was concentrated after drying over Na₂SO₄, and the oily residue purified by rotational chromatography (I). Ethanol, saturated with gas. HCl, was added to the oily product (4) and repeatedly evaporated to dryness; on addition of dry diethyl ether a precipitate of $4 \times HCl$ was obtained; m.p. 63-64 °C (Et2O); RF(1) 0.37; RF(II) 0.68; yield 52%; C15H20N2OxHCl (280.8). Calcd. C 64.2 H 7.54 N 9.98 Found C 64.3 H 7.50 N 9.79.- 1H-NMR ([D6]DMSO): 1.79 (quint, J = 6.6 Hz, 4H, CH2CH2CH2), 2.51 (t, 2H, signal under DMSO signals, Pyr-CH₂), 2.61 (t, J = 7.7 Hz, 2H, PhCH₂), 3.35 (t, J = 6.3 Hz, 2H, CH₂O), 3.36 (t, J = 6.1 Hz, 2H, CH₂O), 5.34* (br. s, 1H, NH), 7.17 (m, 3H, Ph-3-H, Ph-4-H, and Ph-5-H), 7.26 (d, J = 7.3 Hz, 2H, Ph-2-H and Ph-6-H), 7.93 (s, 2H, Pyr-3-H and Pyr-5-H).- IR: 3016 (aromat. CH), 2935 (aliphat. CH), 2659 (NH⁺), 1491, 1451, 1110, 749, 699. EI-MS: 244 (M⁺⁺, 18), 140(8), 125(9), 118(48), 108(100), 95(9), 91(47), 81(53), 77(4), 65(6).

3-Cyclohexylpropyl-3-(4-pyrazolyl)propyl ether (5)

Compound 5 was prepared from 3 and 3-chloropropylcyclohexane; reaction time 20h; purification: rotational chromatography (I); $R_{\rm P}(I) 0.29$; $R_{\rm P}(II) 0.74$; yield 48%; $C_{15}H_{26}N_{2O}$ (250.4) Calcd. C 72.0 H 10.5 N 11.2 Found C 71.9 H 10.3 N 11.2.– ¹H-NMR (CDCl₃): 0.88–1.68 (m, 15H, CH, CH₂), 1,84 (t, J = 6.9 Hz, 2H, PyrCH₂CH₂), 2.58 (t, J = 7.6 Hz, 2H, PyrCH₂), 3.38 (t, J = 6.2 Hz, 2H, CH₂O), 3.42 (t, J = 6.8 Hz, 2H, CH₂O), 7.41 (s, 2H, Pyr-3-H and Pyr-5-H).– EI-MS: 250 (M⁺⁺, 5), 167(1), 154(4), 139(11), 126(6), 108(100), 95(8), 83(5), 81(33).

N-Benzyl-3-[(1-benzylcarbamoyl)-4-pyrazolyl]propyl carbamate (10)

A mixture of 0.81 g (5 mmol) 1xHCl in dry acetonitrile (20 ml) and 0.67 g benzyl isocyanate (5 mmol) was stirred on an oil bath at 70-80 °C for 3 h. The solvent was evaporated in vacuo. The residue was acidified with 20 ml of 1N HCl, the separated precipitate (0.64 g) filtered off, and the water layer washed with diethyl ether $(3 \times 20 \text{ ml})$. The collected ether extracts were dried over Na₂SO₄, evaporated to dryness, combined with the precipitate, and separated by cc (20 g) (CHCl3 : AcOEt, 1:1). Fractions 1-4 were evaporated, recrystallized (EtOH) to afford 0.1 g of 10; m.p. 98-99 °C; RF(I) 0.75; yield 5%; C22H23N4O3 (392.4) Calcd. C 67.3 H 6.17 N 14.3 Found C 66.9 H 6.12 N 14.3.- H-NMR ([D6]DMSO); 1.83 (m, 2H, CH2CH2CH2), 2.50 (t, 2H, PyrCH₂), 3.97 (t, J = 6.3 Hz, 2H, CH₂O), 4.17 (d, J = 5.9 Hz, 2H, PhCH₂NHurethane), 4.41 (d, J = 6.9 Hz, 2H, PhCH2NH-urea), 7.25 (s, 1H, Pyr-3-H), 7.23-7.31 (m, 10 H, Ph-H), 7.32 (s, 1H, Pyr-5-H), 7.71 (s, 1H, CONH-urethane), 8.10 (s, 1H, CONH-urea).-IR: 3345, 3116 (NH), 3053, 3020 (aromat. CH), 2926 (aliphat. CH), 1711 (C=O-urea), 1695 (C=O-urethane), 1527 (C=N), 1241 (O-CO), 740, 697.- MS: *FAB: 393 ([M+H]*, 12), 260(94), 154(98), 136(71), 132(14), 107(41), 91(100), 81(35), 77(40), 63(18).

1-Benzylcarbamoyl-4-(3-hydroxypropyl)pyrazole (8)

Fractions 5–10 of the above synthesis were evaporated, recrystallized (AcOEt : Et2O) to afford 0.49 g 8; m.p. 49–51 °C; RF(I) 0.35; yield 38%; C14HnN3O2 (259.3) Calcd. C 64.9 H 6.59 N 16.2 Found C 65.0 H 6.72 N 16.3. ¹H-NMR (CDCI₃): 1.84 (quint, J = 7.5 Hz, 2H, PyrCH₂CH₂CH₂C), 2.59 (t, J = 7.6 Hz, 2H, PyrCH₂), 3.68 (t, J = 6.3 Hz, 2H, CH₂O), 4.59 (d, J = 6.1 Hz, 2H, PhCH₂), 7.34 (s, 1H, Pyr-3-H), 7.28–7.36 (m, 5H, PhH), 7.44 (s, 1H, Pyr-5-H), 8.03 (s, 1H, CONH).– IR: 3391 (OH), 3322, 3113 (NH), 3023 (aromat. CH), 2935 (aliphat. CH), 1714 (C=O), 1527 (C=N), 1384, 751, 699.– MS: ⁺FAB: 260 ([M+H]^{*}, 9), 243(1), 217(2), 206(2), 165(1), 155(1), 139(4), 132(16), 127(100), 109(11), 103(11), 95(13), 91(83), 81(38), 77(9).

N-Benzyl-3-(4-pyrazolyl)propyl carbamate (6)

The water layer from the synthesis of 10 was made alkaline with solid NaHCO₃, extracted with CH₂Cl₂ (3×30 ml), the organic layer was dried (Na₂SO₄), evaporated to dryness, recrystallized (AcOEt : cyclohexane) to give 0.66 g 6; m.p. 108–109 °C; R_F(I) 0.19; yield 52%; C₁₄H₁₇N₃O₂ (259.3) Calcd. C 64.9 H 6.59 N 16.2 Found C 64.7 H 6.54 N 16.0.– ¹H-NMR ([D6]DMSO): 1.79 (m, 2H, CH₂CH₂CH₂), 2.47 (t, J = 7.7 Hz, 2H, PyrCH₂), 3.96 (t, J = 6.2 Hz, 2H, CH₂O), 4.17 (d, J = 5.7 Hz, 2H, PhCH₂), 7.25 (s, 1H, Pyr-3-H), 7.23–7.31 (m, 5H, Ph-H), 7.41 (s, 1H, Pyr-5-H), 7.69* (br. s, 1H, CONH), 12.50* (br. s, 1 H PyrNH).– IR: 3318, 3171 (NH), 3054, 3023 (aromat. CH), 2931 (aliphat. CH), 1687 (C=O), 1539 (C=N), 1276 (O-CO), 695.– EI-MS: 259 (M⁺⁺, 1), 213(1), 163(1), 150(4), 133(25), 126(8), 108(100), 104(15), 91(40), 81(66), 77(11).

The reaction above, performed with with free base 1, gave after cc separation 10, 6, and 8 in 10%, 7%, and 79% yield, respectively.

N-Phenyl-3-[(1-phenylcarbamoyl)-4-pyrazolyl]propyl carbamate (11)

Compound 11 was prepared by reaction of 1xHCl with phenylisocyanate and similar separation; m.p. 135–137 °C (EtOH); $R_F(I)$ 0.85; yield 16%; $C_{20}H_{20}N_4O_3$ (364.4) Calcd. C 65.9 H 5.54 N 15.4 Found C 65.8 H 5.51 N 15.3.–¹H-NMR ([D6]DMSO): 1.87 (quint, J = 7.5 Hz, 2H, CH₂CH₂CH₂), 2.53 (t, 2H, signals under DMSO signals, PyrCH₂), 4.06 (t, J = 6.5 Hz, 2H, CH₂O), 6.96 (t, J = 7.0 Hz, 1H, Ph-4-H-urethane), 7.16 (t, J 7.2 Hz, 1H, Ph-4-H-urea), 7.27 (m, 2H, Ph-3-H-urethane and Ph-5-H-urethane), 7.36 (m, 2H, Ph-3-H-urea and Ph-5-H-urea), 7.46 (m, 3H, Ph-2-H-urethane, Ph-6-Hurethane, and Pyr-3-H), 7.60 (d, 2H, Ph-2-H-urea and Ph-6-H-urea), 8.10 (s, 1H, Pyr-5-H), 9.05 (br. s, 1H, CONH-urea), 9.64 (s, 1H, CONH-urethane).-IR: 3354, 3318, 3115 (NH), 3052 (aromat. CH), 2951 (aliphat. CH), 1710 (C=O), 1527 (C=N), 1228 (O-CO), 754.– MS: *FAB: 365 ([M+H]⁺, 10), 246(100), 138(15), 127(7), 120(17), 109(98), 93(63), 81(87), 77(38), 65(14).

1-Phenylcarbamoyl-4-(3-hydroxypropyl)pyrazole (9)

Compound 9 was isolated as described for 8; m.p. 87–88 °C (CHCl₃: AcOEt); R_P(1) 0.42; yield 18%; C₁₃H₁₅N₃O₂ (245.2) Calcd. C 63.7 H 6.16 N 17.1 Found C 63.9 H 6.19 N 17.1.– ¹H-NMR (CDCl₃): 1.87 (quint, J = 7.5 Hz, 2H, CH₂CH₂CH₂), 2.63 (t, J = 7.6 Hz, 2H, PyrCH₂), 3.71 (t, J = 6.3 Hz, 2H, CH₂O), 7.16 (t, J = 7.4 Hz, 1H, Ph-4-H), 7.36 (t, J = 7.9 Hz, 2H, Ph-3-H

and Ph-5-H), 7.53 (s, 1H, Pyr-3-H), 7.60 (d, J = 7.8 Hz, 2H, Ph-2-H and Ph-6-H), 8.10 (s, 1H, Pyr-5-H), 9.05 (br. s, 1H, CONH).– IR: 3340 (OH), 3113 (NH), 3051 (aromat. CH), 2937 (aliphat. CH), 1709 (C=O), 1533 (C=N), 1447, 751. EI-MS: 245 (M^{++} , 0.2), 126(15), 119 (100, PhNHCO), 108(16), 91(21), 81(24), 64(4).

N-Phenyl-3-(4-pyrazolyl)propyl carbamate (7)

Compound 7 was isolated as described for 6; m.p. 137–138 °C (AcOEt : cyclohexane); $R_F(1)$ 0.25; yield 57%; $C_{13}H_{15}N_{3}O_2$ (254.2) Calcd. C 63.7 H 6.16 N 17.1 Found C 63.5 H 6.08 N 16.9–¹H-NMR ([D6]DMSO): 1.87 (quint, J = 7.3 Hz, 2H, CH₂CH₂CH₂), 2.53 (t, J = 7.4 Hz, 2H, PyrCH₂), 4.08 (t, J = 6.5 Hz, 2H, CH₂O), 6.98 (t, J = 7.3 Hz, 1H, Ph-4-H), 7.27 (t, J = 7.8 Hz, 2H, Ph-3-H and Ph-5-H), 7.46 (d, J = 7.8 Hz, 4H, Ph-2-H, Ph-6-H, Pyr-3-H, and Pyr-5-H), 9.64 (s, 1H, CONH), 12.56* (br. s, 1H, NH).- IR: 3325, 3151 (NH), 3046 (aromat. CH), 2949 (aliphat. CH), 1692 (C=O), 1525 (C=N), 1231 (O-CO). EI-MS: 245 (M^{**}, 5), 137(15), 126(15), 119(88), 108(100), 93(46), 91(27), 81(91), 77(8), 64(14).

The reaction above, performed with free base 1, gave after cc separation 11, 7, and 9 with 20%, 6%, and 69% yield, respectively.

Pharmacology

The new compounds were tested *in vitro* for H₃-receptor antagonist activity in an assay with K^+ -evoked release of [³H]histamine with synaptosomes of rat brain cortex ¹⁾. The *in vivo* efficiency of the compounds toward H₃-receptors was evaluated by their effect on the central N^+ -methylhistamine level. This level was measured in whole brain of mice after peroral administration of the compounds. They were also tested for H₁-receptor antagonist activity by in the classical isolated guinea-pig ileum assay according to a modified procedure of *Lennartz et al.*⁸⁾ and screened for H₂-receptor antagonism of the histamine-stimulated increase in heart rate on the spontaneously beating guinea-pig right atrium ^{9,10)}. In each case a minimum of two determinations were performed. The presented data are mean values. Compound 4 was applied as solution in H_2O with an equivalent amount of HCl; compounds 5-7 as solutions in H_2O : EtOH (1:1) mixtures with an equivalent amount of HCl; compounds 8-11 as solutions in DMSO.

References

- 1 J.-M. Arrang, M. Garbarg, J.-C. Schwartz, Nature (London) 1983, 302, 832-837.
- 2 J.-M. Arrang, M. Garbarg, J.-C. Lancelot, J.-M. Lecomte, H. Pollard, M. Robba, W. Schunack, J.-C. Schwartz, *Nature (London)* **1987**, 327, 117–123.
- 3 R. Lipp, H. Stark, W. Schunack in *The Histamine Receptor* (Ed. J.C. Schwartz and H.L. Haas) Ser.: Receptor Biochemistry and Methodology, Wiley-Liss Inc., New York, 1992, vol. 16, 57-72.
- 4 R. Leurs, H. Timmerman in *Progress in Drug Research* (Ed. E. Jucker) Birkhäuser Verlag, Basel, **1992**, *39*, 127–165.
- 5 J.-C. Schwartz, J.-M. Arrang, M. Garbarg, J.-M. Lecomte, C.R. Ganellin, A. Fkyerat, W. Tertiuk, W. Schunack, R, Lipp, H. Stark, K. Purand, PCT Int. Appl. WO 93 14,070 (10.1.1992); Chem. Abstr. 1994, 120, 107004c.
- 6 K. Kieć-Kononowicz, X. Ligneau, H. Stark, J.-C. Schwartz, W. Schunack, Arch. Pharm. (Weinheim) 1995, 328, 445–450.
- 7 R.G. Jones, M.J. Mann, J. Am. Chem. Soc. 1953, 75, 4048-4052.
- 8 H.G. Lennartz, M. Hepp, W. Schunack, Eur. J. Med. Chem. Chim. Ther. 1978, 13, 229–234.
- 9 A. Buschauer, S. Postius, I. Szelenyi, W. Schunack, Arzneim.-Forsch. 1985, 35, 1025–1029.
- 10 J.M. van Rossum, Arch. Int. Pharmacodyn. Ther. 1963, 143, 299-330. [Ph317]