Synthesis of Original Trifluoromethylated 6-Aryl-pyridazines Fused with Thiazolidine or 1,2,4-Triazole

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Abstract: An efficient synthesis of original 8-trifluoromethyl-7*H*-thiazolo[3,2-*b*]- and 1,2,4-triazolo[4,3-*b*]pyridazines is described. Starting from the 4-trifluoromethyl-4,5-dihydropyridazin-3-one, the methodology involves a five-membered ring closure, based on the reaction of a bis(electrophilic) reagent with an exocyclic heteroatom linked to position 3 and the endocyclic nitrogen at position 2 of the pyridazine nucleus.

Key words: fluorine, nitrogen heterocycles, pyridazine derivatives, thiazolo[3,2-*b*]pyridazine, 1,2,4-triazolo[4,3-*b*]pyridazine

The 6-aryl-1,2,4-triazolo[4,3-*b*]pyridazine moiety is an interesting pharmacophore, well-known for its anxiolytic and antihypertensive properties.² Some of its derivatives are also useful agents for the treatment of asthma.³ Moreover, because fluorine incorporation on a molecule could considerably modify its physico-chemical and biological properties, the use of fluorine in potential drugs is increasing.⁴ For instance, CL 218,872 represents such a trifluoromethylated pharmacophore possessing the ability to bind the benzodiazepine receptors in rat brain (Figure 1), which appears to be useful against anxiety.⁵





Several trifluoromethylated 1,2,4-triazolo[4,3-*b*]pyridazines have already been synthesized^{2b,6} but never at the position proposed here. Our previous report dealt with the synthesis of 4-trifluoromethylated pyridazines from perfluoroketene dithioacetal **1** in an original and efficient three-step pathway.⁷ The current paper addresses the synthesis of more sophisticated bis(heterocycles) **4** and **8–10** (Scheme 1). Two general methodologies can be considered to prepare these targets based on the priority order of ring construction. Rather than a five- then six-membered ring construction,⁸ we chose to first build the pyridazine

SYNTHESIS 2006, No. 1, pp 0103–0106 Advanced online publication: 27.10.2005 DOI: 10.1055/s-2005-918433; Art ID: Z11705SS © Georg Thieme Verlag Stuttgart · New York moiety due to our easy and efficient synthesis of 4,5-dihydropyridazin-3-ones.⁷

Thus compound **2** was the common starting material for the syntheses of both 7*H*-thiazolo[3,2-*b*]pyridazines and 1,2,4-triazolo[4,3-*b*]pyridazines. The presence of two nucleophilic centers, one exocyclic heteroatom linked to position 3 and the endocyclic nitrogen at position 2, would allow double nucleophilic substitution with a bis(electrophile).





A thionation reaction with Lawesson's reagent⁹ was applied to compound **2** to give the corresponding 4,5-dihydropyridazin-3-thione **3** (Scheme 2). Then, **3** was reacted with methyl α -bromoacetate giving the 7*H*-thiazolo[3,2-*b*]pyridazine **4** in a totally regioselective reaction. The ring closure was then performed by an in situ amidification reaction. The presence of an acidic hydrogen at position 4 of **3** (α to the trifluoromethyl group), avoids the formation of a quaternary ammonium salt in the final product.¹⁰



Scheme 2 *Reagents and conditions:* (a) Lawesson's reagent, toluene, reflux, 2 h; (b) BrCH₂CO₂Me, DMF, 100 °C, 5 h

The 1,2,4-triazolo[4,3-*b*]pyridazines synthesis is depicted in Scheme 3. First, the pyridazin-3-one **5**, obtained by oxidation of **2** with copper(II) chloride, was reacted with phosphorous oxy chloride to yield the 3-chloropyridazine **6**.⁷ Then nucleophilic addition of hydrazine to **6** followed by chloride elimination, led to 3-hydrazinopyridazine **7** in



Scheme 3 Reagents and conditions: (a) $CuCl_2$, CH_3CN , reflux, 7 h; (b) $POCl_3$, 70–80 °C, 4 h; (c) NH_2NH_2 · H_2O , CH_3CN , 80 °C, 3 h; (d) AcOH, 100 °C, 6 h; (e) 1,1'-carbonyldiimidazole, CH_3CN , reflux, 4 h; (f) BrCN, EtOH, r.t., 5 h; (g) HCl (6 N), *i*-PrOH, r.t., 20 min

good yield. To obtain 1,2,4-triazolo[4,3-b]pyridazines, the triazolo ring must be formed from the reaction of a 1,1-bis(electrophile) reagent with the 1,4-bis(nucleophile) 7. In this way, the 3-methyl-1,2,4-triazolo[4,3-b]pyridazine 8 was prepared by heating 7 in an acetic acid medium. The reaction of 7 with 1,1'-carbonyldiimidazole vielded the heterocycle 9 which exists in two tautomeric forms. Analytical data highlight the predominance of the keto form. The IR spectrum shows the carbonyl vibration at 1724 cm⁻¹ and the NH at 3135 cm⁻¹. Moreover, the ¹H NMR spectrum in DMSO- d_6 shows a broad singlet at 13.1 ppm corresponding to the amide proton; such observations were consistent with what had been reported previously.¹¹ The reaction of 7 with cyanogen bromide gave the 1,2,4-triazolo[4,3-b]pyridazine 10a with a free primary amine function.¹² Then, simple treatment with an acidic solution of isopropanol, quantitatively yielded its dihydrochloride form 10b.

The presence of an aryl bromide on these compounds both allows further chemical transformations like palladiumcatalyzed reactions and could have biological relevance. Indeed Albright and co-workers reported antihypertensive effects related to the presence of an electron-withdrawing group on the phenyl ring of such compounds.^{2a}

The procedures depicted allow an efficient access, under mild conditions, to new biologically interesting compounds such as **4** and **8–10**. The presence of a trifluoromethyl group, in such an original position on 6-arylpyridazines fused with thiazolidine and 1,2,4-triazole make them good candidates for further biological evaluation. ppm relative to an internal standard: TMS (0.00 ppm) or CHCl₃ (7.27 ppm) for ¹H and ¹³C NMR and fluorotrichloromethane (0.00 ppm) for ¹⁹F NMR spectra. Coupling constants are reported in Hz. All reactions were monitored by TLC with Merck silica gel 60 F254 0.25 mm plates. IR spectra were recorded with a Avatar 320 Fourier transform spectrometer. LRMS were obtained with either a mass spectrometer coupled with gas chromatography (GC-MS) Thermoquest Trace GC 2000 series, in EI mode, or a mass spectrometer quadripole Micromass 2MD (ESI+) coupled with liquid chromatography Waters (LC-MS). HRMS were recorded on a Q-TOF Micromass spectrometer in positive electrospray mode (ES+, EC 30 V). Elemental analyses were performed with a Perkin-Elmer CHN 2400 apparatus. Petroleum ether used had a bp range 40–60 °C.

For analytical data of compounds 5 and 6 see reference 7.

6-(4'-Bromophenyl)-4-trifluoromethyl-4,5-dihydro-2*H*-py-ridazin-3-thione (3)

To a solution of 4,5-dihydropyridazin-3-one **2** (300 mg, 0.9 mmol) in toluene (5 mL), was added Lawesson's reagent (416 mg, 1.0 mmol). The mixture was stirred at 110 °C for 2 h. After cooling, the residue was washed with brine (4 mL). The organic layer was extracted with Et_2O (2 × 5 mL), dried over MgSO₄, filtered, and evaporated in vacuo. The product was purified by silica gel chromatography to give a yellow solid.

Yield: 255 mg (84%); mp 220–222 °C; $R_f 0.6$ (petroleum ether-EtOAc, 80:20).

IR (KBr): 3253 (NH), 2942, 1584, 1476, 1342 cm⁻¹.

¹H NMR (acetone- d_6): δ = 3.00 (br s, 1 H, NH), 3.28 (dd, J = 7.7, 18.3 Hz, 1 H, 5-H), 3.42 (dd, J = 3.3, 18.3 Hz, 1 H, 5-H), 4.1 (m, 1 H, 4-H), 7.67 (d, J = 8.5 Hz, 2 H, Ar-H), 7.85 (d, J = 8.5 Hz, 2 H, Ar-H).

¹³C NMR (acetone-*d*₆): δ = 21.7 (q, *J* = 2.7 Hz, 5-C), 47.8 (q, *J* = 26.7 Hz, 4-C), 125.3 (CBr), 125.6 (q, *J* = 281.2 Hz, CF₃), 128.8, 132.7 (Ar-C), 135.0 (1'-C), 152.7 (6-C), 185.3 (3-C).

¹⁹F NMR (acetone- d_6): $\delta = -67.9$ (d, J = 9.5 Hz).

Anal. Calcd for $C_{11}H_8BrF_3N_2S$: C, 39.19; H, 2.39; N, 8.31. Found: C, 39.14; H, 2.26; N, 8.33.

MPs were determined on a Büchi apparatus and are uncorrected. ¹H, ¹³C, and ¹⁹F nuclear NMR spectra were recorded with either a Bruker AC-250 or AC-400 spectrometer. Chemical shifts are reported in

6-(4'-Bromophenyl)-8-trifluoromethyl-7*H*-thiazolo[3,2-*b*]py-ridazin-3-one (4)

To a solution of 4,5-dihydropyridazin-3-thione **3** (200 mg, 0.6 mmol) in DMF (5 mL) was added methyl α -bromoacetate (68 μ L, 0.7 mmol). The mixture was stirred at 100 °C for 5 h. After cooling, DMF was evaporated. The mixture was diluted in Et₂O (10 mL) and neutralized, at 0 °C, with a sat. soln of NaHCO₃. After decantation and separation, the aqueous layer was washed with Et₂O (3 × 10 mL). The combined organic layers were dried over Na₂SO₄, filtered, and evaporated in vacuo. The product was purified by silica gel chromatography to give a solid.

Yield: 129 mg (57%); mp 190–192 °C; $R_f 0.4$ (petroleum ether-EtOAc, 80:20).

IR (KBr): 1720, 1668, 1612, 1381 cm⁻¹.

¹H NMR (DMSO-*d*₆): δ = 3.62 (s, 2 H, 7-H), 4.07 (s, 2 H, 2-H), 7.71 (d, *J* = 8.5 Hz, 2 H, Ar-H), 7.85 (d, *J* = 8.5 Hz, 2 H, Ar-H).

¹³C NMR (DMSO- d_6): δ = 23.1 (7-C), 30.0 (2-C), 91.9 (q, *J* = 33.8 Hz, 8-C), 124.7 (q, *J* = 269.7 Hz, CF₃), 124.9 (CBr), 128.5, 131.6 (Ar-C), 133.4 (1'-C), 135.9 (NCS), 150.7 (6-C), 165.9 (3-C).

¹⁹F NMR (DMSO- d_6): $\delta = -61.8$ (s).

GC-MS (EI): m/z (%) = 378 [M⁺ + 1], 376 (100) [M⁺ - 1], 349, 309 [M⁺ + 1 - CF₃], 221 [M⁺ - PhBr], 183.

Anal. Calcd for $C_{13}H_8BrF_3N_2OS$: C, 41.40; H, 2.14; N, 7.43. Found: C, 41.37; H, 1.95; N, 7.27.

6-(4'-Bromophenyl)-3-hydrazino-4-trifluoromethylpyridazine (7)

To a solution of 3-chloropyridazine **6** (500 mg, 1.5 mmol) in MeCN (10 mL), was added NH₂NH₂·H₂O (360 μ L, 7.5 mmol). The mixture was heated at 80 °C for 3 h. The solvent was removed in vacuo. The resulting residue was diluted in CH₂Cl₂ (10 mL) and washed with a sat. soln of NaHCO₃ (8 mL). After decantation and separation, the aqueous layer was washed with CH₂Cl₂ (2 × 10 mL). The combined organic layers were dried over Na₂SO₄, filtered, and concentrated in vacuo. The residue can be chromatographed on silica gel or used directly for the next step. Purification yielded a white solid.

Yield: 425 mg (85%); mp 91 °C.

¹H NMR (DMSO- d_6): $\delta = 4.7$ (br s, 2 H, NH₂), 7.66 (d, J = 8.0 Hz, 2 H, Ar-H), 8.02 (d, J = 8.0 Hz, 2 H, Ar-H), 8.08 (s, 1 H, 5-H), 8.4 (br s, 1 H, NH).

¹³C NMR (DMSO- d_6): δ = 112.3 (m, 4-C), 121.6 (m, 5-C), 122.5 (CBr), 122.6 (q, *J* = 273.3 Hz, CF₃), 127.7, 131.8 (Ar-C), 134.8 (1'-C), 149.0 (6-C), 153.6 (3-C).

¹⁹F NMR (DMSO- d_6): $\delta = -66.4$ (s).

6-(4'-Bromophenyl)-3-methyl-8-trifluoromethyl-1,2,4-triazolo[4,3-*b*]pyridazine (8)

3-Hydrazinopyridazine 7 (100 mg, 0.3 mmol) was heated at 100 °C in an excess of AcOH (0.5 mL) for 6 h. The mixture was cooled to 0 °C, diluted with EtOAc (5 mL), and neutralized with a sat. soln of Na₂CO₃ (3 mL). After separation, the aqueous layer was washed with EtOAc (3×10 mL). The combined organic layers were dried over MgSO₄, filtered, concentrated in vacuo, and recrystallized from MeCN to give a white solid.

Yield: 86 mg (80%); mp >220 °C; R_f 0.4 (cyclohexane–EtOAc, 50:50).

¹H NMR (CDCl₃): δ = 2.93 (s, 3 H, CH₃), 7.74 (d, *J* = 8.4 Hz, 2 H, Ar-H), 7.77 (s, 1 H, 7-H), 7.91 (d, *J* = 8.4 Hz, 2 H, Ar-H).

¹³C NMR (CDCl₃): δ = 10.0 (CH₃), 115.7 (q, *J* = 4.8 Hz, 7-C), 120.9 (q, *J* = 274.0 Hz, CF₃), 126.6 (CBr), 127.3 (q, *J* = 36.9 Hz, 8-C), 128.7 (Ar-C), 132.2 (1'-C), 132.8 (Ar-C), 138.6 (NCN), 148.4, 152.0 (3-C, 6-C).

¹⁹F NMR (CDCl₃): $\delta = -65.3$ (s).

6-(4'-Bromophenyl)-8-trifluoromethyl-1,2,4-triazolo[4,3-*b*]py-ridazin-3-one (9)

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To a solution of 3-hydrazinopyridazine 7 (85 mg, 0.25 mmol) in MeCN (1.5 mL), was added 1,1'-carbonyldiimidazole (83 mg, 0.50 mmol). The mixture was refluxed for 4 h. After cooling to r.t., the residue was washed with brine (2 mL). The organic layer was then extracted with EtOAc (3×3 mL), dried over Na₂SO₄, filtered, concentrated in vacuo, and recrystallized from MeCN.

Yield: 75 mg (84%); *R*_f 0.3 (cyclohexane–EtOAc, 60:40).

IR (KBr): 3135 (NH), 1724 (CO) cm⁻¹.

¹H NMR (DMSO-*d*₆): δ = 7.77 (d, *J* = 8.4 Hz, 2 H, Ar-H), 8.05 (d, *J* = 8.4 Hz, 2 H, Ar-H), 8.16 (s, 1 H, 7-H), 13.1 (br s, 1 H, NH).

¹³C NMR (DMSO-*d*₆): δ = 120.9 (q, *J* = 273.5 Hz, CF₃), 121.1 (q, *J* = 4.8 Hz, 7-C), 124.8 (CBr), 125.2 (q, *J* = 35.8 Hz, 8-C), 129.1, 132.1 (Ar-C), 132.5 (1'-C), 133.0 (NCN), 149.2, 149.3 (3-C, 6-C). ¹⁹F NMR (DMSO-*d*₆): δ = -64.1 (s).

LC-MS (ES+): m/z (%) = 361 [M + 1 + H⁺], 359 [M - 1 + H⁺].

HRMS: m/z calcd for $C_{12}H_6BrF_3N_4O [M + H]^+$: 358.9755; found: 358.9770.

3-Amino-6-(4'-bromophenyl)-8-trifluoromethyl-1,2,4-triazo-lo[4,3-*b*]pyridazine (10a)

A solution of 3-hydrazinopyridazine **7** (2.85 g, 8.6 mmol) and BrCN (1.09 g, 10.3 mmol) in EtOH (80 mL) was stirred at r.t. for 5 h. The mixture was then cooled with an ice bath and the pH adjusted to 10 by the addition of a soln of KOH (2 M). The precipitate formed was removed by filtration, washed with cold Et_2O , and recrystallized from MeCN to yield a yellow crystalline powder.

Yield: 2.52 g (82%); mp >220 °C.

IR (KBr): 3425 (NH₂), 3098, 1653, 1138 cm⁻¹.

¹H NMR (DMSO-*d*₆): δ = 6.97 (br s, 2 H, NH₂), 7.80 (d, *J* = 8.4 Hz, 2 H, Ar-H), 8.05 (s, 1 H, 7-H), 8.24 (d, *J* = 8.4 Hz, 2 H, Ar-H).

¹³C NMR (DMSO-*d*₆): δ = 115.1 (q, *J* = 4.9 Hz, 7-C), 121.4 (q, *J* = 273.6 Hz, CF₃), 124.7 (q, *J* = 35.6 Hz, 8-C), 125.0 (CBr), 129.4, 132.0 (Ar-C), 132.6 (1'-C), 134.6 (NCN), 149.9, 151.3 (3-C, 6-C).

¹⁹F NMR (DMSO- d_6): δ = -63.3 (s).

LC-MS (ES+): m/z (%) = 360 [M + 1 + H⁺], 358 [M - 1 + H⁺].

HRMS: m/z calcd for $C_{12}H_7BrF_3N_5$ [M + H]⁺: 357.9915; found: 357.9906.

3-Amino-6-(4'-bromophenyl)-8-trifluoromethyl-1,2,4-triazo-lo[4,3-*b*]pyridazine Dihydrochloride (10b)

Compound **10a** (2.52 g, 7.0 mmol) was stirred in HCl–*i*-PrOH (6 N; 10 mL) at r.t. for 20 min. Alcohol was then removed and the resulting crystals were washed with cold Et_2O (10 mL), filtered, and dried in vacuo.

Yield: 3.03 g (100%).

IR (KBr): 3420 (NH), 3039, 2655, 1694, 1349, 1146 cm⁻¹.

¹H NMR (DMSO- d_6): δ = 7.5 (m, 2 H, NH₂), 7.80 (d, J = 8.4 Hz, 2 H, Ar-H), 8.18 (s, 1 H, 7-H), 8.24 (d, J = 8.4 Hz, 2 H, Ar-H).

Anal. Calcd for $C_{12}H_9BrCl_2F_3N_5$: Cl, 16.4. Found: Cl, 16.2.

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