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From cells, to mice, to target: Characterization of NEU-1053 (SB-443342) and its analogs for treatment of human African trypanosomiasis.

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ABSTRACT

Human African Trypanosomiasis is a neglected tropical disease that is lethal if left untreated. Existing therapeutics have limited efficacy and severe associated toxicities. 2-(2-(((3-((1*H*-Benzo[d]imidazol-2-yl)amino)propyl)amino)methyl)-4,6-dichloro-1*H*-indol-1-yl)ethan-1-ol (NEU-1053) has recently been identified from a high throughput screen of >42,000 compounds as a highly potent and fast acting trypanocidal agent capable of curing a blood stream infection of *T. brucei* in mice. We have designed a library of analogs to probe the SAR and improve the predicted CNS exposure of NEU-1053. We report the activity of these inhibitors of *Trypanosoma brucei*, the efficacy of NEU-1053 in a murine CNS model of infection, and identification of the target of NEU-1053 via X-ray crystallography.

KEYWORDS

Trypanosoma brucei, methionyl-tRNA synthetase, medicinal chemistry

INTRODUCTION

Insect-borne trypanosomal diseases are a menace to human health. Human African trypanosomiasis (HAT, or sleeping sickness) is a neglected tropical disease caused by two subspecies of *Trypanosoma brucei* for which current therapeutics are toxic and inconvenient. Though there are two compounds in clinical trials, SCYX-7158¹ and fexinidazole,² given the clinical failure rate for infectious diseases,³ it is prudent to continue the search for new drugs.

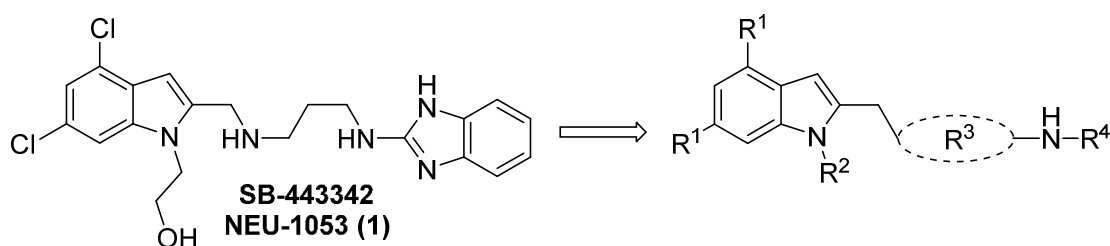
To that end, we recently described a high-throughput screening campaign performed as part of an industry-academic partnership between GlaxoSmithKline, Spanish National Research Council (CSIC), and Northeastern University, in which we uncovered 798 inhibitors of *T. brucei* cellular proliferation. We also reported *in vitro* drug metabolism,

physicochemical properties, and pharmacokinetics data, plus kinase selectivity data for key analogs.⁴ Included in that report was NEU-1053 (SB-443342, **1**), a singleton compound identified in the screen that showed rapid and irreversible proliferation inhibition of *T. brucei*, showed good plasma exposure, and cured a bloodstream infection in a *T. brucei rhodesiense* mouse model of HAT. As a singleton hit with no other analogs included in the screening campaign, there was no SAR information apparent from the HTS. Besides looking to better understand the SAR of this series, we wished to explore the various structural regions of the compound in order to identify effective analogs with more attractive physicochemical properties. We report those efforts here, definitively demonstrate the mechanism of action for **1**, and describe the results of an *in vivo* efficacy experiment for this compound in a murine model of Stage II HAT.

RESULTS

The overall strategy for the SAR exploration of **1** is shown in **Figure 1**. First, noting that the chlorine atoms on the indole provide substantial contribution to molecular weight and lipophilicity, we were interested to understand their importance. Second, the hydroxyethyl substituent on the indole nitrogen seemed to be a potential metabolic liability, and exploration of this region was needed. Third, the linker between the two aromatic systems in the molecule needed to be probed in terms of length, vector, and rigidity. Lastly, looking to reduce the size of the molecule, we wished to better understand the requirements of the 2-aminobenzimidazole eastern end by replacing the benzimidazole functionality. We first describe the synthesis of these analogs, and will then discuss the impact on potency and properties for this series.

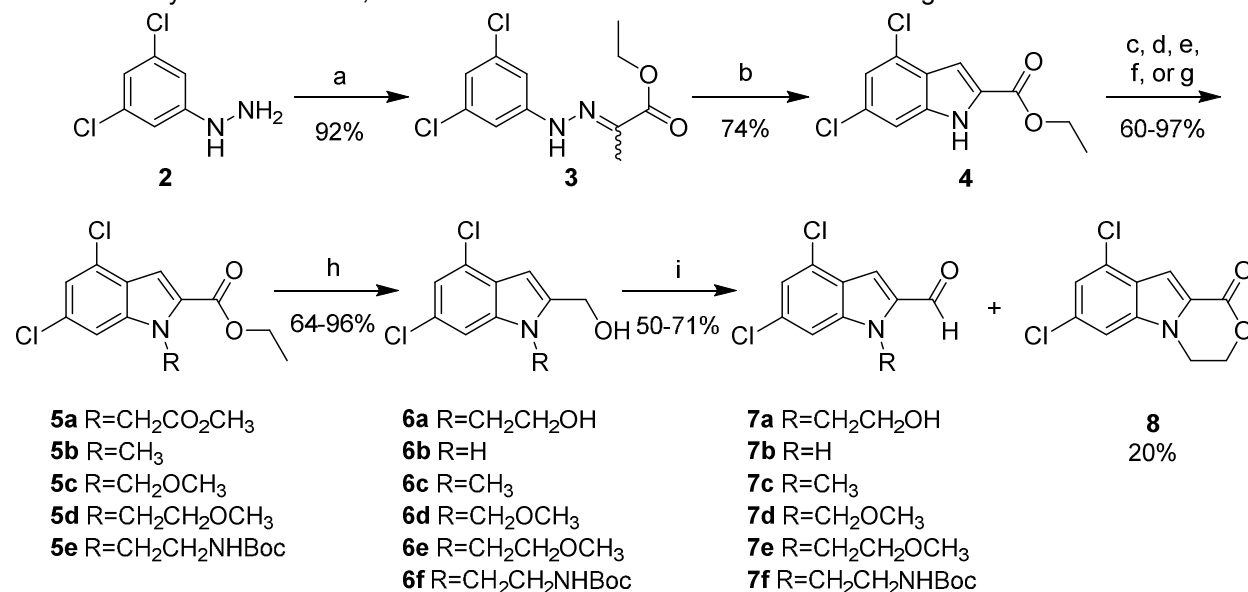
Figure 1. **1** SAR regions of interest



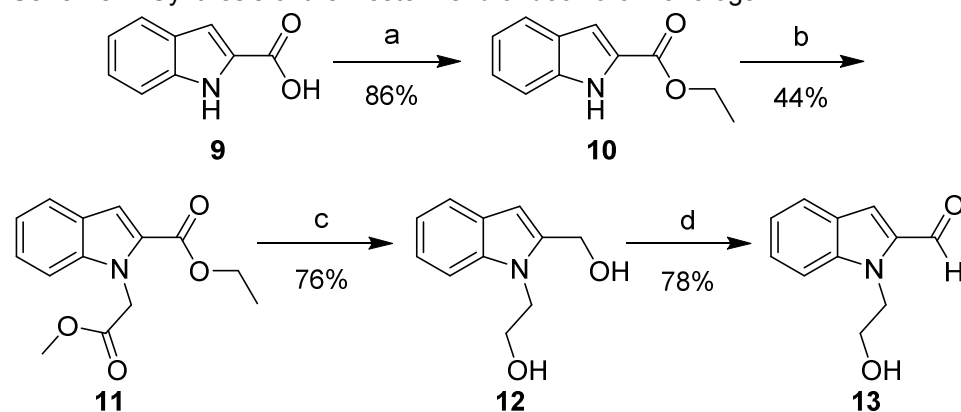
The synthesis of **1** and its 3,5-dichloroindole analogs commenced with the condensation of hydrazine **2** and ethyl pyruvate to yield a mixture of *E* and *Z* isomers of **3** (**Scheme 1**). Cyclization of **3** generated 3,5-dichloroindole **4**, which could be recrystallized from H₂O/ethanol. Alkylation with the appropriate alkyl halides or tosylates^{5, 6} and subsequent reduction with DIBAL gave corresponding *N*-substituted 2-(hydroxymethyl)indoles **6a-f** in good to high yields. Oxidation of the benzylic alcohol with freshly prepared MnO₂ following the Attenburrow procedure⁷ produced aldehydes **7a-f** in 50-71% yield. The oxidation of diol **6a** also generated lactone **8** in 20% yield, likely stemming from over-oxidation of the intermediate lactol. This material could be reverted to **6a** via reduction

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or used for further analog synthesis (*vide infra*). The non-halogenated **13** was synthesized in a similar manner from the commercially available indole **9** (**Scheme 2**).

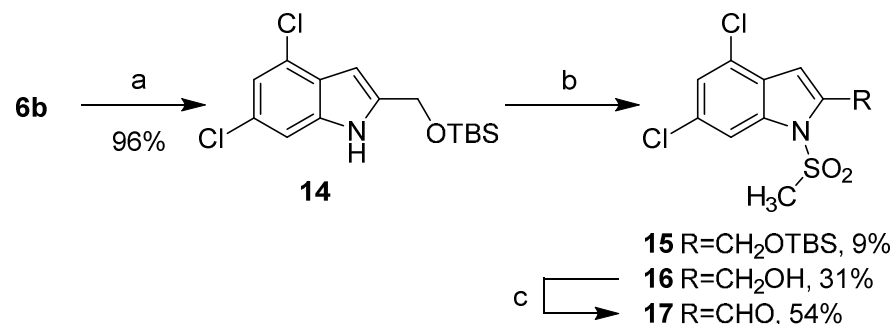
Scheme 1. Synthesis of the 3,5-dichloroindole western end of **1** and its analogs.

Reagents and conditions: (a) ethyl pyruvate, EtOH, reflux, 1 hr; (b) PPA, 130 °C, 10 min; (c) methyl bromoacetate, K₂CO₃, DMF, rt, o.n.; (d) NaH, DMF, 0 °C, 30 min; then CH₃I, 0 °C→rt, 3 hrs; (e) NaH, DMF, 0 °C, 30 min; then CH₃OCH₂CH₂Br, 0 °C→rt, 18 hrs; (f) NaH, DMF, 0 °C, 30 min; then CH₃OCH₂CH₂Br, 0 °C→rt, 18 hrs; (g) K₂CO₃, DMF, BocNH(CH₂)₂OTs, rt, o.n.; (h) DIBAL, THF, rt, 1 hr; (i) MnO₂, THF, rt, 2 hrs.

Scheme 2. Synthesis of the western end of dechloro-1 analogs.

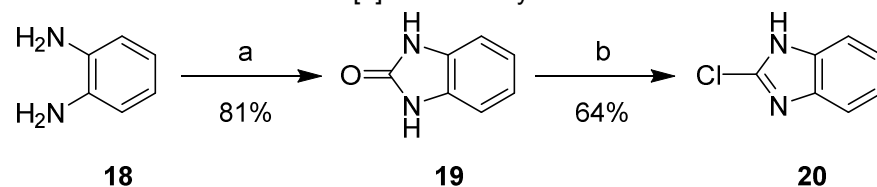
Reagents and conditions: (a) H₂SO₄, EtOH, reflux, o.n.; (b) methyl bromoacetate, K₂CO₃, DMF, rt, 3 hrs; (c) LiAlH₄, THF, 0 °C, 30 min, rt, 1.5 hrs; (d) MnO₂, CH₂Cl₂, rt, o.n.

The alcohol moiety of compound **6b** was protected as the TBS ether, and the sulfonylation of this compound was performed using the same protocol for **5c-f**; this reaction also effected desilylation, providing **16** in 31% yield (**Scheme 3**). The free alcohol was oxidized to the corresponding aldehyde with MnO₂ in modest yield.

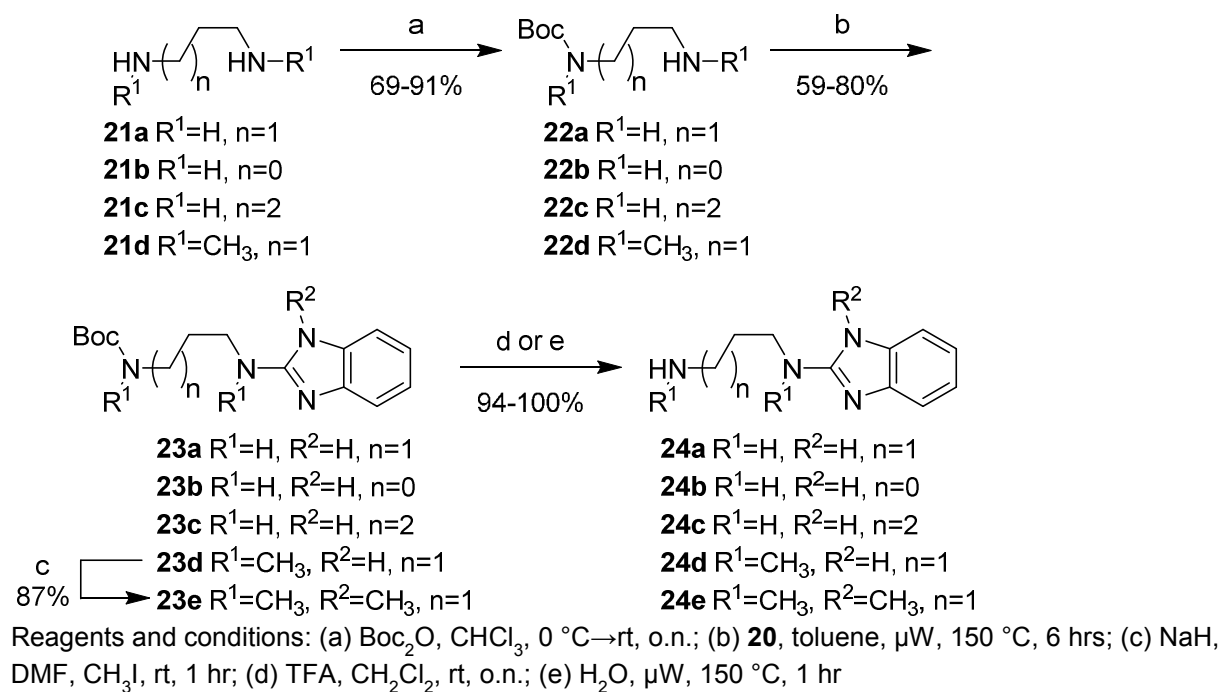
Scheme 3. TBS protected *N*-substituted indole synthesis.

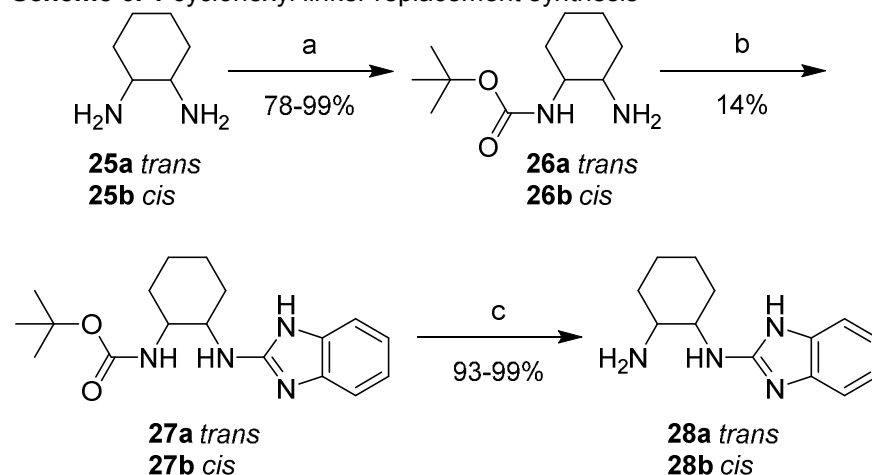
Reagents and conditions: (a) TBSCl, imidazole, CH₂Cl₂, rt, 1 hr; (b) NaH, DMF, rt, 30 min; CH₃SO₂Cl, rt, 1 hr; (c) MnO₂, CH₂Cl₂, rt, 2 hrs.

Aminoalkyl-2-aminobenzimidazole intermediates were prepared as shown in **Schemes 4-7**. Briefly, **20** was formed via heating of **18** in the presence of urea followed by deoxy-chlorination with POCl₃. Mono-Boc protected diamines **22a-d**, **26a-b**, **29**, and **32**, prepared according to known procedures⁸⁻¹³ or commercially available, were coupled to **20** using microwave irradiation at 150 °C. Removal of the Boc group was effected under either acidic conditions to yield the bis-TFA salt, or via microwave promoted thermolysis to give the free base. Attempts to couple **20** with **21a** directly were successful in generating product, though the purification of **24a** was complicated due to its high water solubility. This was overcome by the introduction a Boc group, allowing purification via extraction followed by chromatography. The isolation of **24** was accomplished by simple concentration of the reaction medium.

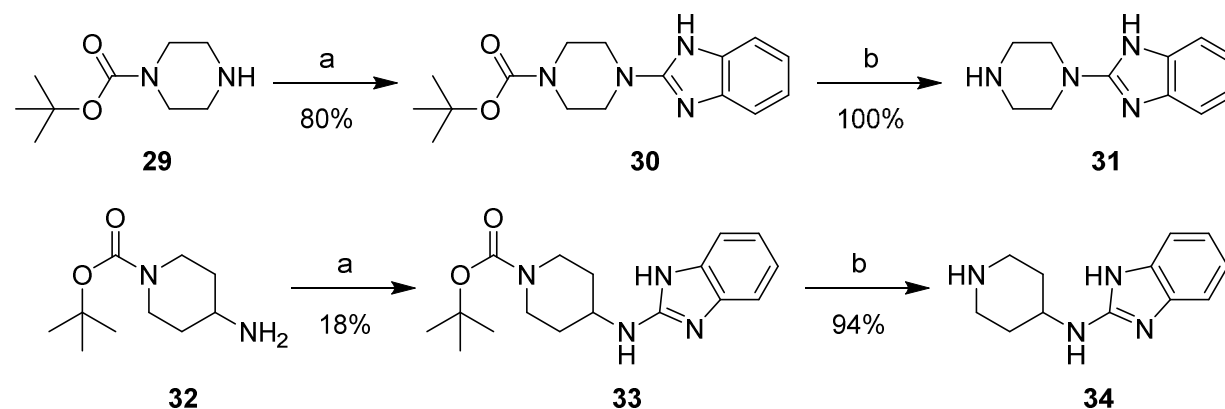
Scheme 4. Chloro-1*H*-benzo[d]imidazole synthesis

Reagents and conditions: (a) urea, (CH₂OH)₂, 140 °C, 1 hr, 170 °C, 7 hrs; (b) POCl₃, reflux, 2 hrs.

Scheme 5. Acyclic eastern half synthesis

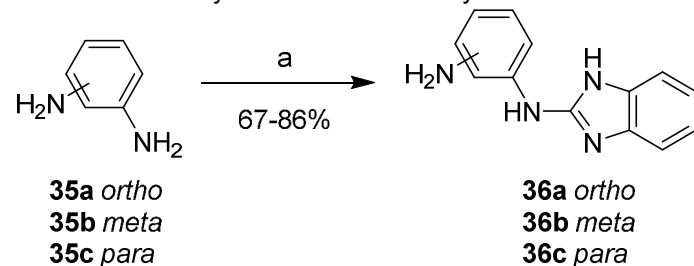
Scheme 6. 1 cyclohexyl linker replacement synthesis

Reagents and conditions: (a) Boc_2O , CH_2Cl_2 , $0\text{ }^\circ\text{C} \rightarrow \text{rt}$, o.n.; (b) **20**, toluene, μW , $150\text{ }^\circ\text{C}$, 6 hrs; (c) TFA, CH_2Cl_2 , rt, o.n.

Scheme 7. 1 cyclic linker analog synthesis

Reagents and conditions: (a) **20**, toluene, μW , $150\text{ }^\circ\text{C}$, 6 hrs; (b) TFA, CH_2Cl_2 , rt, o.n.

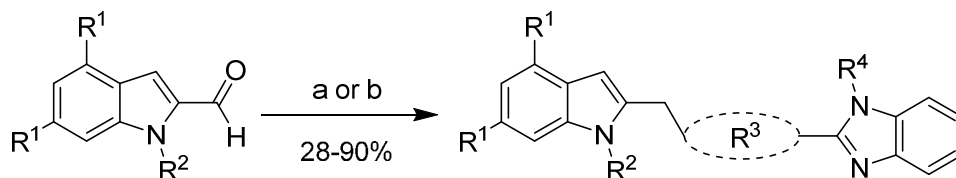
2-(aminophenyl)aminobenzimidazoles **36a-c** were prepared and purified without Boc protection under microwave heating (**Scheme 8**). Reaction times were reduced to 15 minutes by increasing the temperature from $150\text{ }^\circ\text{C}$ to $200\text{ }^\circ\text{C}$.

Scheme 8. Phenylenediamine linker synthesis

Reagents and conditions: (a) **20**, xylenes, μW , $200\text{ }^\circ\text{C}$, 15 min.

Reductive amination of 2-indolecarboxaldehydes **7a-f**, **13**, and **17** with **24a** was carried out with NaBH_3CN under either mildly basic conditions with the bis-TFA salt of **24a**, or mildly acidic conditions with the free base (**Scheme 9**, product structures in **Tables 1** and **3**). The same reaction conditions were used to couple **7a** with amines **24**, **28**, **31**, **34**, and **36** (**Table 4**).

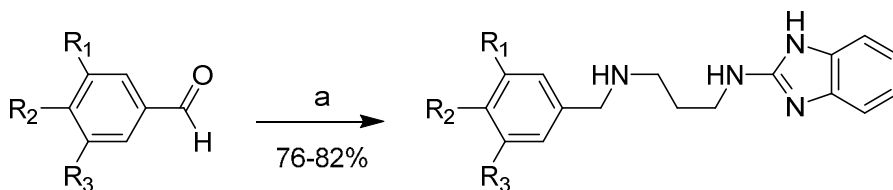
Scheme 9. Synthesis of **1** analogs via reductive amination



Reagents and conditions: (a) amine, KOAc, NaBH_3CN , CH_3OH ; (b) amine, AcOH, NaBH_3CN , CH_3OH

The synthesis of analogs **38**, substituting the dichloroindole scaffold with chlorophenyl rings, are shown in **Scheme 10**, via reductive amination from commercially available chlorobenzaldehydes **37**.

Scheme 10. Synthesis of chlorophenyl **1** analogs



37a $\text{R}^1=\text{R}^2=\text{H}$, $\text{R}^3=\text{Cl}$

37b $\text{R}^1=\text{R}^3=\text{H}$, $\text{R}^2=\text{Cl}$

37c $\text{R}^1=\text{R}^3=\text{Cl}$, $\text{R}^2=\text{H}$

38a $\text{R}^1=\text{R}^2=\text{H}$, $\text{R}^3=\text{Cl}$

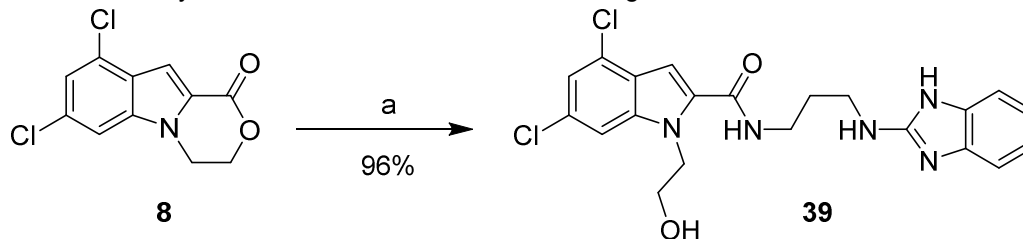
38b $\text{R}^1=\text{R}^3=\text{H}$, $\text{R}^2=\text{Cl}$

38c $\text{R}^1=\text{R}^3=\text{Cl}$, $\text{R}^2=\text{H}$

Reagents and conditions: (a) **24a**, AcOH, NaBH_3CN , CH_3OH

The side product **8** (**Scheme 1**) was used to produce **39** via ester aminolysis with the free base of **24a** and microwave irradiation (**Scheme 11**).

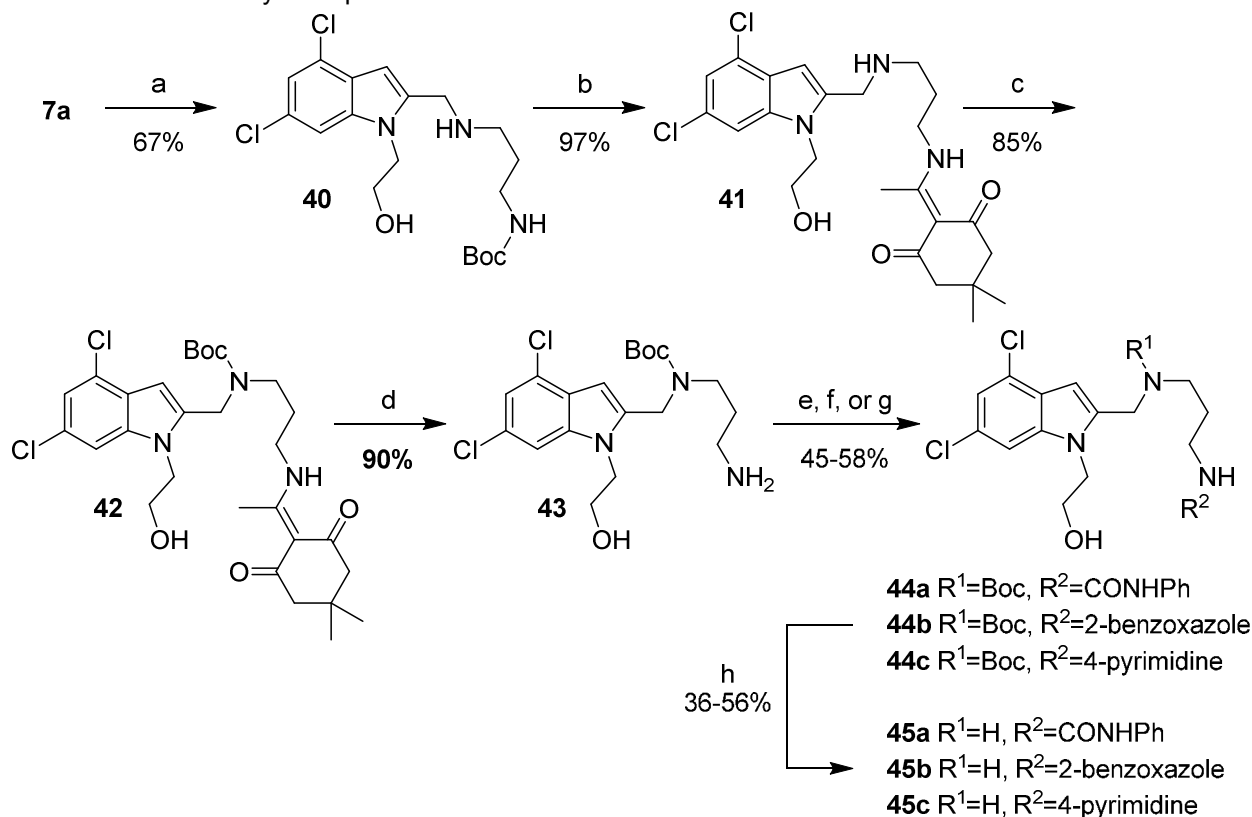
Scheme 11. Synthesis of **38** via ester-amide exchange



Reagents and conditions: (a) **24a**, DMF, μW , 150°C , 1 hr

To synthesize analogs varying the benzimidazole of **1**, aldehyde **7a** was first reductively aminated with **22a** (**Scheme 12**). Exchange of the Boc protecting group for the acid-stable N-(1-(4,4-dimethyl-2,6-dioxocyclohexylidene)ethyl) (Dde) protecting group provided **41** in high yield. The secondary amine was Boc protected and the enamine removed upon heating in the presence of hydrazine monohydrate. The free primary amine of **43** was then reacted with phenylisocyanate to provide **44a**, 2-chlorobenzoxazole (**44b**), or 4-chloropyrimidine (**44c**). Deprotection of the Boc-protected secondary amines under acidic conditions gave the final analogs in modest yields.

Scheme 12. Heterocyclic replacements on the eastern half of **1**



Reagents and conditions: (a) **22a**, AcOH, NaBH₃CN, CH₃OH, rt, 12 hr; (b) 3M aq. HCl, THF, rt, 1 hr, 80 °C, 1 hr; then DDE-OH, (*i*-Pr)₂NEt, CH₂Cl₂, CH₃OH, rt, 12 hr; (c) Boc₂O, K₂CO₃, CH₃CN, H₂O, rt, 12 hr; (d) H₂NNH₂·H₂O, CH₃CN, 45 °C, 3 hr; (e) PhCNO, CH₃CN, rt, 12 hr; (f) 2-chlorobenzoxazole, (*i*-Pr)₂NEt, CH₃CN, 80 °C, 12 hr; (g) 4-chloropyrimidine, (*i*-Pr)₂NEt, CH₃CN, μW, 150 °C, 2 hr; (h) 4M HCl in 1,4-dioxane, rt, 6 hr.

***In vitro* assessment against *T. brucei*.**

Indole halogens. Removal of both chlorides on the indole ring resulted in an 78x loss in potency (**46**, EC₅₀: 75 nM, **Table 1**). A significant reduction in MW (432→364) and clogP (3.60→2.39) was effected however, resulting in the highest CNS

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multiparameter optimization (MPO) score¹⁴ among these analogs. CNS exposure is essential to effectively treat stage II HAT when the parasites have invaded the CNS and crossed the blood brain barrier (BBB). Noting that a CNS MPO score ≥ 4 is suggestive of CNS penetration, **46** score was calculated to be 4.3. This compound also retained a high lipophilic ligand efficiency (LLE, pEC_{50} -cLogP) of 4.71 (the desirable range for LLE is ≥ 4).¹⁵ Replacement of the aminomethyl group of **1** with an amide (**39**) resulted in a 520-fold loss in activity. This correlates well with the conclusions from the linker SAR in that the basicity of the dialkylamine is essential for high potency.

Table 2 shows the results of replacing the dichloroindole scaffold with simplified chlorophenyl rings. Compound **38c**, retaining the 3,5-dichloro motif showed only a slight reduction in potency (EC_{50} =16 nM). Removal of one of the chlorine atoms (**38a**) resulted in an 8-fold loss in activity (EC_{50} =0.15 μ M) compared to **38c** and migration of the chloride to the *para* position (**38b**) resulted in a further 6-fold loss in activity (EC_{50} =0.96 μ M). Our observation that the indole moiety in this class of compounds can be replaced with a simple dichlorophenyl group may provide a useful path forward in the event that toxicity due to indole metabolism is observed.

Indole N-substituents. Modifications to the 2-hydroxyethyl group of **1** displayed mixed results (**Table 3**). Truncations (**47a**, **47b**) or a shift of the oxygen atom to an internal position (**47c**) resulted in less than a 10-fold loss in activity. Larger groups (**47d**, **47e**, **47f**) comparatively caused a far larger loss in activity ($>300\times$). Replacement of the alcohol with an amine (**47g**) led to around a 200-fold loss in activity. MPO scoring for these compounds was similar ($\leq \Delta 0.3$) to **1** except for the large increase resulting from addition of the methanesulfonamide group (**47e**, MPO=4.1) and large decrease apparent in the Boc-protected **47f** (MPO=1.5). Generally, the smaller indole *N*-substituents remained highly active while those that increased steric bulk resulted in a drop in potency by 2-3 orders of magnitude.

Linker. Contracting or extending the diamino linker by a methylene unit (**48a** and **48b**) did not lead to a loss in potency (EC_{50} =5.0 nM each, **Table 4**) compared to **1**. Compound **48a** did however possess a slightly improved MPO score (3.8). Capping both amines of the linker with methyl groups was not tolerated (**48c**, EC_{50} =0.98 μ M). The trimethylated compound (**48d**) showed approximately an additional 3-fold loss in activity (EC_{50} =2.2 μ M).

The tolerance for a chain of 2-4 carbons between amines suggests adequate space within the biological target(s) of action for conformational variations of the linker. The *para*-, *meta*-, and *ortho*- phenylenediamine linkers, designed to restrict the conformation of the linker, all showed a dramatic loss in activity. The *ortho*-linked **48e** was the most active (EC_{50} =0.42 μ M), followed closely by the *meta*-linked **48f** (EC_{50} =0.96 μ M). The *para*-linked compound, **48g**, was an additional order of magnitude less active (EC_{50} =5.5 μ M). These data suggested either that (1) a bent orientation of the linker is closest to the active conformation of **1**, or (2) the basicity of these nitrogens is important.

Compounds **48h-k** further probed the tolerance for reduced linker flexibility and diamine orientation. Three of the four saturated cyclic linked analogs, **48h-j**, remained sub-micromolar in activity. Compounds **48h** and **48i** were within 2-fold of one another,

showing no preference for *cis* or *trans* substitution of the cyclohexyl ring, and were essentially equipotent to the aromatic analog **48e**. Compound **48k**, bearing a piperazinyl linker, was $>5\ \mu\text{M}$, compared to the 4-aminopiperadinyllinked **48j** ($\text{EC}_{50}=0.40\ \mu\text{M}$). This difference in activity is likely not a result of linker reduction as **48a** retained essentially all activity compared to **1**. Rather, taken together with the poor activity of **48c** and **48d**, the reduced activity of **48k** illustrates the importance of (1) the hydrogen bonding donor motif and (2) a high degree of flexibility in the linker for high potency.

Benzimidazole replacements. Any replacements for the benzimidazole ring that lacked a hydrogen bond donor were not tolerated (**Table 5**, **45b** and **45c**). Specifically, changing to an isosteric benzoxazole (**45b**) led to around an 800-fold loss in activity compared to **1**. Compound **45c**, also devoid of a hydrogen bond donating 4-pyrimidine ring, possessed activity approximating **45b** ($0.79\ \mu\text{M}$ vs. $0.27\ \mu\text{M}$). Reintroduction of a hydrogen bond donor with a urea N-H (**45a**), which could recapitulate that of the benzimidazole, regained the high potency observed for **1**.

All analogs of **1** were screened against MRC5-SV2 human lung cells to assess toxicity. Most compounds possessed a selectivity index in the 10-50x range. Compounds in the single digit nanomolar range against *T. brucei* displayed a higher selectivity index in the range of 900-7,000x.

Physicochemical and ADME properties. The calculated physicochemical properties of compounds **1**, **38a-c**, **39**, **45a-c**, **46**, **47a-g**, and **48a-k** are presented in **Table 6**, color coded in terms of desirability. Compound **47e** is noteworthy among analogs replacing the 2-hydroxyethyl group for its high MPO score above 4.0, due to its reduced clogP and pK_a (calculated using JChem for Excel, Chemaxon, Inc.) Analogs **48g**, **48f**, **48e**, **48h**, and **48i** bearing phenylenediamine and 1,2-diaminocyclohexyl linkers possess poor MPO scores as a result of high molecular weights, clogP, and clogD values.

An excellent MPO score (4.3) is calculated for **46**, which resulted from reduced molecular weight compared to **1** (363.5, $\Delta 68.9$), clogP (2.39, $\Delta 1.21$), and clogD (0.36, $\Delta 1.23$); this results simply from the removal of the chlorine atoms from the indole. In addition to the improved MPO scores of many compounds in this series, several compounds retained LE values ≥ 0.30 and LLE values ≥ 4.0 , indicating a good balance of size and lipophilicity.

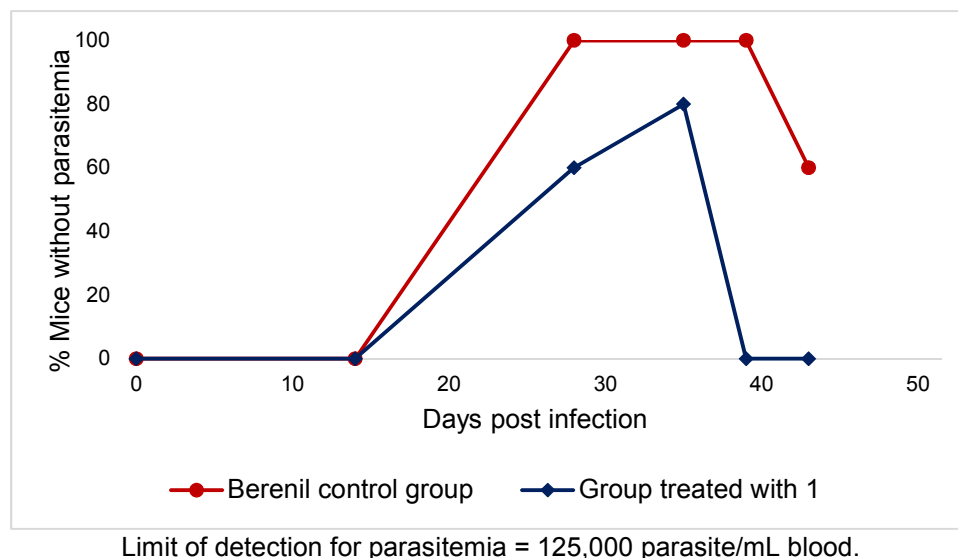
In vitro absorption, distribution, metabolism, and excretion (ADME) properties were collected for several compounds and are tabulated in **Table 7**. Few compounds improved the aqueous solubility to an appreciable extent, with **46** as the only significant exception at $298\ \mu\text{M}$. Compound **46** was also the only analog to show reduced human plasma protein binding (PPB) to $<99\%$, or to reduce the $\log D_{7.4}$ below 3.0 among those tested. Of the 16 compounds assessed for human liver microsome clearance (HLM Cl_{int}), only 4 analogs possessed lower clearance than **1**. Compounds **47a** and **47b** displayed the lowest human microsomal clearance, implicating the 2-hydroxyethyl group may indeed be a metabolic handle as we expected. Others with high microsomal clearance are **47d**, **47e**, **47f**, **48c**, **48e**, and **48h**; the Boc protected analog **47f** and the dimethylamino analog **48c** both showing rapid metabolism (142 and $135\ \mu\text{L}/\text{min}/\text{mg}$ respectively). Similarly, few compounds possessed good rat hepatocyte clearance, with

only **47g** ($Cl_{int} = 7.96 \mu\text{L}/\text{min}/10^6$) and **48k** ($Cl_{int} = 12.8 \mu\text{L}/\text{min}/10^6$) showing any significant improvement over **1**.

Pharmacokinetics study. Though we previously reported plasma pharmacokinetics, we did not determine CNS drug levels. Thus, female BALB/c mice were injected with a single, 10 mg/kg intraperitoneal dose of **1**, and plasma and brain levels were measured, showing a brain-to-plasma ratio of 0.39, with drug levels of 31.6 ng/g of brain tissue measured at the 24 hour time point. (The PK parameters are summarized in the Supporting Information, **Tables S1-S3** and **Figure S1**). With this evidence in hand, we proceeded to assess this compound in a mouse model of Stage II HAT.

Murine CNS infection model. We advanced **1** to a mouse model of a CNS infection with *T. b. brucei* (GVR 35 strain). On day 0, 2 groups of 5 mice each were infected with *T. b. brucei* and checked for parasitemia on day 14. One group, taken as the control, was treated with a single dose of 40 mg/kg Berenil on day 21. From days 22 to 25 the control group was treated with the vehicle DMSO in PBS. The other group of mice was treated with **1** at a dose of 20 mg/kg/day on days 21-25. Following a 2-day hiatus, treatment with **1** was resumed on days 28-32. Parasitemia was checked from both groups via examination of tail blood on days 14, 28, 35, 39, and 43. The results are reported in **Figure 2** and **Table S-4**.

Figure 2. Mice without parasitemia in a mouse model of CNS infection with *T. b. brucei*

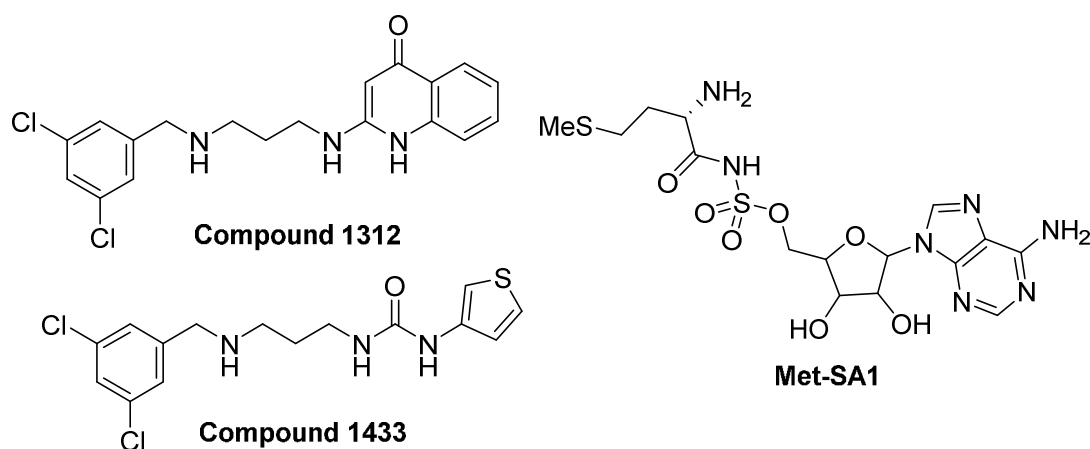


On day 28, 2 days after the 1st regimen of treatment was completed, 3/5 mice treated with **1** showed no detectable parasitemia. Following the 2nd round of treatment, this number increased to 4/5 mice showing no detectable parasitemia on day 35. When checked on day 39 however, all mice treated with **1** showed a full relapse of parasites in the blood. The control group comparatively showed no detectable levels of parasitemia when checked on days 28, 35, and 39 in all 5 mice, and only 2 mice showing a relapse on day 43. Since recrudescence was observed even in animals where the bloodstream infection was cleared by **1**, we interpret this result to suggest that compound **1** does not

have sufficient antiparasitic activity in the brain to cure Stage 2 HAT. This may be attributed in part to the high PPB (>99.9%) and rapid clearance (rat hepatocyte Cl_{int} =24.9 μ L/min/ 10^6 cells) of **1**.

Target Identification. Though the potent activity of **1** was uncovered from a set of compounds selected on the basis of their likely kinase inhibition activity, we noted that the structure of **1** was quite similar to a class of previously reported *T. brucei* methionyl-tRNA synthetase (MetRS) inhibitors (such as compound **1312**, **Figure 3**).¹⁶ *T. brucei* has a single MetRS and is necessary for cell growth as previously reported using RNA interference.¹⁶ We wished to determine whether **1** inhibited *T. brucei* MetRS.

Figure 3. Structures of previously published *T. brucei* MetRS inhibitors (**1312**¹⁶, **1433**¹⁷, and **Met-SA1**¹⁸)



Biochemical assessment. A functional assay for the *T. brucei* MetRS enzyme based on ATP-depletion was previously described.¹⁹ Compound **1** was determined to be a highly potent inhibitor, with an IC_{50} below the lower detection limit of the assay (<25 nM) (**Figure 4**). The positive control compound, **Met-SA1**¹⁸ **Figure 3**, is also a potent inhibitor with an IC_{50} of <25 nM.

Figure 4. *T. brucei* MetRS inhibition assay. IC_{50} s of both the positive control, **Met-SA1**, and compound **1** are <25 nM (the lower detection limit of the assay as described in the results and discussion). Error bars represent standard error of the mean of three replicates. Pentamidine, a negative control, had no *T. brucei* MetRS inhibition up to 10 μ M (results not shown).

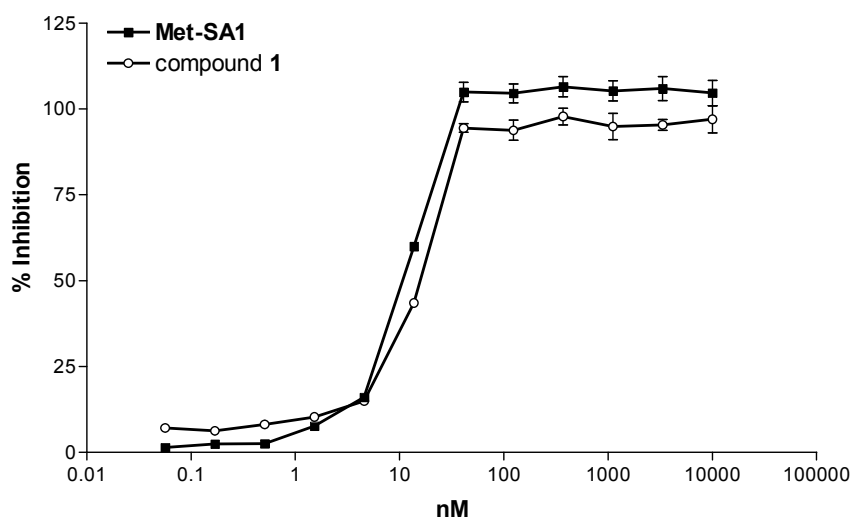
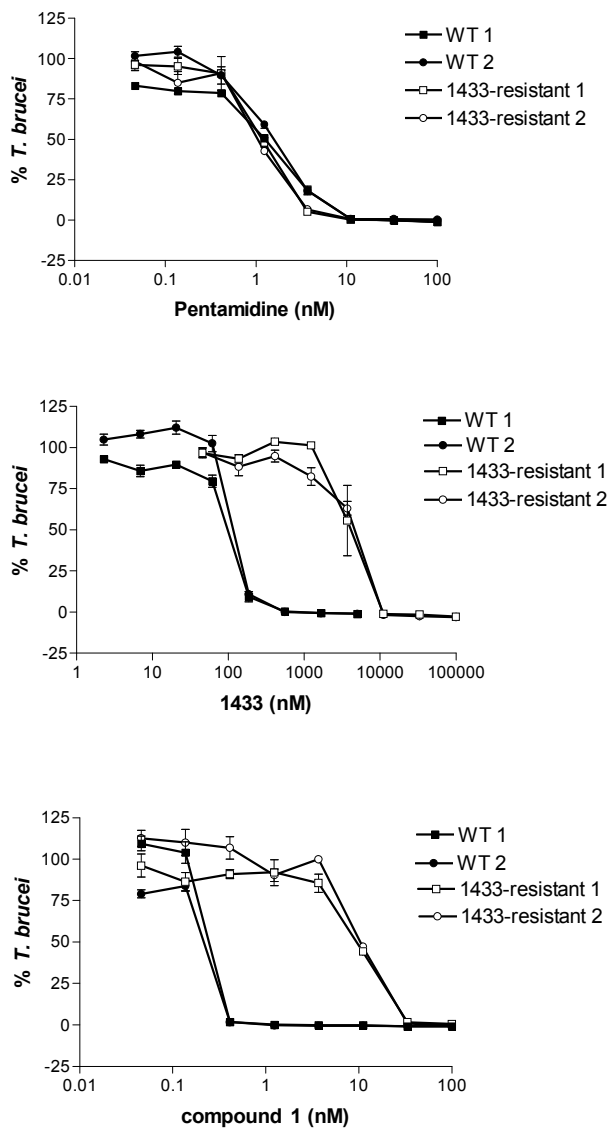


Figure 5. Wild-type and **1433**-resistant *T. brucei* growth inhibition assays. Results from two independent assays, as indicated with the 1 and 2. Error bars represent standard error of the mean of three replicates.



Potency shift in resistant parasites. To further test if compound **1** is targeting the *T. brucei* MetRS, a *T. brucei* growth inhibition cell assay was conducted with a strain that is resistant to MetRS inhibitors. The resistant strain was generated by serial *in vitro* passage of wild-type *T. brucei* with the MetRS inhibitor **1433** (**Figure 3**) and was shown to have ~35-fold upregulated expression of MetRS mRNA.¹⁸ The EC₅₀ of **1** and **1433** are ~107-fold and ~48-fold less potent against the **1433**-resistant strain compared to the

wild-type (**Figure 5** and **Table 8**), consistent with the hypothesis that **1** acts by inhibiting the MetRS enzyme. Pentamidine does not inhibit *T. brucei* MetRS (data not shown) and does not show a difference in EC₅₀ values between the **1433**-resistant and wild-type strains.

Structural biology. Even though crystals of TbruMetRS•Met would disintegrate after ~20 seconds in contact with a solution of **1**, it was possible to replace the bound methionine with **1** in the binding site using short soaking times.²⁰ The structure in complex with **1** was determined at 2.6 Å resolution and deposited in the Protein Data Bank with PDB identifier 5TQU. Data collection and refinement statistics are shown in **Table S5** (supporting information) and the electron density map for the compound is shown in **Figure 6A**. Compound **1** binds to TbruMetRS in a manner similar to previously-reported compound-bound structures.^{20, 21} Characteristically, two pockets are involved in binding of the inhibitor (**Figure 6B**): the enlarged methionine pocket (EMP), formed mainly by hydrophobic residues including the ones engaged in methionine binding, and the “auxiliary pocket” (AP), not present in the methionine bound state.²⁰ The EMP, formed mainly as a result of the movement of TbruMetRS residues Val473, Trp474 and Phe522, is occupied by the 4,6-dichloroindolyl-ring moiety which also interacts with residues Ile248 and Tyr481. The AP is occupied by the benzimidazole moiety which interacts with His289, Gly290 and Val473, while establishing a hydrogen bond with Asp287 through one of the nitrogens of the benzimidazole ring with a distance of 3.0 Å. The flexible linker of **1** forms hydrophobic interactions with the aromatic group of residue Tyr250 and a hydrogen bond through the exocyclic amine of the benzimidazole moiety with the catalytic Asp287 with a distance of 2.8 Å (**Figure 6C**). In a novel interaction pattern for TbruMetRS inhibitors, the OH group in the hydroxyethyl substituent of the 4,6-dichloroindolyl-ring moiety makes a hydrogen bond with residue Asp287 with a distance of 2.6 Å. As a result, Asp287 plays a central role in establishing polar contacts with the three different regions of **1**: the linker, the benzimidazole moiety and the OH group in the hydroxyethyl moiety. Furthermore, at least one water molecule is part of what seems to be a crucial hydrogen bonding network involving Asp287 and **1** (**Figure 6C**).

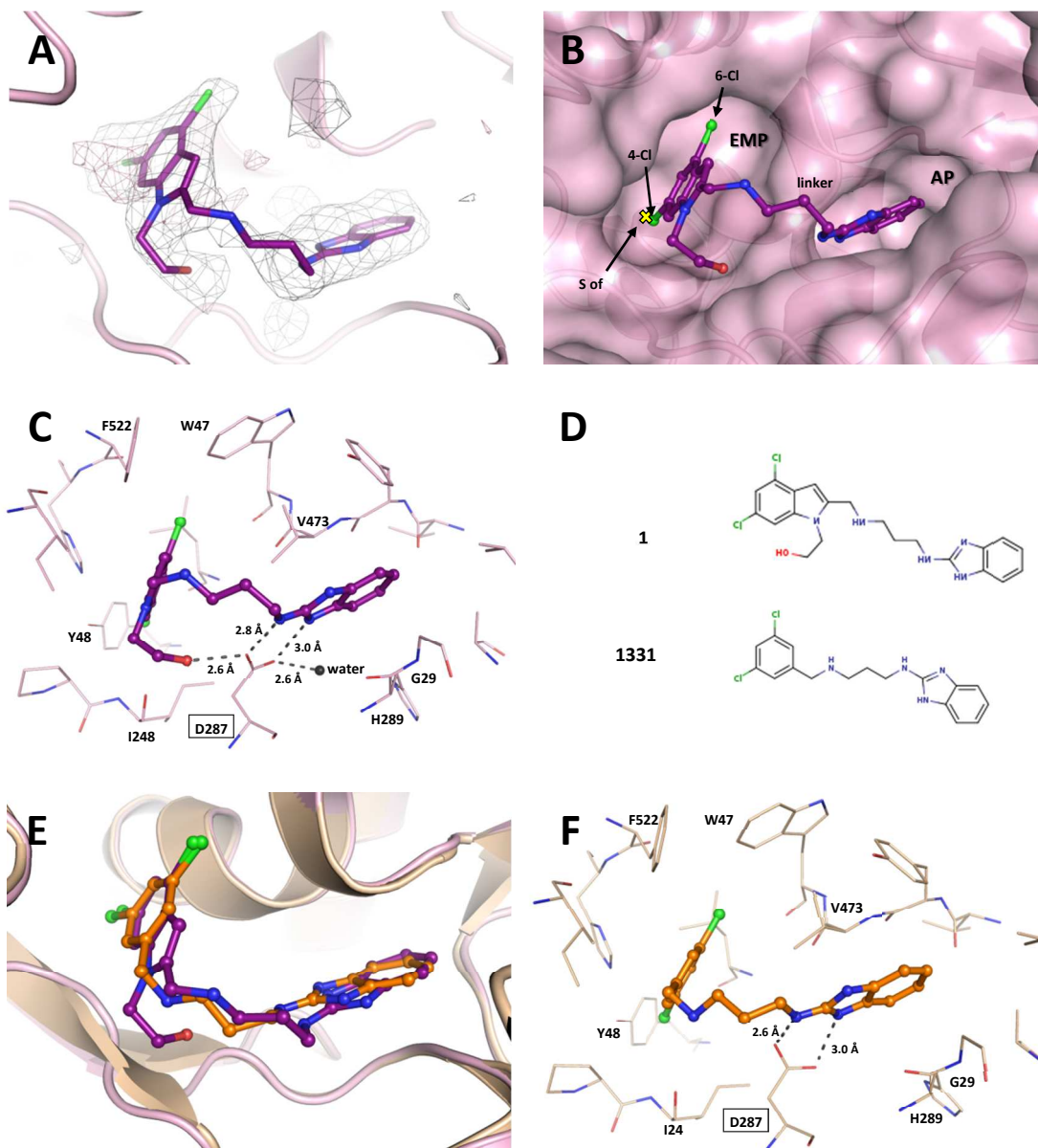
When comparing the new structure to previously reported TbruMetRS complexes with inhibitors bound, the overall binding mode of **1** is quite similar. Taking the TbruMetRS•**1331** structure (**Figure 6D**) as a representative of previously determined inhibitor bound structures, it appears that the two ring systems of each inhibitor occupy similar positions. For instance, the two chlorine atoms are only 0.8 and 0.3 Å (Cl-4 and Cl-6, respectively) apart in the two structures (**Figure 6E**) and the linkers follow a similar path. The key difference is the extra interaction of the hydroxyl group of **1** with Asp287 (compare **Figures 6C** and **6f**), representing one additional hydrogen bond between the inhibitor and the target enzyme.

CONCLUSIONS

In summary, we report the structure-activity relationships for the various regions of **1**, and described its activity in a mouse model of CNS infection of *T. brucei*. In addition, the

compound's mechanism of action via *T. brucei* MetRS was conclusively elucidated by a combination of biochemical, cellular, and crystallographic techniques. Despite the excellent *in vitro* anti-parasitic potency of **1**, the lack of CNS activity in the late-stage model of HAT necessitates further optimization of CNS permeability and pharmacokinetics; this work is ongoing.

Figure 6. Binding of **1** to TbruMetRS. (A) The TbruMetRS•**1** structure (PDB: 5TQU) with the difference electron density map calculated by omitting the inhibitor, contoured at the 3σ level (positive density in grey, negative density in red). (B) General features of the **1** binding mode. The position of the methionine sulfur atom in the TbruMetRS•methionine structure (PDB: 4EG1)²⁰ is shown with a yellow cross. The protein surface and the two pockets, EMP and AP, where the inhibitor is bound are shown. (C) Hydrogen bond network in the TbruMetRS•**1** structure. The label for residue Asp287 is highlighted with a rectangle. The hydroxyethyl substituent of the 4,6-dichloroindolyl-ring moiety establishes an extra hydrogen bond with residue Asp287. (D) Chemical structures of compound **1** and the previously reported TbruMetRS inhibitor 1331. (E) Superposition of compounds **1** and 1331 (PDB: 4EG7)²⁰ bound to TbruMetRS. (F) Hydrogen bond interactions between inhibitor and enzyme in the TbruMetRS•1331 structure.



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work benefitted from remote access to resources at the Stanford Synchrotron Radiation Lightsource supported by the U.S. Department of Energy Office of Basic Energy Sciences under Contract No. DE-AC02-76SF00515 and by the National Institutes of Health (P41GM103393).

SUPPORTING INFORMATION

Experimental details for the pharmacokinetics experiments, chemical syntheses, X-ray crystallography, and *in vitro* and *in vivo* biological methods are available in the Supporting Information. In addition, annotation of compounds with NEU registry numbers and SMILES strings are provided. This information is available free of charge via the Internet at <http://pubs.acs.org/>.

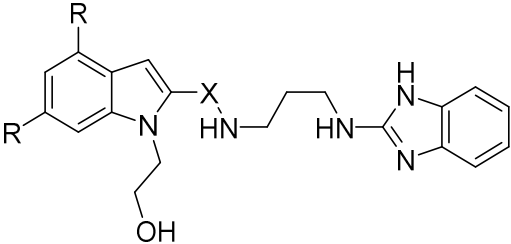
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TABLES

Table 1. Activity of modified indole analogs of 1 against *T. brucei*



Compound	R	X	<i>T. brucei</i> EC ₅₀ (nM) ^a	MRC5-SV2 TC ₅₀ (μM) ^b	Selectivity (TC ₅₀ /EC ₅₀)
1	Cl	CH ₂	0.96	12	
39	Cl	CO	500	18	36
46	H	CH ₂	75	27 ^c	360

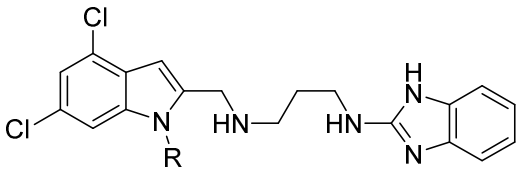
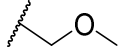
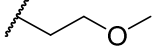
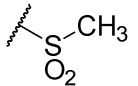
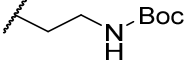
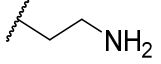
^a SD <0.20. ^b SD <1.5 unless otherwise noted. ^cSD=9.3

Table 2. Activity of chlorophenyl analogs of **1** against *T. brucei*

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Compound	R	<i>T. brucei</i> EC ₅₀ (nM) ^a	MRC5-SV2 TC ₅₀ (μM) ^b	Selectivity (TC ₅₀ /EC ₅₀)
38a	<div></div>	150	22	147
38b	<div></div>	960	22	23
38c	<div></div>	16	22	1375

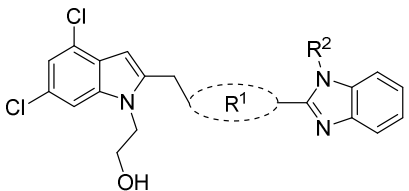
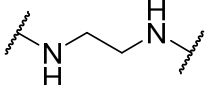
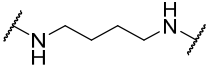
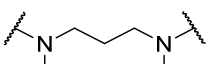
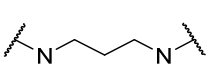
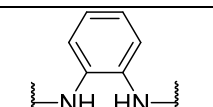
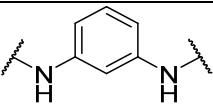
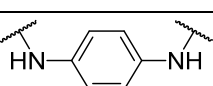
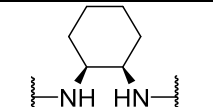
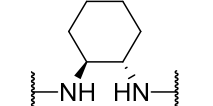
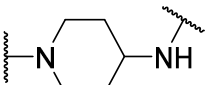
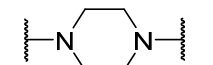
^a SD <0.20. ^b SD <3.0.

Table 3. Activity of indole N-substituted analogs of **1** against *T. brucei*

				
Compound	R	<i>T. brucei</i> EC ₅₀ (nM) ^a	MRC5-SV2 TC ₅₀ (μM) ^b	Selectivity (TC ₅₀ /EC ₅₀)
47a	H	9.0	8.8	978
47b	CH ₃	5.0	9.8	1960
47c		8.0	7.4	925
47d		620	10	16
47e		340	1.4	4.1
47f		790	8.1	10
47g		200	8.4	42

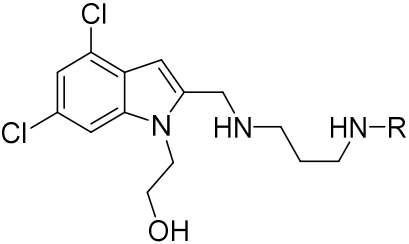
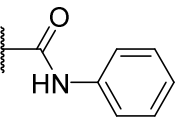
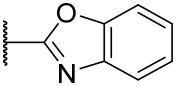
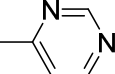
^a SD <0.20. ^b SD <3.0.

Table 4. Activity of linker analogs of **1** against *T. brucei*

					
Compound	R ¹	R ²	<i>T. brucei</i> EC ₅₀ (nM) ^a	MRC5-SV2 TC ₅₀ (μM) ^b	Selectivity (TC ₅₀ /EC ₅₀)
48a		H	5.0	26 ^c	5200
48b		H	5.0	11	2200
48c		H	980	22	22
48d		CH ₃	2200	10	4.5
48e		H	420	25	60
48f		H	960	21	22
48g		H	5500	23	4.2
48h		H	680	14	21
48i		H	340	17	50
48j		H	400	12	30
48k		H	>5000	50	--

^a SD <0.55. ^b SD <3.0 unless otherwise noted. ^c SD=8.8

Table 5. Activity of chlorophenyl analogs of **1** against *T. brucei*

<div></div>				
Compound	R	<i>T. brucei</i> EC ₅₀ (nM) ^a	MRC5-SV2 TC ₅₀ (μM) ^b	Selectivity (TC ₅₀ /EC ₅₀)
45a	<div></div>	2.0	14	7000
45b	<div></div>	790	26	33
45c	<div></div>	270	>50	--

^a SD <0.20. ^b SD <3.0.

Table 6. Physicochemical properties of 1 and its analogs

Cmpd	MW (g/mol) ^a	clogP ^{a,c}	clogD ^{a,d}	H-bond donors ^a	pKa ^{b,c}	TPSA (Å ²) ^{b,c}	MPO Score ^e	LE	LLE
1	432.4	3.60	1.59	4	9.29	82.48	3.5	0.31	5.57
38a	314.8	3.44	1.57	3	9.15	57.32	4.4	0.31	3.38
38b	314.8	3.44	1.53	3	9.19	57.32	4.4	0.27	2.49
38c	349.3	4.04	2.33	3	8.98	57.32	4.0	0.33	3.65
39	446.3	3.11	2.98	4	6.98	94.97	3.7	0.21	3.19
45a	435.4	3.28	1.36	4	9.34	82.90	3.7	0.30	5.42
45b	433.3	3.66	1.78	3	9.29	79.83	3.7	0.21	2.45
45c	394.3	2.30	0.42	3	9.29	79.58	4.3	0.25	4.27
46	363.5	2.39	0.36	4	9.31	82.48	4.3	0.26	4.71
47a	388.3	4.06	2.06	4	9.28	73.11	3.6	0.31	4.04
47b	402.3	4.29	2.28	3	9.29	62.25	3.4	0.31	4.11
47c	432.4	4.35	2.36	3	9.27	71.48	3.2	0.28	3.75
47d	446.4	4.24	2.23	3	9.29	71.48	3.2	0.21	1.96
47e	466.4	2.88	2.08	3	8.01	96.39	4.1	0.22	3.59
47f	531.5	4.98	2.97	4	9.29	100.58	1.5	0.17	1.12
47g	431.4	3.49	-0.73	4	9.83	89.89	3.3	0.23	3.21
48a	418.3	3.54	1.88	4	8.97	82.48	3.8	0.30	4.76
48b	446.4	4.11	1.82	4	9.60	82.48	3.0	0.28	4.39
48c	460.4	4.62	3.43	2	8.54	61.52	3.0	0.20	1.48
48d	474.4	4.84	3.66	1	8.54	50.66	3.0	0.18	0.81
48e	466.4	5.34	5.07	4	7.38	79.15	2.2	0.20	1.06
48f	466.4	5.34	5.06	4	7.39	79.15	2.2	0.19	0.76
48g	466.4	5.34	5.04	4	7.42	79.15	2.2	0.16	-0.14
48h	472.4	4.95	2.94	4	9.35	82.48	2.1	0.20	1.51
48i	472.4	4.95	2.94	4	9.35	82.48	2.1	0.19	1.22
48j	458.4	3.93	3.61	3	7.38	70.31	3.2	0.21	2.47
48k	444.4	4.38	4.30	2	6.73	60.32	3.2	0.18	0.92

^a Values that are within the MPO more desired ranges are green and outside the less desired range are red. ^b Calculated using ChemAxon JChem for Excel. ^c Calculated at pH 7.4 using ChemAxon JChem for Excel. ^d Calculated using MPO calculator previously reported.¹⁴

Table 7. ADME measurements of **1** and analogs

Compound	Aqueous Solubility (uM)	Human PPB (%)	HLM Cl_{int} (μ L/min/mg)	HLM $t_{1/2}$ (min)	Rat Hepatocyte Cl_{int} (μ L/min/ 10^6)	Rat Hepatocyte $t_{1/2}$ (min)	logD 7.4
1	6.33	> 99.9	24	*	24.9	*	4.0
46	298	98.61	22.6	*	81.4	*	2.0
47a	4.02	99.8	15.5	*	17.2	*	4.3
47b	12.2	99.8	12.9	*	29.3	*	4.5
47c	< 1	99.9	34.2	*	33.6	*	4.0
47d	3.77	> 99.9	43.8	*	49.7	*	4.1
47e	19	*	80.9	8.6	98.8	7	*
47f	32	*	142	4.9	33.3	20.8	*
47g	*	*	*	*	7.96	87.1	*
48a	21	99.8	36	*	30.7	*	4.3
48b	4.26	99.9	22.5	*	17	*	3.7
48c	11.7	> 99.9	135	*	61.7	*	4.0
48d	*	*	*	*	> 300	< 2.3	*
48e	*	> 99.9	51.6	*	100	*	4.9
48f	< 1	> 99.9	33.9	*	22.5	*	4.3
48g	1.39	> 99.9	27.3	*	18.7	*	4.7
48h	11	> 100	64.2	*	115	*	4.0
48j	0.3	100	*	*	38.8	*	4.4
48k	< 1	99.0	21.1	*	12.8	*	4.9

* Not obtained

Table 8. Wild-type and 1433-resistant *T. brucei* growth inhibition assays. EC₅₀s represent averages and standard error of the mean of two independent assays.

	Strain		
Compound	WT EC ₅₀ (nM)	1433 resistant EC ₅₀ (nM)	1433 resistant EC ₅₀ / WT EC ₅₀
Pentamidine	1.5 ± 0.2	1.1 ± 0.1	0.73
1433	98 ± 1	4700 ± 230	48
1	0.13 ± 0.01	14 ± 1	108

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