

Enantioselective Synthesis of 4-Aminotetrahydroquinolines via 1,2-Reductive Dearomatization of Quinolines and Copper(I) Hydride-Catalyzed Asymmetric Hydroamination

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Supporting Information

ABSTRACT: A 1,2-reductive dearomatization of quinolines and copper(II) acetate monohydrate/(R,R)-Ph-BPE/P(ptolyl)₃-catalyzed enantioselective hydroamination sequence was developed, affording diverse 4-amino-1,2,3,4-tetrahydroquinolines with high levels of enantioselectivity in either a stepwise or one-pot fashion. Pleasingly, internal *cis*-cyclic



alkenes, which are challenging substrates in copper hydride-catalyzed enantioselective hydroamination reactions, were transformed efficiently under mild conditions.

hiral tetrahydroquinolines (THQs) are widespread in pharmaceuticals and natural products (Figure 1a).¹ Their



Figure 1. Synthesis of chiral 1,2,3,4-tetrahydroquinolines via dearomatization of quinolines.

efficient construction has long been a goal for both organic and medicinal chemists.² In addition to well-established cyclization reactions such as the Pavarov reaction,^{1c} dearomatization of quinolines represents a straightforward and attractive method toward these scaffolds. Notable examples include asymmetric (transfer) hydrogenation and other reduction reactions (Figure 1b).³ Despite considerable progress, the enantioselective incorporation of THQs with amine functionalities, which are also ubiquitous components in natural products and drug molecules, remains less explored.^{3k-m,o} This might be attributed to the fact that the strong coordination and poisoning capacity of

the substrates and reduced products lead to potential deactivation of the catalysts. In addition, installation of the amine moiety on the quinoline is required prior to the asymmetric transformation, which somehow renders this protocol impractical. Considering the significant importance of chiral amine-substituted THQs, as exemplified by 4-amino-1,2,3,4-tetrahydroquinolines, which show various biological activities, including CETP inhibition,⁴ NMDA receptor antagonism,⁵ and potent bradykinin antagonism,⁶ the development of a straightforward approach directly starting from abundant quinolines is highly desirable.⁷

Reissert-type nucleophilic addition constitutes a commonly used strategy for direct functionalization of quinolines and offers a flexible choice to introduce functionalities such as nitrile, alkyne, and so on (Figure 1b).8 However, to the best of our knowledge, the engagement of amine moieties is problematic and remains unknown to date. To circumvent the challenge, we envisioned that an umpolung tactic using electrophilic amines might be adopted. This hypothesis was inspired by pioneering studies by the Buchwald group^{9a} and the Hirano and Miura groups,⁹⁶ who seminally reported copper hydride-catalyzed enantioselective hydroamination reactions independently in 2013. Since then, a series of chiral amines were obtained in good to excellent yields, regioselectivities, and enantioselectivities from readily available alkenes and alkynes.⁹⁻¹¹ More recently, our group disclosed that this powerful copper catalysis could be extended to benzofurans, for which a benzofuran ring opening/ enantioselective hydroamination cascade was observed.⁹ On the basis of the above considerations and continuing with our interest in value-added functionalization of aromatic com-

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Table 1. Optimization of the Reaction Conditions^a

		$\begin{array}{c} \begin{array}{c} \begin{array}{c} H \end{array} \\ \hline \\ \hline \\ CICO_{2}Me \end{array} \end{array} \xrightarrow{(H)} \begin{array}{c} \begin{array}{c} \\ \hline \\ \\ \end{array} \xrightarrow{(NBR_{2}7}(CuH) \end{array} \xrightarrow{(VBR_{2}7}(VH) \end{array}$			
		1a 2a	le	└O₂Me 4aa	
	(R)-BINAP L1	Ph ₂ Ph ₂ Ar = 3.5- ^t Bu-4-OMe-C ₆ H ₂ (<i>R</i>)-DTBM-SEGPHOS	MeO PPh ₂ MeO PPh ₂ (R)-MeO-BIPHEP L3	(R,R)-Me-DuPHOS L4	
	R = Ph, (<i>R</i> , <i>R</i>)-F R = Ph, (<i>R</i> , <i>R</i>)-F	Ph-BPE L5 Pr-BPE L6 C	it ₂ N 3ab	O NBn ₂ 3ac	
entry	ligand	conc. (M)	3	yield of 4aa (%	$(6)^{b}$ ee of 4aa $(\%)^{c}$
1	L1	0.5	3aa	33	42
2	L2	0.5	3aa	72	18
3	L3	0.5	3aa	59	57
4	L4	0.5	3aa	6	14
5	L5	0.5	3aa	95	80
6	L6	0.5	3aa	20	12
7	L5	0.1	3aa	93	84
8	L5	0.02	3aa	32	84
9	$L5 + PPh_3$	0.1	3aa	78	86
10	$L5 + P(p-tolyl)_3$	0.1	3aa	83	87
11	$L5 + P(p-tolyl)_3$	0.1	3ab	85	87
12	$L5 + P(p-tolyl)_3$	0.1	3ac	85	87
13 ^d	$L5 + P(p-tolyl)_3$	0.1	3ac	87	88
$14^{d,e}$	$L5 + P(p-tolyl)_3$	0.1	3ac	94	88
15 ^{<i>d</i>,<i>e</i>,<i>f</i>}	$L5 + P(p-tolyl)_3$	0.1	3ac	88	89
16 ^{<i>d</i>,<i>e</i>,<i>f</i>,<i>g</i>}	$L5 + P(p-tolyl)_3$	0.1	3ac	92	90

NBn

^{*a*}Reaction conditions: Cu(OAc)₂ (5 mol %), ligand (5.5 mol %), secondary ligand (11 mol %, if used), **2a** (0.20 mmol), **3** (0.24 mmol), and (MeO)₂MeSiH (0.8 mmol) in THF at 40 °C for 20 h. Catalyst was preprepared by mixing Cu(OAc)₂, ligand and (MeO)₂MeSiH together in THF. ^{*b*}Isolated yields. ^{*c*}Determined by SFC analysis. ^{*d*}1.4 equiv of **3ac** was used. ^{*e*}Cu(OAc)₂·H₂O was used instead of Cu(OAc)₂, ^{*f*}The reaction was carried out at room temperature. ^{*g*}Cu(OAc)₂·H₂O (2.0 mol %), L5 (2.2 mol %), P(*p*-tolyl)₃ (4.4 mol %). ^{*h*}Cu(OAc)₂·H₂O (1.0 mol %), L5 (1.1 mol %), P(*p*-tolyl)₃ (2.2 mol %), 48 h.

pounds, we explored whether copper hydride catalysis could provide a general solution toward the synthesis of challenging amine-substituted THQs with high enantioselective induction (Figure 1c). Herein we report our results from this study.

At the outset, we envisioned that enantioselective hydroamination of quinolinium salts might be directly enabled by copper hydride catalysis. However, the initial attempts all failed, along with the observation that the starting materials remained. This is largely due to the difficulties associated with resonance stabilization caused by aromaticity. Inspired by the elegant examples of stepwise reduction/enantioselective catalysis,¹² we therefore used a sequence as an alternative strategy, namely, 1,2reductive dearomatization of quinolines followed by copper hydride-catalyzed asymmetric hydroamination of the resultant 1,2-dihydroquinolines. Nevertheless, this method would be very challenging since the internal *cis*-cyclic alkenes are well-known as challenging and problematic substrates in enantioselective hydroamination catalyzed by copper hydride.^{11f}

Accordingly, *N*-protected 1,2-dihydroquinoline 2a, synthesized from quinoline (1a) via 1,2-reductive dearomatization, was used as the substrate in the hydroamination reaction (Table 1).

Upon treatment of 2a with 1.2 equiv of hydroxylamine ester 3aa in a THF solution of CuH catalyst (obtained from $Cu(OAc)_{2}$, (R)-BINAP, and (MeO)₂MeSiH, 0.5 M) at room temperature for 20 h, the desired product 4aa was obtained in 33% yield with 42% ee (entry 1, Table 1). Encouragingly, after screening of various ligands, we found that utilization of (R,R)-Ph-BPE (L5) gave 4aa in 95% yield with 80% ee (entry 5, Table 1). Slightly improved enantioselectivity was achieved when the substrate was diluted to a concentration of 0.1 M (84% ee; entry 7, Table 1), but further decreasing the concentration resulted in a poor yield (32%; entry 8, Table 1). Gratifyingly, the utilization of a secondary ligand¹³ was found to be beneficial to the reaction, and $P(p-tolyl)_3$ provided the best results (87% ee; entry 10, Table 1). Notably, different dibenzylamine precursors (3aa–ac) exerted little impact on the reaction outcome (entries 11 and 12, Table 1). 1-Adamantyl acid-derived hydroxylamine ester 3ac was chosen as the reaction partner because of its slightly improved efficiency and relatively ready availability and cheapness. Further investigations of the substrate ratio, copper source, reaction temperature, and catalyst loading were conducted (entries 13–17, Table 1). It was identified that the reaction of **2a** with **3ac** (1.4 equiv) in the presence of Cu(OAc)₂. H₂O (2 mol %), (*R*,*R*)-Ph-BPE L**5** (2.2 mol %), P(*p*-tolyl)₃ (4.4 mol %), and (MeO)₂MeSiH (0.8 mmol) at room temperature gave the best results (92% yield, 90% ee; entry 16, Table 1).

With the optimized conditions in hand, we next explored the scope of 1,2-dihydroquinolines (Scheme 1). Various protecting



^{*a*}Reaction conditions: as in entry 16 of Table 1. ^{*b*}Isolated yields are shown. ^{*c*}The ee values were determined by HPLC or SFC analysis. ^{*d*}20 h at rt and then 12 h at 40 °C.

groups on the nitrogen atom of the 1,2-dihydroquinoline, such as $-CO_2Me_1$, $-CO_2Et_1$, $-CO_2Bu_1$, $-CO_2Bu_2$, were well-tolerated, and good to excellent yields (71-93%) and enantioselectivities (82-90% ee) were achieved (4aa-fa). However, the Ts group had a detrimental effect on the reactivity, leading merely to the observation of recovered substrate 2g (16% yield, 91% ee; 4ga). The reactions proceeded smoothly with 1,2-dihydroquinolines bearing different substituents at the 6-position (-CO₂Me, -CF₃, -F, -Cl, -Br, -I, -Ph, 2-thienyl, -Me, -OMe, -SMe), in all cases providing the corresponding THQs in good yields (71-94%) with excellent enantioselectivities (87-92% ee) regardless of their electronic properties (4ha-ra). Furthermore, substrates bearing a methyl substituent at other positions were evaluated (4sa-va). Low conversion was observed when the methyl group was incorporated at the 3position (25% yield, 93% ee, >19:1 dr; 4sa), while the reaction was inhibited with a methyl group at the 4-position. These decreased reactivities might be attributed to the unfavorable steric hindrance encountered in the hydrocupration step. It was remarkable to observe the formation of 4sa in a highly enantioselective and diastereoselective fashion. As expected, 5and 7-methyl substituted 1,2-dihydroquinolines were suitable substrates, leading to the isolation of **4ua** and **4va** in 97% yield with 84% ee and 78% yield with 85% ee, respectively.

Next, the amine electrophiles were examined. As illustrated in Scheme 2, various nitrogen sources bearing varied benzylic

Scheme 2. Investigation of Amine Electrophiles^{*a,b,c*}



^{*a*}Reaction conditions: as in entry 16 of Table 1. ^{*b*}Isolated yields are shown. ^{*c*}The ee values were determined by HPLC or SFC analysis. ^{*d*}4-(*N*,*N*-Diethylamino)benzoic acid-derived hydroxylamine ester was used. ^{*e*}1-Adamantanecarboxylic acid-derived hydroxylamine ester was used. ^{*f*}2.0 equiv of **3** was used. ^{*g*}Benzoic acid-derived hydroxylamine ester was used.

moieties $(4-ClC_6H_4CH_2-, 4-MeOC_6H_4CH_2-, 4$ PhC₆H₄CH₂-, 2-BrC₆H₄CH₂-, 2-thienylmethyl, 2-benzofuranylmethyl) were well-accommodated, delivering the corresponding chiral tertiary amines (4ab-ag) in 46–94% yield with 91-92% ee. The reaction is also applicable to substrates 3h and 3i containing stereocenters adjacent to the nitrogen with opposite configuration. The two diastereoisomers 4ah and 4ai were obtained in 66% yield with 11:1 dr and in 71% yield with 18:1 dr, respectively, indicating that catalyst control process operates. Interestingly, the cyclopropane motif, which is of significance in drug discovery,¹⁴ could be introduced through this hydroamination process (82% yield, 92% ee; 4aj). The robustness of this reaction was further demonstrated by the successful incorporation of morpholine and benzylamine on THQs with excellent enantioselectivity (45% yield, 89% ee, 4ak; 48% yield, 92% ee, 4al).

Finally, we achieved the enantioselective construction of 4amino-1,2,3,4-tetrahydroquinolines directly from quinolines in a one-pot fashion. However, the initial results were rather disappointing, leading to the isolation of 4aa in 18% yield. With modified reaction conditions (KBH₄ was used for the 1,2reduction of quinolines), the results were significantly improved, as shown in Scheme 3. The desired THQs were obtained in satisfactory yields (55–90%) with good enantioselectivities (87–90% ee). Notably, there is no need for isolation or

Scheme 3. One-Pot Synthesis^{*a,b,c*}



^{*a*}Reaction conditions: 1 (0.5 mmol), ClCO₂Me (0.6 mmol), and KBH₄ (0.5 mmol) were stirred at room temperature in MeOH (0.5 M) overnight. A solution of Cu(OAc)₂:H₂O (4.0 mol %), L5 (4.4 mol %), P(*p*-tolyl)₃ (8.8 mol %), (MeO)₂MeSiH (1.0 mmol), and **3ac** (0.25 mmol) in THF (2.5 mL, 0.1 M) was added, and the reaction was continued at rt for 20 h. ^{*b*}Isolated yields are shown. ^{*c*}The ee values were determined by HPLC or SFC analysis. ^{*d*}3.0 equiv of 1 was used.

purification of the dihydroquinoline intermediate during the process.

To our delight, the protocol was also amenable to scale-up and diverse transformations. As shown in Scheme 4, the reaction of



21 with **3ac** on a 5 mmol scale delivered **41a** in 86% yield (2.01 g) with 88% ee (eq 1). The absolute configuration of product **41a** was determined by X-ray crystallographic analysis of its enantiomeric pure hydrochloride salt (**41a**·HCl). Transformations of **41a**, including Suzuki coupling (eq 2) and Sonogashira coupling (eq 3), occurred smoothly to incorporate alkene and alkyne functionalities, respectively, on the THQ skeleton in high yields without any erosion of enantiopurity (86% yield, 88% ee, **5**; 95% yield, 87% ee, **6**). Besides, the $-CO_2Me$ group could be easily removed by treatment of **41a** with potassium hydroxide in a methanol/water mixture at 60 °C (84% yield, 89% ee; 7) (eq 4). Importantly, the introduced functionalities such as alkene, alkyne, and N-H could offer easy handles for further modifications.

In conclusion, we have developed a 1,2-reductive dearomatization of quinolines and copper(II) acetate monohydrate/ (R,R)-Ph-BPE/P(p-tolyl)₃-catalyzed enantioselective hydroamination sequence. Starting from readily available substituted quinolines, structurally diverse THQs bearing amine moieties were obtained in good to excellent yields with high enantioselectivities. Importantly, this method represents a rare example of highly enantioselective hydroamination of internal *cis*-cyclic alkenes. Compatibility with gram-scale reaction and versatile manipulations of the resultant THQs further enhance the synthetic utility of the current method.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.9b02034.

Experimental procedures and analysis data for all new compounds (PDF)

Accession Codes

CCDC 1903797 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by e-mailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, U.K.; fax: +44 1223 336033.

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Notes

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