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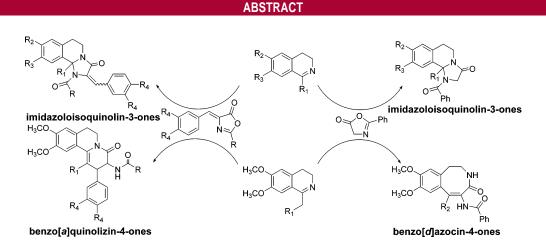
Three Distinct Reactions of 3,4-Dihydroisoquinolines with Azlactones: Novel Synthesis of Imidazoloisoquinolin-3-ones, Benzo[*a*]quinolizin-4-ones, and Benzo[*d*]azocin-4-ones

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A facile and direct synthetic entry to tricyclic imidazoloisoquinolin-3-ones and benzo[a]quinolizin-4-ones is reported based on the ring annulation of 1-unsubstituted and 1-substituted dihydroisoquinolines with azlactones under neutral conditions in a one-step procedure. Bicyclic 2,3-dihydrobenzo[d]azocin-4-ones were also prepared using simple azlactone and 1-substituted dihydroisoquinolines in a one-pot reaction.

Azlactones¹ are versatile precursors for the asymmetric syntheses of α -amino acid derivatives, lactam, and arylpyruvic acid units which have been used for the synthesis of tetrahydro- β -carbolines² and lamellarin alkaloids.³ Schulzeines A–C, new α -glucosidase inhibitors, isolated from the marine

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sponge *Penares schulzei*, were the first three benzo[a]-quinolizin-4-ones containing an amide moiety at the C-3 position.⁴

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Within the class of ring-fused isoquinolines, there have been no reports on the synthesis and biological activity of imidazoloisoquinolines. However, the related triazoloisoquinolines were reported to have some interesting pharmaceutical and agricultural properties.⁵ The berberine,⁶ emetine, and related ipecac alkaloids,⁷ all containing benzo[*a*]quinolizine moieties,⁸ are reported to possess interesting biological activities. Several methods for the preparation of the benzo-[*a*]quinolizine ring system have been reported in the literature.⁹ Benzazocine was found as a structural component of pentacyclic alkaloids which exhibited highly potent cytotoxicity.¹⁰ It was also synthesized for biological study as an eight-membered B-ring of colchicine analogues.¹¹

Various 1-substituted dihydroisoquinolines have been used for the synthesis of benzazocine,¹² benzo[*a*]quinolizines,¹³ and thiazolo[2,3-*a*]isoquinolin-3-ones¹⁴ related to imidazoloisoquinolin-3-ones.

Our group has been interested in the synthesis of some pyrroloisoquinoline alkaloid derivatives.¹⁵ We now report a

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convenient synthesis of tricyclic imidazoloisoquinolin-3-ones 1, benzo[*a*]quinolizin-4-ones 2, and benzo[*d*]azocin-4-ones 3 (Figure 1). To generate heterocyclic structures relevant to

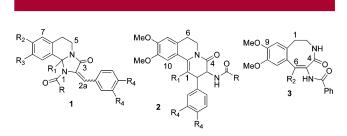


Figure 1. Structures of imidazoloisoquinolin-3-ones 1, benzo[*a*]-quinolizin-4-ones 2, and benzo[*d*]azocin-4-ones 3.

the alkaloid targets, we have investigated the cyclocondensation of azlactones with various dihydroisoquinolines, both unsubstituted and 1-substituted.

Azlactones **4** were prepared directly from the reaction of benzaldehydes and *N*-acetylglycine or hippuric acid in the presence of sodium acetate and acetic anhydride.^{1a} They were obtained in moderate yields after recrystallization from ethanol. The required 3,4-dihydroisoquinolines **5** and **6** were synthesized by the well-established Bischler–Napieralski reaction starting from the arylethylamine derivatives which were converted to the corresponding amide derivatives and then cyclized to imines **5** and **6** using POCl₃.¹⁶

With both key starting materials in hand, the reaction of the simple dihydroisoquinolines with azlactones in acetonitrile was investigated. When 3,4-dihydroisoquinoline **5a** was treated with azlactone **4a** (entry 1, Table 1) in acetonitrile under reflux for 12 h, a single product was obtained in good yield (88%). The product was characterized as imidazoloisoquinolin-3-one **1a** on the basis of spectroscopic and analytical data with a singlet at δ 6.56 (C-10b) in the ¹H NMR spectrum and the amide groups at 1713 and 1668 cm⁻¹ in the IR spectrum. To further demonstrate the scope of this cyclocondensation reaction, the reaction of various azlactones **4a**-**d** and dihydroisoquinolines **5a**-**c** was investigated and the corresponding imidazoloisoquinolin-3-ones **1b**-**h** were obtained in yields ranging from 4 to 94% as shown in Table 1.

The mechanism for the formation of 1 is proposed to involve the acyl iminium salt 7 formed by the reaction of the imine group of the dihydroisoquinoline with the carbonyl group of azlactone followed by subsequent C–N bond formation to provide the imidazoloisoquinolin-3-ones 1(Scheme 1).

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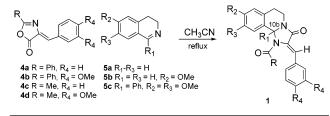
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 Table 1.
 Preparation of Imidazoloisoquinolin-3-ones 1

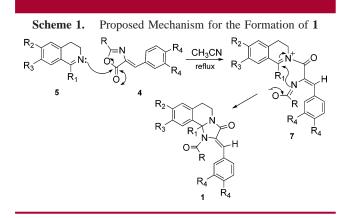


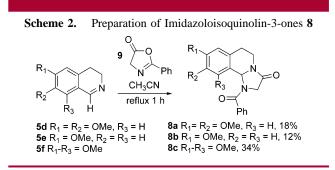
entry	azlactones	isoquinolines	time (h)	yield of 1 (%)
1	4a	5a	12	1a (88)
2	4a	5b	3	1b (94)
3	4b	5a	24	1c (78)
4	4a	5c	48	$1d (17)^a$
5	4c	5a	24	1e (44)
6	4c	5 b	3	1f (64)
7	4d	5a	24	1g(47)
8	4c	5c	48	$1h (4)^b$

 a 78% recovery of imine 5c and 77% recovery of azlactone 4a. b 80% recovery of imine 5c and 50% recovery of azlactone 4c.

The following observations were made from the above reaction. The presence of an electron-donating group at position 6 of the dihydroisoquinoline **5** (R_2) increased the rate of the reaction presumably by increasing the nucleo-philicity of the imine group, as shown in entries 2 and 6 where the reaction went to completion within 3 h. However, the presence of an electron-donating group on the aromatic ring of the azlactone deactivated the carbonyl reactivity resulting in a much slower rate of reaction as shown in entries 3 and 7 where a longer reaction time (24 h) was required. As expected, when the 1-position of the dihydroisoquinoline **5** is substituted with a phenyl group, only a low yield of the product was obtained as in entries 4 and 8. In general, azlactones **4** with a methyl side chain (R = Me) instead of a phenyl group gave lower yields of the product.

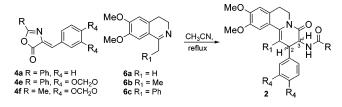
We next turned our attention toward the synthesis of other imidazoloisoquinolin-3-ones $8\mathbf{a}-\mathbf{c}$ by treatment of 3,4dihydroisoquinolines $5\mathbf{d}-\mathbf{f}$ with a simple azlactone 9 in refluxing acetonitrile for 1 h. As indicated in Scheme 2, the low yield of products $8\mathbf{a}-\mathbf{c}$ was observed. This could be due to the formation of other byproducts and decomposition of the simple azlactone 9.





Furthermore, a completely different pathway was observed when azlactones **4** were treated with various 1-alkylsubstituted 3,4-dihydroisoquinolines **6a**–**c** where an imine– enamine equilibrium is possible. For example, when 1-methyl dihydroisoquinoline **6a** ($R_1 = H$) was treated with azlactone **4a** in acetonitrile under reflux for 2 h, the benzo[*a*]quinolizine-4-one **2a** was obtained in 89% yield as shown in entry 1, Table 2. The structure of the product was fully

 Table 2.
 Preparation of Benzo[a]quinolizin-4-ones 2



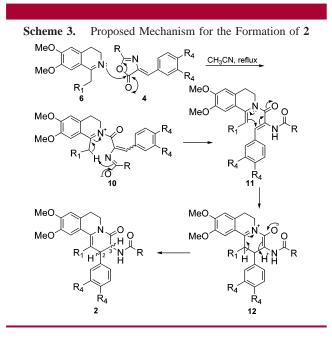
entry	azlactones	isoquinolines	yield of 2 (%, cis/trans) ^a
1	4a	6a	2a (89, 74:26)
2	4a	6b	2b (67, 39:61)
3	4a	6c	2c (83, 24:76)
4	4e	6a	2d (91, 78:22)
5	4e	6b	2e(77, 48:52)
6	4e	6c	2f (88, 44:56)
7	4f	6a	2g(74, 51:49)
8	4f	6b	2h (80, 36:64)
^a All rea	ction times are 2 h	l.	

supported by spectroscopic data with absorption in the IR spectrum at 1683 and 1654 cm⁻¹ for the two amide groups. The ratio of the cis/trans isomers was found to be 74:26 as judged by ¹H NMR (J = 7.6 Hz for the cis isomer and 14.3 Hz for the trans isomer).

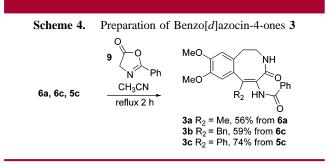
The reaction was found to proceed well with other 1-ethyl and 1-benzyl dihydroisoquinoline derivatives (**6b**,**c**) and differently substituted azlactones **4** giving yields varying from 67 to 91% as shown in Table 2 entries 2–8. It was found that in the case where $R_1 = H$ in **6a** the cis product predominated. When the steric bulk of the substituents increased ($R_1 = Me$, Ph), the trans isomer became the major product.

The formation of 2 is proposed to involve the acyliminium salt 10, analogous to compound 7, formed by the ring-

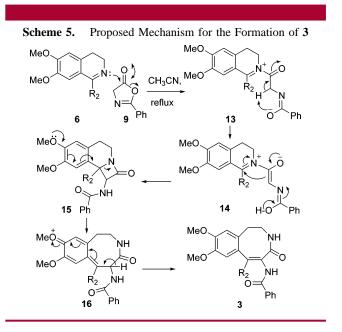
opening reaction at the carbonyl of azlactone 4 with C–N bond formation. With the alkyl substitution at the C-1 position, the acyliminium salt 10 will be readily converted to the corresponding enamide intermediate 11 which undergoes C–C bond formation to give the product 2 via the intermediate 12 as shown in Scheme 3.



A completely different pathway was observed when various 1-substituted 3,4-dihydroisoquinolines **5c**, **6a**, and **6c** were treated with the azlactone **9** in acetonitrile at reflux for 2 h, and benzo[d]azocin-4-one derivatives **3** were instead obtained in moderate yield as shown in Scheme 4. The



mechanism of the reaction was proposed to involve acyl iminium salt **13** which could lead to imidazoloisoquinolin-3-ones **8**, similar to the conversion of intermediate **7** to imidazoloisoquinolin-3-ones **1**, when 3,4-dihydroisoquinolines are unsubstituted. However, in the 1-substituted 3,4dihydroisoquinolines, the formation of imidazoloisoquinolin-3-one was not favorable, and the iminium salt **13** could undergo proton transfer to generate enolate **14** followed by the β lactam ring formation to form lactam **15**. Alternatively, the lactam could be envisaged to form by the addition of the amido ketene, derived from the azlactones **9**, to imine. Cleavage of the C–N bond in the strained lactam assisted by the lone pair of an electron on oxygen could lead to intermediate 16 which could aromatize to give the benzo-[*d*]azocin-4-one derivatives **3** (Scheme 5). All compounds



were fully characterized, and for compound **3a**, the amide absorptions were observed at 1666 and 1631 cm⁻¹. Dihydroisoquinolines with a phenyl substituent at C-1 (**5C**) seem to work slightly better than the C-1 alkyl-substituted dihydroisoquinolines (**6a** and **6c**).

In summary, we have devised a direct, highly efficient route with very simple reaction conditions to imidazoloisoquinolin-3-ones **1**, benzo[a]quinolizine-4-ones **2**, and benzo-[d]azocin-4-ones **3**. The imidazoloisoquinolin-3-ones **1** could be readily obtained by the cyclocondensation reaction of azlactones **4** with C-1 unsubstituted 3,4-dihydroisoquinolines **5**. However, under similar conditions, the C-1 substituted 3,4-dihydroisoquinolines **6** led directly to benzo[a]quinolizine-4-ones **2** and benzo[d]azocin-4-ones **3** depending on the nature of C-1 substituents and azlactone substrates. Compounds **2** lend themselves to conversion to various products, and we are applying this methodology to the synthesis of related alkaloids and other biologically important benzo[a]quinolizin-4-one derivatives.

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Supporting Information Available: Synthetic procedures and spectral data for all new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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