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Total Syntheses of De-branched Nagstatin and Its Analogs Having Glycosidase Inhibiting Activities

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Abstract: De-branched nagstatin and its analogs have been synthesized from protected L-ribo- and xylofurances through the inter- and intra-molecular nucleophilic reactions with the imidazole moieties.

An increasing awareness of the vital role played by carbohydrates in biological processes has stimulated interest in the syntheses of glycosidase inhibitors.¹⁾ Nagstatin (1), which is an N-acetyl- β -D-glucosaminidase inhibitor isolated from fermentation broth of *Streptomyces amakusaensis*²⁾, attracted our attention because of our program in developing new glycosidase inhibitors.³⁾

In this paper we describe the total syntheses of a variety of nagstatin analogs, which feature a general method of entry into the imidazole-having nitrogenous carbohydrates through the intermolecular and intramolecular nucleophilic reactions with the imidazole moieties.

First of all, de-branched nagstatin 2 and its hydroxy analog 3 were effectively synthesized from methyl Lribofuranoside.⁴⁾ O-Benzylation (BnBr, NaH, DMF, 0°C) followed by acid hydrolysis (0.05N HCl, dioxane, 105°C) gave the protected ribose 7 in 90% overall yield, Reaction of 7 (THF, 0°C, 0.5h) with lithiated Ntritylimidazole.⁵⁾ which was prepared from N-tritylimidazole and n-BuLi (THF.-5°C), gave the L-allose derivative 86,7) [47%, mp 62~67°C(amorphous solid), [a]D -111°(c 1.0, CHCl3)] and L-altro derivative 96,7) [40%; mp 132.5~133.5°C(EtOAc), $[\alpha]_D$ -31°(c 1.0, CHCl₃)]. The configurations of their C-2 positions⁷) were determined by the NMR studies of the following cyclized compounds. Both compounds 8 and 9 were converted into 2 and 3 as follows. De-N-tritylation and the S_N2 -type intramolecular cyclization of 8 were effectively realized in one-pot by reaction with BnSO₂Cl in pyridine at -15°C for 1.5h to give preferentially the 5-Osulforylated compound followed by treatment with Ac₂O at 65°C for 1.5h to give the desired acetate 10⁶) [75%. oil, $[\alpha]_D$ -3.0°(c 1.0, CHCl₃)], which was de-O-acetylated (MeONa, MeOH, rt, 1.5h) to the nitrogenous Dtalose analog 11⁶) [84%; mp 77~78°C(hexane-EtOAc), $[\alpha]_D$ -7.8°(c 1.0, CHCl₃)]. The effective de-N-tritylation seemed to be affected by the producing pyridinium acetate and was supported by the stepwise conversions of 13 and 20 into 14 and 21 as shown later. The inversion of the hydroxyl group in 11 with HN3 (n-Bu3P, DEAD, THF, PhMe, rt, 0.5h)⁸) afforded the azido derivative 12 [68%, oil, $[\alpha]_D$ +98°(c 1.2, CHCl₃)], which was subjected to hydrogenolysis (H2, Pd-C, AcOH) and N-acetylation (Ac2O, MeOH) leading to the N-acetyl-Dgalactosamine analog 2⁶) [65%, mp 210~212°C(decomp.), $[\alpha]_D + 109°(c \ 1.3, H_2O)$], which was corresponding to de-branched nagstatin. The optical rotation, the coupling constants of the ¹H-NMR, and glycosidase inhibiting activities⁹⁾ of 2 were much the same as those of nagstatin (1), indicating that natural nagstatin has the D-galacto structure 1^{2} .

Alternatively, 12 was prepared from the other isomer 9. Treatment of 9 with BnSO₂Cl (Py, -10°C, 1h) to give 13 ⁶[86%; mp~60°C(amorphous solid), $[\alpha]_D$ -43°(*c* 1.4, CHCl₃)] followed by acetylation (Ac₂O, Py, 65°C, 1.5h) gave the cyclized compound 14 (90%; oil, $[\alpha]_D$ +75°(*c* 1.4, CHCl₃)] with de-N-tritylation as described above, which was de-O-acetylated (MeONa, MeOH) to 15 [95%; mp 112~113°C(hexane-EtOAc), $[\alpha]_D$ +38°(*c* 1.0, CHCl₃)]. Reaction of 15 with HN₃ (n-Bu₃P, DEAD, THF, PhMe, rt, 0.5h) gave 12 in 68% yield with retention of the C-2 configuration as expected. The S_N2 replacement of the C-2 equatorial group in carbohydrates has been hardly known to occur because of the ring oxygen, the anomeric substituent and dipolar effects.¹⁰⁾ This retention was confirmed by the fact that 15 was treated with benzoic acid, n-Bu₃P and DEAD (Et₂O, *it*, 0.5h) to give the benzoate 16⁶ [75%. mp 97~98°C(EtOAc), $[\alpha]_D$ +98°(*c* 1.1, CHCl₃)], which was deacylated (MeONa, MeOH) to the starting 15. Hydrogenolysis of 15 (H₂. Pd-C, AcOH, rt, 15h) afforded the nitrogenous D-galactose analog 3 ⁶ [90%; mp~80°C(amorphous solid), $[\alpha]_D$ +29°(*c* 1.6, MeOH)]. Similarly, the enantiomeric N-acetyl-L-galactosamine analog 2' and L-galactose analog 3' were prepared from methyl D-rib ofuranoside by the same procedures as mentioned above: 2'⁶): mp 210~212°C(decomp.), $[\alpha]_D$ -104°(*c* 0.65, M:OH); 3'⁶): mp~80°C(amorphous solid), $[\alpha]_D$ -32°(*c* 0.95, MeOH). Both compounds showed no significant glycosidase inhibiting activities⁹).

Furthermore, nitrogenous N-acetyl-D-glucosamine, D-glucose and D-mannose analogs (4, 5 and 6) were efficiently prepared from methyl L-xylofuranoside⁴⁾ by the similar fashion as described above. Successive Obenzylation (BnBr, NaH, DMF, 0°C) and hydrolysis (0.05N HCl, dioxane, 105°C) gave the protected xylofuranose 17 (92%), which reacted with lithiated N-tritylimidazole (THF, -10°C) to vield the L-gulose analog 18 ⁶⁾[64%, mp 40-44°C(amorphous solid), [α]_D -55°(c 0.87, CHCl₃)] and L-idose analog 19 ⁶⁾[16%; mp 43~46°C(amorphous solid), $[\alpha]_D + 18^\circ(c \ 1.2, CHCl_3)$]. The approach of the imidazole moiety to the C-1 position⁷) of 17 was reasonably controlled by the cis Li-chelation between the C-1 and C-2,¹¹) while the aforesaid reaction of 7 was not affected by such chelation controll probably because of the preferential formation of the cis Li-chelation between the C-2 and C-3. Benzylsulfonylation (BnSO₂Cl, Py, -10°C, 0.5 h) of 18 followed by acetylation with concomitant cyclization (Ac₂O, Py, 50°C, 8 h) gave the nitrogenous D-mannose analog 21⁶[84%, oil, [a]_D-50°(c 0.96, CHCl₃)], which was deacetylated (MeONa, MeOH) to the alcohol 22 ⁽⁰⁾[95%, mp 116.5~117.5°C(hexane-EtOAc), [a]p -4.0°(c 0.95, CHCl3)]. Hydrogenolysis (H2, Pd-C, AcOH) of 22 gave the D-mannose analog 6^{6} [91%, mp 111~114°C(amorphous solid), $[\alpha]_{D}$ -36°(c 1.0, MeOH)]. Mitsunobu inversion of the hydroxyl group of 22 was carried out with HN₃ by the aforesaid conditions to give **23**⁶⁾ [71%, oil, $[\alpha]_D$ +59°(c 1.0, CHCl₃)], which was converted into the nitrogenous N-acetyl-D-glucosamine analog 4 ⁶) [65%; mp 249~251°C(decomp.), $[\alpha]_D$ +52°(c 0.90, H₂O)] by hydrogenolysis and N-acetylation as described above. The azide 23 was also obtained from 24 with retention of the C-2 configuration as described in the preparation of 12 from 15. The intermediate 24 was prepared from 19 by benzylsulfonylation and acetylation with cyclization followed by de-O-acetylation as described in the preparationn of 15 from 9, and also derived from 22 in 85% overall yield by inversion (BzOH, n-Bu3P, DEAD, THF, rt, 0.5h) and de-O-acvlation (MeONa, MeOH). The D-glucose analog 5⁶) was prepared from 24 by hydrogenolysis (H₂, Pd-C, AcOH, rt, 15h): mp 169~174°C(amorphous solid), [α]_D -8.0°(c 0.97, MeOH).

Details on the glycosidase inhibiting activities of the synthesized analogs (2 - 6) will be described in another paper⁹.



1 (Nagstatin)



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- All compounds were purified by silica-gel column chromatography and/or recrystallization, and were fully 6 characterized by spectroscopic means. Optical rotations were measured using a 0.5 dm tube at 22°C. Significant ¹H-NMR spectral data (270 and 400 MHz, CDCl₃, δ ; TMS=0, unless otherwise noted) are the following. $2(CD_3OD)$: $\delta 2.05(3H_s)$, $3.94(1H_s) = 9\&2H_z$, $4.15(1H_s) = 6\&6\& 3H_z$, 4.32(1H,dd,J=3&2Hz), 5.14(1H,d,J=9Hz), 6.95(1H,d,J=1.5Hz), 7.35(1H,d,J=1.5Hz). 3(CD₃OD): δ 3.87(1H.dd.J=7&2Hz), 4.23(1H.dt.J=5&4Hz), 4.38(1H.dd.J=4&2Hz), 4.76(1H.d.J=7Hz), 7.05(1H,d,J=1Hz), 7.36(1H,d,J=1Hz), 4(pyridine- d_5); δ 2.11(3H,s), 4.34(1H,ddd,J=9&5& 2.5Hz), 4.44(1H,dd,J=9&9Hz), 4.57(1H,dd,J=9&9Hz), 5.85(1H,dd,J=9&9Hz), 7.35(1H,d, J=1.5Hz), 7.68(1H,d,J=1.5Hz), $5(D_2O)$; δ 3.85(1H,dd,J=10&9Hz), 3.98(1H,dd,J=10&9Hz), 4.12(1H,m), 7.30(1H.d.J=1.5Hz), 7.43(1H.d.J=1.5Hz). $6(\text{pyridine} - d\varsigma): \delta$ 4.72(1H.d.J=9Hz). 4.40(1H.dd.J=7.5&4Hz), 4.76(1H,ddd,J=9&4&4Hz), 5.04(1H,dd,J=9&7.5Hz), 5.66(1H,d,J=4Hz), 7.39(1H,d,J=1.5Hz), 7.78(1H,d,J=1.5Hz). 8: 8 3.83(1H,dd), 3.92(1H,ddd), 4.00(1H,dd), 4.53 (1H.dd), 6.83(1H.d), 7.09(1H.d), 9 : δ 1.86(1H,dd), 3.50(1H,dd), 3.82(1H,ddd), 4.50(1H,dd), 10 : δ 4.09(1H,dd), 4.21(1H,dd), 4.35(1H,dd), 6.25(1H,d), 7.18(1H,d), 7.20(1H,d). 11 : δ 3.97(1H. dd,J=4&2Hz), 4.24(1H,dd,J=5&2Hz), 4.30(1H,dt,J=6&5Hz), 5.02(1H,d,J=4Hz), 7.07(1H,d, J=2Hz), 7.11(1H,d,J=2Hz), 12: δ 3.81(1H,dd), 4.24(1H,dd), 4.28(1H,dt), 5.00(1H,d), 7.07(1H,d), 7.11(1H.d). 13 : δ 1.50(1H,d), 3.73(1H,dd), 4.46(1H,s), 5.08(1H,ddd). 14: δ 3.98(1H,dd), 4.21 $(1H.dd), 4.45(1H.ddd), 6.22(1H.d), 7.08(1H.d), 7.15(1H.d), 15 : \delta 4.02(1H.dd, J=5.5\&2Hz), 4.35$ (1H,dd,J=5&2Hz), 4.39(1H,ddd,J=7.5&5&3Hz), 5.17(1H,d,J=5.5Hz), 6.98(1H,d,J=1.5Hz), 7.10 (1H,d,J=1.5Hz), **16** : δ 4.12(1H,dd), 4.29(1H,dd), 4.50(1H,ddd), 6.49(1H,d), 7.11(1H,d), **18** : δ 3.74(1H,dd), 3.98(1H,dt), 4,15(1H,dd), 4.51(1H,br.dd), 6.85(1H,d). 19(acetone-d₆): § 3.19(1H,dd), 3.25(1H,dd), 3.54(1H,m), 4.34(1H,dd), 6.82(1H,d), 7.05(1H,d). **20** : δ 4.04(1H,dd), 4.19(1H,dd), 4.54(1H,dd), 5.04(1H,ddd), 6.82(1H,d). 21(acetone-d₆): δ 4.16(1H,dd), 4.24(1H,ddd), 4.29(1H, dd), $6.50(1 \text{ H.d}), 6.98(1 \text{ H.d}), 22 : \delta 3.96(1 \text{ H.dd}, \text{J}=8\&4 \text{ Hz}), 4.15(1 \text{ H.ddd}, \text{J}=6\&6\&3.5 \text{ Hz}), 4.20(1 \text{ H.dd}), 4.20(1 \text{ Hz}), 4.20(1$ $dd_{J}=8\&6Hz$), 5.13(1H,d,J=4Hz), 7.01(1H,d,J=1.5Hz), 7.10(1H,d,J=1.5Hz). 23 : δ 3.88(1H,dd), 3.96(1H,dd), 4.15(1H,ddd), 4.73(1H,d), 7.02(1H,d), 7.13(1H,d). 24(acetone- d_6): δ 3.97(1H,dd, J=7.5&6Hz), 4.06(1H,dd,J=7.5&7.5Hz), 4.30(1H,m),4.79(1H,d,J=6Hz), 6.95(1H,s), 7.19(1H,s).
- 7. The carbon-numbering protocol and nomenclature conveniently parallel those of carbohydrates.
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