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The synthesis of 2-oxo-2,3-dihydro-1H-thieno[3,4-d]imidazole derivatives with substituents at N¹, N³, C⁴, and C⁶ (I)-(III) was described in a previous paper [1]. In the present paper we studied the reduction of (I)-(III) with triethylsilane in CF₃COOH as described in [2-4]. The reduction of the 3- and 4-methyl-2-oxo-2,3-dihydro-6-(δ -carbomethoxybutyl)-1H-thieno[3,4-d]imidazoles (I) and (II), and subsequent alkaline hydrolysis of the intermediate methyl ethers (IV) and (V), gave in yields of 82 and 72% the 3- and 4-methyl-2-oxo-6-(δ -carboxybutyl)hexahydrothieno[3,4-d]imidazoles (VI) and (VII), which on the basis of the following data were assigned a complete cis configuration.



 $R=R^{2}=H$, $R^{1}=Me$ (I), (IV), (VI); $R=R^{1}=H$, $R^{2}=Me$ (II), (V), (VII); $R=R^{1}=Me$, $R^{2}=H$ (III), (IX); $R=R^{1}=R^{2}=H$ (VIII), (X), (XIV); $R^{3}=(CH_{2})_{4}COOMe$ (I)-(V), (X), (XIV); (CH₂)₄COOH (VI)-(IX).

Compounds (VI) and (VII) are stable when heated with dilute H_2SO_4 solution, similar to biotin (VIII) with a cis-coupling of the rings [5]. The complete cis configuration of (VI) is also indicated by the fact that the methylation of (VI) with a mixture of CH₂O and HCOOH gives the N,N'-dimethyl derivative (IX), which is identical with the methylation product of the authentic d2-biotin (VIII). Compound (VII) in its melting point, IR spectrum, and TLC data coincides with the previously described high-melting stereoisomer of 2-oxo-4-methyl-6-(δ -carboxybutyl)hexahydrothieno[3,4-d]imidazole, which has a complete cis configuration [6].

The IR spectra of (VI)-(VIII) have certain characteristics that can be used to identify them. Compounds (VI) and (VIII) give three bands in the region of the stretching vibrations of carbonyl groups (1635-1710 cm⁻¹), while (VII) gives two absorption bands. Two bands in the region of higher frequencies (1695-1745 cm⁻¹) are observed for the corresponding methyl ethers (V) and (X).

The high stereoselectivity of the cis-reduction of (I) and (II) can be explained if it is assumed that the approach of the HSiEt₃ to the intermediate protonated forms (XI) and (XII), and also the approach of the proton to the dihydrothiophene derivative (XIII), is from the less shielded side. The number and position of the substituents both exert a substantial influence on the ease of reducing the 2-oxo-2,3-dihydro-1H-thieno[3,4-d]imidazole system. As can be seen from the above given data, (I) and (II), which contain electrondonor groups at N³, C⁴, and C⁶, which facilitate the formation of cations (XI) and (XII), are reduced very smoothly. The reduction of compound (XIV), which lacks alkyl substituents at N¹, N³, and C⁴, proceeds with great difficulty and, after hydrolysis of the intermediate methyl ether (X), leads only to traces of biotin (VIII). However, unsatisfactory results are also obtained in the reduction of compound (III) with CH₃ groups on both of the nitrogen atoms N¹ and N³, which probably hinder the approach of HSiEt₃ to the intermediate cation (XII).

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EXPERIMENTAL

The IR spectra were taken as KBr pellets on a UR-20 spectrometer, while the PMR spectra were taken in CF₃COOH solution on a Tesla B5497 instrument, using HMDS as the internal standard. The TLC was run on Silufol plates in the system: 3:2 ethyl acetate (EA)-alcohol (the spots were detected with aqueous $K_3Fe(CN)_6$ solution, and then with iodine vapors).

 $2-0xo-3-methyl-6-(\delta-carboxybutyl)hexahydrothieno[3,4-d]imidazole (VI). A mixture of$ 0.3 g of 2-oxo-2, 3-dihydro-3-methyl-6-(δ-carbomethoxybutyl)-1H-thieno[3,4-d]imidazole (I) [1] and 0.8 ml of HSiEt; in 3 ml of CF; COOH was heated for 30 h at 50° (here and subsequently the bath temperature), evaporated in vacuo, and the residue was mixed with 1 ml of conc. HCl and 5 ml of alcohol, kept for 12 h at 20°, and evaporated. The residue was treated with a solution of 0.5g of KOH in 5 ml of alcohol, left standing for 12 h at 20°C, and evaporated again. The residue was dissolved in 2 ml of water and acidified with dilute HC1 solution (1:1) to pH \sim 1. The precipitate was filtered, washed in succession with water and acetone, and dried in the air. We obtained 0.28 g of crude (VI), which was purified by heating with 3 ml of 10% H2SO4 solution for 1 h at 140°. After cooling to 20°, the precipitate was filtered, washed with water, dissolved in a solution of 0.16 g of Na₂CO₃ in 50 ml of water; 0.3 g of AB-172 P resin was added, and the mixture was refluxed for 5 min. The filtrate was cooled to 20° and acidified with dilute HCl solution (1:1) to pH vl. The precipitate was filtered, washed in succession with water and acetone, and dried in the air. We obtained 0.24 g (82%) of (VI), mp 205-206°, R_f 0.53. Infrared spectrum (v, cm⁻¹): 1188, 1198, 1235, 1269, 1290, 1312, 1326, 1338, 1412, 1468, 1522, 1635, 1680, 1700, 2865, 2935, 3290. PMR spectrum (δ, ppm): 1.24 m (CH₂CH₂CH₂), 2.06 d (CH₂CO), 2.52 s (CH₃N), 2.64 s (CH₂S), 2.90 m (CHS), 4.00 m, 4.18 m (NCHCHN) [2].

<u>1,8-Dimethyl-2-oxo-6-(&-carboxybutyl)hexahydrothieno[3,4-d]imidazole (IX)</u>. A mixture of 0.5 g of imidazole (VI) and 1 g of formalin in 5 g of HCOOH was heated for 40 h at 100°, evaporated in vacuo, and the residue was treated with a 1:1 EA-MeOH mixture. The precipitate was filtered, washed with EA, and dried in the air. We obtained 0.2 g (38%) of (IX), mp 140-142° (from MeOH), Rf 0.71. Infrared spectrum (ν , cm⁻¹): 765, 1191, 1230, 1265, 1296, 1315, 1414, 1464, 1508, 1678, 1718, 2930, 3120. PMR spectrum (δ , ppm): 1.23 m (CH₂CH₂CH₂), 2.06 d (CH₂CO), 2.50 s (CH₃N), 2.60 s (CH₃N), CH₂S), 3.02 m (CHS), 3.88 m and 4.12 m (NCHCHN). Found: C 52.58; H 7.36; N 10.34; S 11.62%. C₁₂H₂₀N₂O₃S. Calculated: C 52.90; H 7.31; N 10.30; S 11.76%. The (IX), obtained by the methylation of d*l*-biotin (VIII), under the above described conditions, had the same properties.

 $\frac{2-0\text{xo}-4-\text{methyl}-6-(\delta-\text{carboxybutyl})\text{hexahydrothieno}[3,4-d]\text{imidazole (VII)}.$ The reduction of 0.3 g of 2-0x0-2,3-dihydro-4-methyl-6-(δ -carboxybutyl)-1H-thieno[3,4-d]imidazole (II) [1] with HSiEt₃, as described above, gave 0.28 g of crude (VII). After purification we iso-lated 0.21 g (72%) of (VII) with mp 264-265°, R_f 0.63. Infrared spectrum (ν , cm⁻¹):

1198, 1256, 1320, 1330, 1467, 1495, 1660, 1710, 2870, 2932, 3056, 3270. PMR spectrum (δ, ppm): 0.96 t (CH₃CH), 1.30 m (CH₂CH₂CH₂), 2.06 d (CH₂CO), 3.02 m (2CHS), 3.78 m and 4.18 s (NCHCHN). Found: C 51.30; H 7.08; N 10.67; S 12.55%. C₁₁H₁₈N₂O₃S. Calculated: C 51.16; H 6.95; N 10.85; S 12.40%. The obtained (VII) did not depress the mixed melting point with an authentic specimen (mp 260-261°) [6] and had the same IR spectrum and R_f.

Infrared spectrum of d7-biotin (VIII) (v, cm⁻¹): 1164, 1201, 1215, 1236, 1268, 1290, 1325, 1412, 1432, 1468, 1488, 1655, 1690, 1710, 2875, 2940, 3074, 3245.

Infrared spectrum of d-biotin (v, cm⁻¹): 1020, 1079, 1102, 1120, 1142, 1158, 1206, 1240, 1272, 1295, 1320, 1342, 1432, 1465, 1484, 1645, 1694, 1710, 1955, 2480, 2840, 2856, 2860, 2900, 2921, 2935, 2960, 3310, 3365.

 $\frac{2-0xo-4-methyl-6-(\delta-carbomethoxybutyl)hexahydrothieno[3,4-d]imidazole (V). With stirring and cooling in ice, to 14 ml of MeOH was gradually added 3.5 ml of SOCl₂, and then 0.7 g of (VII), and the mixture was kept for 72 h at 20° and then evaporated in vacuo. The residue was treated with excess aqueous Na₂CO₃ solution and then extracted with EA. The extract was dried over MgSO₄, the solvent was removed, and the precipitate was washed with ether and dried in the air. We obtained 0.44 g (60%) of (V), mp 205-206° (from H₂O), R_f 0.65. Infrared spectrum (v, cm⁻¹): 1144, 1212, 1465, 1695, 1738, 2942, 3400. PMR spectrum (<math>\delta$, ppm): 0.97 t (CH₃CH), 1.31 m (CH₂CH₂CH₂), 2.06 t (CH₂CO), 3.03 m (2CHS), 3.37 s (CH₃O), 3.79 m and 4.18 m (NCHCHN). Found: C 53.01; H 7.47; N 10.53; S 11.73%. C₁₂H₂₀N₂O₃S. Calculated: C 52.90; H 7.31; N 10.30; S 11.76%.

The methyl ether of d*l*-biotin [6, 7] was obtained in a similar manner in 66% yield, mp 130-132° (from H₂O). Infrared spectrum (ν , cm⁻¹): 1175, 1208, 1468, 1710, 1745, 2950, 3278.

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CONCLUSIONS

We synthesized the cis-3- and 4-methyl-2-oxo-6-(δ -carboxybutyl)hexahydro[3,4-d]imidazoles by the reduction of the 3- and 4-methyl-2-oxo-2,3-dihydro-6-(carboxybutyl)-lHthieno[3,4-d]imidazoles with HSiEt₃ in CF₃COOH and subsequent alkaline hydrolysis of the intermediate methyl ethers.

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