

# A Practical Synthesis of 2-{4-[4-Fluoro-3-(trifluoromethyl)phenyl]-2-(piperidin-4-yl)-1*H*-imidazol-1-yl}-*N,N*-dimethylethanamine

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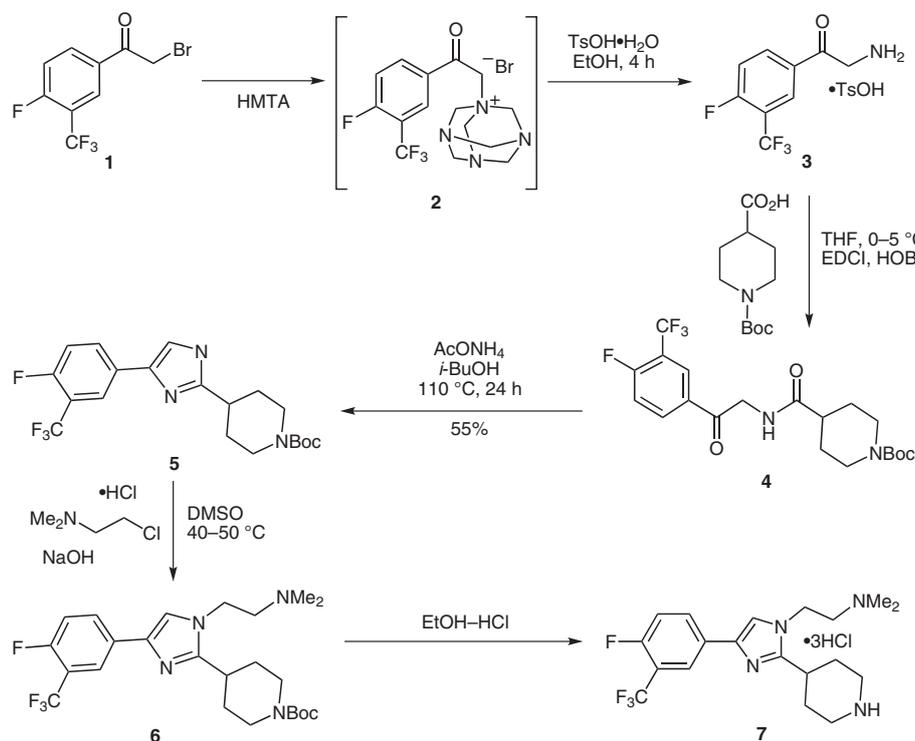
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**Abstract:** A practical synthesis of the title compound was accomplished by hydrogenation of 2-{4-[4-fluoro-3-(trifluoromethyl)phenyl]-2-(pyridin-4-yl)-1*H*-imidazol-1-yl}-*N,N*-dimethylethanamine. The latter was obtained by *N*-alkylation of 4-{4-[4-fluoro-3-(trifluoromethyl)phenyl]-1*H*-imidazol-2-yl}pyridine. Treatment of *N*-{2-[4-fluoro-3-(trifluoromethyl)phenyl]-2-oxoethyl}isonicotinamide hydrochloride with ammonium acetate in acetic acid provided 4-{4-[4-fluoro-3-(trifluoromethyl)phenyl]-1*H*-imidazol-2-yl}pyridine. Coupling of 2-amino-1-[4-fluoro-3-(trifluoromethyl)phenyl]ethanone 4-methylbenzene sulfonate with piperidine 4-carboxylic acid using either T<sub>3</sub>P or EDCI-HOBt provided *N*-{2-[4-fluoro-3-(trifluoromethyl)phenyl]-2-oxoethyl}isonicotinamide hydrochloride.

**Key words:** imidazole, EDCI, T<sub>3</sub>P, ammonium acetate, hydrogenation

Substituted imidazoles are important heteroaromatic compounds, which not only exhibit a broad range of biological activities such as antifungal properties,<sup>1–10</sup> analgesic activity,<sup>2,3</sup> anti-inflammatory activity inhibiting the p38 MAP kinase or cytokine release,<sup>4,5</sup> and antiallergic activity,<sup>6</sup> but also are important building blocks found in naturally occurring compounds.<sup>7–10</sup> Therefore, we were interested in a robust synthesis of 2-{4-[4-fluoro-3-(trifluoromethyl)phenyl]-2-(piperidin-4-yl)-1*H*-imidazol-1-yl}-*N,N*-dimethylethanamine. A survey of literature revealed that synthesis of the title compound can be accomplished in six steps from 2-bromo-1-[4-fluoro-3-(trifluoromethyl)phenyl]ethanone (**1**) (Scheme 1).<sup>11</sup>

The existing synthetic approach to **5**, while well documented, resulted in poor isolated yield of **5** because of oligomerization of the starting material **4** under reaction



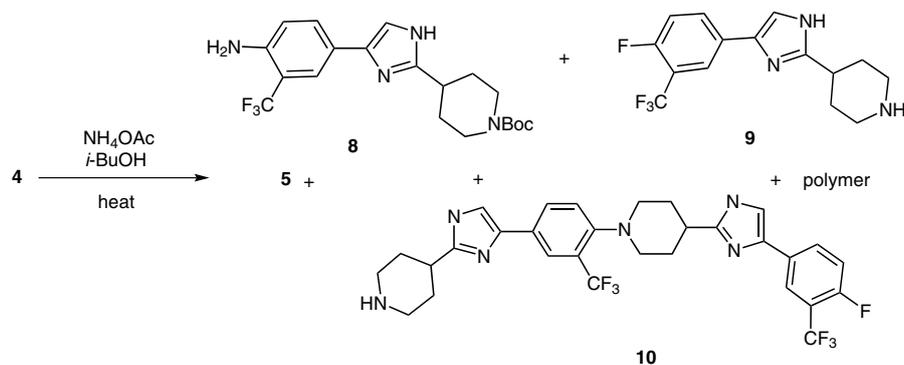
**Scheme 1** Literature synthesis of compound **7**

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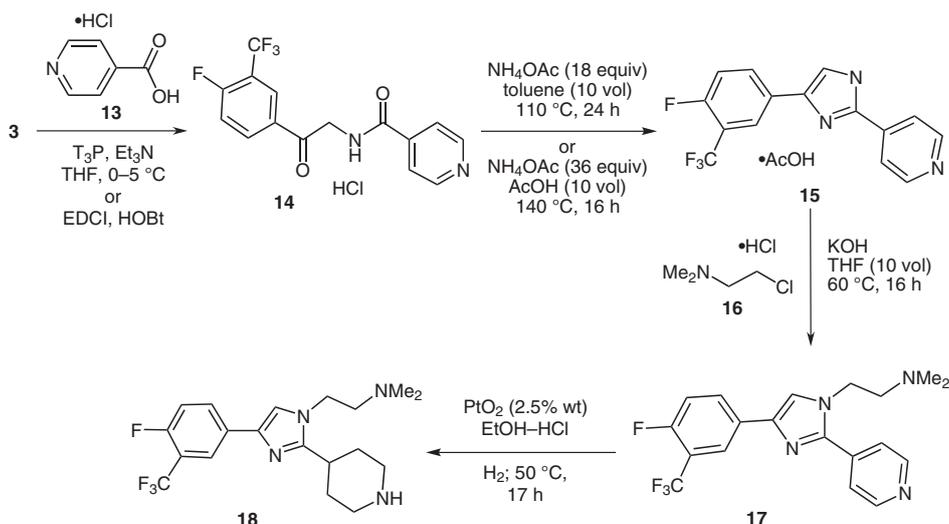


**Scheme 2** Impurities formed during scale-up of **5**

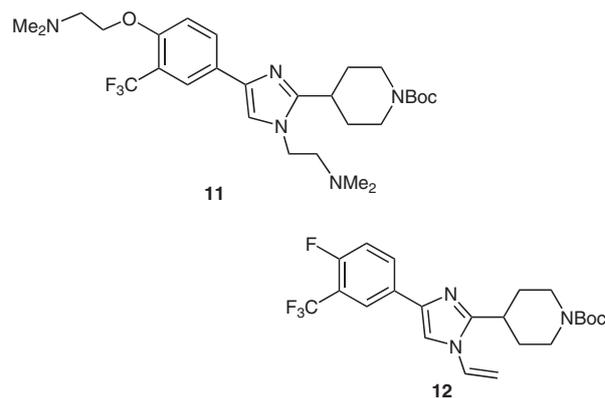
conditions due to its condensation with **9** derived by the thermal deprotection of Boc group. Thus, in addition to the formation of desired **5**, deprotected product **9**, and significant quantities of aniline analogue **8** along with dimer **10** were observed (Scheme 2). These additional impurities generated in this process along with the specific need for protected imidazole **5** posed challenges during scale-up with respect to reproducibility and purification. Further, N-alkylation of **5** in DMSO at 100 kg scale also resulted in the formation of undesired impurities at significantly high level (8–12%, Figure 1), which made the purification of **6** very difficult resulting in a poor isolated yield.

In order to avoid the above mentioned issues, we were interested in developing a robust synthesis of title compound **18** from **3**. Herein, we report a practical synthesis of title compound as shown in Scheme 3.

Thus, synthesis of **3** was accomplished following the literature procedure.<sup>11</sup> Coupling of acid chloride of **13** with **3** resulted in poor conversion. However, synthesis of **14** was successfully achieved in good yield and purity via activation of acid **13** either by using  $\text{T}_3\text{P}$  (2,4,6-triisopropyl-1,2,3,2,4,6-trioxatriphosphorinane-2,4,6-trioxide) or EDCI-HOBt (Table 1).



**Scheme 3** Synthesis of compound **18**



**Figure 1** Impurities **11** and **12** observed during synthesis of **6**

Imidazole ring construction from ketoamides using ammonium acetate has been known in literature.<sup>12</sup> Thus, keeping the literature methods in view, synthesis of **15** was explored in various solvents with excess of ammonium acetate (Table 2). As determined by HPLC, the formation of **15** was successful in acetic acid, propanoic acid, and toluene. However, azeotropic removal of water was required in the synthesis of **15** using toluene as a solvent system. Based upon the yield data, we preferred acetic

**Table 1** A Solvent and Acid Activating Agents Screening Study in the Synthesis of **14**

Entry	Substrate	Acid activating reagent	Solvent	Base <sup>a</sup>	Temp (°C)	Yield (%) <sup>b</sup>
1	<b>13</b> acid chloride	none	CH <sub>2</sub> Cl <sub>2</sub>	Et <sub>3</sub> N	25	30
2	<b>13</b> acid chloride	none	CH <sub>2</sub> Cl <sub>2</sub>	Et <sub>3</sub> N	reflux	20
3	<b>13</b>	EDCI-HOBT	THF	NMM	23–30	88
4	<b>13</b>	T <sub>3</sub> P in EtOAc	EtOAc–THF	Et <sub>3</sub> N	0–5	79
5	<b>13</b>	T <sub>3</sub> P in EtOAc	EtOAc–THF	NMM	0–5	75

<sup>a</sup> NMM: *N*-Methylmorpholine.

<sup>b</sup> Isolated yield.

acid as a solvent for the synthesis of **15** using ammonium acetate at 140 °C under pressure. This resulted in excellent conversion of **15** as determined by HPLC. The reaction workup involved addition of water effecting in crystallization of **15**. Thus, synthesis of **15** was demonstrated at 100 gram scale, which provided product in 89% yield with excellent purity.

A screening study was performed to find a suitable experimental condition for N-alkylation of **15** with **16** (Table 3). Impact of organic and inorganic bases was studied. HPLC data from this study indicated that use of sodium hydride as a base in THF provided the best result with respect to the formation of **17** (Table 3). Both cesium carbonate and powdered potassium hydroxide provided similar outcome in the synthesis of **17**. As the use of NaH on large scale poses challenges with respect to safety and handling, we were interested in the development of a process with the use of alkali metal hydroxide. Thus, a study

was further completed to find the optimum amount of potassium hydroxide required for the formation of **17** (Table 3, entries 9–14). Based upon the data in Table 3, alkylation of **15** using THF as a solvent and powdered KOH provided a mixture of **17** and **19** in ratio of 91:6 along with undesired impurities (**20** and **21**, Scheme 4). The possible pathway for the formation of **20** may involve the reaction of **17** with **16** to form quaternary salt **20**, which upon elimination under basic conditions leads to formation of **21** (Scheme 5).

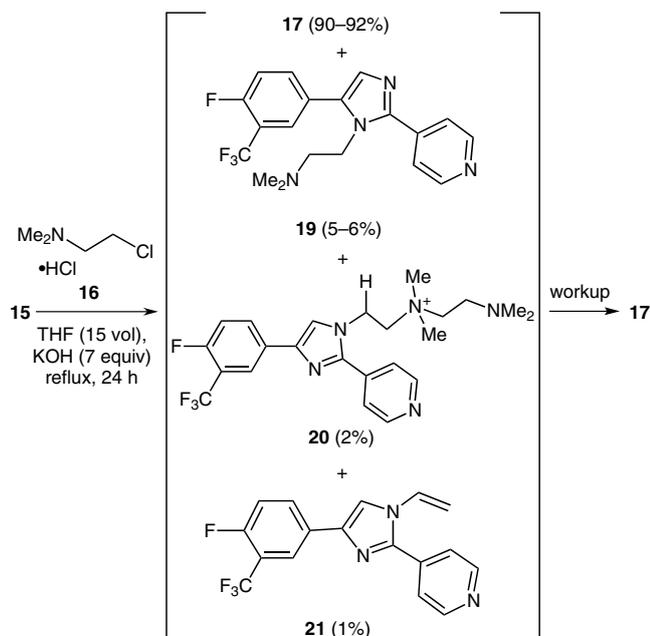
Keeping the complexity of alkylation reaction in view, a workup involving treatment of reaction mixture with activated carbon followed by crystallization was designed to obtain pure compound **17** in 78% yield. Absorption of **19** was observed on the carbon and it was determined that 30% of carbon was required for removal of **19**. N-Alkylation of **15** was performed at 300 g scale, which provided **17** in 77% isolated yield with excellent purity.

**Table 2** A Solvent Screening Study in the Synthesis of **15**

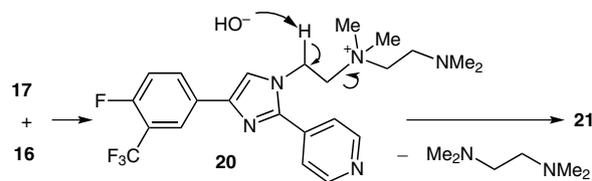
Entry	NH <sub>4</sub> OAc (equiv)	Solvent	Volume <sup>a</sup>	Temp (°C)	Time (h)	Formation of <b>15</b> by HPLC (%)	Yield (%) <sup>b</sup>
1	20	EtOH	20	reflux	24	20	NI
2	20	<i>i</i> -PrOH	20	reflux	24	32	NI
3	50	AcOH	20	110	24	57	NI
4	50	AcOH	20	140	10	94	NI
5	50	AcOH	10	140	10	91	NI
6	36	AcOH	10	140	12	91	NI
7	36	EtCO <sub>2</sub> H	10	140	10	93	NI
8	36	AcOH	10	140	16	93	89
9	36	AcOH	10	140	16	94	90
10	20	AcOH	10	140	16	68	NI
11	12	AcOH	10	140	16	52	NI
12	21	toluene	21	111	24	86	78

<sup>a</sup> Reaction volume is based upon the quantity of **15** (g) used in the reaction, for example, for entry 9 amount of **15** used was 100 g (1 equiv); thus 10 solvent volumes correspond to 10 × 100 = 1 L; NI = product not isolated.

<sup>b</sup> Isolated yield.

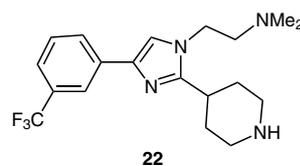


**Scheme 4** Impurities observed during **15** alkylation



**Scheme 5** Possible pathway for the formation of impurity **21**

Our attempts to reduce the pyridine ring of **17** by catalytic hydrogenation using 5% palladium on carbon in acetic acid at room temperature and 50 psi gave a mixture of desired product **18** along with corresponding 4-des-fluoro analogue **22** (Figure 2) after 30 hours.<sup>13</sup>



**Figure 2** Structure of **22**

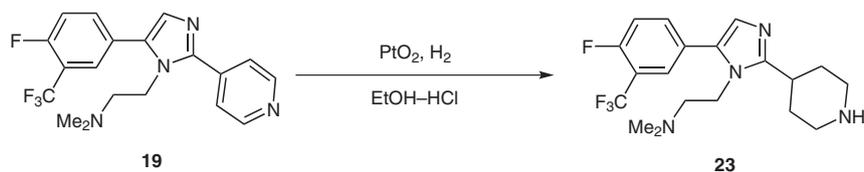
**Table 3** Imidazole **25** N-Alkylation Screening Study and Optimized Conditions Data

Entry	Starting material and base (equiv)		Reaction conditions	In situ analysis of N-alkylation by HPLC <sup>b</sup>					Yield of <b>17</b> (%)		
	<b>15</b>	<b>16</b>		Base	Solvent (vol) <sup>a</sup>	Temp (°C), time (h)	<b>15</b> (%)	<b>17</b> (%)		<b>21</b> (%)	<b>19</b> (%)
1	1	2	NaH (10.0)	THF (20)	63–65, 18	ND	94.1	2.1	1.05	0.8	80
2	1	2	DIPEA (3.5) K <sub>2</sub> CO <sub>3</sub> (3.5)	DMF (20)	63–65, 18	52.4	35.4	0.3	10.1	0.3	NI
3	1	2	<i>t</i> -BuOK (7.0)	THF (20)	63–65, 18	56	–	–	–	–	NI
4	1	2	Cs <sub>2</sub> CO <sub>3</sub> (7.0)	THF (20)	63–65, 18	ND	92.3	1.5	4.3	0.8	76
5	1	2	NaH (6.0)	THF (20)	63–65, 18	ND	82.2	1.8	13.1	0.5	66
6	1	2	powder KOH (7)	THF (20)	63–65, 18	ND	90.3	0.7	5.4	ND	77
7	1	2	flake KOH (7)	THF (20)	63–65, 18	0.03	91.5	1.7	5.42	0.8	75
8	1	2	powder NaOH (3)	DMSO (15)	60–65, 16	79.8	11.0	0.6	7.36	0.9	NI
9	1	2	powder KOH (7)	THF (15)	63–65, 18	ND	89.8	1.4	5.4	0.1	74
10	1	2	powder KOH (6)	THF (15)	63–65, 18	0.7	90.5	1.6	5.5	ND	74
11	1	2	powder KOH (5)	THF (15)	63–65, 18	0.7	87.3	1.9	7.9	0.4	73
12	1	2	powder KOH (4)	THF (15)	63–65, 18	46.4	30.4	2.6	15.7	1.7	NI
13	1	2	powder KOH (3)	THF (15)	63–65, 18	74.1	1.2	1.1	20.6	0.7	NI
14	1	2	powder KOH (6)	THF (15)	63–65, 18	0.6	92.1	1.0	6.4	0.6	77
15	1	2	Cs <sub>2</sub> CO <sub>3</sub> (7.0)	THF (20)	63–65, 18	ND	91.2	0.6	4.8	0.3	74
16	1	2	powder KOH (6)	THF (15)	63–65, 18	0.6	92.1	1.0	6.4	0.6	77
17	1	2	powder KOH (6)	THF (15)	63–65, 18	0.8	89.1	0.6	6.7	0.3	77
1	1	2	NaH (10.0)	THF (20)	63–65, 18	ND	94.1	2.1	1.1	0.8	78

<sup>a</sup> Reaction volume is based upon the quantity of **15** (g) used in the reaction (as explained in footnote a of Table 2).

<sup>b</sup> ND: not detected.

<sup>c</sup> Isolated yield. NI: not isolated.

Scheme 6 Synthesis of **23**Table 4 Hydrogenation of **17** Using PtO<sub>2</sub> Catalyst

Entry	Starting material and conditions					Temp (°C), time (h)	In situ analysis by HPLC ( <b>18</b> ), (%)	Yield (%) <sup>a</sup>
	<b>17</b> (g)	EtOH (L)	Catalyst	Concd HCl (mL)	H <sub>2</sub> (MPa)			
1	36	0.50	PtO <sub>2</sub> (5%)	21	1.4–1.6	48–52, 28	96.9	91
2	200	2.8	PtO <sub>2</sub> (2.5%)	105	1.4–1.6	48–52, 22	98.9	92

<sup>a</sup> Isolated yield.

However, we were successful in catalytic reduction of **17** using PtO<sub>2</sub> (2.5%) in ethanol under acidic conditions (Table 4).<sup>14</sup> Removal of catalyst by filtration followed by neutralization of reaction solution with sodium hydroxide provided **18** in good yield with excellent purity. In a similar fashion, isomer **23** was also prepared via hydrogenation of **19** (Scheme 6); the latter was obtained by chromatographic purification.

In conclusion, we have described a practical synthesis of 2-{4-[4-fluoro-3-(trifluoromethyl)phenyl]-2-(piperidin-4-yl)-1H-imidazol-1-yl}-N,N-dimethylethanamine.

<sup>1</sup>H, <sup>19</sup>F, and <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub> using a Bruker Avance 400 MHz spectrometer. Chemical shifts ( $\delta$ ) are expressed in ppm downfield relative to TMS (0 ppm) and coupling constants ( $J$ ) are given in Hz. Standard abbreviations are used to describe the signal patterns. High-resolution mass spectra (HRMS) were obtained using a Waters GCT Premier TOF mass spectrometer with EI source.

#### N-{2-[4-Fluoro-3-(trifluoromethyl)phenyl]-2-oxoethyl}isonicotinamide Hydrochloride (**15**)

The tosylate salt **3** (392 g, 1.00 mole) was combined with isonicotinic acid (135 g, 1.09 mol) and THF (5.5 L). The resultant slurry was cooled to 0–5 °C. T<sub>3</sub>P (50% in EtOAc; 1001 g, 1.57 mol) diluted in THF (0.75 L) and Et<sub>3</sub>N (510 g, 4.94 mol) diluted in THF (0.75 L) were added concurrently over 0.5 h to give a yellow, homogeneous solution (addition of T<sub>3</sub>P solution was slightly faster than Et<sub>3</sub>N). The reaction was quenched with brine (3 L) and the aqueous layer was back-extracted with EtOAc (0.5 L). The combined organic layers were washed with brine (2 × 3 L), dried (MgSO<sub>4</sub>), filtered, and concentrated to give a yellow oil. The crude oil was dissolved in EtOAc (3 L), cooled to 5–10 °C, and HCl gas was bubbled in for 1 h. The resultant slurry was stirred at 15 °C for 2 h. The mixture was concentrated to 1.5 L, and MeOH (1.5 L) was added. The slurry obtained was stirred at r.t. for 6 h. The material was filtered and dried under vacuum; yield: 280 g (77%); yellow solid; purity: 99.2%; mp 117–119 °C.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 4.87 (d,  $J$  = 5.3 Hz, 2 H), 7.73 (t,  $J$  = 9.7 Hz, 1 H), 7.78 (d,  $J$  = 5.7 Hz, 2 H), 8.35 (d,  $J$  = 7.0 Hz, 1 H), 8.43 (m, 1 H), 8.75 (d,  $J$  = 6.2 Hz, 2 H), 9.3 (t,  $J$  = 5.3 Hz, 1 H).

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 193.5, 165.6, 160.8, 150.8, 141.1, 136.0, 135.9, 132.3, 127.8, 121.7, 118.7, 118.4, 47.1.

<sup>19</sup>F NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = –60.3, –108.6.

HRMS:  $m/z$  calcd for C<sub>15</sub>H<sub>10</sub>F<sub>4</sub>N<sub>2</sub>O<sub>2</sub>: 326.0678; found: 326.0668.

#### 4-{4-[4-Fluoro-3-(trifluoromethyl)phenyl]-1H-imidazol-2-yl}pyridine Acetic Acid Salt (**15**)

Compound **15** (100 g, 0.276 mol) was combined with NH<sub>4</sub>OAc (758 g, 9.83 mol) and AcOH (1 L) and the mixture was heated in an autoclave at 140 °C for 20 h. The mixture was concentrated to remove AcOH, and H<sub>2</sub>O (2 L) was added. The mixture was stirred at 20 °C for 2 h. The material was filtered and washed with H<sub>2</sub>O (0.5 L). The crude product was purified by reslurrying in EtOAc (0.5 L) and heptanes (1 L). The material was filtered and dried under vacuum; yield: 228 g (89%); brown solid; purity: 99.4%; mp 233–235 °C.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 1.89 (s, 3 H), 7.54 (t,  $J$  = 9.7 Hz, 1 H), 7.92 (d,  $J$  = 6.2 Hz, 2 H), 8.10 (s, 1 H), 8.22 (d,  $J$  = 5.7 Hz, 2 H), 8.65 (d,  $J$  = 5.7 Hz, 2 H), 11.95 (s, 1 H), 13.22 (s, 1 H).

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 172.4, 159.2, 156.7, 150.8, 144.3, 140.0, 137.3, 131.9, 131.1, 123.0, 119.4, 118.1, 118.0, 117.4, 21.5.

<sup>19</sup>F NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = –60.0, –119.4.

HRMS:  $m/z$  calcd for C<sub>17</sub>H<sub>13</sub>F<sub>4</sub>N<sub>3</sub>O<sub>2</sub>: 367.0944; found: 367.0941.

#### 2-{4-[4-Fluoro-3-(trifluoromethyl)phenyl]-2-(pyridin-4-yl)-1H-imidazol-1-yl}-N,N-dimethylethanamine (**17**)

The imidazole **15** (192 g, 0.523 mol) was combined with KOH (176 g, 3.14 mol) and THF (3 L). The mixture was heated to 30–40 °C and stirred for 1 h. 2-Chloro-*N,N*-dimethylethanamine hydrochloride (**16**; 150 g, 1.04 mol) was added and the reaction mixture was heated at 60–65 °C for 20 h. The mixture was filtered and the cake washed with EtOAc (0.4 L). Activated carbon (30 wt%) was added to the filtrate and the mixture was stirred at 60–65 °C for 3 h. The mixture was filtered through Celite, and the cake rinsed with EtOAc (0.8 L). The filtrate was concentrated to a brown solid (175 g). The crude material was slurried in EtOAc (0.1 L) and heptanes (1 L) at r.t. for 16 h. The material was filtered and dried under vacuum; yield: 151.8 g (77%); yellow solid; purity: 99.3%; mp 122–126 °C.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 2.07 (s, 6 H), 2.59 (t,  $J$  = 6.5 Hz, 2 H), 4.21 (t,  $J$  = 6.5 Hz, 2 H), 7.53 (t,  $J$  = 10.1 Hz, 1 H), 7.71 (dd,  $J$  = 4.4 Hz, 2 H), 8.12 (br t, 3 H), 8.70 (dd,  $J$  = 4.4 Hz, 2 H).

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = 150.6, 145.3, 138.3, 138.0, 131.7, 131.6, 131.0, 130.9, 123.1, 118.2, 118.0, 59.2, 45.6, 45.5, 40.4.

<sup>19</sup>F NMR (DMSO-*d*<sub>6</sub>):  $\delta$  = –60.1, –119.3.

HRMS:  $m/z$  calcd for C<sub>19</sub>H<sub>18</sub>F<sub>4</sub>N<sub>4</sub>: 378.1468; found: 378.1462.

**2-{[4-[4-Fluoro-3-(trifluoromethyl)phenyl]-2-(pyridin-4-yl)-1H-imidazol-1-yl]-N,N-dimethylethanamine (18)}**

Compound **17** (143 g, 0.378 mol) was charged to an autoclave followed by EtOH (1.1 L). PtO<sub>2</sub> (2.5%) was added, followed by aq 2 N HCl (0.38 L, 0.755 mol). The mixture was stirred at 70 °C under 140 psi H<sub>2</sub> for 24 h. The mixture was filtered through Celite and the cake was rinsed with EtOH (2 × 0.25 L). The filtrate was concentrated to afford a white solid. The crude material was dissolved in H<sub>2</sub>O (1.7 L) and EtOH (0.43 L) and cooled to 0–5 °C. By maintaining a temperature of 0–5 °C, the pH of the mixture was adjusted to 10–11 with aq 2 N NaOH. The resultant slurry was held at 0–5 °C for 2 h. The product was filtered, rinsed with cold EtOH–H<sub>2</sub>O (1:5, 0.8 L), and dried under vacuum (60–70 °C); yield: 149 g (88%); white solid; purity: 99.0%; mp 92–95 °C.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): δ = 1.70 (br q, 4 H), 2.17 (s, 6 H), 2.54 (m, 4 H), 2.81 (m, 1 H), 3.00 (d, *J* = 11.9 Hz, 2 H), 3.99 (t, *J* = 6.6 Hz, 2 H), 7.44 (t, *J* = 10.1 Hz, 1 H), 7.69 (s, 1 H), 8.01 (d, *J* = 7.0 Hz, 2 H).

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>): δ = 32.32, 33.59, 43.08, 45.30, 46.08, 59.45, 116.38, 117.29, 117.40, 121.45, 124.15, 129.88, 132.11, 135.91, 152.21, 155.72, 158.19.

<sup>19</sup>F NMR (DMSO-*d*<sub>6</sub>): δ = –60.0, –120.6.

HRMS: *m/z* calcd for C<sub>19</sub>H<sub>24</sub>F<sub>4</sub>N<sub>4</sub>: 384.1937; found: 384.1929.

**2-{[5-[4-Fluoro-3-(trifluoromethyl)phenyl]-2-(pyridine-4-yl)-1H-imidazol-1-yl]-N,N-dimethylethanamine (19)}**

The active carbon cake containing crude **19** from the preparation of **17** was added to MeOH (1 L) and refluxed for 1 h. The material was filtered and the filtrate was concentrated to obtain 33 g of a red solid, which was purified by preparative HPLC; yield: 1.9 g; white solid; purity 98.2%; mp 157–161 °C.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): δ = 2.18 (s, 6 H), 2.68 (t, *J* = 5.7 Hz, 2 H), 4.38 (t, *J* = 5.7 Hz, 2 H), 7.43 (t, *J* = 9.7 Hz, 1 H), 7.83 (s, 1 H), 8.11 (m, 4 H), 8.42 (d, *J* = 6.6 Hz, 2 H).

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>): δ = 45.01, 55.66, 58.74, 116.42–117.34, 118.47, 121.87, 124.24, 130.08, 132.51, 133.24, 142.75, 147.95, 149.31, 155.49, 157.99.

<sup>19</sup>F NMR (DMSO-*d*<sub>6</sub>): δ = –60.0, –121.2.

HRMS: *m/z* calcd for C<sub>19</sub>H<sub>18</sub>F<sub>4</sub>N<sub>4</sub>: 378.1468; found: 378.1472.

**2-{[5-[4-Fluoro-3-(trifluoromethyl)phenyl]-2-(piperidin-4-yl)-1H-imidazol-1-yl]-N,N-dimethylethanamine (23)}**

Reduction of **19** using method outlined for compound **18** provided **23**; yield: 4.9 g (64%); white solid; purity: 99.6%; mp 166–169 °C.

<sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>): δ = 1.71 (br q, 2 H), 1.84 (br d, *J* = 11.4 Hz, 2 H), 2.00 (d, *J* = 9.7 Hz, 2 H), 2.12 (s, 6 H), 2.33 (m, 4 H), 2.62 (m, 1 H), 2.90 (d, *J* = 11.0 Hz, 2 H), 7.43 (t, *J* = 9.7 Hz, 1 H), 7.65 (s, 1 H), 8.01 (d, *J* = 7.0 Hz, 2 H).

<sup>13</sup>C NMR (DMSO-*d*<sub>6</sub>): δ = 30.78, 35.56, 45.55, 53.47, 56.28, 56.86, 113.25, 117.14, 117.33, 121.45, 121.86, 121.90, 124.15, 130.03, 130.11, 132.37, 136.72, 152.13, 155.68, 158.17.

<sup>19</sup>F NMR (DMSO-*d*<sub>6</sub>): δ = –60.0, –120.7.

HRMS: *m/z* calcd for C<sub>19</sub>H<sub>24</sub>F<sub>4</sub>N<sub>4</sub>: 384.1937; found: 384.1907.

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