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# Photoinduced Copper-Catalyzed Site-Selective C(sp<sup>2</sup>)-C(sp) Cross-Coupling via Aryl Sulfonium Salts

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**ABSTRACT:** The classical Sonogashira reaction of aryl electrophiles in the presence of Pd catalysts has been well established as a potent method for arylalkyne synthesis. However, the site-selective  $C(sp^2)-C(sp)$  cross-coupling strategy using a non-noble-metal catalyst is rare. An efficient alternative approach for the synthesis of arylalkynes via a Cu-catalyzed Sonogashira-type reaction promoted by visible light is described. This method enables site-selective alkynylation from aryl sulfonium salts derived from diverse arenes to a set of arylalkynes with high selectivity and high functional-group compatibility. Moreover, rapid alkynylation of drug molecules is demonstrated.

rylalkyne motifs are prevalent in the synthesis of **A**pharmaceuticals, macrocycles, molecular organic materials, and natural products.<sup>1</sup> Arylalkynes have been prepared from aryl halides via Sonogashira cross-coupling,<sup>2</sup> and the aryl coupling partners have been subsequently extended to diazonium salts,<sup>3a,b</sup> ammonium salts,<sup>3c</sup> hydrazides,<sup>3d</sup> and carboxylic acids<sup>3e</sup> (Figure 1A). Despite these important successes, a considerable challenge in this field is the siteselective installation of these synthetic linchpins (i.e., halogen, amino, hydrazine, and carboxyl groups) into arenes.<sup>4</sup> Moreover, noble-metal salts as catalysts hamper the practicability of this approach. In fact, because of their low cost and low toxicity, base metals such as Cu and Fe have attracted considerable attention as Sonogashira reaction catalysts over the past decade.<sup>5</sup> In these methods, however, the aryl donor is identified as only aryl halides; late-stage alkynylation of complex organomolecules, in particular, remains elusive. Alternatively, the non-noble-metal-catalyzed site-selective Sonogashira reaction with arenes as inexpensive and readily available feedstocks is of great value but has rarely been explored.

In some recent elegant works, Ritter and co-workers have described the one-step conversion of certain simple arenes into aryl sulfonium salts, which are robust arylating reagents for coupling with a wide range of functional-group precursors.<sup>6</sup> In this method, various functional groups are efficiently incorporated into simple arenes. More importantly, this approach enables the late-stage functionalization of complex molecules and drugs via the aryl sulfonium salts. A related



Figure 1. Reaction development.

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alkynylation reaction with aryl sulfonium salts would complement this strategy and provide an indispensable method for the synthesis of arylalkynes. Notably, the groups of Ritter<sup>4a</sup> and Zhang<sup>7</sup> independently reported the alkynylation of aryl sulfonium salts. However, the use of Pd and heating is inevitable (Figure 1A). We surmised that a method with mild conditions and without Pd as a catalyst would enhance the substrate compatibility and increase the chemoselectivity. Over the past few years, photoredox/Cu cooperative catalysis strategies, which greatly enhance the capability of Cu-catalyzed reactions, have been recognized as an intriguing alternative to cross-coupling reactions.<sup>8</sup> Inspired by these works, we reasoned that aryl sulfonium salts<sup>9</sup> other than aryl halides can be used as versatile electrophiles for dual photoredox/ copper-catalyzed  $C(sp^2)-C(sp)$  bond formation (Figure 1B). Furthermore, the Cu acetylide complexes, upon coordination of Cu<sup>I</sup> salts with acetylenes, will substantially decrease the energy ( $\lambda$  > 350 nm) needed for the photoexcitation and prolong the lifetime (on the order of microseconds) of the excited intermediates,<sup>10</sup> resulting in sufficient reducing ability  $[E_{1/2} = -1.77 \text{ V} \text{ vs the saturated calomel electrode (SCE)}]^{10c}$ to reduce the aryl sulfonium ( $E_{1/2} = -1.05$  V vs the SCE).<sup>4b</sup> A proposed reaction pathway is depicted in Figure 2. Initially, the



**Figure 2.** Possible pathways for photoinduced Cu-catalyzed Sonogashira cross-coupling reaction.

coordination of the terminal alkyne to a  $Cu^{I}$  species (A) generates Cu acetylide intermediate B. The excitation of intermediate B generates a photoexcited complex C, which reduces the aryl sulfonium salt through a single-electron transfer (SET) process, affording a  $Cu^{II}$  complex (D). Finally, D is expected to undergo reductive elimination to give the arylalkyne and regenerate A, closing the catalytic cycle.

The generation of a Cu acetylide intermediate is the key step in this reaction, and related control experiments have been conducted. First, 4-methylphenyl-copper acetylide 2g' was used as a coupling partner for this reaction (Figure 3A). The desired product **3ag** was obtained in good yield. After that, the



Figure 3. Control experiments.

coupling of aryl sulfonium salt 1a with alkyne 2k was investigated by employing 2g' as the catalyst. Two different products were observed (1:3.5 3ag:3ak; ratio determined by NMR), which provides better support for an on-cycle copper acetylide species (Figure 3B).

Table 1. Reaction Optimization<sup>a</sup>

PhO-TT*E 1a 0.2 mmol	$3F_4^{-} + \underbrace{\qquad}_{2a} \underbrace{\qquad}_{1.5 \text{ equiv}}^{\text{CuCl (10 mol%)}} F_4^{-} \underbrace{\qquad}_{2*7W \text{ blue LEDs}}^{\text{CuCl (10 mol%)}} F_4^{-}$	0- <b>C</b> -Br 3aa
entry	variation from the "standard" condition	ons yield (%)
1	none	81
2	no CuCl	0
3	no light	0
4	no K <sub>2</sub> CO <sub>3</sub>	0
5	MeCN instead of DMSO	32
6	DMF instead of DMSO	25
7	DBU instead of K <sub>2</sub> CO <sub>3</sub>	43
8	K <sub>3</sub> PO <sub>4</sub> instead of K <sub>2</sub> CO <sub>3</sub>	73
9	CuBr instead of CuCl	65
10	CuI instead of CuCl	72

<sup>a</sup>Reaction conditions: 1a (0.2 mmol), 2a (0.3 mmol), CuCl (10 mol %), and  $K_2CO_3$  (0.6 mmol) in DMSO (2.0 mL) at 25 °C, irradiation with 2  $\times$  7 W blue LED bulbs under  $N_2$  for 20 h. Yields of isolated products.

We investigated the Sonogashira reaction of aryl sulfonium salt 1a with 1-Br-4-ethynylbenzene 2a in the presence of a Cu salt as a catalyst under blue light-emitting diode (LED) irradiation. The optimal reaction conditions were determined (Scheme 1; see Tables S1-S7). The desired arylalkyne 3aa was obtained in good yield under the standard conditions [81% (Table 1, entry 1)]. Control reactions established that the Cu catalyst, base, and light were all essential for the coupling (Table 1, entries 2-4). When dimethyl sulfoxide (DMSO) was replaced with other solvents, the desired product was obtained in lower yields (Table 1, entries 5 and 6; see Table S2).  $K_2CO_2$  was identified as the best base among those investigated (Table 1, entries 7 and 8; see Table S3). Among various Cu salts, CuCl was the best choice (Table 1, entries 9-11; see Table S4). We considered that the ligand may affect the activity of the Cu acetylate complexes and therefore screened different commercially available ligands. The corresponding results show that the use of ligands had little effect on the reaction; only the use of a Cu catalyst (CuCl; no ligand co-additive) as the catalyst was a better choice (see Table S5).

With the optimized reaction conditions in hand, we explored a wider variety of aryl sulfonium salts (Scheme 1, 1b-1s) as the electrophiles. The use of aryl sulfonium salts instead of aryl halides as electrophiles rendered the current Sonogashira coupling method amenable to diverse arenes. Functional groups such as halogen (3da), allyl (3ga), ester (3ha), and cyano (3ia) were all tolerated well. Monosubstituted aryl sulfonium salts (1b-1e) underwent the cross-coupling reaction with 1-Br-4-ethynylbenzene 2a to give the desired products in good yields. In addition, challenging substrates that bear two or three functional groups on their aromatic rings (1f-1j) were also suitable substrates. The corresponding products were obtained in moderate to good yields (3fa-3ja). Highly symmetrical alkynes are an important set of building pubs.acs.org/OrgLett

Letter





<sup>*a*</sup>Reaction conditions: 1a-1s (0.2 mmol), 2a-2m (0.3 mmol), CuCl (10 mol %), and K<sub>2</sub>CO<sub>3</sub> (0.6 mmol) in DMSO (2.0 mL) at 25 °C, irradiation with 2 × 7 W blue LED bulbs under N<sub>2</sub> for 20 h. Yields of isolated products. <sup>*b*</sup>CuCl (30 mol %). <sup>*c*</sup>With 0.4 mmol of terminal alkyne (2n-2p).

blocks in organic synthesis. 1-Ethynyl-4-methoxybenzene 2h was selected as an alkyne coupling partner to react with 4methoxy aryl sulfonium salt 1b, and 1,2-bis(4-methoxyphenyl) ethyne 3bh was obtained with ease. 4-Alkyl aryl sulfonium salts were also obtained in good yields (3kh-3mh). The protocol could also be extended to the late-stage alkynylation of drugs (3na-3sa), which was difficult to accomplish otherwise, demonstrating the potential of this work for the rapid modification of pharmaceuticals.

Subsequently, the coupling of 4-phenoxy aryl sulfonium salt 1a with a series of terminal alkynes was evaluated, and the results were summarized in Scheme 1. Cu acetylide was regarded as the key catalyst of this reaction according to the possible pathways described in Figure 2. During the experi-

ment, we found that changing the alkyne affected the activity of the Cu acetylide intermediate such that the aryl sulfonium salt could not be completely converted in some cases. Fortunately, this problem was solved well by simply increasing the amount of CuCl to 30%. The reactions with phenyl-substituted acetylenes containing diverse electronic properties at the *para* position (2a-2g) furnished the corresponding arylalkynes in moderate to good yields (40-83%). Notably, 1-Br-4ethynylbenzene and 1-I-4-ethynylbenzene (2a and 2d,respectively), which are incompatible with the conventional approach, proceeded well in this coupling, demonstrating its excellent chemoselectivity. The yields of the desired products were maintained when the methoxy group was present at the *meta* or *ortho* position of the phenyl (3aj or 3ak, respectively). To our delight, the coupling was successful for terminal alkynes with other groups such as thiophene, pyridine, alkyl, and substituted alkyl groups (2l-2p).

To further explore the prospective use of this methodology, we carried out the scale-up synthesis of arylalkynes. As shown in Scheme 2A, **3aa** was obtained in 78% yield when 1.0 mmol

## Scheme 2. Scale-up and Transformations

A) Scale-up experiment



aryl sulfonium salt **1a** was treated with 1-Br-4-ethynylbenzene **2a** in the presence of 10 mol % CuCl and 3.0 equiv of  $K_2CO_3$ . Arylalkynes make up an important class of building blocks in organic synthesis.<sup>11</sup> We therefore examined the applications of arylalkynes in macrocyclic construction. Isoquinolinium salt **4bh** was obtained in 70% yield by treatment of arylalkyne **3bh** with 2-phenylpyridine through oxidative C–H activation and annulation (Scheme 2B). Next,  $\alpha$ -pyrone derivative **4ap** was delivered in 78% yield via Ir-catalyzed electrochemical vinylic C–H annulation of 2-phenylacrylic acid with arylalkyne **3ap** (Scheme 2C).

In summary, the photoinduced site-selective Cu-catalyzed Sonogashira coupling of aryl sulfonium salts with terminal alkynes was demonstrated. This coupling method enables rapid transformation of readily available arenes into arylalkynes, which are important building blocks in synthetic chemistry. The successful alkynylation of arenes, many of which contain sensitive functional groups (e.g., alkene, ester, cyano, or heterocyclic groups), underscores the high chemoselectivity and functional-group tolerance of this method. Moreover, the site-selective alkynylation of drug molecules was realized. This work substantially expands the scope of the metal-catalyzed  $C(sp^2)-C(sp)$  coupling reaction and provides a facile synthesis strategy for the construction of diversely functionalized arylalkynes.

### ASSOCIATED CONTENT

#### Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.0c02364.

Experimental procedures, synthesis of the starting materials, and compound characterization data (PDF)

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## Notes

The authors declare no competing financial interest.

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