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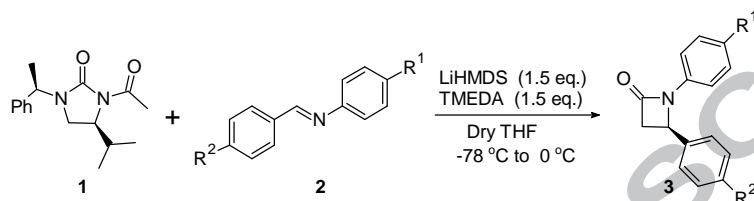
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## Graphical Abstract

**A highly efficient stereoselective synthesis of  $\beta$ -lactams**

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## A highly efficient stereoselective synthesis of $\beta$ -lactams

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### ABSTRACT

An efficient strategy for a one-pot, single step synthesis of  $\beta$ -lactams employing an imidazolidinone based chiral auxiliary with various aldimines *via* asymmetric Mannich-type reaction has been described.

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### Introduction

The azetidin-2-one scaffold, also known as  $\beta$ -lactam, constitutes numerous biologically important drug molecules and natural products.<sup>1</sup> Besides  $\beta$ -lactam antibiotics, stereoselective construction of the  $\beta$ -lactam core generates considerable research interest as versatile building blocks for the synthesis of numerous organic molecules.<sup>2</sup> Examination of the literature offers various approaches<sup>3</sup> for the construction of the  $\beta$ -lactam core such as Staudinger reaction,<sup>4</sup> enolate-imine condensation,<sup>5</sup> carbene insertion reaction,<sup>6</sup> Kinugasa reaction<sup>7</sup> and asymmetric catalysis.<sup>3b</sup>

Most of these approaches are associated with drawbacks such as poor yields, multiple steps and low regio- and stereoselectivity.<sup>8</sup> Considering these drawbacks and based on our keen interest<sup>9</sup> in the asymmetric synthesis of bioactive molecules, we decided to synthesize  $\beta$ -lactams using a chiral auxiliary mediated approach. The present research discloses an efficient auxiliary mediated strategy for the stereoselective synthesis of these molecules in a single step. The chiral auxiliary employed in this study was synthesized by a procedure developed in our laboratory (Scheme 1).<sup>9b</sup> To standardize the reaction conditions, a model reaction was studied in which kinetic deprotonation of *N*-acetyl auxiliary **1** (1.0 equiv.) was performed using LiHMDS (1.0 equiv.) in dry THF at -78 °C (Scheme 1) followed by addition of imine electrophile **2a** (1.05 equiv.) at the same temperature. The reaction failed to afford the alkylated product. Very slow progress of the reaction was witnessed when the reaction temperature was increased to 0 °C.

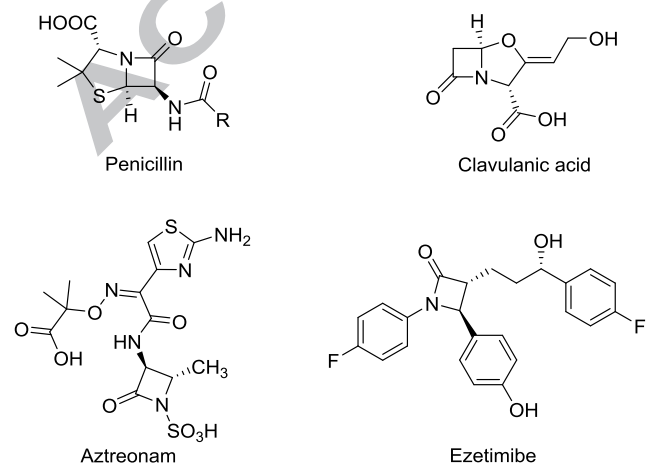
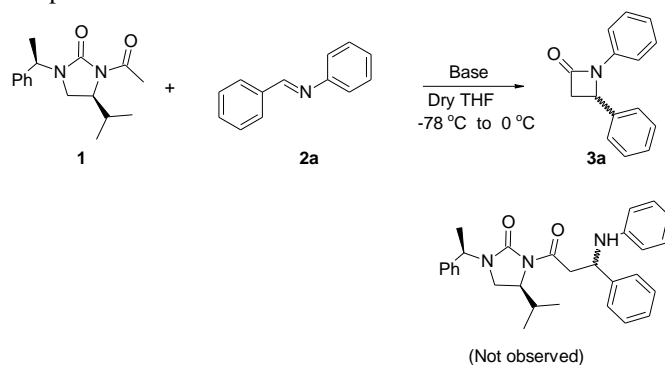


Figure 1. A few biologically active molecules containing a  $\beta$ -lactam ring



Scheme 1. Synthesis of  $\beta$ -lactam **3a** using different bases

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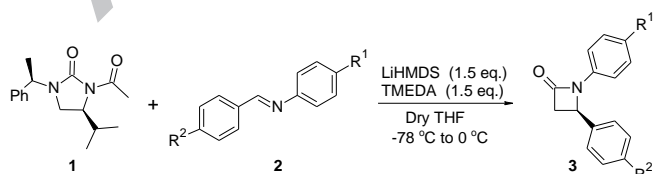
Quite surprisingly the reaction resulted in the formation of a  $\beta$ -lactam in low yield (Table 1) instead of the expected alkylated adduct. Increase in temperature above 0 °C led to degradation of the desired product. A study to optimize the stoichiometry of LiHMDS indicated 1.5 equivalents to be the best, as the yields did not improve at increased loadings. A similar result was obtained when the reaction was attempted with LDA (Table 1; entry 4). Contrary to this, other bases such as *n*-BuLi, NaHMDS and KHMDS gave very sluggish reactions and did not afford the desired product (Table 1; entries 5-7). The available literature reports<sup>10</sup> and our previous experience<sup>9c</sup> strongly support the utility of the additive TMEDA as a cation chelator. Thus the hexamethyldisilylamide anion would be sufficiently basic for proton abstraction. Further attempts were made to enhance the yield of the desired  $\beta$ -lactam by using TMEDA as an additive. In a separate reaction, addition of equimolar amounts (1.5 equiv.) of LiHMDS and TMEDA to a solution of the acetylated auxiliary **1** in THF at -78 °C followed by introduction of the imine electrophile **2a** resulted in improvement of the yield from 40% to 65% (Table 1; entry 8). On the contrary a similar reaction with equimolar amounts (1.5 equiv.) of LDA and TMEDA afforded only 40% yield of the desired  $\beta$ -lactam (Table 1; entry 9).

**Table 1.** Optimization of reaction conditions<sup>a</sup>

Entry	Reagent	Equivalent	Solvent	Yield (%)
1	LiHMDS	1.0	THF	35
2	LiHMDS	1.5	THF	40
3	LiHMDS	2.0	THF	30
4	LDA	1.5	THF	30
5	NaHMDS	1.5	THF	nd
6	KHMDS	1.5	THF	nd
7	<i>n</i> -BuLi	1.5	THF	nd
8	LiHMDS, TMEDA	1.5	THF	65
9	LDA, TMEDA	1.5	THF	40

<sup>a</sup> **1** (1.0 equiv.), **2a** (1.05 equiv.), Temp. -78 °C to 0 °C.

The developed protocol was further extended to different aromatic imines bearing electron donating and withdrawing groups to generalise the scope of the reaction (Scheme 2). The reactions afforded the  $\beta$ -lactam derivatives with ease and good yield, providing excellent stereoselectivity (Table 2). Contrary to this, the imines derived from aliphatic aldehydes and/or aliphatic amines (Scheme 3) did not afford the desired product<sup>11</sup> (Table 3).



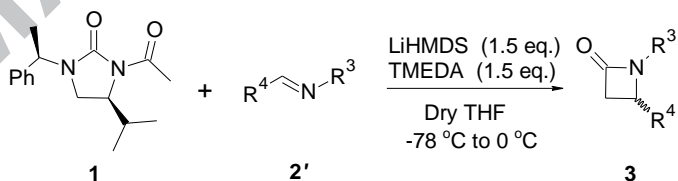
**Scheme 2.** Synthesis of aryl substituted  $\beta$ -lactams

The stereoselectivity of the developed protocol was established by comparing the specific rotation of the synthesized  $\beta$ -lactam **3n** with the reported<sup>5b</sup> value [observed  $[\alpha]_D^{25} = -89.5$  (c 0.11, CHCl<sub>3</sub>), reported for the antipode  $[\alpha]_D = +87.8$  (c 0.115, CHCl<sub>3</sub>)] and was found to be in good agreement. The enantiomeric excess of all the synthesized compounds were evaluated by chiral HPLC

**Table 2.** Synthesis of aryl substituted  $\beta$ -lactams<sup>a</sup>

Entry	R <sup>1</sup>	R <sup>2</sup>	Product	Yield (%) <sup>b</sup>	ee <sup>c</sup>
1	H	H	<b>3a</b>	65	90:10
2	H	F	<b>3b</b>	62	>99
3	H	Cl	<b>3c</b>	63	100
4	H	Br	<b>3d</b>	64	100
5	F	H	<b>3e</b>	63	85:15
6	OCH <sub>3</sub>	H	<b>3f</b>	60	100
7	H	OCH <sub>3</sub>	<b>3g</b>	61	100
8	CH <sub>3</sub>	H	<b>3h</b>	62	93:07
9	CH <sub>3</sub>	OCH <sub>3</sub>	<b>3i</b>	61	100
10	OCH <sub>3</sub>	OCH <sub>3</sub>	<b>3j</b>	60	100
11	CH <sub>3</sub>	CH <sub>3</sub>	<b>3k</b>	63	88:12
12	CH <sub>3</sub>	Br	<b>3l</b>	62	78:22
13	F	OBn	<b>3m</b>	63	100
14	F	CH <sub>3</sub>	<b>3n</b>	64	78:22

<sup>a</sup> **1** (1.0 equiv.), LiHMDS (1.5 equiv.), TMEDA (1.5 equiv.), **2** (1.05 equiv.), THF, Temp. -78 °C to 0 °C, <sup>b</sup>Isolated yield, <sup>c</sup>determined by chiral HPLC using Chiralcel OD-H column; *n*-Hex / *i*-PrOH = 95:5, 1.0 mL/min, 254 nm.



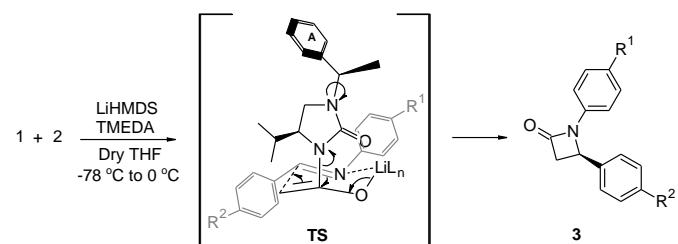
**Scheme 3.** Synthesis of alkyl substituted  $\beta$ -lactams

**Table 3.** Synthesis of alkyl substituted  $\beta$ -lactams

Entry	R <sup>3</sup>	R <sup>4</sup>	Product	Yield
1	Pr	Ph	<b>3o</b>	Nil
2	Ph	Pr	<b>3p</b>	Nil
3	C <sub>6</sub> H <sub>11</sub>	Pr	<b>3q</b>	Nil

<sup>a</sup> **1** (1.0 equiv.), LiHMDS (1.5 equiv.), TMEDA (1.5 equiv.), **2'** (1.05 equiv.), THF, Temp. -78 °C to 0 °C.

analysis (Table 2). The observed stereoselectivity in acetate alkylation reaction may be envisaged by the proposed transition state model (TS). Deprotonation of the *N*-acetyl auxiliary at -78 °C using LiHMDS/TMEDA affords the *Z*-enolate which upon reaction with an imine electrophile would lead to a six-membered chelated transition state (Scheme 4).



**Scheme 4.** Proposed transition state for the stereoselective synthesis of  $\beta$ -lactams

This would give rise to  $\beta$ -lactam **3** as the major product. Further the stoichiometric use of TMEDA increases the basicity of the generated enolate and assists in the *in situ* formation of  $\beta$ -lactam in good yield and stereoselectivity.

## Conclusion

The present work describes a chiral auxiliary mediated highly efficient, single step synthesis of  $\beta$ -lactams stereoselectively. The developed protocol afforded the desired products with good yield and high selectivity.

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## Supplementary data

Supplementary material which includes experimental procedures and compound data can be found in the online version.

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**Highlights**

- Chiral auxiliary mediated asymmetric Mannich-type reactions with aldimines.
- One-pot, single step synthesis of  $\beta$ -lactams.
- Transition state model for the formation of  $\beta$ -lactams.
- Good overall yield and high enantioselectivity.

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